

## Synthesis and Biological Studies of some 2-Amino-3-cyano-4-aryl-6-(2'-hydroxy-4'-n-butoxy-5'-H/nitrophenyl)pyridines<sup>†</sup>

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Cyanopyridines exhibit various biological activities<sup>1</sup>. With a view to studying their biological activity, some new cyanopyridines (2) were synthesised. 2'-Hydroxy-4'-n-butoxychalcones (1) and malononitrile were reacted (1:1 molar ratio) in presence of ammonium acetate to give 2-amino-3-cyano-4,6-disubstituted-pyridines (2) through Michael reaction<sup>2</sup> with the elimination of water and hydrogen (Scheme 1). Previously reported chalcones were prepared<sup>3</sup> from 2-hydroxy-4-n-butoxy-5-nitroacetophenone or 2-hydroxy-4-n-butoxyacetophenone and various araldehydes. The structure of the compounds have been supported by elemental analysis, ir and nmr spectral studies.

The compounds (2a—1) were screened for antibacterial activity at a concentration of 50  $\mu\text{g ml}^{-1}$  in

DMF by cup-plate method<sup>4</sup> against gram-positive bacteria *Staphylococcus aureus* and gram-negative bacteria *Escherichia coli* and were compared with chloromycetin and penicillin G. Most of the compounds were found less active or inactive against both the bacteria.

### Experimental

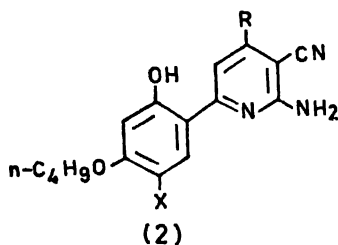
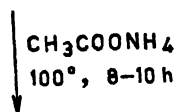
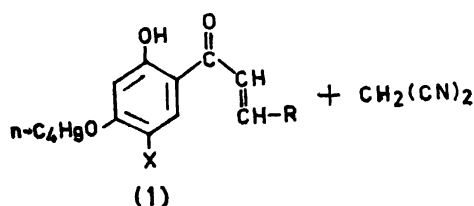
All the melting points were determined in open capillaries and are uncorrected. The ir spectra of the compounds were recorded on a Perkin-Elmer 577 spectrophotometer and pmr spectra ( $\text{CDCl}_3$ ; TMS as internal standard) on a XL-100A (100.1 MHz) spectrometer.

<sup>†</sup>Presented at the 28th Annual Convention of Chemists held at Calcutta, 1991.

TABLE 1—PHYSICAL DATA OF COMPOUNDS (2)\*

Compd. no.	R	X	Mol. formula	M.p. °C	Yield %
2a	Phenyl	NO <sub>2</sub>	C <sub>22</sub> H <sub>20</sub> O <sub>4</sub> N <sub>4</sub>	165	35
b	2'-Bromophenyl	NO <sub>2</sub>	C <sub>22</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Br	190	45
c	4'-Chlorophenyl	NO <sub>2</sub>	C <sub>22</sub> H <sub>19</sub> O <sub>4</sub> N <sub>4</sub> Cl	200	46
d	2',4'-Dichlorophenyl	NO <sub>2</sub>	C <sub>22</sub> H <sub>18</sub> O <sub>4</sub> Cl <sub>2</sub>	227	50
e	2'-Methoxyphenyl	NO <sub>2</sub>	C <sub>22</sub> H <sub>22</sub> O <sub>5</sub> N <sub>4</sub>	170	30
f	3',4'-Dimethoxyphenyl	NO <sub>2</sub>	C <sub>24</sub> H <sub>24</sub> O <sub>6</sub> N <sub>4</sub>	178	37
g	4'-Methylphenyl	NO <sub>2</sub>	C <sub>23</sub> H <sub>22</sub> O <sub>4</sub> N <sub>4</sub>	205	42
h	3',4'-Methylenedioxyphenyl	NO <sub>2</sub>	C <sub>23</sub> H <sub>20</sub> O <sub>6</sub> N <sub>4</sub>	168	40
i	4'-N-Dimethylaminophenyl	NO <sub>2</sub>	C <sub>24</sub> H <sub>26</sub> O <sub>4</sub> N <sub>6</sub>	185	35
j	4'-Bromophenyl	H	C <sub>22</sub> H <sub>20</sub> O <sub>4</sub> N <sub>3</sub> Br	205	38
k	2'-Chlorophenyl	H	C <sub>22</sub> H <sub>20</sub> O <sub>4</sub> N <sub>3</sub> Cl	195	40
l	3',4',5'-Trimethoxyphenyl	H	C <sub>25</sub> H <sub>27</sub> O <sub>6</sub> N <sub>3</sub>	180	33

\* Elemental analyses found satisfactory.

X = NO<sub>2</sub>/H, R = Aryl  
Scheme 1

**2-Amino-3-cyano-4,6-disubstitutedpyridines (2):** A mixture of chalcone (0.1 mol), malononitrile (0.1 mol) and ammonium acetate (0.8 mol) was refluxed in ethanol (30 ml) for 8–10 h on a water-bath. The cooled contents were then poured on ice with constant stirring and the resulting yellow solid was

washed with water and the residue was crystallised from ethanol (Table 1):  $\nu_{\max}$  3 300–3 462 (NH<sub>2</sub>), 3 450–3 560 (OH) and 2 220–2 225 cm<sup>-1</sup> (C≡N),  $\delta$  7.1–7.4 (ArH of pyridine nucleus), 7.2–7.8 (br s, NH<sub>2</sub>) and 4.8–5.2 (s, OH).

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