Fatty Acid Composition of Spilanthes paniculata

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Manuscript received 24 August 1987, accepted 22 December 1987

As a part of our programme to explore the herbal drugs from Indian medicinal plants, we chemically investigated an endemic herb of Tripura, *Spilanthes paniculata* Wall. ex DC¹. In earlier communication we reported² the presence of amino acids in the plant. Now we report the fatty acid composition in the aerial parts of the plant.

Aerial parts of the plant, S. paniculata (collected from the area of greater Agartala in September 1985) were air-dried, milled and it (1 kg) was extracted with petrol (b.p. $60 - 80^{\circ}$) in a Soxhlet for 48 h. The petrol extract was concentrated to a gummy residue (6.5 g). The crude residue (6.0 g) was chromatographed through a silica gel column (60-120 mesh). The latter fraction of petrol eluate and earlier fraction of petrol – benzene (9:1) eluate on evaporation of solvent afforded an oily residue (1.5 g). The chemical characteristics of this oil were determined by the usual methods : iodine value (Wij's/30 min), 24.5; saponification equivalent (0.5 N alcoholic KOH), 191.5. The oil (1.0 g) was saponified with methanolic 1 N KOH (25 ml) for 2 h in an atmosphere of N_{g} . The solvent MeOH was removed under reduced pressure and the residue diluted with $H_{g}O$. The unsaponifiable matter was recovered by extracting the aqueous layer with ether. The ether layer was washed with H.O. dried with anhydrous Na SO, and evaporated to get a residue (0.3 g). The mixture of fatty acids was obtained by acidification of the aqueous layer with dilute HCl to pH 2-3 followed by extraction with solvent ether. The ether layer was washed, dried in the usual manner and evaporated to get an oil of fatty acids (0.6 g).

Me-esters of fatty acids were prepared with $CH_{g}N_{g}$ in the usual manner and were purified through silica gel column chromatography. The purified esters mixture was analysed by glc technique on a Pye Unicam 104 gas chromatograph equipped with a FID and a 10% PEGA column. Operating conditions were inj. port temperature 200°, column temperature 190°, FID temperature 210°, and carrier gas N_g at flow rate 50 ml min⁻¹. The Me-esters were identified³ by co-glc with authentic samples. The ECL values and relative retention time (with respect to stearic acid) as well as relative abundance (in terms of peak areas) of the Me-esters are given in Table 1.

The presence of unsaturated Me-esters was also confirmed by the catalytic hydrogenation of a part of ester mixture with 10% Pd – C in EtOAc followed by glc analysis under identical condition when the peaks of 10: 1, 12: 1, 12: 2, 14: 1, 14: 2, 16: 1,18: 1 and 18: 2 acids were not observed in the

TABLE 1—EQUIVALENT CHAIN LENGTHS (ECL), RELATIVE RETENTION TIME (RRT) AND RELATIVE ABUNDANCE (RA) OF MO-ESTERS			
Me-esters of fatty acids	EOL	RRT (18 : 0)	RA %
Capric (10 : 0)	10.00	0.07	2.4
Decamonoenoic (10:1)	10.75	0.09	0.8
Lauric (12:0)	12.00	0.14	5.0
Dodecamonoenoic (12:1)	12.65	0.17	0.6
Dodecadienoic (12:2)	13.55	0.23	0.4
Myristic (14 . 0)	14.00	0.27	8.0
Tetradecamonenoic (14:1)	14.55	0,32	2.2
Tetradecadienoic (14:2)	15.50	0.44	2.2
Pentadecanoic (15:0)	15.00	0.38	8.8
Palmitic (16:0)	16 00	0.52	55.5
Palmitoleic (16:1)	16.50	0.61	0.4
Margaric (17:0)	17.00	0.72	0.6
Steario (18 : 0)	18.00	1.00	18.5
Oleic (18:1, n-9)	18.35	0,12	8.8
Linoleic (18:2, n-6)	18.95	1.96	0.9
Unidentified			04

chromatogram and relative intensities of the corresponding saturated acids were increased.

Acknowledgement

The authors thank R.S.I.C., Bose Institute. Calcutta, for glc analysis and U.G.C., New Delhi, for financial assistance.

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Amino Acids of Ajuga macrosperma

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Manuscript received 30 November 1987, accepted \$2 December 1987

THE genus Ajuga of the family Labiatae comprises of 40 species¹ which are found in temperate regions of the world. Only 5 species² are found in India. Ajuga macrosperma Wall., an annual hairy herb having flowers in spicate whorls and blue corolla, is found in the upper plain area of greater Agartala. We investigated chemically the whole plant of A. macrosperma and detected the presence of twelve amino acids among which six are essential for human system. The identification and relative yields of these amino acids are reported here for the first time. The whole plants of *A. macrosperma* (collected from Suryamaninagar, Agartala in December, 1986) were air-dried and milled. The milled plant material (300 g) was extracted with petrol (b.p. $60-80^{\circ}$) and rectified spirit successively in a soxhlet for 48 h each. The alcoholic extract was concentrated to a gummy residue (20 l g). Chromatography^{8,4} of a part of the residue (1.3 g) through Amberlite IR 120 (H⁺) and elution of the column for neutral and basic amino acids with water and aqueous 4 N HCl⁵ led to the identification of lysine, arginine, hydroxyproline and valine by co-pc with authentic samples.

Another portion of the crude residue (1.6 g) was chromatographed through Amberlite IRA 400 (OH⁻) and elution of the column with water and 2 N acetic acid for neutral and acidic amino acids, led to the identification of glycine, proline, methionine, phenylalanine, leucine, valine, tyrosine, aspartic acid and glutamic acid by co-pc with authentic samples. Valine was found also from the column of cation-exchange resin.

All these amino acids were isolated as their ninhydrin complexes by preparative $pc^{5.6}$ on Whatmann (3 mm) samples using solvent system n-BuOH-HOAc-H₂O (12:3:5) as developer. Their relative yields ($\mu g g^{-1}$) of the air-dried whole plant based on colorimetric estimation are as follows: proline, 67.9; hydroxyproline, 42 8; methionine, 23.8; tyrosine, 18 4; phenylalanine, 15.3; leucine, 14.2; arginine, 13.5; lysine, 9.2; valine, 9.2; glutamic acid, 2.3; aspartic acid, 1.3; and glycine, 1.3.

Acknowledgement

Sincere thanks are due to Dr. N. K. Chakraborty, M. B. B. College, Agartala, for identification of the plant species, to Mr. A. Nath, ex-student of this University for collection of plant material and to U.G.C., New Delhi, for financial assistance.

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Chemical Constituents of the Bark and Leaves

of Pterospermum heyneanum Wall.

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Manuscript received 13 March 1987, revised 3 September 1987, accepted 11 December 1987

PTEROSPERMUM heyneanum Wall. (Syn. P. xylocarpum) is reputed for its medicinal use¹. Isolation of kaempferol, its 3-O-galactoside, kaempferide-7-O-glucoside and β -sitosterol were earlier reported^{2,5} from the stem cuttings of P. heyneanum Wall. In continuation of our work on the plants of Sterculiaceae family⁴, we now report the chemical constituents of the bark and leaves of P. heyneanum.

The air-dried and powdered bark (1.5 kg) of *P.* heyneanum was extracted successively with n-hexane, chloroform and methanol. The plant waxes were removed from the hexane extract by repeated treatment with methanol and the wax-free residue (6 g) yielded n-triacontanol (100 mg), m p. $84-5^{\circ}$; taraxerone (120 mg), m.p. $240-42^{\circ}$; friedelin (150 mg), m.p. $258-59^{\circ}$; taraxerol (100 mg), m.p. $292-93^{\circ}$; and β -sitosterol (200 mg), m.p. $135-36^{\circ}$ on column chromatography. The isolated compounds were identified by their physical and spectral (ir and ¹H nmr) characteristics and direct comparison with the authentic samples.

The residue (4 g) from the chloroform extract on chromatographic resolution over silica gel gave β -sitosterol-3-O- β -D-glucoside (250 mg), compound A (25 mg) and aurantiamide acetate^{s-7} (60 mg). The residue from the methanol extract yielded only an additional quantity of β -sitosterol glycoside.

Compound A was obtained as a pale yellow oil but it showed homogeneity on the (Si-gel, Rt 0.76, $C_{e}H_{e}$ - EtOAc, 9:1; fluorescent yellow spot when sprayed with methanolic sulphuric acid and heated to 80°). It responded to L-B test for steroids, developed yellow colour with TNM and showed positive Zimmermann colour reaction. Its ir (CHCl_a) spectra exhibited bands at 3 450 (OH), 1 720 and 1 680 cm⁻¹ (C=O), and the uv spectra (MeOH) showed maxima at 242 (log ϵ , 4.2) and 273 nm (3.7). Its ¹H nmr spectra showed signals for 4 methyls at δ 0.71-0.98 and two hydrogens of a α,β -unsaturated carbonyl system at δ 5.95 and 6 55 (d, J 10 Hz each) in addition to a 1H multiplet at δ 5.22 and a 1H broad singlet at δ 5.68. From the foregoing observations, compound A is considered to be a \prec,β -unsaturated ketosterol whose complete characterisation could not be carried out for want of material.

Aurantiamide acetate, $C_{2V}H_{28}N_2O_4$ (*M*⁺ 444) was obtained from C_8H_6 – EtOAc (9:1) eluate and it crystallised from benzene as colourless feathery needles, m.p. 185 – 86°, $[{\boldsymbol{x}}]_D - 26.5^\circ$; ν_{max} 3 300, 1730, 1 660, 1 255, 745, 690 cm⁻¹. The compound was characterised from the results of hydrolysis (6 N