

The effect of metallicity on the CH₄ and CO quenched abundance in H-dominated atmospheres

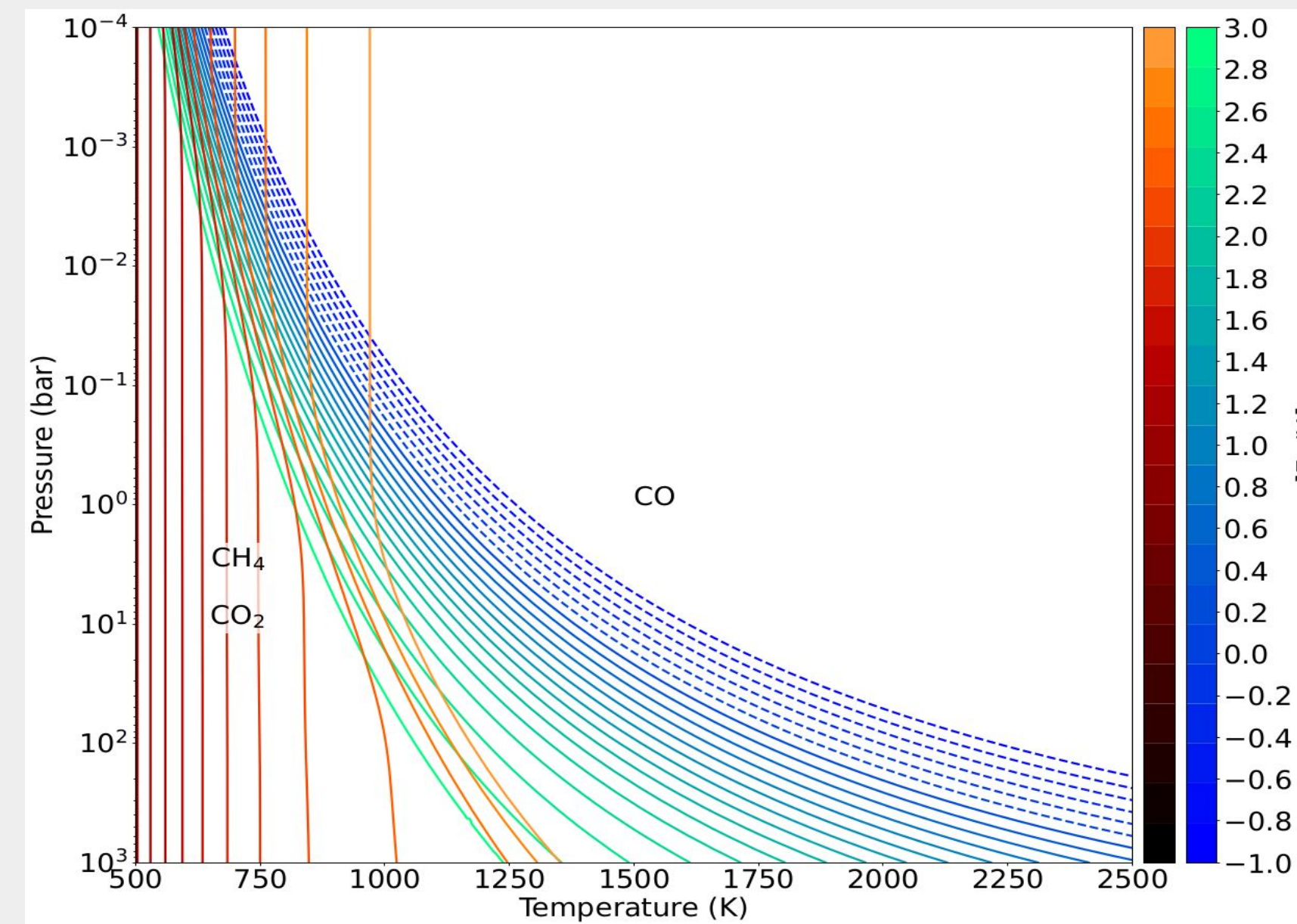
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Introduction

- The atmospheric metallicity greatly influences the composition of exoplanet atmospheres.
- We calculate the chemical equilibrium abundance, chemical time scale and vertical mixing time scale to find the quenched curve in our parameter space (temperature: 500-2500 K, pressure: 10⁻⁴-10³ bar, metallicity: 0.1-1000 × solar metallicity).
- We find the quenched curve for various transport-strength in our parameter range by comparing the chemical and vertical mixing timescales.
- To benchmark our quenched curve, we compare the result of quenching approximation with the 1D photochemistry-transport model.
- We also found that the quenching approximation is a powerful tool to constrain atmospheric parameters.

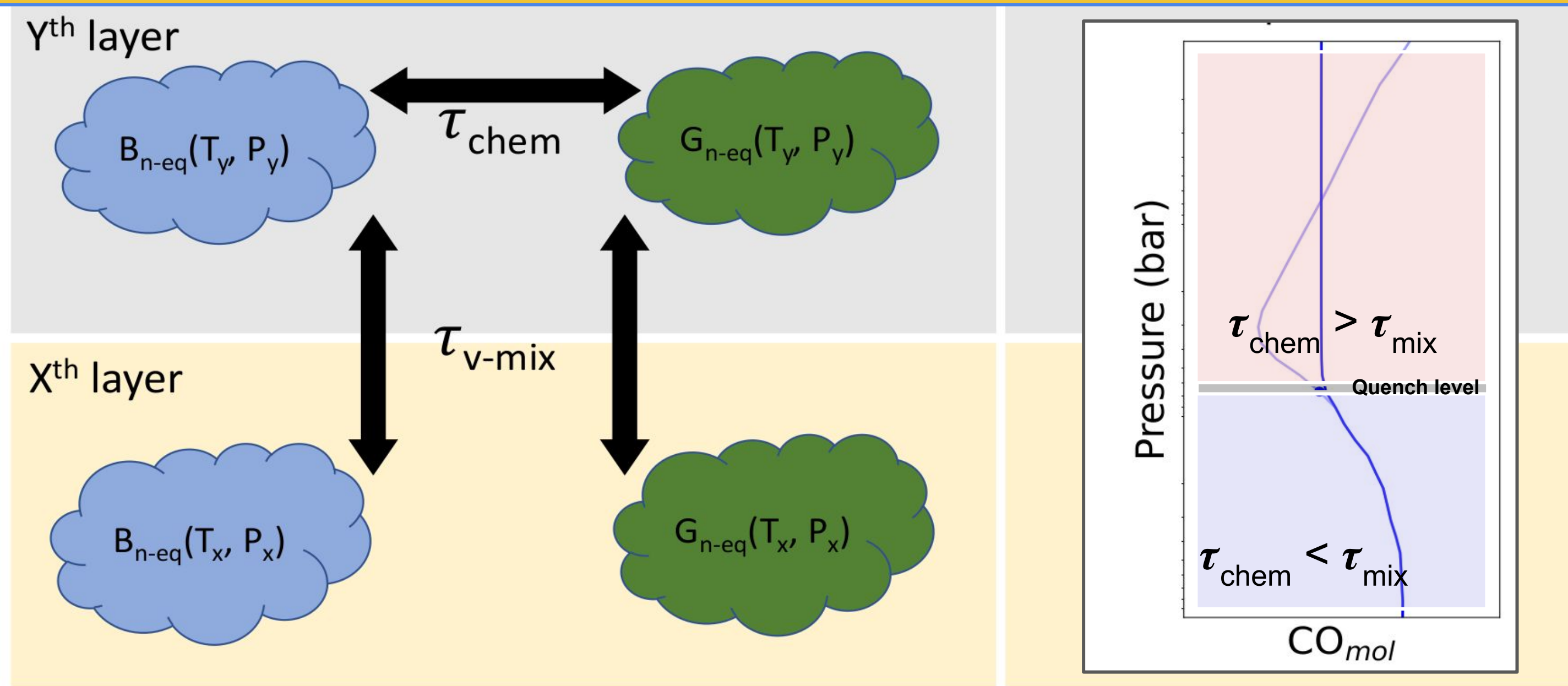
Equilibrium calculation, and mixing processes



Contours of CH₄/CO = 1 and CO₂/CO = 1 for assorted metallicity values are shown. The cyan to blue lines delimitate the region where CH₄ (left) or CO (right) is the dominant species, and the orange to black lines delimitate the region where CO₂ (left) or CO (right) is the dominant species.

- The CO, H₂O, and CO₂ abundances increase with metallicity; however, the CH₄ abundance increases only in the CH₄ dominant region.
- The equal-abundance curve of CH₄ and CO divides the temperature-pressure space of CH₄, CO, CO₂, H₂O into two regions.
- In the CO dominant region, CH₄, CO and CO₂ vary as $\approx [\text{Fe}/\text{H}]^0$, $\approx [\text{Fe}/\text{H}]^1$, $\approx [\text{Fe}/\text{H}]^2$ respectively.
- In the CH₄ dominant region, the increment in CH₄ is proportional to the metallicity, while CO and CO₂ increase more rapidly with metallicity than in the first region.

Quenching Approximation

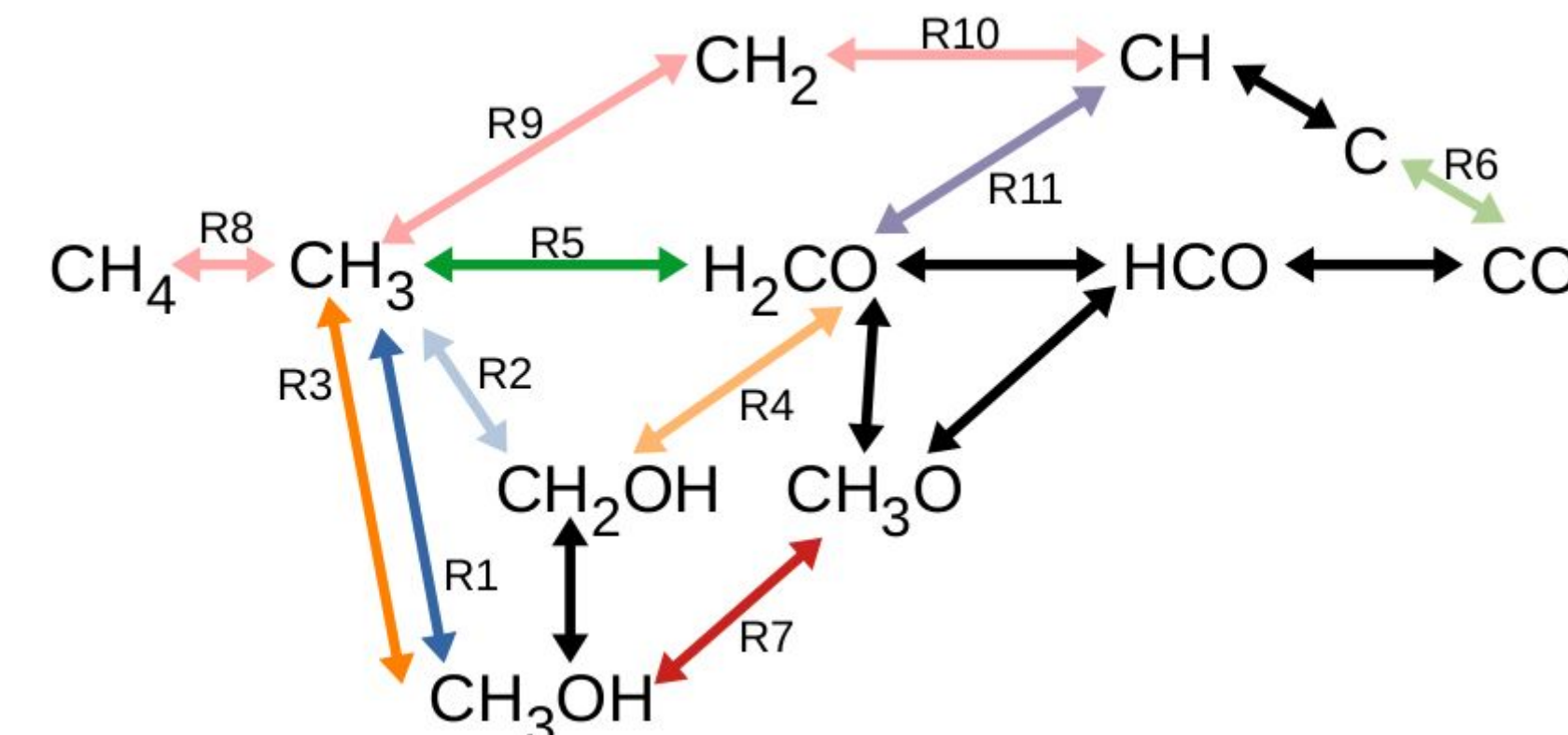


Quenching Approximation (contd.)

Chemical time scale (τ_{chem})

$$\tau_{\text{CO}} = \frac{[\text{CO}]}{\text{Reaction rate of RLS}} + \tau_{\text{H}_2} \times \frac{3[\text{CO}]}{\text{H}_2}, \quad \tau_{\text{CH}_4} = \frac{[\text{CH}_4]}{\text{Reaction rate of RLS}} + \tau_{\text{H}_2} \times \frac{3[\text{CO}]}{\text{H}_2}$$

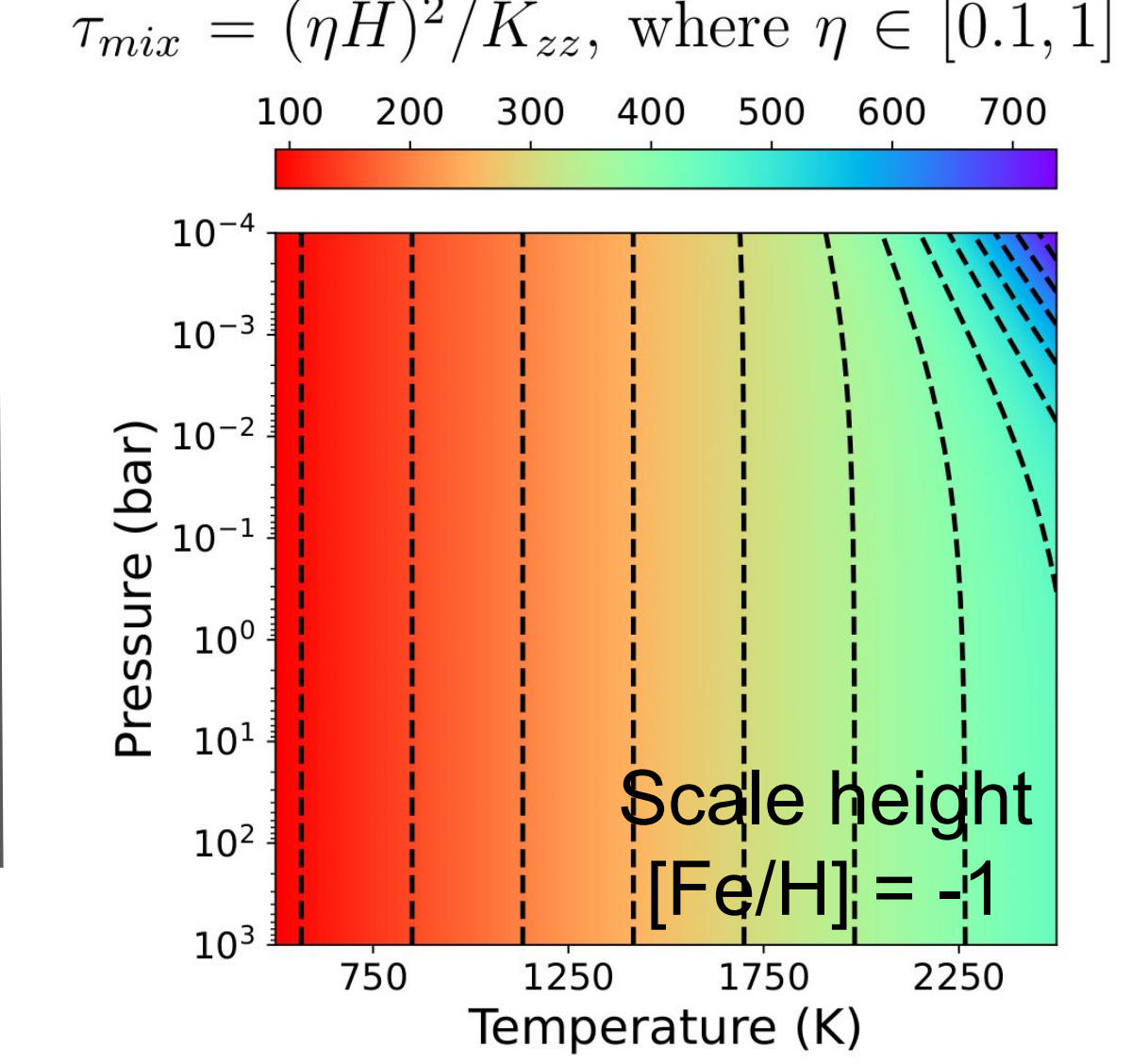
RLS is the rate limiting step, which is the slowest reaction in the fastest conversion scheme. We find the RLS in each of our grid points for the conversion of CH₄-CO-H₂O-CO₂ using an in house developed network analysis tool.



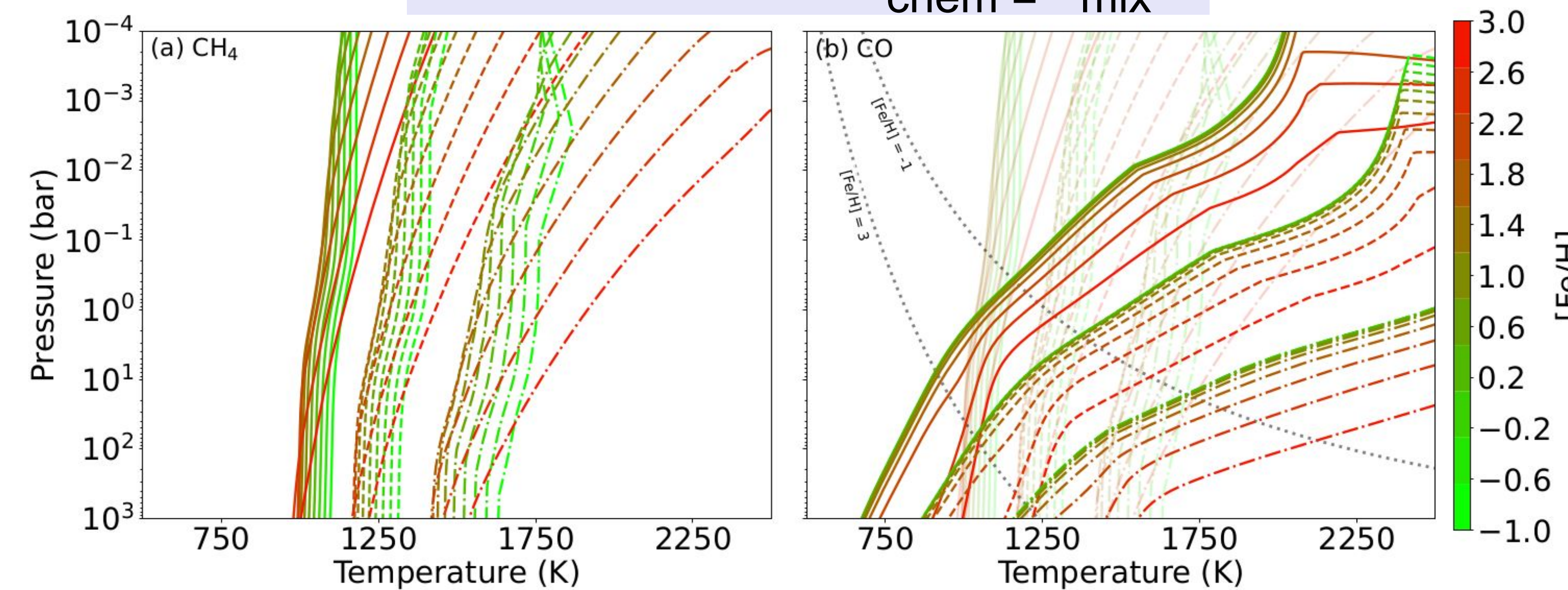
Mixing time scale (τ_{mix})

$$\tau_{\text{mix}} = L^2 / K_{\text{zz}}, \quad H = \frac{K_b T}{\mu g}$$

$$\tau_{\text{mix}} = (\eta H)^2 / K_{\text{zz}}, \quad \text{where } \eta \in [0.1, 1]$$

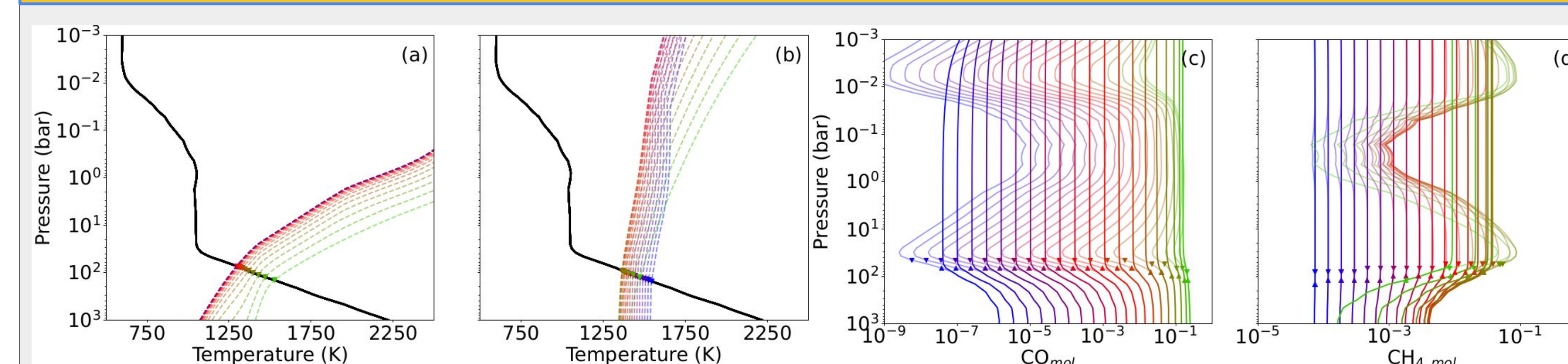


Quenched Curve $\tau_{\text{chem}} = \tau_{\text{mix}}$



Contour lines on which the vertical mixing and chemical conversion timescales are equal for CH₄ and CO, respectively. The three sets of lines are for different values of K_{zz} coefficients (solid: 10⁴ cm² s⁻¹, dotted: 10⁸ cm² s⁻¹, and dashed-dotted: 10¹² cm² s⁻¹). Each set of lines runs from green to red (shown in the color bar), representing eleven different metallicity values.

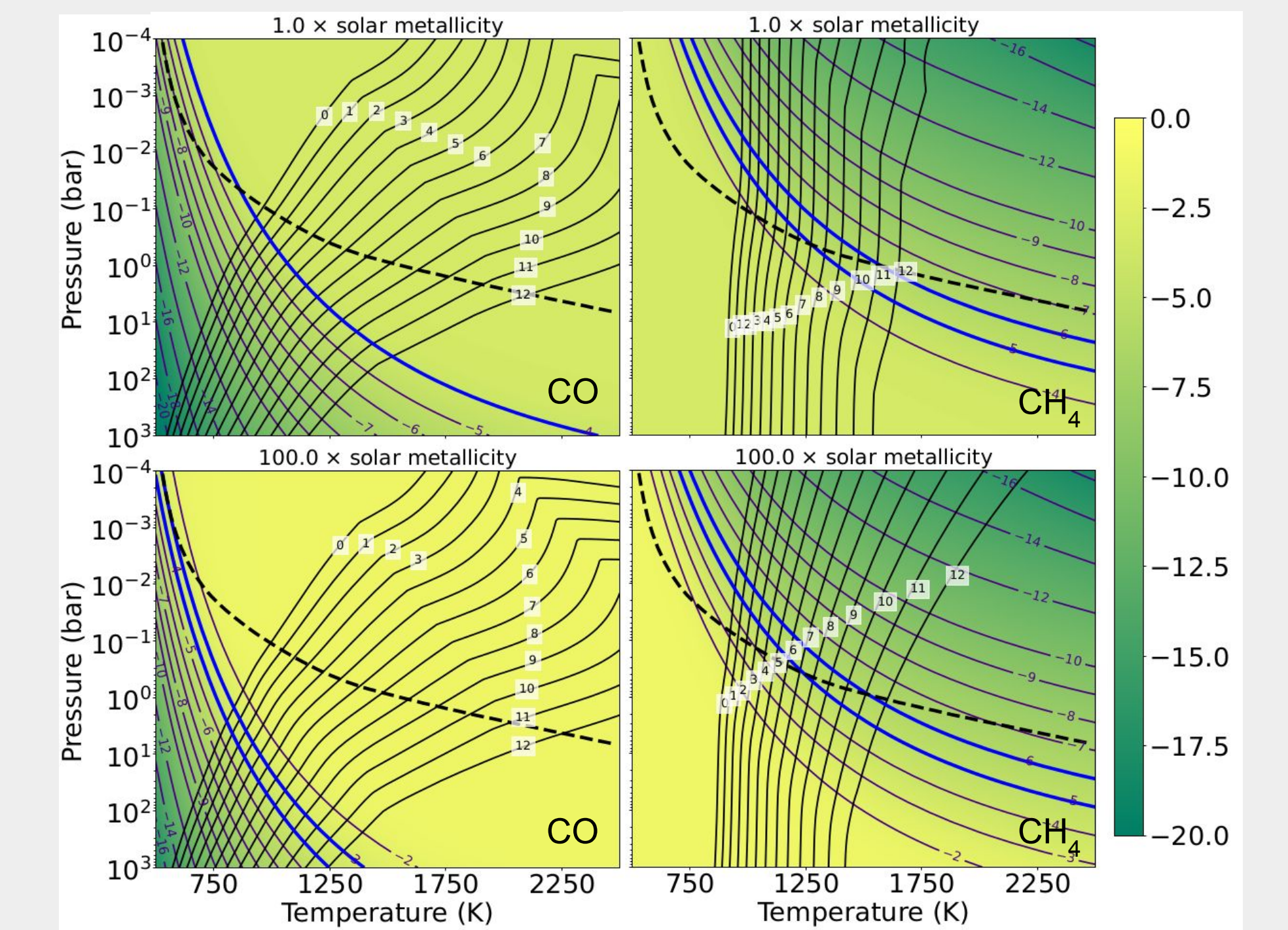
Benchmarking



In (a) and (b) we overplot the thermal profile with the quenched curve of CO and CH₄. In (c) and (d) the solid colored lines are the output of the in house developed 1D photochemistry-transport model and the corresponding faded colored lines are the equilibrium abundances. The upper and lower triangles are the quench levels calculated using 0.1H and 1H as the mixing length.

Constraining the parameters

The observed abundance of molecules and the thermal profiles of the exoplanets are overplotted with the color-mesh plot of the equilibrium mixing ratio and quenched curve of the molecules.



Simultaneous constraints on the observed CH₄ and CO abundance conclude that the solar metallicity can explain the abundance along with K_{zz} = 10⁷ to 10¹⁰ cm² s⁻¹. The same conclusion has been made by Moses et al. (2016) with a kinetic/transport model. They use K_{zz} = 4 × 10⁷ cm² s⁻¹, the solar metallicity, and C/O = 0.66 to best fit the observed abundance.

Conclusion

- In the CO dominant region, CH₄, CO and CO₂ vary as $\approx [\text{Fe}/\text{H}]^0$, $\approx [\text{Fe}/\text{H}]^1$, $\approx [\text{Fe}/\text{H}]^2$ respectively, and in the CH₄ dominant region CH₄ varies as $\approx [\text{Fe}/\text{H}]^1$
- For a fixed K_{zz} value, the CO quenched curve shifts towards the lower atmosphere with increasing metallicity, and CH₄ quenched curve shows complex behaviour.
- The quenching approximation is a powerful tool that can constrain the metallicity and transport strength without solving the transport model.

References

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- Chemical timescale
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- Vertical mixing timescale
Moses, J. I., Visscher, C., Fortney, J. J., et al. 2011, *ApJ*, 737, 1
- Constraining the parameters
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