

4-(SUBSTITUTED) BENZYLTHIOSEMICARBAZONES

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4-Substituted benzylthiosemicarbazones have been prepared by the action of 4-(substituted) benzylthiosemicarbazides on aryl aldehydes.

Following the discovery of the tuberculostatic activity of *p*-acetamidobenzaldehyde thiosemicarbazone by Domagk *et al.* (*Naturwiss.*, 1946, **33**, 315) a large number of thiosemicarbazones of aryl and hetero aldehydes and ketones have been prepared. 4-Chloro- and 4-bromo-benzylthioureas have been reported to show tuberculostatic activity (Mrs. Shah, Ph. D. thesis, Bombay Univ., 1957). This observation prompted us to prepare thiosemicarbazones containing a substituted benzyl group. 4-(Substituted) benzylthiosemicarbazones have been prepared by the condensation of 4-(substituted) benzylthiosemicarbazides with aromatic aldehydes.

EXPERIMENTAL

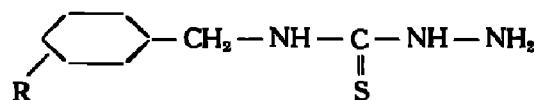
4-(Substituted) benzylthiosemicarbazides were prepared by the action of benzyl isothiocyanates (Trivedi *et al.*, this *Journal*, 1956, **33**, 423; 1958, **35**, 658) on hydrazine hydrate in cold (Buu Hoi *et al.*, *J. Chem. Soc.*, 1956, 2160).

4-(Substituted) benzylthiosemicarbazones were prepared by the action of 4-(substituted) benzylthiosemicarbazides on aromatic aldehydes in ethanolic solution (Buu Hoi *et al.*, *loc. cit.*).

All the compounds were crystallised from dilute ethanol.

TABLE I

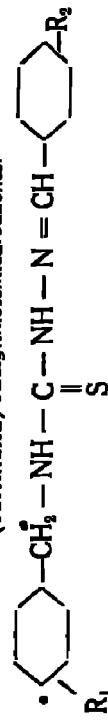
4-(Substituted) benzylthiosemicarbazides.



No.	R.	M.P.	Formula.	% Sulphur. Found.	% Sulphur. Calc.
1.	<i>o</i> -Chloro	141°	C ₈ H ₁₀ N ₃ ClS	15.0	14.9
2.	<i>p</i> -Chloro	156°	C ₈ H ₁₀ N ₃ ClS	14.9	14.9
3.	<i>p</i> -Bromo	135°	C ₈ H ₁₀ N ₃ BrS	12.4	12.3
4.	<i>o</i> -Methyl	152°	C ₈ H ₁₂ N ₃ S	16.3	16.4
5.	<i>m</i> -Methyl	106°	C ₉ H ₁₂ N ₃ S	16.3	16.4
6.	<i>p</i> -Methyl	156°	C ₉ H ₁₂ N ₃ S	16.5	16.4
7.	2: 4-Dimethyl	129°	C ₁₀ H ₁₄ N ₃ S	15.2	15.3
8.	3: 4-Dimethyl	176°	C ₁₀ H ₁₄ N ₃ S	15.4	15.3
9.	2: 5-Dimethyl	181°	C ₁₀ H ₁₄ N ₃ S	15.4	15.3

4-(SUBSTITUTED) BENZYLTHIOSEMICARBAZONES

TABLE II
4-(Substituted) benzylthiosemicarbazones.



R_1	M.P.	Formula.	% Sulphur. Found.	% Sulphur. Calc.								
		$R_1 = o\text{-Chloro.}$			$R_1 = p\text{-Chloro.}$		$R_1 = o\text{-Bromo-}$		$R_1 = p\text{-Bromo-}$			
H	178°	$C_{10}H_{11}N_3S$	10.6	10.5	151°	$C_{10}H_{11}N_3ClS$	10.5	10.5	138°	$C_{10}H_{11}N_3BrS$	9.1	9.2
<i>o</i> -OH	199°	$C_{10}H_{10}ON_2ClS$	10.1	10.0	192°	$C_{10}H_{10}ON_2ClS$	10.0	10.0	185°	$C_{10}H_{10}ON_2BrS$	8.8	8.8
<i>p</i> -OMe	169°	$C_{10}H_{10}ON_2ClS$	9.6	9.6	170°	$C_{10}H_{10}ON_2ClS$	9.7	9.6	129°	$C_{10}H_{10}ON_2BrS$	8.6	8.5
<i>o</i> -Cl	195°	$C_{10}H_9N_3Cl_2$	9.6	9.5	145°	$C_{10}H_9N_3Cl_2S$	9.5	9.5	147°	$C_{10}H_9N_3BrClS$	8.5	8.4
<i>p</i> -Cl	184°	$C_{10}H_9N_3Cl_2S$	9.5	9.5	165°	$C_{10}H_9N_3Cl_2S$	9.6	9.5	151°	$C_{10}H_9N_3BrClS$	8.5	8.4
2:4-Cl ₂	209°	$C_{10}H_8N_3Cl_2S$	8.7	8.6	175°	$C_{10}H_8N_3Cl_2S$	8.7	8.6	185°	$C_{10}H_8N_3BrCl_2S$	7.8	7.7
<i>p</i> -OH	220°	$C_{10}H_8ON_2ClS$	10.0	10.0	220°	$C_{10}H_8ON_2ClS$	10.0	10.0	209°	$C_{10}H_8ON_2BrS$	8.9	8.8
<i>p</i> -(CH ₃) ₂ N	166°	$C_{10}H_9N_3ClS$	9.3	9.2	174°	$C_{10}H_9N_3ClS$	9.3	9.2	162°	$C_{10}H_9N_3BrS$	8.3	8.2
		$R_1 = o\text{-Me.}$			$R_1 = p\text{-Me.}$		$R_1 = o\text{-Me.}$		$R_1 = p\text{-Me.}$			
H	132°	$C_{10}H_{11}NS$	11.4	11.3	113°	$C_{10}H_{11}NS$	11.4	11.3	132°	$C_{10}H_{11}N_3S$	11.4	11.3
<i>o</i> -OH	181°	$C_{10}H_{10}ON_2S$	10.7	10.7	152°	$C_{10}H_{10}ON_2S$	10.7	10.7	201°	$C_{10}H_{10}ON_2S$	10.7	10.7
<i>p</i> -OMe	150°	$C_{10}H_{10}ON_2S$	10.2	10.2	166°	$C_{10}H_{10}ON_2S$	10.3	10.2	168°	$C_{10}H_{10}ON_2S$	10.3	10.2
<i>o</i> -Cl	178°	$C_{10}H_9N_3ClS$	10.2	10.1	155°	$C_{10}H_9N_3ClS$	10.2	10.1	172°	$C_{10}H_9N_3ClS$	10.2	10.1
<i>p</i> -Cl	175°	$C_{10}H_9N_3ClS$	10.1	10.1	158°	$C_{10}H_9N_3ClS$	10.1	10.1	173°	$C_{10}H_9N_3ClS$	10.1	10.1
2:4-Cl ₂	209°	$C_{10}H_8N_3Cl_2S$	9.2	9.1	166°	$C_{10}H_8N_3Cl_2S$	9.2	9.1	177°	$C_{10}H_8N_3Cl_2S$	9.2	9.1
<i>p</i> -OH	196°	$C_{10}H_8ON_2S$	10.8	10.7	162°	$C_{10}H_8ON_2S$	10.8	10.7	215°	$C_{10}H_8ON_2S$	10.8	10.7
<i>p</i> -(CH ₃) ₂ N	149°	$C_{10}H_9N_3S$	9.8	9.8	105°	$C_{10}H_9N_3S$	9.9	9.8	150°	$C_{10}H_9N_3S$	9.7	9.8

TABLE II—*contd.*

R_2 .	Formula.	M.P.	% Sulphur.		M.P.	% Sulphur.		M.P.	% Sulphur.	
			Found.	Calc.		Found.	Calc.		Found.	Calc.
$R_1 = 3:4\text{-Dimethyl}$.										
H	$C_{17}H_{19}N_3S$	121°	10.7	10.8	182°	10.9	10.8	125°	10.9	10.8
<i>o</i> -OH	$C_{17}H_{19}ON_3S$	171°	10.3	10.2	207°	10.3	10.2	201°	10.3	10.2
<i>p</i> -OMe	$C_{19}H_{21}ON_3S$	142°	9.8	9.8	161°	9.9	9.8	141	9.8	9.8
<i>o</i> -Cl	$C_{17}H_{18}N_3Cl_2S$	184°	9.8	9.7	150°	9.7	9.7	184	9.8	9.7
<i>p</i> -Cl	$C_{17}H_{18}N_3Cl_2S$	143°	9.8	9.7	188°	9.8	9.7	175	9.7	9.7
2:4-Cl ₂	$C_{17}H_{17}N_3Cl_2S$	185°	8.8	8.7	208°	8.8	8.7	200°	8.8	8.7
<i>p</i> -OH	$C_{17}H_{19}ON_3S$	164°	10.2	10.2	205	10.3	10.2	206	10.2	10.2
<i>p</i> -(CH ₂) ₂ N	$C_{19}H_{24}N_4S$	175°	9.5	9.4	175°	9.5	9.4	146	9.5	9.4

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