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## Chalcones. XVIII : Potential Germicides Derived from Hydroxy Acetonaphthones

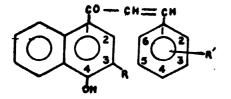
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Manuscript received 10 October 1972; revised 28 August 1973; accepted 11 August 1975

CHALCONES have variable germicidal<sup>2-17</sup>, bactericidal,<sup>1,19,20</sup> fungicidal<sup>11</sup> and carcinogenic<sup>12</sup> activity. These effects may be either due to interaction with the germs under investigation or for killing bacterias by inhibiting the supporting microorganisms or by direct action. Moreover, methoxy and hydroxy chalcones<sup>21</sup> can check up the destruction of adrenaline. Furthermore, naphthyl and phenanthryl chalcones possess potential bactericidal and carcinogenic<sup>22</sup> activity. Considering these facts various hydroxy naphthyl and nitro hydroxy naphthyl chalcones have been synthesised expecting enhanced germicidal effect. The present communication is in pursuit of our work<sup>4-8</sup> on syntheses of naphthyl chalcones of biochemical importance.

4-Hydroxy-1-acetonaphthone and 4-hydroxy-3-nitro-1-aceto-naphthone have been condensed in presence of 10% alcoholic caustic potash solution with various aryl aldehydes to form compounds of the type (I) and (II).



Type (I) R=H and R'=H; 2-,3-,-4-Cl; 2-,3-.4-Me; and 2-,3-,4-NO<sub>2</sub>.

Type (II)  $R=NO_2$  and R'=5-Br-2-OH; 2-Me; 5-Br-3:4-(OCH<sub>3</sub>)<sub>2</sub> and 5:6-benzo.

The condensation of 4-hydroxy-1-acetonaphthone and 4-hydroxy-3-nitro-1-acetonaphthone with chloro, bromo, methyl and methoxy benzaldehydes were obtained in good yields at room temperature within 24-50 hr. The interaction with nitrobenzaldehydes under the same conditions gave dark coloured products of unknown composition. The same reaction when carried at 10°-15° gave yellow coloured crystalline compounds in 75-80% yield. This abnormal behaviour of nitrobenzaldehydes may be due to its sensitiveness towards alkali solution at elevated temperatures.

The compounds were characterised by the vivid colours obtained on wetting with conc. sulfuric acid, elemental analyses, and by preparing a few 2:4-DNPs. The elemental analyses for C, H, and N for synthesised compounds were within  $\pm 0.5\%$  of the calculated value.

# Pharmacology

The agar-cup method was used to investigate germicidal activity against *E. coli* and *S. aureus* of the synthesised chalcones. The inocula were obtained from agar solidified nutrient broth media and growth was checked at  $(32^{\circ}\pm2^{\circ})$  after 24 hr. None of the compounds described in this paper were found to be significantly active at a concentration of  $100\pm15\mu$ g ml.

## Experimental

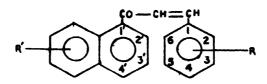
**Preparation** of 4-hydroxy-3-nitro-1-acetonaphthone<sup>23</sup>: It was prepared by the nitration of 4-hydroxy-1-acetonaphthone<sup>28</sup> in glacial acetic acid.

Preparation of Naphthyl chalcones – Naphthyl chalcone analogues were obtained by dropwise mixing 10% caustic potash solution (10 to 15 ml.) to an equimolar aldehyde free alcoholic solution of the required naphthyl ketone and aryl aldehydes at low temperature with constant stirring. The dark coloured reaction mixture on keeping at room temperature deposited crude compound. It was collected and crystallised from alcohol. The final purity was checked by thin layer chromatography.<sup>18</sup>

2:4-dinitrophenylhydrazone derivatives were prepared by grinding the ethanolic hydrochloric acid solution of the required compound with the reagent at 100° and orange to red coloured derivatives were collected and purified from ethyl acetate unless otherwise mentioned (Table 1). The derivatives were analysed satisfactorily for nitrogen.

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### TABLE 1—PHYSICAL PROPERTIES AND ANALYTICAL DATA OF NAPHTHYLCHALCONES



R'	R	Yield %	Colour & crystals	M. <b>P.</b> °C of		Halochromism with Conc.	Analyses
				Chal.	2:4-DNP	H <sub>2</sub> SO <sub>4</sub>	
4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH 4'-OH-3'-NO <sub>2</sub>	5-Br-3, 4-	54	Yellow rods Violet prisms Shininig rods L. Y. rods Y. Plates Y. rods C.Y. needles Y. Plates L. Y. Plates Ochre Plates Y. needles Tiny needles	191 217 156 194 196 157 177 214 190 181 159 94		Red Dark Red Red Blood red Brown Red Dark red Brown Bloodred Greenish Y. Greenish Blood red Pink	C,H C,H,N C,H,N C,H,N C,H,N C,H,N C,H C,H C,H C,H
4'-OH-3'-NO <sub>3</sub>	(OCH <sub>8</sub> ) 5:6-Benzo	57 51	Y. Globules Brown Needles	146 153		Dirty Yellow	C,H,N C,H

 All melting points are uncorrected since they were recorded in open capillary. Crystallised from : a = dioxan ; £ = Benzene C.=Canary, L.=Lemon, Y.=Yellow

### Acknowledgement

Thanks are due to Director, Harcourt Butler Technological Institute, Kanpur for providing necessary facilities, Mr. A. H. Siddiqui of I. I. T. Kanpur, for microanalyses, and Mrs. Bimla Misra for her assistance during the whole work.

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# Studies in some $\beta$ -Ketoester Complexes

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### Manuscript received 28 April 1975 ; revised 12 September 1975 ; accepted 13 September 1975

THE study of binary metal  $\beta$ -diketone complexes have been carried out earlier.<sup>1,2</sup> It is known that in  $\beta$ -diketone complexes there is metal ligand  $\pi$ -interaction and formation of a six membered planar ring with-