

How the Assignment of Protonation States in Distal Residues May Alter Protein-Ligand Binding in Molecular Dynamics Simulations

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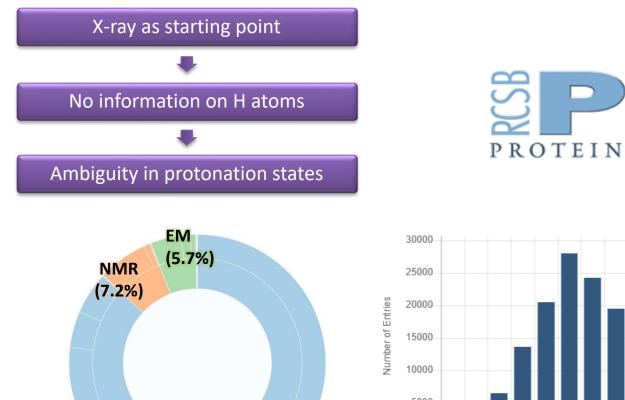
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3/05/2022 Student Webinar: BioExcel Spring School 2022



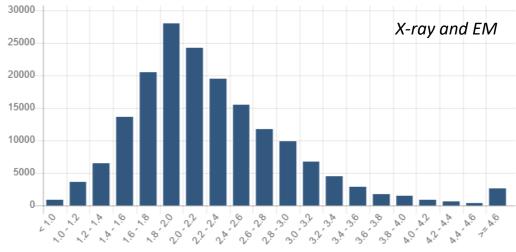
INTRODUCTION

- $\circ~$ On the starting point of molecular dynamics
- \circ Object of study
- The His57 hypothesis



X-Ray

(86.9%)

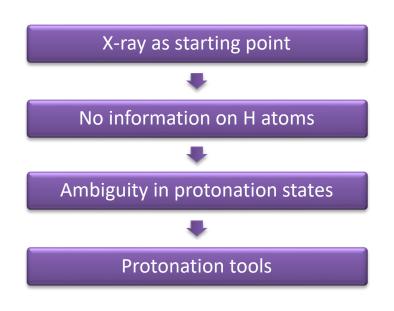


DATA

BANK

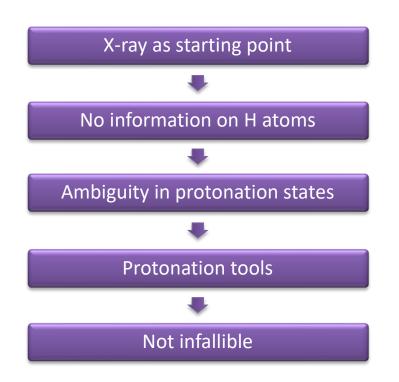
Resolution (Angstrom)

From: https://www.rcsb.org/stats/



Protonation Tools:

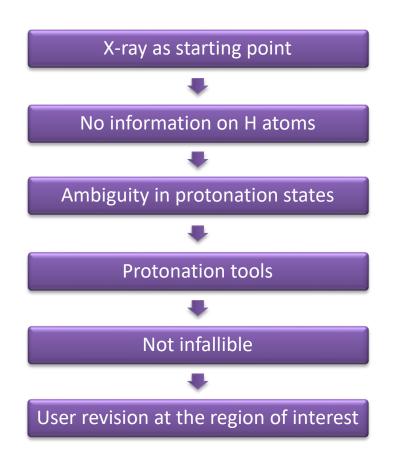
- MD modeling software
- H++
- O PROPKA

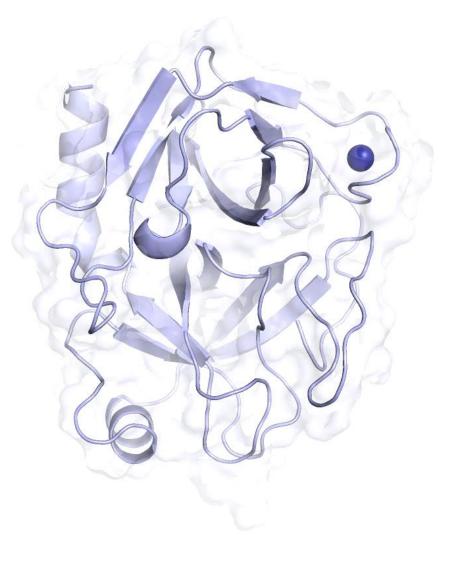


Protonation Tools:

- MD modeling software
- H++
- O PROPKA

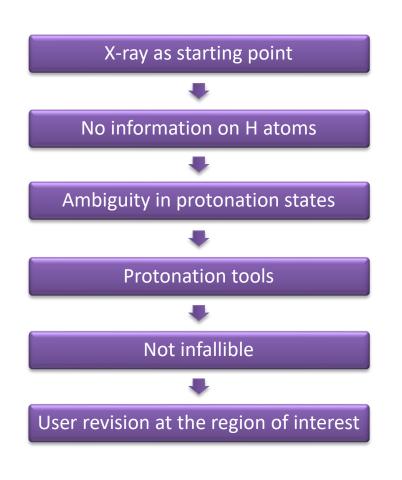
Comput. Theor. Chem., 2012, 1000, 75-84

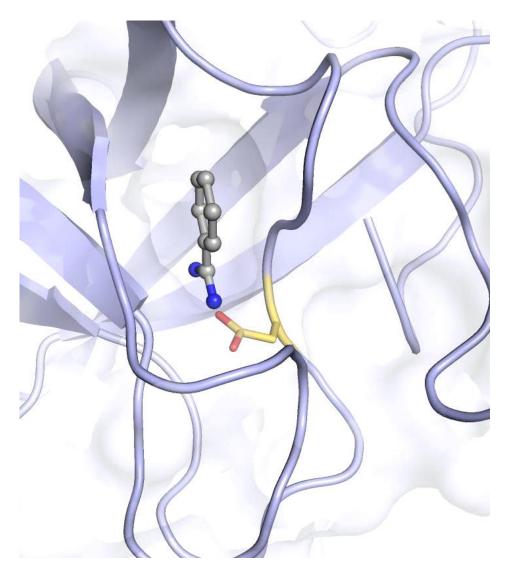


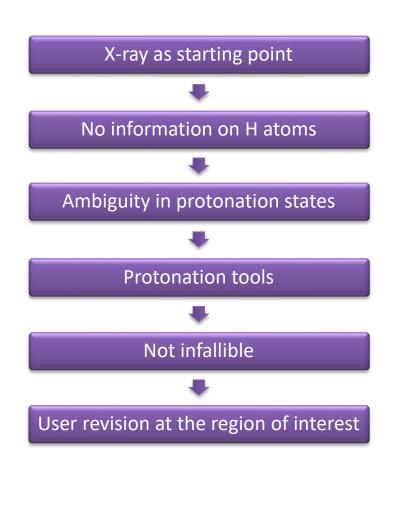


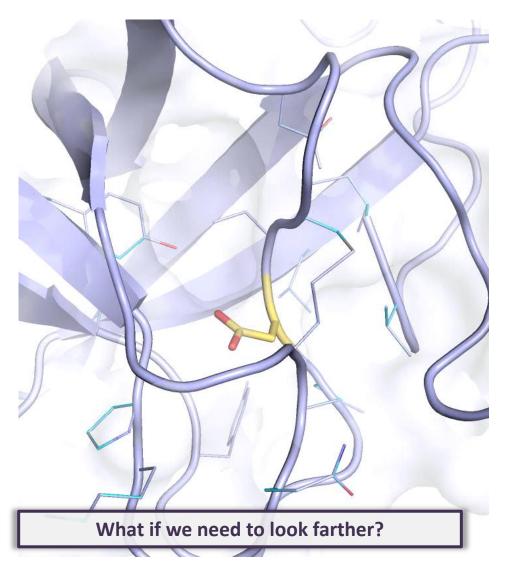
Comput. Theor. Chem., 2012, 1000, 75-84

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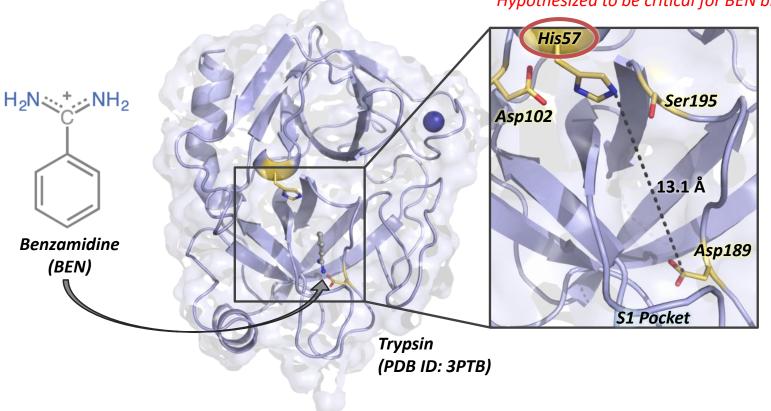








Object of Study

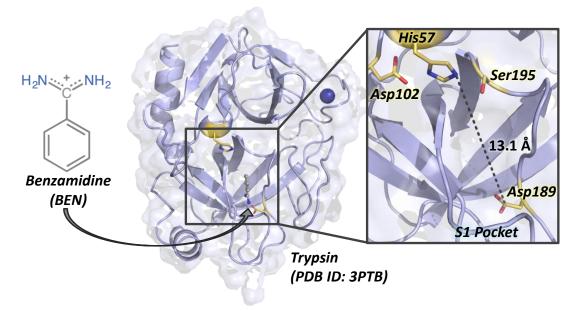


Hypothesized to be critical for BEN binding

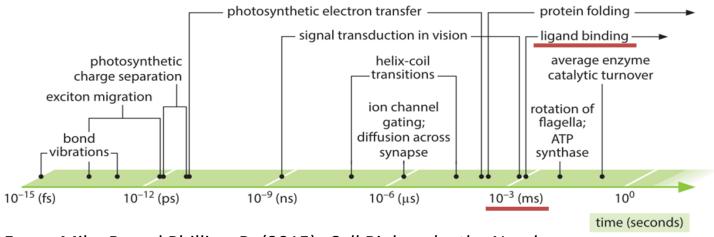
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Object of Study



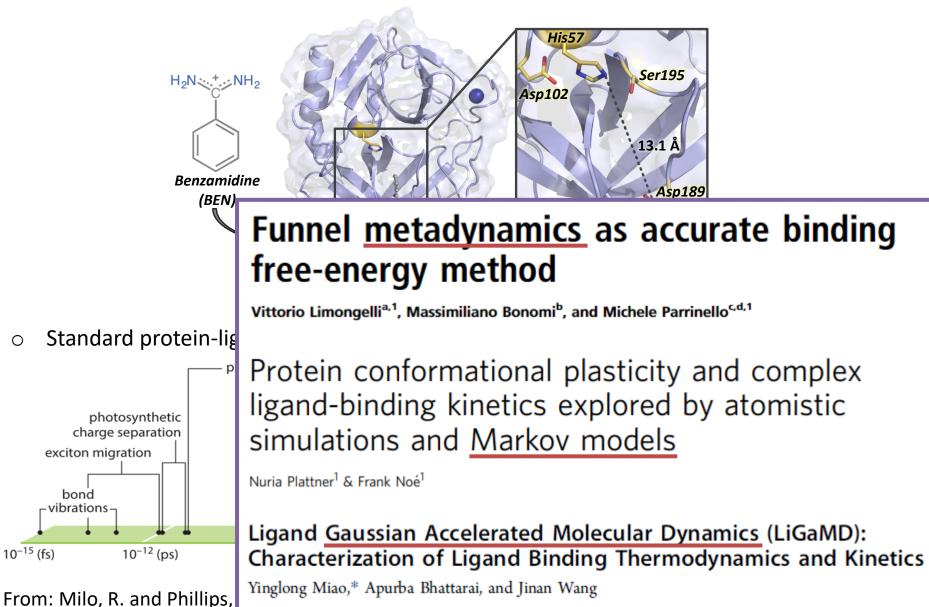
• Standard protein-ligand model



From: Milo, R. and Phillips, R. (2015). Cell Biology by the Numbers

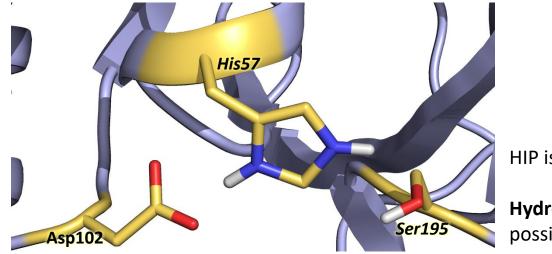
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Object of Study



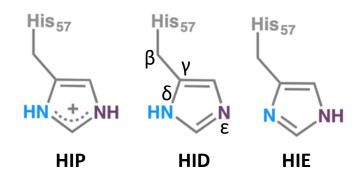
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The His57 Hypothesis



HIP is the suggested one

Hydrogen-bond analysis - HID is also a possibility



Tool	рН 7.0	рН 8.0
H++ (3.0)	HIP	HIP
PROPKA3	HIP	HID

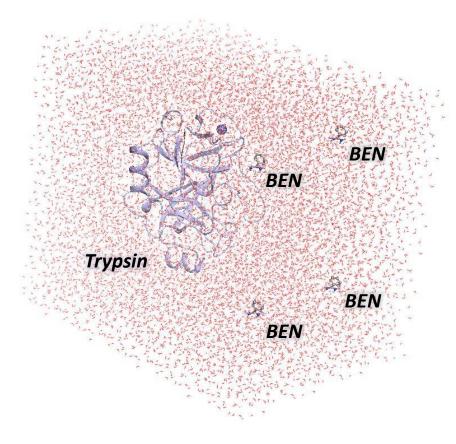
pH of activation [7-9]

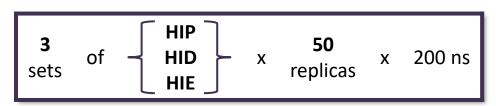
METHODOLOGY

- Spontaneous molecular dynamics
- Spontaneous constant-pH molecular dynamics

Spontaneous Molecular Dynamics

• Spontaneous MD simulations in explicit solvent.





AMBER18

- AMBER-ff14SB Trypsin | GAFF BEN
- Cubic box with 12Å of TIP3P water molecules
- \circ 300K | time step of 2 fs

Spontaneous Constant-pH Molecular Dynamics

• **Discrete Constant-pH** in explicit solvent:

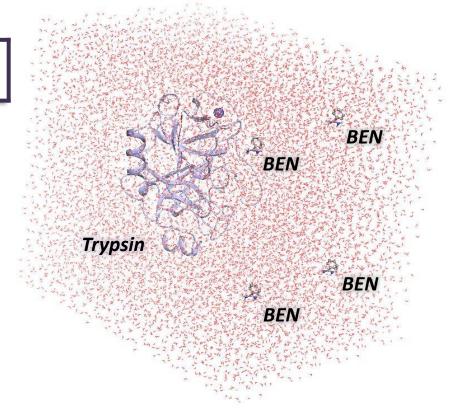
- User selects which residues are allowed to titrate
- After a set of MD steps, these residues may change their protonation state via Metropolis Monte Carlo attempts
- Accounts for the equilibrium



pH of activation [7-9]

AMBER18, constant-pH parameters:

- \circ $\,$ All His were allowed to titrate
- 50 MD steps before protonation change attempt
- 100 solvent relaxation steps
- GB parameters CpHMD was parametrized for



RESULTS AND DISCUSSION

- Conventional MD
- Free-energy landscape
- Binding Pathway
- Constant-pH
- Binding Pathway is Altered by His57

Conventional MD

Binding Events[%] =

Binding Events [%]

replicas w/binding

#total replicas

HID

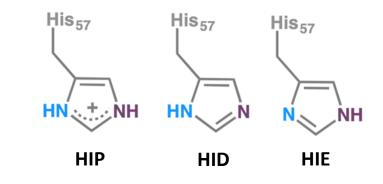
48

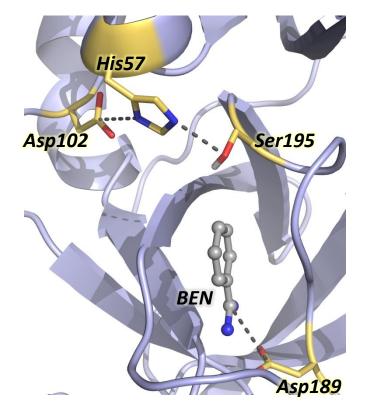
HIE

44

HIP

10



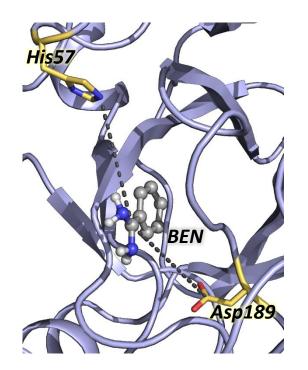


His57 alters the number of encountered binding events

	HIP	HID	HIE	X-Ray
d <i>(His57-Asp102)</i> [Å]				3.5
d(<i>His57-Ser195</i>) [Å]	3.1 ± 0.5	3.9 ± 0.4	4.6 ± 0.9	3.0
d(<i>BEN-Asp189</i>) [Å]	3.8 ± 0.1	3.8 ± 0.1	3.8 ± 0.1	3.9

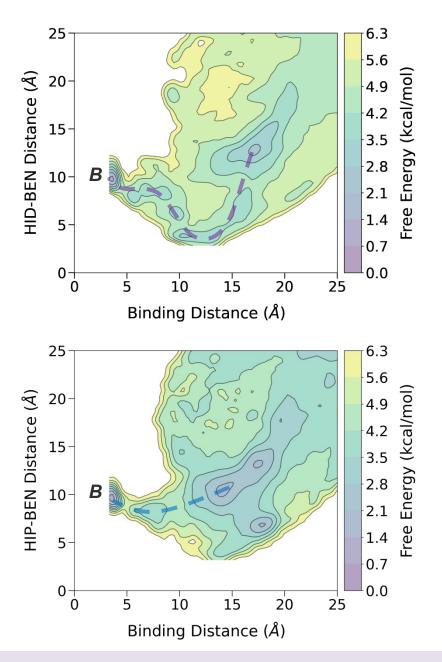
• HIE diverges too much from the crystal structure

Free Energy Landscape

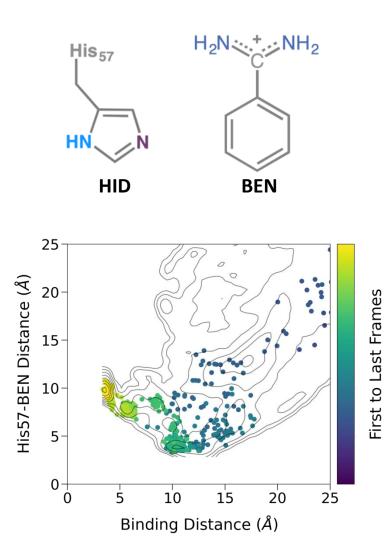


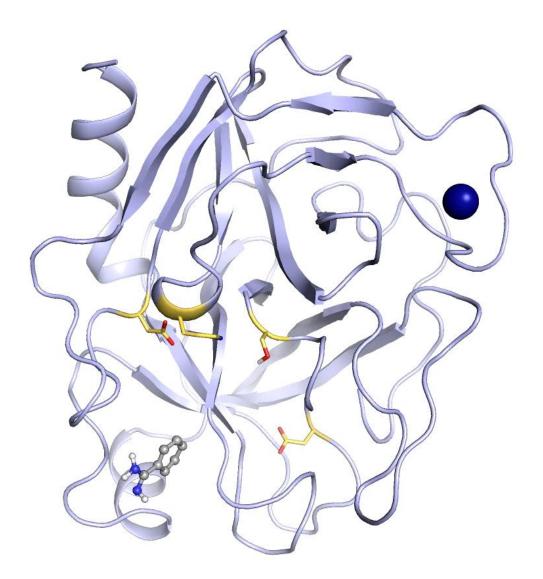
 $\Delta G = -RTln(P)$

• Two different binding pathways

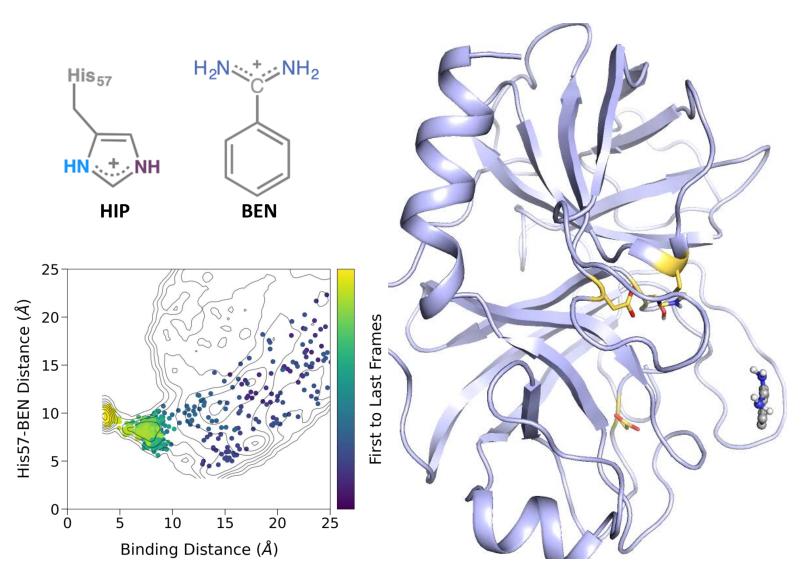


Binding Pathway





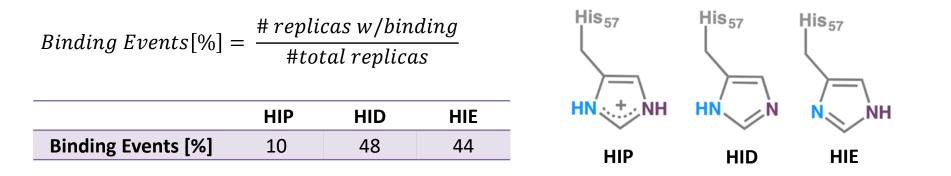
Binding Pathway



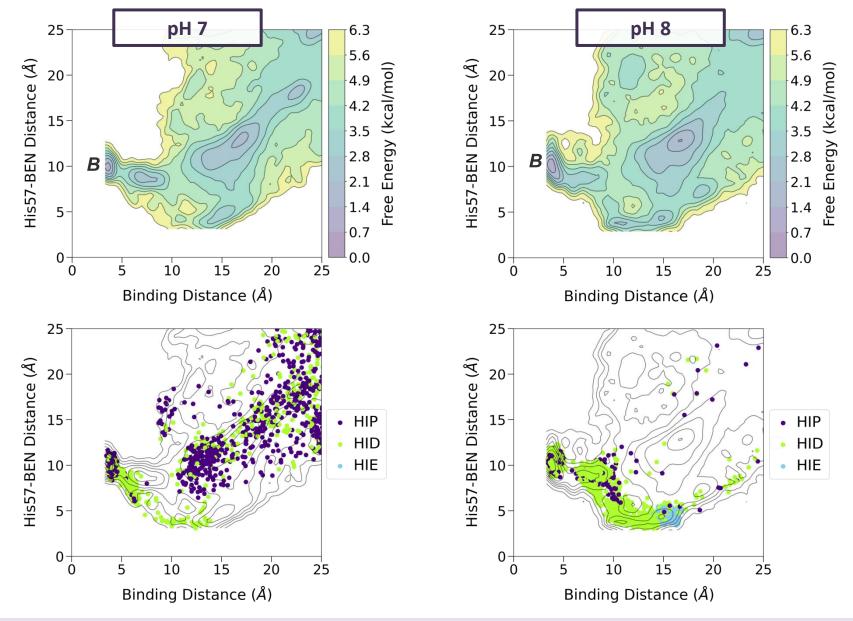
Constant-pH

	рН 7.0	рН 8.0
Binding Events [%]	23	27
Populations [%] HIP – HID – HIE	74-26-0	42-57-0

- Discrete Constant-pH in explicit solvent:
 - User selects which residues are allowed to titrate
 - After a set of MD steps, these residues may change their protonation state via Metropolis Monte Carlo attempts

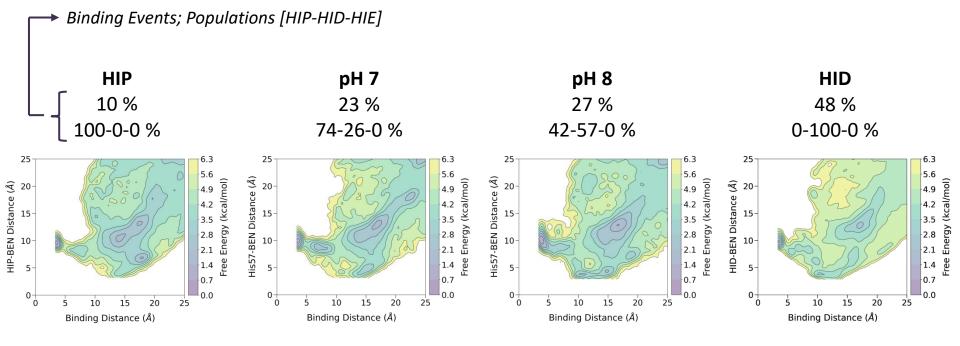


Constant-pH



Results and Discussion **[14 / 17]**

Binding Pathway is Altered by His57



- With increasing HID population, another binding pathway close to His57 appears
- This binding pathway is responsible for the increase in the amount of binding events

CONCLUSIONS

• Concluding remarks

Concluding Remarks

- Assignment of protonation states distal to the binding site can influence the computational characterization of protein-ligand binding.
- In spontaneous binding simulations and other path-relevant studies, the impact of protonation states in residues found along the pathway should always be considered; especially when modeling difficult or not sufficiently-known systems.
- Simulating at pH 7 and with fixed protonation states is a common procedure in MD simulations. These assumptions cannot always be made:
 - Over 60% of binding events involve protonation states.
 - Enzymes are sensitive to pH changes and catalytic residues commonly change their protonation states at different stages of the catalytic cycle.

Consider the system's relevant pH, and whether to employ constant-pH methods.



ACKNOWLEDGMENTS

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Carla Calvó-Tusell Jordi Soler



THANK YOU for your ATTENTION!



H. Girame, M. Garcia-Borràs, F. Feixas, bioRxiv, 2022 (doi: 10.1101/2022.04.30.490145)

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Constant-pH

- Accounts for the equilibrium of protonation states.
 - Discrete Constant-pH Metropolis Monte Carlo
 - Continuous Constant-pH

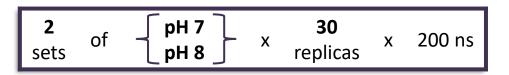
AMBER18

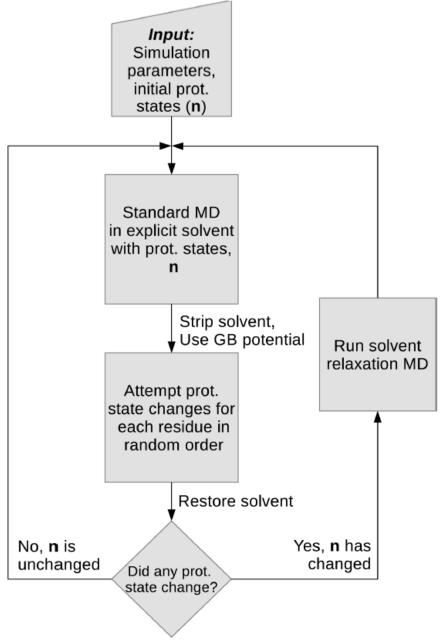
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From: J. Chem. Theory Comput., 2014, 10, 1341-1352

Methodology [7 / 17]