# **Development of Diffusive Gradients in Thin-Films (DGT) for the** measurements of gold, silver and platinum in aquatic systems Y. W. Jia<sup>1</sup>, M. Elskens<sup>1</sup>, N. Brion<sup>1</sup>, Y. Gao<sup>1</sup>

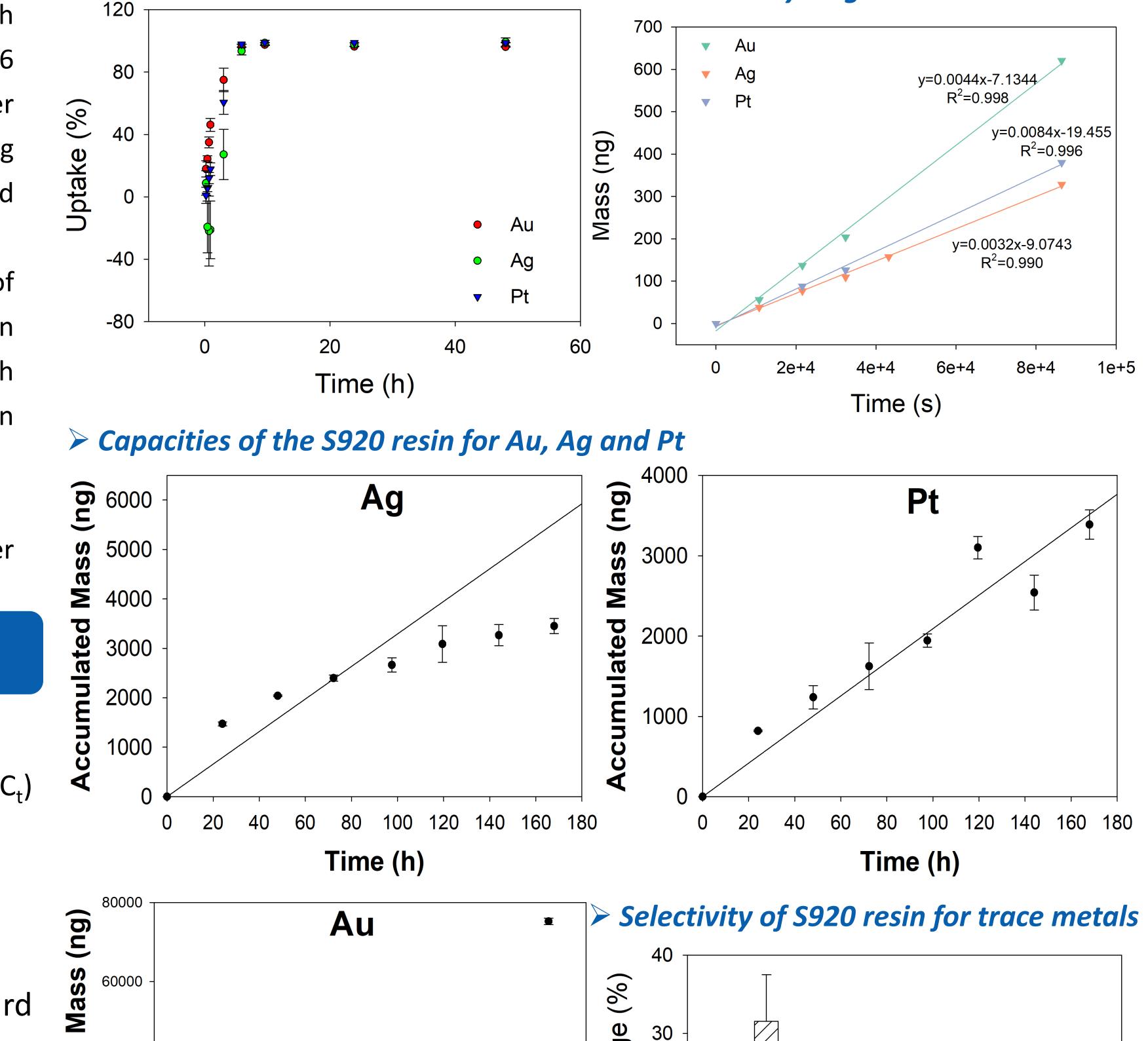
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# INTRODUCTION

Backgrounds: Au, Ag and Pt are the top three precious metals (PM) entering the wastewater treatment plant (WWTP) in Brussels<sup>[1]</sup>. Although 52-99% of them can be removed after treatment, there is still 0.28, 24.6 and 0.04 kg of Au, Ag and Pt being discharged into the Zenner River per year. Considering the toxicity of Au, Ag and Pt nanoparticles reducing from Au-, Ag- and Pt-ions by the cells in biotas<sup>[2]</sup>, PM in the effluent could risk the biotas in the Zenner River.

## RESULTS

- > Uptake of Au, Ag and Pt by S920 resin
- > Diffusion coefficients of Au, Ag and Pt in the hydrogel



- *Current status:* The concentrations of PM are under the limitation of detection of ICP-MS by grab sampling. DGT can overcome the limitation of traditional grab sampling method by passive accumulation, which provides the quantitative *in-situ* measurements of labile metals level in the aqueous system<sup>[3]</sup>.
- *Objectives:* 1) develop a DGT which is specified for Au, Ag and Pt

2) to evaluate the risk of available Au, Ag and Pt in the Zenner

River.

# **MATERIAL AND METHODS**

#### **Uptake kinetics**

- Standard solutions (Au, Ag, Pt) were taken before ( $C_0$ ) and after ( $C_t$ ) S920 resin immersed at different time points.
- Uptake (%) =  $\frac{C_o C_t}{C_o} * 100\%$

#### **Capacities of the S920 resin for Au, Ag and Pt**

- For Au, S920 resin gels were immersed in the Au standard solutions in a range of 10-10000  $\mu$ g/L.
- For Ag and Pt, S920 DGTs were immersed in 200 µg/L standard solution (Ag, Pt) for 7 days. 2 DGTs were sampled every day.

### Diffusion coefficients of Au, Ag and Pt

S920 DGTs were immersed in 20 µg/L standard solution (Au, Ag, Pt) for 3, 6, 9, 12 and 24 h.

•  $C_{DGT} = \frac{M \cdot \Delta g}{D \cdot A \cdot t} \circ C_w$ , concentration of a target compound  $\circ$  $\circ$  *M*, mass of a target on the resin gel  $\circ$ D, diffusion coefficient • *A*, exposure area  $\circ \Delta g$ , the thickness of diffusive gel • *T*, exposure time

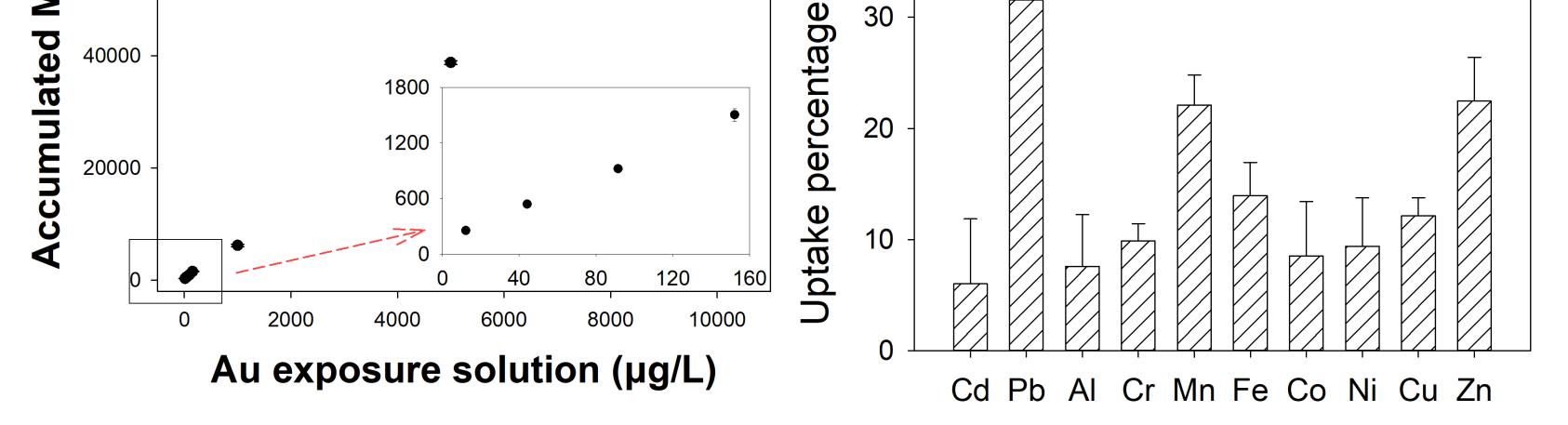
#### Selectivity of S920 resin for trace metals

S920 resin gels were immersed in wastewater for 24 h.

#### **Elution and Analysis**

MS.

- All the targets on the S920 resin gels were eluted by 1 mL AR at 65°C for 12h.
- Samples were diluted 10-1000 times before analysis by ICP-



# CONCLUSIONS

**Q**S920 resin has rapid uptake for Au, Ag and Pt and showed sufficiently high capacities for the binding of them.

**D**The effective diffusion coefficients of Au, Ag and Pt in agarose diffusive gel are 6.35E-06, 8.40E-06 and 3.49E-06 cm<sup>2</sup>/s at 25°C, respectively.

**S**920 resin has strong selectivity for Au, Ag and Pt.

# ACKNOWLEDGEMENTS



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[1]Brion, N., Verbanck, M. A., Bauwens, W., Elskens, M., Chen, M. & Servais, P. (2015). Assessing the impacts of wastewater treatment implementation on the water quality of a small urban river over the past 40 years. Environmental Science and Pollution Research, 22(16), 12720-12736.

[2] Etschmann, B., Brugger, J., Fairbrother, L., Grosse, C., Nies, D. H., Martinez-Criado, G., & Reith, F. (2016). Applying the Midas touch: Differing toxicity of mobile gold and platinum complexes drives biomineralization in the bacterium Cupriavidus metallidurans. Chemical Geology, 438, 103-111.

[3] Zhang, H., & Davison, W. (1995). Performance characteristics of diffusion gradients in thin films for the in situ measurement of trace metals in aqueous solution. Analytical chemistry, 67(19), 3391-3400.