



SYNTHESIS AND ANTITUBERCULAR EVALUATION OF CERTAIN PYRROLE DERIVATIVES

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ABSTRACT

A series of 2-aryl-5-[4-(1*H*-pyrrol-1-yl)phenyl]-1,3,4-oxadiazole derivatives (**VIa-g**) have been synthesized in good yields. These compounds were synthesized with an approach to reduce the growing anti-tubercular resistance and to develop more potent and less side effects having antitubercular activity. The reaction of 2,5-dimethoxytetrahydro furan with 4-aminobenzoate (**II**) in presence of ethanol, which yields ethyl 4-pyrrol-1-yl benzoate (**III**). This ethyl-4-pyrrol-1-yl benzoate (**III**) on reaction with hydrazine hydrate produced 4-pyrrol-1-yl benzoic acid hydrazide (**IV**). This carbohydrazide on treatment with different substituted benzoyl chlorides gave intermediates (**Va-g**), which on cyclisation with P₂O₅ in the presence of DMF yielded of 2-aryl-5-[4-(1*H*pyrrol-1-yl)phenyl]-1,3,4-oxadiazoles **VI(a-g)**. Structure of newly synthesized pyrrole derivatives were confirmed on the basis of physicochemical and spectral data (IR, ¹H-NMR, ¹³C-NMR and Mass spectra). All the synthesized compounds were screened for their antitubercular activity using Microplate Alamar Blue Assay (MABA) method. Compounds showed anti-tubercular activity at MIC values between 1.6 to 12.5 µg/ml, compounds **VIb**, **VIc** and **VI d** showed highest activity of 1.6 µg/ml. Selected compounds were evaluated for anti-bacterial activity against gram positive (*S. aureus*) and gram negative (*E. coli*). Compounds showed moderate activity. Physicochemical properties of the selected compounds satisfied with the Lipinski's rule of 5.

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INTRODUCTION

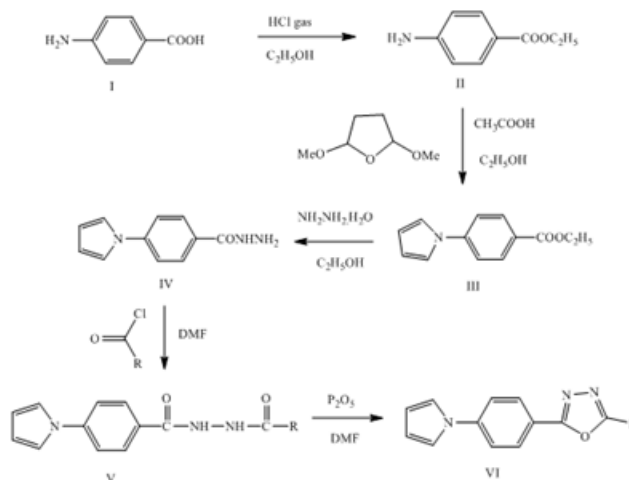
Tuberculosis (TB), a disorder, caused by *Mycobacterium tuberculosis* (*M. tuberculosis*), is the second leading disease causing the death along with HIV [1]. In the first update on the tuberculosis since the UN General Assembly High Level meeting on Sep 26, 2018, in which All UN Member States committed to end the global TB epidemic by 2030, WHO published the Global Tuberculosis Report on Oct 17, 2019. The report details data from 202 countries and territories that account for more than 99% of the world's population and estimated the number of TB cases. Ahead of first 2020 milestone, TB still accounts for the highest mortality from any infectious diseases worldwide, even surpassing HIV/AIDS, causing 1.5 million deaths in 2018. According to WHO's report, the current pace of change is not enough: the global cumulative rate of reduction for TB incidence was only 6.3% between 2015 and 2018, which is much less than that of the 2020 milestone of 20% [2]. TB is a major threat; killing about 2 million people each year. The World Health Organisation estimates that 1 billion people will be newly infected in the period 2000-2020, resulting in 35 million more deaths. TB is currently the leading killer of youths, women and AIDS patients in the world [3].

Pyrrole is an important heterocycle of the plant and animal kingdom as a subunit of chlorophyll in plant cells, hemin and vitamin B₁₂ in animal cells. First isolated in 1857 from the product of bone pyrolysis, it showed biological activities that are characteristic of haemoglobin [4]. Earlier, pyrrole derivatives have shown *in vitro* antitubercular activity [5,6] and recently, much of the research was carried out on anti-TB drug design using pyrrole as templates for the synthesis [7,8] including molecular modelling along with laboratory investigations [8,9]

Experimental

All the chemicals used in the synthetic experiment were purchased from Sigma-Aldrich, S. D. Fine-Chem Limited and spectrochem Pvt. Ltd. Solvent used were of reagent grade and they were purified and dried by standard methods. Melting points were determined using the Shital-digital programmable melting point apparatus and are uncorrected. FTIR spectra in KBr pellets were recorded on a Bruker FTIR spectrophotometer. The ¹H spectra were recorded on a Bruker AVANCE II at 400 MHz, chemical shifts are expressed in parts per million (*ppm*) relative to TMS. The abbreviations used to describe the peak patterns are: (s) singlet, (d) doublet, (t) triplet and (m) multiplet. Mass spectra (MS) were recorded in a JEOL GCMATE II GC – Mass spectrometer and Shimadzu QP 20105 GC-Mass spectrometer. Analytical thin layer chromatography (TLC) sheets of silica gel 60F254 (Merck, Darmstadt, Germany) visualized by long- and short- wavelength ultraviolet (UV) lamps.

SCHEME



R = 3-Br C₆H₄, 4-F C₆H₄, 3-F C₆H₄, 2-F C₆H₄, 2-Cl C₆H₄, 3-Cl C₆H₄, 4-Cl C₆H₄

Synthesis of ethyl 4-pyrrol-1-yl benzoate (III):

2,5-Dimethoxytetrahydrofuran (16 g, 0.12 mol) was added to ethyl 4-aminobenzoate (16.5 g, 0.1 mol) in ethanol (100 ml) and this mixture was heated at reflux in presence of catalytic amount of acetic acid (0.05 ml) for 3 hr. After removal of solvent, product is filtered and recrystallized from ethanol.

Synthesis of 4-pyrrol-1-yl benzoic acid hydrazide (IV):

Compound IV was synthesized by refluxing a mixture of ethyl 4-pyrrol-1-yl benzoate 3.22 g, 0.015 mol) with hydrazine hydrate (10ml) in absolute ethanol (10ml) for 3 hr. The reaction mixture was cooled and crystalline mass obtained was recrystallized from ethanol to secure yellow crystals.

General procedure for the synthesis of N-(aryl carbonyl)-4-(1H-pyrrol-1-yl)benzhydrazide derivatives (Va-g)

A mixture of 4-pyrrol-1-yl benzoic acid hydrazide 1mmol and substituted acyl chlorides 1.5 mmol in 10 ml of dry DMF was refluxed for 7-8 hr (monitored by TLC). The reaction mixture was cooled and slowly quenched onto crushed ice with stirring and neutralized with saturated sodium bicarbonate solution. The solid which separated was filtered, washed with cold ethanol, dried and recrystallized from aqueous DMF to give the products.

General procedure for the synthesis of 2-aryl-5-[4-(1H-pyrrol-1-yl) phenyl]-1,3,4-oxadiazole derivatives (VIa-g):

To a suspension of V(a-g) (0.001 mol) in dry DMF (50 ml), phosphorous pentoxide (2.0 g) was added portion wise with stirring at room temperature and the reaction mixture was refluxed for 4 hr. The reaction mixture was cooled and then filtered. The resulting solid that separated was washed with water, dried and recrystallized from aqueous DMF to give the products.

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(3-bromophenylphenyl)-1,3,4-oxadiazole (VIa)

IR (KBr) ν_{\max} , cm^{-1} : 1644.99 (C=N), 1068.44 (C-O-C), 2918.57 (Ar CH=CH Stre); ^1H NMR (400MHz, δ ppm, DMSO- d_6): 7.50-8.17 (m, 12H's of pyrrole-C₂, C₃, C₄, C₅-H, bridging phenyl- C₂, C₃, C₅, C₆-H and phenyl-C₂, C₄, C₅, C₆-H); ^{13}C NMR (500MHz, δ ppm, DMSO- d_6): 118.45, 119.50, 120.44, 121.70, 126.46, 126.89, 127.41, 130.02, 136.69, 142.04, 165.23, 164.34.

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(4-fluorophenyl)-1,3,4-oxadiazole (VIb)

IR (KBr) ν_{\max} , cm^{-1} : 1604.81 (C=N), 1101.05 (C-O-C), 2924.34 (Ar CH=CH Stre); ^1H NMR (400MHz, δ ppm, DMSO- d_6): 7.36-8.08 (m, 12H pyrrole-C₂, C₃, C₄, C₅-H, bridging phenyl-C₂, C₃, C₅, C₆-H and phenyl-C₂, C₃, C₅, C₆-H); MS (EI, DMSO- d_6): Found = 306.22 [M^+], Calcd = 305.31.

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(3-fluorophenyl)-1,3,4-oxadiazole (VIc)

IR (KBr) ν_{\max} , cm^{-1} : 1647.04 (C=N), 1071.28 (C-O-C), 2923 (Ar CH=CH Stre); ^1H NMR (400MHz, δ ppm, DMSO- d_6): 6.33 (t, 2H pyrrole-C₃, C₄-H), 7.53 (t, 2H pyrrole-C₂, C₅-H) and 7.60-8.03 (m, 8H, bridging phenyl-C₂, C₃, C₅, C₆-H and phenyl- C₂, C₄, C₅, C₆-H).

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(2-fluorophenyl)-1,3,4-oxadiazole (VIId)

IR (KBr) ν_{\max} , cm^{-1} : 1637.46 (C=N), 2922.5 (Ar CH=CH Stre); ^1H NMR (400MHz, δ ppm, DMSO- d_6): 6.33-8.12 (m, 12H pyrrole-C₂, C₃, C₄, C₅-H, bridging phenyl-C₂, C₃, C₅, C₆-H and phenyl-C₃, C₄, C₅, C₆-H).

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(2-chlorophenyl)-1,3,4-oxadiazole (VIe)

R (KBr) ν_{\max} , cm^{-1} : 1642.52 (C=N), 2922.55 (Ar CH=CH Stre), 1182089 (C-O-C).

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(3-chlorophenyl)-1,3,4-oxadiazole (VIIf)

IR (KBr) ν_{\max} , cm^{-1} : 2846.91 (Ar CH=CH Stre), 1686 (C=O), 1094 (C-O-C).

2-(4-(1H-pyrrol-1-yl)phenyl)-5-(4-chlorophenyl)-1,3,4-oxadiazole (VIIf)

IR (KBr) ν_{\max} , cm^{-1} : 2922.41 (Ar CH=CH Stre), 1645.20 (C=O).

Biological activity:**Antitubercular activity**

MIC values were determined for pyrrole derivatives against *M. tuberculosis* H37Rv strain by Microplate Alamar Blue assay (MABA) [10] using INH as the standard drug. The 96 wells plate received 100 μL of Middlebrook 7H9 broth and serial dilution of compounds were made directly on the plate with drug concentrations of 0.2, 0.4, 0.8, 1.6, 3.125, 6.25, 12.5, 25, 50 and 100 $\mu\text{g}/\text{mL}$. Plates were covered, sealed with parafilm and incubated at 37 $^{\circ}\text{C}$ for 5 days. Then, 25 μL of freshly prepared 1:1 mixture of alamar blue reagent and 10% Tween 80 were added to the plate and incubated for 24 hr. A blue color in the well was interpreted as no bacterial growth and pink color was scored as the growth. The MIC was defined as the lowest drug concentration, which prevented the color change from blue to pink. Table 2 reveals antitubercular activity (MIC) data.

Antibacterial activity

MIC determination of the tested compounds was investigated by aside-by-side comparison with norfloxacin and ciprofloxacin against Gram-positive (*S. aureus*) and Gram-negative bacteria (*E. Coli*) by the broth microdilution method [11,12]. Serial dilution of the test compounds and reference drugs were prepared in Muller-Hinton agar. Drugs (10mg) were dissolved in dimethylsulfoxide (DMSO, 1 mL). Further progressive dilutions with the molten Muller-Hinton agar were performed to obtain the required concentrations of 0.2, 0.4, 0.8, 1.6, 3.125, 6.25, 12.5, 25, 50 and 100 mg/mL . The tubes were inoculated with 105 cfu mL^{-1} (colony forming unit/mL) and incubated at 37 C for 18h. MIC was the lowest concentration of the tested compound that yielded no visible growth on the plate. To ensure that solvent had no effect on bacterial growth, a control was performed with the test medium supplemented with DMSO at the same dilutions as used in the experiments and DMSO had no effect on the microorganisms in the concentration studied. Table 3 reveals antibacterial activity (MIC values) data.

RESULT AND DISCUSSION

Synthesis

Synthetic route adapted to obtain the target compounds are depicted in scheme. FTIR, ¹H NMR, ¹³C NMR and mass spectral data are in agreement with the proposed structures of all the synthesized compounds. The physicochemical properties of the newly synthesized compounds are depicted in Table 1.

Table 1. Physical data of the synthesized compounds:

Comp.	R	Yield (%)	Recrystallisation solvent	M.P.(^o C)	Rf *	Molecular Formula	Mol. weight
VIa	3-Br C ₆ H ₄	64	DMF	246-249	0.881	C ₁₈ H ₁₂ BrN ₃ O	287.33
VIb	4-F C ₆ H ₄	76	DMF	252-254	0.843	C ₁₇ H ₁₂ FN ₃ O	305.31
VIc	3-F C ₆ H ₄	66	DMF	263-265	0.804	C ₁₇ H ₁₂ FN ₃ O	305.31
VIId	2-F C ₆ H ₄	56	DMF	250-252	0.854	C ₁₈ H ₁₂ FN ₃ O	305.31
VIe	2-Cl C ₆ H ₄	55	DMF	248-250	0.818	C ₁₈ H ₁₂ ClN ₃ O	321.76
VIIf	3-Cl C ₆ H ₄	69	DMF	233-235	0.846	C ₁₈ H ₁₂ ClN ₃ O	321.76
VIg	4-Cl C ₆ H ₄	70	DMF	251-255	0.803	C ₁₈ H ₁₂ ClN ₃ O	321.76

All the synthesized compounds of pyrrolyl oxadiazole derivatives (**VIa-g**) were obtained in a good yield and their structures were established by NMR spectral analysis, FTIR spectra of a compound **VIa** showed the presence of C=N peak at 1644.99 cm⁻¹ and Ar CH=CH stretch at 2918.57 cm⁻¹, further the structure of the compound **VIa** was confirmed by ¹H NMR spectra which showed the presence of 12 hydrogens of pyrrole-C₂, C₃, C₄, C₅-H, bridging phenyl- C₂, C₃, C₅, C₆-H and phenyl-C₂, C₄, C₅, C₆-H, as multiplet at 7.50-8.17 ppm. ¹³C NMR spectra showed a peak at 118.45 ppm for pyrrole C₃, C₄, 119.50 ppm for bridging phenyl C₂, C₆, 120.44 pyrrole C₂, C₅, 121.70 for bridging phenyl C₄, 126.46 for bridging phenyl C₃, C₅, 126.89 for phenyl C₅, 127.41 for phenyl C₂, 130.02 for phenyl C₁, 136.69 for phenyl C₃, C₆, 142.04 for bridging phenyl C₁, for 165.23 oxadiazole-C₅ and 164.34 oxadiazole-C₂.

Biological activity

Antitubercular Activity

All the synthesized compounds (**VIa-g**) were screened against *M. tuberculosis* H37Rv strain which showed MIC value ranging from 1.6-12.5 µg/ml. Among all the screened compounds, compounds **VIb**, **VIc**, **VIId** showed highest activity at MIC value of 1.6 µg/ml. The compounds **VIe**, **VIIf**, **VIg** showed antitubercular activity at MIC value of 12.5 µg/ml. The compounds **VIb**, **VIc**, **VIId** showed antitubercular activity at MIC value of 3.25 µg/ml respectively (Table 2).

Antibacterial Activity

The results of antimicrobial activities (expressed in MIC) of the compounds against Gram-positive and Gram-negative bacteria are illustrated in Table 2. The activity of ciprofloxacin is used for comparison. Pyrrolyl oxadiazole derivatives have shown antibacterial activity between MIC of 25-50 µg/ml. Compound **VIc** showed highest activity against Gram-positive bacteria with MIC value of 25 µg/ml. Compound **VIb** showed highest activity against Gram-negative bacteria with MIC value of 25 µg/ml. The compound screened for antibacterial activity against gram-positive (*S. aureus*) and gram negative bacteria (*E. coli*) using broth microdilution method.

Table 2: In vitro antitubercular and antibacterial activity results of newly synthesized compounds.

Comp.	In vitro antitubercular activity MIC value (µg/ml)	In vitro antibacterial activity MIC value (µg/ml)	
		Gram +ve (<i>S. aureus</i>)	Gram -ve (<i>E. coli</i>)
VIa	3.25	--	--
VIb	1.6	50 µg/mL	25 µg/mL
VIc	1.6	25 µg/mL	50 µg/mL
VIId	1.6	--	--
VIe	12.5	--	--
VIIf	12.5	--	--
VIg	12.5	--	--
Pyrazinamide	3.12 µ/ml	--	--
Streptomycin	6.25 µ/ml	--	--
Ciprofloxacin	--	2 µg/mL	2 µg/mL

CONCLUSION

Novel series of 2-aryl-5-[4-(1H-pyrrol-1-yl) phenyl]-1,3,4-oxadiazole derivatives (**VIa-g**) were synthesized and identified as potent antitubercular and antibacterial agents. Among all the compounds **VIb**, **VIc** and **VIId** have displayed significant activity against *M. tuberculosis* with MIC value 1.6 µ/ml. These compounds will be useful as the lead compounds for developing antitubercular and antibacterial agents.

LIST OF ABBREVIATIONS

⁰ C	-Degree centigrade
DMSO	-Dimethyl sulfoxide
FTIR	-Fourier Transfer Infrared Spectroscopy
NMR	-Nuclear Magnetic Resonance
ppm	-Parts per million
MP	-Melting point
Rf	-Retention factor
Min	-Minutes
H	-Hours
Mol	-Mole
TLC	-Thin Layer Chromatography
MIC	-Minimum Inhibitory Concentration
DMF	-NN-Dimethylformamide

Conflict of Interest

All the authors have no conflict of interests.

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