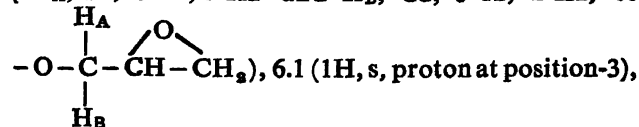


3.35 (1H, m,  $-\text{OCH}_2-\text{CH}-\text{CH}_2$ ), 3.95 and 4.3 (1H<sub>A</sub>, dd,  $J$  12, 5 Hz and H<sub>B</sub>, dd,  $J$  12, 2 Hz, of



6.1 (1H, s, proton at position-3), 6.8 (1H, s, at position-5) and 7.3–7.55 (2H, m, proton at position-7 and -8).

**4-Methyl-7-(2-hydroxy-3-isopropylaminopropoxy) coumarin (3):** A mixture of 4-methyl-7-(2,3-epoxypropoxy)coumarin (1.0 g) and isopropylamine (1 ml) in methanol was refluxed for 6 h on a steam-bath. The solvent was then evaporated to afford 3 which was crystallised from benzene–petroleum ether, (0.8 g), m.p. 140°;  $\delta$  1.1 (6H, d,  $-\text{CH}(\text{CH}_3)_2$ ), 2.4 (3H, s,  $\text{CH}_3$ , at position-4), 2.55–2.95 (5H, m,  $\text{N}-\text{CH}_2\text{CH}_2\text{NHOH}$ ), 4.0 (2H, d of  $\text{ArOCH}_2$ ), 6.21 (1H, s, proton at position-3) and 7.0–7.8 (3H, m, ArH).

Similarly, other compounds were prepared (Table 1).

#### Acknowledgement

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#### References

1. B. N. C. PRICHARD and P. M. S. GILLAM, *Br. Med. J.*, 1964, 2, 727, *Am. J. Cardiol.*, 1966, 18, 387.

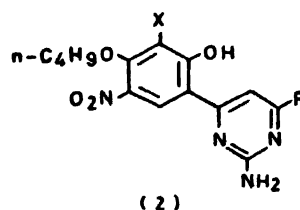
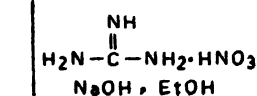
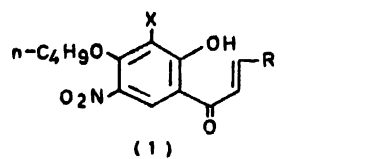
### Preparation and Antimicrobial Activity of 2-Amino-4-(2'-hydroxy-3'-bromo/H-4'-n-butoxy-5'-nitrophen-1'-yl)-6-arylpurimidines

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**2-AMINOPYRIMIDINES**<sup>1</sup> are known for their physiological importance. With a view to get better therapeutic activity, some new 2-aminopyrimidines (2) were synthesised and evaluated for their antimicrobial activity. The compounds have been prepared<sup>2</sup> by the reaction of 2'-hydroxy-3'-bromo-4'-n-butoxy-5'-nitrochalcones and 2'-hydroxy-4'-n-butoxy-5'-nitrochalcones (1) with guanidine nitrate in the presence of aqueous sodium hydroxide (40%) in ethanol (Scheme 1). Previously reported 2'-hydroxy-3'-bromo-4'-n-butoxy-5'-nitrochalcones<sup>3</sup> and 2'-hydroxy-4'-n-butoxy-5'-nitrochalcones<sup>4</sup> have been prepared from 2-hydroxy-3-bromo-4-n-butoxy-5-nitroacetophenone or 2-hydroxy-4-n-butoxy-5-



X=Br/H  
R=aryl  
Scheme 1

nitroacetophenone and various araldehydes and evaluated their biological activity<sup>3,4</sup>. The structure of the compounds has been supported by elemental analysis and ir spectral data. The products have been screened for their antimicrobial activity.

**Antimicrobial activity:** The compounds were screened for antibacterial and antifungal activities using cup-plate agar diffusion method<sup>5</sup>. The testing was carried out at a concentration of 50  $\mu\text{g}$  using gram-positive bacteria *Staphylococcus aureus*, gram-negative bacteria *Escherichia coli* and fungi *Aspergillus niger*. The compounds showed only mild antibacterial activity (inhibition 20–30%). The compounds 2 having bromo and chloro substituents in benzene ring were found to be more active. Maximum activity was observed in compounds (2) when R=4-tolyl, 2- and 4-chlorophenyl, 2,4-dichlorophenyl against *S. aureus* and 2- and 4-bromophenyl, 4-methoxyphenyl against *E. coli*. The activity was not comparable with the known antibiotic like chloromycetin under similar condition. The most antifungal active compound of the series was Sl. no. 1 (Table 1); it inhibited the growth of *A. niger* to the extent of 80%. Most of the compounds showed moderate antifungal activity (inhibition 60–75%) and a few showed weak activity (inhibition < 55%) against *A. niger*. In general, the presence of chlorine atom in R decreased fungitoxicity.

#### Experimental

All the melting points were determined in open capillaries and are uncorrected. The ir spectra

TABLE 1—ANALYTICAL AND PHYSICAL DATA OF 2-AMINO-4,6-DIARYLPYRIMIDINES (2)<sup>a</sup>

Sl. no.	R	X	Mol. formula	M.p. °C	Yield %
1.	Phenyl	Br	C <sub>20</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Br	205	70
2.	2'-Chlorophenyl	Br	C <sub>20</sub> H <sub>18</sub> O <sub>4</sub> N <sub>4</sub> BrCl	190	75
3.	4'-Chlorophenyl	Br	C <sub>20</sub> H <sub>18</sub> O <sub>4</sub> N <sub>4</sub> BrCl	240	78
4.	2',4'-Dichlorophenyl	Br	C <sub>20</sub> H <sub>17</sub> O <sub>4</sub> N <sub>4</sub> BrCl <sub>2</sub>	260	85
5.	4'-Methylphenyl	Br	C <sub>21</sub> H <sub>21</sub> O <sub>4</sub> N <sub>4</sub> Br	185	65
6.	3',4'-Methylenedioxyphenyl	Br	C <sub>21</sub> H <sub>19</sub> O <sub>6</sub> N <sub>4</sub> Br	162	70
7.	3',4',5'-Trimethoxyphenyl	Br	C <sub>22</sub> H <sub>25</sub> O <sub>7</sub> N <sub>4</sub> Br	155	65
8.	3',4'-Dimethoxyphenyl	Br	C <sub>22</sub> H <sub>23</sub> O <sub>6</sub> N <sub>4</sub> Br	180	70
9.	4'-Methoxyphenyl	Br	C <sub>21</sub> H <sub>21</sub> O <sub>5</sub> N <sub>4</sub> Br	135	65
10.	2'-Methoxyphenyl	Br	C <sub>21</sub> H <sub>21</sub> O <sub>5</sub> N <sub>4</sub> Br	201	60
11.	4'-N-Dimethylamino-phenyl	Br	C <sub>23</sub> H <sub>27</sub> O <sub>4</sub> N <sub>5</sub> Br	210	62
12.	Phenyl	H	C <sub>20</sub> H <sub>20</sub> O <sub>4</sub> N <sub>4</sub>	190	70
13.	2'-Bromophenyl	H	C <sub>20</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Br	235	75
14.	4'-Bromophenyl	H	C <sub>20</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Br	250	80
15.	2'-Chlorophenyl	H	C <sub>20</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Cl	202	85
16.	4'-Chlorophenyl	H	C <sub>20</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Cl	231	80
17.	2',4'-Dichlorophenyl	H	C <sub>20</sub> H <sub>18</sub> O <sub>4</sub> N <sub>4</sub> Cl <sub>2</sub>	270	90
18.	2'-Methoxyphenyl	H	C <sub>21</sub> H <sub>23</sub> O <sub>5</sub> N <sub>4</sub>	185	70
19.	4'-Methoxyphenyl	H	C <sub>21</sub> H <sub>23</sub> O <sub>5</sub> N <sub>4</sub>	208	65
20.	3',4'-Dimethoxyphenyl	H	C <sub>22</sub> H <sub>24</sub> O <sub>6</sub> N <sub>4</sub>	160	68
21.	3',4',5'-Trimethoxyphenyl	H	C <sub>22</sub> H <sub>26</sub> O <sub>7</sub> N <sub>4</sub>	175	70
22.	4'-Methylphenyl	H	C <sub>21</sub> H <sub>23</sub> O <sub>4</sub> N <sub>4</sub>	198	75
23.	3',4'-Methylenedioxyphenyl	H	C <sub>21</sub> H <sub>21</sub> O <sub>6</sub> N <sub>4</sub>	217	70
24.	4'-N-Dimethylamino-phenyl	H	C <sub>23</sub> H <sub>27</sub> O <sub>4</sub> N <sub>5</sub>	165	60

<sup>a</sup>All compounds gave satisfactory N analysis.

(KBr) were taken on a Perkin-Elmer spectrophotometer.

2-Amino-4-(2'-hydroxy-3'-bromo-4'-n-butoxy-5'-nitrophen-1'-yl)-6-phenylpyrimidine (2): A mixture of 2'-hydroxy-3'-bromo-4'-n-butoxy-5'-nitrochalcone (1; X=Br; 1.00 g) and guanidine nitrate (0.30 g) in ethanol (50 ml) was refluxed and an aqueous solution of sodium hydroxide (40%, 5 ml) added to it portionwise during 3 h. The reflux was continued further for 6 h and the resulting solid on cooling was filtered and crystallised from aqueous DMF, (70%), m.p. 205° (Found: C, 52.17; H, 4.21; N, 12.12. C<sub>20</sub>H<sub>19</sub>O<sub>4</sub>N<sub>4</sub>Br calcd. for: C, 52.29; H, 4.14; N, 12.20%);  $\nu_{\max}$  (KBr) 3 460, 3 330 (NH), 1 630 (C=N) and 3 200 cm<sup>-1</sup> (OH phenolic).

Similarly, other pyrimidines (Table 1) were prepared.

2-Amino-4-(2'-hydroxy-3'-H-4'-n-butoxy-5'-nitrophen-1'-yl)-6-phenylpyrimidine (2): A mixture of 2'-hydroxy-4-n-butoxy-5'-nitrochalcone (1; X=H, 1.00 g) and guanidine nitrate (0.30 g) in ethanol (50 ml) was refluxed and an aqueous solution of sodium hydroxide (40%; 5 ml) added to it portionwise during 3 h. The reflux was continued further for 8 h and the resulting yellow solid on cooling was filtered and crystallised from aqueous DMF, (75%), m.p. 190° (Found: C, 63.20; H, 5.12; N, 14.82. C<sub>20</sub>H<sub>20</sub>O<sub>4</sub>N<sub>4</sub> calcd. for: C, 63.16; H, 5.26;

N, 14.74%);  $\nu_{\max}$  (KBr) 3 455, 3 335 (NH), 1 620 (C=N) and 3 200 cm<sup>-1</sup> (OH phenolic).

Similarly, other pyrimidines (Table 1) were prepared.

#### Acknowledgement

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#### References

1. A. KRENTZBERGER and S. L. ROEHLING, *Chem. Abstr.*, 1975, 83, 78767; G. B. BENNET, R. B. MASON, L. J. ALDEN and J. B. ROACH, *J. Med. Chem.*, 1978, 21, 623; K. K. WEINHERDT and M. MARN, *Chem. Abstr.*, 1981, 95, 97837; V. KARABANOV, L. P. PRIKARCHIKOVA, V. BOLDYREV, I. A. NASYR, I. G. VALADIMIRSEV, V. M. CHERKASOV, N. I. ZHURAVAKAYA and V. P. BORISENKO, *Chem. Abstr.*, 1981, 94, 15141.
2. N. R. EL-RAYYES, *J. Heterocycl. Chem.*, 1982, 19, 415.
3. M. D. ANKHIWALA and H. B. NAIK, *J. Indian Chem. Soc.*, communicated.
4. M. D. ANKHIWALA and H. B. NAIK, *J. Inst. Chem., Calcutta*, communicated.
5. F. KAVANAGH, "Analytical Microbiology", Academic, New York, 1968, p. 125.

### Pyrimidine Antagonists. Part-III. Synthesis of some 2,4-Diamino-5-substituted-phenylazo-6-substituted-aminopyrimidines and Evaluation of their Anticancer Property against Leukemia P388

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PURINE and pyrimidine possess widerange biological activities<sup>1</sup>.

The presence of amino groups in C-2 and C-4 positions of the pyrimidine ring and also the presence of hydroxyl and/or amino groups in C-2 and C-4 positions having a bulky substituted-amino group in C-6 position<sup>2</sup> have been found to exhibit activity against *Streptococcus faecalis* ATCC 8043. The present investigation aims to keep the 2,4-diaminopyrimidine moiety intact and then substitute new group at the phenylmoiety at the C-6 position having phenylazo group in C-5 position.

In previous communications<sup>3</sup> the present author reported the synthesis of some 2,4-diamino-6-substituted-aminopyrimidines and 2,4-diamino-6-substituted-amino-5-arylazopyrimidines.

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