

Biogenic Fluorescent Protein-Silk Fibroin Phosphors for Highly Performing Light-Emitting Diodes

Verónica Fernández-Luna^{ab}, Juan P. Fernández-Blázquez^a, Miguel A. Monclús^a, Francisco Javier Rojo^c, Rafael Daza^c, Daniel Sanchez-deAlcazar^d, Aitziber L. Cortajarena^d, and Rubén D. Costa^{a*}

Received 00th January 20xx,
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

This work presents a new family of bio-hybrid light-emitting diodes (Bio-HLEDs) using all-bio color down-converting coatings that combine silk fibroin (SF) as a packaging matrix and fluorescent proteins (FPs) as emitters. The thermal, mechanical, optical, and luminescent features of the FP-SF bio-phosphors were optimized with respect to the blend composition and the FP stability under thermal and irradiation stress. Bio-HLEDs with FP-SF filters feature stabilities of 1.2 h and 500 h (200 mA) at efficiencies of ~40 lm/W for on-chip and remote configurations, respectively. This is significantly superior to devices with FP-polymer bio-phosphors – *i.e.*, 1.8 min (on-chip) and 300 h (remote) at <35 lm/W. The excellence of the biogenic phosphors is attributed to i) the ideal optical features of the SF films, ii) the highly efficient emission of the FPs, and iii) the high thermal- and photo-stabilities of FPs in the SF matrix. Finally, we revealed that the emission deactivation process of FP-SF phosphors is related to a very slow protonation of the ionic form of the FP-chromophore along with large morphological changes due to thermal- and photo-induced FP motions. Overall, this work shows the promising prospect of fully biogenic phosphors en route to designing highly stable bio-based lighting devices.

1. Introduction

The excellent mechanical and thermal features of silk fibroin (SF) is fueling its wide use in tissue engineering, drug delivery, diagnostics, and sensing.^{1,2} For instance, the SF from the *Bombyx mori* cocoon feature high tensile strength values around 300–740 MPa^{3,4} along with a breaking elongation of 15% and breaking toughness of 6×10^4 J/kg.⁵ These properties are highly dependent on the spinning rate of the silk fiber, being able to reach similar mechanical properties to those obtained with the *Nephila* spider dragline silk with a breaking strength of 1.3 GPa.³

Recently, the optical features of SF including a refractive index of 1.54^{6,7} and a high light-guiding efficiency,^{8,9} in concert with outstanding environmental and thermal stabilities¹⁰ are attracting increasing attention in the Green Photonics field.^{1,11–18} In this context, the first SF-lasers were provided by Omenetto and co-workers, who reported a cocoon silk 1D distributed feedback lasers with a stilbene as a laser dye, showing an optical pump threshold intensity of $180 \mu\text{J}/\text{cm}^2$ (427 nm).¹⁹ Bêche and co-workers described the integration of pristine spider silk fiber in a photonic chip made of polymer microstructures.²⁰ They also demonstrated efficient optical fibers with different arrangements, showing optical losses of 10.5 dB/cm. In this line,

Sun *et al.* studied the linear and nonlinear optical properties of natural silk fibers,²¹ indicating that unprocessed silk featured an average waveguide losses of 2.8 dB/mm. In a followed up work, Wang and colleagues proposed the application of recombinant spider silk protein and silkworm silk for optical waveguide.²² This work demonstrated that spider silk present smoother surface, higher refractive index ($n = 1.70$) and a lower optical loss in air (0.8 dB/cm) compared to the silkworm protein ($n = 1.52$ and 1.7 dB/cm).

Despite of all the above interesting findings in lasers and light-guiding applications using different type of SF derivatives, SF-based light-emitting diodes (LEDs) has barely been investigated. In 2015, Zhang and co-workers transformed SF films into electrodes for an array of LEDs embedding silver nanowires into the SF network.²³ Recently, SF hydrogels have been used to control the light distribution and light-extraction efficiency in LEDs. These lenses featured a high thermal stability (70 °C) and low refractive index ($n = 1.35$). However, the stability of the bare SF-lenses was of <150 h under illumination.²⁴

Herein, we describe a new use of SF as a matrix to stabilize fluorescent proteins (FPs) to design biogenic phosphors as a new family of bio-hybrid light-emitting diodes (Bio-HLEDs).^{25,26} Bio-HLEDs promise to replace inorganic phosphors (IPs) – *i.e.*, color down-converting materials based on rare-earth elements and Cd-based emitters – for environmentally friendly bio-phosphors. IPs are, indeed, considered one of the most critical roadblocks towards the sustainable development of the LED technology. This is related to i) their negative ecological impact – *i.e.*, mining, lack of recycling, and use of toxic elements,^{27,28} and ii) the visual and non-visual health effects under long expositions to blue light – *i.e.*, irreversible damage of photoreceptors in young children and elderly people along with the alteration of the natural biorhythm of the body (circadian rhythm).^{29–32}

^aIMDEA Materials Institute, Calle Eric Kandel 2, 28906 Getafe, Spain.
E-mail: ruben.costa@imdea.org

^bDepartamento de Física de Materiales, Universidad Autónoma de Madrid, 28049 Madrid, Spain.

^cUniversidad Politécnica de Madrid, Centro de Tecnología Biomédica, Campus Montegancedo, 28223 Pozuelo de Alarcón, Spain.

^dCenter for Cooperative Research in Biomaterials (CIC biomaGUNE), Basque Research and Technology Alliance (BRTA), Paseo de Miramon 182, 20014, Donostia San Sebastián, Spain.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

Up to date, there are two types of bio-phosphors, namely i) biogenic emitters like FPs stabilized in polymer matrices (FP-polymer)^{33–40} and metal organic frameworks (FP-MOFs)^{41,42} and ii) artificial emitters stabilized in biogenic matrices like DNA,⁴³ proteins,^{24,44–49} polysaccharides,^{50–52} and cellulose.^{50,51,53} Table S1 summarizes the most important figures of merit of the above contributions.

As above described, SF has widely been used as scaffold to embedded different artificial emitters, such as QDs, AuNCs, dyes, *etc.*^{8,17} There are, however, a few works combining SF with FPs *via* genetic engineering^{54,55} and solid-state.⁵⁶ None of them has, surprisingly, been applied to solid-state lighting.⁵⁷ Having in mind the excellent optical features of SF films and the efficient color down-conversion of FPs, this combination could be of utmost interest to develop a third family of fully biogenic phosphors (packaging and emitters) towards highly efficient bio-based lighting systems. This constitutes the major thrust of the work at hand.

In this context, we provide a straightforward preparation and optimization of FP-SF color down-converting coatings and their application in highly performing Bio-HLEDs. Compared to Bio-HLEDs based on FP-polymer,³⁹ the FP-SF bio-phosphors provide devices featuring enhanced stabilities – *i.e.*, 1.2 h vs. 1.8 min and >500 h vs. <300 h at 200 mA for on-chip and remote configurations, respectively – and slightly better efficiencies – *i.e.*, 40 lm/W vs. 35 lm/W.³⁹ These results also stand out among the prior-art Bio-HLEDs as summarized in Table S1.

The excellence of the new class of fully biogenic phosphors is attributed to i) the ideal optical features of the SF films, ii) the highly efficient emission of the FPs, and iii) the very slow deactivation mechanism related to a protonation of the ionic form of the FP chromophore in concert with large morphological changes due to thermal- and photo-induced FP motion. Overall, this work shows the promising prospect of fully biogenic phosphors *en route* to designing the third family of Bio-HLEDs towards highly stable bio-based lighting devices.

2. Results and discussion

2.1. Preparation and Characterization of SF-based Coatings

Flexible and self-standing FP-free SF-films were prepared by drop-casting water solutions with 3, 6, and 9 wt.% of SF derived from *B. mori* onto a teflon surface and dried under room conditions – Figure 1. The solutions were prepared with lyophilized degummed SF using the procedure reported elsewhere [see](#) experimental section for details.⁵⁸ The SF-films presented a similar thickness (*ca.* 30 μm), while optical microscope pictures and atomic force microscopy (AFM) measurements showed a homogenous morphology (Figure S1) with a root means square (RMS) roughness values of 7.5 ± 1.7 nm, 2.2 ± 0.7 nm, and 1.7 ± 0.1 nm for 3, 6, and 9 wt.% SF films, respectively.

The thickness, weight, and morphology of the SF-films did not change after 1 month under ambient storage conditions, indicating that the water content holds constant. Likewise, the transmittance and color features of fresh and 1-month-old films

were similar regardless of the SF amount – Figures 1 and S2. In general, SF-films featured transmittances >80% in the visible range and a refractive index of 1.5, indicating good optical properties for color down-converting packaging purposes. In particular, the transmittance of 3 wt.% SF-films was 10 % higher than those with 9 wt.%. This small loss may be related to the increase of scattering events related to the number of crystalline domains.^{59,60} Finally, all the SF-films featured a similar whitish color with x/y CIE color coordinates of 0.33/0.34 – Figure S2.

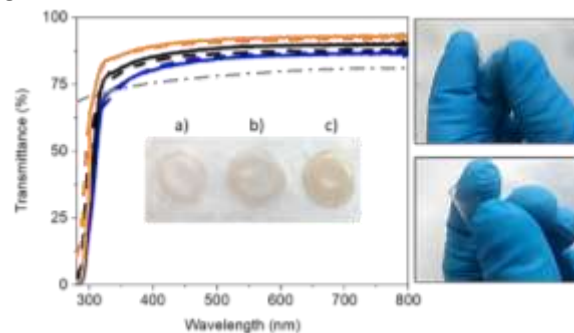


Figure 1. Transmittance spectra of fresh (solid line) and 1 month-old (dashed line) SF-films with 3 (orange), 6 (black) and 9 (blue) wt.% SF compared to a fresh polymer (branch/linear PEO) film (dashed-dotted line). Pictures of 3 (a), 6 (b) and 9 (c) wt.% SF-films (inset) and those highlighting the flexibility of 3 wt.% SF-films (right) are also shown.

Thermal properties of SF-films were studied using thermogravimetric analysis (TGA) under inert atmosphere and modulated differential scanning calorimetry (MDSC). TGA showed three well-defined regions for all samples – Figure S3 and Table 1: i) loss of free water accounting for <10% (35–150 $^{\circ}\text{C}$), ii) <5 % loss of trapped water (150–250 $^{\circ}\text{C}$), and iii) fibroin degradation (>250 $^{\circ}\text{C}$).⁶¹ MDSC, in heat flow profiles – Figure S3, presented a broad endothermic peak starting at 50 $^{\circ}\text{C}$ that corresponds to the water evaporation. This endothermic peak was followed by an endo-exo transition peaking at 170 $^{\circ}\text{C}$ and 210 $^{\circ}\text{C}$, respectively. The endothermic process is associated with the glass transition of the silk as it is observed in reversible heat flow. This provides a high molecular motion of the α -helix fibroin region, while the exothermic band is linked to the β -sheet formation from random coil.^{62–67} Finally, the films presented an endothermal band peaking at 260 $^{\circ}\text{C}$ that corresponds to the decomposition of the β -sheet fibroin.^{63,66,68} The mechanical features – *i.e.*, hardness (H) and reduced modulus (E); Table 1 – were studied using *quasi*-static and cyclic nanoindentation assays at maximum applied loads of 4 and 12 mN, respectively. The H and E features reached constant values in the depth range of 400–600 nm. Within this displacement range, the tip penetration is enough to avoid surface roughness, tip rounding, and compliance effects on the measured H and E values. All the films showed similar H values (~ 0.50 GPa), while E values were close to 10 GPa for 6 and 9 wt.% SF-films and ~ 7.8 GPa for 3 wt.% films – Figure S4. These values are in line with previous reports, suggesting that the morphology of the SF-films consist of a well-interconnected fibroin network.⁶⁹

Table 1. Thermal, optical, and mechanical features of fresh, 1-month-old, heated, and irradiated films with different amounts of SF.

Sample	T _g (°C)	T ^a (%)	x/y CIE coordinates	H (GPa)	E (GPa)
3 wt. %	182	92/91	0.33/0.34	0.48±0.05	7.76±0.57
6 wt. %	181	87/85	0.32/0.33	0.52±0.05	10.06±0.81
9 wt. %	182	84/84	0.32/0.34	0.48±0.07	10.02±0.98
3 wt. % heated ^b	185	88	0.32/0.33	0.6±0.1	10.11±0.61
3 wt. % irradiated ^c	181	79	0.40/0.43	0.52±0.05	8.73±0.60

^aTransmittance of SF-films in the visible region fresh/1 month-old; ^bHeated at 50 °C for 24 h under dark conditions; ^cIrradiated with a 450 nm LED for 24 h under dark conditions.

2.2. Preparation and Characterization of FP-SF Coatings

In light of the high transmittance, the ease handling of the aqueous solutions, and the degree of flexibility of 3 wt.% SF-films, they were doped with the enhanced green fluorescent protein (eGFP) as a color down-converting material. Similar to the SF-films, the FP-doped films were prepared with an homogeneous aqueous solution of SF-eGFP with a total concentration 1.5 mg/mL and a FP:SF mass ratio of 1:20. This mass ratio was previously optimized to ensure a quantitative conversion efficiency of the LED excitation source in Bio-HLEDs – *vide infra*.

To the naked eye, the FP-SF films showed the characteristic pale yellowish white color appearance and strong green emission of the eGFP – Figures 2, S2, and Table 2. Close inspection of the photoluminescence features indicated an increase of the photoluminescence quantum yields (ϕ) from 50% (aqueous solution) to 65% (film) and the excited state lifetimes (τ) from 2.55 to 2.97 ns, respectively. These values account for enhanced radiative and non-radiative rate constants, while confirming that the chromophore of the eGFP is present in its ionic form ($\lambda_{\text{exc}}=490$ nm).^{70–72} Indeed, excitation at 375 nm did not reveal any emission feature of the neutral form of the chromophore at the wavelength range of 400–475 nm – Figure 2. However, the overall emission band of the FP-SF films was slightly red-shifted with a maximum wavelength emission ($\lambda_{\text{max}}=490$ nm) of 520 nm compared to that in solution (510 nm). This suggests that during the drying process the SF-FP interaction is enforced, leading to a slightly distorted β -barrel that promotes small structural changes of the microenvironment surrounding the chromophore without changing its ionic form. This behavior has also been identified in FP-polymer bio-phosphors as shown in

Figure 2 and Table 2. However, this feature is a key aspect towards highly stable Bio-HLEDs, since the kinetics of the photo-induced deactivation of the chromophore from its ionic to neutral forms is significantly slowdown in this distorted conformation promoted by the interaction with the surrounding matrix.³⁹

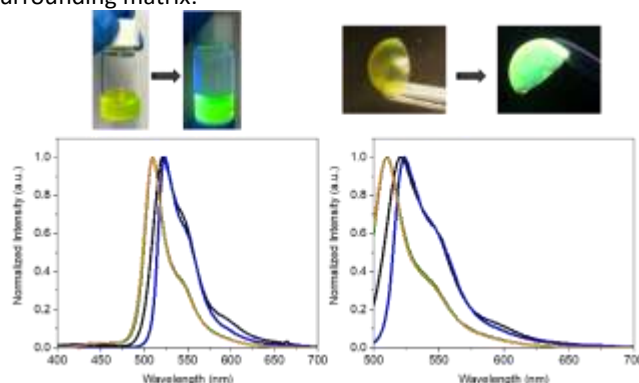


Figure 2. Top: Pictures of the FP-SF aqueous solutions (left) and films (right) under room and UV irradiation. Bottom: Emission spectra at excitation wavelengths of 375 nm (left) and 490 nm (right) of FP-SF aqueous solutions (green), FP aqueous solution (orange), FP-SF films (black), and FP-polymer films (blue).

Further confirmation of the FP-SF interaction was acquired from the thermal and mechanical features of these films. On one hand, MDSC assays did not display the characteristic endo-exo transition between 170 and 200 °C in SF-films – *vide supra*, as it was overlapped with the broad endothermic process related to the protein degradation – Figure S5. On the other hand, the presence of FP leads to stiffer films as both, H and E values, increase compared to those of pristine SF-films – Tables 1 and 2. TGA assays indicated that FP-SF films contain lower amount of water – *i.e.*, < 1 or 2% – but enough to have an influence on mechanical properties. However, the morphology of fresh FP-SF films showed the same homogenous morphology to that of SF-films – Figure S6. The absence of a clear T_g is rationalized by the presence of the FP. Owing to the strong FP-SF interaction, the global molecular motion of the α -helix fibroin region is restricted, avoiding the β -sheet formation turns to a random coil. As such, the enhanced mechanical behaviour is related to the presence of FP that slightly reduces the water content after drying and interacts with the SF matrix – *vide supra*, leading to significant plasticizer effect.

Table 2. Mechanical, optical, and photoluminescent properties of FP-SF solution, FP-polymer, and FP-SF bio-phosphors.

Sample	H (GPa)	E (GPa)	x/y CIE coordinates	λ_{\max}^d (nm)	ϕ^e (%)	τ^f (ns)	k_r ($\times 10^8 \text{ s}^{-1}$)	k_{nr} ($\times 10^8 \text{ s}^{-1}$)
FP-SF solution ^a	-	-	-	510	50	2.55	1.96	1.96
FP-polymer	-	-	-	524	70	2.03	3.45	1.48
FP-SF	0.56 \pm 0.02	9.76 \pm 0.11	0.31/0.33	520	65	2.97	2.19	1.18
FP-SF heated ^b	0.17 \pm 0.02	1.35 \pm 0.20	0.33/0.35	524	47	2.47	1.82	2.23
FP-SF irradiated ^c	0.17 \pm 0.02	0.13 \pm 0.01	0.40/0.43	520	60	2.90	2.07	1.38

^aMili-Q water; ^bHeated at 50 °C for 24 h under dark conditions; ^cIrradiated with a 450 nm LED for 24 h under dark conditions; ^dExcitation at 375 nm; ^eExcitation at 485 nm; ^fAverage excited-state lifetimes at 450 nm excitation.

2.3. Preparation and Characterization of Bio-HLEDs with FP-SF Coatings

Bio-HLEDs were prepared using the named on-chip configuration, in which the commercial blue-emitting chip (450 nm; WINGER electronics) was directly covered with the above SF and FP-SF filters prepared with a dome form (25 μm thick and 7 mm in diameter) as shown in Figures 2 and 3 – see experimental section for preparation details. For reference purposes, devices with FP-polymer bio-phosphors are also discussed.

At first, the changes of the electroluminescence spectrum and luminous efficiency were monitored upon increasing the applied current from 5 to 200 mA – Figures 3 and S8. Regardless of the applied current, the maximum luminous efficiencies (*ca.* 20 lm/W at 30–50 mA) that are dominated by the blue-emitting electroluminescence spectrum of both, commercial LEDs and the Bio-HLEDs with SF-filters, exhibited similar values – Figure S8. This corroborates the excellent optical features of SF-films for color down-conversion purposes in hybrid LEDs – *vide supra*. In contrast, Bio-HLEDs with FP-SF filters featured slightly higher luminous efficiencies (*ca.* 40 lm/W at 30–50 mA) caused by their better transmittance values compared to the polymer films – *i.e.*, mixture branch/linear PEO as shown in Figure 1 and see experimental section. The device emission is dominated by an electroluminescence spectrum centered at 520 nm that corresponds to the FP emission. Common to all the devices, the maximum luminous efficiency is followed by a significant reduction caused by the loss of the internal quantum efficiency of the electron-hole recombination at the emitting chip – Figure 3.⁷³

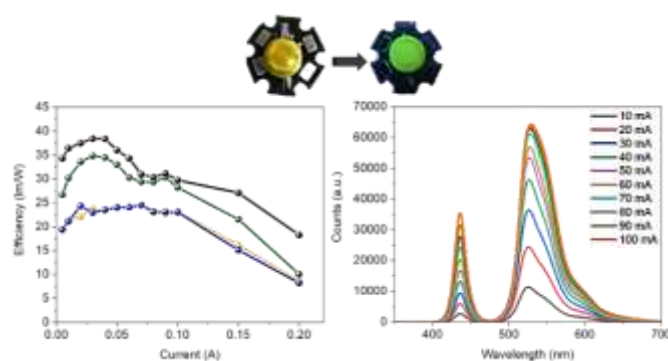


Figure 3. Top: Pictures of Bio-HLEDs with FP-SF filters. Bottom: Luminous efficiency vs. applied current (left) for a commercial 450 nm LED (blue), Bio-HLEDs with SF-films (orange), Bio-HLEDs with FP-SF-films (black), and Bio-HLEDs with FP-polymer (green). The electroluminescence spectra (right) of Bio-HLEDs with FP-SF-films at different driving currents are also shown.

Likewise Bio-HLEDs with FP-polymer filters, the major limitation of using FP-SF filters is the photo-induced heat generation, reaching a maximum temperature of 43 °C at 200 mA – Figure 4. This contrasts with the maximum temperature of <30 °C at 200 mA for commercial LEDs and Bio-HLEDs with SF filters. The heat generation in Bio-HLEDs has recently been attributed to the photo-induced vibrational and rotational FP motions.³⁹ Upon strong photon flux fluencies, FPs act as a hot spot that transmits the heat to the polymer surrounding until the melting temperature of the polymer network is reached. This strongly weakens the FP-matrix interaction provoking a quick and irreversible FP denaturation. Thus, devices with FP-polymer coatings feature a linear increase of the temperature with the applied current up to values close to 60 °C at 200 mA – Figure 4.³⁹ In contrast, Bio-HLEDs with FP-SF coatings featured i) a much lower maximum temperature and ii) a different photo-induced temperature generation behavior that consists of two linear regimes going from 0 to 100 mA reaching 40 °C and from 100 to 200 mA, in which the temperature slightly increases up to 42 °C. This behavior is also noted for devices with SF-filters and commercial LEDs. Thus, SF filters show an effective heat dissipation at high applied currents that is attributed to i) the higher thermal conductivity of the SF films compared to those of the polymer coatings – *i.e.*, ~6.50 W/mK vs. ~0.3–0.5 W/mK

–⁷⁴ and ii) the enhanced mechanical features that constrains FP motion.

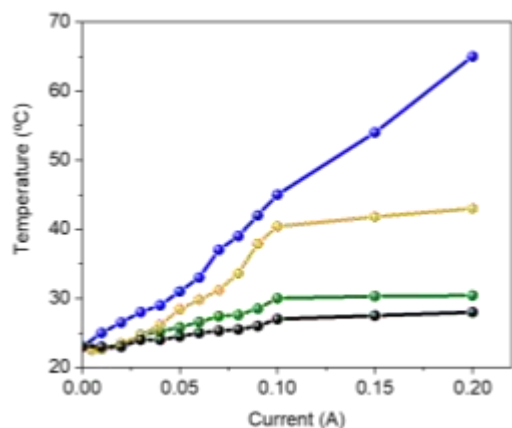


Figure 4. Changes on the temperature vs. applied current for commercial 450 nm LED (black) and Bio-HLEDs with FP-polymer (blue), FP-SF (orange), and SF (green) bio-phosphors.

The Bio-HLED stability – *i.e.*, t_{50} or time to reach half of the initial emission intensity – was studied at high driving currents of 200 mA, monitoring the intensity decay of the color down-converting band and the temperature of the filters – Figure 5. The emission decay obeyed a bi-exponential behavior. The first decay regime went hand-in-hand with a quick increase of the filter temperature without showing any change in the emission band – Figure 5. The second regime was characterized by a slow decay in both, temperature and emission intensity, that are associated to a slightly blue-shifted emission band (<5 nm). Here, the emission plateaus at *ca.* 30 % of the initial value for more than 16 h. This behavior significantly contrasts with that of Bio-HLEDs with FP-polymers, in which the emission band is red-shifted in >15 nm and the emission intensity is completely vanished – Figure 5. Indeed, Bio-HLEDs with FP-SF filters featured two orders of magnitude higher t_{50} value (1.2 h) than those with FP-polymer filters (1.8 min) – Figure 5. All-in-all, these findings suggest that FP-SF bio-phosphors provide Bio-HLEDs with a superior performance caused by a less severe heat generation and FP deactivation processes than those noted in FP-polymer filters.

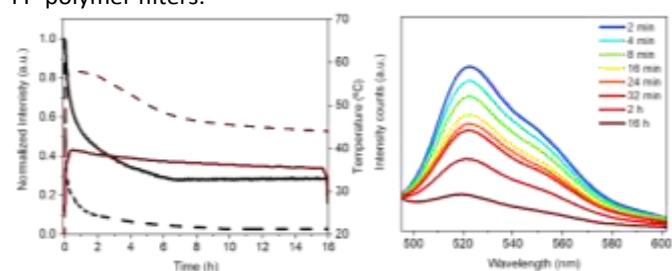


Figure 5. Typical performance of 'on-chip' Bio-HLEDs operating at high-applied currents of 200 mA. Left: Overall emission intensity decay (black) along with temperature changes (red) over time for Bio-HLEDs with FP-SF (solid line) and FP-polymer (dashed line). Right: Electroluminescence spectra of Bio-HLEDs with FP-SF phosphors over time.

2.4. Deactivation Mechanisms in Bio-HLEDs with FP-SF Coatings

To determine the degradation mechanism in FP-SF filters towards long-living Bio-HLEDs, we decided to decouple the

thermal and irradiation stresses, monitoring the changes of the photophysical, thermal, mechanical, and morphological features of the filters.

2.4.1. Temperature Induced Degradation Mechanism

The isothermal stability assays were carried out placing fresh FP-SF filters onto a dark heating station (50 °C for 24 h), monitoring their emission features upon exciting at 450 nm for 5 s every 10 minutes – Figures 6 and S9. During the heating process, the loss percentage of emission intensity (I_l) associated to a quenching rate (k_q) is obtained. This is followed by a cooling down process to room temperature under dark conditions, in which the gain percentage of the emission intensity (I_g) at a certain recovery rate (k_{re}) is noted. Typically, there is a further emission intensity recovery (I_{sh}) that is related to a self-healing process taking place at room temperature and dark conditions over long periods of time.

In particular, heating led to a quick emission loss ($I_l = 65\%$; $k_q = 0.5 \text{ h}^{-1}$) followed by a slight reduction of the emission intensity (<5%) after 24 h. The emission band shape held constant, showing a slightly blue-shifted maximum emission wavelength (<5 nm; Figure 6) that suggests a partial change on the microenvironment of the protein chromophore – *vide infra*. Cooling provoked a quick gain in emission intensity ($I_g = 40\%$; $k_{re} = 0.2 \text{ h}^{-1}$) that accounts for a recovery up to 70% of the fresh emission intensity of a FP-SF coating. This is slightly enhanced after the heated FP-SF coatings were kept in dark and room temperature for *ca.* 200 h, reaching a final I_{sh} value of ~80%. Both recovery processes did not show any significant change of the emission band shape. As a reference purpose, FP-polymer coatings showed a more pronounced and quicker thermal emission quenching ($I_l = 73\%$; $k_q = 0.97 \text{ h}^{-1}$), while the sum of both thermal and self-healing recoveries do not present a significant higher I_{sh} value than the FP-SF films – see inset Table 3 and Figure 6 and Figure S10.

To get more insights into the degradation mechanism, the photophysical features of fresh and heated FP-SF filters were compared – Figure S10 and Table 2. In short, several studies have shown that the emission deactivation of FP could be related to i) changes in the nature of the emitting excited state *via* excited state proton transfer (ESPT) – *i.e.*, neutral, anionic, and intermediate forms,⁷⁵ ii) population of dark states – *i.e.*, dark state, triplet or *cis/trans* isomerization,⁷⁶ iii) photooxidative reddening,⁷⁷ and iv) reversible photochromism.^{78,79} The last two processes are discarded as they lead to the formation of highly emissive red-emitting species that are not noted in SF-FP filter.

Excitation at 375 nm revealed the band shape of the ionic chromophore ($\lambda_{max} = 522 \text{ nm}$; corroborated at excitation at 490 nm) and the arising presence of its neutral form ($\lambda_{max} = 430 \text{ nm}$) – Figure S11. This implies that the deactivation mechanism is related to a slow transformation of the FPs featuring highly emissive ionic forms of the chromophore to those with the poorly emissive neutral form. This is driven by the partial distortion of the β -barrel caused by temperature-induced variations of the external environment that promotes small

changes in the chromophore cavity leading to a slow ESPT processes. This is further confirmed by the decrease in ϕ and τ (47% and ca. 2.47 ns (λ_{exc} = 450 nm), respectively) after thermal stress compared to fresh SF-FP filters – Table 2. Please, notice that the deactivation via the population of dark states and/or fully denaturation for the proteins lead to a lack of emission.⁷⁶ Finally, AFM and optical images of the heated films revealed a dramatic change in the film morphology, showing a higher degree of heterogeneity and roughness that is strongly juxtaposed to that of heated FP-free SF films – Figures S6 and S7. For instance, heated SF-films exhibited similar thermal and mechanical features with a slightly loss in transmittance of <5% compared to the fresh SF-films – Figures S2, S5 and Table 1. These findings indicate that a small degree of short range rearrangement might be present in heated SF-films. In stark contrast, the morphological changes of the heated FP-SF films translated into, for example, a significant reduction of H and E values compared to fresh FP-SF films – *i.e.*, H: 0.17 GPa vs. 0.56 GPa; E: 1.35 GPa vs. 9.76 GPa, respectively – Table 2.

Table 3. Thermal deactivation parameters associated to the isothermal assay at 50 °C under dark conditions of FP-polymer and FP-SF bio-phosphors.

Sample	I_i (%)	k_q (h ⁻¹)	I_g (%)	k_{re} (h ⁻¹)	I_{sh} (%)
FP-polymer	73	0.97	40	0.67	>80
FP-SF	65	0.5	40	0.2	>80

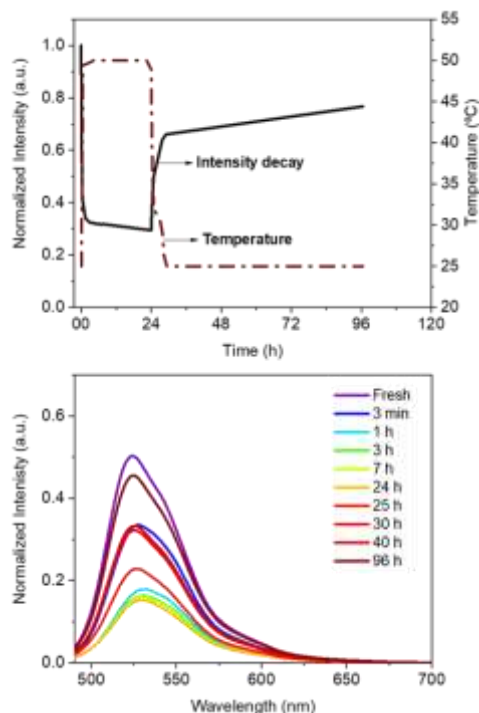


Figure 6. Thermal stability of FP-SF bio-phosphors at a constant temperature (50 °C) for 24 h and after cooling, monitoring the emission intensity (top), and emission band shape (bottom) upon excitation with a blue LED (450 nm) for 5 s every 10 min. Inset: Table 3 summarizing the most relevant parameters in FP-SF films vs FP-polymer in the emission intensity changes and the quenching and recovery rates during the isothermal assay.

All-in-all, the use of a SF packaging matrix enhances the thermal stability of the embedded FPs compared to those with

polymers. The temperature induced deactivation mechanism seems to be similar through a small distortion of the β -barrel promoting the protonation of the ionic chromophore form. This process should also be promoted by long range morphological arrangements of the FP-SF network upon heating.

2.4.2 Irradiation Induced Degradation Mechanism

To avoid the photo-induced heat generation at high applied currents, we fabricated the named remote Bio-HLEDs, in which the FP-SF filters are placed at 2 cm distance from the emitting chip. At 200 mA, the emission intensity and electroluminescent emission band of these Bio-HLEDs were monitored – Figure 7. The decay of the emission intensity followed four well-defined regimes. First, a small emission decrease is related to the slightly increase in temperature that raised during the first 24 h and held fairly constant at <30 °C for the whole measurement. This is followed by a second emission decrease of <10%, reaching a plateau that holds for 120 h. In both regimes, the emission band shape remains constant. At this point, the emission intensity exhibited another 18% loss until 150 h along with a slightly blue-shifted emission (521 nm). These emission features kept constant until 450 h, at which another slow decay of 11 % took place. This is in concert with a blue shifting of λ_{max} to 516 nm and a new emission plateau at approximately 40% of the initial intensity for >1,000 h. This resulted in a device lifetime (t_{50}) of around 500 h that is much higher than the previous record noted for Bio-HLEDs with FP-polymer (<300 h).³⁹ This is in perfect agreement with the stability enhancement noted for on-chip Bio-HLEDs operating at the same high applied currents – *vide supra*.

The spectroscopic changes of the FP-SF coatings were also obtained at each plateau regime – Figure S11 and Table 4. As above stated, the emission features slightly changed during the first 120 h – *i.e.*, ϕ of 55% and τ of 2.55 ns, indicating that the majority of the FPs presented the ionic chromophore form; though a slightly conversion to the neutral chromophore form is noted – Figure S11. However, the most critical aspect is the large morphological changes that might be promoted by FP motion upon constant excitation – Figures S6 and S7. Similar to the isothermal assays, the mechanical and optical features of the irradiated FP-SF films were dramatically affected by the morphological changes – Figure S2 and Table 2. For reference purposes, irradiated SF-films did not show any change – Figures S2 and S6. Thus, the initial loss in the device intensity might be mainly related to i) the slight increase in temperature and ii) changes in the transmittance and refractive index values caused by the large morphological arrangements.

During the third regime (150-450 h), the emission band shape is significantly broadened and slightly blue-shifted – Figures S11 and 7. This is in concert with a reduction of the ϕ and τ values associated to the arising of the emission band of the neutral form of the chromophore – Figure S11. Both findings indicate that a partial distortion of the β -barrel promoting the deactivation of the chromophore to its neutral form. Finally, this is even more pronounced during the fourth regime (500-1000 h), showing a further broadening of the ionic emission band, a reduction of the ϕ , and an increase of the emission band of the neutral chromophore form – Figure S11 and Table 4. Hence, Bio-

HLEDs with FP-SF filters showed enhanced stabilities that FP-polymer coatings, since the deactivation process is related to a very slow protonation of the ionic form of the FP chromophore in concert with large morphological changes upon irradiations.

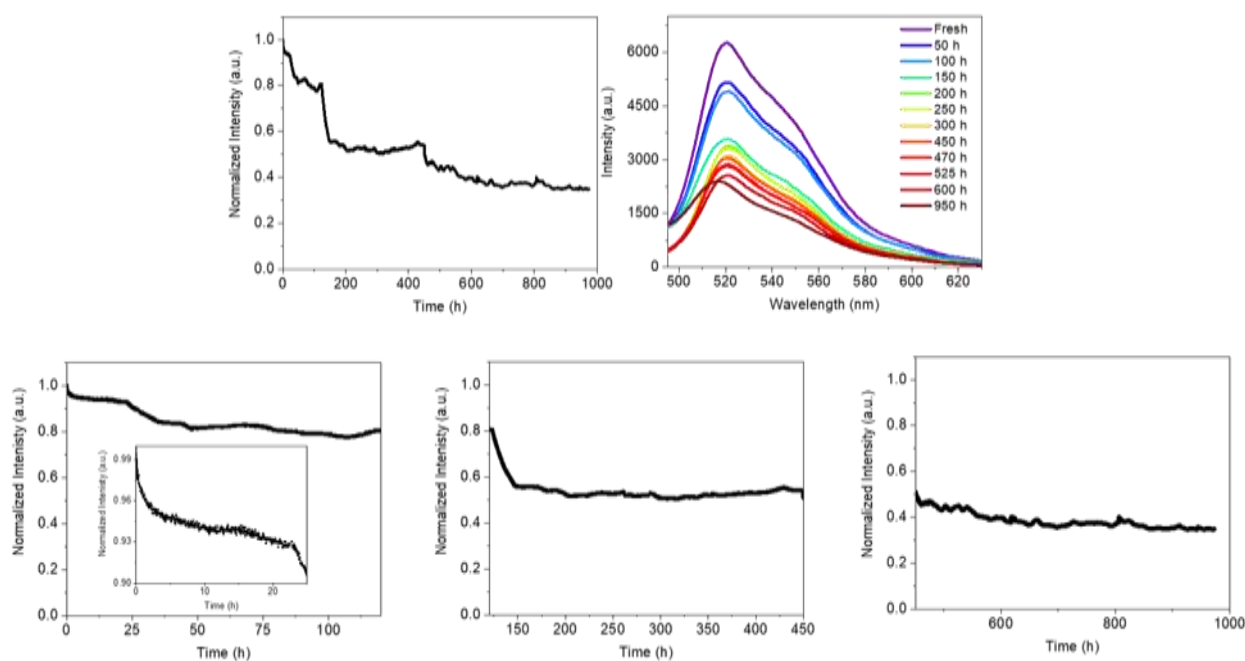


Figure 7. Typical performance of remote (2 cm) FP-SF based Bio-HLEDs operating at a high applied currents at 200 mA. Top: Overall emission intensity decay (left) and the electroluminescence spectra changes (right) over time. Bottom: Emission intensity decay divided into four regimes, namely 0-24 h (inset graph), 24-120 h (left), 120-450 h (centre), and 450-1000 h (right).

Table 4. Photoluminescence parameters of fresh FP-SF films and those used in remote Bio-HLED (2 cm) at different degradation stages.

Sample	λ_{\max}^a (nm)	ϕ (%)	τ^c (ns)	k_r ($\times 10^8 \text{ s}^{-1}$)	k_{nr} ($\times 10^8 \text{ s}^{-1}$)
FP-SF fresh	520	65	2.97	2.19	1.18
"On-chip"	520	25	2.58	0.97	2.90
100 h	521	55	2.55	1.86	1.52
375 h	522	50	2.58	1.68	1.68
600 h	521	44	2.21	1.99	2.53
975 h	516	30	1.82	1.65	3.85

^aExcitation at 375 nm; ^bExcitation at 485 nm; ^cAverage excited-state lifetimes at 450 nm excitation

3. Conclusions

This work presents the first study dealing with fully biogenic phosphors for HLEDs. In particular, the preparation and optimization of self-standing color filters combining SF as a packaging matrix and FP as emitters is presented. The major interest on SF-FP bio-phosphors lies on i) the ideal optical features of the SF films, ii) the highly efficient emission of the FPs, and iii) the high thermal- and photo-stabilities of FPs in the SF matrix. This leads to Bio-HLEDs with enhanced stabilities of 1.2 h (on chip) and >500 h (remote) operating at high applied

currents of 200 mA compared to those using FP-polymers as bio-phosphors – *e.g.*, 1.8 min (on-chip) and <300 h (remote). Noteworthy, the device performance of this novel class of fully biogenic phosphor based Bio-HLEDs stands out among others Bio-HLEDs based on biogenic matrices embedding artificial emitter^{43,50,51,80,81} and biogenic emitters in MOFs,^{41,42} polymers,^{33–36,39,40,47,82} and dry films^{48,49,53} as summarized in Table S1.

The major limitations of this family of bio-phosphors were also deciphered. On one hand, the presence of FPs into the SF matrix strongly affects the thermal and mechanical features compared to pristine SF-films. This is even more dramatic under

thermal and irradiation stress, leading to large morphological changes due to FP motion. On the other hand, the SF matrix effectively enhances both thermal and photo-stabilities of the FPs, but the deactivation mechanism *via* the protonation of the ionic chromophore form is still active. It is therefore highly desired to control the content and water motion in these coatings. Both limitations could be tackled i) using SF of different lengths, ii) chemically cross-linking of the SF matrix, and iii) bio-conjugating the FP-SF moieties. These actions are on-going in our laboratories.

Overall, the major thrust of the work at hand is to assert the brilliant prospect of a new family of all-bio phosphors *en route* to biogenic lighting applications.

4. Experimental section

4.1. Materials

All materials were purchased from Sigma-Aldrich and used without further treatment. Teflon molds were prepared and cut forming a curve coating with a width of 8.0 mm and depth of 5 mm.

4.2. Protein Expression and Purification

eGFP was expressed in *Escherichia coli* DH10b cells grown in Luria–Bertani (LB) liquid medium for 24 h at 37 °C. The cells were harvested by centrifugation at 4500 rpm for 15 min and the pellets were resuspended in lysis buffer (300 mM NaCl, 50 mM Tris/HCl pH 7.4) with 1 mg/ml lysozyme, 5 mM β -mercaptoethanol and 5 μ l of DNase stock solution. The lysate was sonicated 5 min with 30 sec of intervals and 40% of amplitude, and then centrifuged at 10000 rpm for 45 min. The supernatant solution was loaded into Ni²⁺ HisTrap™ column, and the protein was eluted with 300 mM NaCl, 300 mM imidazole, 50 mM Tris/HCl pH 7.4. The fractions containing the protein were pooled and dialyzed into Milli-Q water. The purity of the samples was determined by SDS/PAGE and spectrometrically. The protein was concentrated to 10 mg/ml then stored in aliquots at -20 °C.

4.3. Preparation of Silk Fibroin Powder

In accordance with previous works, *B. mori* cocoons, kindly provided by IMIDA (Murcia, Spain), were cut in small pieces and soaked in a bottle with 0.2% (w/v) Na₂CO₃ solution, at a concentration of cocoons/solution of 2% (w/v). They were degummed by boiling for 30 min in autoclave (121 °C, 103.4 kPa), followed by 20 min drying under ambient conditions. During the degumming process, sericin is removed from cocoons and fibroin is the only remaining protein. After the procedure, silk fibroin fibers were repeatedly rinsed in tap water, followed by rinsing in distilled water and dried overnight at room temperature. Then, the dried fibers were dissolved in 9.3M lithium bromide (LiBr) solution at a concentration fibers/solution of 20% (w/v) during 4 hours at 60 °C under gentle stirring. LiBr was cleaned from the solution by dialysis against 50 mM Tris base, pH 8 and 4 °C water for 72 h using BioDesignDialysis Tubing™ (MWCO 3.5 KDa). After dialyzing, the solution was centrifuged at 5000 rpm for 20 min and the supernatant was frozen at -80 °C and lyophilized. Finally, silk

fibroin (SF) obtained was grinded to make powder and it was sealed and stored at -20 °C for further used.

4.4. Preparation of Protein Coatings

The FP-polymer coatings were prepared as follows. A mixture of 4:1 ratio with TMPE:PEO was added and stirred (300 rpm) to an eGFP solution (0.45 mg) under ambient conditions overnight, forming a hydrogel. The FP-polymer films were obtained after a gentle vacuum process from 200 to 3 mbar for 4 h. The FP-free coatings were also prepared following the above description using without eGFP for reference purposes. The FP-SF coating were prepared as follows. First, a SF solution of 3 wt.% was dissolved in a final volume of 300 μ l of Milli-Q water. The mixture was stirred (400 rpm) for 3 h under ambient conditions. The eGFP solution (0.45 mg) was added to the SF preparation, and stirred under mild conditions (400 rpm), the final mixture was poured into the Teflon molds with a dome-like design, and left to dry under ambient conditions overnight, forming a perfectly curved thin film. The SF coatings were prepared using the protocol above-named, dissolving different wt.% of SF into 300 μ l of Mili-Q water. Once the solutions were prepared, they were stored at -4 °C until further use.

4.6. Characterization Techniques

The photophysical studies were carried out using a FS5 Spectrofluorometer (Edinburgh Instruments) with the SC-10 module for solid samples, the SC-30 Integrating Sphere to determine ϕ , and the 450 nm time-correlated single photo-counting or TCSPC (64.3 ps pulse width) module to determine τ . The measurements were performed at room temperature. The FP-coatings were also measured under constant heating at 50 °C in dark conditions. Both, the devices after the photo- and thermal-processes were spectroscopically analyzed using the spectrofluorometer above-mentioned. Emission spectra under excitation of 375 nm and 490 nm as well as ϕ were measured. The time decay values were monitored and then adjusted to a biexponential decay fit using Origin 8b. To calculate the average lifetime for each FP-coating, the following equation was used:⁸³

$$\langle \tau \rangle_f = \frac{\int_0^x t \sum a_i \exp\left(-\frac{t}{\tau_i}\right) dt}{\int_0^x \sum a_i \exp\left(-\frac{t}{\tau_i}\right) dt} = \frac{\sum a_i \tau_i^2}{\sum a_i \tau_i} \quad (1)$$

Where a_i (λ) is the amplitude fractions and τ_i are the lifetimes. The SF-films were subjected to constant heating at 50 °C for 24 h, and the changes of the emission were also monitored. Both the devices after the photo- and thermal-processes were spectroscopically analyzed using the aforementioned spectrofluorometer. Transmittance spectra was acquired with a UV-Vis spectrometer UV-2600 (Shimadzu), using a wavelength range 200–800 nm, scan speed medium, threshold 0.01 and a slit width of 2.0.

Thermogravimetric curves were measured with a TGA-Q50 (TA instruments) analyzer under N₂ atmosphere with a flow rate of 90 mL/min, and a temperature ramp of 10 °C/min from room temperature up to 450 °C. Modulated differential scanning calorimetry was performed with a Q200 (TA instruments) with a modulation of 0.48 °C every 60 s and a temperature ramp of 3 °C/min from 20 to 300 °C.

Atomic force microscopy (AFM) was performed with a Park XE150 instrument in ambient conditions, using Si cantilevers with a nominal spring constant of 40 N/m, and a tip radius < 10 nm. Topography images (45 x 45 μm scan size) were obtained by scanning the samples surface in non-contact mode, with a resolution of 512 pixels and a scan rate of 0.5 Hz. Roughness data was calculated using the image analysis XEI Park Systems software. Further characterization of the film morphology was performed using an optical microscope Olympus BX51 with and Olympus camera adaptor U-TV0.5XC-3-8. The films were placed onto a quartz slide and analyzed at 10x magnification. An augmentation of 10X was used to take the images.

The mechanical properties of the material were measured by applying the instrumented nanoindentation technique using a Hysitron TI 950 TriboIndenter equipped with a pyramidal Berkovich diamond indenter, with a tip radius of ~ 350 nm. The area function for the tip was calibrated from indentations into a fused silica reference sample of known elastic modulus. Two types of load-controlled tests were performed: (1) "load-partial-unload" tests, up to a maximum load of 12 mN, which were used to identify penetration depths for which hardness and modulus values are not affected by roughness and tip rounding effects and (2) quasi-static indentation tests, up to a maximum load of 4 mN and performed using load-hold-unload cycle times of 10-30-3 s, which were used to study the effects of creep on the measured properties. Arrays of 2x7 indentations were performed on each sample, with a spacing between indents of 25 μm in both, vertical and horizontal directions. Reduced elastic modulus (E) and hardness (H) were obtained from the analysis of load-displacement curves by the Oliver and Pharr method,⁸⁴ using the following equations:

$$E = \frac{\sqrt{\pi}}{2\sqrt{A}} S \quad (2)$$

$$H = \frac{P_{\max}}{A} \quad (3)$$

Where, P_{\max} is the peak indentation load, A is the projected area, and S is the contact stiffness value.

4.5. Device Preparation and Characterization

The bio-phosphors were placed at different distances from the 450 nm LED (WINGER Electronics) and irradiated at 200 mA at room conditions. The Bio-HLEDs were characterized using a Keithley 2400 as a current source, while the changes in the electroluminescence spectrum were monitored using an AVS-DESKTOP-USB2 (Avantes) in conjunction with a calibrated integrated sphere Avasphere 30-Irrad, while changes in the bio-phosphor temperature were monitored using a thermographic camera T430sc (FLIR).

Acknowledgements

[**] V.F-L. and R.D.C. acknowledge the program "Ayudas para la atracción de talento investigador—Modalidad 1 of the Consejería de Educación, Juventud y Deporte—Comunidad de Madrid with the Reference No. 2016-T1/IND- 1463." R.D.C. acknowledges Spanish MINECO for the Ramón y Cajal program (RYC-2016-20891) and the Europa Excelencia program

(ERC2019-092825). R.D.C. acknowledges the 2018 Leonardo Grant for Researchers and Cultural Creators from BBVA Foundation and the European Union's Horizon 2020 research and innovation programme under grant agreement No 863170. A.L.C. acknowledges the European Research Council ERC-CoG-648071-ProNANO, ERC-PoC-2018-841063-NIMM, and the Spanish Ministry of Economy and Competitiveness (BIO2016-77367-R). This work was performed under the Maria de Maeztu Units of Excellence Program from the Spanish State Research Agency – Grant No. MDM-2017-0720 (CIC biomaGUNE).

References

- 1 L. D. Koh, Y. Cheng, C. P. Teng, Y. W. Khin, X. J. Loh, S. Y. Tee, M. Low, E. Ye, H. D. Yu, Y. W. Zhang and M. Y. Han, *Prog. Polym. Sci.*, 2015, **46**, 86–110.
- 2 C. Zhou, F. Confalonieri, N. Medina, Y. Zivanovic, C. Esnault, T. Yang, M. Jacquet, J. Janin, M. Duguet, R. Perasso and Z. Li, *Nucleic Acids Res.*, 2000, **28**, 2413–2419.
- 3 Z. Shao and F. Vollrath, *Nature*, 2002, **418**, 741–741.
- 4 G. H. Altman, F. Diaz, C. Jakuba, T. Calabro, R. L. Horan, J. Chen, H. Lu, J. Richmond and D. L. Kaplan, *Biomaterials*, 2003, **24**, 401–416.
- 5 M. A. Wilding and J. W. S. Hearle, ed. J. C. Salamone, CRC Press, Boca Raton, Florida, 1996, pp. 8307–8322.
- 6 H. Perry, A. Gopinath, D. L. Kaplan, L. D. Negro and G. Omenetto, *Adv. Mater.*, 2008, **20**, 3070–3072.
- 7 A. Bucciarelli, V. Mulloni, D. Maniglio, R. K. Pal, V. K. Yadavalli, A. Motta and A. Quaranta, *Opt. Mater.*, 2018, **78**, 407–414.
- 8 H. Tao, D. L. Kaplan and F. G. Omenetto, *Adv. Mater.*, 2012, **24**, 2824–2837.
- 9 B. S. T. Parker, P. Domachuk, J. Amsden, J. Bressner, A. Lewis, D. L. Kaplan and F. G. Omenetto, *Adv. Mater.*, 2009, **21**, 2411–2415.
- 10 X. Huang, G. Liu and X. Wang, *Adv. Mater.*, 2012, **24**, 1482–1486.
- 11 F. G. Omenetto and D. L. Kaplan, *Science*, 2010, **329**, 528–531.
- 12 M. Humar, S. J. J. Kwok, M. Choi, A. K. Yetisen, S. Cho and S. H. Yun, *Nanophotonics*, 2017, **6**, 414–434.
- 13 D. D. N. Rockwood, R. R. C. Preda, T. Yücel, X. Wang, M. L. Lovett and D. L. Kaplan, *Nat. protoc.*, 2011, **6**, 1612–1631.
- 14 M. Irimia-Vladu, *Chem. Soc. Rev.*, 2014, **43**, 588–610.
- 15 F. G. Omenetto and D. L. Kaplan, *Nat. Photonics*, 2008, **2**, 641–643.
- 16 G. S. Perrone, G. G. Leisk, T. J. Lo, J. E. Moreau, D. S. Haas, B. J. Papenburg, E. B. Golden, B. P. Partlow, S. E. Fox, A. M. S. Ibrahim, S. J. Lin and D. L. Kaplan, *Nat. Commun.*, 2014, **5**, 3385.
- 17 N. Lin, L. Cao, Q. Huang, C. Wang, Y. Wang, J. Zhou and X. Y. Liu, *Adv. Funct. Mater.*, 2016, **26**, 8885–8902.
- 18 S. Shabahang, S. Kim and S. Yun, *Adv. Funct. Mater.*, 2018, **28**, 1706635.
- 19 S. Toffanin, S. Kim, S. Cavallini, M. Natali, V. Benfenati, S. Toffanin, S. Kim, S. Cavallini, M. Natali, M. Muccini and F. G. Omenetto, *Appl. Phys. Lett.*, 2012, **101**, 091110.

- 20 N. Huby, V. Vié, A. Renault, S. Beauvils, T. Lefèvre, F. Paquet-Mercier, M. Pézolet and B. Bêche, *Appl. Phys. Lett.*, 2013, **102**, 123702.
- 21 S. Kujala, A. Mannila, L. Karvonen, K. Kieu and Z. Sun, *Sci. Rep.*, 2016, **6**, 22358.
- 22 X. Qiao, Z. Qian, J. Li, H. Sun, Y. Han, X. Xia, J. Zhou, C. Wang, Y. Wang and C. Wang, *ACS Appl. Mater. Interfaces*, 2017, **9**, 14665–14676.
- 23 N. Qi, B. Zhao, S.-D. Wang, S. S. Al-Deyab and K.-Q. Zhang, *RSC Adv.*, 2015, **5**, 50878–50882.
- 24 R. Melikov, D. A. Press, B. G. Kumar, I. B. Dogru, S. Sadeghi, M. Chirea, İ. Yilgör and S. Nizamoglu, *Sci. Rep.*, 2017, **7**, 7258.
- 25 V. Fernández-Luna, P. B. Coto and R. D. Costa, *Angew. Chem., Int. Ed.*, 2017, **57**, 8826–8836.
- 26 E. Fresta, V. F. Luna, P. B. Coto and R. D. Costa, *Adv. Funct. Mater.*, 2018, **28**, 1707011.
- 27 J. Casamayor, D. Su and Z. Ren, *Light. Res. Technol.*, 2018, **50**, 801–826.
- 28 M. Bergamos and R. Höltig, in *Recycling options for LEDs and LED products. Electronics Goes Green 2012+*, IEEE, Berlin, Germany, 2012.
- 29 I. Jaadane, P. Boulenguez, S. Carré, M. Savoldelli, C. Martinsons and A. Torriglia, *Free Radic. Biol. Med.*, 2015, **84**, 373–384.
- 30 J. Hye Oh, S. Ji Yang and Y. Rag Do, *Light Sci. Appl.*, 2014, **3**, e141.
- 31 D. Van Norren and T. G. M. F. Gorgels, *Photochem. Photobiol.*, 2011, **87**, 747–753.
- 32 European Commission, Scientific Committee, a) https://ec.europa.eu/health/scientific_committees/scheer/docs/scheer_q_002.pdf (accessed: January 5, 2019). b) https://ec.europa.eu/health/scientific_committees/emerging_en (accessed: January 5, 2019).
- 33 L. Niklaus, H. Dakhil, M. Kostrzewa, P. B. Coto, U. Sonnewald, A. Wierschem and R. D. Costa, *Mater. Horiz.*, 2016, **3**, 340–347.
- 34 M. D. Weber, L. Niklaus, M. Pröschel, P. B. Coto, U. Sonnewald and R. D. Costa, *Adv. Mater.*, 2015, **27**, 5493–5498.
- 35 L. Niklaus, S. Tansaz, H. Dakhil, K. T. Weber, M. Pröschel, M. Lang, M. Kostrzewa, P. B. Coto, R. Detsch, U. Sonnewald, A. Wierschem, A. R. Boccaccini and R. D. Costa, *Adv. Funct. Mater.*, 2017, **27**, 1601792.
- 36 C. F. Aguiño, M. Lang, V. Fernández-Luna, M. Pröschel, U. Sonnewald, P. B. Coto and R. D. Costa, *ACS Omega*, 2018, **3**, 15829–15836.
- 37 S. Nizamoglu, *SDÜ Fen Bilim. Enstitüsü Derg.*, 2016, **20**, 490–495.
- 38 D. A. Press, R. Melikov, D. Conkar, E. N. Firat-Karalar and S. Nizamoglu, *Adv. Photonics 2016 (IPR, NOMA, Sensors, Networks, SPPCom, SOF)*, 2016, NoTu2D.3.
- 39 V. Fernández-luna, D. S. Alcázar, J. P. Fernández-blázquez, A. L. Cortajarena, P. B. Coto and R. D. Costa, *Adv. Funct. Mater.*, 2019, **29**, 1904356.
- 40 A. Espasa, M. Lang, C. F. Aguiño, D. Sanchez-dealcazar, J. P. Fernández-blázquez, U. Sonnewald, A. L. Cortajarena, P. B. Coto and R. D. Costa, *Nat. Commun.*, 2020, **11**, 1–10.
- 41 X. Wang, Z. Li, W. Ying, D. Chen, P. Li, Z. Deng and X. Peng, *J. Mater. Chem. C*, 2020, **8**, 240–246.
- 42 X. Wang, Y. Guo, Z. Li, W. Ying, D. Chen, Z. Deng and X. Peng, *RSC Adv.*, 2019, **9**, 9777–9782.
- 43 M. S. P. Reddy and C. Park, *Sci. Rep.*, 2016, **6**, 32306.
- 44 K. Benson, A. Ghimire, A. Pattammattel and C. V. Kumar, *Adv. Funct. Mater.*, 2017, **1702955**, DOI:10.1002/adfm.201702955.
- 45 N. Hendler, B. Belgorodsky, E. D. Mentovich, M. Gozin and S. Richter, *Adv. Mater.*, 2011, **23**, 4261–4264.
- 46 I. Yuruker, S. U., Arik, M., Tamdogan, E., Melikov, R., Nizamoglu, S., Press, D. A., & Durak, in *Volume 3: Advanced fabrication and manufacturing; emerging technology frontiers; energy, health and water- applications of nano-, micro- and mini-scale devices; MEMS and NEMS; technology update talks; thermal management using micro channels, jets, sprays*, ASME, San Francisco, California, USA, 2015.
- 47 A. Aires, V. Fernández-luna, J. Fernandez-cestau, R. D. Costa and A. L. Cortajarena, *Nano Lett.*, 2020, DOI:10.1021/acs.nanolett.0c00324.
- 48 S. An, H. S. Jo, Y. Il Kim, K. Y. Song, M.-W. Kim, K. B. Lee, A. L. Yarin and S. S. Yoon, *Nanoscale*, 2017, **9**, 9139–9147.
- 49 S. Pratap, R. Mallem, K. Im, J. Lee, C. Park and P. chao, *Opti. Mater.*, 2019, **95**, 109270.
- 50 D. Zhou, H. Zou, M. Liu, K. Zhang, Y. Sheng, J. Cui, H. Zhang and B. Yang, *ACS Appl. Mater. Interfaces*, 2015, **7**, 15830–15839.
- 51 H. Tetsuka, A. Nagoya and R. Asahi, *J. Mater. Chem. C*, 2015, **3**, 3536–3541.
- 52 M. Sun, S. Qu, Z. Hao, W. Ji, P. Jing, H. Zhang, L. Zhang, J. Zhao and D. Shen, *Nanoscale*, 2014, **6**, 13076–13081.
- 53 J. Gotta, T. Ben Shalom, S. Aslanoglou, A. Cifuentes-Rius and N. H. Voelcker, *Adv. Funct. Mater.*, 2018, **28**, 1706967.
- 54 T. Iizuka, H. Sezutsu, K. Tatematsu, I. Kobayashi, N. Yonemura, K. Uchino, K. Nakajima, K. Kojima, C. Takabayashi, H. Machii, N. Kuwabara, T. Nakamura, K. Yoshii and T. Tamura, *Adv. Funct. Mater.*, 2013, **23**, 5232–5239.
- 55 J. Chao, D. Wook, O. Joo, S. Kim, C. Seok, J. Ren, H. Yoo, S. Yoon, J. Eun, Y. Ri and H. Kweon, *Biomaterials*, 2015, **70**, 48–56.
- 56 S. Putthanarat, R. K. Eby, R. R. Naik, S. B. Juhl, M. A. Walker, E. Peterman, S. Ristich, J. Magoshi, T. Tanaka, M. O. Stone, B. L. Farmer, C. Brewer and D. Ott, *Polymer*, 2004, **45**, 8451–8457.
- 57 B. Zhu, H. Wang, W. R. Leow, Y. Cai, X. J. Loh, M. Han and X. Chen, *Adv. Mater.*, 2016, **28**, 4250–4265.
- 58 L. Fernández-García, N. Marí-Buyé, J. A. Barios, R. Madurga, M. Elices, J. Pérez-Rigueiro, M. Ramos, G. V. Guinea and D. González-Nieto, *Acta Biomater.*, 2016, **45**, 262–275.
- 59 I. Chul, H. Yong, K. Gill and Y. Hwan, *Biol. Macromol.*, 2003, **33**, 203–213.
- 60 I. Chul, H. Kweon, Y. Hwan and S. Hudson, *Biol. Macromol.*,

- 2001, **29**, 91–97.
- 61 Y. Gotoh, M. Tsukada, T. Baba and N. Minoura, *Polymer*, 1997, **38**, 487–490.
- 62 H. Kweon and Y. H. Park, *J. Appl. Polym. Sci.*, 2001, **82**, 750–758.
- 63 H. Y. Kweon, I. C. Um and Y. H. Park, *Polymer*, 2000, **41**, 7361–7367.
- 64 H. Y. Kweon and Y. H. Park, *J. Appl. Polym. Sci.*, 1999, **73**, 2887–2894.
- 65 G. Freddi, P. Monti, M. Nagura, Y. Gotoh and M. Tsukada, *J. Polym. Sci. Part B Polym. Phys.*, 1997, **35**, 841–847.
- 66 M. Tsukada, G. Freddi, Y. Gotoh and N. J. Kasai, *Polym. Sci. B Polym. Phys.*
- 67 M. Nagura, S. Yamazaki and M. Tsukada, in *Proceedings of the 7th International Wool Textile Conference*, 1985, p. 345.
- 68 M. Tsukada, *J. Polym. Sci. Part B Polym. Phys.*, 1986, **24**, 457–460.
- 69 D. M. Ebenstein, J. Park, D. L. Kaplan and K. J. Wahl, *MRS Online Proc. Libr. Arch.*, 2004, **841**, 1–6.
- 70 N. V. Visser, J. W. Borst, M. A. Hink, A. Van Hoek and A. J. W. G. Visser, *Biophys. Chem.*, 2005, **116**, 207–212.
- 71 A. Ghisaidoobe and S. Chung, *Int. J. Mol. Sci.*, 2014, **15**, 22518–22538.
- 72 M. Cotlet, J. Hofkens, M. Maus, T. Gensch, M. Van der Auweraer, J. Michiels, G. Dirix, M. Van Guyse, J. Vanderleyden, A. J. W. G. Visser and F. C. De Schryver, *J. Phys. Chem. B*, 2002, **105**, 4999–5006.
- 73 T. Taki and M. Strassburg, *ECS J. Solid State Sc.*, 2020, **9**, 015017.
- 74 Y. Xue, S. Lofland and X. Hu, *Polymers*, 2019, **11**, 456.
- 75 G. Jung, S. Mais, A. Zumbusch and C. Bräuchle, *J. Phys. Chem. A*, 2000, **104**, 873–877.
- 76 W. Weber, V. Helms, J. A. McCammon and P. W. Langhoff, *Proc. Natl. Acad. Sci. USA*, 1999, **96**, 6177–6182.
- 77 A. M. Bogdanov, E. A. Bogdanova, D. M. Chudakov, T. V. Gorodnicheva, S. Lukyanov and K. A. Lukyanov, *Nat. Methods*, 2009, **6**, 859–860.
- 78 C. C. David, P. Dedecker, G. D. Cremer, N. Verstraeten, C. Kint, J. Michiels and J. Hofkens, *Photochem. Photobiol. Sci.*, 2012, **11**, 358–363.
- 79 R. Ando, H. Mizuno and A. Miyawaki, *Science*, 2004, **306**, 1370–1373.
- 80 N. Hendler, E. D. Mentovich, B. Belgorodsky, D. Rimmerman and S. Richter, *ChemPhysChem*, 2013, **14**, 4065–4068.
- 81 K. Benson, A. Ghimire, A. Pattammattel and C. V. Kumar, *Adv. Funct. Mater.*, 2017, **27**, 1702955.
- 82 C. Ezquerro, E. Fresta, E. Serrano, E. Lalinde, J. García-Martínez, J. R. Berenguer and R. D. Costa, *Mater. Horizons*, 2019, **6**, 130–136.
- 83 A. Sillen and Y. Engelborghs, *Photochem. Photobiol.*, 1998, **67**, 475–486.
- 84 W. C. Oliver and G. M. Pharr, *J. Mater. Res.*, 1992, **7**, 1564–1583.