



Scheme 1

Experimental

The compounds were characterised by elemental analysis and ir spectral data (Table 1). Mps. were taken in open capillary tubes and are uncorrected.

TABLE 1—PHYSICAL DATA OF THE HYDRAZONES*

Sl. no.	R	R ₁	M.p. °C	Yield %	Colour
1.	CH ₃	CH ₃	235	30	Yellow
2.	CH ₃	CH ₂ COCH ₃	204	80	Yellow
3.	CH ₃	CH ₂ C ₆ H ₅	196	50	Pale yellow
4.	H	CH=CHC ₆ H ₅	174	75	Yellow
5.	C ₆ H ₅	C ₆ H ₅	186	40	Yellow
6.	C ₆ H ₅	C ₆ H ₅	159	50	Yellow
7.	H	<i>o</i> -NO ₂ C ₆ H ₅	205	70	Yellow
8.	H	<i>m</i> -NO ₂ C ₆ H ₅	185-88	65	Orange
9.	H	<i>p</i> -NO ₂ C ₆ H ₅	243	80	Orange
10.	H	<i>o</i> -ClC ₆ H ₅	204	45	Pale yellow
11.	H	<i>o</i> -OHC ₆ H ₅	208-09md	50	Pale yellow
12.	H	<i>p</i> -OHC ₆ H ₅	251	80	Pale yellow
13.	CH ₃	2-OHC ₆ H ₅	246	60	Orange
14.	C ₆ H ₅	COC ₆ H ₅	226	80	Yellow
15.	C ₆ H ₅	CH ₂ COCH ₃	186	50	Light orange
16.	R = R ₁ = Isatin		296md	50	Orange
17.	R = R ₁ = Acenaphthene-quinone		300	35	Deep brown
18.	R = R ₁ = 5-Chloroisatin		300	60	Orange
19.	R = R ₁ = 5-Bromoisatin		300	50	Orange
20.	R = R ₁ = 5,7-Dinitro		300	70	Orange

* All compounds gave satisfactory C, H and N analyses.

Synthesis and Biological Activities of Quinazoline-4-thioglycolic Acid Hydrazones

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HYDRAZONES exhibit various biological activities¹. Present paper reports the synthesis of hydrazones from quinazoline 4-thioglycolic acid hydrazide and aldehydes or ketones (Scheme 1). All the compounds have been screened for their antimicrobial activities.

Quinazoline 4-ethylthioglycolate: A solution of ethyl thioglycolate (3.2 g, 0.026 mol) in dry benzene (70 ml) was added dropwise during 5 min a benzene solution of 4-chloroquinazoline² (50 g, 0.302 mol) and the mixture was refluxed for 4 h. The resulting orange coloured solid was washed with benzene and then with 10% sodium hydrogen carbonate solution and recrystallised from methanol as yellow rectangular crystals (60%), m.p. 80–82°.

Quinazoline 4-thioglycolic acid hydrazide: Quinazolin 4-thioglycolate in 95% alcohol (20 ml) was added an alcoholic solution (1 ml) of hydrazine hydrate with stirring and the mixture was refluxed for 4 h. Pale yellow precipitate that separated after cooling was recrystallised from 95% alcohol, (70%), m.p. 177–78°.

The hydrazones: A mixture of quinazoline 4-thioglycolic acid hydrazide (0.01 mol) in ethanol (25 ml) and ethanolic solution of the

appropriate aldehydes and ketones (0.01 mol) was refluxed for about 5 h. The solid that separated on cooling was recrystallised from ethanol.

Results and Discussion

The hydrazones of quinazoline 4-thioglycolic acid are crystalline in nature and stable under dry condition.

The strong intensity ir band present in the compounds at $3\ 400\text{--}3\ 100\text{ cm}^{-1}$ can be attributed to the NH stretching frequency. A moderately strong band at $1\ 700\text{--}1\ 600\text{ cm}^{-1}$ in the spectra of all the hydrazones is attributed to the stretching frequency of C=O group. Almost all the compounds exhibit C=N stretching frequency at $1\ 600\text{--}1\ 580\text{ cm}^{-1}$.

All the compounds were screened against *S. aureus*, *S. albus*, *Proteus*, *Pyocyanea*, *E. coli* and *Klebsiella* bacteria using agar plate disk diffusion technique³. Compounds of sl. nos. 4, 5 and 16 were found to possess highest inhibitory action (8–12 mm) against the growth of pathogenic bacteria, and compound of sl. no. 5 showing highest action (>12 mm) against *Pyocyanea*. Decreasing sensitivity was found in case of hydrazones from propiophenone (sl. no. 6), 2-hydroxyacetophenone (sl. no. 13) as well as aliphatic ketone (sl. no. 1), which are moderately active against *Pyocyanea*.

Compounds of sl. no. 10 and 17 from *o*-chlorobenzaldehyde and acenaphthenequinone exhibit moderate activities (6–10 mm) against *Proteus*, *Pyocyanes* and *E. coli*.

Compounds no. of sl. no. 4, 9 and 18 were found moderately active (6–8 mm) against *Klebsiella*.

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