

Nanotubes from the misfit layered compound (SmS)_{1.19}TaS₂: Atomic structure, charge-transfer and electrical properties

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(Dedicated to Professor John B. Goodenough on the Occasion of his 100th year Birthday. Until hundred and twenty like twenty –a proverb in Hebrew)

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Abstract

Misfit layered compounds MX-TX₂, where M, T= metal atoms; X = S, Se or Te, (MLC) and their nanotubes are of significant interest due to their rich chemistry and unique quasi-1D structure. In particular, Ln-TX₂ (Ln= rare-earth atom) constitute a relatively large family of MLC, from which nanotubes have been synthesized. The properties of MLCs can be tuned by the chemical and structural interplay between LnX and TX₂ sub-layers and alloying of each of the Ln, T and X elements. In order to engineer them to gain desirable performance, a detailed understanding of their complex structure is indispensable. **MLC nanotubes are a relatively new-comer and offer new opportunities. In particular, like WS₂ nanotubes before, the confinement of the free-carriers in these quasi-1D nanostructures and their chiral nature offer intriguing physical behavior.** High-resolution transmission electron microscopy in conjunction with a focused ion beam are engaged to study SmS-TaS₂ nanotubes and their cross-sections at the atomic scale. The atomic resolution images distinctly reveal that Ta is in trigonal prismatic coordination with S atoms in a hexagonal structure. Furthermore, the position of the sulfur atoms in both the SmS and the TaS₂ sub-lattices, is revealed. X-ray photoelectron spectroscopy, electron-energy loss spectroscopy and X-ray absorption spectroscopy are carried out. These analyses conclude that charge transfer from the Sm to the Ta atoms leads to filling of the Ta 5d_{z²} level, which is confirmed by density functional theory (DFT) calculations. Transport measurements show that the nanotubes are semimetallic with resistivities in the range of 10⁻⁴ Ω□cm at room temperature and magnetic susceptibility measurements show a superconducting transition at 4 K.

Introduction

Misfit layered compounds (MLCs) are a class of two-dimensional (2D) materials receiving considerable attention due to their unique structure; crystallographic diversity and chemically tailorable characteristics (*vide infra*).¹⁻⁸ Among the MLCs, the chalcogenides-based MLCs are of special interest due to their metallic and semiconducting properties. The chalcogenides-based MLCs with the general formula $(MX)_{1+y}_m(TX_2)_n$ with $M = \text{Sn, Sb, Pb, Bi}$; Ln -rare earth, Y ; $T = \text{Ta, Nb, V, Cr}$ and X is a chalcogen atom S, Se, Te constitute a superstructure of alternating slabs of distorted rocksalt MX and hexagonal TX_2 structural units (see Fig. 1a-d). For the most common case, it is abbreviated as MX-TX_2 ($m=n=1$). Another shortened notation for MLC often used in the literature is $(O-T)$, indicating orthorhombic ($O = \text{MX}$) and trigonal prismatic ($T = \text{TX}_2$) coordination respectively.

With their unique structure and tunable properties, chalcogenides-based MLCs offer potential applications in thermoelectrics.⁹⁻¹³ In order to tune the desired character of MLC such as electronic conductivity or phonon scattering and improve their thermoelectric performance, a detailed structural understanding of each sublayer (MX and TX_2) atom by atom is highly desirable. Further, the periodic modulation due to non-stoichiometry and misfit strain can create strain waves and lead to vacancies in the structure (especially in the rocksalt unit), which may induce Anderson localization in these compounds.¹⁴ Advancement in sub-angstrom resolution electron microscopy and growth techniques^{15, 16} in recent times have prompted research into their local structure, charge transfer characteristics, and spectroscopic properties. The inception of misfit nanotubes over the last decade brought about another twist to their study.¹⁷⁻¹⁹

Due to their 1D structure and chiral nature, nanotubes of inorganic layered compounds, like WS_2 offer intriguing physical properties,^{20, 21} making the study of MLC nanotubes also highly warranted. MLC nanotubes of the kind SnS-SnS_2 were first obtained serendipitously by laser ablation of SnS_2 powder.²² Later-on rare-earth-based LnS-TaS_2 (Ln = rare-earth atom) MLCs and their nanotubes have been studied quite extensively.^{17, 23-27} LnS-TaS_2 MLC constitute a large family of compounds and are rather interesting owing to the significant charge transfer from the LnS unit to TaS_2 . The unique optical and magnetic properties of rare-earth compounds offer numerous potential applications, making the study of lanthanide-based MLC nanotubes highly warranted.

Among the LnS-TaS_2 family, SmS-TaS_2 is of special interest due to the exotic physical properties of their binary sulfide constituents (*vide infra*).^{7, 8, 28, 29} The binary SmS shows switching behavior, i.e. an ability to undergo reversible pressure-induced (at 6.5 kbar) semiconductor-to-metal transition at room temperature, which can be switched back to its original phase upon heating.^{28, 30-32} As a result Sm can accommodate two ground states configurations, either nonmagnetic Sm^{2+} (semiconducting SmS) with $4f^6$ (7F_0) configuration or in the magnetic Sm^{3+} (metallic SmS) with $4f^5$ ($^6H_{5/2}$) configuration.³³ Interestingly, in addition to the properties of SmS , the abundant electron coupling interactions in the 2H-TaS_2 lead to compelling physical phenomena such as layer-dependent charge density waves²⁹ and interfacial superconductivity.³⁴ The combination of these physical properties from the respective binary phases and the atomic-scale manipulation of charge transfer in $(\text{SmS})_{1.19}\text{TaS}_2$ can tune the electronic quantum phases, which could result in new physical observations. The binary SmS crystallizes in a rocksalt structure with the space group $Fm-3m$ ($a = 5.97 \text{ \AA}$, for Sm^{2+} , and $a = 5.57 \text{ \AA}$, for the high-pressure phase, Sm^{3+}), where Sm atoms are octahedrally coordinated to S atoms. 2H-TaS_2 is a layered compound with hexagonal lattice ($a = 3.314 \text{ \AA}$, $c = 12.097 \text{ \AA}$, $P6_3/mmc$) in which each Ta atom is bound to six sulfur atoms in a trigonal

prismatic configuration.³⁵ The structural motifs of SmS-TaS₂ can be described in the orthorhombic space group *Fm2m* with *FF* centering,⁷ in which SmS and TaS₂ are stacked layer by layer periodically along *c*-axis (*a*=3.29 Å, *b*=5.67 Å and *c*=22.50 Å). Given the unit-cell dimensions of each sub-lattice, the misfit ratio- *y* of the compound (SmS)_{1+y}TaS₂ can be found from the formula $1+y=2a_{\text{TaS}_2}/a_{\text{SmS}}$ to be 1.19. This factor represents also the deviation from the stoichiometry of the two sub-units in the MLC. Previously, the SmS-TaS₂ MLC was investigated using scanning tunneling micro(spectro)scopy and it was concluded that the outermost layer on the surface of the cleaved crystal is SmS.³⁶ On the other hand, high-resolution transmission electron microscopy (HRTEM) revealed that the outermost layer of MLC is the TaS₂ layer.³⁷ Also, the (SmS)_{1.19}TaS₂ surface was found to be semimetallic.^{36, 38}

The charge transfer, from LnS to TaS₂ stabilizes MLC structures and alters the electronic properties of the two sub-units, appreciably.³⁹ The amount of charge transfer can be tuned by alloying the rocksalt (LnS) unit with other rare earth or hetero-atoms.^{40, 41} The role of the charge transfer from the Ln atoms to the Ta atoms has been elucidated.^{25, 41} In particular, since the work function of the LnS sub-unit is smaller than that of the hexagonal TaS₂, charge transfer occurs from the rare-earth atom to the partially occupied 5*d*_{z²} level of the tantalum atom. This charge transfer modifies the effective valence state of the rare-earth atom (2+) closer to the more stable 3+ state. Furthermore, the 5*d*_{z²} level of the Ta atom, which dominates the density of states (DOS) at the Fermi level (*E_f*), is getting almost filled by the charge transfer.^{39, 42} This effect leads to an increase in the density of occupied states at the *E_f* and reduces the hole conductivity. The charge transfer also stabilizes 2H (trigonal prismatic coordination) polytype of TaS₂ preventing its transformation into 1T with octahedral coordination and the ensuing charge density wave transition. This conjecture was supported by Raman spectroscopy of individual nanotubes.²⁷ Nonetheless, direct evidence from transmission electron microscopy with the sub-angstrom resolution was missing so far and is presented here.

In the present work, nanotubes (and flakes) of the misfit compound (SmS)_{1.19}TaS₂ were studied in detail to address the structural aspects atom by atom and correlate it with its properties. High-resolution transmission electron microscopy with the sub-Ångstrom resolution was employed here to elucidate the atomic arrangements in the lattice of (SmS)_{1.19}TaS₂ nanotubes, which was not available before. Moreover, by using dual-beam focused ion beam microscopy (FIB), lamellae of such nanotubes were prepared to enable direct imaging of the superstructure from the axial *b* direction (nanotube growth axis). This analysis yields the lattice structure of such MLC nanotubes in unprecedented detail and provides a pathway to correlate the structure with the physical properties of such 1D nanostructures.

X-ray photoelectron spectroscopy (XPS) and high-resolution electron energy loss spectroscopy (HR-EELS) were employed here to elucidate the core levels and valence band structure of the (SmS)_{1.19}TaS₂ MLC nanotubes and flakes, and pure 2H-TaS₂. It was shown that the 5*d*_{z²} level of Ta in MLC is getting filled-up upon coupling with SmS and the charge transfer from Sm→Ta. The filling of Ta 5*d*_{z²} and valence conversion of Sm²⁺ to Sm³⁺ upon charge transfer was further confirmed by X-ray absorption (XAS) studies. These observations were validated with DFT calculations. Magnetic susceptibility measurements at low temperature show that (SmS)_{1.19}TaS₂ is superconducting below 5K while the four-probe electrical measurements at room temperature show a semimetallic behavior of the nanotubes.

Results and discussion

As illustrated in **Fig. 1a** and **e**, the rocksalt SmS structure in the MLC is modulated compared to pristine SmS (**Fig. 1d**) whereas the hexagonal TaS₂ is almost undistorted resembling bulk 2H-TaS₂. The SmS slab is made of only half a unit cell (along the *c*-axis) between the two TaS₂ units of MLC. The Sm is coordinated to only five sulfur atoms (instead of six in the binary SmS) within the unit (**Fig. 1e**) and consequently may have a strong dipolar interaction with sulfur atoms of the adjacent

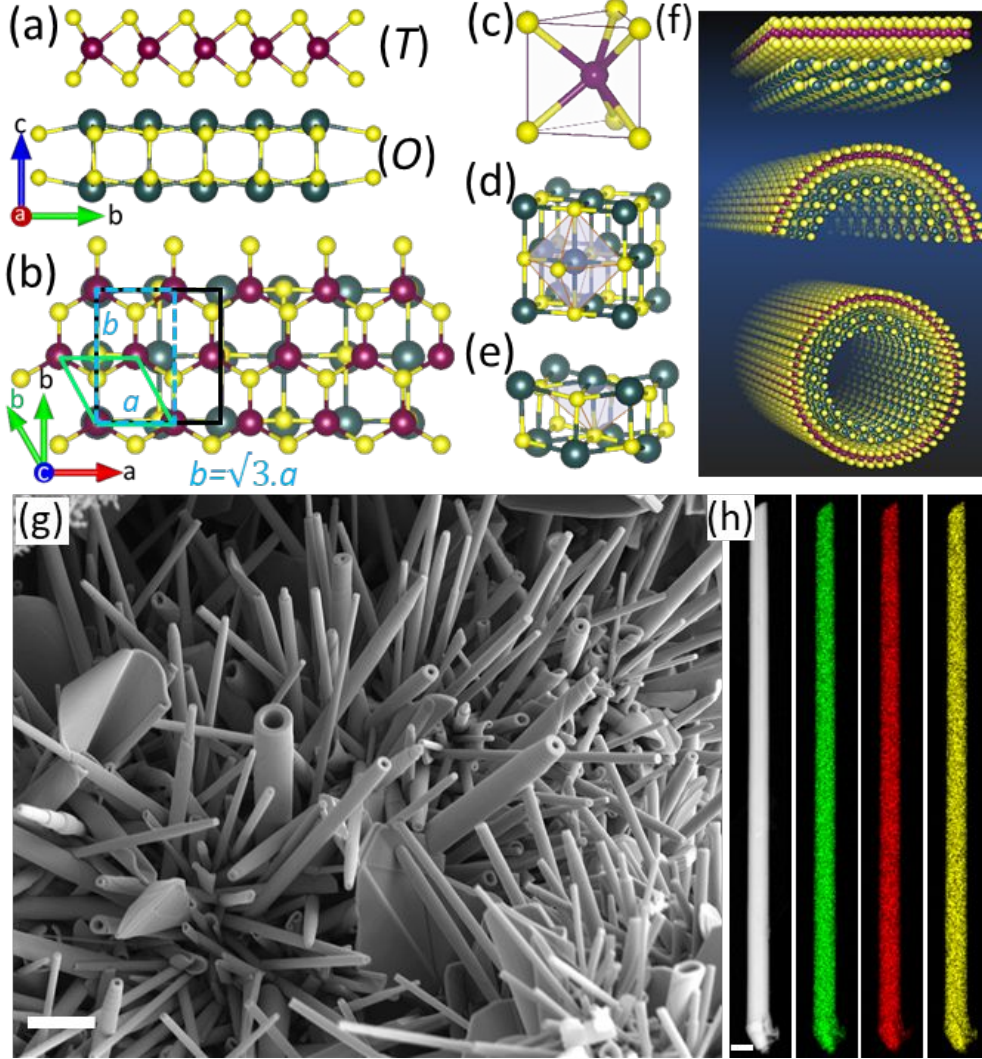


Fig. 1. Structure, morphology, and chemical analyses of misfit $(\text{SmS})_{1.19}\text{TaS}_2$. Schematic drawing of the MLC lattice structure projected in *a*-*b* plane (a) and along the *c*-direction (b), respectively. In *a*-*b* plane, the incommensurate-*a*, and commensurate-*b* crystallographic axes of the ortho-hexagonal unit cell ($b = \sqrt{3}a$) were marked in cyan color. The O and T represent orthorhombic and trigonal prismatic coordination of Sm in the rocksalt unit and Ta in the hexagonal unit and are graphically represented in (c) and (d), respectively. Single-layer MLC slab constitutes half a unit cell of rocksalt structure and is graphically shown in (e) with corresponding coordination of Sm. (f) Graphical rendering of the formation mechanism of an MLC nanotube via misfit strain relaxation (folding) and seaming of the rim atoms. (g) SEM image of $(\text{SmS})_{1.19}\text{TaS}_2$ flakes and nanotubes obtained by quenching high-temperature CVT reaction to ambient conditions, scale bar is 2 μm . (h) STEM HAADF images of a single $(\text{SmS})_{1.19}\text{TaS}_2$ MLC nanotube and corresponding SEM-EDS chemical maps (Sm- green, Ta- red, and S- yellow), scale bar 200 nm.

TaS₂. MLC compounds tend to roll into nanotubes due to the misfit strain (between the rocksalt MX and hexagonal TX₂ layers) and seaming of the dangling bonds at the rim atoms.⁴³ The folding mechanism is schematically depicted in **Fig. 1f**. The MLC nanotubes of (SmS)_{1.19}TaS₂ were produced by a well-established chemical vapor transport protocol with slight modification in growth temperature to improve the yield. SEM micrographs of as-obtained (SmS)_{1.19}TaS₂ powder displayed in **Fig. 1h** and **S1a** show nanotubular structures as a major product. The other common byproducts such as flakes and nanoscrolls with similar chemical compositions (as confirmed by SEM-EDS) were also observed. Semi-quantitative EDS analysis showed that the stoichiometry of the nanotubes is (SmS)_{1.05}TaS₂. The majority of the nanotubes display constant diameter along the tube axis whereas few of them showed telescopic contour with varying diameters along the tube axis. SEM analysis revealed that the MLC tubes are grown perfectly under the established reaction conditions. It will be interesting to know what thermodynamic and kinetic factors could lead to either type of morphology such as nanotube/nanoscroll but none of the *ex-situ* measurements could reveal those conditions. Statistical analysis of the nanotubes size distribution was carried out using SEM images. This analysis revealed that the nanotubes display varying lengths and the majority of them falling within the range of 100 to 200 nm in diameter. **Fig. 1h** shows low magnification scanning transmission electron microscopy (STEM) and STEM-EDS chemical maps of single (SmS)_{1.19}TaS₂ nanotube (see also **Fig. S1b**). The nanotube is 200 nm in diameter and the size of the tube is uniform along its entire length. The chemical maps reveal the uniform distribution of samarium (red), tantalum (green), and sulfur (yellow) elements throughout the tube. Quantitative STEM-EDS analysis of several nanotubes and flakes (**Fig. S1c,d**) show the stoichiometry (SmS)_{1.08}TaS₂ and (SmS)_{1.05}TaS₂, respectively, which are quite comparable to the theoretical value 1+y=1.19. Deviations are attributed to the experimental error as well as structural defects (*vide infra*). No signal, which could be associated with oxidation of the nanotube core was obtained. Slight and somewhat nonuniform surface oxide layer (< 1 nm) was occasionally observed.

X-ray diffraction (XRD) of (SmS)_{1.19}TaS₂ powder exhibits a strong diffraction pattern with a highly preferred orientation along the *c*-direction (see **Fig. S2**). The observed patterns are consistent with an earlier report⁷ and the interlayer spacing calculated from the (002) periodicity is 11.3 Å equivalent to *c*/2 of *FF* centered (SmS)_{1.19}TaS₂ misfit lattice. The strongly preferred orientations along <001> direction are characteristics of freestanding (*O-T*) superstructures which are grown seamlessly along the *c*-direction. In addition to the regular (001) peaks with periodicities of 11.3 Å, the XRD pattern exhibits weaker reflections with periodicities of 17.2 Å, typical for the (*O-T-T*) superstructure.³⁷ Indeed the purely (*O-T-T*) order in a nanotube/flakes generates a new compound i.e. (SmS)_{1.19}(TaS₂)₂. Here, the (*O-T-T*) order was interspersed, sporadically, between the (*O-T*) superstructure and was probably caused by the defects in the SmS unit. The relative intensity of (002) planes of (*O-T*) and (*O-T-T*) structure in XRD yields 4% of (*O-T-T*) layers in the overall compound. It is worth mentioning that a systematic transformation from (*O-T*), i.e. (LaS)_{1.14}(TaS_xSe_{1-x})₂ to (*O-T-T*)- (LaS)_{1.14}(TaSe₂)₂ MLC occurred upon increasing the selenium to the sulfur ratio in the asymmetric misfit system.³⁷ Further, the X-ray diffraction data (**Fig. S2**) does not reveal any characteristic peaks for impurities, such as binary sulfides (SmS/TaS₂) and elemental Sm/Ta/S in the reaction products within its sensitivity limit.

Transmission electron microscopy (TEM) analysis reported in previous studies²⁷ did not have sufficient resolution to reveal the finest details of the structure of these nanotubes. Therefore, in the present work, the nanotubes were analyzed via TEM techniques with the highest possible resolution. Several (SmS)_{1.19}TaS₂ nanotubes were examined here. The results for one such

nanotube are displayed in **Fig. 2** (see also **Fig. S3** and **S4**). A low magnification image of the nanotube in the inset of **Fig. 2a** shows a constant diameter of 170 nm along its entire length. The

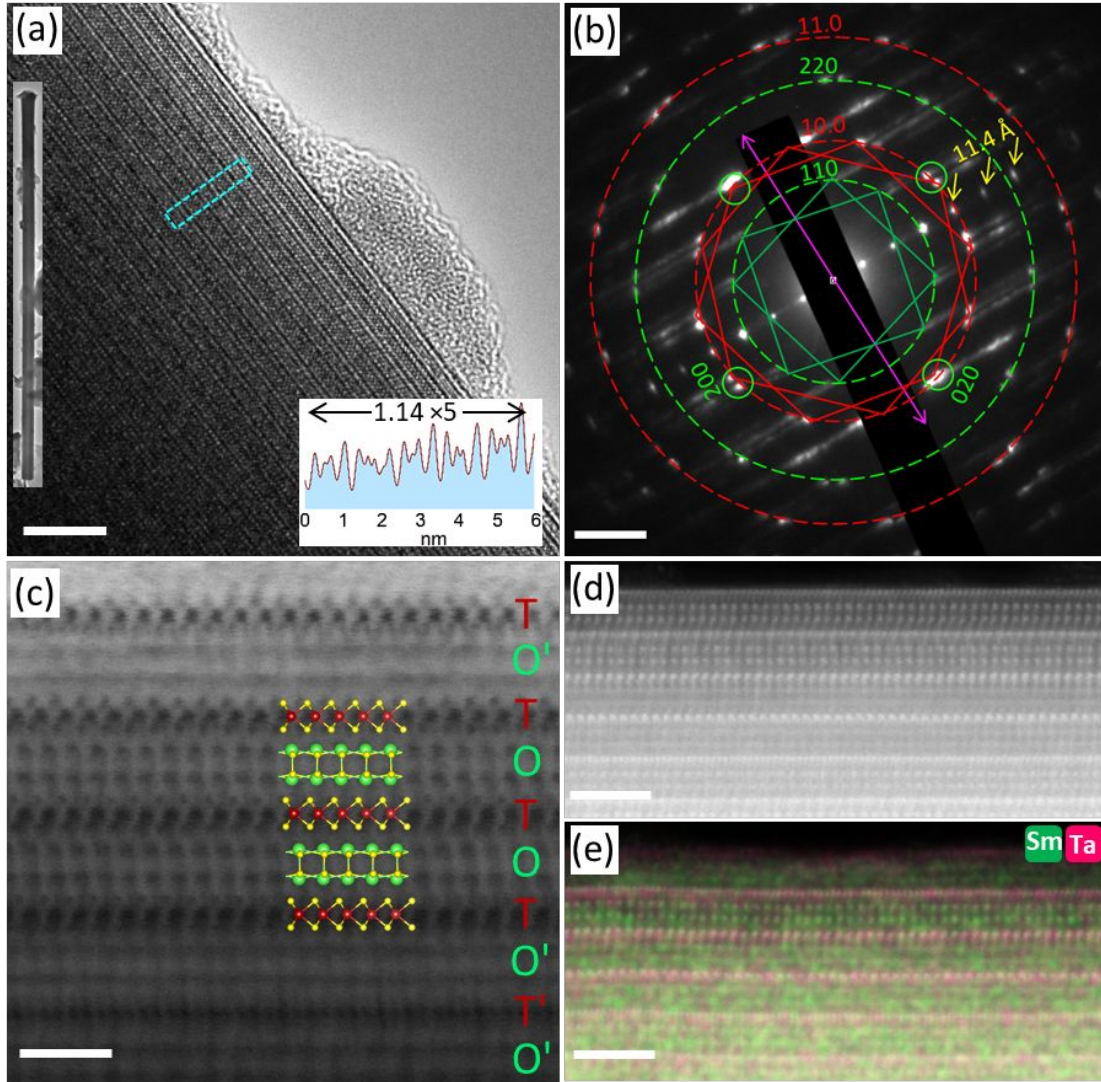


Fig. 2. TEM images, electron diffraction and chemical analyses of a $(\text{SmS})_{1.19}\text{TaS}_2$ nanotube. (a) High-resolution TEM image. The periodic stacking of SmS and TaS₂ layers in the misfit structure is revealed with TaS₂ as the outermost layer. Scale bar is 5 nm. The low magnification image of the nanotube (diameter 170 nm) and an intensity line profile perpendicular to the tube axis are shown in insets. (b) SAED pattern acquired from part of an individual nanotube. The sets of diffraction spots corresponding to SmS and TaS₂ are marked with green and red dotted circles. The respective Miller indices are indicated. Small yellow arrows indicate basal plane reflections and the tubule axis is marked by the purple double-headed arrow. The two sets of four pairs (110) reflections of the rocksalt SmS subsystem are marked by rotated green squares. The two sets of six pairs (10.0) reflections of the orthohexagonal TaS₂ sub-lattice are marked by red hexagons. These sets of reflections are rotated by 30° with respect to each other. (c) Atomically resolved HR-STEM image of a few (O-T) layers near the surface of the nanotube. The scale bar is 1 nm. The corresponding atomic model is overlaid on the HR-STEM image. The 30° rotation of (O-T)(O-T)' layers is clearly visible. Yellow, red, and green spheres represent S, Ta, and Sm atoms respectively. (d) STEM-HAADF image and overlaid (e) STEM-EDS elemental maps Sm (red) and Ta (cyan) of a few layers from the surface of the tube, scale bar 2 nm.

high-resolution TEM image of the SmS-TaS₂ superstructure reveals a periodic stacking sequence of SmS and TaS₂ along the c-direction. The outermost layer is TaS₂, which is true for all the

nanotubes analyzed here as well as other Ln-based misfit nanotubes. Note that potential sources of damage prior to investigation, such as plasma cleaning, were avoided to preserve the pristine surface of the nanotubes. Previously, an STM study of the cleaved surface of SmS-TaS₂ shows that SmS is the surface layer,³⁶ but that observation may have resulted from the cleavage process.³⁶ The surfaces of the nanotubes analyzed here are almost intact and unlike LaS-TaS₂,⁴⁴ do not show any, or little oxidation. The intensity profile drawn perpendicular to the tube axis (along *c*) shown in the inset reveals that the periodicity of the single-layer MLC unit is 11.4 Å, which is in close agreement with the (001) reflection of the corresponding (*O-T*) structure observed from XRD.

Selected area electron diffraction (SAED) pattern of an individual (SmS)_{1.19}TaS₂ nanotube is displayed in **Fig. 2b**. The intense and distinguished spots indicate ordered stacking of the SmS and TaS₂ layers in the misfit lattice. The ED pattern reveals a pair of four-fold and a pair of six-fold periodicity for rocksalt SmS and hexagonal TaS₂ sublattices respectively, which are rotated by 30° with respect to each other. Two sets of four pairs of spots that are azimuthally equally distributed with the interplanar spacings of 1.85 Å and 3.7 Å (on the green circle) were assigned to the (110) and (220) planes of rocksalt SmS. Two sets of six pairs of spots (on the red circles) with the interplanar spacings 1.6 and 2.85 Å are attributed to hexagonal TaS₂. These azimuthally equally distributed sets of quartet and sextet spots (marked by green squares and red hexagons-rotated by 30°) indicate two folding vectors for both SmS and TaS₂ layers in the nanotube. This results in the formation of super periodicity of the kind (*O-T*)(*O-T*)' and is reminiscent of the *CF* and *FF* superstructure in MLC (*vide infra*).⁴⁵ The angular splitting of the spots into pairs is due to a small chiral angle of the nanotube (≈ 3 deg). The (020) reflections of SmS, which coincide with the (10.0) reflections of TaS₂ (marked by a small green circle), reveal the common commensurate *b*-axis. The (020) and (200) reflections of SmS are approximately placed on the same dotted circle and are perpendicular to each other. This observation confirms that the *a* and *b* lattice parameters of SmS are (almost) equal. In the present nanotube, the common commensurate *b* axis appears 15° off from the nanotube growth axis. In many other nanotubes, the commensurate *b* axis coincide with the tube axis (see **Fig. S3**). The basal plane reflections indicated by yellow arrows reveal the periodicity of 11.4 Å along the *c*-direction. In general, the nanotubes of (SmS)_{1.19}TaS₂ prepared in this study exhibit a high degree of crystallinity, and the superstructure of SmS and TaS₂ is well preserved.

The atomic structure of the nanotubes was analyzed with high-resolution scanning transmission electron microscopy (HR-STEM) and STEM- energy-dispersive X-ray spectroscopy (EDS) (**Fig. 2c-e** and **Fig. S4**). The atomically resolved STEM bright-field (BF) image in **Fig. 2c** reveals that the nanotubes are comprised of (*O-T*) and (*O-T*)' layers with high stacking order. The contrast difference between (*O-T*) and (*O-T*)' layers is clearly visible indicating the two different crystallographic orientations of the misfit layers. The interatomic distances in the projection reveal different orientations of the TaS₂ in the nanotube, the viewing directions are along <10.0> and <11.0>. Similarly, the orientation of the SmS layer can be linked to <100> and <110> directions which are in line with the ED results. In many nanotubes of this kind, the (*O-T*) pairs seem to be tilted in 30° with respect to the adjacent (*O-T*)' wall forming thereby a super-periodicity of the kind (*O-T*)(*O-T*)' as confirmed by ED. The reason for the tilting of (*O-T*) layers is unknown, though. A possible explanation is that quenching from the high temperature to room temperature in the synthesis may have arrested the reorganization of layers in the nanotube. Alternatively, the misfit lattice may prefer this unique orientation to minimize the misfit strain. The two different orientations in the nanotube tend to alternate along the common *c*-axis leading to double periodicity with 23 Å as the unit distance of the supercell (**Fig 2c** and **2d**). The first two layers being

(*O-T*) i.e. $\langle 10.0 \rangle$ TaS₂ and $\langle 100 \rangle$ SmS and the adjacent layers are oriented 30° designated as (*O-T*)', i.e. $\langle 11.0 \rangle$ TaS₂ and $\langle 110 \rangle$ SmS, respectively. There is stronger contrast of the (*O-T*) layers over the (*O-T*)' layers. The difference in contrast is related to orientation-dependent channeling phenomena and not to composition. Careful analysis of the atomic arrangement reveals that TaS₂ lattice consists of a chevron-type pattern where the sulfur atoms are coordinated with the Ta atoms in trigonal prismatic fashion, i.e. 1H polytype. Generally, quenching of TaS₂ from such a high temperature would lead to 1T polytype.⁴⁶ The stability of 1H-TaS₂ in the (SmS)_{1.19}TaS₂ MLC can be attributed to strong charge transfer from SmS to TaS₂, which is further confirmed by spectroscopic techniques (*vide infra*). The SmS subunit is arranged in a distorted rocksalt structure. In the $\langle 100 \rangle$ viewing direction of the SmS the sulfur atoms sit at a small projected distance from the samarium, and consequently in this orientation sulfur and samarium positions are not as well resolved as below (see Fig. 4b). The representative atomic models of $\langle 10.0 \rangle$ TaS₂ and $\langle 100 \rangle$ SmS are overlaid on the HR-STEM image for clarity. HR-STEM (DF) image and the corresponding HR-STEM-EDS

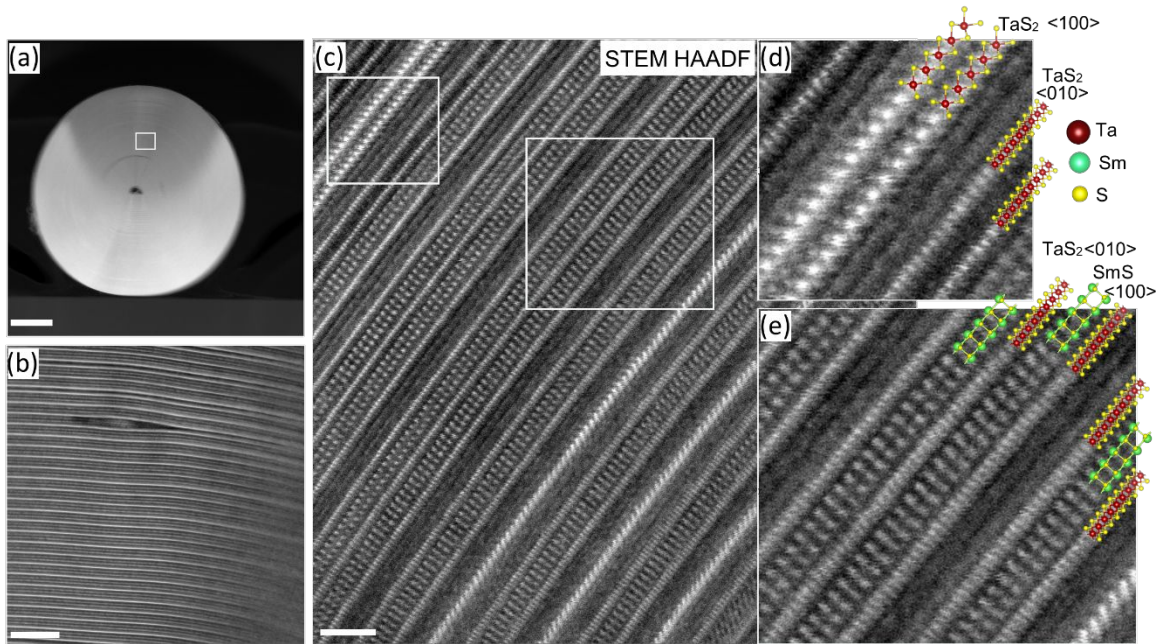


Fig. 3. STEM analysis of cross-sectioned FIB lamella of (SmS)_{1.19}TaS₂ nanotube (a) Low magnification STEM-BF image of nanotube lamella, scale bar is 100 nm. (b) High magnification STEM-HAADF image of a portion of the cross-section showing the basic MLC backbone (*O-T*) structure, the defect layer of SmS ending abruptly is seen, scale bar is 5 nm. (c) Atomic resolution STEM-HAADF image revealing the rocksalt SmS and trigonal prismatic TaS₂ layers, the rotation between the layer (*O-T*) and (*O-T*)' is evident, scale bar is 2 nm. *The SmS and TaS₂ layers which are in-phase to the electron beam focus correspond to (*O-T*) and those are in defocus correspond to (*O-T*)' orientation.* (d and e) Atomic-resolution images from the area marked in image (c), projections of atomic models of double hexagonal TaS₂ layers and distorted rocksalt SmS are overlaid on the STEM image.

maps overlaid on the STEM image are displayed in Fig. 2d-e. The STEM-EDS confirms that Sm and Ta are in antiphase relationship with each other. Since the sulfur atoms are coordinated to both samarium and tantalum atoms, the sulfur maps show a uniform distribution due to channeling phenomena in the vicinity of the heavier Ta and Sm atomic helices and do not yield any extra information here.

To understand the structural details further, nanotubes were sliced into thin lamella using a dual-beam focused ion beam (FIB) microscope and transferred onto a TEM grid. **Fig. 3a** shows a low-magnification STEM-ADF image of one such cross-section (the tube diameter is around 500 nm). A magnified image in **Fig. 3b** shows an edge dislocation and adjacent to it, a misfit structure with double layer periodicity (*O-T-T*). It is believed that the two features occur in the vicinity to each other due to strain fields induced by the dislocation. Generally, the confined volume and curvature of the nanotubes, confer larger density of defects than in flakes.³⁷ Moreover, the fact that each layer in the MLC nanotube contains a different number of atoms, induces strain, which can be relaxed via defect formation. **Figs. 3c-e** show high-magnification STEM-BF images of a portion of the lamella. Indeed the top exploded area includes three such repeating (*O-T-T*) units (**Fig. 3d**). The repeating two TaS₂ layers appear in trigonal prismatic coordination, i.e. the 2H polytype arrangement. The rotation of 30° in the (*O-T-T*) arrangement is also evident, i.e. TaS₂ <01.0>/SmS<100> and TaS₂ <01.0>/SmS<110>. The fogging of some of the MLC layers in the HRSTEM images can be ascribed to the stacking faults, i.e. rotation of the (*O-T*) pair with respect

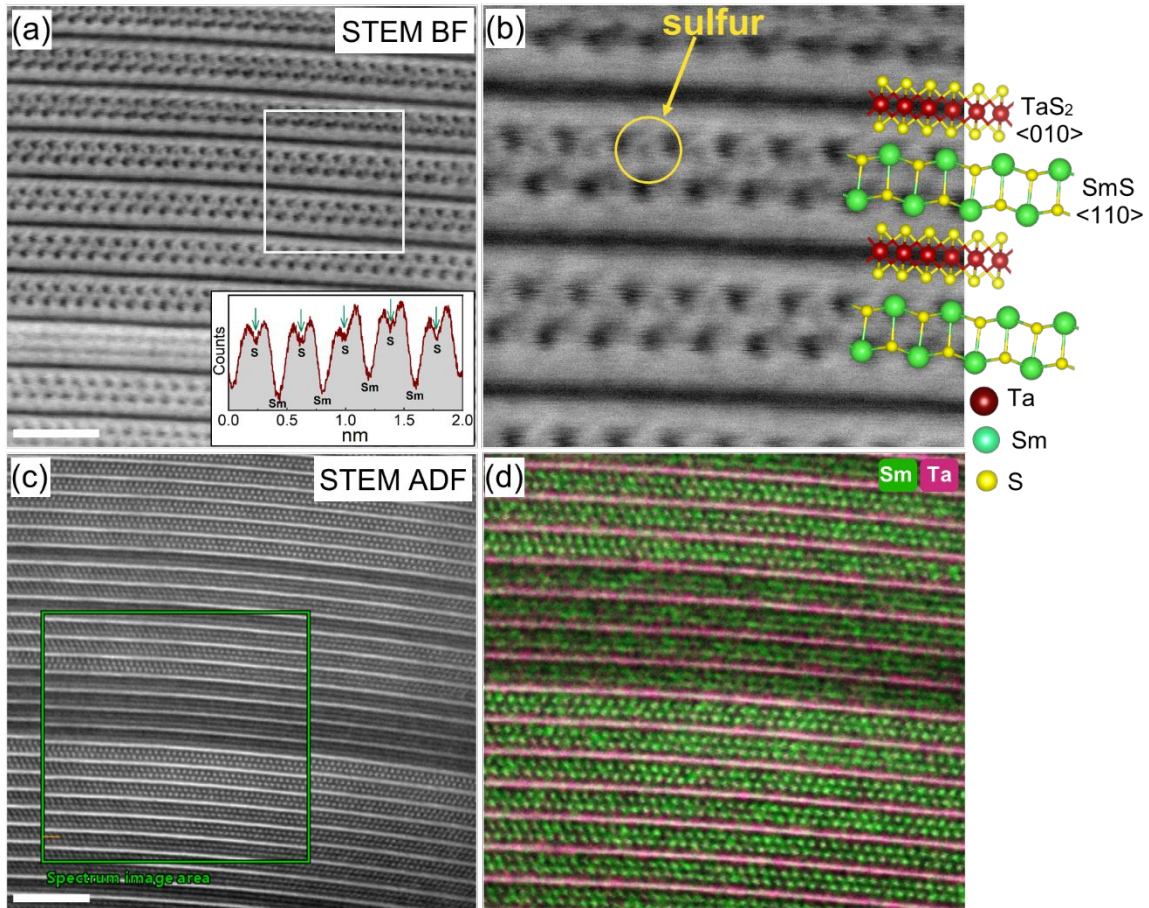


Fig 4. (a) Atomic resolution HR-STEM-BF image of a portion of the nanotube lamellae, a magnified image in (b) succinctly reveal the sulfur atoms adjacent to samarium atoms in the rock salt unit. The intensity profile is drawn from one of the SmS layers shown in the inset of (a) and it shows clear modulation of S and Sm, scale bar 2 nm. (c-d) STEM-ADF and corresponding STEM-EDS analyses of (SmS)_{1.19}TaS₂ nanotube lamellae. EDS mapping shows the clear antiphase correlation between Sm and Ta atoms, scale bar 5 nm. Sulfur atoms were distributed uniformly across the lamellae and are presented in the SI, Fig S6.

to the (*O-T*)' or (*O-T-T*) and (*O-T-T*)' layers. The different crystallographic orientation of the two pairs leads to a small scattering of the incoming electron beam and blurring of the TEM image. On

the other hand, the ordered structure from other portions of the lamellae consists of purely (*O-T*) superstructures (**Figs. 3e** and **S5**), but the orientation of the layers varies from one pair to another. This observation reinforces the proposition that the (*O-T-T*) superstructure in this lamella indeed resulted from the defects formed by the SmS layer. Schematic rendering of the SmS $\langle 110 \rangle$ and TaS₂ $\langle 01.0 \rangle$ atomic models corresponds very well to the underlying STEM-BF image.

Figs. 4 a, b and **Fig. S5** show HR-STEM-BF images and exploded views of a lamella from a different region of the same nanotube. The (*O-T*) superstructure order is strictly followed, extended over many layers. Since the SmS layer is oriented along $\langle 110 \rangle$, the Sm and S atoms in the rocksalt structure are skewed and are therefore clearly visible in a magnified STEM-BF image (**Figs. 4b**). The intensity profile drawn from one of the SmS layers (insert of **Fig 4a**) reveals clear modulation of the S and Sm atoms. Note that since the rocksalt structure is distorted, the samarium and sulfur atoms do not share a common plane, i.e. the sulfur atoms are displaced towards the center of the SmS subunit. The high-resolution STEM-EDS images (**Fig. 4d** and **S6**) show that the two Sm and Ta layers are in antiphase relationship with each other. The even sulfur distribution on the entire lattice is evident (see **Fig. S6a**). The intensity profiles of the EDS chemical maps presented in the **Fig. S6b** show two Sm atomic layers (of the rocksalt unit) in between the Ta layers and their periodic modulation. The EDS sulfur profile reveals two atomic layers of trigonal prismatic coordinated sulfur adjacent to the Ta atomic layer, and the sulfur from SmS layers are also distinguishable from that of the TaS₂ layer. **No indication for any oxide formation in the core of the nanotube was observed.** The HR-STEM and STEM-EDS results presented here revealed atomically each Sm, Ta, and S atoms and their position in the MLC lattice as well as their distribution, which was not available before. The Ta-S are in trigonal prismatic coordination in the hexagonal lattice while the Sm-S atoms are coordinated in a distorted rocksalt lattice with orthorhombic symmetry. Equipped with these insights, the stability of the misfit lattice gained upon charge transfer from the SmS slab to TaS₂ was studied by combined XPS, EELS, and XAS analyses. These analyses were further corroborated by theoretical calculations, thereby shedding light on the structure-property relationships.

(SmS)_{1.19}TaS₂ powder containing nanotubes and flakes was densely spread over a carbon tape and XPS spectra were collected from the sample surface. **Fig. 5** presents XPS measurements of the (SmS)_{1.19}TaS₂ misfit samples in comparison with 2H-TaS₂ flakes. The core-level Ta 4*f* spectrum of pure TaS₂, **Fig. 5a**, shows pronounced oxidation as evident from the high energy shoulders at ~24 and 26 eV of Ta 4*f* doublet.⁴⁷ This signal is believed to arise from the platelet edges that point upwards in the powder grains. Curve fitting details of the Ta 4*f* doublet in TaS₂ and (SmS)_{1.19}TaS₂ are presented in **Fig. S7**. Remarkably, the Ta line appears far more homogeneous in the misfit sample, which indicates that nanotubes suffer much less edge oxidation than the (TaS₂) platelets. The sulfur S 2*p* core-level spectra of TaS₂ and (SmS)_{1.19}TaS₂ misfit are presented in **Fig. 5b**. Co-existence of the two misfit constituents is manifested by the S 2*p* spectrum (see also SI **Fig. S8**). Here, as expected two leading sulfur chemical states are observed, attributed to S in the trigonal prismatic TaS₂ and the rocksalt SmS ingredients. For the reference TaS₂ sample, the SmS component is missing from the S 2*p* line, while other components arise due to platelet edge oxidation (see also Ta 4*f* line). The 3*d* core-levels of samarium, observed at 1084 (Sm 3*d*_{5/2}) and 1110 eV (Sm 3*d*_{3/2}), are consistent with literature reports of the Sm³⁺ chemical state (see **Fig. S9**).⁴⁸ No signature of Sm²⁺ at energies 1073 and 1100 eV, were observed.⁴⁸ Earlier reports of single-crystal (SmS)_{1.19}TaS₂ suggested Sm²⁺-Sm³⁺ valance fluctuation, but no evidence of that notion is seen here. Further support for this conjecture, i.e. the existence of pure Sm³⁺ and the absence of

valence fluctuations, was provided by the XAS analysis (*vide infra*). Quantitative analysis of the chemical composition showed a Sm/Ta ratio of 1.12, which is slightly lower than the theoretical value- 1.19.

The XPS valence band spectra presented in **Fig. 5c** show a clear difference between the reference (TaS_2) and the $(\text{SmS})_{1.19}\text{TaS}_2$ structure. The missing feature at 4-8 eV is attributed to the lack of Sm contribution. Differences at the top of the valence band, just below the Fermi energy (zero binding energy), are seen as well. The valence band spectrum of $(\text{SmS})_{1.19}\text{TaS}_2$, exhibits broadening plus a shift of the MLC spectrum to higher binding energies compared to the pure TaS_2 . This result suggests that the Fermi level of the misfit was ‘pushed’ upwards and the misfit became less *p*-type, as compared to TaS_2 . These differences are complemented by the work function (WF) shift shown in **Fig. 5d**. Here, a clear difference in the onset of the secondary electron emission, about 270 meV in magnitude, is seen between the reference and the MLC. The latter result suggests that the WF of bulk SmS is lower than that of TaS_2 , hence when brought into contact with TaS_2 (in the MLC), electron density is expected to be transferred from the SmS to the TaS_2 layers. Consequently, the valence band of the semiconducting SmS is partially depopulated (by charge transfer) and thus becomes conductive as well. Interestingly, the Ta $4f_{7/2}$ binding energy of the TaS_2 -related component in the misfit compound is similar to the one in pure TaS_2 . This fact indicates that the donated charge contributes a charge density to the Ta atoms exclusively, such that both the Fermi energy and the Ta-core levels are equally affected (see also SI).

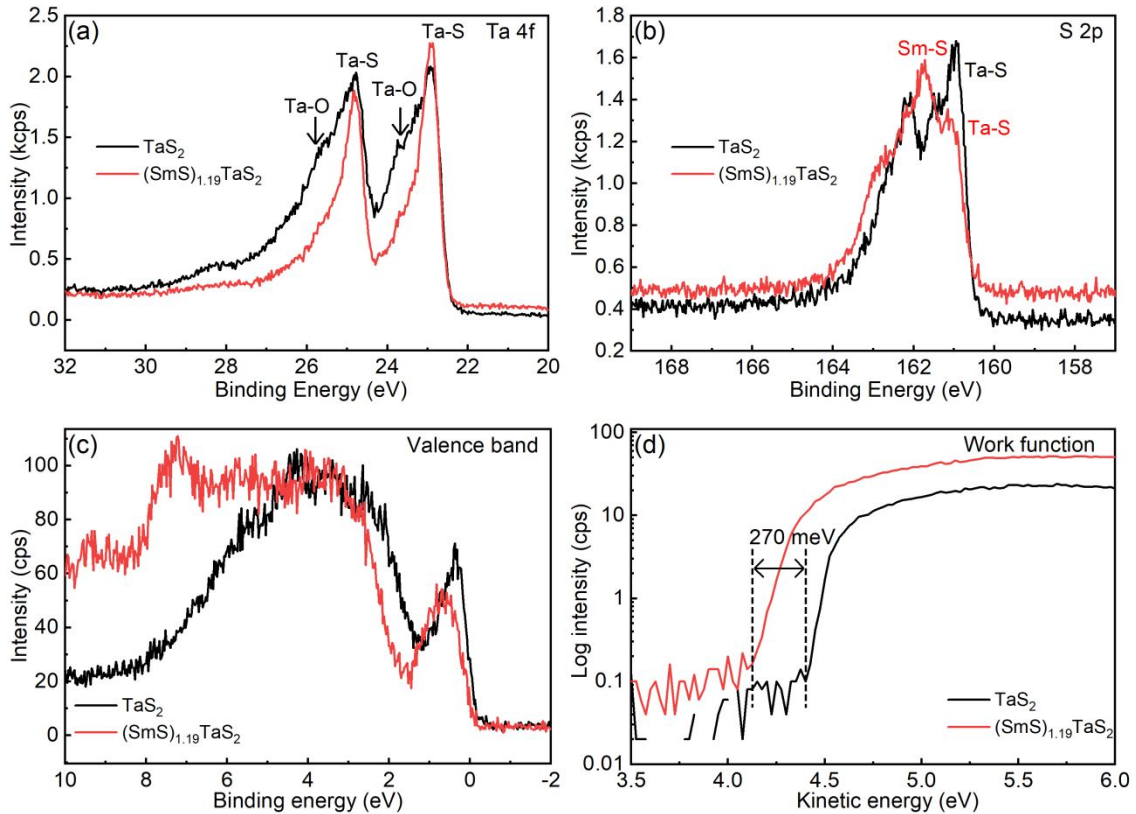


Fig. 5. X-ray photoelectron spectroscopic results of TaS_2 platelets (black) and the $(\text{SmS})_{1.19}\text{TaS}_2$ misfit nanotubes (red). (a) the Ta $4f$ doublet; (b) the S $2p$ doublet; (c) the valence band spectral region; (d) the onset of secondary electron emission, given on a log-scale, from which the sample work function is extracted for $(\text{SmS})_{1.19}\text{TaS}_2$ and TaS_2 , respectively.

Notably, XPS measurements can also provide rich information on the electrical properties of the probed samples.⁴⁹⁻⁵² In the present study, tested under extreme positive and negative charging conditions, both TaS₂ and SmS-TaS₂ are found to be very good conductors. Advantageously, these measurements, done *in-situ* in the XPS chamber, offer contactless electrical characterization. Yet, as an electrical probe, these measurements are better suited for semiconductors and insulators, yielding only limited sensitivity to the differences between the conductivity levels of metals. Notwithstanding this reservation, the conductivity of the misfit sample was found to be very similar to that of the metallic TaS₂ platelets. Hence, this observation suggests, in agreement with previous reports,⁷ that band filling by the Sm-to-Ta charge transfer is incomplete and the metallic conductivity is preserved in both constituents of the MLC.

Monochromated low-loss EEL spectra of TaS₂ and (SmS)_{1.19}TaS₂ misfit compound are presented in **Fig. 6**. Hyperspectral data were recorded with a scanning focused probe obtaining both spatial and a high energy resolution (better than 90 meV) to resolve low-energy excitations in the near-infrared region. **Fig. 6a** displays an exemplary spectrum of a (SmS)_{1.19}TaS₂ nanotube, obtained as the sum of multiple spectra in a region of interest across the central part of the nanotube. The elastic contribution to the spectrum was subtracted using the mirrored left-hand tail of the zero-loss-peak (ZLP).⁵³ The remaining tail at the lowest energies in the inelastic part of the spectrum contains Cerenkov losses and surface losses that are flattened out here because of a large collection angle. The inelastic part of the low loss spectrum is plotted for the central part of a (SmS)_{1.19}TaS₂ nanotube, the edge of a (SmS)_{1.2}TaS₂ platelet and a 2H-TaS₂ platelet in **Fig. 6b**. The reference spectrum of 2H-TaS₂ is very distinct from those of the misfit compound: A strong transition at around 1 eV observable all across the platelets of TaS₂ is almost absent in the misfit layered compound. To understand the origin of this peak and the reason for its absence in the

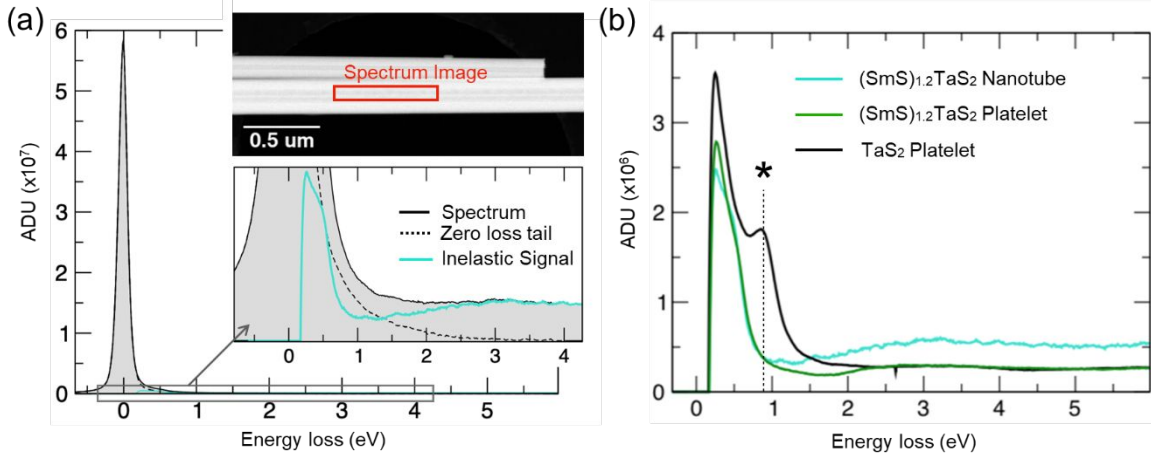


Fig. 6. Low-loss EEL spectra. (a) An exemplary spectrum of a (SmS)_{1.19}TaS₂ nanotube. The insets show an annular dark-field image of the nanotube and a magnified part of the spectrum with the extracted inelastic signal. The spectrum is a sum of spectra obtained in a region of interest (red rectangle) close to the tube axis, where the incident direction of the electron beam is close to the *c*-axis direction of the MLC. For background subtraction the tail of the elastic zero-loss peak was reflected from the high-energy side. (b) Inelastic part of the EEL low loss signal for three samples, the center part of a (SmS)_{1.19}TaS₂ nanotube (as in (a)), a (SmS)_{1.19}TaS₂ platelet and a 2H-TaS₂ platelets. The platelets were transmitted in the direction of the *c*-axis. The strong peak at about 1 eV energy loss (marked by an asterisk) in 2H-TaS₂ is associated with a transition from occupied *S* 3*p* states into unoccupied Ta 5*d*_{z²} states. The peak is absent in the MLC, in the platelet and the nanotube.

misfit compound one has to refer to the classical dielectric formalism of the loss function.^{54, 55} Accordingly, the EEL low-loss function is related to the so-called volume loss function, i.e. $\text{Im}(-1/\epsilon)$, where ϵ is the frequency-dependent dielectric function. The imaginary (ϵ_2) and real (ϵ_1) part of the frequency-dependent dielectric functions for bulk 2H-TaS₂, obtained from DFT calculations are presented in **Fig. S10**. A peak in ϵ_2 of 2H-TaS₂ just above 1 eV upon longitudinal excitation along the *c*-axis can be possibly ascribed to a transition from occupied S 3*p* states into unoccupied Ta 5*d*_{z²} states above the Fermi level.

Such an S 3*p* to Ta 5*d* transition is indeed likely to apply for the ~1 eV loss in the TaS₂ EEL spectrum, because it does not violate the dipole selection rules. Yet, the intensity of this peak is particularly high and, therefore, the related transition is suspected to be of a significant intra-atomic character, other than inter-atomic. This transition is enabled by intra-atomic transitions between, e.g. 5*d* Ta 3/2 and 5/2 states. Remarkably, the 1 eV EELS peak is practically absent from the (SmS)_{1.19}TaS₂ nanotubes and platelets. This result can be attributed to an almost complete filling of the Ta 5*d*_{z²} states in the misfit compound. The DFT calculations further substantiate an almost complete occupation of these Ta *d* states (*vide infra*). In agreement with the EEL data, the theoretical dielectric function of (SmS)_{1.2}TaS₂, obtained by DFT calculations does not show the excitation at about 1 eV (**Fig. S10**) owing to the charge transfer from the Sm 4*f*-band to the Ta 5*d*_{z²} states in the misfit compound.

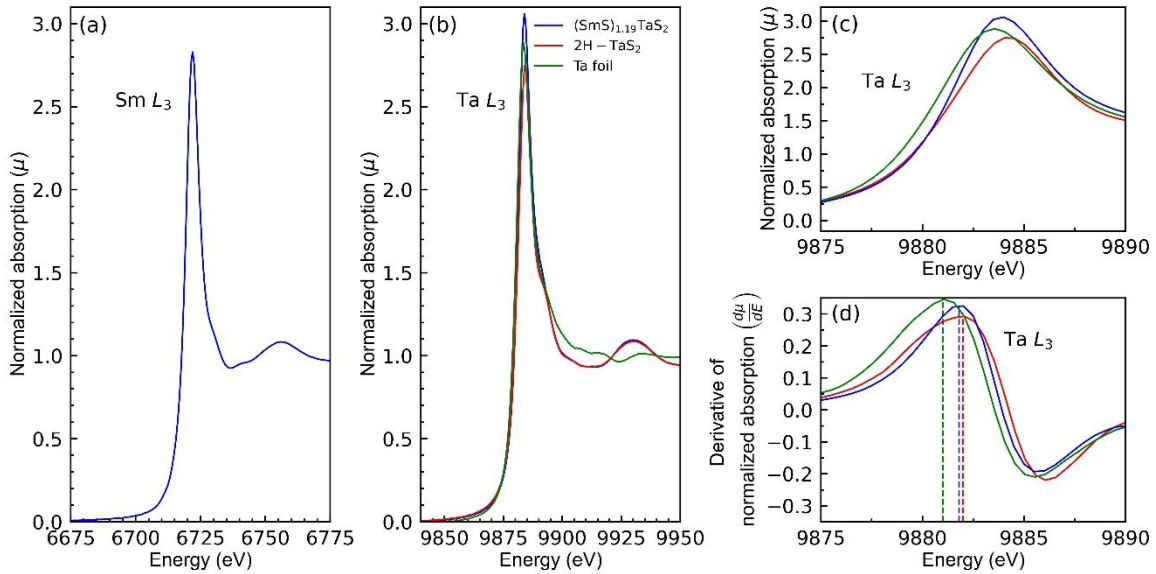


Fig. 7. XANES spectra of (SmS)_{1.19}TaS₂ and TaS₂ powders dispersed in the polymer matrix and pure (99.99%) Ta foil collected in a transmission geometry at PETRA III P23 beamline. Spectral regions for Sm-L₃ (a) and Ta-L₃ (b, c) edges. (d) Derivative of the normalized absorption in the vicinity of Ta L₃ edge. XANES in (a), (b) and (c) spectra are normalized to the edge step and the energy scale is calibrated with pure Ta and Mn foils.

Unlike the surface-sensitive XPS, XAS analysis provides the finest structural, chemical state and charge transfer insights of the bulk. **Fig. 7** shows Sm L₃ and Ta L₃ XANES spectra of (SmS)_{1.19}TaS₂ misfit compound (nanotube + flakes) in comparison with 2H-TaS₂ and Ta foil (99.99%) collected in transmission geometry. The shape of the Sm L₃ XANES spectrum (**Fig. 7a**) and the position of the ‘white line’ maxima at 6722 eV, (edge position 6719.5 eV, corresponding to 2*p* to 5*d* transition) is characteristic of Sm³⁺ state.⁵⁶ No signature of Sm²⁺ ‘white line’ in the range 6711-6713 eV was found, indicating that there is no mixed/intermediate valence state of Sm, which in turn signifies

a strong charge transfer.⁵⁶⁻⁵⁸ This observation is in-line with the XPS and also with the literature reports.^{7, 8}

The shape of the Ta L₃ XANES spectrum (**Fig. 7b**) resembles the 2H-TaS₂ spectra, and the position of the ‘white line’ at 9884 eV is close to that of the 2H-TaS₂ structures.^{59, 60} However, precise examination of the ‘white line’ (**Fig. 7c**) or the absorption edge position from the derivative of the normalized absorption (**Fig. 7d**), shows that there is a small shift between 2H-TaS₂ (9881.8 ± 0.1 eV) and the (SmS)_{1.19}TaS₂ (9882 ± 0.1 eV) Ta L₃ edges. It is known that the position of the Ta L₃ edge depends on the valence state of Ta: the higher the valence (oxidation state) the stronger Ta L₃ edge shifts towards higher X-ray energies.^{61, 62} **Fig. 7d** shows that the position of Ta L₃ edge of pure Ta foil has slightly lower value (oxidation state = 0, edge position 9881 ± 0.1) than that of TaS₂ and (SmS)_{1.19}TaS₂. The small difference between the L₃ edges of TaS₂ and (SmS)_{1.19}TaS₂ (0.2 ± 0.1 eV) can-not be over-interpreted, since some tantalum-oxide may have been occurred on the surface of the tubes (flakes). The (SmS)_{1.19}TaS₂ Ta L₃ ‘white’ line intensity (3.01 ± 0.01), which corresponds to the transition from the 2p core level to unoccupied Ta 5d states (d_{z²} band), is higher than the ‘white line’ intensity value measured for pristine 2H-TaS₂ (2.75 ± 0.01).^{59, 63} Interestingly, the intercalation of pristine 2H-TaS₂ by pyridine⁵⁹ or hydrazine⁶³ increases the intensity of the ‘white line’ up to the value of 2.5 and 2.9, respectively. In the case of (SmS)_{1.19}TaS₂, the charge transfer from SmS to TaS₂ would lead to a similar increase in the white line intensity of Ta L₃ in comparison with pristine 2H-TaS₂. In general, XAS investigations show that the white line intensity grows with electron-donating intercalates and diminishes with electron-withdrawing intercalates.^{59, 63} The fundamental reason for such a change is yet to be understood.

Density Functional Theory (DFT)

DFT calculations were employed to get an insight into the electronic structure of the SmS-TaS₂ MLC and to compare it to the parent SmS and TaS₂ phases. As a misfit model, the approximant (SmS)_{1.20}TaS₂ was chosen, in which a supercell included one SmS slab (12 SmS units) and one TaS₂ layer (10 TaS₂ units). A preliminary geometry optimization yielded the lattice parameters for misfit as $a = 17.08$ Å, $b = 5.79$ Å and $c/2 = 11.53$ Å. The in-plane parameters fitting for the TaS₂ sub-lattice were close to that of bulk 2H-TaS₂ (calc. $a = 3.35$ Å). On the other hand, the SmS sublattice showed a slight contraction compared to the bulk compound. Little peculiarity can be observed in the distortion of the SmS slab within SmS-TaS₂ misfit, when compared to the LaS slab within the LaS-TaS₂ misfits studied earlier.²⁶ The angles of the S-Sm-S configurations in this slab vary in the range 80-89° with the S atoms retracting into the slab, which is in agreement with the atom positions in the atomic-resolution STEM-BF images (see ball-and-stick model in **Fig. 8**).

While the calculations within Local Density Approximation or Generalized Gradient Approximation (LDA or GGA) ascribe a semi-metallic character for bulk SmS, the present LDA+U calculations describe this compound as a semiconductor with direct Γ - Γ transition type and a fundamental band gap of 0.69 eV (**Fig. 8a**, and **Fig. S11**). The latter is consistent with the scattering in the available experimental and theoretical data. The experimentally reported gap is 0.15 eV⁶⁴ or 0.4 eV⁶⁵ whereas the calculated values are 0.25 eV⁶⁵ or 0.71 eV.⁶⁶ The bandgap edges arise from the highly intense and strongly localized band of occupied Sm4f-states and from the shallow band of unoccupied Sm6s-states. The valence band of the occupied S3p-states is found at 2.5-5.5

eV below the top of the Sm4*f*-band. In general, such a DOS profile characterizes SmS as a lattice with a highly ionic character. The nominal oxidation states of the elements are Sm²⁺ and S²⁻ and the Sm4*f* states do not participate in the chemical bonding. Remarkably, the occupation and composition of the conduction band in the electronic structure of SmS is different from that of the isostructural LaS. In the latter, the Fermi level is hosted at a shallow band of La5*d*-states as shown in **Fig. S11**.

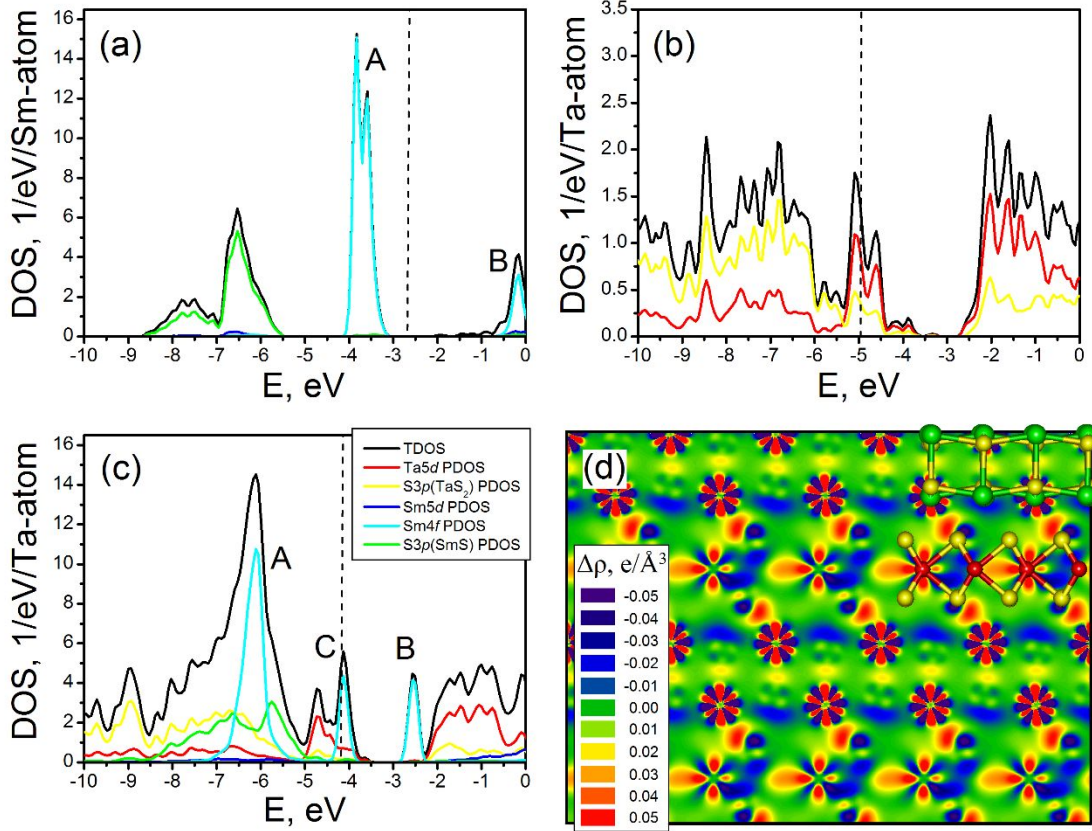


Fig. 8. Electronic densities-of-states (DOSs) for bulk fcc-SmS (a) 2H-TaS₂ (b) and misfit (SmS)_{1.20}TaS₂ (c). Fermi level is drawn as a dashed line. Panel (d) depicts the electronic density redistribution map after the (SmS)_{1.20}TaS₂ crystal assembly from SmS slabs and TaS₂ monolayers. Sm, S and Ta atoms of the ball-and-stick models are painted in green, yellow and red, respectively. Corresponding spin-resolved band structures are plotted on **Fig. S11**. DFT LDA+U calculations.

According to the present and also earlier calculations, bulk 2H-TaS₂ is a metal, where the Fermi level is hosted at the band of well-localized Ta5*d*_{z²}-states, see **Fig. 8b**. The wide valence band is composed of a mixture of dominating S3*p*- and secondary Ta5*d*-states responsible for the covalent Ta-S bonding. The wide conduction band is dominated by the Ta5*d*- and secondary S3*p*-states and is separated from the Ta5*d*_{z²}-band by a gap of ~0.5 eV. The latter reflects the well-documented tendency of TaS₂ to act as an acceptor within misfit compounds, (almost) filling the 5*d*_{z²}-band similar to 4*d*_{z²}- or 5*d*_{z²}-bands in the semiconducting MoS₂ or WS₂.

In analogy with the (La,Y)S-TaS₂ misfits studied in the past²⁵, (SmS)_{1.19}TaS₂ possesses a metal-like character in the framework of LDA+U calculations (**Fig. 8c**). Here, the metallic properties of SmS-TaS₂ arise due to the charge transfer from Sm4*f*- to Ta5*d*_{z²}-states within individual SmS and TaS₂ components acting as the donor and the acceptor of electron density, respectively. The absolute

value of the Fermi level in SmS-TaS₂ is found to be in-between the Fermi levels of the parent binary compounds (**Fig. S11**). The calculated effective charge on Sm atoms increases from +0.53 e in SmS to the average value of +0.81 e in (SmS)_{1.20}TaS₂, while the effective magnetic moments on all these atoms decrease from 6.97 μ_B to 6.49 μ_B on the average, respectively.

In contrast to (La,Y)S-TaS₂ misfits, the DOS profile of SmS-TaS₂ misfit can not be assembled from the DOS profiles for the individual SmS and TaS₂ in a simplified rigid-band model. The charge transfer in SmS-TaS₂ is accompanied by remarkable reorganization of the Sm4*f*-states compared to the pristine SmS compound (see bands A, B, C in **Fig. 8a, c**). The Fermi level is hosted at the shoulder of a Sm4*f*-band (C-band), which is split-off from the main occupied Sm4*f*-band (A-band). This new C-band is also well localized and appears in the pristine SmS compound, nearly in the middle between occupied A-band and unoccupied B-band. Noteworthy, the A-band in SmS-TaS₂ is aligned with both the occupied S3*p*-band of SmS and the occupied S3*p*-band of TaS₂, which may point to additional strengthening of Sm-S bonding both within the SmS layer and at the SmS | TaS₂ interface. Indeed, mapping the electron density redistribution within (SmS)_{1.20}TaS₂ unveils not only an enhancement of electron density at the Ta atoms (four-lobed patterns of Ta5*d*-orbitals in **Fig. 8d**), but also a charge redistribution within the Sm atoms (six-lobed patterns of Sm4*f*-orbitals). Furthermore, an enhancement of electron density between the Sm atoms and the S atoms of TaS₂ is visible, which is responsible for the rise of a new coordinate Sm-S bonding (red “blobs” between Sm and S). A slight enhancement of the electron density between Sm and S atoms within the SmS unit can be also observed.

To complement the data of low-loss EEL spectroscopy, the frequency-dependent dielectric functions $\epsilon(\omega) = \epsilon_1(\omega) + i\epsilon_2(\omega)$, where $\epsilon_1(\omega)$ and $\epsilon_2(\omega)$ are the real and imaginary parts of the function, respectively, have been calculated using the same pseudopotential DFT method for in-plane (xx) and out-of-plane (zz) scattering on both 2H-TaS₂ and (SmS)_{1.20}TaS₂ compounds (**Fig. S10**). The results for TaS₂ are in semi-qualitative agreement with the optical properties elucidated from the EELS analysis (**Fig. 6**) and more sophisticated full-potential plane-wave calculations.⁶⁷ The origin of the signal at ~1 eV in the inelastic part of the low-loss EEL spectrum along *c*-axis of TaS₂ and the corresponding maximum of $\epsilon_2^{zz}(\omega)$ function at ~1.5 eV in the calculations, are related to the electron transfer into the half-filled Ta5*d*_{*z*²}-band, see **Fig. S10a**. Contributions to these transitions arise from the S 3*p* and Ta 5*d* states (**Fig. 8b**), noting that both of them are dipole-allowed ($\Delta j = \pm 1$) transitions. The real part of the dielectric function, ϵ_1 , influences as well the 1 eV region. Yet, in common to all related transitions, the role of the empty Ta 5*d*_{*z*²} states is dominant. In contrast, for the misfit structure, the ~1 eV region is modified significantly, due to the charge transfer discussed above. Accordingly, the dielectric function calculated for (SmS)_{1.20}TaS₂ confirms the disappearance of the EEL signal from the ~1 eV regime, when applied along the *c*-axis of the MLC (**Fig. S10b**).

Magnetic measurements

The ac susceptibility measured at frequencies 1, 100, and 1000 Hz and zero dc magnetic field show a strong diamagnetic signal below 3.9 K (**Fig. 9a**), indicating that (SmS)_{1.19}TaS₂ undergoes superconducting transition below 5 K. This transition temperature is appreciably higher than that occurring in 2H-TaS₂ (0.63 K).⁴⁵ The presence of diamagnetic signals is also verified by zero-field cooled dc magnetization measured at a magnetic field of 20 Oe (**Fig. 9a**, inset) and the transition temperature obtained by both ac and dc measurements is in excellent agreement. **The low-temperature ac susceptibility shows a large volume fraction ~45%, of the superconducting phase, indicating the bulk superconducting in nature.** The isothermal magnetization, *M*(*H*), measured at 2 K shows butterfly-shaped hysteresis loops (**Fig. 9b**) possibly associated with Type-II

superconductivity.⁶⁸ The observed behavior was verified three times for $(\text{SmS})_{1.19}\text{TaS}_2$ sample prepared in different batches. No superconductivity transition was observed in the magnetic measurements of GdS-TaS_2 tubes and flakes, which were prepared in a similar way (see Fig. S12). The iso-field, iso-thermal magnetic and ac susceptibility measurements show that GdS-TaS_2 ordered antiferromagnetically at temperature 7 K (Fig. S12).

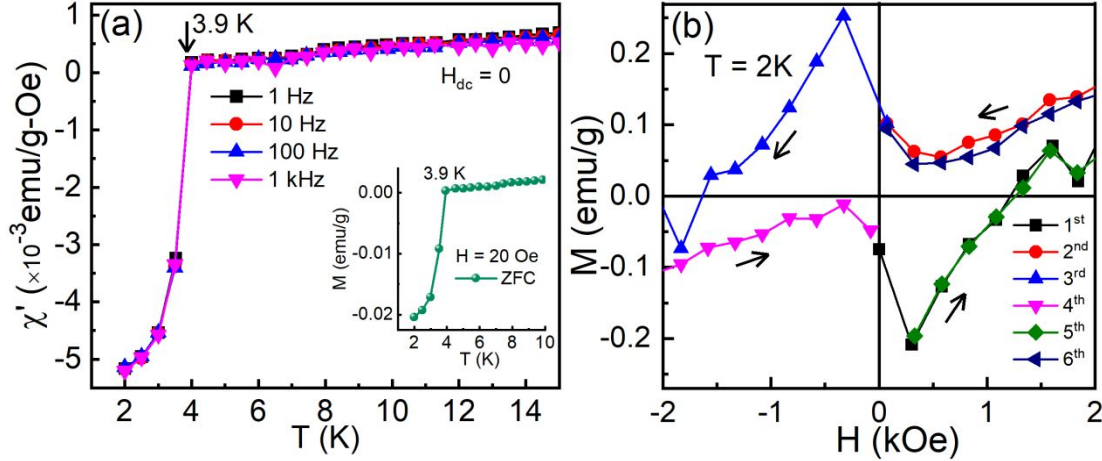


Fig. 9. Temperature-dependent magnetic susceptibility of $(\text{SmS})_{1.19}\text{TaS}_2$ MLC, (a) Real part of the ac magnetic susceptibility measured at frequencies, $f = 1, 10, 100$ and 1000 Hz and $H_{dc} = 0$. Inset in (a) shows zero-field cooled dc magnetization as a function of temperature measured at $H_{dc} = 20$ Oe. (b) Zero field cooled isothermal magnetization measured at $T = 2$ K. The hysteresis loops were carried out as $0 \rightarrow +10$ (first cycle) $\rightarrow -10$ (second and third cycles) $\rightarrow +10$ (fourth and fifth cycles) $\rightarrow 0$ kOe (sixth cycle). Only the part of hysteresis loops was shown for clarity.

Charge density waves (CDW) and superconductivity (SC) co-exist in 2H-TaS_2 .⁶⁹ It is well understood that the intercalation of alkali metals such as Li/Na and pyridine would enhance the superconducting transition temperatures and suppress the CDW transition.⁶⁹ The intercalation of small quantities of Li or pyridine in 2H-TaS_2 was shown to enhance the SC critical temperature- T_c from 0.8 K to 3.5 K.⁶⁹ A similar effect can be anticipated in misfit compounds, whereby the enhancement in T_c of $(\text{SmS})_{1.19}\text{TaS}_2$ compared to pure TaS_2 is reminiscent of the strong charge transfer from SmS to TaS_2 . The superconductivity in a series of MLCs was studied before using heat capacity and magnetic susceptibility analyses.⁷⁰ The authors did not find any superconductivity in LaS-NbS_2 and SmS-NbS_2 , which was attributed to the strong charge transfer between the layers and the stronger polar coupling between the layers, compared to other MLCs. Note that no evidence for the presence of pure tantalum impurities ($T_c = 4.4$ K) was obtained by any of the techniques used in this study. Magnetic susceptibility of SmS-TaS_2 under relatively high magnetic fields (0.875 T), and did not exhibit a superconductivity transition.⁷ The χ^{-1} vs temperature curve in the interval $100 \text{ K} < T < 300 \text{ K}$ shows good agreement (R-factor = 0.99987) with the Curie-Weiss equation (Fig. S13), the derived Curie constant is $C = 0.0627(6) \text{ cm}^3 \text{ K mole}^{-1}$. The effective magnetic moment is equal to $\mu_{\text{eff}} = 0.71 \mu_B$ while the calculated value of μ_{eff} for Sm^{3+} is $0.84 \mu_B$.⁷¹ Thus samarium in $(\text{SmS})_{1.2}\text{TaS}_2$ MLC exists in Sm^{3+} state with $4f^5$ ($^6\text{H}_{5/2}$) electronic configuration.

Transport measurements

Transport properties of individual LaS-TaS₂ nanotubes were recently reported.⁴⁴ In support of the DFT calculations, the LaS-TaS₂ nanotubes were found to be semimetallic. Given the fact that the Sm-based MLC is chemically more stable than the LaS-TaS₂ and the improved handling of the

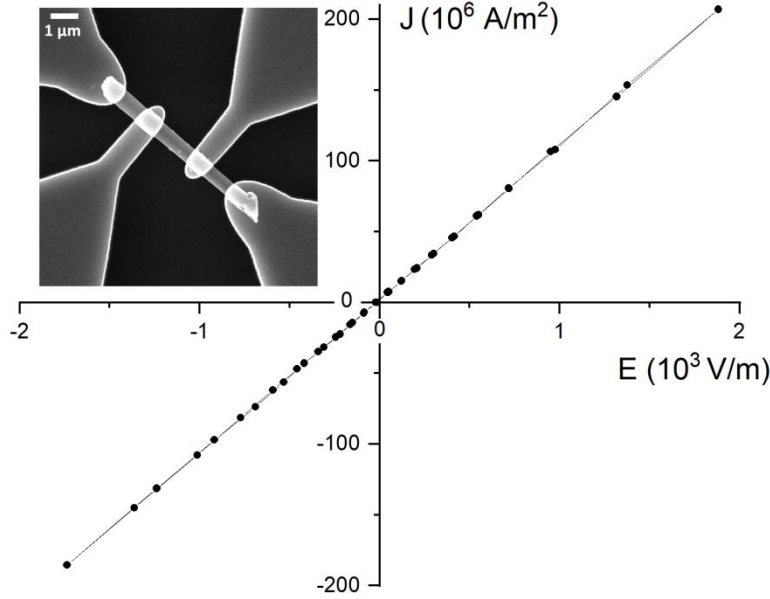


Fig. 10. A typical I-V curve of a single (SmS)_{1.19}TaS₂ nanotube (550 nm in diameter and 7 μm long), giving resistivity $\rho = 0.92 \times 10^{-3} \Omega \square \text{cm}$. Several devices of this kind were prepared, exhibiting resistivities between $(0.40 - 0.95) \times 10^{-3} \Omega \square \text{cm}$. The inset is an SEM image of nanotube (in the middle) with four contact probes fabricated using EBL, showing the geometry of devices used for (room temperature) electrical transport measurements.

nanotubes, surface oxidation of the nanotubes were generally less of an issue here. Several devices of this kind were prepared and measured (Fig. 10). The specific resistivities derived from the measurements are in the range of $(0.40 - 0.95) \times 10^{-3} \Omega \square \text{cm}$. These resistivity values are similar to the room temperature resistivity of bulk 2H-TaS₂, which is on the order of $5 \times 10^{-3} \Omega \square \text{cm}$.⁷² In fact these values are comparable to the values reported in Ref. 7. The somewhat counterintuitive low resistivity of SmS-TaS₂ nanotubes, which is comparable to that of 2H-TaS₂ flakes can be possibly attributed to their quasi-1D structure, which leads to a reduced scattering of the charges. Preliminary cathodoluminescence (CL) measurements of individual nanotubes in cryogenic temperatures (-150 K) were carried-out within the SEM. The CL spectra revealed f - f luminescence with a peak at 695 nm (see Fig. S14), which were absent in the background. The 695 nm luminescence peak has been previously assigned to Sm²⁺ and is attributed to the local reduction of Sm³⁺ by the electron beam.⁷³

Conclusions

Herein, the structure and some physical properties of nanotubes and flakes formed from the misfit layered compound (SmS)_{1.2}TaS₂ were studied. High-resolution transmission electron microscopy (HR-STEM) was used to shed light on the structure of the nanotubes in detail. Their cross-sectional lamella were prepared and studied via HR-STEM providing unprecedented resolution of the lattice-atoms in the nanotube. In particular, the trigonal prismatic arrangement of the 2H-TaS₂ was clearly visible. X-ray photoelectron spectroscopy and X-ray absorption and electron energy loss spectroscopy analyses coupled with density functional theory calculations

indicated that strong charge transfer from the samarium 4f level to the 5d_{z²} levels in TaS₂ leads to the partial filling of these energy levels. This charge transfer has several implications. First, the Sm atom is present as Sm³⁺ valence state in the MLC lattice. Also, the SmS-TaS₂ exhibits a comparable conductivity to bulk 2H-TaS₂, which is vindicated through four probe transport measurements. Finally, the charge transfer suppresses the charge density wave phase of TaS₂, promoting thereby the superconductivity of this MLC with T_c of 3.8-4.4 K compared to 0.8 K for bulk 2H-TaS₂. This study sheds new light on the structure-properties relations in MLC and their nanotubes in particular. In particular, the present study demonstrates that Ln-based MLC nanotubes are likely to exhibit intriguing 1D quantum physical phenomena at cryogenic temperatures.

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Supporting Information

Experimental and characterization details, computation details, XRD, HR-STEM, STEM-EDS, magnetic measurements, electrical characterization, frequency-dependent dielectric function, Band structure of LaS-TaS₂ and SmS-TaS₂, magnetic susceptibility fitting of $\chi(T)$ Curie-Weiss equation and cathodoluminescence.

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TOC Graphic

