# Advanced Healthcare Materials Graphene Oxide-doped Gellan Gum-PEGDA hydrogel mimicking the mechanical and lubrication properties of articular cartilage

--Manuscript Draft--









Pisa, August 03, 2020

Dear Editor,

We would like to submit our paper entitled "*Graphene Oxide-doped Gellan Gum-PEGDA hydrogel mimicking the mechanical and lubrication properties of articular cartilage*" to be considered for publication in the prestigious journal Advanced Healthcare Materials.

The paper provides an original contribution to the field of nanocomposite materials applied for the cartilage tissue engineering, a domain that is receiving a growing attention.

In the paper, we present a possible new strategy to repair chondral defects by mimicking the mechanical-zone specific and lubrication properties of the articular cartilage tissue, combining gellan gum (GG), poly (ethylene glycol)-diacrylate (PEGDA) and graphene oxide (GO). The results reported in the manuscript highlight the potential use of a multilayered synthetic injectable or surgically implantable filler for restoring AC defects. We believe these findings will be of interest to the readers of your journal.

The challenges addressed by this paper stand at the interface between biomedical science and materials engineering, thus we think the paper has the potential to raise the interest of a large pool of scientists working in these domains.

We declare that this manuscript (or any part of it) is original, has not been published before and is not currently being considered for publication elsewhere.

As Corresponding Author, I confirm that this manuscript has been read and approved for submission by all the named authors.

Please let me know of your decision at your earliest convenience.

Thank you in advance for your time and consideration.

Yours sincerely,

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Articular cartilage (AC) is a specialized connective tissue able to provide a low-friction gliding surface supporting shock-absorption, reducing stresses, and guaranteeing wear-resistance thanks to its structure and mechanical and lubrication properties. Being an avascular tissue, AC has a limited ability to heal defects. Nowadays, conventional strategies show several limitations, thus resulting limited in properly restoring chondral defects. Several tissue engineering approaches have been proposed to restore the AC's native properties without reproducing its mechanical and lubrication properties yet.

This work reports the fabrication of a bilayered structure made of gellan gum (GG) and poly (ethylene glycol) diacrylate (PEGDA), able to mimic the mechanical and lubrication features of both AC superficial and deep zones. Through appropriate combinations of GG and PEGDA, cartilage Young's modulus was effectively mimicked for both zones. Graphene oxide was used as a dopant agent for the superficial hydrogel layer, demonstrating a lower friction than the nondoped counterpart. The bilayered hydrogel's anti-wear properties were confirmed by using a knee simulator, following ISO 14243. Finally, *in vitro* tests with human chondrocytes confirmed the absence of cytotoxicity effects.

The results shown in this paper open the way to a multilayered synthetic injectable or surgically implantable filler for restoring AC defects.

#### **1. Introduction**

Cartilage is a specialized avascular and aneural connective tissue capable of bearing mechanical stresses without permanent distortion. Three types of cartilage tissue (elastic, fibrocartilage and hyaline) are distributed in different parts of the human body, designated to perform specific functions depending on their structural composition.<sup>[1]</sup> In particular, hyaline cartilage, also named articular cartilage (AC), is the most common type of cartilage that is present in hips,

elbows, shoulders and knee joints. It covers the opposing bone surfaces within each joint, providing a low-friction gliding surface for the articulation supporting shock-absorption, distributing loads, reducing stresses on the subchondral bone, and guaranteeing wearresistance.[2]

AC is an anisotropic and viscoelastic tissue. From a functional and structural viewpoint, it can be divided into four zones, which vary in extracellular matrix (ECM) composition, density, collagen fibers assembling, phenotype and chondrocyte activity.<sup>[2b, 3]</sup> The superficial zone, also named the tangential zone, is in direct contact with the synovial fluid of the joint. It is composed of elongated chondrocytes, present with high density and organized parallel to the articular surface.<sup>[4]</sup> This layer provides the smooth gliding surface guaranteeing lubrication and low wear of the joint, exhibiting a lower tensile strength and stiffness than the deeper zones in terms of compressive properties.[2b] The middle zone, also named transitional zone between superficial and deep ones, is featured by a low density of spherical-shaped chondrocytes and collagen fibrils loosely packed and obliquely aligned to the articular surface. This zone guarantees the resistance to compression forces due to the presence of thicker collagen fibers. The deep zone, also termed radial zone, possesses spherical-shaped chondrocytes in a columnar fashion parallel to the collagen fibers and perpendicular to the joint axis. This zone represents the bridge between cartilage and bone and provides the highest resistance to compressive forces during the movement of the joint. Finally, separated by a tidemark line and featured by the lowest density of chondrocytes, the calcified zone represents the last layer of AC that, functionally, anchors the cartilage to the bone<sup>[2b]</sup>. However, overall, the AC can be considered divided in two macro-regions: a superficial and a deep one.

The mechanical properties of AC differ from layer to layer.<sup>[5]</sup> In particular, the compression Young's modulus is in the range of  $280 \pm 160$  kPa to  $730 \pm 260$  kPa for the superficial and deep regions, respectively. This is mainly due to their different permeability to the synovial fluid

flow.[6] Overall, compression tests performed on the human knee AC exhibited a full-thickness compression Young's modulus of  $581 \pm 168$  kPa.<sup>[5, 6b, 7]</sup>

Lubrication is another crucial property of AC. The superficial zone of AC provides a lowfriction gliding surface for the synovial joint, due to an extremely low coefficient of friction between cartilage and cartilage  $(0.001 - 0.03)^{[6b]}$ . However, damages to the superficial area may lead to a rapid wearing of AC and subsequent cartilage breakdown.<sup>[4]</sup>

AC damages constitute the most common injuries of the knee joints.[8] At the origin of the damage, there are mechanical injuries (*i.e.*, adventitious or sport-related traumas)<sup>[9]</sup> or agerelated degeneration (*i.e.,* osteoarthritis (OA)). However, also risk factors such as obesity and genetic predisposition increase the probability to incur in a progressive AC degeneration.<sup>[10]</sup> Two types of defects can mainly occur: chondral (partial and full-thickness) and osteochondral defects.<sup>[9a, 11]</sup> Chondral defects only affect AC and do not extend to the underlying subchondral bone, while osteochondral lesions involve both cartilage and the subchondral bone.<sup>[12]</sup> In osteochondral injuries, blood cells and mesenchymal progenitor cells have access to the damaged cartilage from the subchondral bone, and the healing process leads to the formation of weaker fibrous cartilage.[9c] Differently, chondral defects cannot self-regenerate at all. Conventional strategies, such as auto-, allo-, and xeno-grafts or joint replacement full implants, have been adopted for cartilage repair, showing several limitations:<sup>[13]</sup> osteochondral autografts, which derive from lighter-load-bearing areas of the patient's joint, suffers from limited availability and risk of donor site morbidity; $[14]$  limitations of osteochondral allograft transplantation include low supply, short shelf-life and the need to find a fresh *post-mortem* tissue from a very young donor;<sup>[14b]</sup> osteochondral xenografts might show an immunogenic response;[15] prostheses, necessary in case of irreparable AC damage, must be implanted via invasive surgery and the post-intervention recovery is long and complicated. In addition, the

eventual migration of prosthetic components or the loosening of implants frequently result in adjacent healthy tissue damage and, in some cases, in the need for a new surgery.<sup>[9c]</sup> Tissue engineering (TE) approaches have evolved to restore damaged native tissues by targeting the development of tissue substitutes *in vitro*. [16] Commonly, such a process involves the combination of proper biomaterials, stem cells and growth factors that make up the main elements of three-dimensional (3D) engineered constructs. Scaffolds with a precise microarchitecture can support cell growth and the production of new ECM and, in the meantime, degrade to let the formation of the targeted biological tissue.<sup>[17]</sup> However, many TE approaches exhibit limitations in terms of clinical translation, due to their relative inefficiency in reestablishing and mimicking the native tissue structure. In particular, it is difficult to reproduce the mechanical properties of the hyaline cartilage *in vitro* and, at the same time, to provide the scaffold with appropriate lubrication properties and let the scaffold degrade overtime without affecting such properties.[18]

Multilayer structures were developed to regenerate osteochondral defects, which were composed of subchondral and cartilage zones.<sup>[19]</sup> Osteochondral structures typically show poor mechanical properties and low cell differentiation in the bone and chondral direction, thus resulting relatively ineffective in repairing the defects and in restoring the native tissue. It is not straightforward, with TE approaches, to reconstitute a multilayer structure featured by mechanical properties reflecting the ones of the natural cartilage.[20] Furthermore, TE constructs have to deal with a complex regulatory pathway before entering the clinics, being advanced therapy medicinal products bearing human cells. Fully synthetic (acellular) materials have the advantage of a simpler certification pathway, in addition to the fact that mechanical properties can be varied with higher flexibility, since cells are not embedded in the material and there is no need to keep them alive through a highly porous and soft environment.

Thus, synthetic acellular biomaterials can represent a valid alternative for the repair of cartilage defects and the restoration of tissue functionalities and anatomical structures.<sup>[16a]</sup> The achievement of multilayered structures mimicking the zone-specific mechanical and lubrication properties of the AC would be highly promising for AC repair.[21] This approach can be used to produce a kind of hemiarthroplasty, such as common metal substitutes, that could reconstitute *in situ* the AC surface and also its overall mechanical properties.

Hydrogels are water-swollen materials that can be considered optimal candidates for being injected or placed in the joint and to form 3D structures *in situ*. [22] In the state-of-the-art, some materials such as poly(ethylene oxide terephthalate)/poly-(butylene terephthalate) (PEOT/PBT)<sup>[23]</sup>, poly( $\varepsilon$ -caprolactone) (PCL)<sup>[24]</sup>, collagen<sup>[25]</sup>, chondroitin sulphate (CS) and gelatin (G) microribbon  $(\mu RB)^{[20]}$ , have been shaped in multi-layers and proposed as possible synthetic grafts for AC repair. However, none of them effectively mimicked both the mechanical and lubrication properties of native cartilage. Thus, further efforts are needed in this field.

Gellan gum (GG) is a water-soluble anionic polysaccharide produced by the bacterium *Sphingomonas elodea*; the repeating unit of the polymer is a tetrasaccharide, which consists of two residues of D-glucose and one of each residue of L-rhamnose and D-glucuronic acid.<sup>[26]</sup> GG has been widely studied in the TE field because of its properties such as biocompatibility, biodegradability and ductility. In particular, its use has been particularly advantageous for the cartilage TE field because of its structural similarity with native AC glycosaminoglycans due to the presence of glucuronic acid residues in its repeating unit.<sup>[19c, 27]</sup> This material is thermosensitive and can be ionically crosslinked, interacting with monovalent and divalent ions, producing hard and translucent gels in the presence of metallic ions, including sodium, calcium and magnesium ones.<sup>[27a, 28]</sup>. GG has many advantages over other hydrogels, including shearthinning properties and gel formation at physiological temperatures, which make it a qualified

candidate for being injected into cartilage defects. However, the mechanical properties of such a hydrogel are far from those related to the target cartilage zones.[29]

In this paper, we explored the addition of poly (ethylene glycol) diacrylate (PEGDA) to improve both stiffness and toughness of GG. PEGDA is a synthetic polymer used as a prepolymer solution for the formation of a crosslinked polymeric network upon light exposure and the presence of a photo-initiator.[30] The combination of GG and PEGDA allows playing with both noncovalent and covalent crosslinking methods, to achieve an interpenetrated polymeric network able to effectively recover its shape after loading, as well as to show adequate compressive strength. As mentioned, the achievement of optimal lubrication properties is also a significant challenge, which is needed to improve the performance of the cartilage substitutes. Carbon-based nanomaterials have been used in the literature to improve the mechanical and electrical properties of bare materials. Recently, graphene oxide (GO) has been claimed as an ideal nanomaterial for cartilage TE due to its chondro-inductive properties when embedded into polymeric formulations. In fact, GO can provide a cell-friendly microenvironment able to enhance chondrogenic differentiation.<sup>[31]</sup> On the other hand, GO exhibits peculiar selflubricating and anti-wear properties,  $[32]$  thanks to a high surface-to-volume ratio, a low effective threshold resulting from its nanometer dimensions, which leads to optimizing the performance of polymer composites.[33]

We hypothesize that, by combining GG, PEGDA and GO, we could be able to develop a bilayered hydrogel well mimicking both the mechanical features and the lubrication properties of native AC. Such a nanocomposite hydrogel could be used as an injectable filler or a surgically implantable substitute in chondral defects (Figure 1) showing no cytotoxicity and guaranteeing suitable resistance to wear.



#### *In situ* bilavered hydrogels injection and crosslinking



**Figure 1.** Depiction of the concept targeted in this work. By combining GG, PEGDA and GO, two hydrogels with specific properties, mimicking the ones of superficial and deep cartilage, respectively, are formulated. Then, the superficial and deep cartilage-mimicking hydrogels could be injected in situ and sequentially crosslinked using UV light (photo-crosslinking) and  $CaCl<sub>2</sub>$  solution (ionic crosslinking).

#### **2. Results and discussion**

#### **2.1 Mechanical properties and swelling ratio**

All combinations of GG and PEGDA at different concentrations and crosslinking approaches presented in the following are summarized in Table S1. Figure 2 reports the results of the mechanical characterization of hydrogels combining GG and PEGDA at different concentrations, aimed at mimicking the superficial and deep cartilage. Data show a relevant increase in the Young's modulus when PEGDA concentration increased from 10 (GG/PEGDA10) to 15 (GG/PEGDA15) % w/v. This range of values was then compared with the mechanical properties of superficial and deep cartilage, featured by a Young's modulus of  $280 \pm 160$  kPa and  $730 \pm 260$  kPa, respectively.<sup>[6b]</sup> The first layer included the superficial and

the middle zones areas of a healthy cartilage, whereas the second layer included the deep zone of a healthy cartilage.



**Figure 2.** Mechanical characterization of GG/PEGDA10 and GG/PEGDA15 hydrogels, in terms of: Young's modulus, fracture stress, fracture strain, toughness and swelling ratio. For the Young's modulus, the blue areas represent the range of stiffness reported in O'Connell et al.<sup>[6b]</sup> for the native articular cartilage. \*p < 0.05, \*\*p < 0.01, n = 5.

Analyzed samples: GG/PEGDA10\_5minUV, GG/PEGDA10\_5minUV\_10minMgCl2, GG/PEGDA10\_10minUV, GG/PEGDA10\_10minUV\_10minMgCl2, GG/PEGDA15\_5minUV, GG/PEGDA15\_5minUV\_10minMgCl<sub>2</sub>, GG/PEGDA15\_10minUV, GG/PEGDA15\_10minUV\_10minMgCl2.

The effects of UV exposure time and ionic crosslinking with  $MgCl<sub>2</sub>$  were analyzed for each PEGDA concentration. The Young's modulus of GG/PEGDA10 hydrogels photo-crosslinked for 5 min using a UV light source emitting at 365 nm at an intensity of 40 mW cm<sup>-2</sup>, was significantly improved by ionic crosslinking from an average value of  $122 \pm 16$  kPa (without ionic crosslinking, GG/PEGDA10\_5minUV) to  $182 \pm 65$  kPa (with ionic crosslinking GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub>) ( $p < 0.05$ ). Differently, no statistical difference was found between hydrogels photo-crosslinked for 10 min at the same conditions of wavelength and intensity described above (without ionic crosslinking, GG/PEGDA10\_10minUV), showing a Young's modulus of  $139 \pm 33$  kPa, and hydrogels similar, but subjected to ionic crosslinking (with ionic crosslinking GG/PEGDA10\_10minUV\_10minMgCl2), resulting in a modulus of  $166 \pm 42$  kPa. On the other hand, the Young's modulus of GG/PEGDA15 hydrogels photocrosslinked for 5 min (same conditions as before) improved from an average value of 738  $\pm$ 142 kPa (without ionic crosslinking, GG/PEGDA15\_5minUV) to 801  $\pm$  152 kPa (with ionic crosslinking, GG/PEGDA15\_5minUV\_10minMgCl<sub>2</sub>), but without any statistical difference. Finally, increasing the time of photopolymerization to 10 minutes, a statistically significant difference ( $p < 0.05$ ) in the Young's modulus was found, between absence (715  $\pm$  163 kPa, GG/PEGDA15 10minUV) and presence (780  $\pm$  190 kPa, GG/PEGDA15\_10minUV\_10minMgCl2) of ionic crosslinking.

These results, together with the ones shown in Figure S1, demonstrate that the PEGDA concentration can modulate the rather low mechanical properties of GG, in agreement with other reports.[34] Indeed, the properties of PEGDA can be varied by acting on its molecular weight, concentration, photo-initiator concentration and crosslinking parameters such as exposure time.<sup>[35]</sup> The scientific literature also confirms the relevant role of ionic-crosslinking in the improvement of hydrogels Young's moduli.<sup>[36]</sup> Only one study focused on analyzing the blend of GG and PEGDA (Mn: 6000), even if its primary purpose was the evaluation of the

blend printability. In this study, 1.5 % GG blended with 10 % PEGDA was analyzed and a comparison between the Young's moduli of this material in different conditions (noncrosslinked, photo-crosslinked and, double (photo- and ionic) was made. The Young's modulus evaluated through compression tests was improved from 40 kPa to 60 kPa and to 184 kPa, respectively, for the above-mentioned crosslinking conditions.<sup>[36]</sup> In our case, the combination between GG and PEGDA and the possibility to increase the UV exposure time and PEGDA concentration allowed achieving values closer to the ones featuring the different zones of the human AC, namely  $280 \pm 160$  kPa (superficial) and  $730 \pm 260$  kPa (deep), evidenced with blue regions in the graphs of Figure 2, which constitutes an original result, for this material.

The adoption of a dual crosslinking strategy (optical and chemical one) also allowed to modulate other important mechanical features of the hydrogels. The fracture stress of GG/PEGDA10 hydrogels photo-crosslinked for 5 min was significantly improved from 127  $\pm$ 27 kPa (without ionic crosslinking, GG/PEGDA10 5minUV) to 194  $\pm$  27 kPa (with ionic crosslinking, GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub>) ( $p < 0.01$ ). Similarly, the fracture stress for hydrogels photo-crosslinked for 10 min was significantly improved from  $112 \pm 23$  kPa (GG/PEGDA10 10minUV) to 173  $\pm$  26 kPa using the ionic crosslinking (GG/PEGDA10\_10minUV\_10minMgCl<sub>2</sub>) ( $p < 0.01$ ). The same behavior was found in GG/PEGDA15 hydrogels in which the ionic crosslinking significantly improved the fracture stress of photo-crosslinked materials ( $p < 0.01$ ). Interestingly, the fracture strain of both GG/PEGDA10 and GG/PEGDA15 hydrogel formulations did not show any significant difference between the groups, ranging from values slightly larger than 40 % for the hydrogels mimicking the superficial cartilage and slightly smaller than 40 % for the hydrogels mimicking the deep cartilage. Such values can be considered suitable for the application in the cartilage TE field, being superior to the ones possessed by the native  $AC$ .<sup>[5]</sup>

Moreover, the toughness of both GG/PEGDA10 and GG/PEGDA15 hydrogels resulted statistically different between samples treated with or without ionic crosslinking (*p < 0.01*). The toughness of GG/PEGDA10 was improved of around 1.5 fold when ionic crosslinking was applied for both photopolymerization times, whereas the toughness of GG/PEGDA15 hydrogels was improved significantly from  $64 \pm 14$  kPa (GG/PEGDA15  $-5$ minUV) to  $119 \pm 34$ kPa (GG/PEGDA15\_5minUV\_10minMgCl<sub>2</sub>) (5 min of photopolymerization) ( $p < 0.01$ ). The toughness values of GG/PEGDA15\_10minUV and GG/PEGDA15\_10minUV\_10minMgCl<sup>2</sup> (10 min photopolymerization) were not significantly different, being  $136 \pm 34$  kPa and  $140 \pm 140$ 50 kPa, respectively.

The hydrogel formulations described above showed interesting mechanical properties, which had not been deeply investigated, so far, in the state-of-the-art concerning this material type. Interestingly, the results highlighted that ionic crosslinking with  $MgCl<sub>2</sub>$  in addition to DMEM led to a statistically significant improvement of the Young's modulus, fracture stress and toughness. Besides, 5 min of UV exposure for each layer allowed keeping the crosslinking time lower while maintaining an optimal polymerization degree. The contribution of ionic crosslinking proved more effective than the UV exposure time. In fact, an overall increase in the mechanical properties of the hydrogels was observed, when comparing the samples crosslinked with the same UV light timing but with the additional ionic crosslinking with MgCl2. This enhancement is probably due to an increase in the number of divalent cations ( $Mg^{2+}$ ) that promoted the aggregation of GG chains and formed a more robust polymer network structure. Among the candidate divalent cations,  $MgCl<sub>2</sub>$  was chosen because it proved to have a positive effect on cartilage tissues affected by degenerative diseases, like osteoarthritis.<sup>[37]</sup>

Several blend compositions with various mechanical properties have been developed in the cartilage TE field, so far. Strontium alginate (Alg-Sr) and strontium alginate/chondroitin sulfate (Alg/CS-Sr) hydrogels have been proposed as interesting blends with tunable mechanical

properties for cartilage TE. Both blends showed intriguing biological features, but their compression Young's moduli were far from the target values featuring the natural AC (less than 60 kPa for Alg-Sr and less than 120 kPa for Alg/CS-Sr).[38] Another study focused on the development of an interpenetrating hydrogel made of Agar and polyacrylamide (PAAm), which were physically and ionically crosslinked. The resulting hydrogels exhibited a linear stiffness that improved by increasing the Agar concentration up to 313 kPa (30 mg/mL), a value which is far from the desired one (581  $\pm$  168 kPa<sup>[5, 6b, 7]</sup>), if full-thickness cartilage tissue substitution is targeted.<sup>[39]</sup> Fully synthetic hydrogels, composed of poly (2-acrylamido-2methylpropanesulfonic acid) (PAMPS) and poly (N-isopropylacrylamide-co-acrylamide) [P(NIPAAm-co-AAm)], were also evaluated as potential off-the-shelf materials for cartilage replacement.<sup>[40]</sup> These hydrogels have been developed without reproducing cartilage-specific zones and achieving 1 MPa as a target Young's modulus: this value could be too high for human cartilage tissue substitution. Moreover, these hydrogels were hard to be developed *in situ* due to reactions that typically produce toxic compounds.

In addition to the mechanical analyses, the swelling ratio of the different samples was calculated, to get hints on the crosslinking density of each material formulation and to verify if there were any significant changes due to different concentrations and/or different crosslinking parameters. Figure 2 also shows that the different crosslinking parameters, while keeping the same PEGDA concentration, did not significantly modify the swelling degree. On the other hand, different PEGDA concentrations produced different swelling ratios. This indicates a higher crosslinking density for GG/PEGDA15 hydrogels, as also confirmed by mechanical properties. Based on the results obtained, we chose the best candidates (GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> and  $GG/PEGDA15$  5minUV 10minMgCl<sub>2</sub>), namely the ones showing mechanical properties as close as possible to the ones of the superficial and deep zones of the  $AC$ .<sup>[7b]</sup>

#### **2.2 Hydrogel doped with graphene oxide**

GO which is known for its remarkable mechanical and lubrication properties, was selected for doping the GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> hydrogel (the one mimicking the superficial AC zone). Previous studies showed that GG is a suitable reducing agent for stabilizing the dispersion of nanofillers in aqueous solutions.<sup>[41]</sup> However, only a few studies have explored the combination of GG and carbon-based dopants to improve the mechanical properties of these hydrogels.[42]

GO was synthesized using the modified Hummer's method.[43] Figure 3a and Figure 3b show SEM images and Raman spectrum (with a laser wavelength of 532 nm) of the layered GO bulk. Raman analysis confirmed the two characteristic bands at 1360 cm<sup>-1</sup> (D band) and 1606 cm<sup>-1</sup> (G band) related to GO. After sonication of the GO in ethanol, a droplet from the dispersion was dropcasted on a Si/SiO<sub>2</sub> wafer and let dry overnight; this allowed to observe flakes with a minimum average thickness of  $4.0 \pm 0.3$  nm which was measured using atomic force microscopy (AFM; Figure 3c). Additionally, we performed high-resolution transmission electron microscopy (HRTEM) to confirm the layered structure of GO at the nanoscale (Figure 3d).



**Figure 3.** Characterization of GO and its inclusion in the hydrogel: a) SEM images of layered GO flakes; b) Raman spectrum at 532 nm; c) AFM measurements of a dispersed GO demonstrated minimal thickness of  $4.0 \pm 0.3$  nm (n = 5); d) HRTEM image, confirming the layered structure of a dispersed GO. e) Analysis of the nanocomposite GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub>: Young's modulus, fracture stress, fracture strain and toughness. For the Young's modulus, the blue areas represent the range of stiffness featuring the superficial articular cartilage, reported in).<sup>[6b]</sup> \*p < 0.05, \*\*p < 0.01, n = 5. Analyzed samples: GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub>, GG/PEGDA10/GO\_5minUV\_10minMgCl2.

GO (0.01 % w/v) was added, as mentioned, to the GG/PEGDA10  $5minUV$  10 $minMgCl<sub>2</sub>$ hydrogel formulation, obtaining a homogeneous dispersion of the nanomaterial in the matrix (Figure S2). This was facilitated by the low concentration used  $(0.01 \% w/v)$ , which also

minimizes the cytotoxicity risk of GO, which is known to be dose-dependent.<sup>[44]</sup> With the concentration used in this paper, previous reports highlighted an increased mesenchymal stem cell (MSCs) growth with respect to the controls, thus resulting beneficial, rather than toxic.<sup>[45]</sup> The mechanical properties of GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> and the corresponding GO-doped hydrogel (GG/PEGDA10/GO\_5minUV\_10minMgCl2) were compared in order to assess the role of GO in modulating the behavior of samples photo-crosslinked for 5 min and subsequently ionically crosslinked with  $MgCl<sub>2</sub>$  (Figure 3e). The use of such a concentration of GO led to a slight increase (not statistically significant) in the hydrogel stiffness (from  $182 \pm$ 26 kPa to  $248 \pm 21$  kPa). However, the stiffness of the nanocomposite was still included in the range featuring the native human superficial cartilage zone.<sup>[6b]</sup> The fracture strain did not change with the addition of the nanofiller. On the other hand, the average value of fracture stress and toughness increased significantly, from  $194 \pm 27$  to  $246 \pm 34$  kPa and from  $36 \pm 3$  to  $41 \pm 9$  kPa, respectively. The toughness of the GO-doped hydrogel showed an improvement, probably due to the higher energy dissipation in the presence of GO nanosheets, highlighting a better mechanical performance than the bare hydrogel. [46] As well, fracture stress increased thanks to the dopant agent, which acted as a reinforcement agent of synthetic cartilage tissue. These results are in line with other pieces of evidence available in the scientific literature.<sup>[47]</sup> For example, in poly (vinyl alcohol) (PVA)/GO nanocomposite hydrogels, a significant reinforcement was ascribed to the formation of a dense and stable crosslinking network at different concentrations of the polymer due to the presence of micron-sized GO ( $\sim 0.3$  % w/v). In our case, the introduction of a relatively low concentration of GO  $(0.01\%$  w/v) did not lead to a statistical difference with the bare hydrogel in term of Young's modulus, thus allowing to remain in the range of the target value for mimicking the superficial AC tissue. Nevertheless, other crucial mechanical properties, such as fracture stress and toughness, were significantly improved, thus highlighting that even a low concentration of GO can guarantee important

reinforcing mechanisms.<sup>[48]</sup> Moreover, we observed that the swelling ratio slightly decreased by adding the GO nanofiller (from  $7.2 \pm 0.2$  to  $6.5 \pm 0.3$ ), but without any statistical significance. Lubrication tests were performed on the superficial cartilage-mimicking samples (GG/PEGDA10\_5minUV\_10minMgCl<sup>2</sup> and GG/PEGDA10/GO\_5minUV\_10minMgCl2) in order to assess if the presence of the GO nanosheets could improve lubrication properties. A depiction of the set-up for the evaluation of the lubrication properties is shown in Figure 4a. The lubrication test was divided into three intervals, according to the Stribeck curve, which provides information on the different friction regimes.[49] In the first interval, the system was allowed to relax before the next applied load; in the second interval, the rotational speed (10 min<sup>-1</sup>) and normal force (0.5 N) were increased (up-curve); finally, the rotational speed was decreased (down-curve). This analysis allowed analyzing the evolution of friction in the static regime and the transition from the static to the kinetic one.



**Figure 4.** a) Depiction of the set-up used for testing of hydrogels (red block). b) Lubrication analysis of the composite hydrogels according to the Stribeck curve: the transition from static to dynamic friction is marked with arrows in the diagram. The blue line represents the coefficient of friction of the articular cartilage. c) Analysis of the wear and the torque applied on the hydrogel through the rheometer. Analyzed samples: GG/PEGDA10\_5minUV\_10minMgCl2, GG/PEGDA10/GO\_5minUV\_10minMgCl<sup>2</sup>

As shown in Figure 4b, the coefficient of friction (COF) gradually grew in the static regime for both GG/PEGDA10 5minUV 10minMgCl<sub>2</sub> and GG/PEGDA10/GO 5minUV 10minMgCl<sub>2</sub> hydrogels and a slightly higher value was observed for the doped sample. Nonetheless, for higher rotational speeds, in the kinetic regime, the COF of GG/PEGDA10/GO **5**minUV\_10minMgCl<sub>2</sub> stabilized, whereas the COF of GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> maintained an increasing trend, reaching significantly higher values. The lubrification provided by the doped formulation was thus effective in stopping the increment of the COF. Moreover, it slightly decreased with rotational speed. Results demonstrate that the non-doped formulation possesses a slightly smaller COF in the static regime than the doped one, compatible with the range found for the AC, while the GO nanofiller dispersed into the hydrogel was able to markedly improve the lubrication properties of the bare formulation in the kinetic regime.

The tribological properties of graphene mainly depend on its structure.<sup>[50]</sup> GO might exhibit a slightly larger COF than graphene caused by the presence of exposed interlayer hydrogen bonds, even if it confers a longer lifetime and wear life because it promotes layer slipping mechanisms. This is extremely important for bear-wearing applications, as in the field of materials for cartilage substitution. As demonstrated by Tang et al.<sup>[51]</sup>, GO with a relatively large lateral size (in our case, approximately 4 μm) guarantees lubricity by the sliding along the basal plane of its crystalline lamellar structure. The authors also analyzed the lubricant behavior of increasing concentrations of GO (from 0.05 % to 1.5 %  $w/v$ ) in a water-based lubricant solution, claiming that there is a sort of threshold (in their case  $0.1 \%$  w/v) above which GO, similarly to other nanomaterials, can decrease its lubricant effect.<sup>[52]</sup> The concentration of GO adopted in this

work (0.01 % w/v) is well below such a threshold value. Also, Meng et al.<sup>[47b]</sup>, investigated different concentrations of GO embedded into PVA nanocomposite hydrogels lubricated with FBS (up to 2 % w/v). Their results showed that the minimum COF (close to 0.07) was found for the GO concentration of 1.5 % w/v, while it resulted higher for all the other concentrations tested (above and below 1.5 % w/v). It is important to underline that the lubrification properties achieved by these authors were inferior to the results reported in this work. Furthermore, the use of such a high concentration of GO (1.5 % w/v) could severely influence cell viability due to cytotoxic effects.

Figure 4c shows the results of a wear test performed through a rheometer: it can be observed that for both samples, GG/PEGDA10\_ **5**minUV\_10minMgCl<sup>2</sup> and GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub>, the thickness of the hydrogel remained relatively constant during the whole test, as well as the applied torque. The oscillation was probably due to the elastic deformation of the hydrogel specimen. Native AC possesses considerable lubricant properties in order to withstand the cyclical stresses due to the daily life activities of people. The addition of GO was demonstrated to be effective in keeping the COF at a value close to the native AC one  $(0.03)$ .<sup>[6b]</sup> Furthermore, the doped formulation did not show any preliminary signs of wear. Lubricity and wear represent crucial factors for many biomedical applications, especially those related to cartilage tissue substitution.<sup>[53]</sup>

#### **2.3 Full cartilage-mimicking bilayered hydrogel**

Figure 5a shows the fabrication procedure of bilayered hydrogels by sequentially adding each specific hydrogel formulation in a polydimethylsiloxane (PDMS) mold. The model described in the following permitted, by using the mechanical data reported in section 3.1, to find appropriate volumetric proportions for building a bilayered hydrogel structure, whose overall properties could be similar to the ones of a healthy human AC (a full-thickness tissue, including all the different zones described in the Introduction section). It has been assumed that the composite material was isotropic and composed of two zones (deep and superficial), which

contribute to the final stiffness of the composite material in direct proportion to their volume fraction.<sup>[54]</sup> The volumetric fraction of deep cartilage ( $f_{DC}$ ) and superficial cartilage ( $f_{SC}$ ) for cylindrical hydrogels is the ratio between the deep or superficial cartilage volume and the total volume, described as:

$$
f_{DC} = \frac{v_{DC}}{v_{tot}} = \frac{\pi r^2 h_{DC}}{\pi r^2 h_{tot}} = \frac{h_{DC}}{h_{tot}}\tag{1}
$$

$$
f_{SC} = \frac{V_{SC}}{V_{tot}} = \frac{\pi r^2 h_{SC}}{\pi r^2 h_{tot}} = \frac{h_{SC}}{h_{tot}}
$$
 (2)

where r is the radius of the crosslinked hydrogel disc,  $h_{DC}$ ,  $h_{SC}$  and,  $h_{tot}$  are the heights of deep cartilage, superficial cartilage and the total disc (deep + superficial), respectively. Moreover, the relationship between deep and superficial volumetric fractions is:

$$
f_{DC} + f_{SC} = 1\tag{3}
$$

The equivalent Young's modulus of the bilayered structure can be expressed by applying the Reuss model as follows:

$$
E_{cartilage} = \frac{E_{DC} E_{SC}}{f_{SC} E_{DC} + f_{DC} E_{SC}} \tag{4}
$$

where  $E_{cartilage}$ ,  $E_{DC}$  and,  $E_{SC}$  are the Young's modulus of the healthy cartilage, the one of the deep cartilage hydrogel, and the one of the superficial hydrogel, respectively.

Substituting the relationship (3) in (4),  $f_{deep}$  can be obtained as:

$$
f_{DC} = \frac{E_{DC} E_{SC}}{E_{cartilage} - E_{SC}} \ast \frac{1}{E_{SC} - E_{DC}}
$$
\n
$$
\tag{5}
$$

Subsequentially,  $f_{DC}$  was substituted in (1) to find  $h_{DC}$  as:

$$
h_{DC} = f_{DC} h_{tot} \tag{6}
$$

Finally, the expression (6) was substituted into the definition of  $h_{tot}$  to calculate  $h_{SC}$  as:

$$
h_{SC} = h_{tot} - h_{DC} \tag{7}
$$

Fixed any height, to calculate the volumetric fractions of deep and superficial layers,  $E_{DC}$  and  $E_{SC}$  were considered equal to 864 kPa and 182 kPa, respectively (data derived from the

mechanical characterization of hydrogels), while  $E_{cartilage}$  was considered equal to 500 kPa, based on literature data.<sup>[6b, 55]</sup>

Using the model, we obtained the volumetric fraction of the superficial (GG/PEGDA10/GO\_**5**minUV\_10minMgCl2) and deep (GG/PEGDA15\_**5**minUV\_10minMgCl2) layers equal to 19.43% and 80.57%, respectively. This analysis partially reflects the real situation of native AC, in which the superficial layer constitutes the 10-20 % of the overall cartilage thickness.[56]



**Figure 5.** (a) Depiction of the sequential fabrication procedure used to assemble the cylindrical bilayered hydrogels into PDMS molds. b) Mechanical analysis of the bilayered hydrogels after 30 days incubation in PBS and H2O2 in terms of: Young's modulus, fracture stress, fracture strain and toughness. c) Analysis of the total mass variation overtime, upon incubation in PBS and  $H_2O_2$ , \*p < 0.05, n = 4.

In the state-of-the-art, similar models were developed for predicting composite hydrogel moduli, once given the mechanical properties of the different constitutive materials. Previous studies

 

reported the use of a model to calculate the volumetric fraction of sol-gel composites.<sup>[57]</sup> Isostress within isostrain models were used to describe the mechanical behaviors of poly(2 acrylamido-2-methylpropanesulfonic sodium) (PNaAMPS) dispersed into neutral polyacrylamide (PAAm) hydrogels at small deformation.[57c] Moreover, an isostress model was developed and validated to predict the mechanical responses of cellulose nanocrystals (CNC) reinforced poly(2-hydroxyethyl methacrylate) (PHEMA) nanocomposite hydrogels.<sup>[57a]</sup> However, no multilayered structures using different materials or different concentrations of blended materials have been modeled to predict their whole elastic modulus, in the field of cartilage substitution.

Once fabricated, the mechanical characterization of the bilayer was performed to compare the consistency of the model results with the experimental ones, also evaluating the stability of the hydrogel after incubation in physiological medium (PBS) or inflammatory-like conditions  $(H<sub>2</sub>O<sub>2</sub>)$  up to 30 days. Interestingly, the average Young's modulus of the bilayer (529  $\pm$  128 kPa) was very close to the value of the healthy AC (500 kPa), used to apply the model. As shown in Figure 5c, the Young's modulus, fracture stress, fracture strain and toughness did not vary after one month in PBS. Slightly more evident differences were found after incubation in  $H_2O_2$ , especially regarding toughness.  $H_2O_2$  accounts for a more aggressive condition, used to predict long‐ term oxidative degradation through a short‐ term test. Indeed, induced oxidative stresses can affect both GG and PEGDA.<sup>[58]</sup> However, even the most affected parameter, namely the toughness, remained close to the one owned by the native AC, namely about 47  $kPa<sup>[59]</sup>$ , also in the worst case. These results demonstrate that one month of incubation produced no marked mass loss in the tested samples, even if there was a slight effect on the material mechanical properties, induced by the oxidative stresses of  $H_2O_2$ . Probably, a change of the polymer chain interactions might occur during the continuous presence of an inflammatory condition.

#### **2.4 Wear test**

To perform a wear test, the bilayered hydrogels described in the previous section were fabricated within a menisci-like phantom, as described in Figure 6a.

The wear test was performed using a four-station displacement control knee joint simulator under fetal bovine calf serum-based solution that mimics the knee synovial fluid. A vertical load was applied and the test run with a frequency of 1.1 Hz.<sup>[60]</sup> The applied load at a frequency of 1.1 Hz described a full walk step that was simultaneously composed of 4 different knee movements (trends), namely axial force, flexion-extension angle, intra-extra tibial rotation, and front-rear motion. Figure 6b reports a general description of the system components and some sequential pictures of the system while performing the test. A movie of the knee simulator during the wear test is reported as Movie S1.



**Figure 6.** (a) Depiction of hydrogels fabrication within a menisci-like phantom. b) Depiction of the knee simulator set-up. Sequential images of the knee simulator movements during the wear test (see also Movie S1). In the zoomed image, it is possible to observe that the meniscus is immersed into fetal bovine calf serum-based solution and loaded by femoral condyles of metal knee prosthesis. c) Macroscopic and brightfield (4x) images of samples before wear test. d) Macroscopic and brightfield (4x) images of samples after 18,000 cycles. e) Macroscopic and brightfield (4x) images of samples after 100,000 cycles. The brightfield images of medial and lateral condyles were captured in the center and on the edge of samples using an optical microscope. Scale bar: 250 µm. f) Roughness measurements using optical profilometer of pretested and tested (100,000 cycles) bilayered hydrogels. g) Surface images captured using optical profilometer of pre-tested (left) and tested (right) bilayered hydrogels.  $n = 4$ .

Macroscopic visual examination of all knee-retrieved components revealed a consistent wear pattern between two subsequentially ISO 14243 standard wear tests, as shown in Figure 6c-e. A similar behavior was observed between the medial and the lateral condyles of the specimens. After 18,000 cycles, we observed that both condyles remained perfectly intact (Figure 5d) both in the center and at the edges, as the pre-test samples (Figure 6c). After 100,000 cycles on the UHMWPE holes, we observed that the medial condyle showed some slight signs of wear without delamination, while the lateral condyle remained intact (Figure 6e). We can suppose that medial condyle hydrogel samples were subjected to delamination due to rubbing with the prosthesis because the body load is transmitted through the menisci in different proportions (70% in the lateral compartment and 50% in the medial compartment).[61] Moreover, this effect can be also partially due to a scarce adhesion between the UHMWPE hole and the hydrogel, being the PE hydrophobic. No statistical differences were found in terms of roughness between 100,000 cycles-tested and pre-tested samples (Figure 6f), while slight signs of wear were observed on the surface of the stressed hydrogel (Figure 6g). Despite the presence in the stateof-the-art of different examples of cartilage-substituting hydrogels, to our knowledge, our study is the first one assessing the functional behavior of the produced hydrogels by using this wear test, adhering to the ISO 14243 standard. Thus, these results shed light on the cyclical resistance of the fabricated hydrogels, which is of crucial importance for assessing their possible success as an osteoarticular tissue substitute, but also highlight the need, in general, for an extensive assessment of the wear behavior of hydrogels, in the state-of-the-art of this research field. The test was conducted by combining all different walk movements following the ISO 14243

standard for 100,000 cycles that corresponded to evaluate standard human continuous walk for around 24 h. Comparing these results with the ones obtained in a study in which natural cartilage was tested, it is worth highlighting that already at 50,000 cycles, the cartilage tissue undergoes damage, if continuously solicited.<sup>[62]</sup> We can thus claim that the fabricated bilayer showed an excellent behavior when subjected to a high number of cycles and can thus be considered a suitable functional cartilage substitute.

It is important to underline that the materials described in this study could be injected directly in chondral defects through syringes-like devices and sequentially crosslinked in situ.<sup>[63]</sup> Indeed, we preliminarily evaluated the printability of both hydrogel formulations mimicking the deep and the superficial cartilage zones, finding encouraging results (Figure S3, Movie S2). However, the potential application of such a bilayered hydrogel does not exclude the fabrication of the construct outside the body and its subsequent surgical implantation (*e.g.,* through arthroscopic procedures). Along this line, the use of computed tomography (CT) data would help in obtaining the exact shape of chondral defects and then to fabricate defect-specific tissue substitutes using customized molds.<sup>[64]</sup>

#### **2.5 Cytotoxicity evaluation**

Finally, to evaluate the cytotoxicity of the superficial and deep hydrogels  $(GG/PEGDA10/GO-5minUV-10minMgCl<sub>2</sub>$  and  $GG/PEGDA15-5minUV-10minMgCl<sub>2</sub>$ , we assessed viability and metabolic activity of human chondrocytes put in contact with the supernatant collected from both hydrogels incubated in culture medium for 24 hours. Cell viability and cell metabolic activity were checked after two and six days. Cell viability assessed by Live and Dead (L/D) assay showed the presence of only viable cells (green stained), almost without dead cells (red stained) at the analyzed time points in both experimental groups, indicating that the hydrogel formulations did not affect cell viability (Figure 7a). The behavior was similar among the cells put in contact with the hydrogels-derived media and the cells cultured as a control group (chondrocyte simply grown in standard culture medium).

Furthermore, metabolic cell activity, evaluated by MTT assay, showed no statistical difference at the two time points between the control and the experimental groups (Figure 7b). After six days, the absorbance values were all lower than the ones found after two days. This result was

due to cell confluence after a week in culture, that determined a reduced metabolic activity in all samples analyzed, without any difference between control and experimental groups.[65]



GG/PEGDA10/GO **5**minUV 10minMgCl<sub>2</sub> supernatant, and treated with GG/PEGDA15\_**5**minUV\_10minMgCl<sup>2</sup> supernatant evaluated after 2 and 6 days. Viable cells are shown in green dead/necrotic cells are shown in red. Scale bars: 200 µm. b) MTT test of the same samples evaluated after 2 and 6 days.  $n = 4$ .

Analyzed samples: GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub>, GG/PEGDA15\_5minUV\_10minMgCl<sup>2</sup>

Cytocompatibility is a fundamental standard evaluation of cell viability for cartilage repair. Our data confirmed that the hydrogels did not affect this important parameter. Only three papers focused on different combinations of  $GG/PEGDA^{[36, 66]}$  and  $GG/GO^{[67]}$ , but cell viability was

evaluated only in GG/PEGDA hydrogels. In line with our data, they have demonstrated that mesenchymal stromal cells embedded in GG/PEGDA showed a high percentage (above 87%) of viable cells.[36] By contrast, other authors have shown that a different GG/PEGDA hydrogel formulation provided a unfavorable environment for mesenchymal stromal cells proliferation associated to an increase of cell apoptosis starting from third day.[66]

#### **3. Conclusion**

We described the development of a bilayered hydrogel able to mimic the mechanical features of native AC and featured by enhanced lubrication properties. The bilayer was based on a combination of GG and PEGDA, and on two combined crosslinking approaches (physical and ionic). The resulting material proved to effectively modulate the mechanical properties of the superficial and the deep components of the bilayer. In this process, the influence of the ionic crosslinking was more relevant than the photo-crosslinking one. Two formulations were identified as the optimal ones to mimic the superficial and deep zones of the human articular cartilage.

GO nanosheets were synthesized in order to improve the lubrication properties of the top layer, thus to mimic the articular cartilage also in terms of low friction. Tribological analyses confirmed the beneficial effect of the addition of GO nanosheets into the top layer, in the kinetic regime. The nanocomposite also guaranteed improved mechanical properties, especially toughness.

By applying the Reuss model, we developed a bilayer structure mimicking the overall mechanical properties of the healthy articular cartilage, using two layers made of different hydrogel formulations and featured by specific thicknesses, which resulted as the output of the model. No cytotoxic effects were found on human chondrocytes up to 6 days, demonstrating the safety of the bilayered hydrogel. Moreover, a wear test confirmed that the bilayered construct can withstand physiologically relevant stresses up to 100,000 cycles. In conclusion,

the proposed biomimetic bilayered hydrogel is a promising candidate as a possible synthetic substitute to be delivered into chondral and osteo-chondral defects in a minimally invasive way and to restore the functional properties of the cartilage. Future *in vivo* tests will demonstrate the pre-clinical and clinical suitability of such a composite material.

### **4. Experimental Section**

#### *Materials:*

Gellan gum (GG) powder (Gelzan®, Merck) and poly (ethylene glycol) diacrylate (PEGDA. Mn: 575, Merck) solution was used for the preparation of hydrogels. 2-hydroxy-4′-(2 hydroxyethoxy)-2-methylpropiophenone (Irgacure 2959, I2959, Merck) was used as photoinitiator, stocked in a mother solution made of dimethyl sulfoxide (DMSO, Merck). Magnesium chloride hexahydrate ( $MgCl<sub>2</sub>-6H<sub>2</sub>O$ , Merck) diluted in deionized water was used to perform the ionic crosslinking of hydrogels. Graphite flakes  $(99\%$ , Alfa Aesar), H<sub>2</sub>SO<sub>4</sub>  $(98\%$ , Merck),  $H_3PO_4$  (98%, Merck), KMnO<sup>4</sup> (99%, Merck) and  $H_2O_2$  (30% Merck) were used to synthesize graphene oxide (GO) sheets. Wear tests were conducted using sodium azide (NaN3, Merck) and ethylenediaminetetraacetic acid (EDTA, Merk). Cell culture studies were conducted using low glucose Dulbecco's Modified Eagle Medium (DMEM, Merk), fetal bovine serum (EuroClone), penicillin/streptomycin (Gibco) and trypsin/EDTA (EuroClone, Merk). Cell viability studies performed using phosphate buffered saline (PBS, Gibco) solution, Live/Dead® Viability/Cytotoxicity Kit (Invitrogen) and metabolic activity by 3-(4,5-Dimethyl-2-thiazolyl)- 2,5-diphenyl-2H-tetrazolium bromide (MTT, Merck).

#### *Preparation of superficial and deep cartilage hydrogels*:

Briefly, GG (1.75 and 1.67 % w/v) was dissolved in deionized water by magnetic stirring at 65°C for 1 h. After dissolution, the solutions were formed by adding PEGDA at a concentration of 15% and 10% w/v, respectively. Both solutions were kept at 75 °C and agitated by magnetic stirring for 1 h. Then, I2959 photo-initiator  $(0.1\% \text{ w/v})$  was added in both solutions by maintaining both temperature control and magnetic stirring active. Finally, each formulation was poured into cylindrical holes (diameter: 6 mm; height: 5 mm) of a custom-made PDMS mold. The photo-crosslinking was performed by UV light exposure using a UV optic fiber  $(\lambda=365 \text{ nm}, \text{Lightning cure LC5}, \text{Hamamatsu Photonics UK}, \text{Ltd})$  at an intensity of 40 mW cm<sup>-</sup> for 5 and 10 minutes. Some hydrogels underwent a further ionic crosslinking step through a first immersion into a MgCl<sub>2</sub> solution (1 % w/v in deionized water) for 10 minutes at room temperature (RO). Then, all the sample was finalized by incubation in DMEM for 24 hours at 37°C. All combinations of materials at different concentrations and crosslinking approaches are summarized in Table S1.

#### *Hydrogels mechanical characterization*:

Each sample type was analyzed in terms of Young's modulus (E), fracture stress, fracture strain and toughness. Uniaxial compression was performed with a Instron Mechanical Testing System (model 2444, Instron, Norwood, MA, USA) equipped with both  $\pm$  10 N and  $\pm$  1000 N load cells at a compression rate of 1 mm min<sup>-1</sup> until reaching the hydrogel breaking point. E derived from the linear region of the stress-strain curve (the first 10% of strain), fitted according to the equations:

$$
\sigma = \frac{F}{A_0} \tag{8}
$$

and

$$
\varepsilon = \frac{\Delta l}{l_0} \tag{9}
$$

where  $\sigma$  and  $\varepsilon$  are the stress and the strain, *F* is the force, *Δl* is the deformation,  $A_0$  is the sample area and  $l_0$  is the sample initial length. E was calculated according to the equation:

$$
E = \frac{\sigma}{\varepsilon} \tag{10}
$$

The fracture stress and fracture strain were assumed as the stress and strain at which the sample broke. The toughness, defined as the amount of strain energy per unit volume that a hydrogel can absorb, was calculated as the area under the stress-strain curve up to the breaking point.

 

The swelling ratio was assessed to measure the amount of water absorbed by each type of hydrogel. The mass of each sample type was measured after incubation in DMEM for 24 hours at room temperature after removing residual liquid using filter paper (W  $_{\text{after swelling}}$ ). Then, each sample was dried entirely, and the mass was weighed ( $W<sub>drv</sub>$ ). The calculation of swelling ratio follows the equation:

$$
SR = \frac{W_{after\,welling}}{W_{dry}}\tag{11}
$$

#### *Graphene oxide synthesis and characterization*:

We synthesized GO, as described in our previous paper.<sup>[68]</sup> In short, we oxidized graphite flakes using the modified Hummer's method.<sup>[69]</sup> A 9:1 mixture of concentrated  $H_2SO_4/H_3PO_4$  (360:40) mL) was added to a mixture of graphite flakes  $(3 \text{ g})$  and KMnO<sub>4</sub>  $(18 \text{ g})$ . The reaction was then heated to 50 °C and stirred for 12 h. Next, the reaction was cooled to room temperature and decanted onto ice (400 mL) with  $30\%$  H<sub>2</sub>O<sub>2</sub> (3 mL). For workup, the mixture was washed by extensive centrifugation cycles with water (12,000 rpm for 4 h) following by a dialysis for a week. When the solution pH was reached to  $pH = 7$ , it was centrifuged again (12,000 rpm for 4 h), and finally dried by lyophilization, yielding 5 g of graphite oxide. The obtained graphite oxide (yellowish solid) was exfoliated to GO sheets by half an hour of ultrasonication (brown solution).

The GO bulk was characterized by field-emission SEM (FESEM; FEI, Helios 600) operating at 5 keV. HRTEM measurements were carried out by a JEOL-2100 instrument operating at 200 keV. HRTEM samples were prepared by dispersing a section of GO bulk in ethanol followed by a gentle sonication for 15 minutes; then we dropped-cast a single droplet of the dispersion on a 300-mesh Cu lacey carbon grid (from SPI). Raman scattering data in the range of 1000- 3000 cm-1 was taken using a micro-Raman instrument (HORIBA Scientific LabRAM HR) in the air at RT. GO bulk sample was excited by a laser with an excitation wavelength of  $\lambda$ ex = 532 nm. AFM measurements were performed by using a Bio FastScan scanning probe

microscope (Bruker AXS). All images were obtained using soft tapping mode with a Fast Scan B (Bruker) silicon probe (spring constant of 1.8 N/m). The resonance frequency of the cantilever was approximately 450 kHz (in the air). The measurements were performed under environmental conditions. The images were captured in the retrace direction with a scan rate of 1.6 Hz. The resolution of the images was 512 samples/line. For image processing, we used Nanoscope Analysis software. The "flatting" and "planefit" functions were applied to each image. We measured GO flakes on  $Si/SiO<sub>2</sub>$  wafer, prepared by a single droplet from the same dispersion of GO as before.

#### *Preparation of nanocomposite hydrogels*:

GO sheets were resuspended in deionized water (0.2 % w/v) and de-agglomerated using an ultrasonic bath (Bransonic 2510, power: 20 W) for 5 minutes before their use. Then, the GO suspension was added to the GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> hydrogel solution to get a final concentration of 0.01% w/v and the blend  $(GG/PEGDA10/GO_5minUV_10minMgCl_2)$ was stirred at 75°C for 30 minutes for obtaining a homogenized solution. After that, the solution was poured into cylindrical holes (diameter: 6 mm; height: 5 mm) of a custom-made PDMS mold and photo-crosslinked upon UV exposure (same parameters described in section 5.2). Later, the ionic crosslinking was performed on some hydrogels by immersing the samples in MgCl<sub>2</sub> solution (1 % w/v in deionized water) for 10 min. Then, the ionic crosslinking was finalized by incubation in DMEM for 24 hours at 37 °C.

A preliminary evaluation of GO nanosheets dispersion in the nanocomposite hydrogel was carried out using an optical microscope (Hirox digital microscope, Hirox Co Ltd.).

#### *Characterization of GO-based nanocomposite hydrogels*

#### *Mechanical characterization*:

Young's modulus, fracture stress, fracture strain and toughness and swelling ratio were measured as described in section "Hydrogels mechanical characterization".

 

#### *Tribological test*:

Tribological tests on crosslinked hydrogels with  $(GG/PEGDA10/GO_5minUV_10minMgCl_2)$ and without GO (GG/PEGDA10\_5minUV\_10minMgCl2) (Figure 4a) were performed using a MCR 102 rheometer (Anton Paar) equipped with the Tribocell T-PTD200 and a Peltier temperature control, the H-PTD200 hood and a disposable measuring system shaft. Customized sample holders were used to keep fixed the specimens on the top of the Tribocell. The test geometry was set as flat-on-flat, in which a glass disc mounted on the shaft was interfaced to the surface of the hydrogel while lubricating the contact between these components with PBS. A representation of the tribological set-up is reported in Figure 4b. The glass disc was pressed against the sample by applying a normal force of 0.5 N. The torque was measured to realize the defined rotational speeds. During all experiments, the temperature was kept constant at 25°C. The test was divided into three parts:

1) The normal force was applied and remained constant ( $F_N = 0.5$  N) during the entire test. The time for adjusting the normal force was kept short (1.5 minutes) to avoid potential evaporation of the PBS buffer.

2) The first and second extended Stribeck curve runs were for running-in. The rotational speed was increased logarithmically from  $10^{-5}$  min-1 to 1 min-1. Therefore, the third run was used for the final evaluation.

3) The wear behavior was investigated at a constant rotation speed ( $Vs = 5$  mm s<sup>-1</sup>) for 30 minutes.

Subsequently, the speed was lowered again. Finally, the COF versus rotational speed and torque/thickness versus time were graphed. The COF was determined as the ratio between the frictional force and the normal force. Several tests consisted of three consecutive runs each were performed.

#### *Development of bilayered hydrogels*

#### *Preparation of the bilayered structure*:

Bilayered structures were fabricated using the custom-made PDMS mold (diameter: 6 mm; height: 5 mm), assuming the volumetric ratios defined into the modeling. Firstly, 113 µl of GG/PEGDA15\_5minUV\_10minMgCl<sup>2</sup> hydrogel solution was poured in the mold to achieve a height of 4 mm. Then, the hydrogel solution was photo-crosslinked upon UV light exposure for 5 minutes and after immersed in MgCl<sub>2</sub> solution (1% w/v, in deionized water) for 10 minutes. Subsequently, 28 µl of the nanocomposite GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub> was poured onto the crosslinked DC layer to achieve a final height of 5 mm. Then, the bilayer was photo-crosslinked upon UV light exposure for 5 minutes and after immersed in  $MgCl<sub>2</sub>$  solution (1% w/v, in deionized water) for 10 minutes. Finally, the crosslinked bilayered structures were incubated for 24 hours at RO in DMEM.

#### *Characterization of bilayered structures*:

Young's modulus, fracture stress and strain and toughness were measured after preparation, as previously described in section 5.3. The degradation of the bilayered structures was evaluated by incubating each sample at 37 °C in PBS (1X) and  $H_2O_2$  (3 % in PBS), according to the ISO10993-13, within an orbital agitator (711CT, Elettrofor).<sup>[70]</sup> A cyclic agitation at 2 rpm was imposed for the whole testing period. The material degradation kinetics was evaluated by monitoring the percentage of dry weight loss over time, after 2 weeks and 1 month. Four independent samples for each material type were tested.

The mechanical performance of the bilayered structures was also analyzed after 1 month of degradation in PBS and  $H_2O_2$  and compared to those owned by the material without being subjected to degradation.

#### *Wear test*:

Wear test was carried out to qualitatively evaluate whether the construct remained intact under repeated cyclical stresses. In particular, the wear test was performed using a four-station

displacement control knee joint simulator (Shore Western, Inc., Monrovia, CA, USA) that consents to simulate the knee movement up to four degree-of-freedom (DOF), as previously reported.[60] Briefly, the menisci were sterilized with ethylene oxide (ETO) gas and tested in conjunction with four CoCrMo alloy femoral and tibial components (size 2; Adler ORTHO, Milan, Italy), consolidated by compression molding (accordingly to ISO 5834/1-2) and ETO sterilized. Four bilayered structures were prepared keeping the same volumetric ratio as described previously into two symmetric holes (diameter: 12 mm; height: 7 mm) created into Ultra-High-Molecular-Weight-Poly-Ethylene (UHMWPE) mobile menisci (Genus mobile bearing, size 2, Adler ORTHO, Milan, Italy). Axial load was applied vertically (perpendicular to the tibial tray), oscillating between 168 and 2600 N following the calculated profile. The applied kinematics was derived from the displacement control simulator. In particular, the flexion/extension angle oscillating between 0◦ (neutral) and 60◦ (flexion) was synchronously with the load; the anterior/posterior translation oscillating between −6.0 mm (neutral) and 6.0 mm (posterior), and the intra/extra-rotation oscillating between −2.0◦ (extra-rotation) and 6.0◦ (intra-rotation). The test was performed using fetal bovine calf serum as a medium and using a frequency of  $1.0 \pm 0.1$  Hz, following the ISO 14243. Each test was performed in the presence of 25% sterile FBS balanced with deionized water and  $0.2\%$  NaN<sub>3</sub> to slow down bacterial growth, and 20 mM EDTA to minimize precipitation of calcium phosphate. The wear test lasted 100.000 cycles. Macroscopic and brightfield (4x) images of all pre-tested and tested sample was captured using a professional camera and optical microscope (Nikon 90i, Nikon, Japan). The brightfield images of medial and lateral condyles were captured in the center and on the edge of samples using an optical microscope.

An optical profiler (Leica DCM8) was used to analyze the roughness of the bilayered hydrogels in the hydrated state before and after the wear test. Z-stacks (scan area  $640 \times 480 \text{ }\mu\text{m}$ 2) were acquired, then data were converted into 2D images provided with a height-related greyscale and

the surface roughness was estimated. The analysis was carried out by using the Gwyddion software [\(http://gwyddion.net/\)](http://gwyddion.net/). The average roughness (Ra) was measured on the acquired images. Three Z-stacks were acquired for each sample type in its medial and lateral condyles. *Cytotoxicity evaluations*:

To perform *in vitro* cytotoxicity tests, the selected crosslinked GG/PEGDA15\_5minUV\_10minMgCl<sub>2</sub> and GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub> hydrogels were incubated for 24 hours in cell culture medium composed of DMEM with 10 % FBS and 100 U/mL penicillin/streptomycin (3 mL/sample). The supernatants were collected and used for the analysis of the cell behavior for the following assays.

Isolated human OA chondrocytes (from male, 59 years old) collected into the biobank of Laboratory of Immunorhematology and Tissue Regeneration (at Istituto Ortopedico Rizzoli) were thawed and expanded in culture. Chondrocytes at passage 3 were seeded in 8 wellschamber slides at a cell density of  $1.2 \times 10^4$  cells/well. Cells were left to adhere for 24 hours, then the culture medium was removed, and the supernatants collected from the GG/PEGDA15\_5minUV\_10minMgCl<sub>2</sub> and GG/PEGDA10/GO\_5minUV\_10minMgCl<sub>2</sub> hydrogels (500 µl/well) were transferred on the chamber slides. An untreated control was performed by simply changing the cell culture medium (DMEM, 10% FBS, 100 U/ml penicillin/streptomycin).

Cell morphology and viability were evaluated after 2 and 6 days by Live/Dead® cell viability assay.[71] Cells were washed twice with PBS and then incubated with Ethidium homodimer-1 (4 μM) and Calcein AM (2 μM) for 45 minutes at 37°C and 5% CO<sub>2</sub>. Labeled cells were then visualized under an optical microscope (Nikon, Japan) for evaluating general morphology, viable (green) and dead (red) cells.

The MTT metabolic activity assay determined the ability of viable cells to reduce the yellow tetrazolium salt (MTT) to blue-colored formazan crystals by mitochondrial enzymes, detectable

with a spectrophotometer. MTT metabolic activity assay was performed at the same time points (2 and 6 days). Cells were washed twice with PBS and then incubated with MTT for 3 hours at 37°C and 5% CO2. Afterward, MTT solution was removed and HCl (0.1 M) was added to cells to permit redox reaction in formazan. Absorbance at 570 nm was evaluated using a spectrophotometer (Tecan, Life Science).

#### *Statistical analysis*:

Normality tests (D'Agostino-Pearson) were performed on all experimental data to assess the type of data distribution. Results with normal distribution were expressed as average values  $\pm$ standard deviations. Data analysis was performed by applying the Student's t-tests to evaluate statistically significant differences between two sample types under analysis, while one-way ANOVA with Tukey's post-test was adopted for multiple comparisons. Statistical analysis was carried out using GraphPad Prism (v 6.0). The significance threshold was set at 5% (\*  $p < 0.05$ ) and 1% (\*\*  $p < 0.01$ ).

#### **Supporting Information**

Supporting Information is available from the Wiley Online Library or from the author.

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### Supporting Information

#### **Graphene Oxide-doped Gellan Gum-PEGDA hydrogel mimicking the mechanical and lubrication properties of articular cartilage**

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Gellan Gum (1.5 % w/v)

Figure S1. Young's modulus of Gellan Gum at 1.5 % w/v, ionically crosslinked with MgCl<sub>2</sub> (1)  $\%$  w/v) for 10 min.



Figure S2. Bright field image of the GG/PEGDA10\_5minUV\_10minMgCl<sub>2</sub> hydrogel with embedded GO.



**Figure S3.** (a-b) Captured images using professional camera (Nikon) performed to evaluate the printed filaments resolution of superficial and deep cartilage-mimicking hydrogels, respectively, using a 22 G needle tip (inner diameter  $= 410 \mu m$ ). Scale bars: 5 mm. c-d) Bright field images captured to measure the printed filaments width. Scale bars:  $250 \text{ µm}$ . The widths were  $548.45 \pm$ 7.67 µm and 589.72  $\pm$  20.26 µm for the superficial and deep cartilage-mimicking hydrogels, respectively. e-f) 3D printed structures made of superficial and deep cartilage-mimicking hydrogels. Scale bars: 2 mm.



#### **Table S1.** List of the tested formulations.



**Movie S1.** In the movie, the wear test is shown. The bilayered hydrogel poured into the meniscilike phantom is stressed by metal knee prosthesis following the ISO 14243. Each simulator's component is shown and described in Figure 6.b.

**Movie S2.** In the movie, the preliminary 3D-printability test using 3Dynamic Omega bioprinter is shown. The nanocomposite hydrogel (GG/PEGDA10/GO) was inserted into 20 mL plastic syringe used as cartridges and extruded through a 22 G tip (inner diameter  $= 410 \mu m$ ) at room temperature on a Petri dish (150\*15 mm) as printing surfaces. The separating distance between the needle and printing surface was set at  $3 \text{ mm}$ , the layer thickness was kept at 0.3 mm, and the printing speed was set at 25 mm min<sup>-1</sup>. Repetier-Host and Slic3r software were used to control the printer and printing parameters, respectively.