

Supporting Information
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Table of Contents

Experimental Procedures	2
Materials	2
Instrumentation and measurements	2
Electrochemical measurements	2
O ₂ evolution experiments	3
Foot of the Wave Analysis (FOWA). <i>k</i> _{obs} Calculation	3
TON Calculation	3
Synthesis of the ligands	3
Synthesis of Cu-complexes	7
Synthesis of complex [(L4)Zn](Na) ₂ ·4H ₂ O	8
Spectroscopic Characterization	9
NMR Spectroscopy	9
Mass-spectrometry	31
UV-Vis spectroscopy	39
X-Ray Crystallography	40
Crystal preparation:	40
Electron Paramagnetic Resonance (EPR)	45
Electrochemical behavior in organic solvents	46
Electrochemical behavior in water	47
Determination of the kinetic constant. Foot of the Wave Analysis (FOWA)	53
O ₂ Evolution experiments	56
Scanning Electron Microscopy	58
Computational studies	60
Cartesian coordinates (Å) and calculated potential energies (atomic units)	71
References	91

SUPPORTING INFORMATION

Experimental Procedures

Materials

All general reagents and chemicals were used as purchased from Sigma-Aldrich, Fluka and Merck chemical companies without further purification unless otherwise stated. The ligand precursor ([2,2'-bipyridine]-6,6'-dicarboxylic acid) and 5-amino-*m*-xylo-4-sulfonic acid were prepared according to the experimental procedure reported in the literature.^{1,2} Air and moisture sensitive reactions were carried out under N₂ or Ar in oven-dried (120 °C) glassware. Evaporation of solvents *in vacuo* was done with a *Büchi Rotavapor R-200* at 40 °C.

Instrumentation and measurements

NMR spectra were measured on a *Bruker AV-500*, *Bruker AV-400*, and *Bruker 300 MHz* spectrometers. All NMR experiments were performed at room temperature in corresponding deuterated solvents and using solvent signals as reference. UV-Vis spectra were measured on a *Cary 50 UV-vis* spectrometer by *Varian Inc.* Electrospray ionization mass spectra (ESI-MS) were performed on an *Agilent Technologies 6130-Quadrupole LC/MS* connected to an *Agilent Technologies HPLC-1200 series*. Samples were dissolved in MeOH or H₂O and injected directly with an auto-sampler. HRMS samples were measured on a *Bruker HPLC-QqTOF Maxis* impact device. Elemental Analysis of the samples was carried out in a *Thermo Finnigan elemental analyzer Flash 1112* model. Electron Paramagnetic Resonance was performed in an *EMX Micro X-band EPR* spectrometer (*Bruker*) at 77 K using a liquid N₂ finger dewar. Data acquisition: perpendicular mode, modulation frequency of 100 KHz, microwave frequency of 9.38 GHz, modulation amplitude of 4 G, a 0.01 ms time constant and 4.12 ms conversion time with a microwave power of 0.556 mW.

Electrochemical measurements

Differential pulse voltammetry (DPV) and cyclic voltammetry (CV) were measured on a *CHI660D* potentiostat using a three-electrode cell. Glassy carbon (GC), Carbon paper (C-paper, *SGL Carbon*, *Sigracet 39 AA*) or boron-doped diamond (BDD) (d = 3 mm) working electrodes were employed while a Pt rod/mesh was used as counter electrode and a Hg/HgSO₄ (K₂SO₄ sat.) or Ag/AgCl (KCl sat.) electrode were used as a reference electrode. Working electrodes were polished with 1 and 0.05 micron alumina paste, washed with distilled water and acetone, and sonicated in acetone for 5 minutes before each measurement. DMF employed for electrochemical measurements was prepared containing the necessary amount of *n*-Bu₄NPF₆ (TBAPF₆) as supporting electrolyte to yield a solution of 0.1 M ionic strength. CVs were typically recorded at different scan rates from 25 to 1000 mV/s. DPVs were recorded with the following parameters: amplitude = 50 mV, step height = 4 mV, pulse width = 0.05 s. All redox potentials in the present work are reported versus NHE by adding 0.648 V to the measured potential.

SUPPORTING INFORMATION

O₂ evolution experiments

Controlled Potential Electrolysis (CPE) experiments were performed at different potentials and different pH values to catalyze the water oxidation reaction by the complexes by using a two-compartment cell closed with a septum. As working electrode large surface BDD electrodes (rectangular shape with 1.5 cm² surface) were used together with a silver/silver chloride (KCl sat.) as a reference electrode. These ones were placed in one of the compartments that was filled with a 1.5 mM solution of the complex (phosphate buffer pH 7, borate buffer pH 9 or phosphate buffer pH 11.6, of 0.1 M ionic strength). In the other compartment, containing only the buffer solution, a mesh platinum counter electrode was used.

The oxygen evolution was monitored with an OXNP type Clark electrode in gas phase (from Unisense Company). The CPE was carried out using an IJ-Cambria CHI-660 potentiostat and was started as soon as the oxygen sensor signal was stable under air atmosphere. During the experiment, solutions of both compartments were vigorously stirred. Calibration of the oxygen sensor was performed after each experiment by adding known amounts of pure oxygen into the cell using a Hamilton syringe. The results of the water oxidation catalysis with the complexes were compared with blank experiments under the same conditions but in the absence of the complex. The Faradaic efficiency was determined according to the total charge passed during the CPE and the total amount of generated oxygen by considering that water oxidation is a 4e⁻ oxidation process.

Foot of the Wave Analysis (FOWA). k_{obs} Calculation

Under catalytic conditions FOWA equation is operative.

$$\frac{i}{i_p} = \frac{n \cdot 2.24 \cdot \sqrt{\frac{R \cdot T}{F \cdot v}} \cdot k_{obs}}{1 + \exp\left[\frac{F}{RT}(E^o_{cat} - E)\right]} \quad \text{Equation (1)}$$

where k_{obs} is the apparent WNA pseudo-rate constant ($k[\text{H}_2\text{O}]$), E^o_{cat} corresponds to the standard potential for the catalytic wave (E^o_{cat} according to the DPVs shown in Figure S46), i is the current in the presence of substrate, i_p corresponds to the peak current of one-electron redox process of the catalyst (extracted from the Cu^{II}/Cu^I couple when available), F is the faradaic constant (96485 C mol⁻¹), T is the temperature (298 K), v is the scan rate (100 mV s⁻¹) and R is 8.314 J mol⁻¹ K⁻¹.³

TON Calculation

The total TON values can be obtained from the oxygen evolution experiment taking in account the total amount of catalyst present in the solution by using equation (2). However, since only the catalyst present in the layer of the solution in contact with the electrode is involved in the water oxidation reaction, this TON value is underestimated. Lin and co-workers adapted it to the formula (3) based on the previous methodology developed by Savéant and co-workers, which gives a more realistic TON value based on the amount catalyst in contact with the electrode.⁴

$$\text{TON} = \frac{\mu\text{mol produced O}_2}{\mu\text{mol cat.}} \quad \text{Equation (2)}$$

$$\text{TON} = \frac{k_{obs} t}{1 + \exp\left[\frac{F}{RT}(E^o_{cat} - E)\right]} \quad \text{Equation (3)}$$

Synthesis of the ligands

The general procedure for the synthesis of ligands, [H₂L1], [H₂L3], [H₂L4]²⁻ and [H₂L5]²⁻ was adapted from the literature⁵ as follows: 500 mg (2.05 mmol) of [2,2'-bipyridine]-6,6'-dicarboxylic acid were suspended in 20 mL of SOCl₂ and the mixture was refluxed at 85 °C under a nitrogen atmosphere during 6 hours. After complete dissolution of the reactant, SOCl₂ was completely removed under vacuum, yielding a white powder corresponding to the acyl chloride derivative. The white solid was re-suspended in 40 mL of dry DCM and the temperature was decreased until 0 °C using an ice bath. Then, 4 eq. of NEt₃ were added dropwise and stirred for 10 minutes. Finally, a previously prepared dispersion of the corresponding phenylamine or naphthylamine (4.1 mmol, 2.0 eq.) in 40 mL of dry DCM were added dropwise to the reaction volume and the mixture was vigorously stirred for 72 h at room temperature. The appearing solid consisted in the corresponding ligand ([H₂L1], [H₂L3], [H₂L4]²⁻ and [H₂L5]²⁻), which was filtered and washed with DCM and Et₂O, yielding the desired product without further purification.

In case of ligands [H₂L2], [H₂L6]²⁻, [H₂L7]²⁻ and [H₂L8]²⁻, the general procedure was adapted as follows. The acyl chloride derivative was re-dissolved in 20 mL of dry DMA and added dropwise to a dispersion of the corresponding amine (4 mmol, 2 eq.) in 8 mL of dry DMA containing 4 eq. of Et₃N. This solution was stirred under a nitrogen atmosphere for 15 h at 50 °C. The appearing solid consisted in the corresponding ligand ([H₂L6]²⁻, [H₂L7]²⁻ and [H₂L8]²⁻) which was filtered and washed with acetone and Et₂O, yielding the desired product

SUPPORTING INFORMATION

without further purification. In case of [H₂L2], the reaction mixture showed no precipitate after 4 days of reaction. The clean solution was mixed with 0.1 M Na₂CO₃ solution, extracted with DCM, dried over MgSO₄, and evaporated to yield [H₂L2] as a white powder.

[H₂L1]·0.8 H₂O

Yield: 695 mg, 1.75 mmol, 86 %. ¹H NMR (DMSO-d₆): δ [ppm] = 10.66 (H₄, s, 2H), 9.22 (H₁, dd, *J* = 6.9, 2.1 Hz, 2H), 8.29 (H₂₋₃, m, 4H), 7.93 (H₅, dd, *J* = 8.71, 1.2 Hz, 4H), 7.4 (H₆, dd, *J* = 8.7 Hz, 4H), 7.16 (H₇, tt, *J* = 7.4, 1.2 Hz, 2H). ¹³C NMR (DMSO-d₆): 162.9 (C₆), 153.9 (C₅), 150.2 (C₁), 139.7 (C₄), 138.6 (C₇), 129.1 (C₉), 125.3 (C₂), 124.7 (C₁₀), 123.6 (C₃), 121.6 (C₈). ESI-MS (MeOH) *m/z* positive mode: 417.0 [H₂L1+Na]⁺. Elemental analysis (% found): C, 70.36; H, 4.86; N, 13.79. Calcd. for C₂₄H₁₈N₄O₂·0.8 H₂O: C, 70.51; H, 4.83; N, 13.70. IR ($\tilde{\nu}$ / cm⁻¹): 3421m, 3347s, 1663m, 1598m, 1530s, 1498w, 1448m, 14312m, 1325m, 1244w, 1077m, 994w, 756m, 687m, 664m, 494w.

[H₂L2]·0.3 H₂O

Yield: 453 mg, 1.00 mmol, 49%. ¹H NMR (500 MHz, CDCl₃): δ [ppm] = 9.92 (H₄, s, 2H), 8.62 (H₁, dd, *J* = 7.9, 1.0 Hz, 2H), 8.42 (H₃, d, *J* = 7.7, 1.0 Hz, 2H), 8.16 (H₂, t, *J* = 7.8 Hz, 2H), 7.46 (H₅, s, 4H), 6.84 (H₇, s, 2H), 2.38 (H₆, s, 12H). ¹³C NMR (126 MHz, CDCl₃): δ [ppm] = 161.7 (C₆), 153.7 (C₁), 149.9 (C₅), 139.1 (C₃, C₉), 137.5 (C₇), 126.6 (C₁₁), 123.9 (C₂), 123.2 (C₄), 117.8 (C₈), 21.6 (C₁₀). ESI-MS (CHCl₃/MeOH) *m/z* positive mode: calcd for [H₂L2+H]⁺, (C₂₈H₂₅N₄O₂): 449.1983, found: 449.1981. Elemental analysis (% found): C, 73.39; H, 5.54; N, 12.55. Calcd for C₂₈H₂₆N₄O₂·0.3 H₂O: C: 73.76; H: 5.88; N: 12.29. IR ($\tilde{\nu}$ / cm⁻¹): 3352s, 3095w, 3015w, 2914m, 2856w, 1683s, 1611m, 1581m, 1553s, 1417s, 1237w, 1178w, 1075w, 996w, 834m, 758m, 685m, 633m.

[H₂L3]·0.5 H₂O

Yield: 1.40 g, 1.42 mmol, 71%. ¹H NMR (400MHz, CDCl₃): δ [ppm] = 10.74 (H₄, s, 2H), 8.78 (H₁, dd, *J* = 7.9, 1.0 Hz, 2H), 8.52 (H₃, dd, *J* = 7.7, 1.0 Hz, 2H), 8.40 (d, *J* = 6.8 Hz, 2H), 8.23 (H₂, t, *J* = 7.8 Hz, 2H), 8.13 (d, *J* = 8.5 Hz, 2H), 7.95 (d, *J* = 8.7 Hz, 2H), 7.77 (d, *J* = 8.2 Hz, 2H), 7.66 – 7.53 (H_{6,9,10}, m, 6H). HRMS-ESI: Calcd for [H₂L3+Na]⁺, (C₃₂H₂₂N₄O₂Na): 517.1635, found: 517.1625. Elemental analysis (% found): C, 76.2; H, 4.5; N, 11.1. Calcd for C₃₂H₂₂N₄O₂·0.5 H₂O: C, 76.3; H, 4.6; N, 11.1. IR ($\tilde{\nu}$ / cm⁻¹): 3370s, 3356s, 3055m, 3012w, 1685s, 1584m, 1536s, 1496s, 1430m, 1405m, 1344m, 1255m, 1133m, 1068m, 996w, 792m, 770m, 735m, 631m.

[H₂L4](Et₃NH)₂·2 H₂O

Yield: 931 mg, 1.68 mmol, 61%. ¹H NMR (DMSO-d₆): δ [ppm] = 10.70 (H₄, s, 2H), 9.24 (H₁, dd, *J* = 7.3, 1.6 Hz, 2H), 8.30 (H₂₋₃, m, 4H), 7.90 (H₅, d, *J* = 8.7 Hz, 4H), 7.65 (H₆, d, *J* = 8.7 Hz, 4H). ¹³C NMR (DMSO-d₆): 162.9 (C₆), 153.9 (C₅), 150.0 (C₁), 144.7 (C₁₀), 139.8 (C₄), 138.6 (C₇), 126.5 (C₉), 125.3 (C₂), 123.7 (C₃), 120.4 (C₈). ESI-MS (MeOH) *m/z* negative mode: 553.0 [H₂L4+H]⁺, 276.1 [H₂L4]²⁻. Elemental analysis (% found): C, 54.6; H, 6.7; N, 10.6. Calcd for C₂₄H₁₆N₄O₈S₂²⁻·2 Et₃NH⁺·2 H₂O: C, 54.5; H, 6.6; N, 10.6. IR ($\tilde{\nu}$ / cm⁻¹): 3333m, 3316m, 3055w, 2933w, 1690s, 1583m, 1522s, 1434m, 1398m, 1275m, 1230w, 1166m, 1126m, 1073m, 893w, 826m, 681m, 548m.

[H₂L5](Et₃NH)₂·1.3 H₂O

Yield: 965 mg, 1.74 mmol, 64%. ¹H NMR (DMSO-d₆): δ [ppm] = 12.55 (H₄, s, 2H), 9.51 (H₁, dd, *J* = 7.3, 1.7 Hz, 2H), 8.67 (H₅, dd, *J* = 8.2, 1.1 Hz, 2H), 8.26 (H₂₋₃, m, 4H), 7.82 (H₈, dd, *J* = 7.7, 1.7 Hz, 2H), 7.43 (H₆, t, *J* = 1.2 Hz, 2H), 7.14 (H₇, td, *J* = 7.5, 1.2 Hz, 2H). ¹³C NMR (DMSO-d₆): 162.4 (C₆), 153.9 (C₅), 149.6 (C₁), 139.9 (C₄), 136.8 (C₁₂), 135.3 (C₇), 130.1 (C₉), 127.6 (C₈), 125.4 (C₂), 123.2 (C₁₋₂), 120.2 (C₃₋₇). ESI-MS (MeOH) *m/z* negative mode: 553.0 [H₂L5+H]⁺, 276.1 [H₂L5]²⁻. Elemental analysis (% found): C, 55.3; H, 6.0; N, 10.6. Calcd for C₂₄H₁₆N₄O₈S₂²⁻·2 Et₃NH⁺·1.3 H₂O: C, 55.4; H, 6.5; N, 10.8. IR ($\tilde{\nu}$ / cm⁻¹): 3422m, 3344m, 3276m, 3001m, 2702m, 2502w, 1682m, 1580s, 1524s, 1425s, 1398w, 1245s, 1222m, 1165s, 1149m, 1124m, 1008m, 904w, 785w, 613m.

[H₂L6](Et₃NH)₂

Yield: 1.29 g, 1.48 mmol, 74 %. ¹H NMR (DMSO-d₆): δ [ppm] = 10.48 (H₄, s, 2H), 9.21 (H₁, d, *J* = 7.9, 1.0 Hz, 2H), 8.87 ((CH₃CH₂)₃NH, s, 2H), 8.25-8.32 (H₂₋₃, m, 4H), 7.56 (H₅, s, 4H), 3.09 ((CH₃CH₂)₃NH, q, *J* = 6.9 Hz, 12H), 2.60 (H₆, s, 12H), 1.17 ((CH₃CH₂)₃NH, tr, *J* = 7.2 Hz, 18H). ¹³C NMR (DMSO-d₆): 162.4 (C₆), 153.9 (C₅), 149.6 (C₁), 139.9 (C₄), 136.8 (C₁₂), 135.3 (C₇), 130.1 (C₉), 127.6 (C₈), 125.4 (C₂), 123.2 (C₁₋₂), 120.2 (C₃₋₇), 62.0 ((CH₃CH₂)₃NH), 25.5 ((CH₃CH₂)₃NH). HRMS-ESI(MeOH): calcd for [H₂L6]²⁻, (C₂₈H₂₄N₄O₈S₂): 304.0523, found 304.0535. Elemental analysis (% found): C, 57.8; H, 6.3; N, 10.2. Calcd for C₂₄H₂₆N₄O₈S₂²⁻·2Et₃NH⁺: C, 58.1; H, 7.0; N, 10.2. IR ($\tilde{\nu}$ / cm⁻¹): 3441br, 3330w, 3084w, 2984m, 1681s, 1644m, 1581m, 1530s, 1436m, 1399m, 1169s, 1082s, 1011s, 681m.

SUPPORTING INFORMATION

[H₂L7](Et₃NH)₂·2 H₂O

Yield: 500 mg, 0.58 mmol, 29 %. ¹H NMR (400 MHz, DMSO-d₆): δ [ppm] = 11.13 (H₄, s, 2H), 9.37 (H₁, dd, *J* = 7.1, 1.8 Hz, 2H), 8.95 (H₇, m, 2H, *H*12), 8.30 (H_{2,3}, m, 4H), 8.05 (H_{6,10}, m, 4H), 7.72 (H₅, d, *J* = 7.7 Hz, 2H), 7.59 (H_{8,9}, m, 4H, *H*13, *H*14), 3.09 ((CH₃CH₂)₃NH, q, 7.3 Hz, 12H), 1.17 ((CH₃CH₂)₃NH, tr, *J* = 7.2 Hz, 18H). ¹³C NMR (101 MHz, DMSO-d₆): δ [ppm] = 163.3 (C₆), 153.7 (C₁), 149.5 (C₅), 142.4 (C₁₀), 139.3 (C₃), 134.5 (C₁₁), 129.8 (C₇), 129.3 (C₁₆), 128.1 (C₁₂), 125.8 (C₁₃₊₁₄), 124.8 (C₂), 124.2 (C₉), 123.2 (C₄), 122.8 (C₁₅), 121.8 (C₈), 62.0 ((CH₃CH₂)₃NH), 25.5 ((CH₃CH₂)₃NH). HRMS-ESI (MeOH): Calcd. for [H₂L7]²⁺, (C₃₂H₂₂N₄O₈S₂): 326.0367, found 326.0373. Elemental analysis (% found): C, 59.25; H, 6.72; N, 9.47. Calcd for C₃₂H₂₂N₄O₈S₂·2Et₃NH⁺·2H₂O: C, 59.18; H, 6.32; N, 9.41. IR ($\tilde{\nu}$ / cm⁻¹): 3398s, 3206s, 3088s, 1685s, 1610m, 1532s, 1508s, 1438w, 1147s, 1041s, 750m, 681m.

[H₂L8](Et₃NH)₂·1.5H₂O

Yield: 774 mg, 1.81 mmol, 88%. ¹H NMR (400 MHz, DMSO-d₆): δ [ppm] = 10.88 (H₄, s, 2H), 9.29 (H₁, dd, *J* = 6.3, 2.7 Hz, 2H), 8.54 (H₁₀, s, 2H), 8.40 – 8.29 (H_{2,3}, m, 4H), 8.14 (H₇, s, 2H), 8.04 (H_{5,6}, m, 4H), 7.86 (H₉, d, *J* = 8.7 Hz, 2H), 7.72 (H₈, dd, *J* = 8.5, 1.5 Hz, 2H), 3.09 ((CH₃CH₂)₃NH, m, 12H), 1.17 ((CH₃CH₂)₃NH, tr, *J* = 7.2 Hz, 18H). ¹³C NMR (101 MHz, DMSO) δ 162.7 (C₆), 153.5 (C₁), 149.6 (C₅), 144.7 (C₁₂), 139.4 (C₃), 136.3 (C₇), 133.1 (C₁₅), 129.3 (C₁₀), 128.9 (C₉), 127.0 (C₁₄), 124.9 (C₂), 124.5 (C₁₃), 123.9 (C₁₁), 123.2 (C₄), 121.9 (C₈), 117.3 (C₁₆), 62.0 ((CH₃CH₂)₃NH), 25.5 ((CH₃CH₂)₃NH). HRMS-ESI (MeOH): calcd for [H₂L8]²⁺, (C₃₂H₂₂N₄O₈S₂): 326.0367, found 326.0362. Elemental analysis (% found): C, 59.6; H, 5.9; N, 9.5. Calcd. for C₃₂H₂₂N₄O₈S₂·2Et₃NH⁺·1.5H₂O: C, 59.8; H, 6.3; N, 9.5. IR ($\tilde{\nu}$ / cm⁻¹): 3310s, 3021s, 2860s, 2776s, 2501m, 1682s, 1580m, 1536s, 1493m, 1438m, 1393m, 1340w, 1241s, 1159s, 1090s, 1019s, 823m, 670m.

SUPPORTING INFORMATION

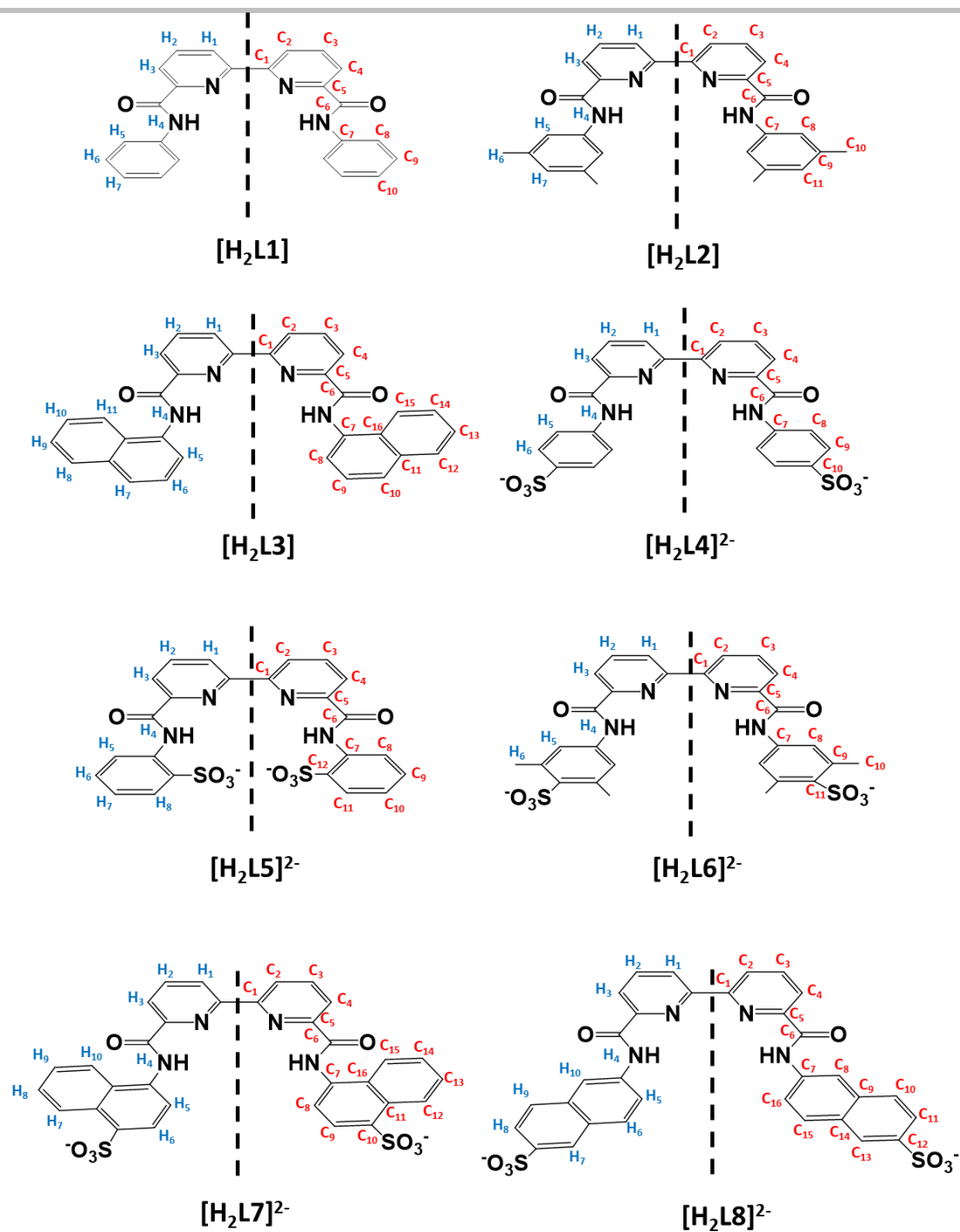


Figure S1. Schematic drawing of the ligands synthesized in this work and their ¹H NMR and ¹³C NMR assignments.

SUPPORTING INFORMATION

Synthesis of Cu-complexes

The general procedure for the synthesis of the copper complexes was adapted from a previous published work.⁶ Typically, 0.2 mmol of the corresponding ligand were suspended in 8 mL of MeOH and stirred during 15 minutes. Afterwards, 7.2 mL (4 eq.) of 0.1 M NaOH_{aq.} were added to the reaction mixture and vigorously stirred during 30 minutes at room temperature. After complete dissolution of the ligand, 0.2 mmol of copper perchlorate hexahydrate dissolved in 8 mL of MeOH were added dropwise to the mixture, which were allowed to react overnight (16 h) at room temperature. Then, the reaction mixture was filtrated and MeOH was evaporated under vacuum, and the remaining solution was diffused with the corresponding solvent (further details about crystallization are explained in Table S1), yielding the corresponding pure copper complexes.

[(L1)Cu]·1.5 H₂O·MeOH

Yield: 38 mg, 0.08 mmol, 42% ESI-MS (MeOH) m/z positive mode: 456.0 [(L1)Cu+H]⁺, 494.1 [(L1)Cu+K]⁺. Elemental analysis (% found): C, 54.6; H, 3.2; N, 10.5. Calcd. for C₂₄H₁₆CuN₄O₂·1.5 H₂O·MeOH: C, 54.8; H, 3.3; N, 10.6. IR ($\tilde{\nu}$ / cm⁻¹): 3396s, 3055w, 2922w, 2837w, 1625m, 1614m, 1577s, 1487m, 1468m, 1354m, 1337m, 1283m, 1142m, 1028m, 747m, 689m, 485w.

[(L2)Cu]·2 H₂O·MeOH

Yield: 41 mg, 0.08 mmol, 42% ESI-MS (MeOH) m/z positive mode: 512.1 [(L2)Cu+Na]⁺, 1023.0 [2(L2)Cu+H]⁺. HRMS-ESI(CHCl₃/MeOH): calcd for [(L3)Cu+H]⁺, (C₂₈H₂₅N₄O₂Cu) : 512.1259, found 512.1268. Elemental analysis (% found): C, 58.5; H, 6.1; N, 9.5. Calcd for C₂₈H₂₄CuN₄NO₂·2 H₂O·MeOH: C, 58.2; H, 5.7; N, 9.4. IR ($\tilde{\nu}$ / cm⁻¹): 3355m, 3031w, 2999w, 2911m, 2860w, 1623m, 1570s, 169m, 1353s, 1262m, 1076m, 1027m, 826m, 767m, 688m, 579w.

[(L3)Cu]·1.5MeOH

Yield: 45 mg, 0.08 mmol, 40% ESI-MS (MeOH) m/z positive mode: 556.1 [(L3)Cu+H]⁺, 578.1 [(L3)Cu+Na]⁺ 1111.1 [2·(L3)Cu+H]⁺. HRMS-ESI(+ve): calcd for [(L3)Cu+H]⁺, (C₃₂H₂₁CuN₄O₂): 556.0955, found: 556.0948. Elemental analysis (% found): C, 66.3; H, 3.9; N, 9.6. Calcd for C₃₂H₂₀CuN₄Na₂O₈S₂·1.5 MeOH: C, 66.6; H, 4.3; N, 9.3. IR ($\tilde{\nu}$ / cm⁻¹): 3655w, 3381m, 3192m, 3051m, 1624m, 1578s, 1565s, 1504m, 1472w, 1392m, 1338m, 1285m, 1037m, 790m, 763m, 705w.

[(L4)Cu]Na₂·3H₂O

Yield: 19 mg, 0.03 mmol, 32% ESI-MS (MeOH) m/z negative mode: 636.0 [(L4)Cu+Na]⁺, 614.0 [(L4)Cu+H]⁻. Elemental analysis (% found): C, 40.2; H, 2.5; N, 7.7. Calcd for C₂₄H₁₄CuN₄Na₂O₈S₂·3 H₂O: C, 40.4; H, 2.8; N, 7.8. IR ($\tilde{\nu}$ / cm⁻¹): 3585s, 3513s, 3460s, 3382m, 3227m, 1621m, 1597s, 1574m, 1497m, 1474m, 1358m, 1199m, 1120s, 1032s, 1007m, 820w, 693m, 570m.

[(L5)Cu]Na₂·4 H₂O·2 MeOH

Yield: 15 mg, 0.02 mmol, 25%. ESI-MS (MeOH) m/z negative mode: 636.0 [(L5)Cu+Na]⁺, 614.0 [(L5)Cu+H]⁻. Elemental analysis (% found): C, 39.2; H, 3.9; N, 6.5. Calcd for C₂₄H₁₄CuN₄Na₂O₈S₂·4 H₂O·2 MeOH: C, 39.2; H, 3.8; N, 7.0. IR ($\tilde{\nu}$ / cm⁻¹): 3585s, 3513s, 3459m, 3417s, 3331m, 3226m, 3097w, 1620w, 1593m, 1574s, 1472m, 1388m, 1293w, 1177m, 1084s, 1013m, 761w, 161m, 569m.

[(L6)Cu]Na₂·2.4H₂O

Yield: 42 mg, 0.06 mmol, 72% ESI-MS (MeOH) m/z negative mode: 609.0 [(L6)Cu+H]⁻. Elemental analysis (% found): C, 44.4; H, 3.9; N, 7.4. Calcd for C₂₈H₂₂CuN₄Na₂O₈S₂·2.4 H₂O: C, 44.3; H, 3.6; N, 7.4. IR ($\tilde{\nu}$ / cm⁻¹): 3404s, 3076w, 2974w, 2937w, 1612w, 1591m, 1557s, 1472m, 1368m, 1305w, 1170s, 1084s, 1016m, 765m, 690m, 648m.

[(L7)Cu]Na₂·7H₂O

Yield: 59 mg, 0.08 mmol, 49% ESI-MS (CHCl₃/MeOH) m/z negative mode: 713.8 [(L7)Cu+H]⁻, 735.8 [(L7)Cu+Na]⁺. HRMS-ESI (CHCl₃/MeOH): calcd for [(L7)Cu]²⁻, (C₃₂H₁₈CuN₄O₈S₂) : 356.4937, found 356.4933. Elemental analysis (% found): C, 43.4; H, 3.2; N, 6.2. Calcd for C₃₂H₁₈CuN₄Na₂O₈S₂·7 H₂O: C, 43.4; H, 3.6; N, 6.3. IR ($\tilde{\nu}$ / cm⁻¹): 3626m, 3358s, 3078w, 1614w, 1575m, 1552s, 1506m, 1420w, 1390s, 176m, 1285w, 1214w, 1182s, 1156m, 1045s, 1156m, 1045s, 1023m, 759m, 688m, 612w.

SUPPORTING INFORMATION

[(L8)Cu]Na₂·6H₂O

Yield: 12 mg, 16 mmol, 20% ESI-MS (MeOH) m/z negative mode: 735.8 [(L8)Cu+Na⁺], 713.8 [(L8)Cu+H⁺]. HRMS-ESI(-ve, MeOH): calcd for [(L8)Cu+Na⁺], (C₃₂H₁₈CuN₄O₈S₂Na): 735.9765, found 735.9763. Elemental analysis (% found): C, 44.2; H, 3.3; N, 6.5. Calcd for C₃₂H₁₈CuN₄Na₂O₈S₂·6 H₂O: C, 44.3; H, 3.5; N, 6.4. IR ($\tilde{\nu}$ / cm⁻¹): 3427s, 3081m, 1673w, 1626m, 1577s, 1530m, 1494m, 1470m, 1377m, 1181s, 1096s, 1035s, 819m, 674m, 623m.

Synthesis of complex [(L4)Zn](Na)₂·4H₂O.

The procedure for the synthesis of the zinc complex was similar to the synthesis of the copper complexes. To a suspension of [H₂L4]²⁻ (200 mg (0.36 mmol) in 16 mL of MeOH), 14.4 mL (4 eq.) of 0.1 M NaOH_{aq.} were added to the reaction mixture and vigorously stirred during 30 minutes at room temperature. After complete dissolution of the ligand, 131 mg (0.36 mmol) of zinc trifluoromethanesulfonate dissolved in 4 mL of MeOH was added dropwise to the mixture, which was allowed to react overnight (16 h) at room temperature. Afterwards, a white precipitate appeared, which was filtrated and washed with H₂O (2 x 2 mL) and dried over vacuum yielding [(L4)Zn]Na₂·4 H₂O as a white solid.

[(L4)Zn]Na₂·4 H₂O

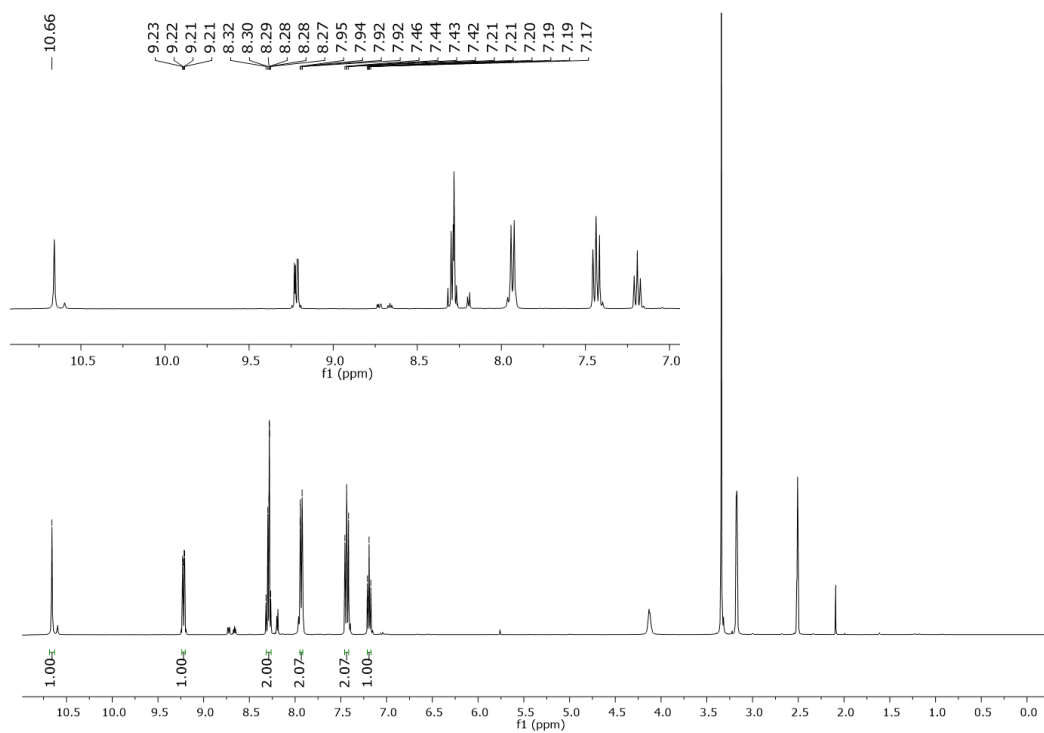
Yield: 193 mg, 0.29 mmol, 81%. ¹H-NMR (D₂O-d₂): δ [ppm] = 8.04 (H_{1,2}, m, 4H), 7.67 (H₃, dd, J= 7.5, 1.3 Hz, 2H), 7.20 (H₄, d, = 8.5 Hz, 4H), 6.36 (H₅, d, J= 8.5 Hz, 4H). Elemental analysis (% found): C, 39.8; H, 3.4; N, 7.6. Calcd for C₂₄H₁₄CuN₄Na₂O₈S₂Zn·4 H₂O: C, 39.3; H, 3.0; N, 7.6.

SUPPORTING INFORMATION

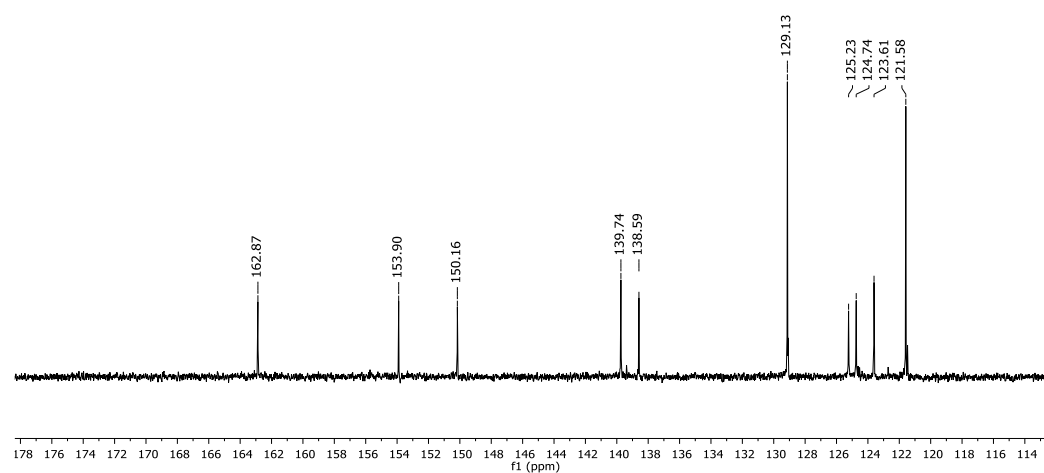
Spectroscopic Characterization

NMR Spectroscopy

A

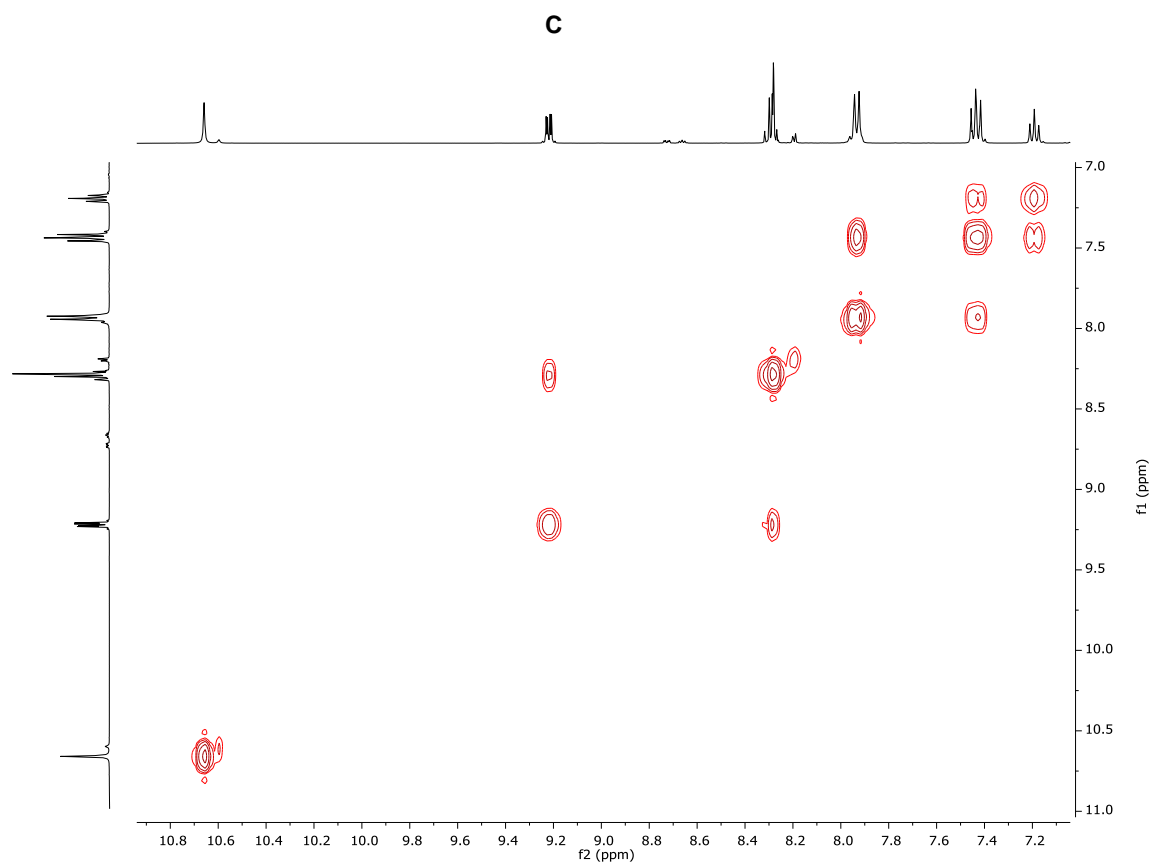


B

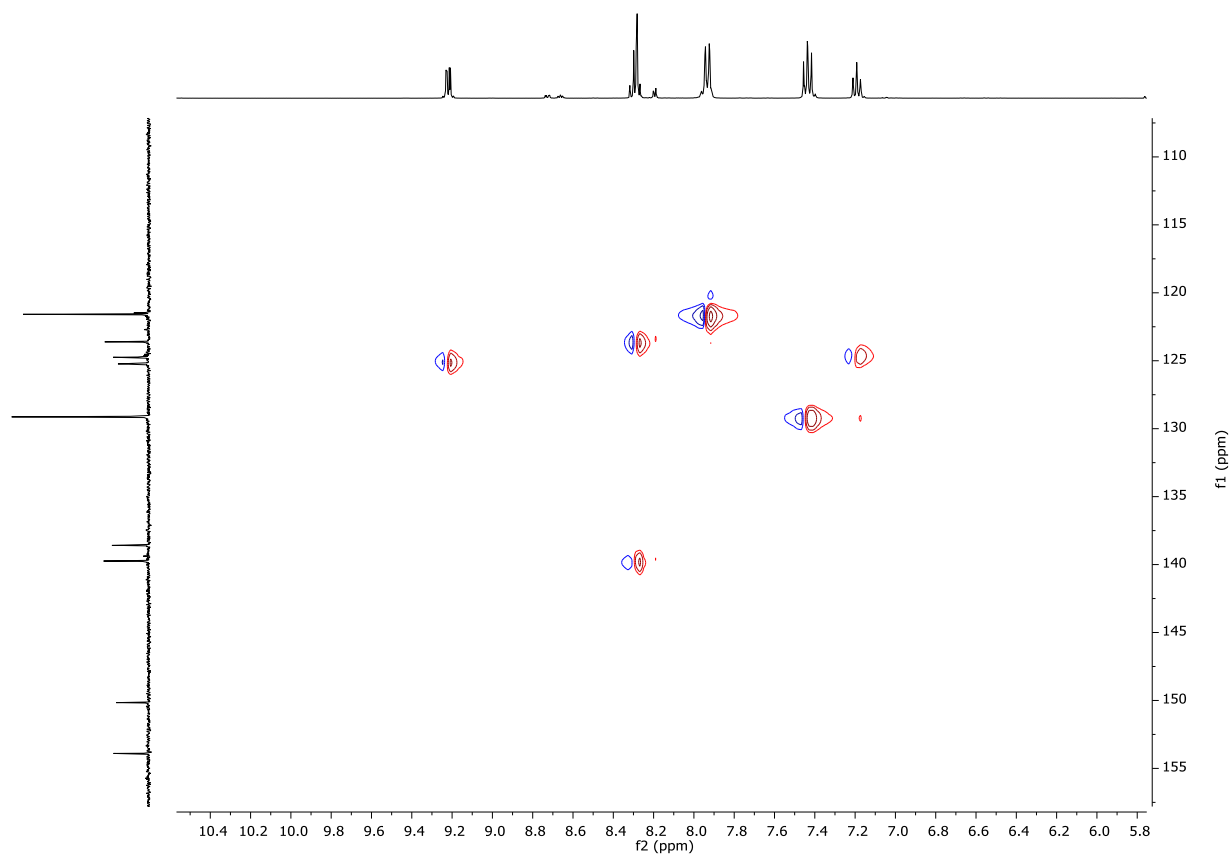


SUPPORTING INFORMATION

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SUPPORTING INFORMATION

E

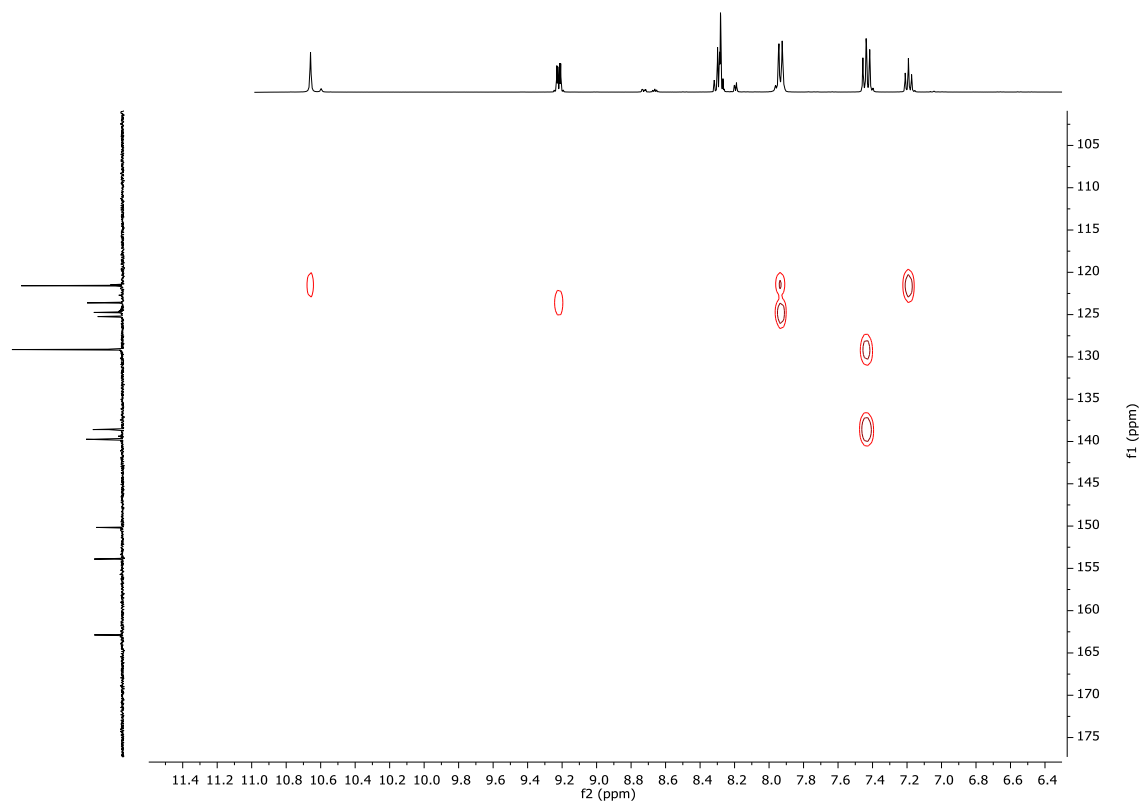
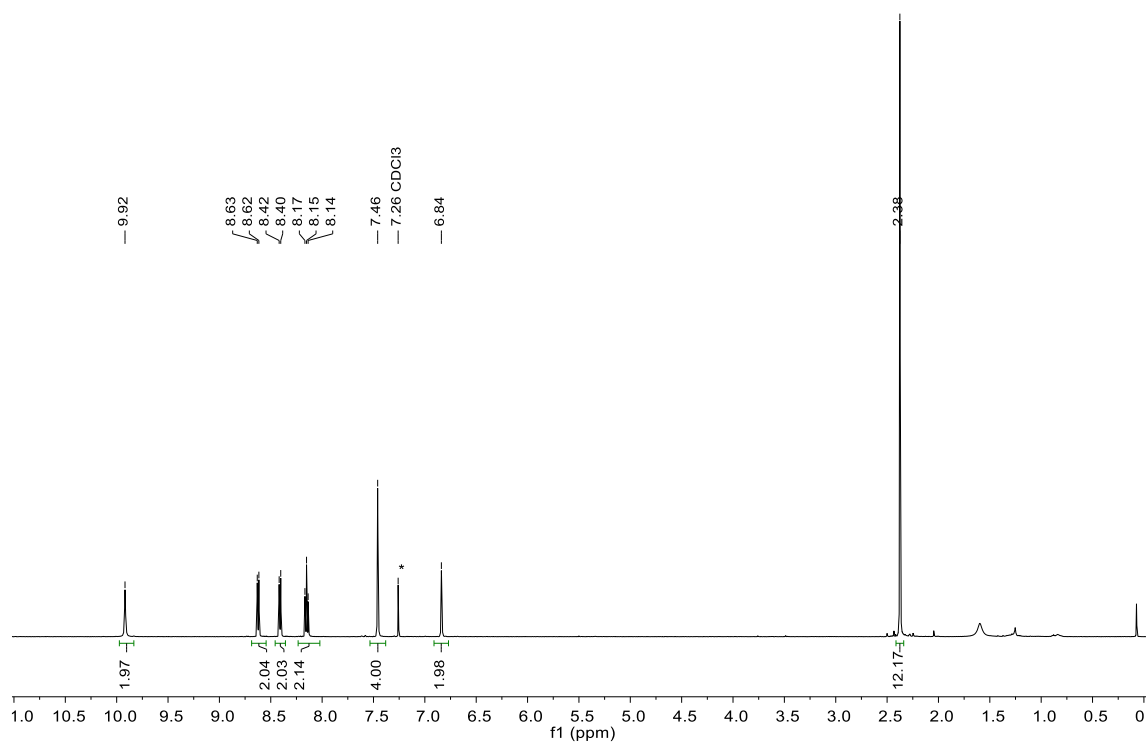


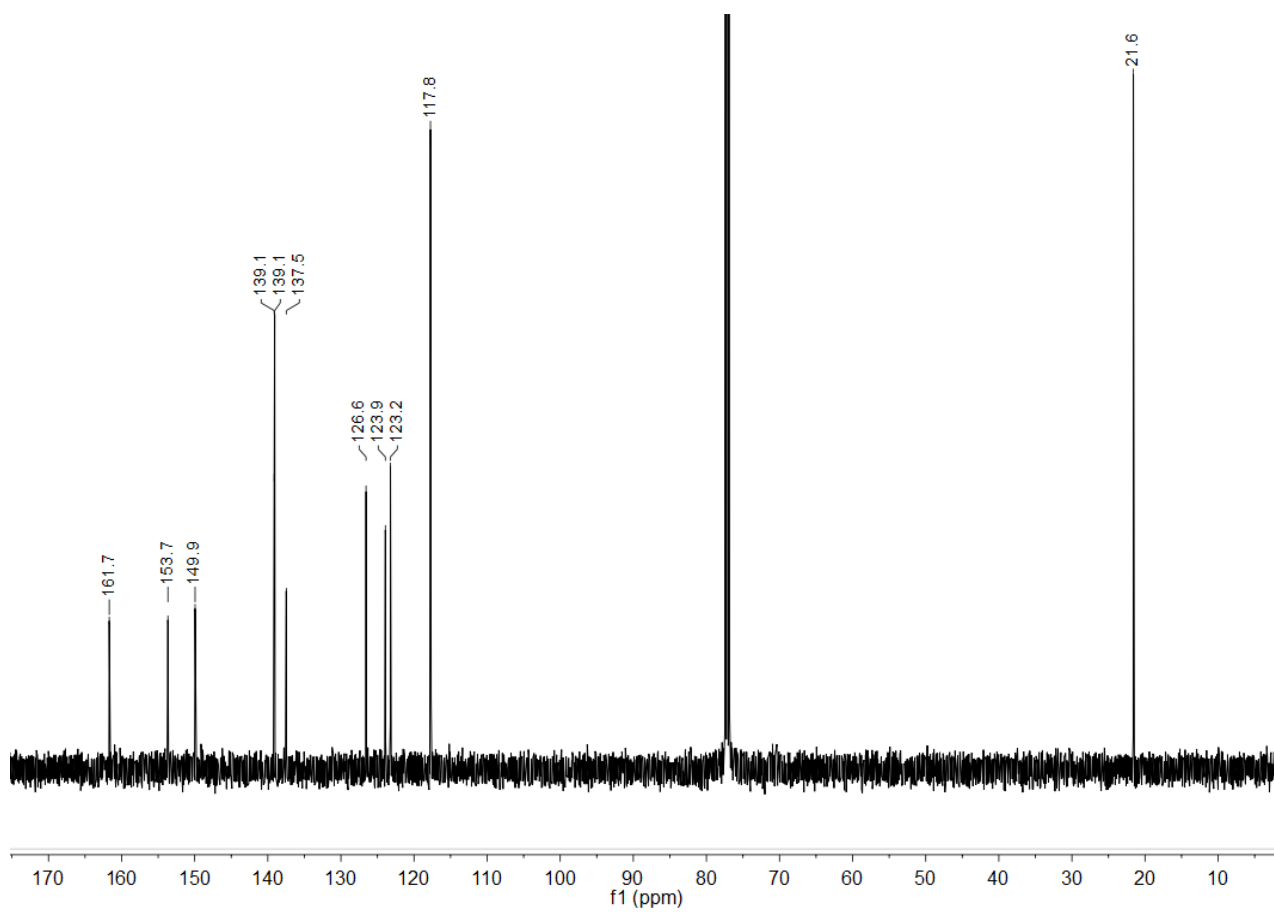
Figure S2. NMR spectra (DMSO-*d*₆, 500 MHz, 25 °C) for H₂L1. (A) ¹H-NMR, (B) ¹³C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A

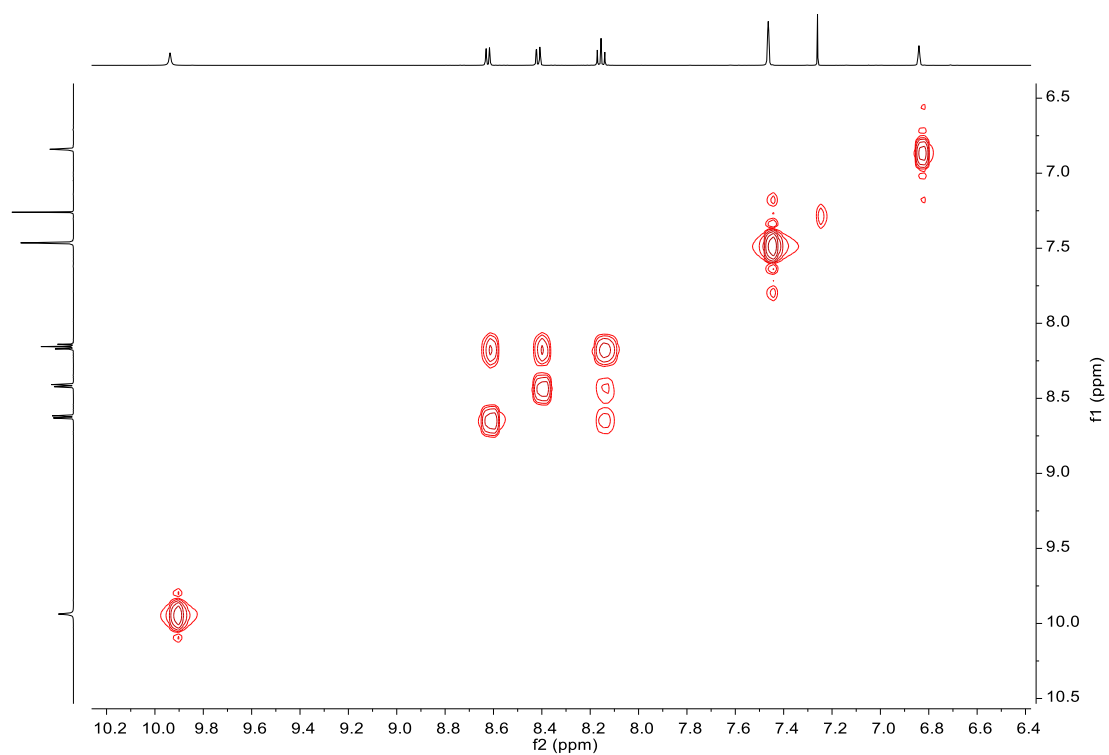


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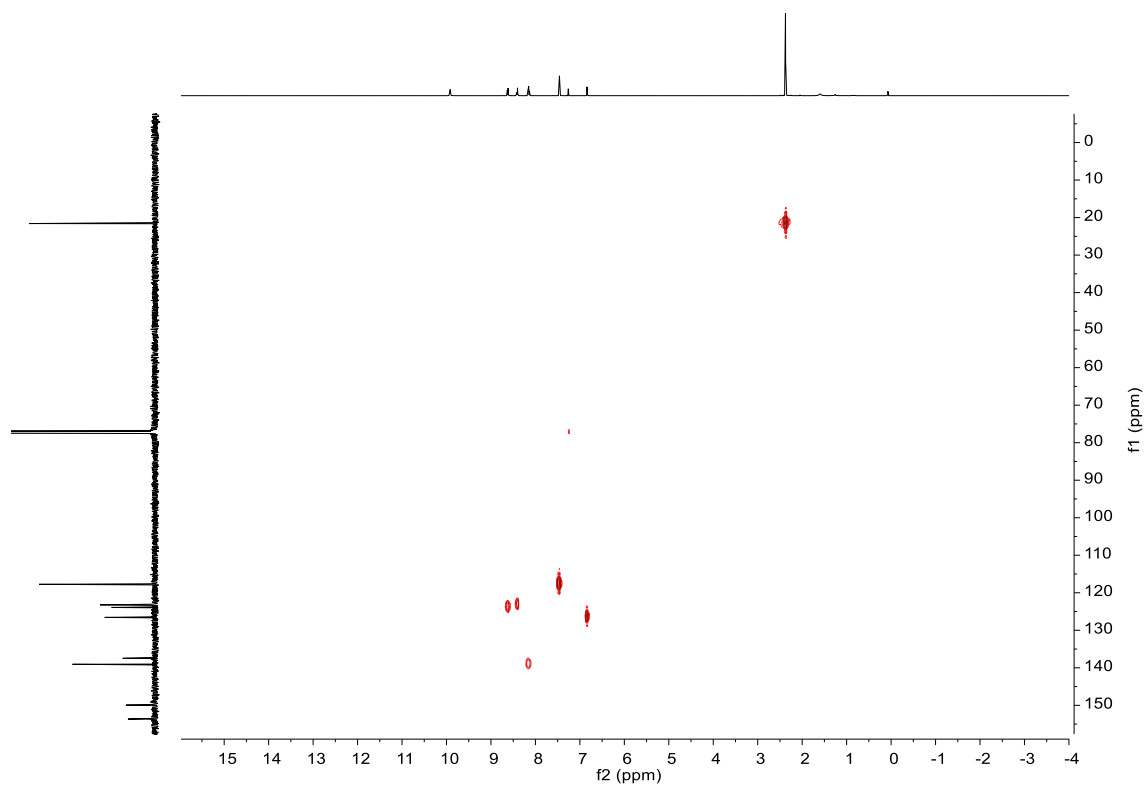


SUPPORTING INFORMATION

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SUPPORTING INFORMATION

E

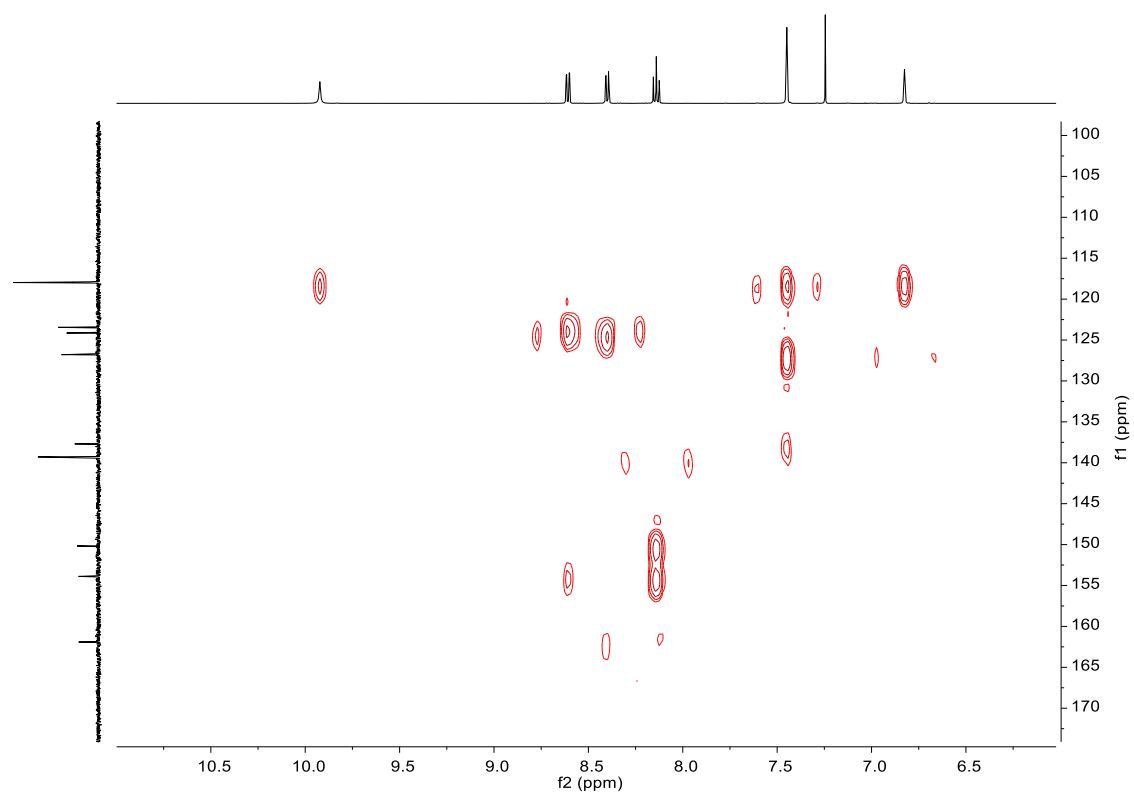
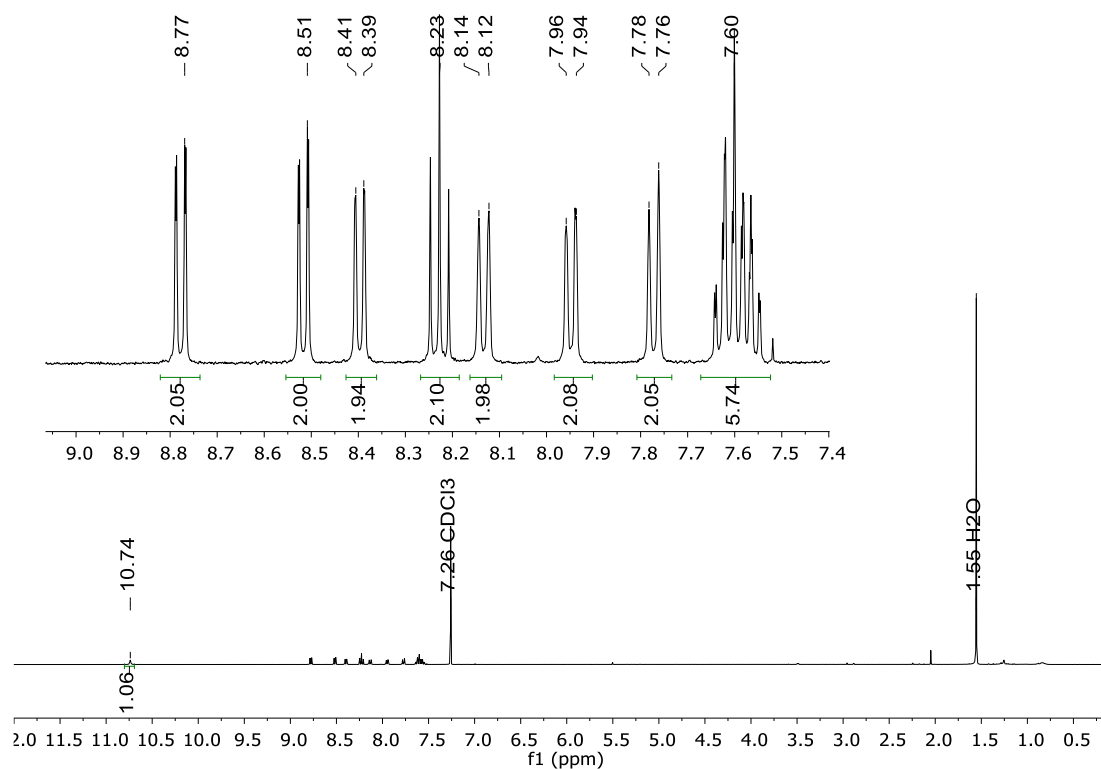


Figure S3. NMR spectra (DMSO-d₆, 500 MHz, 25 °C) for H₂L₂. (A) ¹H-NMR, (B) ¹³C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A



B

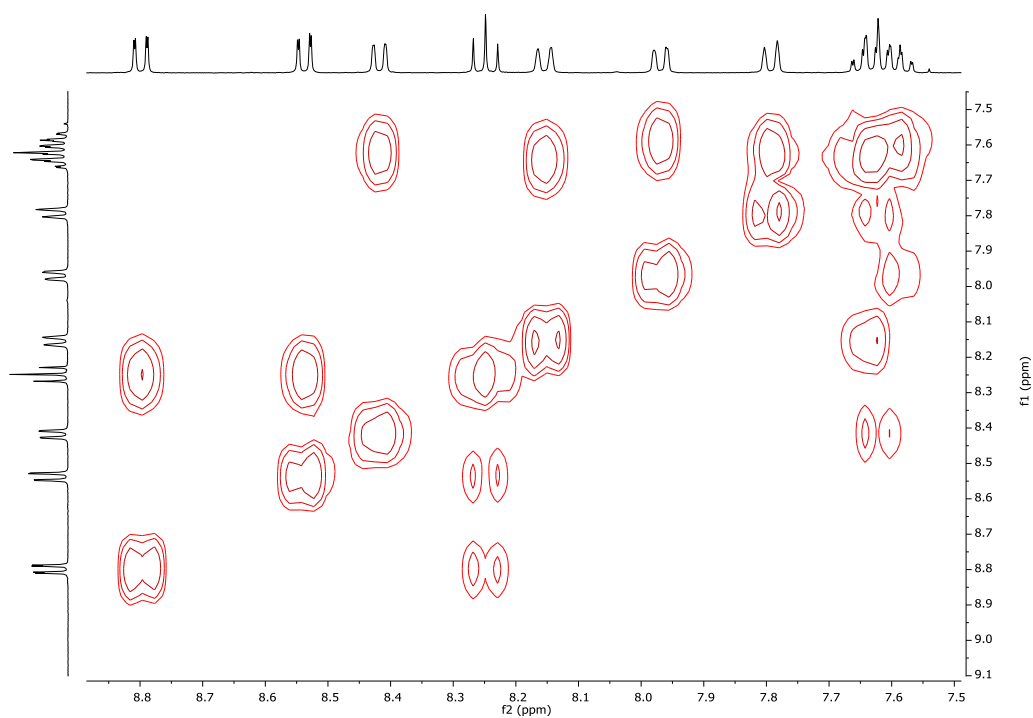
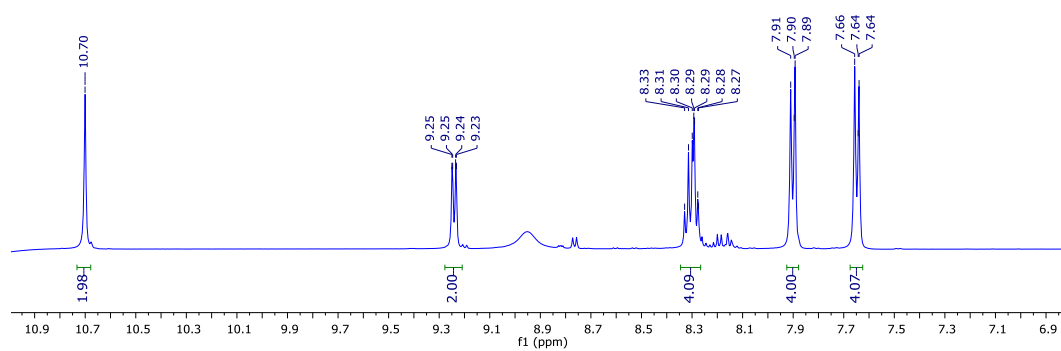


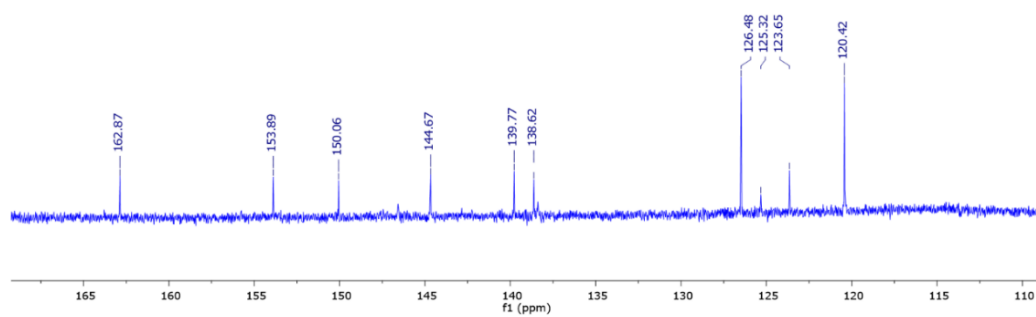
Figure S4. NMR spectra (DMSO- d_6 , 500 MHz, 25 °C) for **H₂L3**. (A) ^1H -NMR, (B) COSY. **Note:** Due to low solubility of the compound, ^{13}C -NMR was not measurable.

SUPPORTING INFORMATION

A

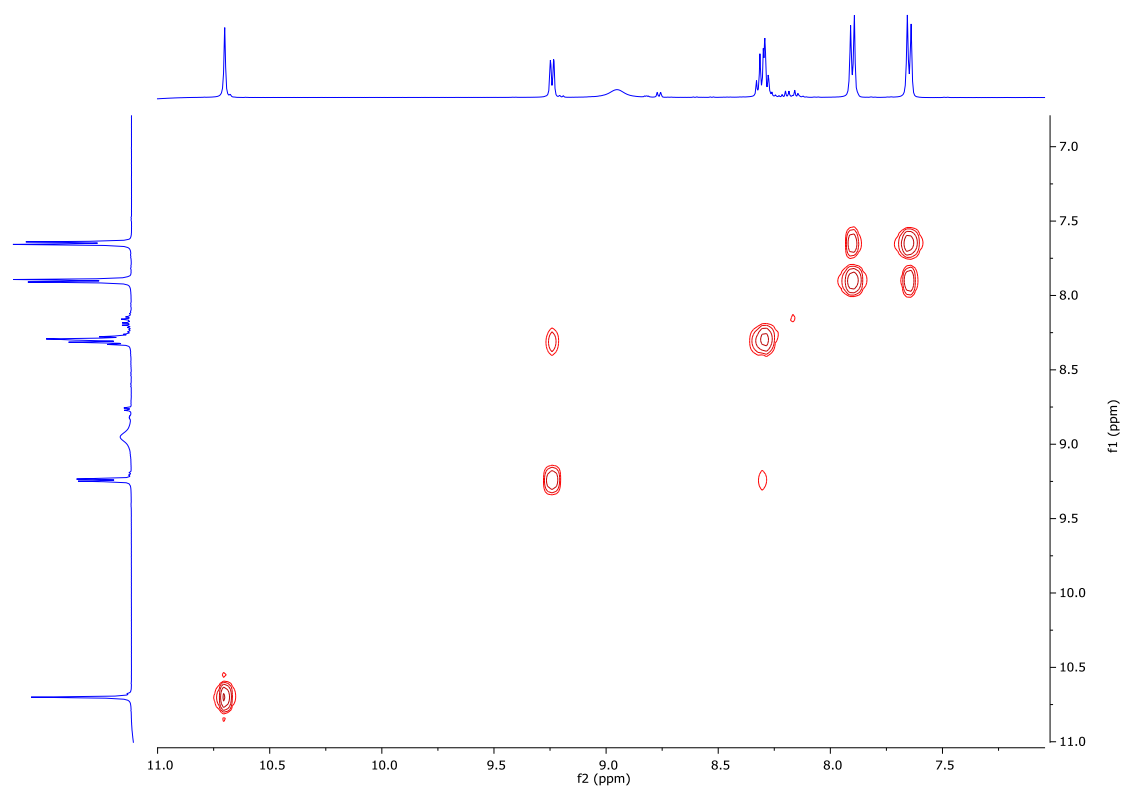


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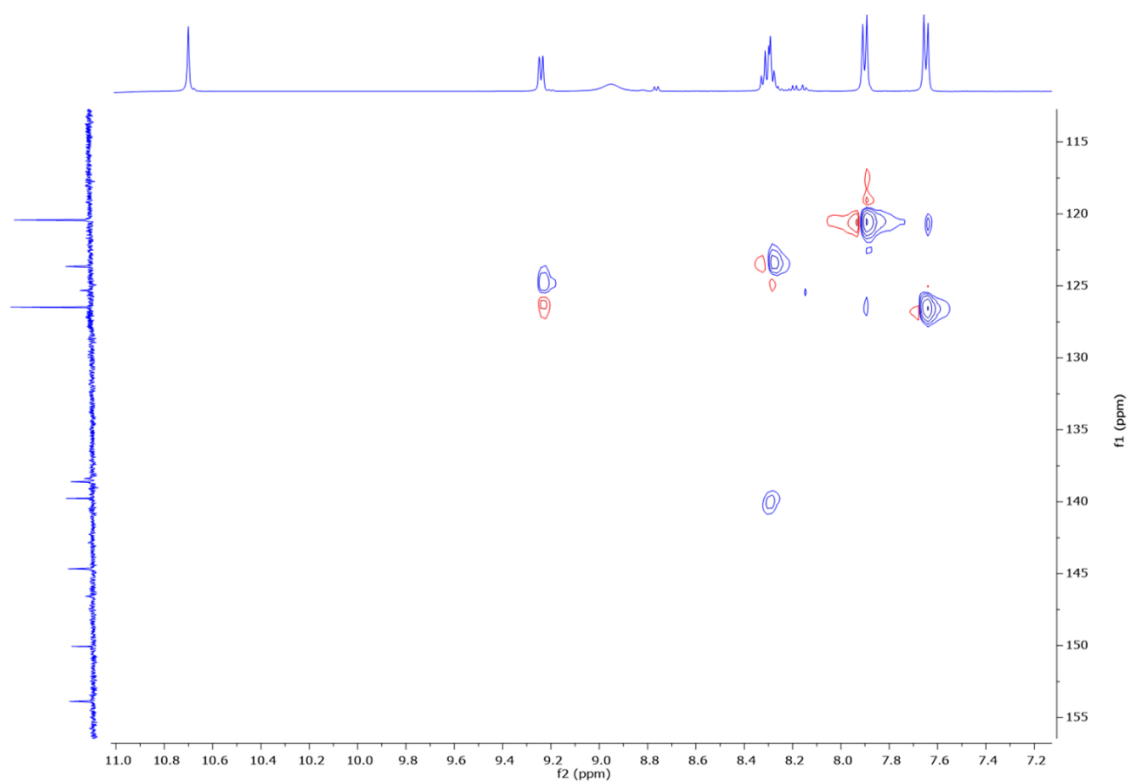


SUPPORTING INFORMATION

C



D



SUPPORTING INFORMATION

E

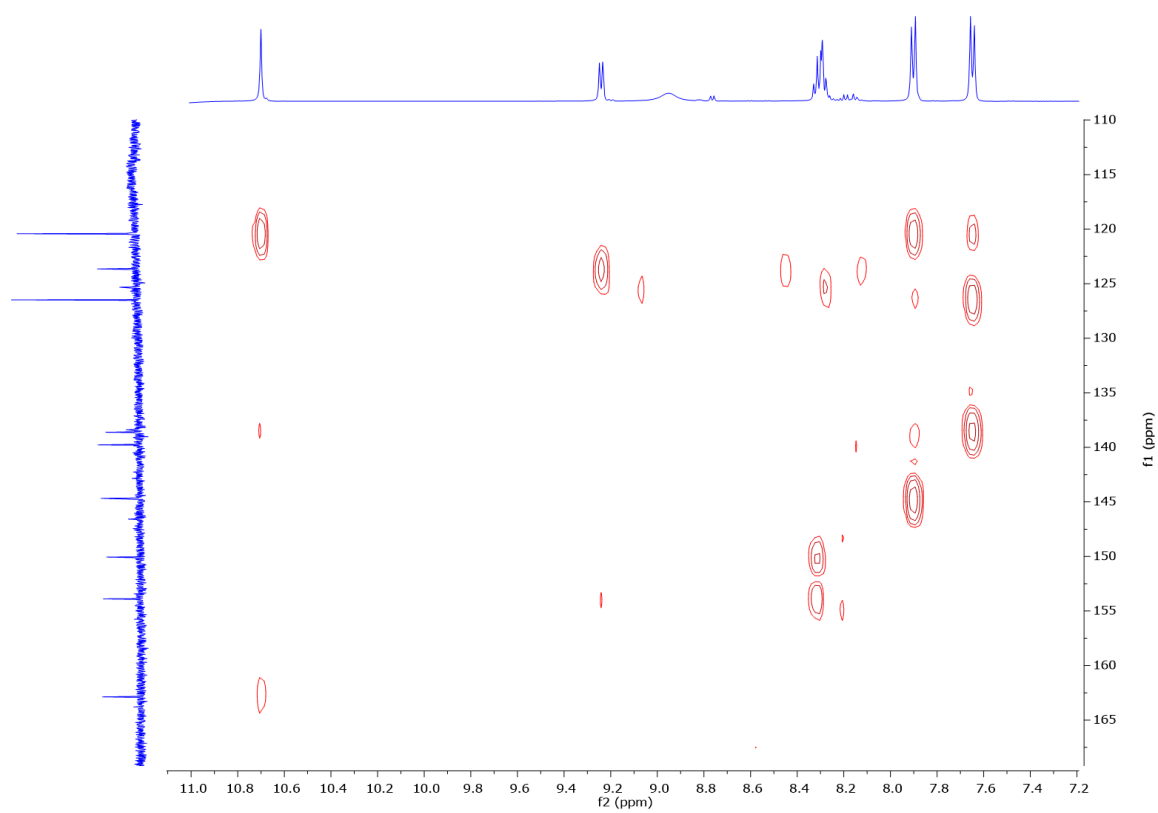
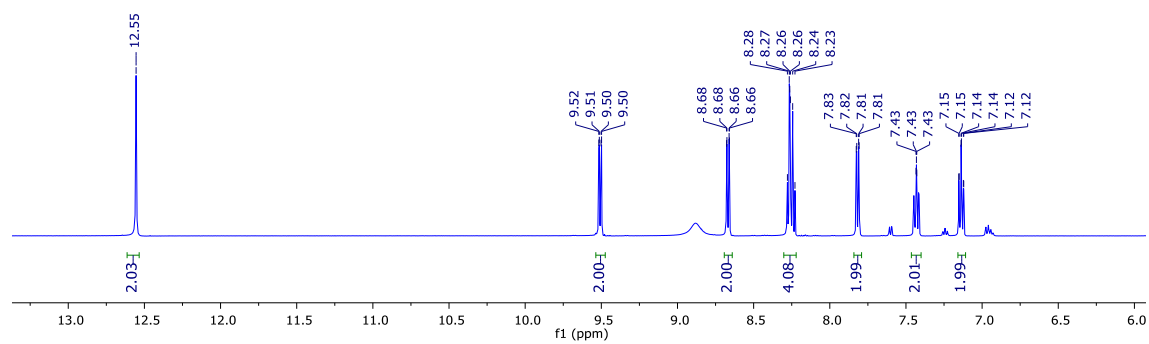


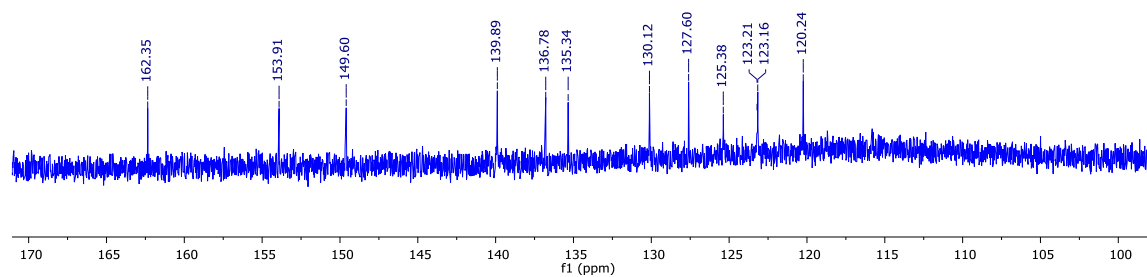
Figure S5. NMR spectra (DMSO- d_6 , 500 MHz, 25 °C) for H_2L4^2- . (A) 1H -NMR, (B) ^{13}C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A

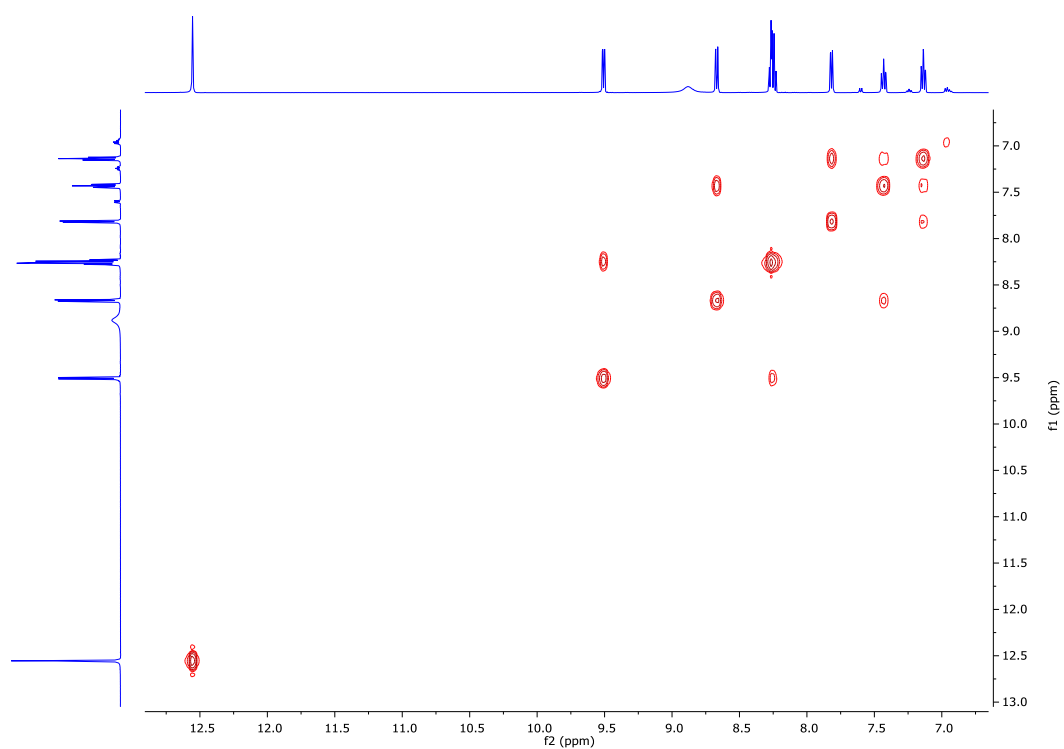


B

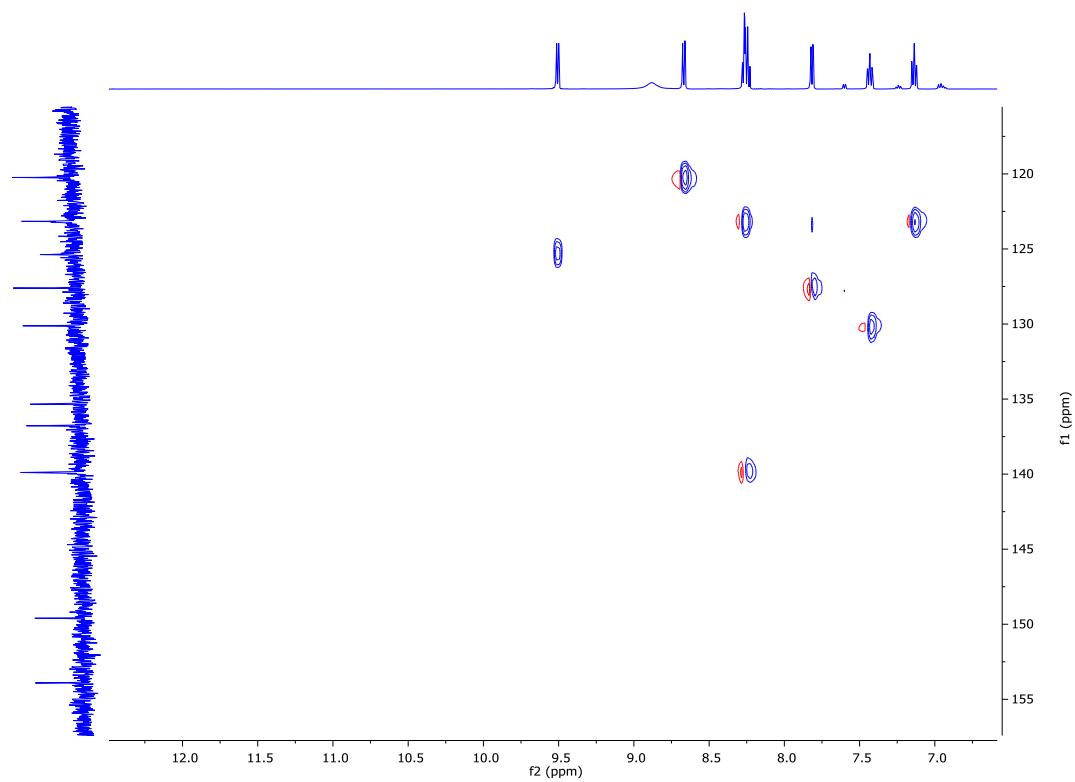


SUPPORTING INFORMATION

C



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SUPPORTING INFORMATION

E

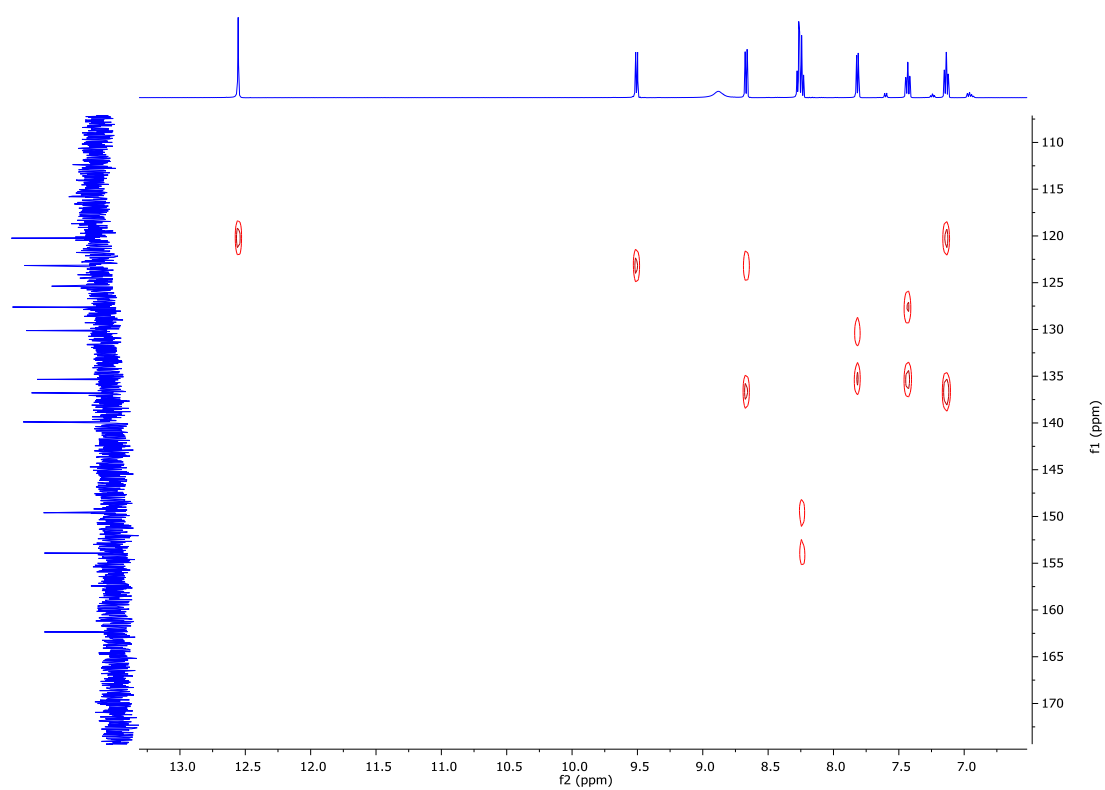
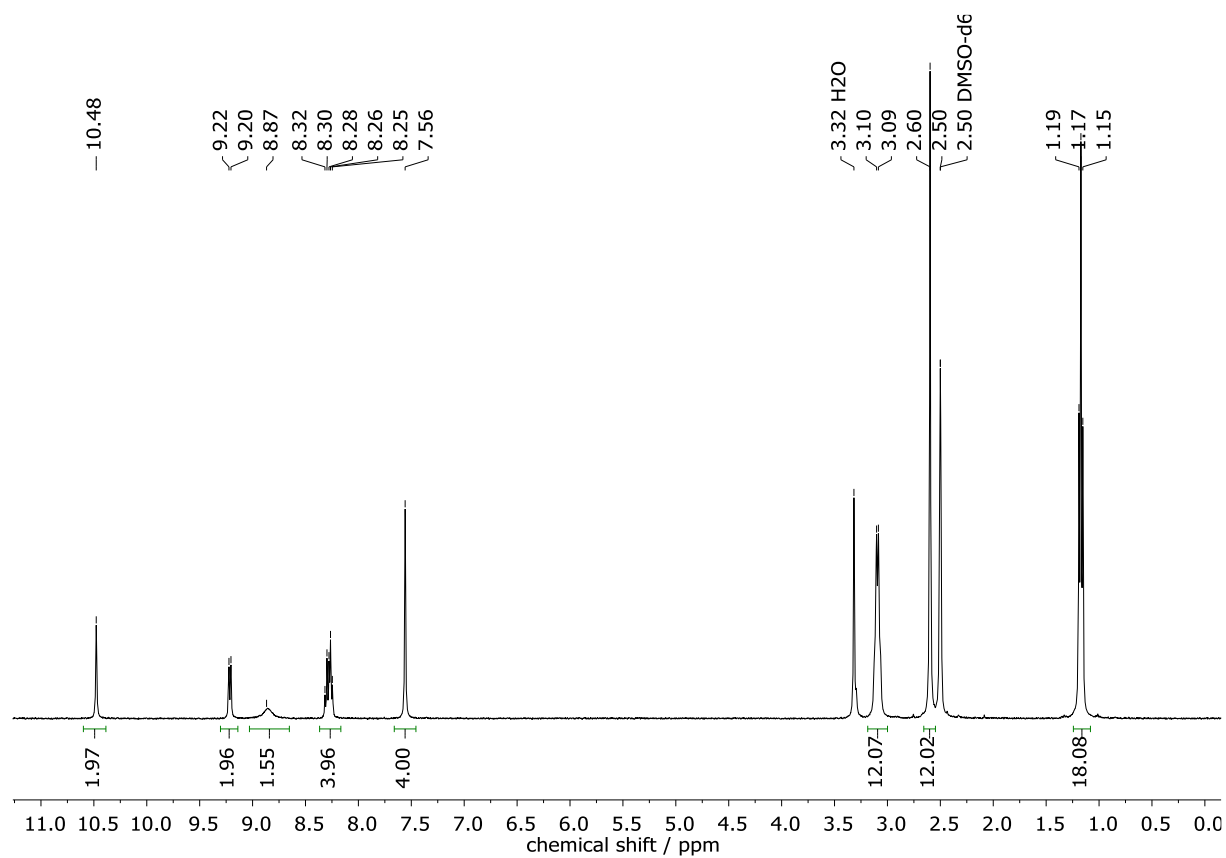


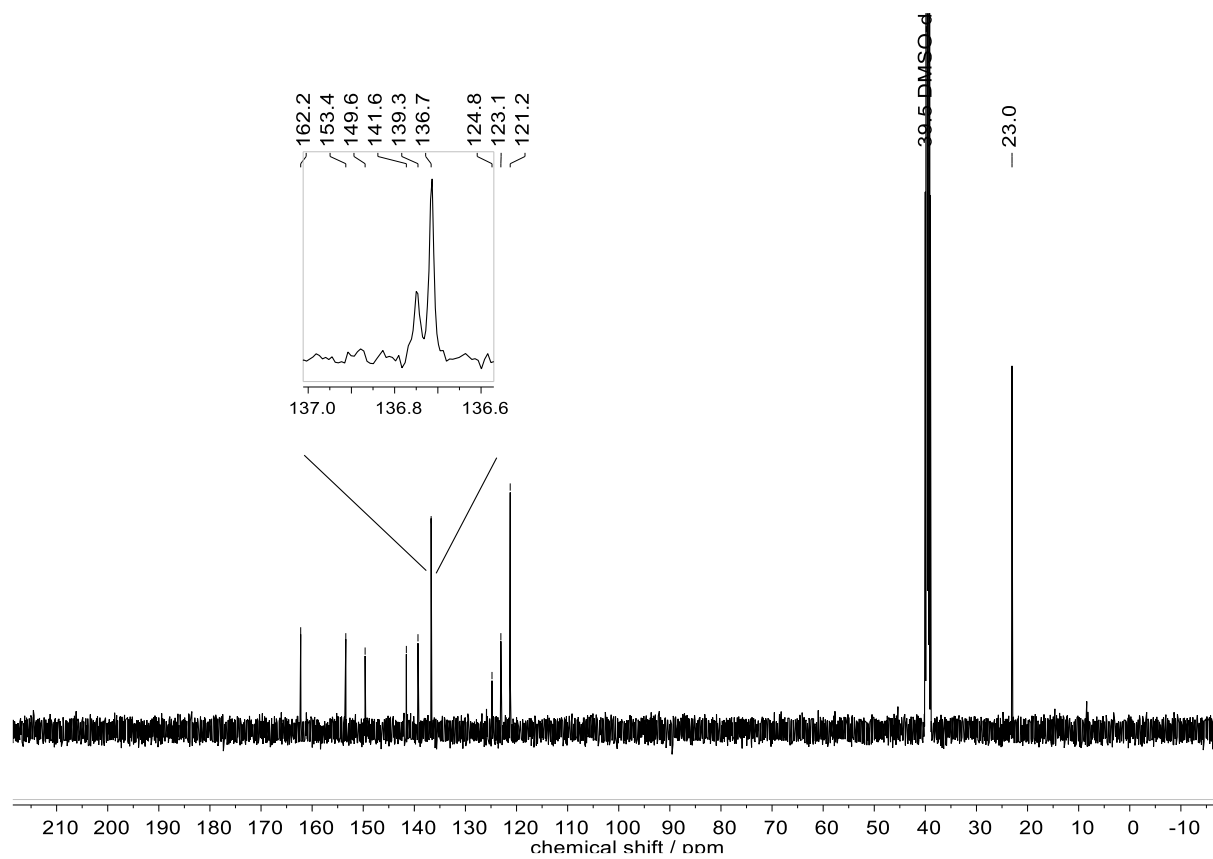
Figure S6. NMR spectra (DMSO-d₆, 500 MHz, 25 °C) for H₂L5²⁻. (A) ¹H-NMR, (B) ¹³C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A

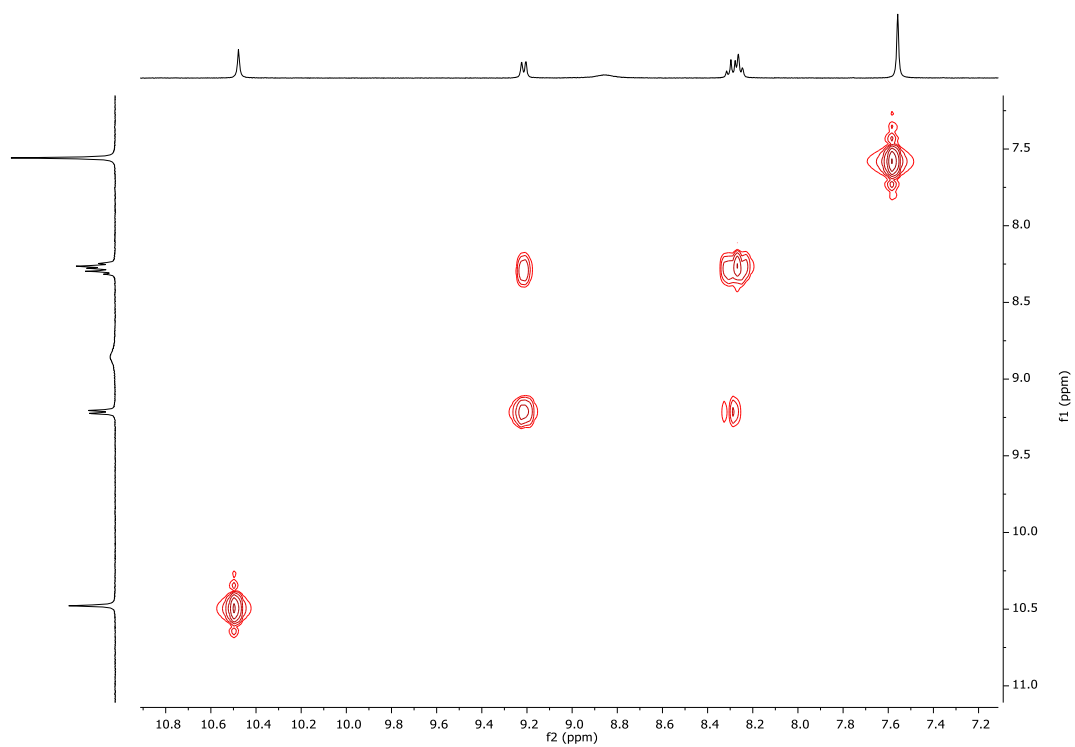


B

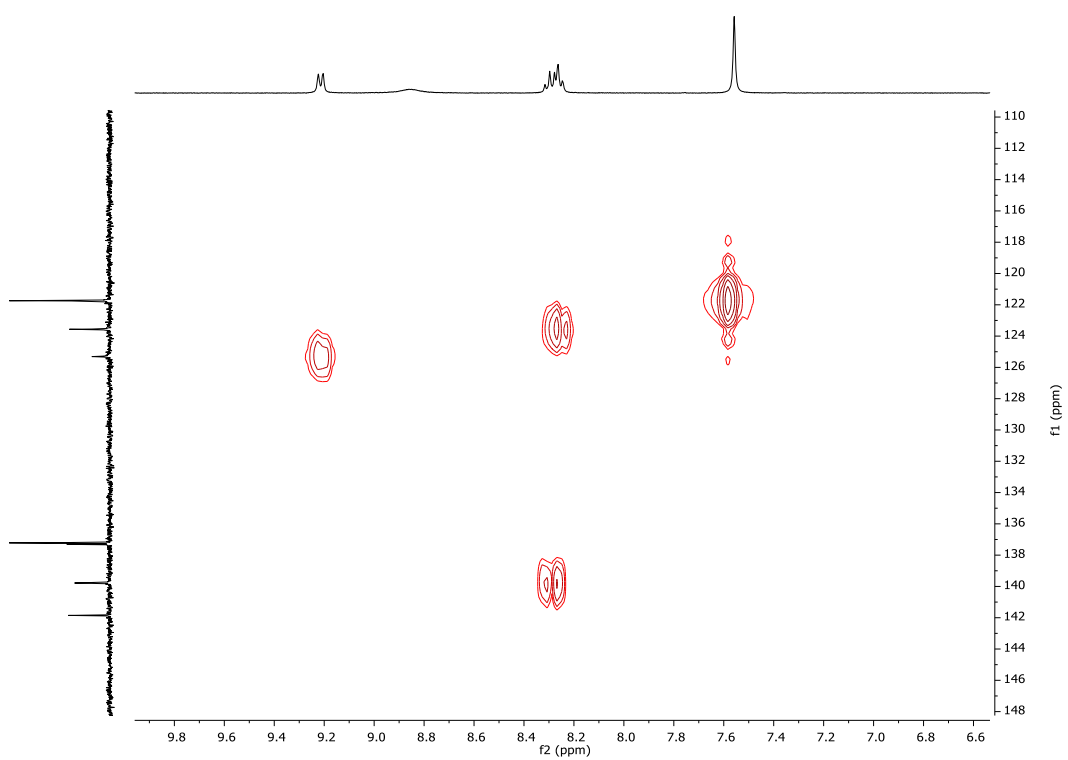


SUPPORTING INFORMATION

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SUPPORTING INFORMATION

E

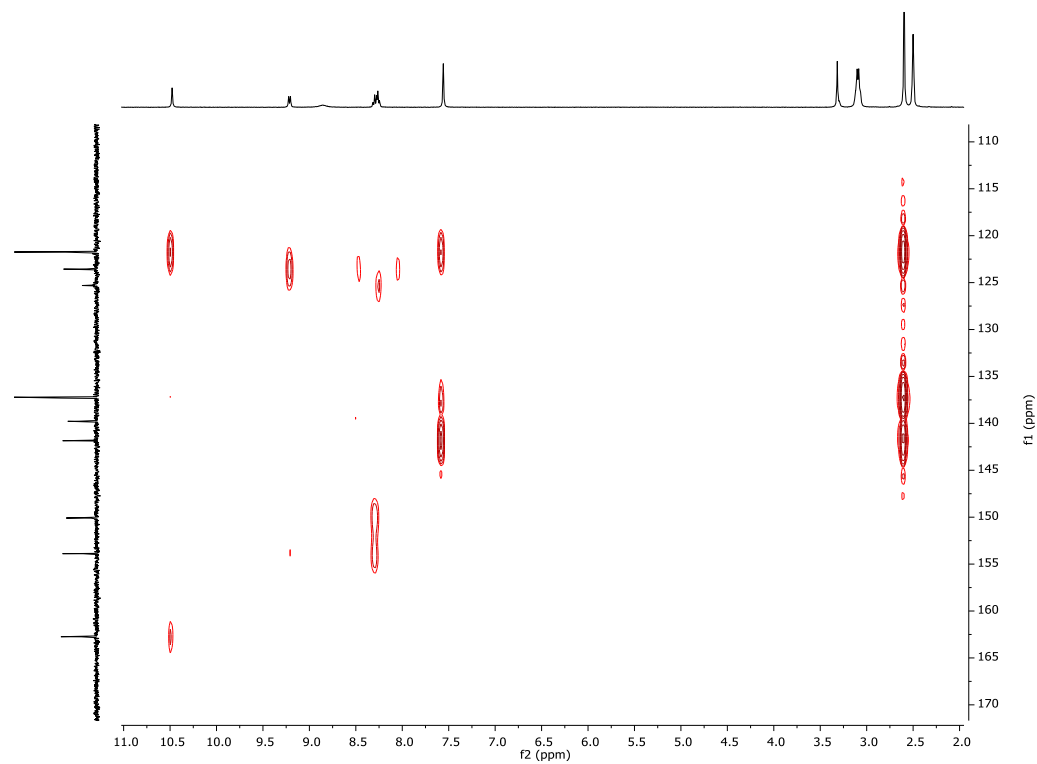
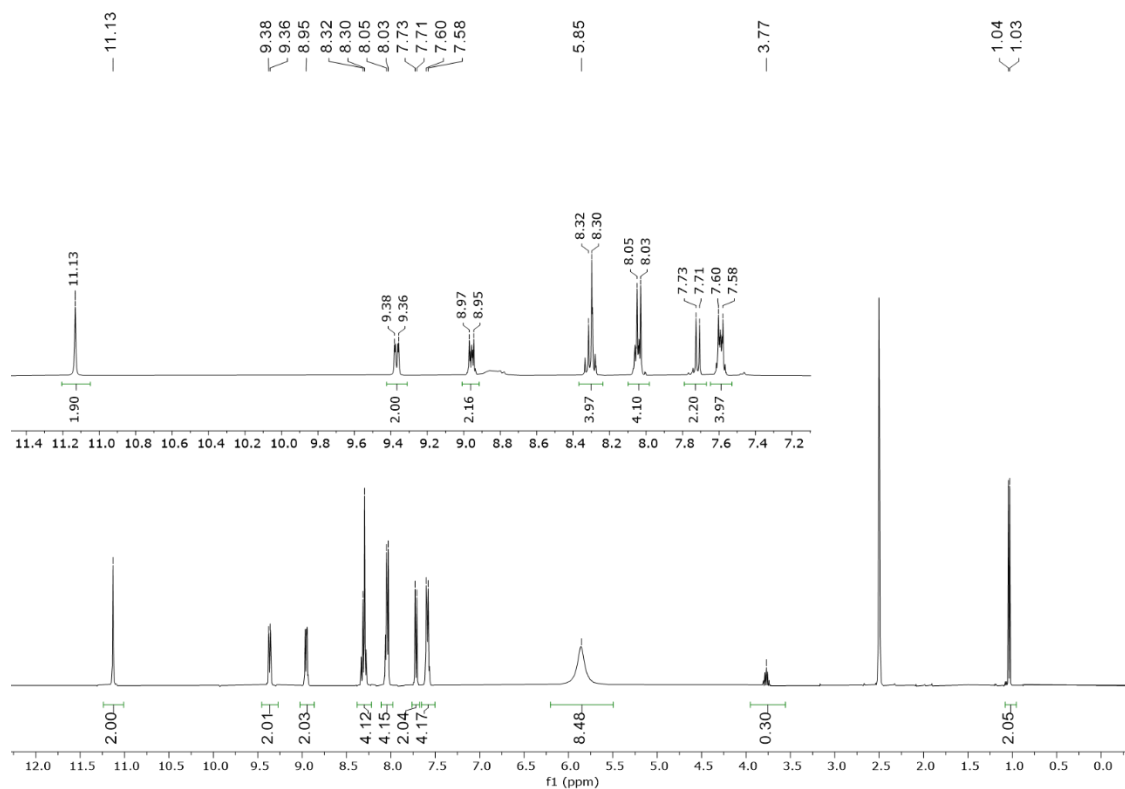


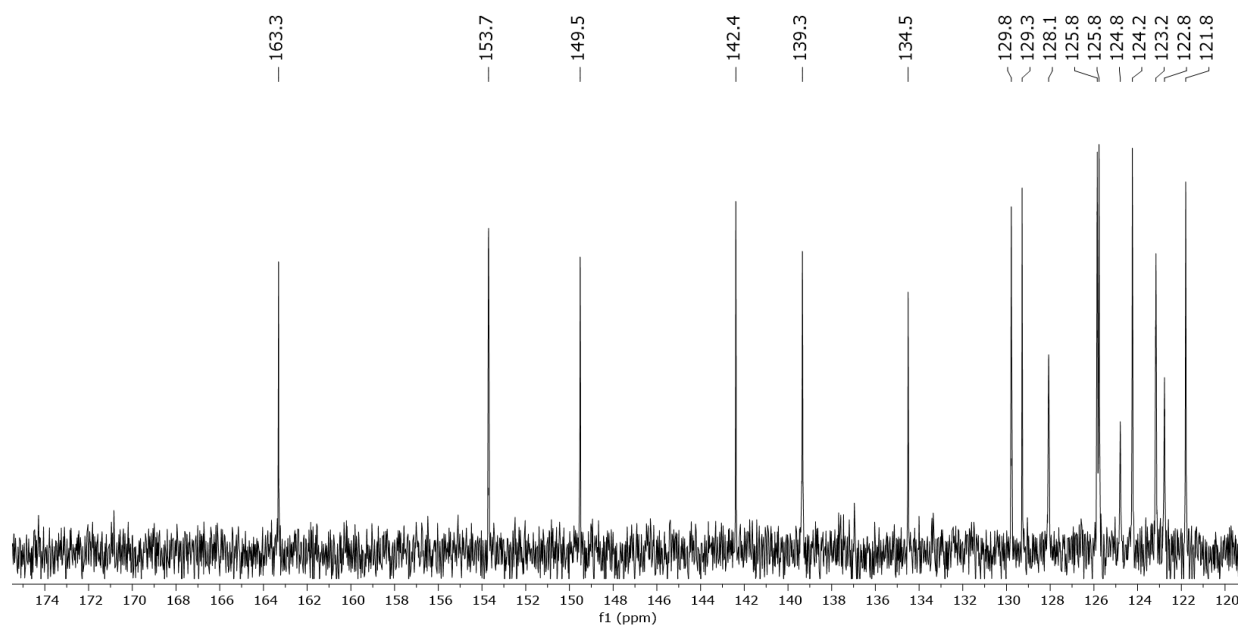
Figure S7. NMR spectra (DMSO- d_6 , 500 MHz, 25 °C) for $\text{H}_2\text{L6}^{2-}$. (A) ^1H -NMR, (B) ^{13}C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A

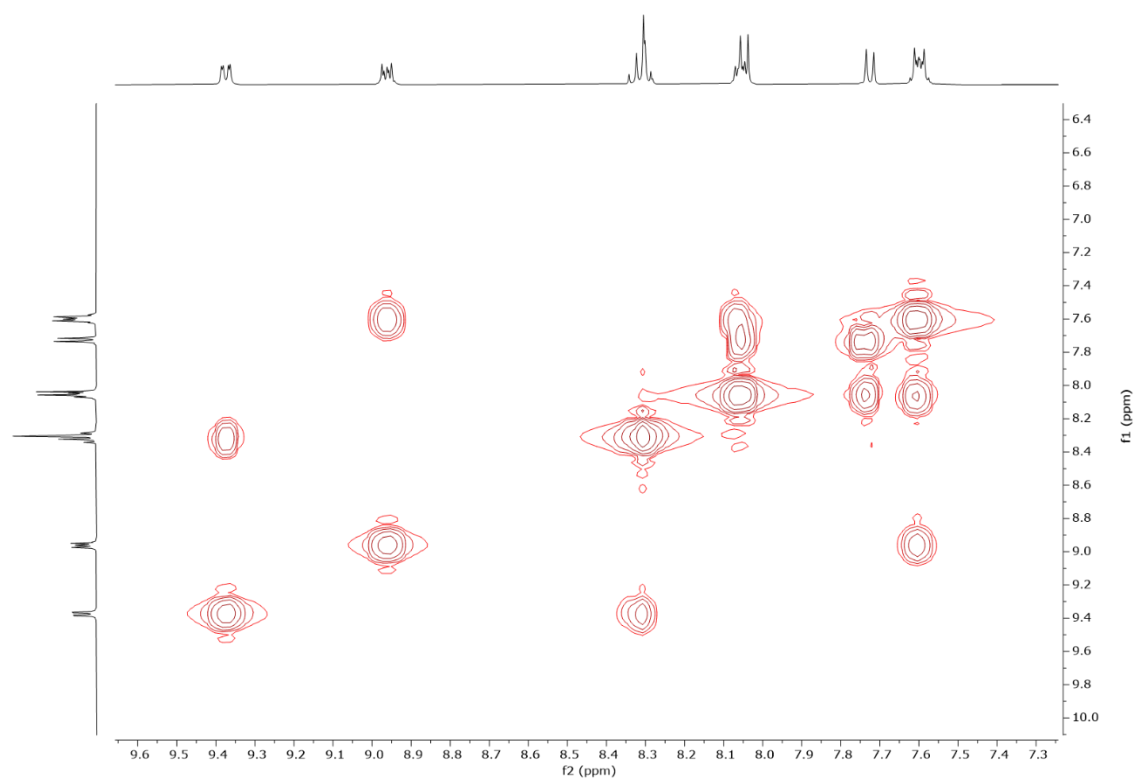


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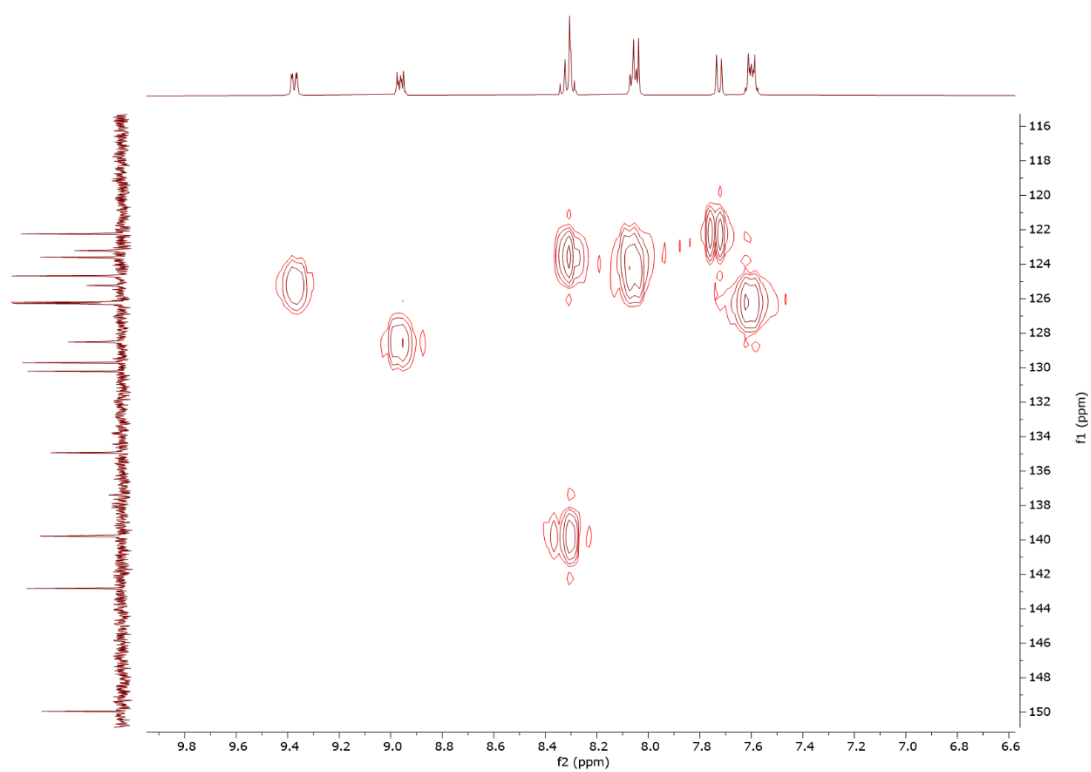


SUPPORTING INFORMATION

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SUPPORTING INFORMATION

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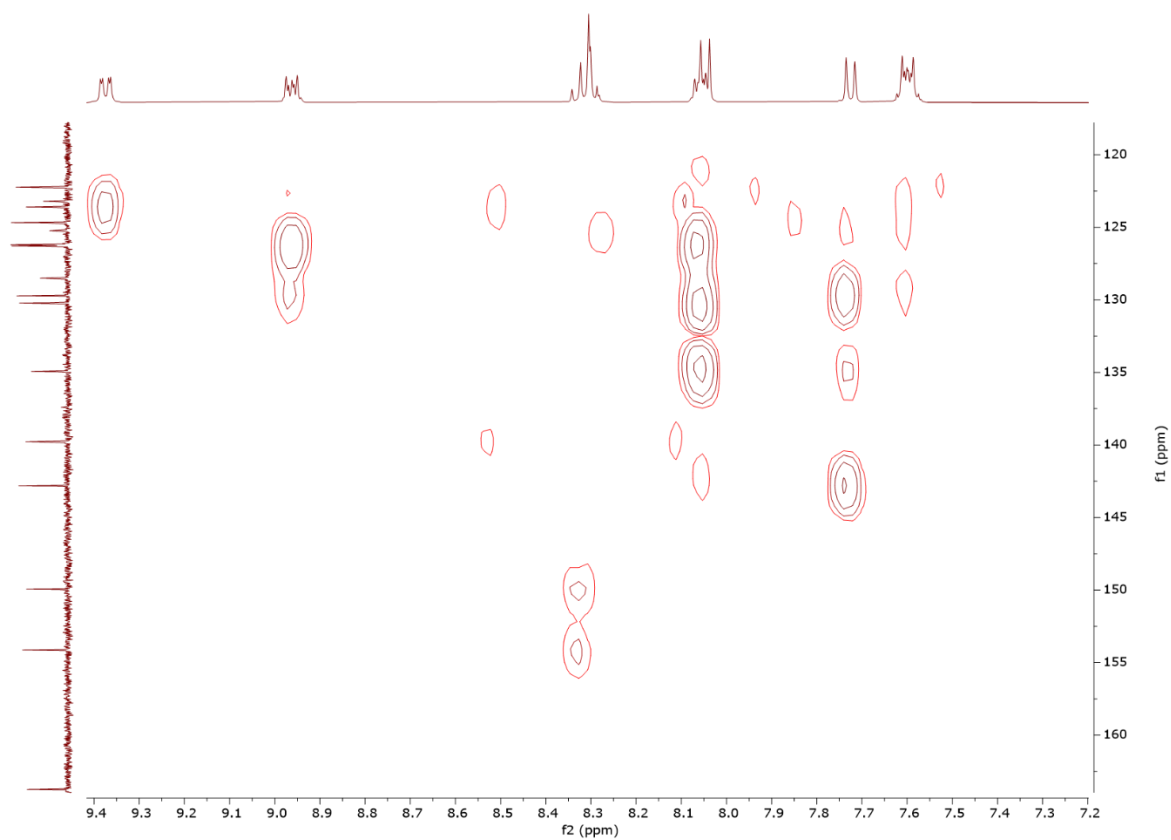
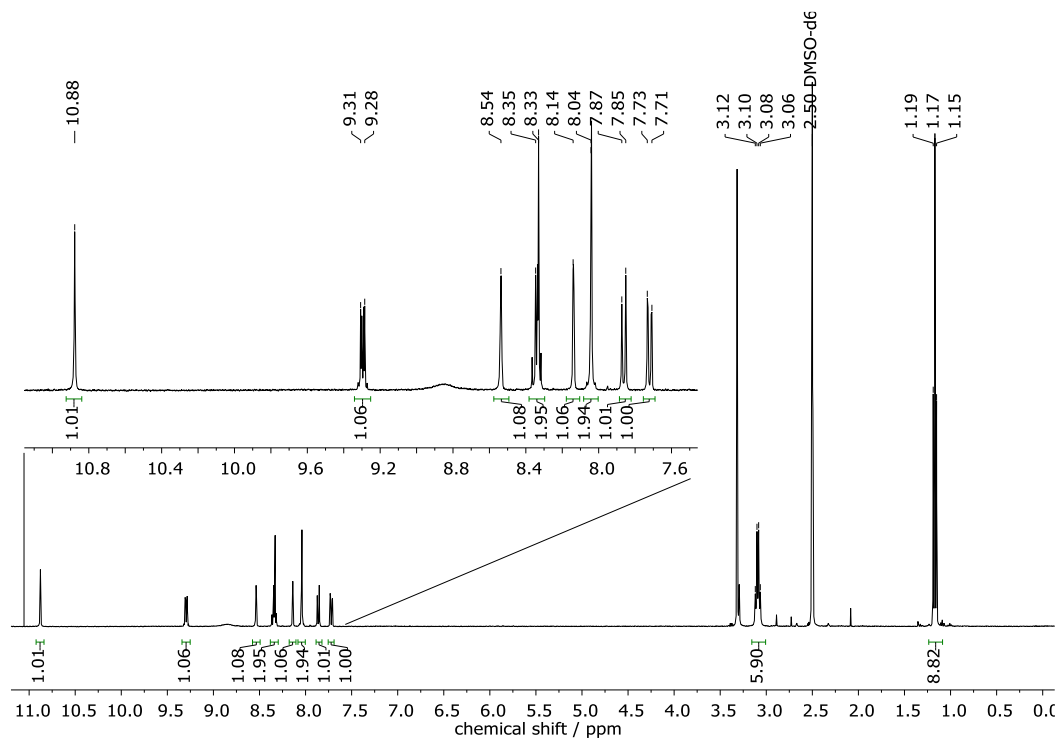


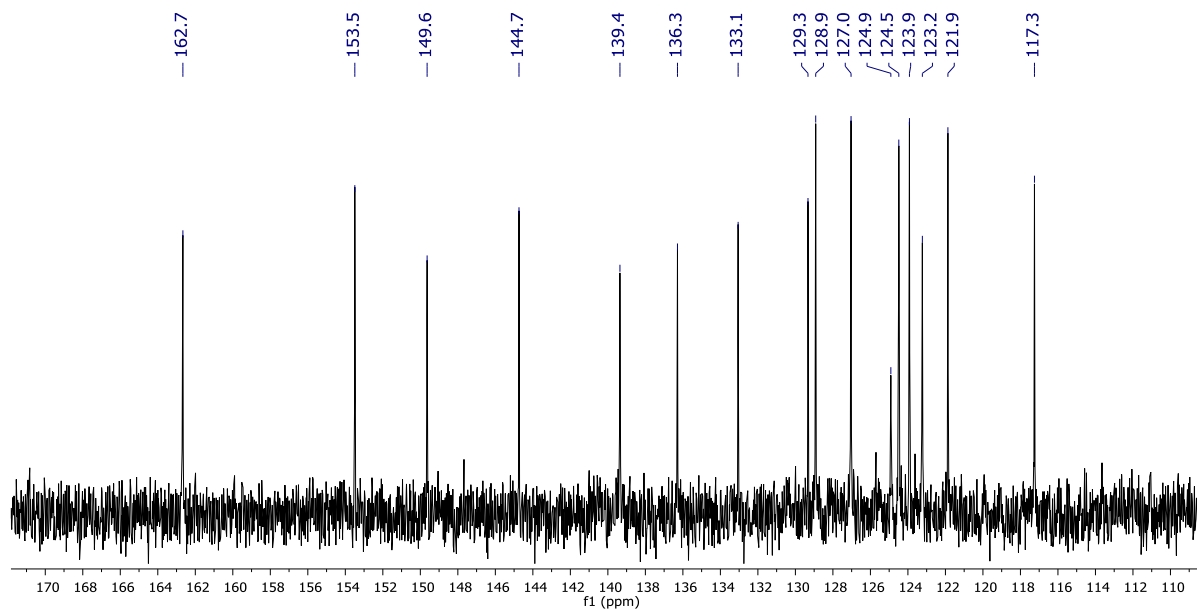
Figure S8. NMR spectra (DMSO- d_6 , 500 MHz, 25 °C) for H_2L7^2 . (A) 1H -NMR, (B) ^{13}C NMR, (C) COSY, (D) HSQC and (E) HMBC.

SUPPORTING INFORMATION

A

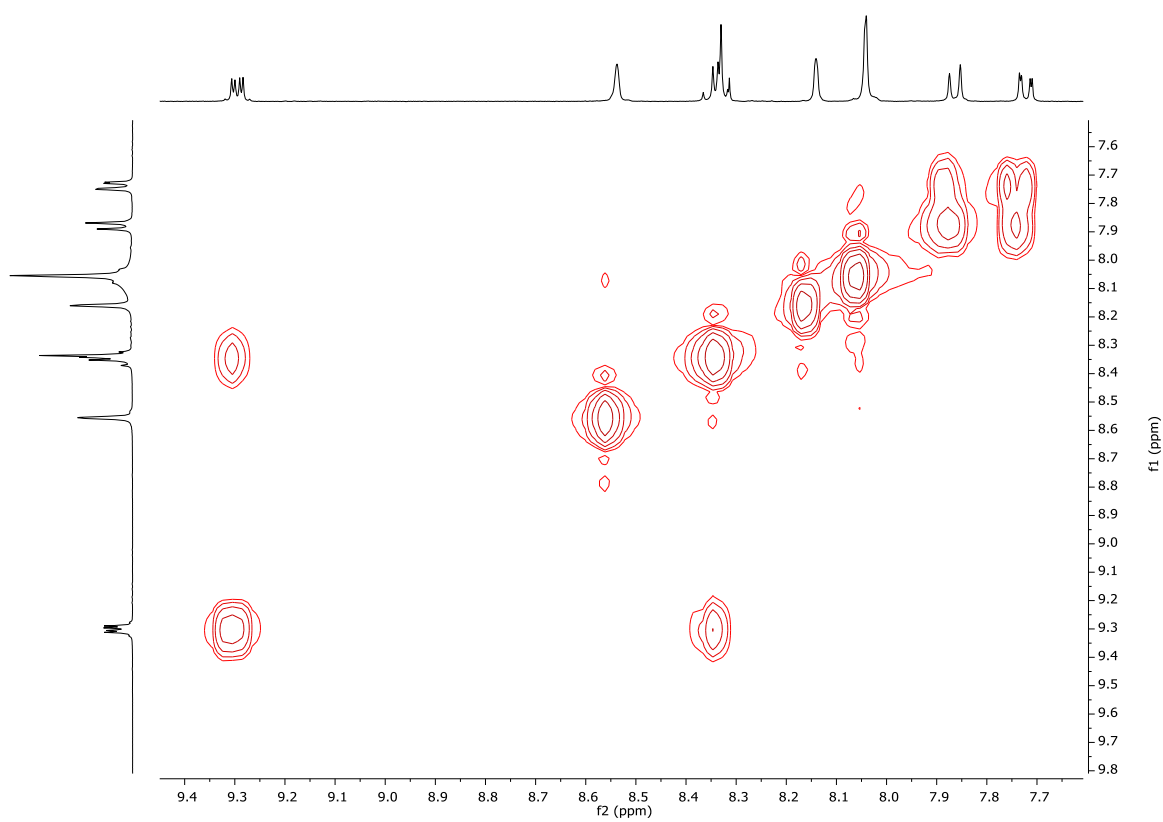


B

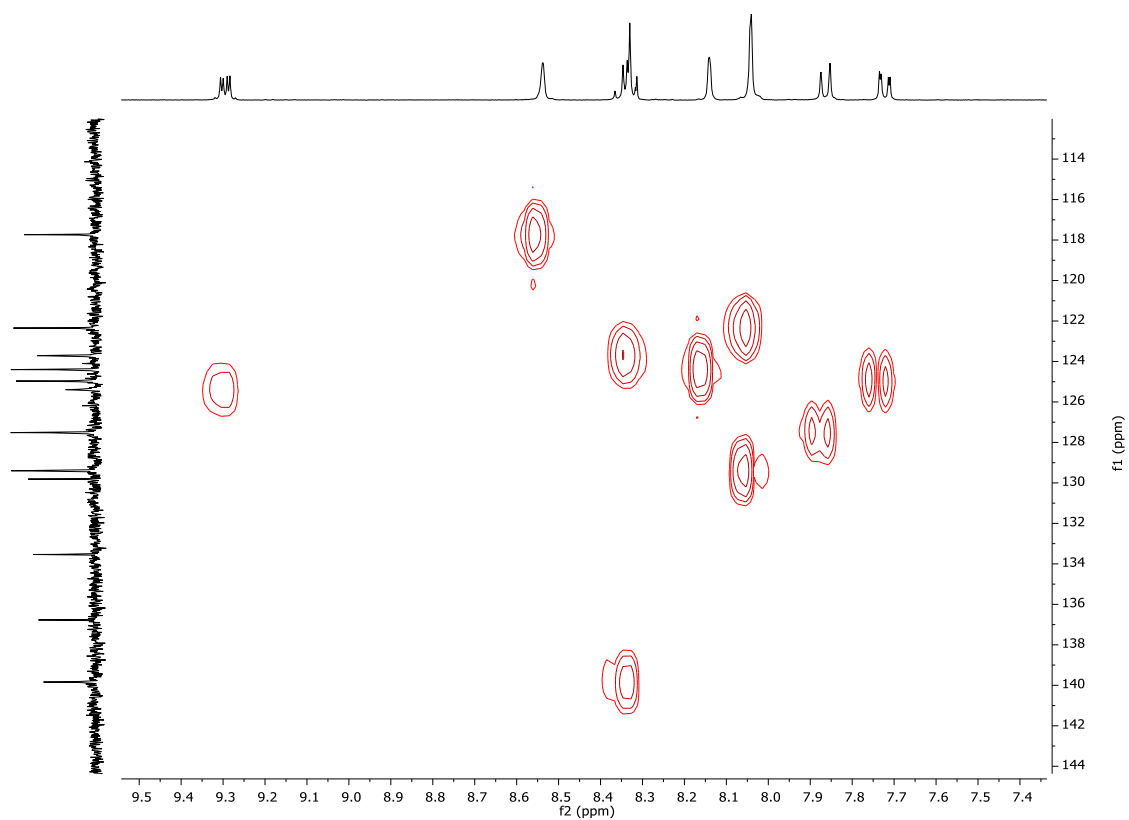


SUPPORTING INFORMATION

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SUPPORTING INFORMATION

E

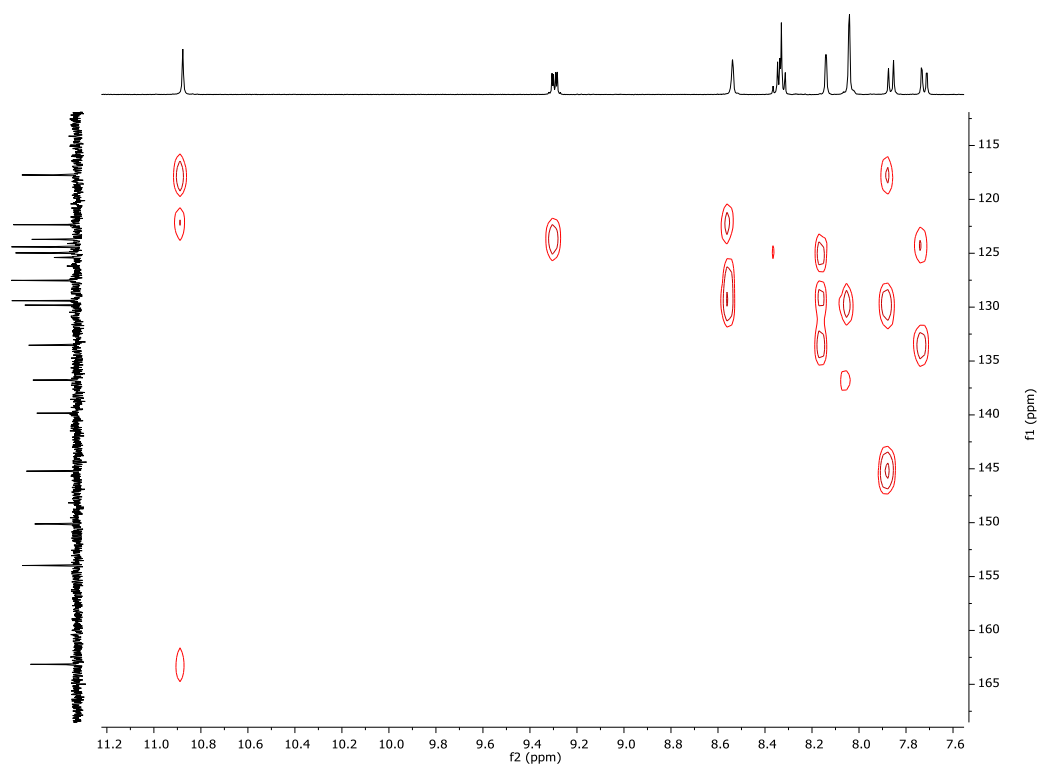


Figure S9. NMR spectra (DMSO- d_6 , 500 MHz, 25 °C) for H_2L8^{2-} . (A) 1H -NMR, (B) ^{13}C NMR, (C) COSY, (D) HSQC and (E) HMBC.

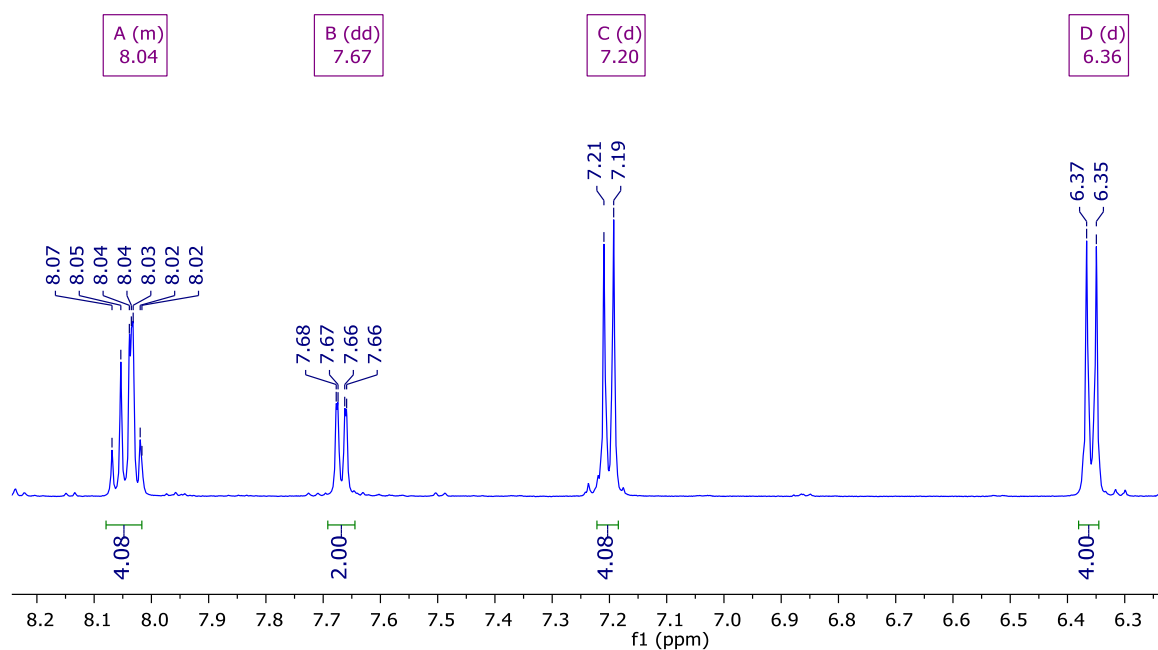


Figure S10. 1H -NMR spectrum of the complex $[(L4)Zn]^{2-}$ (D_2O - d_2 , 500 MHz, 25 °C).

SUPPORTING INFORMATION

Mass-spectrometry

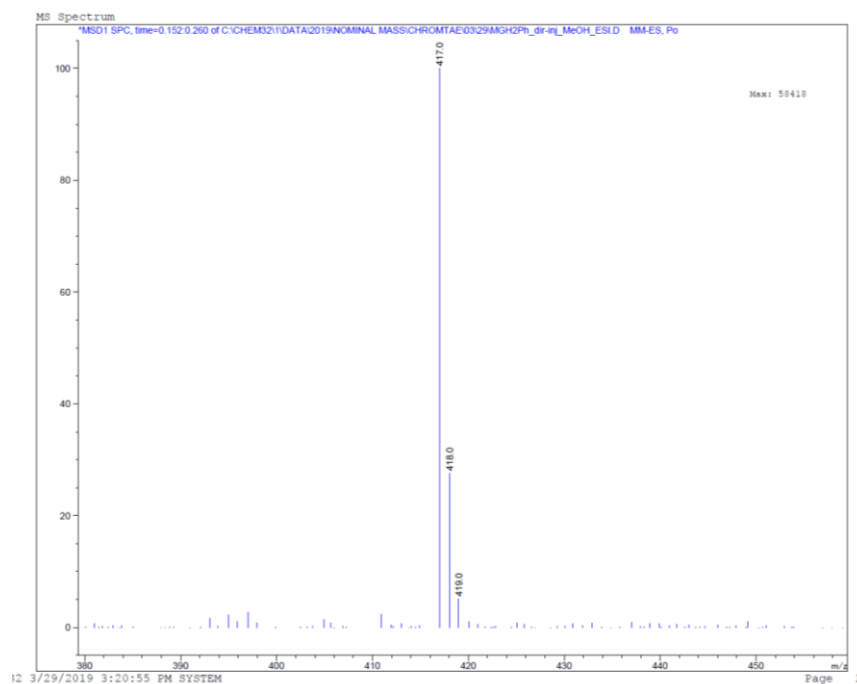


Figure S11. (+)-ESI-MS for **H₂L1**. m/z: 417.0 [**H₂L1**+Na]⁺

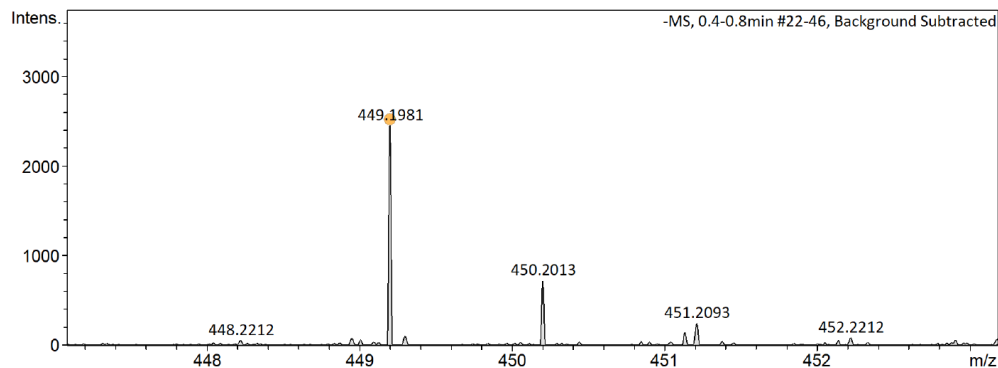


Figure S12. (+)-ESI-HRMS for **H₂L2**. Calcd for [**H₂L2**-H]⁺, (C₂₈H₂₅N₄O₂): 449.1983, found: 449.1981.

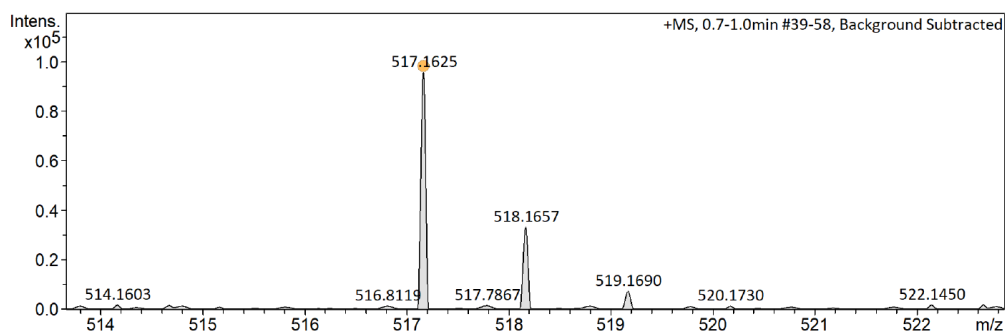


Figure S13. (+)-ESI-HRMS for **H₂L3**. Calcd for [**H₂L3**+Na]⁺, (C₃₂H₂₂N₄O₂Na): 517.1635, found: 517.1625.

SUPPORTING INFORMATION

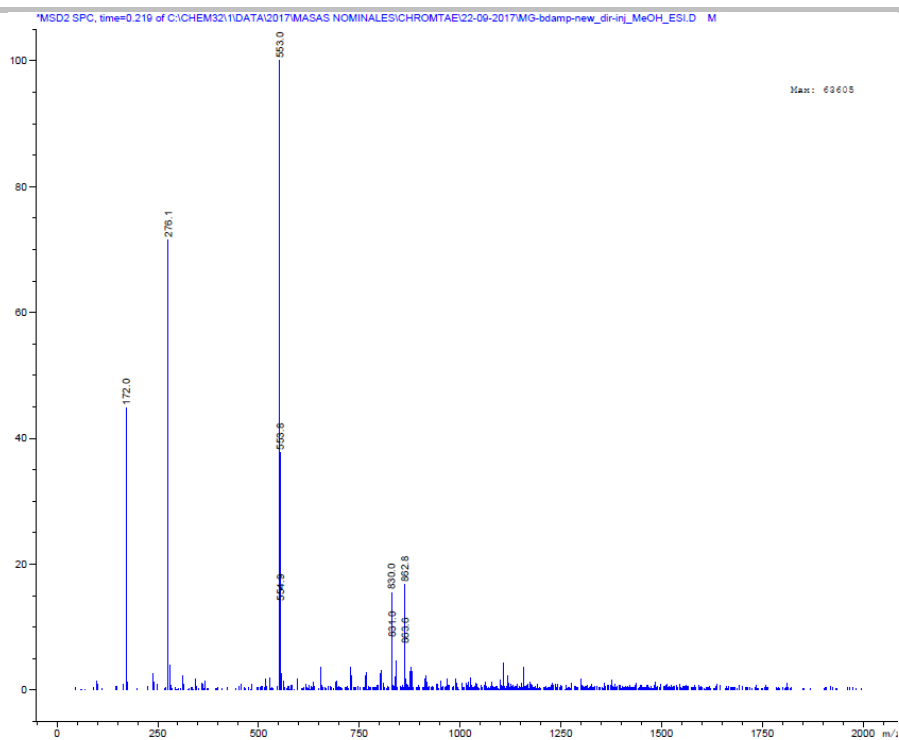


Figure S14. (-)-ESI-MS spectrum for $\text{H}_2\text{L4}^{2-}$. m/z: 553.0 [$\text{H}_2\text{L4}+\text{H}^+$], 276.1 [$\text{H}_2\text{L4}^{2-}$].

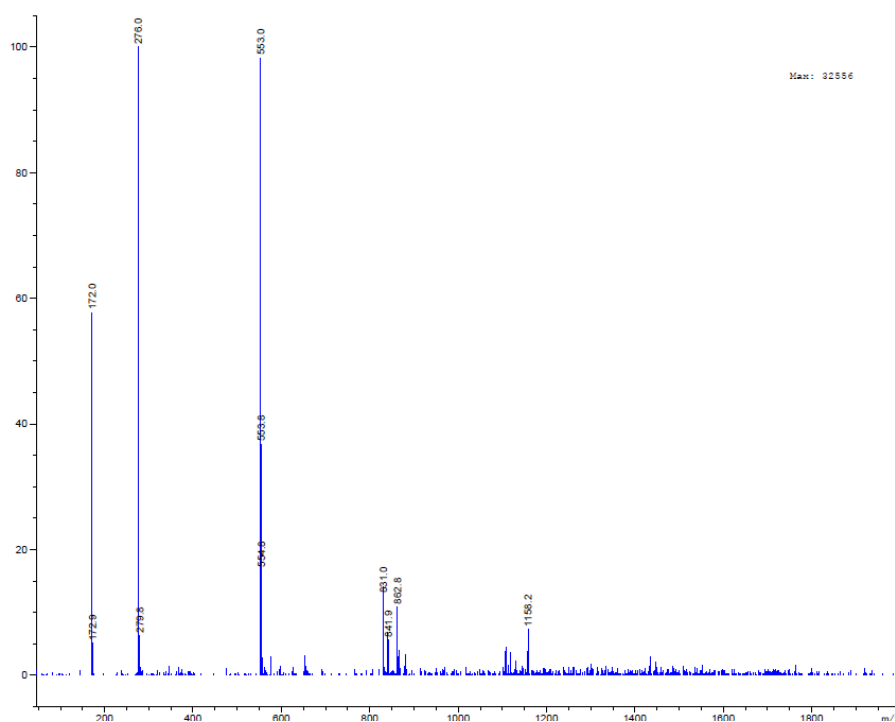


Figure S15. (-)-ESI-MS spectrum for the ligand $\text{H}_2\text{L5}^{2-}$. m/z: 553.0 [$\text{H}_2\text{L5}+\text{H}^+$], 276.1 [$\text{H}_2\text{L5}^{2-}$].

SUPPORTING INFORMATION

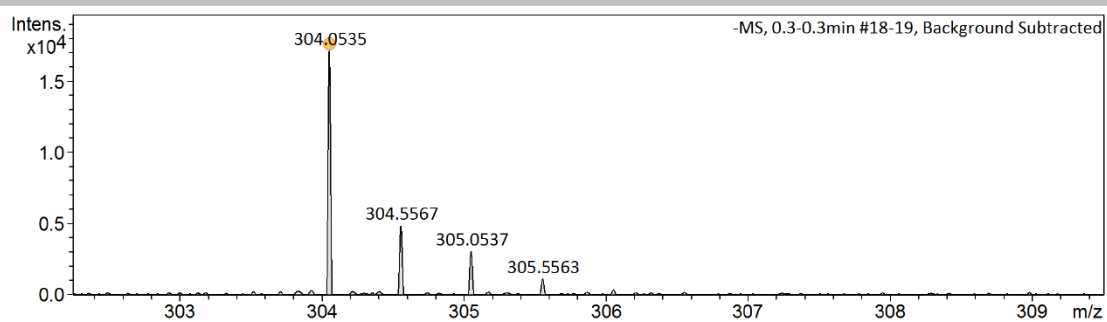


Figure S16. (-)-ESI-HRMS for H₂L6²⁻. Calcd for [H₂L6]²⁻, (C₂₈H₂₄N₄O₈S₂): 304.0523, found 304.0535.

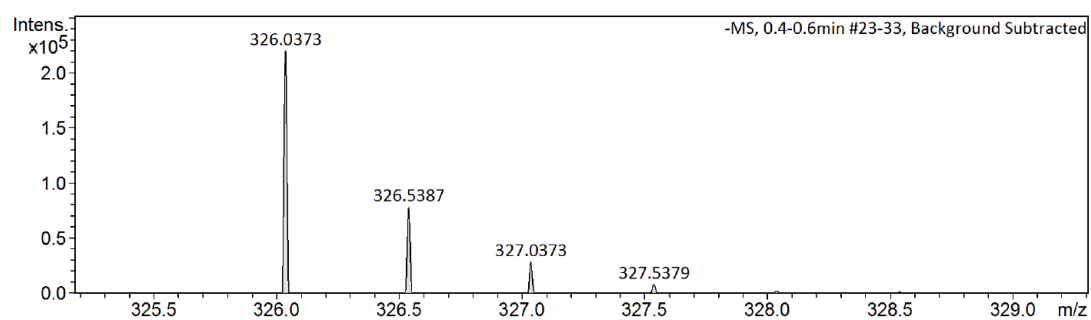


Figure S17. (-)-ESI-MS for H₂L7²⁻. Calcd for [H₂L7]²⁻, (C₃₂H₂₂N₄O₈S₂): 326.0367, found 326.0373.

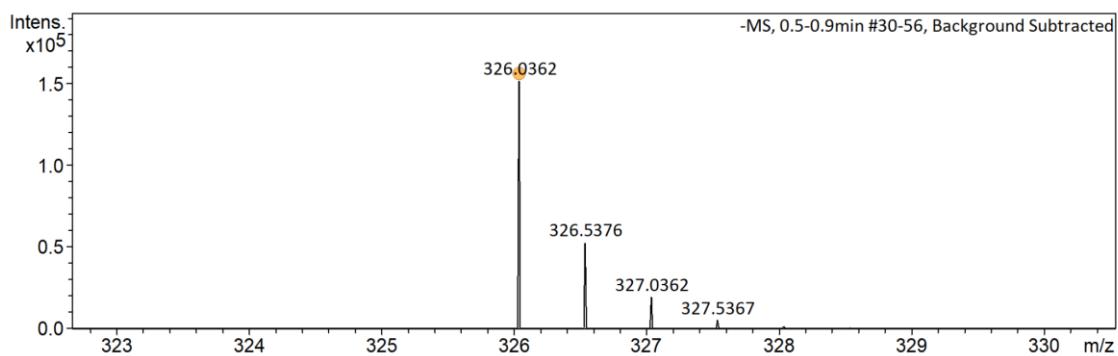


Figure S18. (-)-ESI-MS for H₂L8²⁻. Calcd for [H₂L8]²⁻, (C₃₂H₂₂N₄O₈S₂): 326.0367, found 326.0362.

SUPPORTING INFORMATION

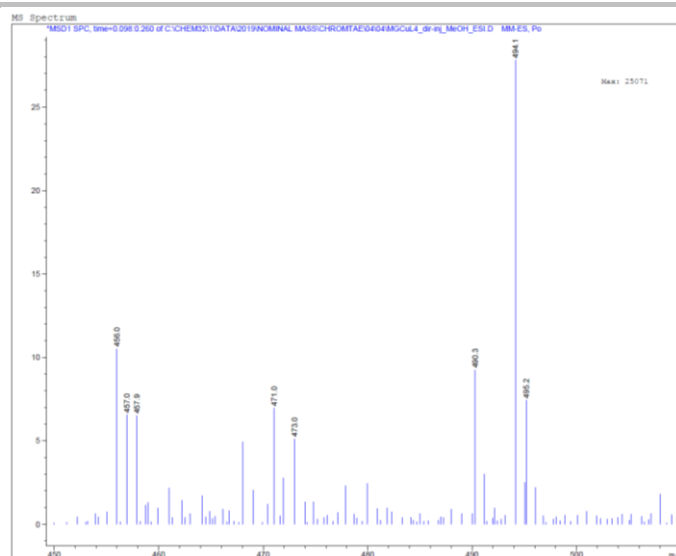


Figure S19. (+)-ESI-MS for [(L1)Cu]. $m/z = 456.0$ [(L1)Cu-H]⁺; $m/z = 494.1$ [(L1)Cu+K]⁺.

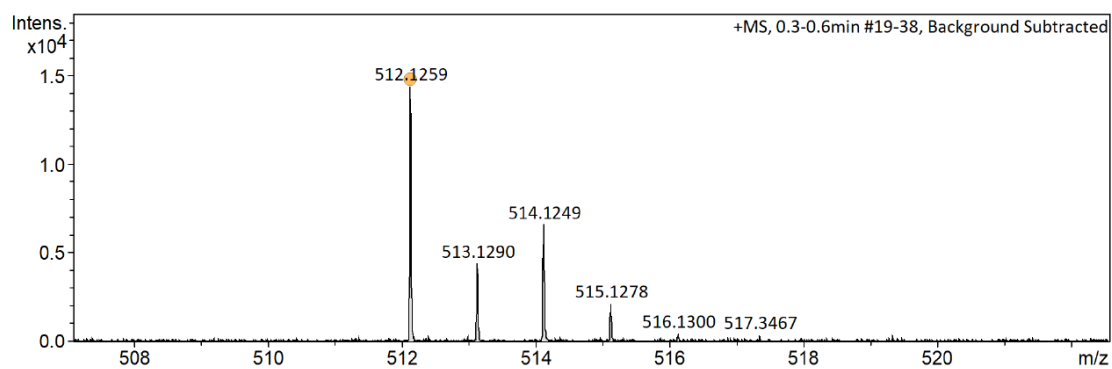


Figure S20. (+)-ESI-HRMS for [(L2)Cu]. Calcd for [(L2)Cu+H]⁺, (C₂₈H₂₅N₄O₂Cu) : 512.1259, found 512.1268.

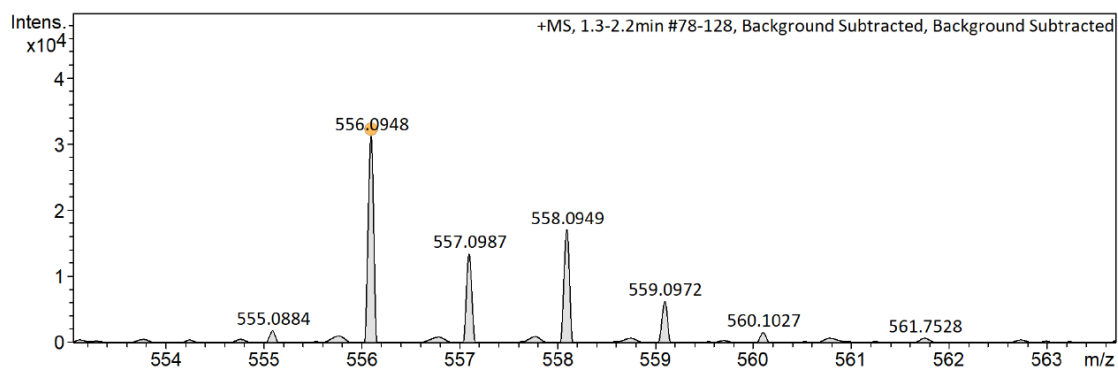


Figure S21. (-)-ESI-HRMS for [(L3)Cu]. Calcd for [(L3)Cu+H]⁺, (C₃₂H₂₁CuN₄O₂): 556.0955, found: 556.0948.

SUPPORTING INFORMATION

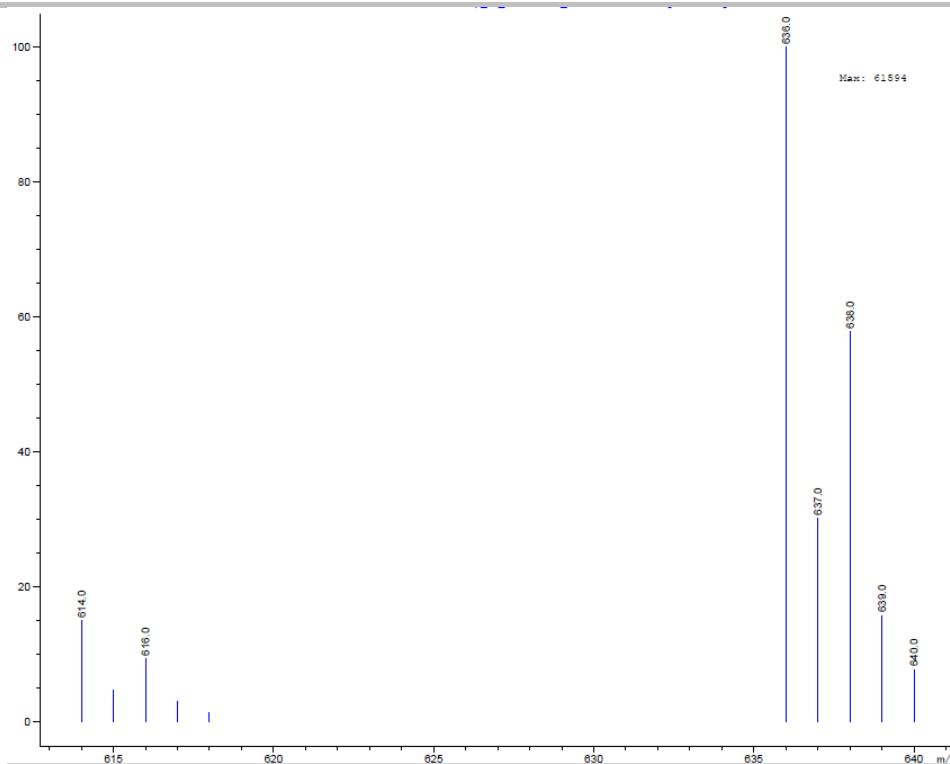


Figure S22. (-)-ESI-MS for $[(L4)Cu]^{2+}$. m/z : 636.0 $[(L4)Cu+Na^+]$, 614.0 $[(L4)Cu+H^+]$.

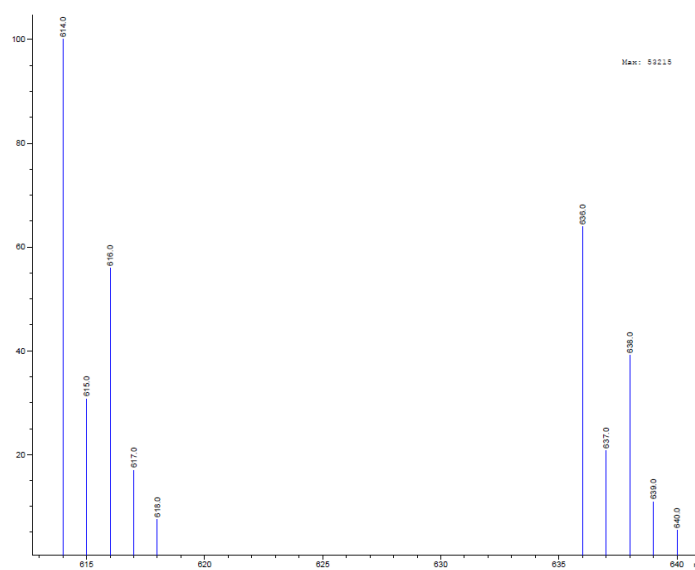


Figure S23. (-)-ESI-MS for $[(L5)Cu]^{2+}$. m/z : 636.0 $[(L5)Cu+Na^+]$, 614.0 $[(L5)Cu+H^+]$.

SUPPORTING INFORMATION

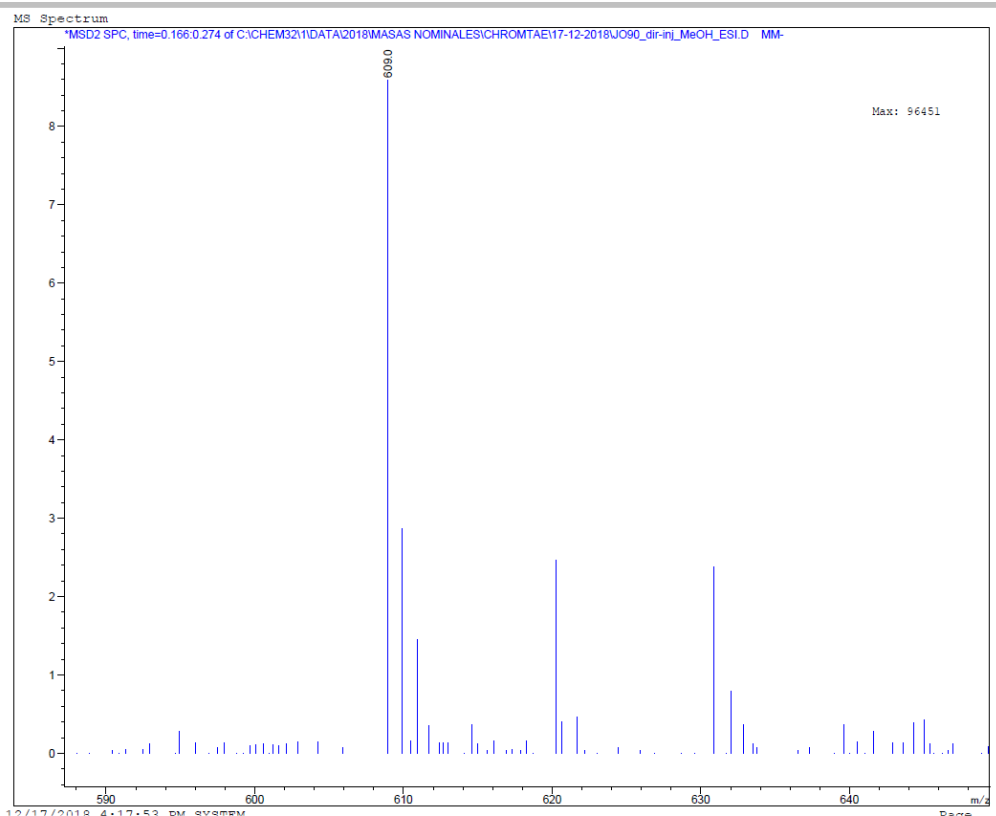


Figure S24. (-)-ESI-MS for $[(L6)Cu]^{2+}$. $m/z = 609.0$ $[(L6)Cu-Na]^+$.

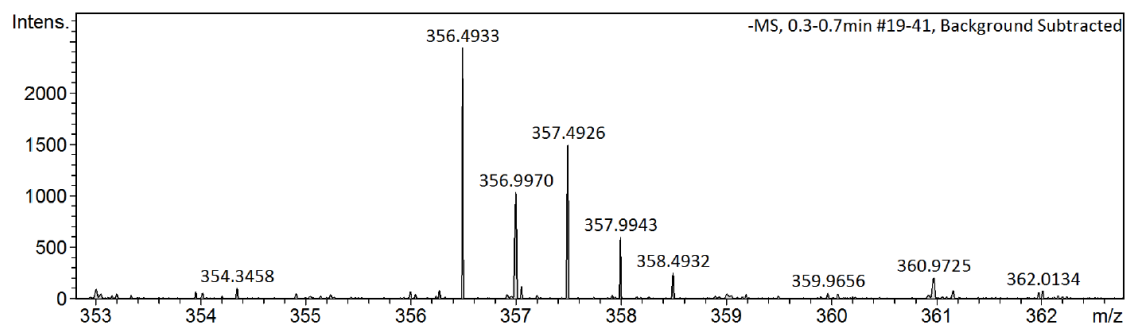


Figure S25. (-)-ESI-HRMS for $[(L7)Cu]^{2+}$. Calcd for $[(L7)Cu]^{2+}$, $(C_{32}H_{18}CuN_4O_8S_2)$: 356.4937, found 356.4933.

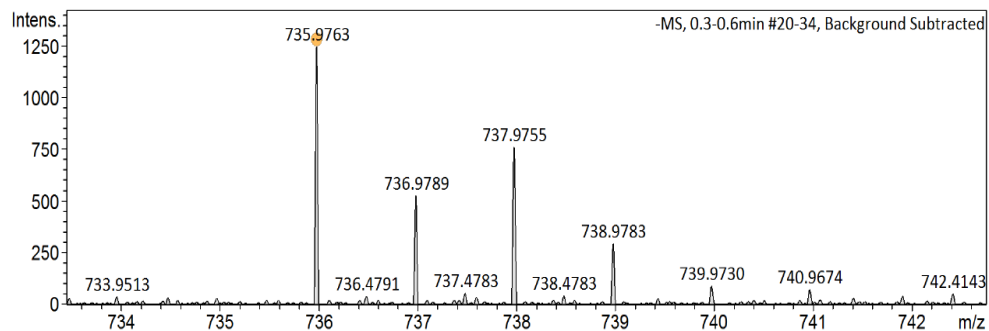


Figure S26. (-)-ESI-HRMS for $[(L8)Cu]^{2+}$. Calcd for $[(L8)Cu+Na]^+$, $(C_{32}H_{18}CuN_4O_8S_2Na)$: 735.9765, found 735.9763.

SUPPORTING INFORMATION

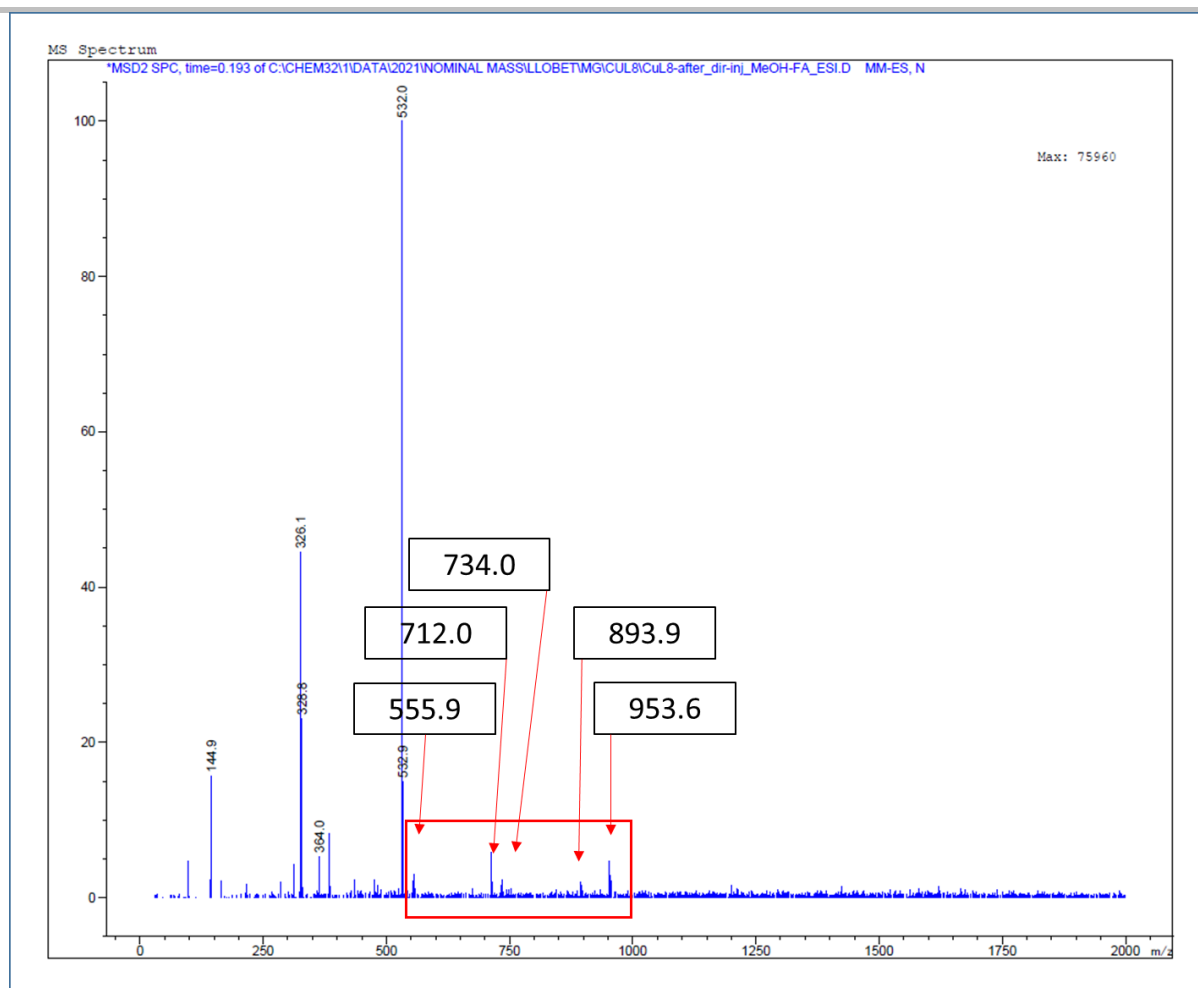


Figure S27. (-)-ESI-MS spectrum for $[(L8)Cu]^{2+}$ after a CPE experiment at 1.45 V vs. NHE in 0.01 M phosphate buffer pH 11.6 during 12 h. The spectrum shows peaks at $m/z=555.9, 712.0, 734.0, 893.9$ and 953.6 can be potentially assigned to molecular Cu degradation products based on the isotopic cracking pattern. The peak at $m/z=712.0$ is assigned in Figure S28.

SUPPORTING INFORMATION

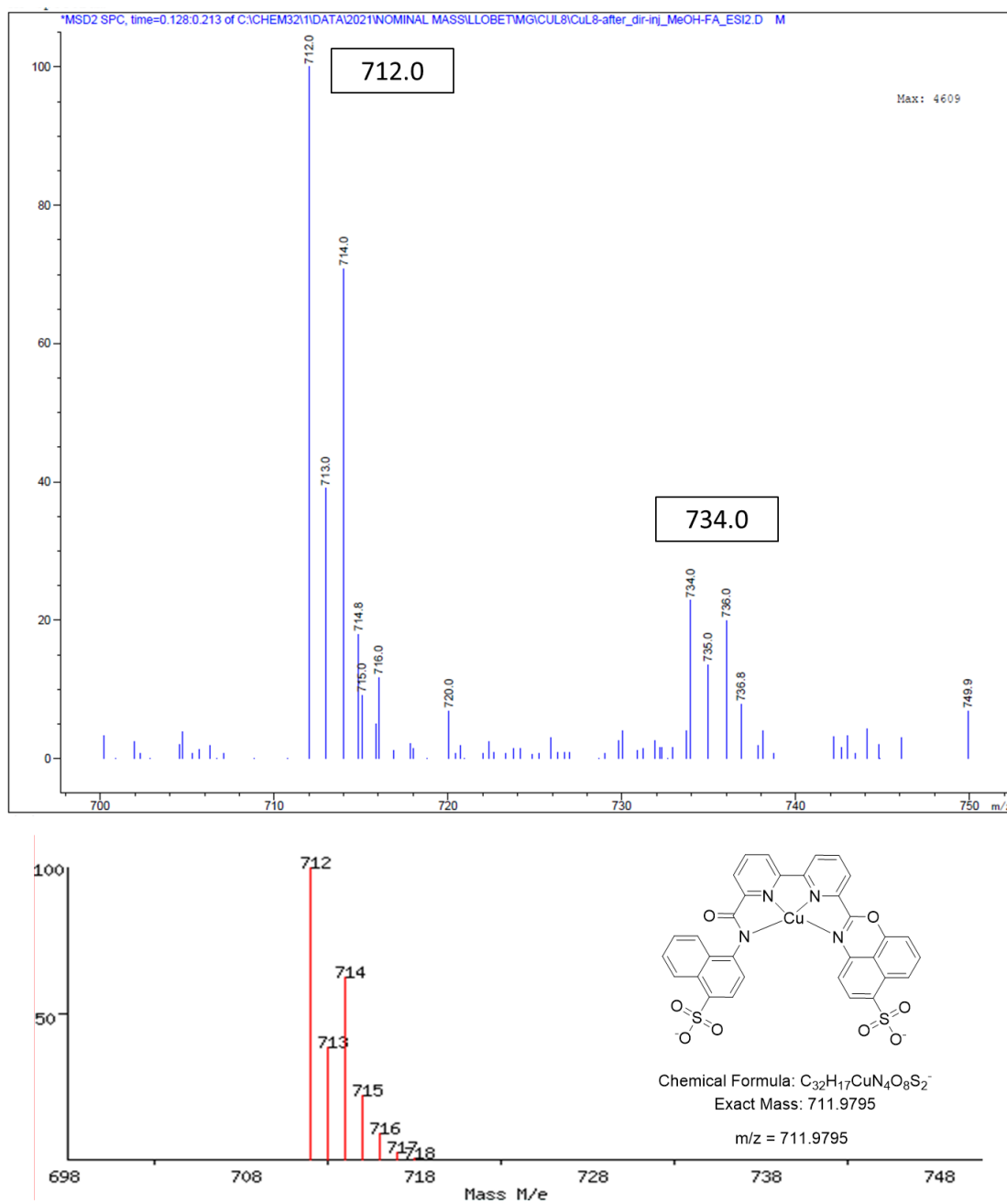


Figure 28. Top, zoom in the m/z= 700-750 region of (-)-ESI-MS spectrum from Figure S27. Bottom, simulation and molecular assignment of the m/z =712 peak.

SUPPORTING INFORMATION

UV-Vis spectroscopy

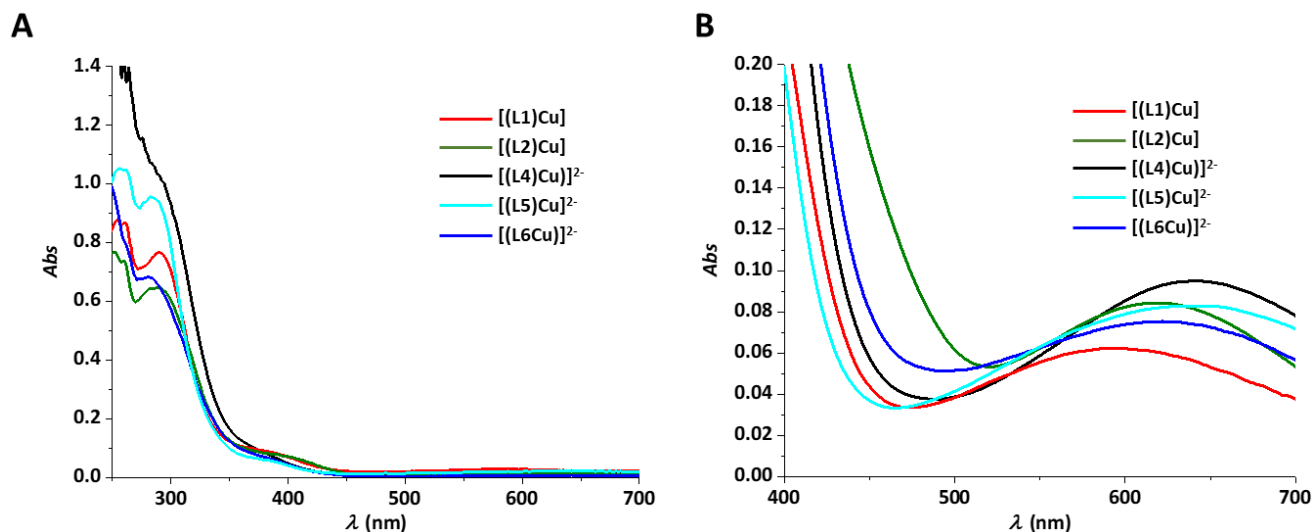


Figure S29. (A) UV-vis spectra of [(L1)Cu] (red line), [(L2)Cu] (green line), [(L4)Cu]²⁻ (black line), [(L5)Cu]²⁻ (light blue line) and [(L6)Cu]²⁻ (blue line). (B) UV-vis spectra enlargement of the visible region. Conditions: [Complex] = ~0.1 mM in 0.1 M phosphate buffer (pH 11.6). **Note:** in order to solve low solubility issues with [(L1)Cu] and [(L2)Cu], 40% of TFE was added to the mixture used for UV-Vis measurements.

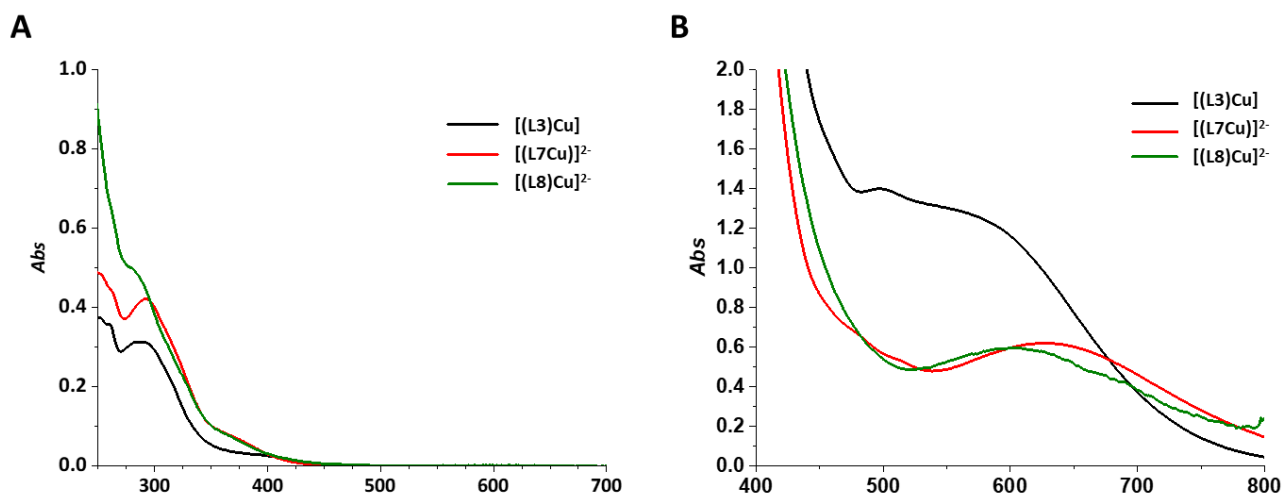


Figure S30. (A) UV-vis spectra of [(L3)Cu] (black line), [(L7)Cu]²⁻ (red line) and [(L8)Cu]²⁻ (green line). (B) UV-vis spectra enlargement of the visible region. Conditions: [Complex] = 0.04 mM (A, $d = 1$ mm) and 1 mM (B, $d = 10$ mm) in 0.1 M phosphate buffer (pH 11.6). **Note:** in order to solve low solubility issues with [(L3)Cu] 40% of TFE was added to the mixture used for UV-Vis measurements.

SUPPORTING INFORMATION

X-Ray Crystallography

Crystal preparation:

Table S1. Summary of the conditions employed for the crystallization of the complexes studied in this work.

Compound	Methodology
$\text{H}_2\text{L5}^{2-}$	Crystals were grown in water by slow diffusion of EtOH.
$[(\text{L1})\text{Cu}]$	Crystals were grown by slow evaporation in MeOH/Toluene.
$[(\text{L2})\text{Cu}]$	Crystals were grown in MeOH by slow diffusion of Et ₂ O.
$[(\text{L3})\text{Cu}]$	Crystals were grown by slow evaporation in MeOH/Toluene.
$[(\text{L4})\text{Cu}]^{2-}$	Crystals were grown in water by slow diffusion of acetone.
$[(\text{L5})\text{Cu}]^{2-}$	Crystals were grown in water by slow diffusion of acetone.
$[(\text{L6})\text{Cu}]^{2-}$	Crystals were grown in MeOH by slow diffusion of layered isopropanol.
$[(\text{L7})\text{Cu}]^{2-}$	Crystals were grown in water by slow diffusion of acetone.

The crystals were selected using a Zeiss stereomicroscope using polarized light and prepared under inert conditions immersed in perfluoropolyether as protecting oil for manipulation.

Data collection: Crystal structure determinations for samples were carried out using an Apex DUO Kappa 4-axis goniometer equipped with an APEX 2 4K CCD area detector, a Microfocus Source E025 luS using MoK α radiation, a Quazar MX multilayer Optics as monochromator and an Oxford Cryosystems low temperature device Cryostream 700 plus ($T = -173$ °C). Full-sphere data collection was used with ω and φ scans. *Programs used:* Bruker Device: Data collection APEX-2,⁷ data reduction Bruker Saint⁸ V1.60A and absorption correction SADABS.⁹

Structure Solution and Refinement: Crystal structure solution was achieved using the computer program SHELXT.¹⁰ Visualization was performed with the program SHELXle.¹¹ Missing atoms were subsequently located from difference Fourier synthesis and added to the atom list. Least-squares refinement on F^2 using all measured intensities was carried out using the program SHELXL 2015.¹² All non-hydrogen atoms were refined including anisotropic displacement parameters.

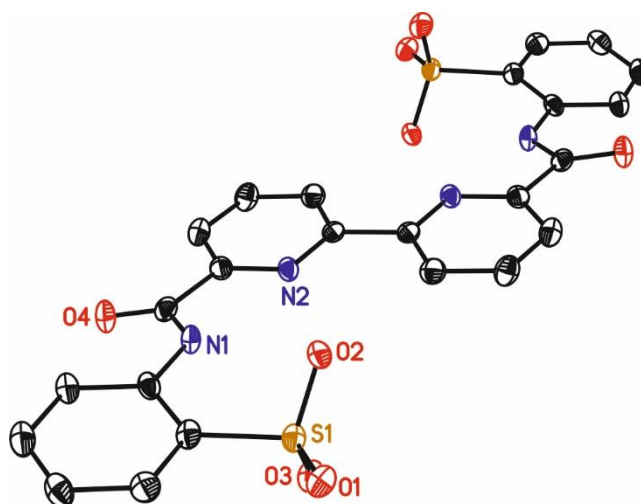


Figure S31. ORTEP representation of $\text{H}_2\text{L5}^{2-}$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; S, yellow.

SUPPORTING INFORMATION

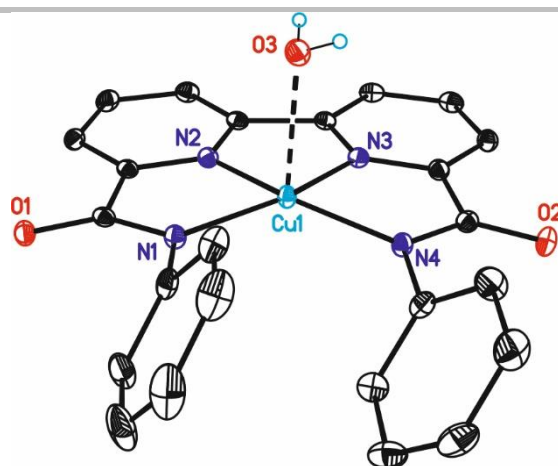


Figure S32. ORTEP representation of $[(L1)Cu]$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; Cu, light blue. The ORTEP drawing shows a contact (2.38 Å) with an oxygen from a H_2O molecule in the apical position.

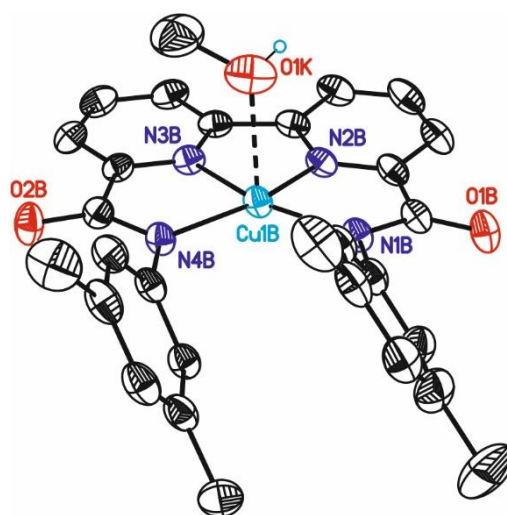


Figure S33. ORTEP representation of $[(L2)Cu]$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; Cu, light blue. The ORTEP drawing shows a contact (2.47 Å) with an oxygen from a $MeOH$ molecule in the apical position.

SUPPORTING INFORMATION

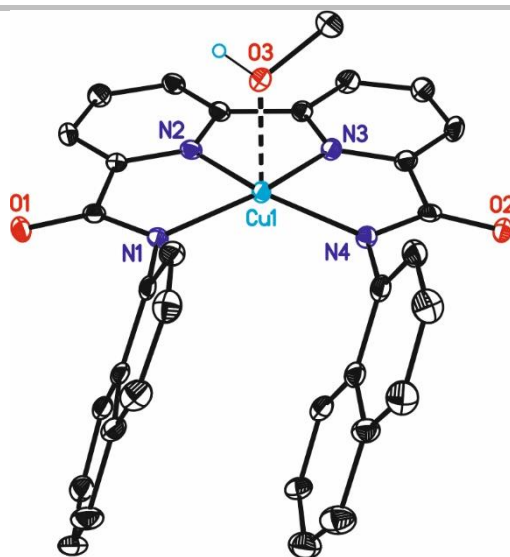


Figure S34. ORTEP representation of $[(L3)Cu]$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; Cu, light blue. The ORTEP drawing shows a contact (2.25 Å) with an oxygen from a MeOH molecule in the apical position.

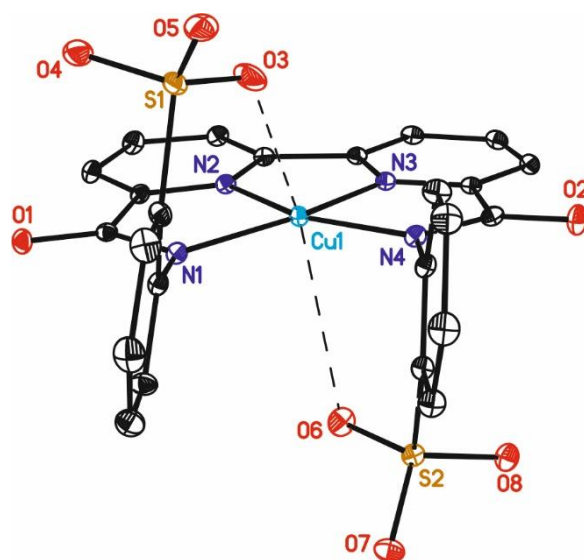


Figure S35. ORTEP representation of $[(L5)Cu]^{2-}$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; Cu, light blue; S, yellow. The ORTEP drawing shows a contact (2.61 Å) between an oxygen from the sulfonate moiety (O3) and the Cu center in the apical position.

SUPPORTING INFORMATION

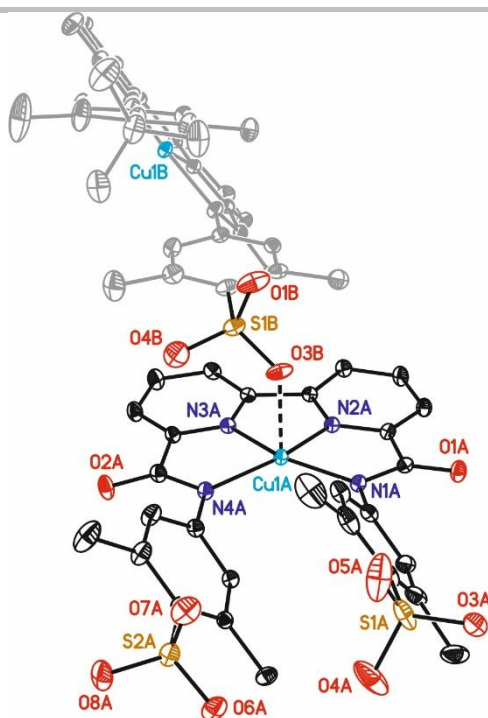


Figure S36. ORTEP representation of $[(L6)Cu]^{2+}$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, black; N, blue; O, red; Cu, light blue; S, yellow. The ORTEP drawing shows a contact (2.33 Å) with a sulfonate moiety from another $[(L6)Cu]^{2+}$ unit in the apical position (grey).

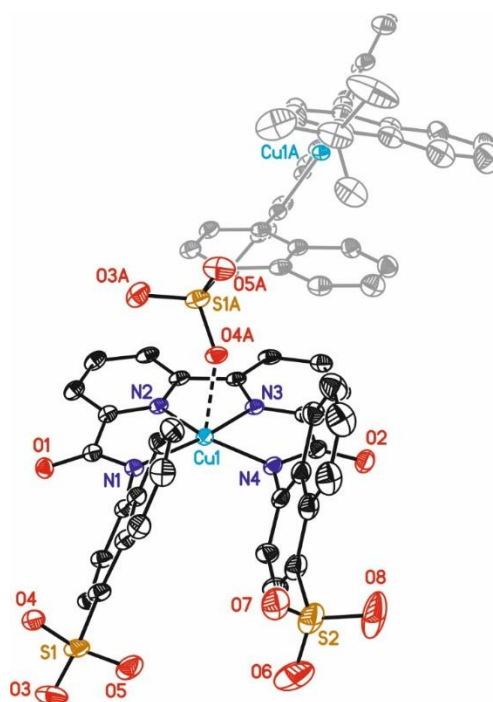


Figure S37. ORTEP representation of $[(L7)Cu]^{2+}$ at 50% probability level. The counter ions, solvent molecules and hydrogen atoms have been omitted for clarity. Color code: C, grey; N, blue; O, red; Cu, light blue; S, yellow. The ORTEP drawing shows a contact (2.38 Å) with a sulfonate moiety from another $[(L7)Cu]^{2+}$ unit in the apical position (grey).

SUPPORTING INFORMATION

Table S2. Selected interatomic bond distances.

	Cu-N _{bpy1}	Cu-N _{bpy2}	Cu-N _{amide1}	Cu-N _{amide2}	Cu-X _{apical}	N _{amide1} -C _{ar1}	N _{amide2} -C _{ar2}
[(L1)Cu]	1.9498(13)	1.9573(13)	1.9833(13)	2.0158(13)	2.3831(14)	1.422(2)	1.433(2)
[(L2)Cu]	1.9522(17)	1.9546(17)	1.9855(17)	1.9913(17)	2.423(3)	1.425(3)	1.423(2)
	1.9475(17)	1.9517(17)	1.9759(18)	1.9828(17)	2.475(2)	1.429(2)	1.421(2)
[(L3)Cu]	1.950(2)	1.9597(19)	1.9920(19)	2.012(5)	2.2455(18)	1.434(3)	1.421(6)
[(L4)Cu] ²⁻	1.9569(14)	1.9508(14)	2.0085(14)	2.0152(14)	2.3069(14)	1.415(2)	1.418(2)
[(L5)Cu] ²⁻	1.9515(14)	1.9473(13)	1.9632(14)	1.9868(14)	2.6110(13)	1.418(2)	1.424(2)
[(L6)Cu] ²⁻	1.9674(17)	1.9602(17)	1.9961(17)	2.0163(17)	2.324(6)	1.419(3)	1.414(3)
	1.9615(17)	1.9554(18)	1.9848(18)	2.0058(18)	2.339(3)	1.420(3)	1.421(3)
[(L7)Cu] ²⁻	1.943(2)	1.964(2)	2.012(2)	1.979(2)	2.382(2)	1.431(3)	1.421(4)

Table S3. Selected interatomic angles.

	N _{bpy1} -Cu-N _{bpy2}	N _{amide1} -Cu-N _{bpy1}	N _{bpy2} -Cu-N _{amide2}	N _{amide1} -Cu-N _{amide2}	N _{bpy1} -Cu-N _{amide2}	N _{bpy2} -Cu-N _{amide1}
[(L1)Cu]	78.02(5)	80.34(5)	80.94(5)	118.99(5)	156.79(6)	157.46(6)
[(L2)Cu]	77.41(7)	80.82(7)	81.09(6)	119.66(7)	158.19(7)	156.42(7)
	77.96(7)	81.09(7)	80.83(7)	119.73(7)	157.90(7)	158.95(7)
[(L3)Cu]	77.99(8)	80.74(8)	80.77(15)	118.44(14)	157.09(19)	156.92(8)
[(L4)Cu] ²⁻	78.27(6)	80.12(6)	80.74(6)	119.69(6)	158.81(6)	155.51(6)
[(L5)Cu] ²⁻	78.49(6)	80.16(6)	80.56(6)	120.78(6)	159.03(6)	158.00(6)
[(L6)Cu] ²⁻	77.43(7)	80.58(7)	80.53(7)	120.23(7)	156.08(7)	157.57(7)
	77.76(7)	80.80(7)	80.58(7)	119.75(7)	156.63(8)	158.13(7)
[(L7)Cu] ²⁻	77.78(10)	81.30(10)	80.62(10)	118.57(10)	157.37(10)	157.09(10)

SUPPORTING INFORMATION

Electron Paramagnetic Resonance (EPR)

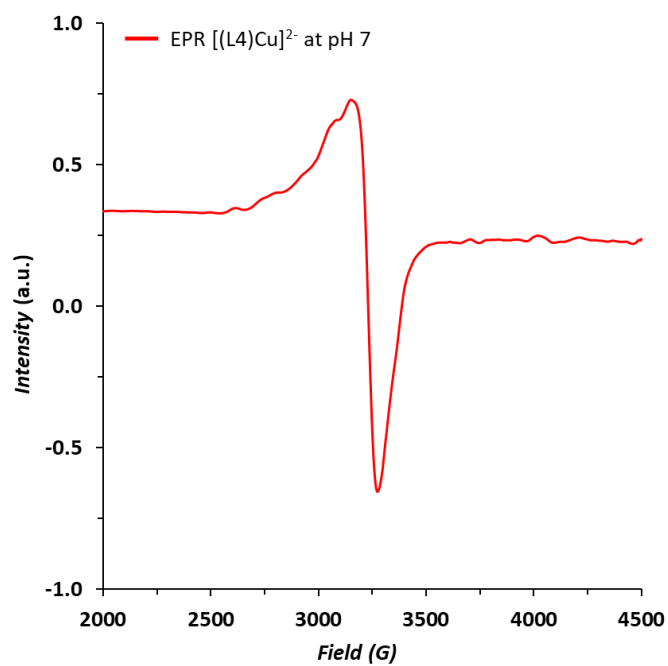


Figure 38. EPR spectra measured for $[(L4)Cu]^{2+}$ in 0.1 M phosphate buffer (pH 7). The spectrum shows the characteristic EPR signal for a Cu^{II} complex with $g_{||} = 2.074$.

Electrochemical behavior in organic solvents

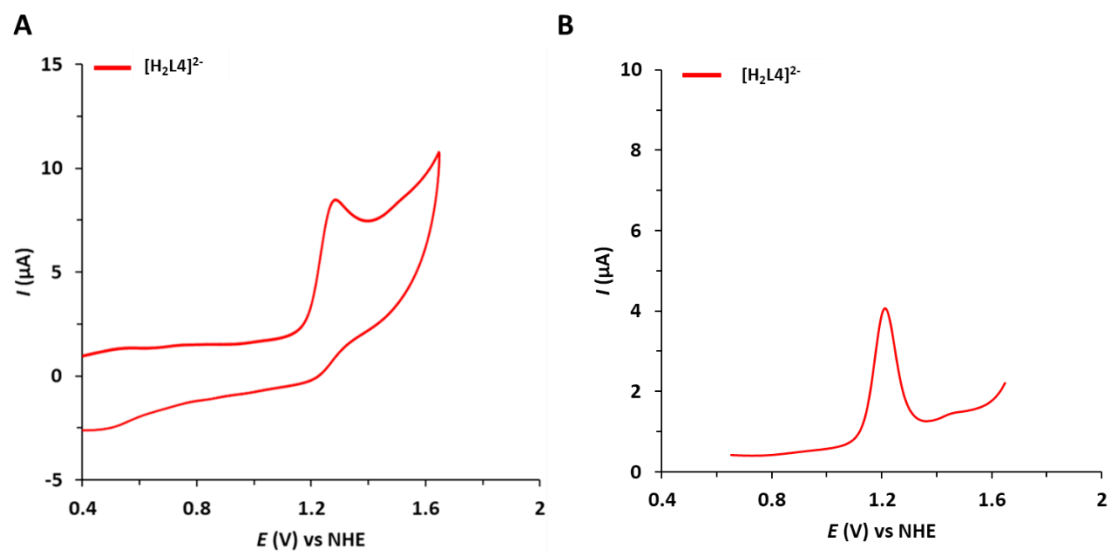


Figure S39. (A) Cyclic Voltammogram of a 1 mM solution of $\text{H}_2\text{L4}^{2-}$ in DMF containing 0.1 M TBAPF₆ at a scan rate of 100 mV/s (B) DPV experiments of a 1 mM solution of $\text{H}_2\text{L4}^{2-}$ in DMF containing 0.1 M TBAPF₆.

SUPPORTING INFORMATION

Electrochemical behavior in water

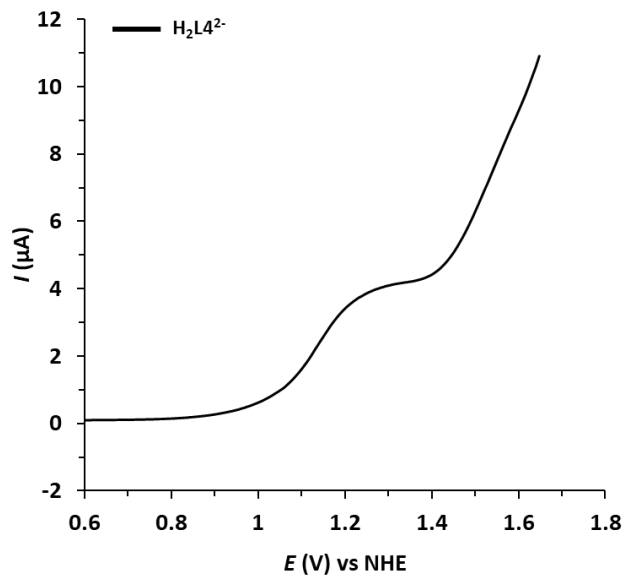


Figure S40. DPV experiment of a 1 mM solution of $[\text{H}_2\text{L4}]^{2-}$ in 0.1 M phosphate buffer pH 11.6. Conditions: scan rate of 100 mV/s, GC as working electrode.

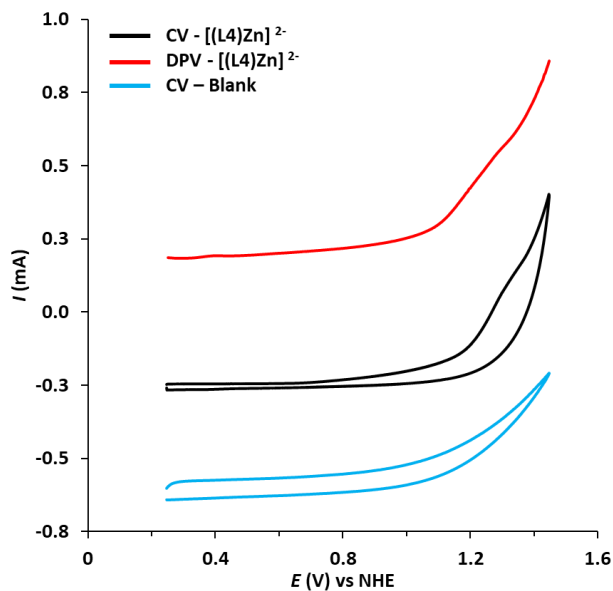


Figure S41. CV (black line) and DPV (red line) experiments of 1 mM solution of $[(\text{L4})\text{Zn}]^{2-}$ in a 0.1 M phosphate buffer pH 11.6. Blue line corresponds to a blank with no complex. Conditions: scan rate of 100 mV/s, GC-paper as working electrode.

SUPPORTING INFORMATION

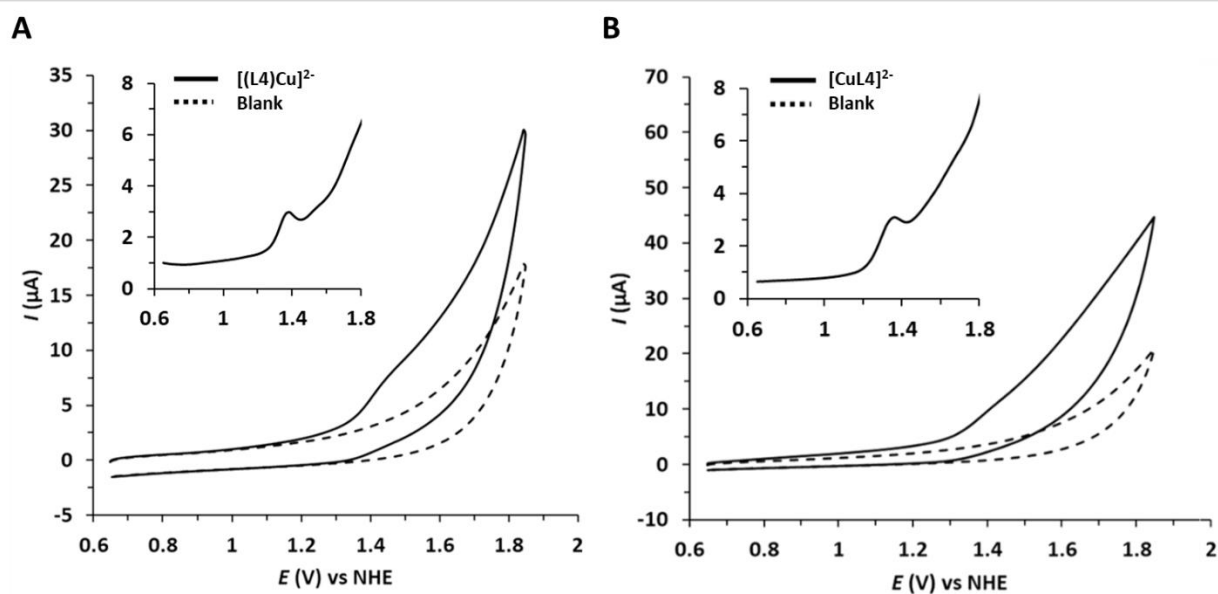


Figure S42. Cyclic Voltammograms of a 1 mM solution of $[(L4)Cu]^{2-}$ at pH 7 (A) and 11.6 (B) in 0.1 M phosphate buffer solutions. *Inset:* Differential Pulse Voltammograms for $[(L4)Cu]^{2-}$ at each pH value. Dashed black line corresponds to a blank with no catalyst. Conditions: scan rate of 100 mV/s, GC as working electrode.

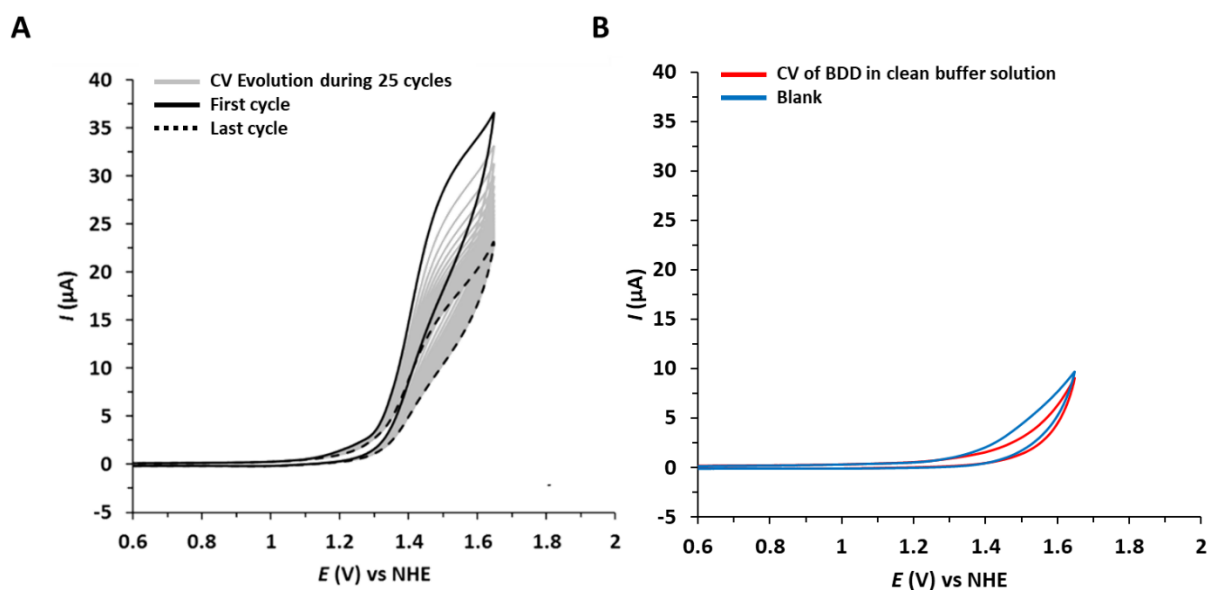


Figure S43. (A) Evolution of CV profile of a 1 mM solution of $[(L4)Cu]^{2-}$ in 0.1 M phosphate buffer (pH 11.6) during 25 cycles. (B) CV measurement with a BDD electrode that has performed the previous 25 cycles of the complex (red line) or of a blank solution (blue line) immersed in a freshly-prepared 0.1 M phosphate buffer pH 11.6. Conditions: scan rate of 100 mV/s, BDD disk as working electrode.

SUPPORTING INFORMATION

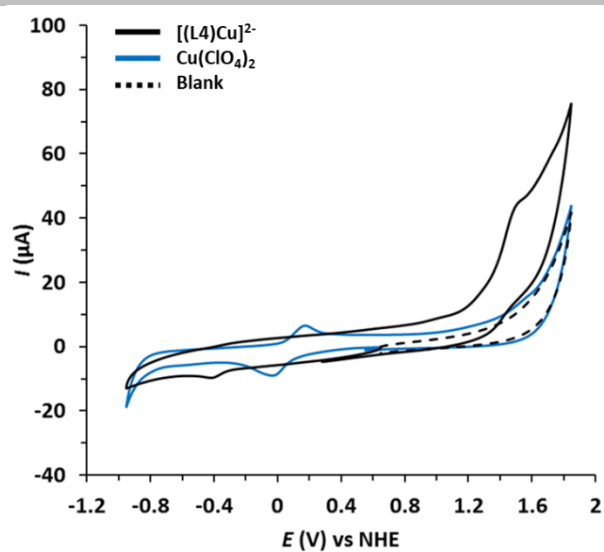


Figure S44. Cyclic Voltammograms of a 1 mM solution of $[(L4)Cu]^{2-}$ in 0.1 M phosphate buffer pH 11.6, showing the electrocatalytic response of the complex (solid black line), of a 1 mM solution of $Cu(ClO_4)_2$ (solid blue line) and the blank (dashed line). Conditions: scan rate of 100 mV/s, BDD disk as working electrode.

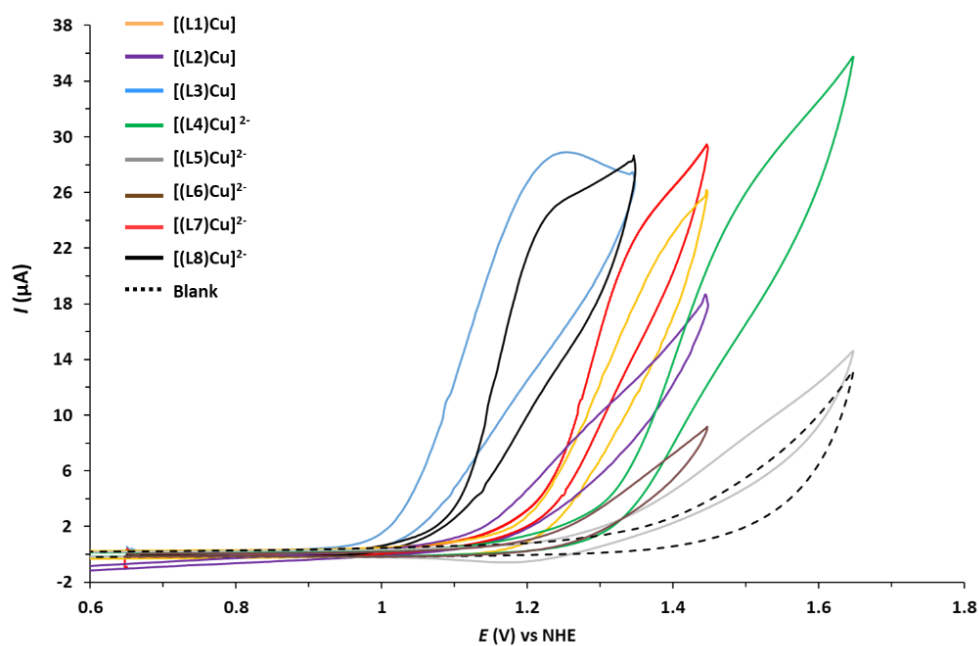


Figure S45. Cyclic Voltammograms for the complexes studied in this work in 0.1 M phosphate buffer pH 11.6. Note: In case of complexes $[(L1-3)Cu]$ the experiments were performed in a mixture of 0.1 M phosphate buffer/TFE (6:4) to fully solubilize the complexes. $[Complex] = \sim 1$ mM. BDD disk as working electrode.

SUPPORTING INFORMATION

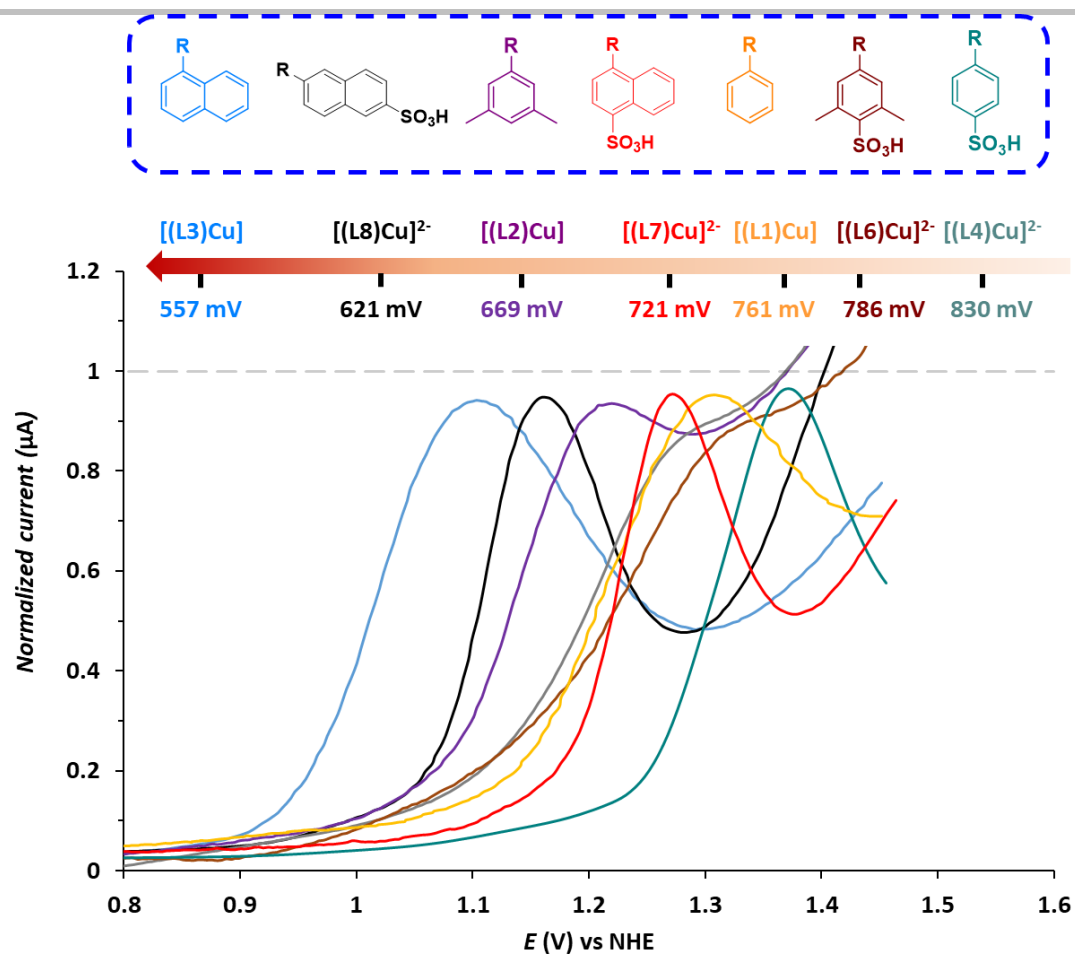
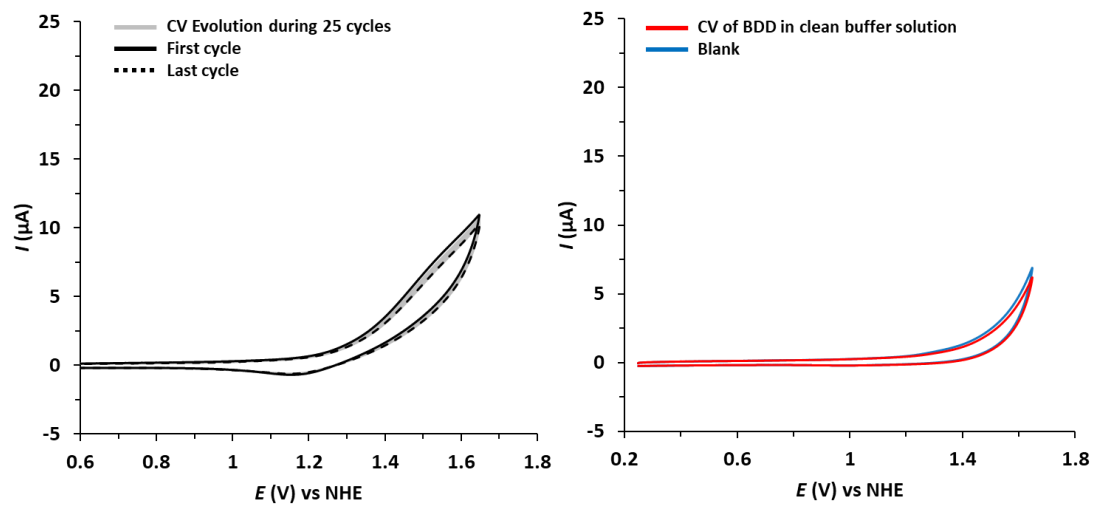


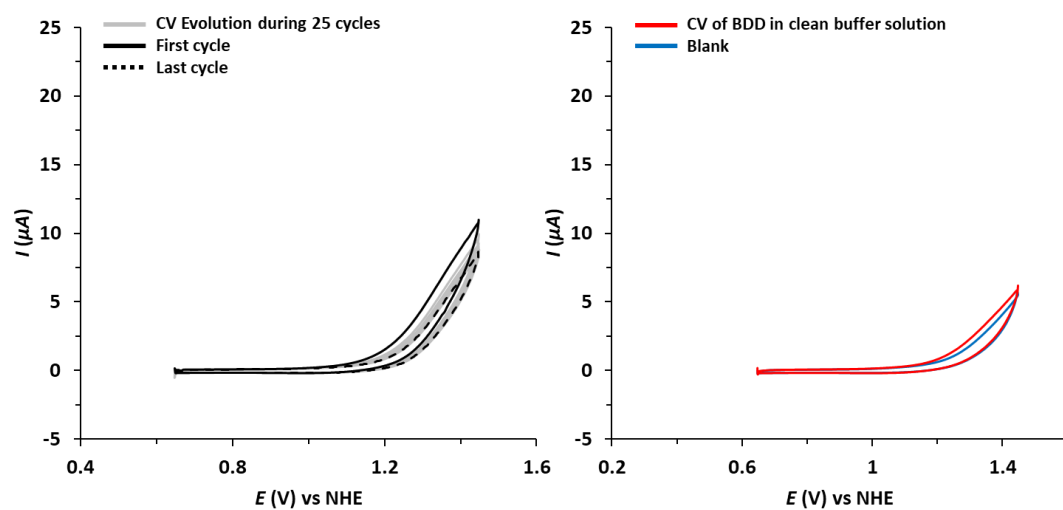
Figure S46. DPV experiments for the complexes studied in this work in 0.1 M phosphate buffer pH 11.6. The current (μA) has been normalized between 0 and 1 values for comparison purposes. *Note:* In case of complexes $[(\text{L}1-3)\text{Cu}]$ the experiments were performed in a mixture of 0.1 M phosphate buffer/TFE (6:4) to fully solubilize the complexes. $[\text{Complex}] = \sim 1$ mM. BDD disk as working electrode. *Top inset:* Influence of the substituents in $[(\text{LN})\text{Cu}]^{n-}$ complexes on the η (mV) of the water oxidation reaction.

SUPPORTING INFORMATION

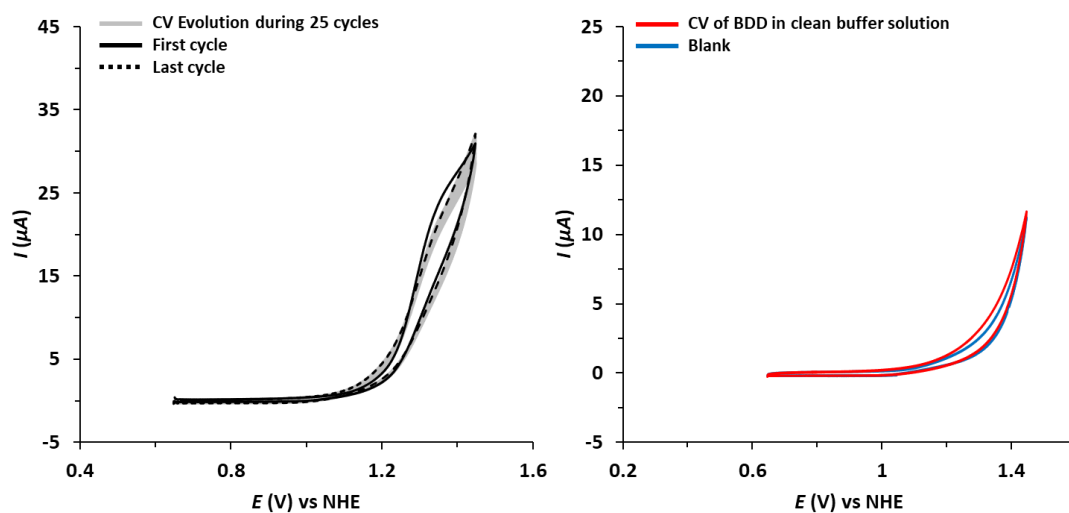
A



B



C



SUPPORTING INFORMATION

D

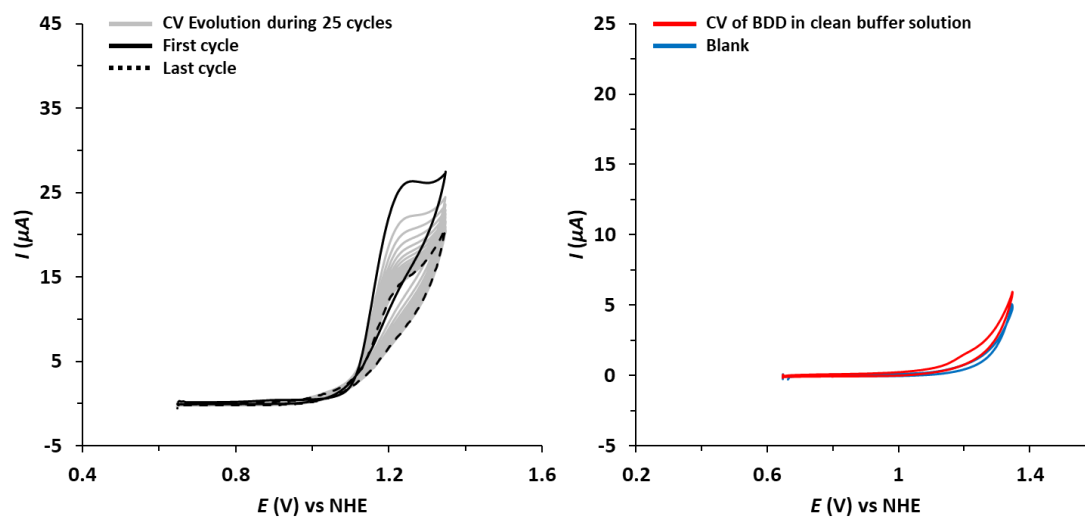


Figure S47. Evolution of CV profile for the complexes in 0.1 M phosphate buffer pH 11.6 after 25 cycles (Left) and CV measurement with a BDD electrode that has performed the previous 25 cycles of the complex (red line) or of a blank solution (blue line) immersed in a freshly-prepared catalyst-free 0.1 M phosphate buffer pH 11.6 (Right). Code: (A) $[(\text{L}5)\text{Cu}]^{2+}$, (B) $[(\text{L}6)\text{Cu}]^{2+}$, (C) $[(\text{L}7)\text{Cu}]^{2+}$ and (D) $[(\text{L}8)\text{Cu}]^{2+}$. Conditions: scan rate of 100 mV/s, $[\text{Complex}] = 1$ mM. BDD as working electrode.

SUPPORTING INFORMATION

Determination of the kinetic constant. Foot of the Wave Analysis (FOWA).

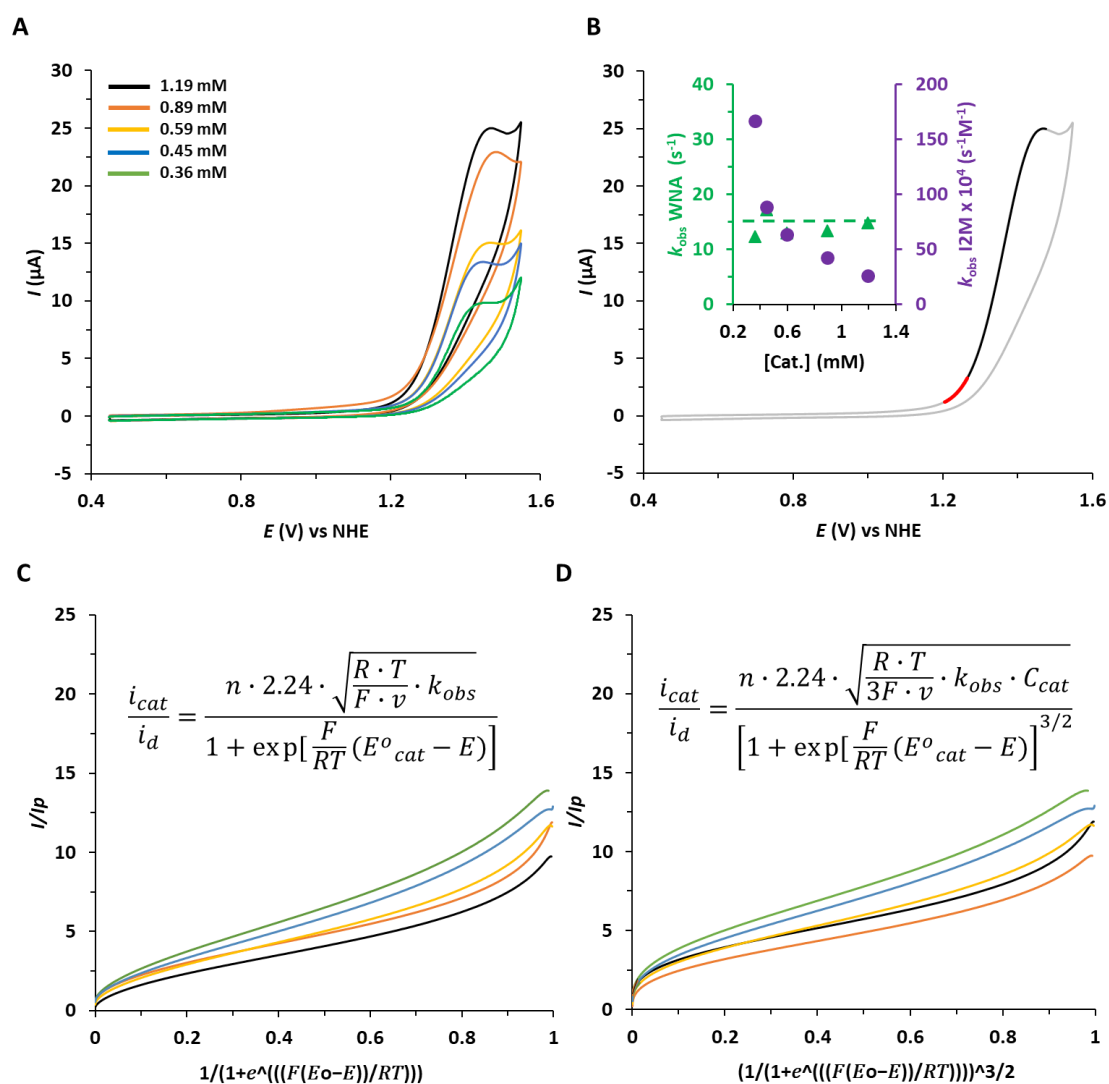
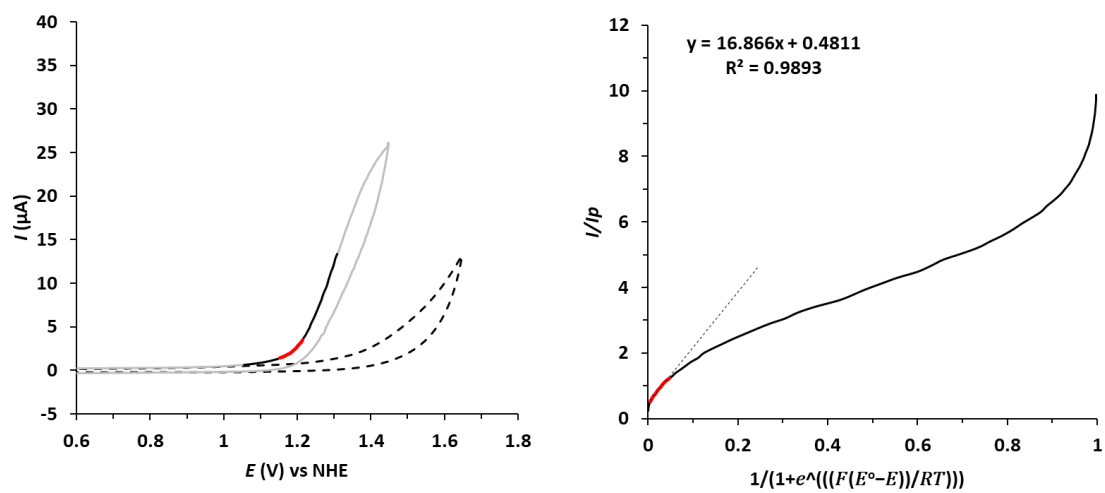


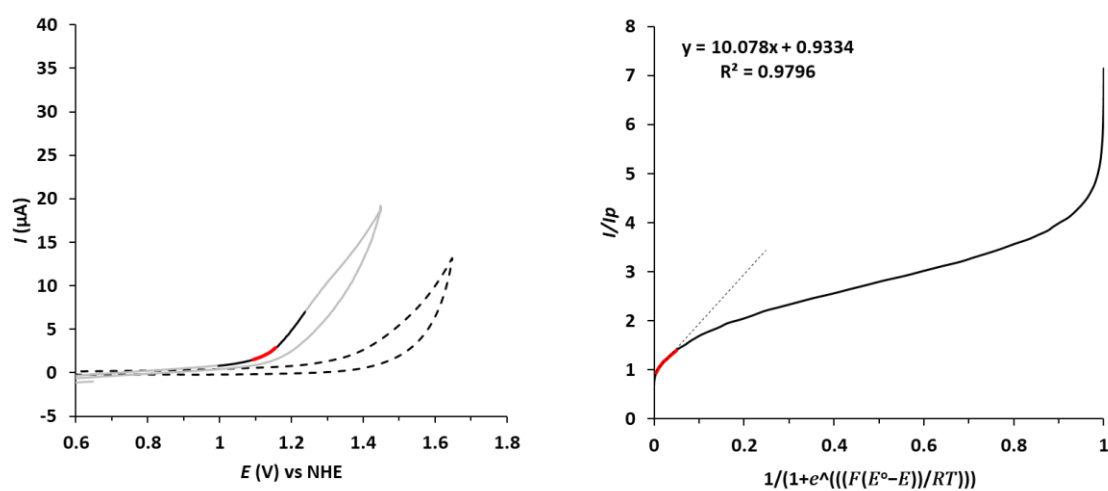
Figure S48. (A) CVs of $[(L4)Cu]^{2+}$ at different concentrations in 0.1 M phosphate buffer pH 11.6 at 100 mV/s scan rate. (B) CV of a 1.19 mM solution of $[(L4)Cu]^{2+}$ in 0.1 M phosphate buffer pH 11.6 at 100 mV/s scan rate (grey line), experimental data used for FOWA analysis (black line) and region used for the determination of k_{obs} (red line) by plotting i/i_p^0 vs. $1/(1+\exp[(F/RT)(E^o_{PQ}-E)])$ for a WNA mechanism and i/i_p^0 vs. $(1/(1+\exp[(F/RT)(E^o_{PQ}-E)]))^{3/2}$ for an I2M mechanism. Inset: Plot of k_{obs} vs. $[(L4)Cu]^{2+}$ assuming a WNA mechanism (green trace) or an I2M mechanism (purple trace). The linear trend indicates that the electrochemical oxidation of water to dioxygen is following a WNA mechanism.³ (C) FOWA region obtained by plotting i/i_p vs. $1/(1+\exp[(F/RT)(E^o_{PQ}-E)])$ for a WNA mechanism and (D) FOWA region obtained by plotting i/i_p vs. $(1/(1+\exp[(F/RT)(E^o_{PQ}-E)]))^{3/2}$ for an I2M mechanism used for the calculation of k_{obs} .

SUPPORTING INFORMATION

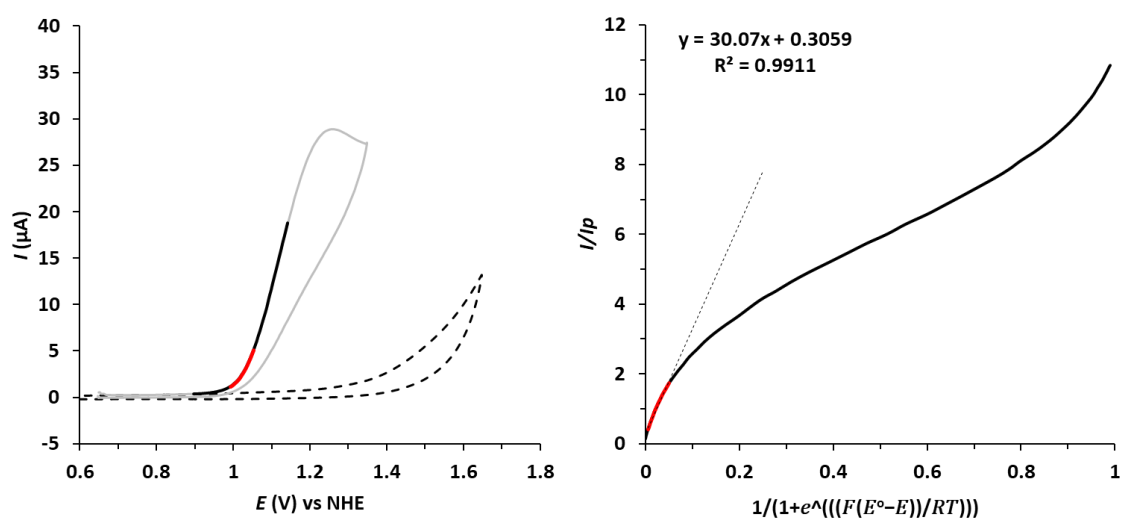
A



B

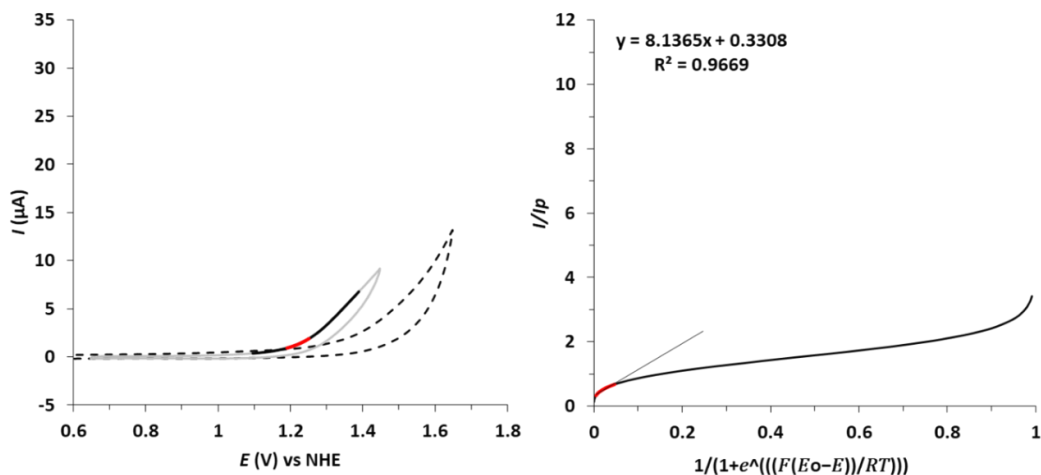


C

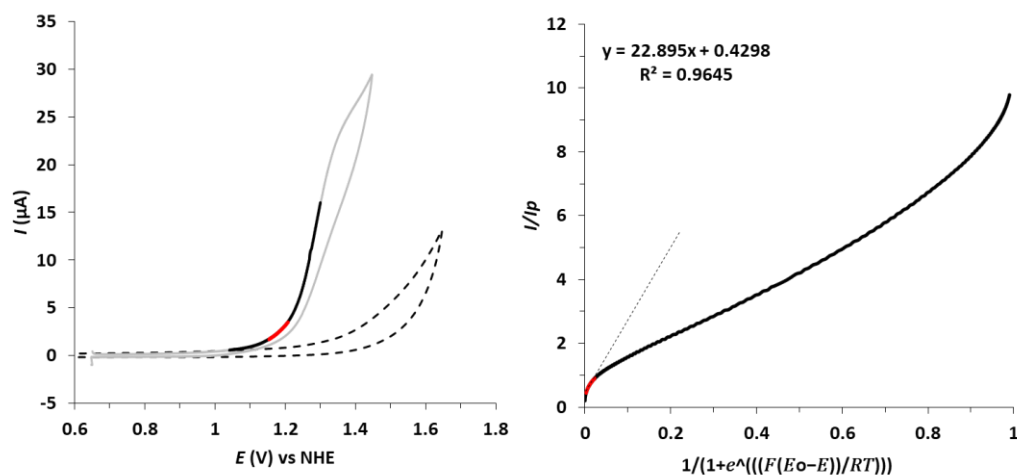


SUPPORTING INFORMATION

D



E



F

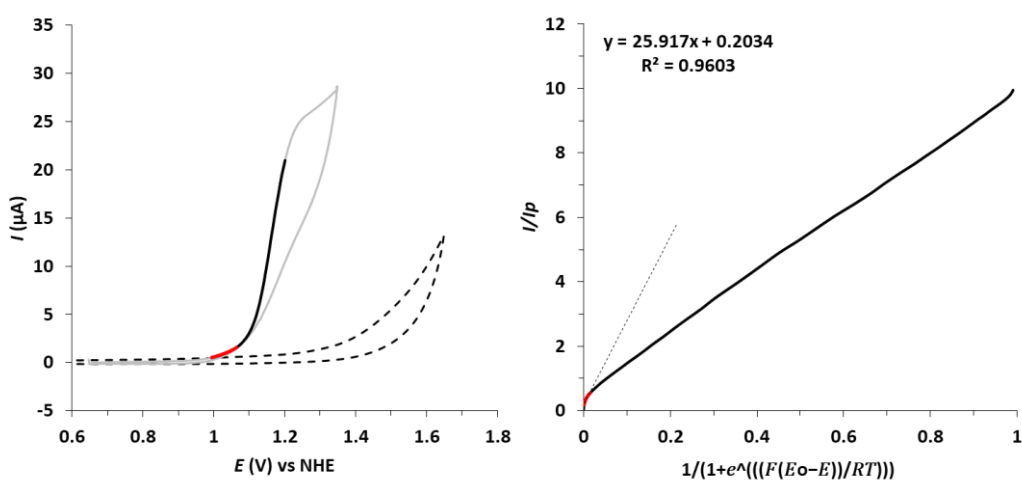


Figure S49. (Left) CVs of 1 mM solution of complexes in 0.1 M phosphate buffer pH 11.6 at 100 mV/s scan rate (grey line) and the blank (dashed line). Code: (A) $[(\text{L}1)\text{Cu}]$, (B) $[(\text{L}2)\text{Cu}]$, (C) $[(\text{L}3)\text{Cu}]$, (D) $[(\text{L}6)\text{Cu}]^{2+}$, (E) $[(\text{L}7)\text{Cu}]^{2+}$, (F) $[(\text{L}8)\text{Cu}]^{2+}$. Solid red line corresponds to the experimental data used for FOWA analysis and solid black line shows the region used for the determination of k_{obs} . (Right) FOWA obtained by plotting i/i_p^0 vs. $1/(1+\exp[(F/RT)(E^0_{\text{PQ}}-E)])$. Note: In case of complexes $[(\text{L}1-3)\text{Cu}]$ the experiments were performed in a mixture of 0.1 M phosphate buffer/TFE (6:4) to fully solubilize the complexes.

SUPPORTING INFORMATION

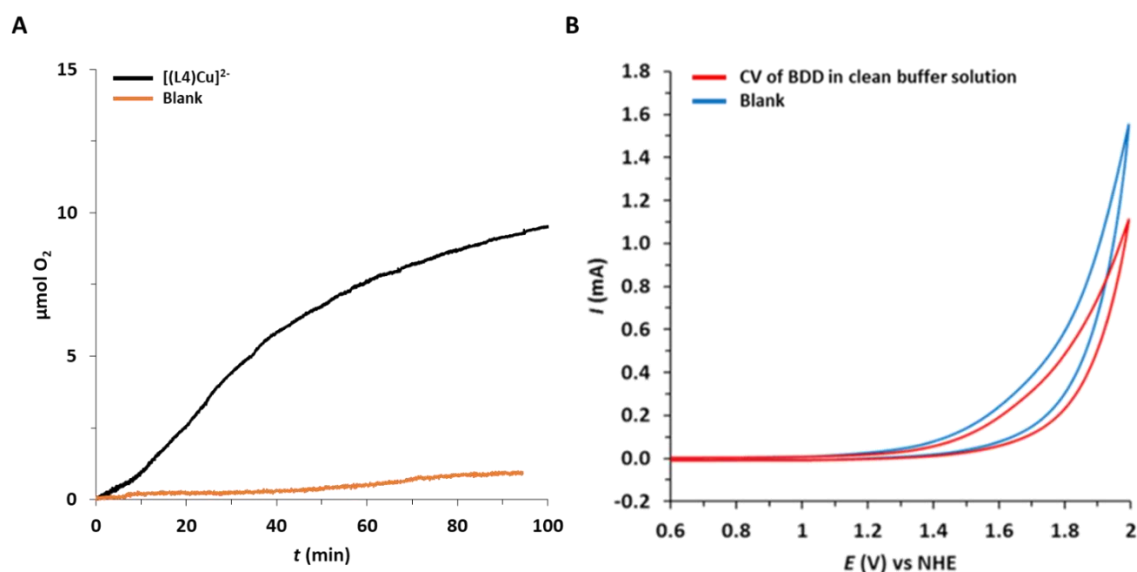
O₂ Evolution experiments.

Figure S50. (A) Oxygen evolution measurements given in $\mu\text{mol O}_2$ vs. time (min) for complex $[(\text{L4})\text{Cu}]^{2-}$ at 1.5 mM concentration ($V_{\text{total}} = 3$ mL) using a Clark probe electrode during a CPE at 1.6 V vs. NHE in 0.1 M phosphate buffer pH 11.6. Blank data in the absence of complex is shown in orange. (B) Comparison of CVs of a blank solution after a CPE at 1.6 V during 95 mins and of the mechanically polished BDD electrode under a blank solution, showing no catalytic response due to the presence of heterogeneous materials deposited onto the surface of the electrode. Conditions: scan rate of 100 mV/s, BDD disk as working electrode, Pt mesh counter electrode and AgCl as reference electrode. $Q = 4.58$ C, moles $e = 4.75 \times 10^{-5}$ mol, $\text{FE} = 76\%$. $\text{TON} = (\mu\text{mol O}_2)/(\mu\text{mol cat.}) = 1.86$. A TON of 58993 was obtained using the methodology developed by Savéant and co-workers based on the electroactive catalyst (i.e. catalyst in contact with the electrode).⁴

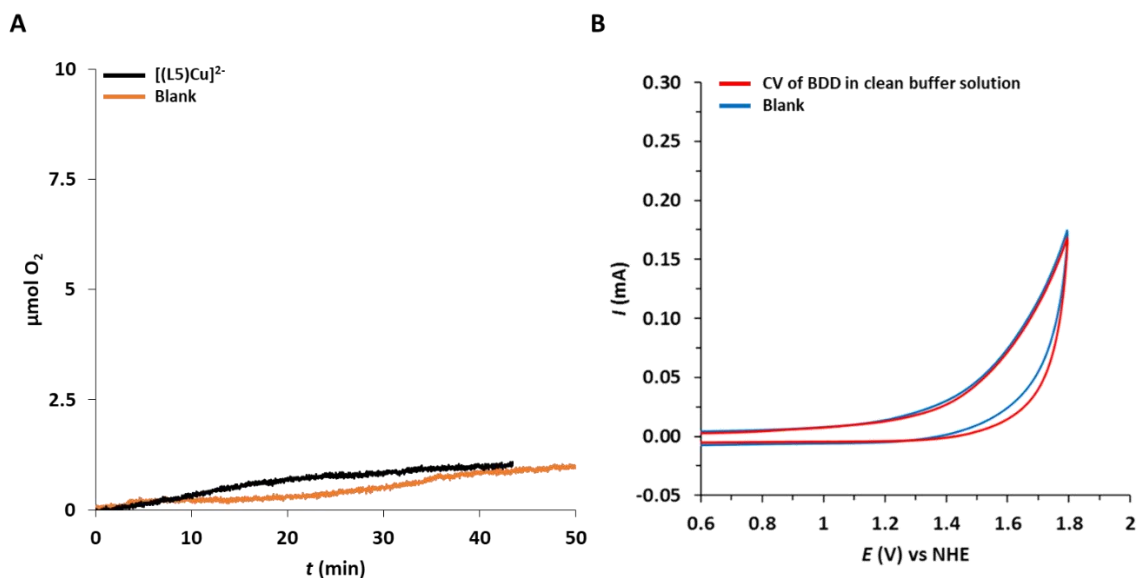


Figure S51. (A) Oxygen evolution measurements given in $\mu\text{mol O}_2$ vs. time (min) for complex $[(\text{L5})\text{Cu}]^{2-}$ at 1.5 mM concentration ($V_{\text{total}} = 3$ mL) using a Clark probe electrode during a CPE at 1.6 V vs. NHE in 0.1 M phosphate buffer pH 11.6. Blank data in the absence of complex is shown in orange. (B) Comparison of CVs of a blank solution after a CPE at 1.6 V during 43 min and of the mechanically polished BDD electrode under a blank solution, showing no catalytic response due to the presence of heterogeneous materials deposited onto the surface of the electrode. Conditions: scan rate of 100 mV/s, BDD disk as working electrode, Pt mesh counter electrode and AgCl as reference electrode.

SUPPORTING INFORMATION

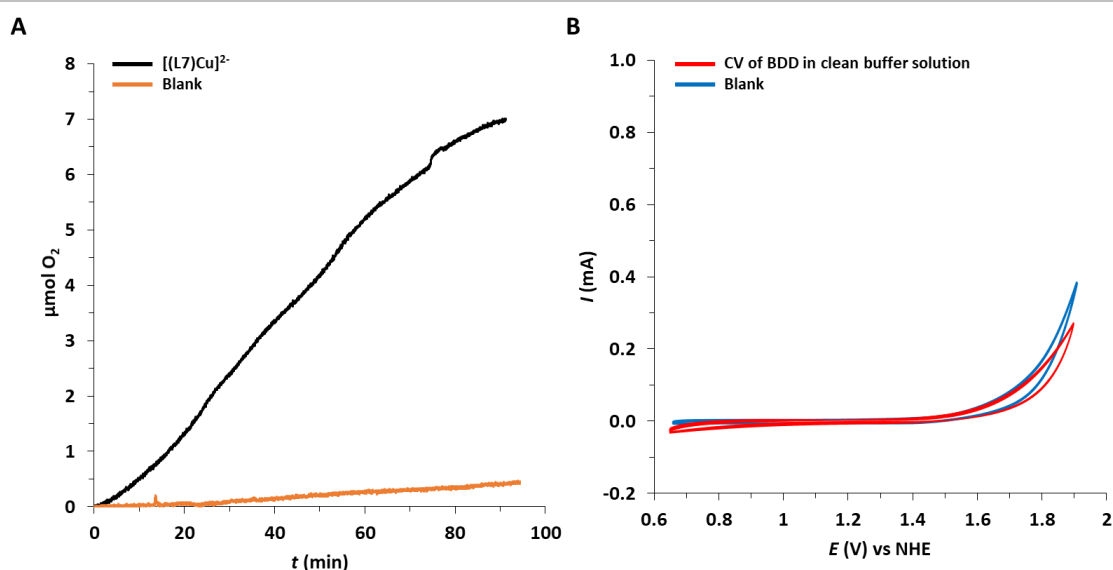


Figure S52. (A) Oxygen evolution measurements given in $\mu\text{mol O}_2$ vs. time (min) for complex $[(\text{L7})\text{Cu}]^{2+}$ at 1.5 mM concentration ($V_{\text{total}} = 3 \text{ mL}$) using a Clark probe electrode during a CPE at 1.45 V vs. NHE in 0.1 M phosphate buffer pH 11.6 V. Blank data in the absence of complex is shown in orange. (B) Comparison of CVs of a blank solution after a CPE at 1.45 V during 94 min and of the mechanically polished BDD electrode under a blank solution. Conditions: scan rate of 100 mV/s, BDD disk as working electrode, Pt mesh counter electrode and AgCl as reference electrode. $Q = 4.98 \text{ C}$, moles $e^- = 5.16 \times 10^{-5} \text{ mol}$, $\text{FE} = 52 \%$. $\text{TON} = (\mu\text{mol O}_2)/(\mu\text{mol cat.}) = 1.35$. TON of 130409 was obtained using the methodology developed by Savéant and coworkers based on the electroactive catalyst (i.e. catalyst in contact with the electrode).⁴

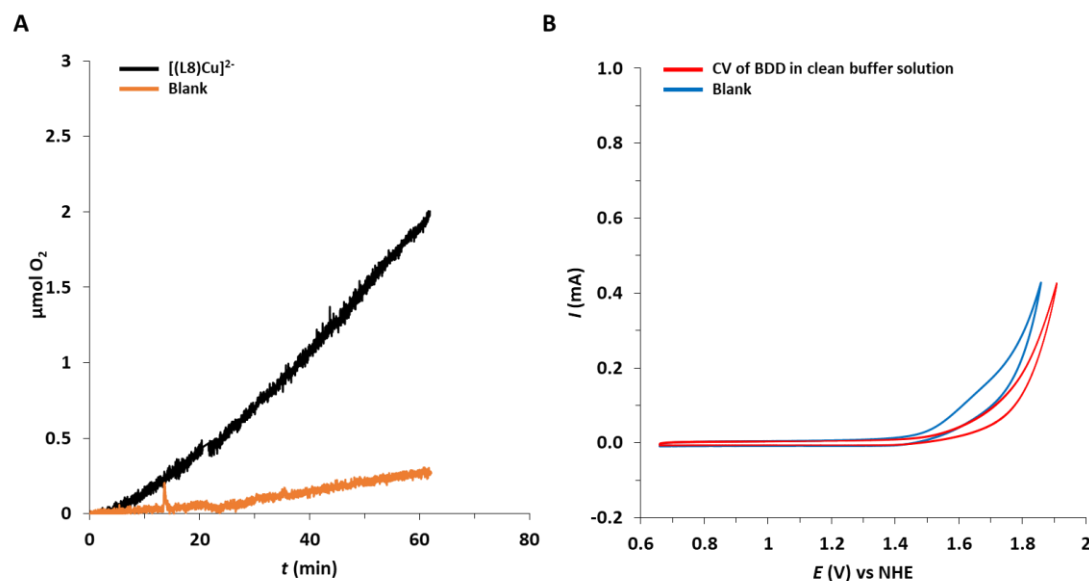
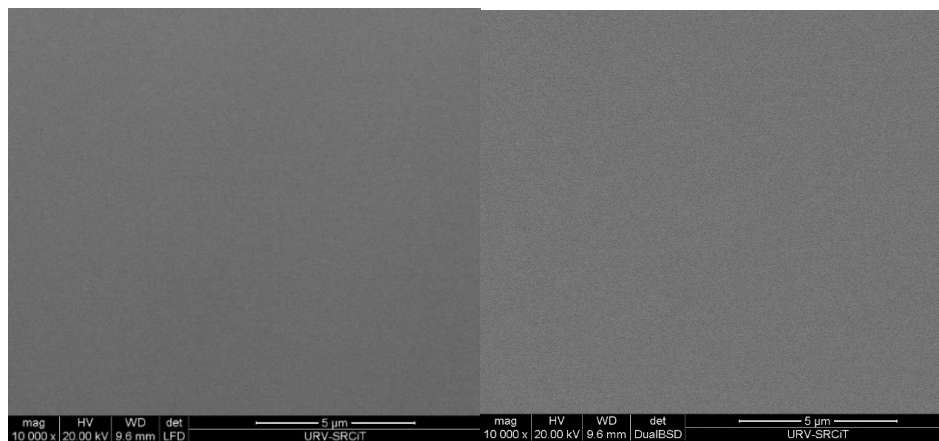


Figure S53. (A) Oxygen evolution measurements given in $\mu\text{mol O}_2$ vs. time (min) for complex $[(\text{L8})\text{Cu}]^{2+}$ at 1.5 mM concentration ($V_{\text{total}} = 3 \text{ mL}$) using a Clark probe electrode During a CPE at 1.45 V vs. NHE in 0.1 M phosphate buffer pH 11.6. Blank data in the absence of complex is shown in orange. (B) Comparison of CVs of a blank solution after a CPE at 1.45 V during 95 min and of the mechanically polished BDD electrode under a blank solution. Conditions: scan rate of 100 mV/s, BDD disk as working electrode, Pt mesh counter electrode and AgCl as reference electrode. $Q = 1.84 \text{ C}$, moles $e^- = 1.9 \times 10^{-5} \text{ mol}$, $\text{FE} = 40 \%$. $\text{TON} = (\mu\text{mol O}_2)/(\mu\text{mol cat.}) = 0.4$. A TON of 97964 was obtained using the methodology developed by Savéant and coworkers based on the electroactive catalyst (i.e. catalyst in contact with the electrode).⁴

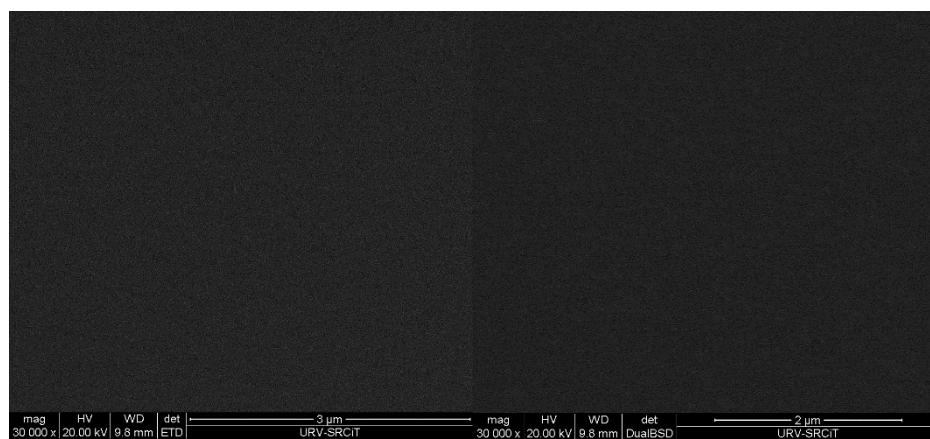
SUPPORTING INFORMATION

Scanning Electron Microscopy

A



B



C

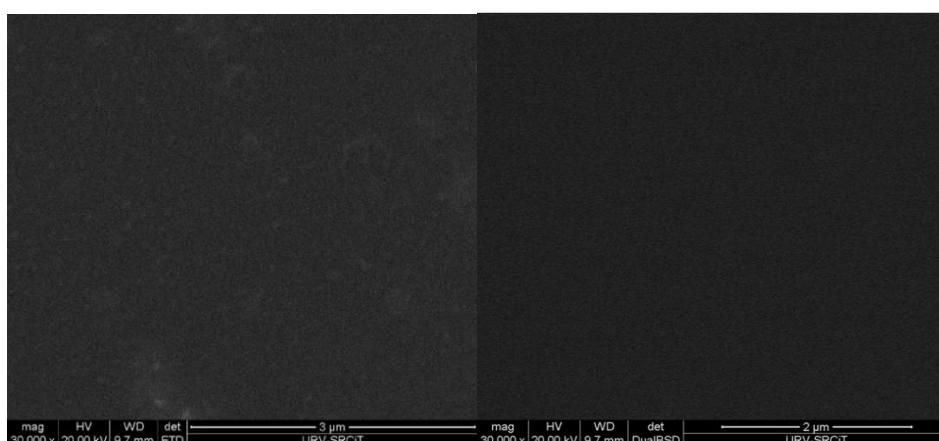


Figure S54. SEM micrographs (left) and corresponding back-scattered electron micrographs (right) of a glassy carbon plate after a 30 minute electrolysis of a 1.5 mM solution of the complexes in phosphate buffer pH 11.6 at 1.6 V vs. NHE. Code: (A) $[(L4)Cu]^{2+}$, (B) $[(L7)Cu]^{2+}$, (C) $[(L8)Cu]^{2+}$. There is no appreciable presence of copper oxide nanoparticles or deposited materials on the electrode.

SUPPORTING INFORMATION

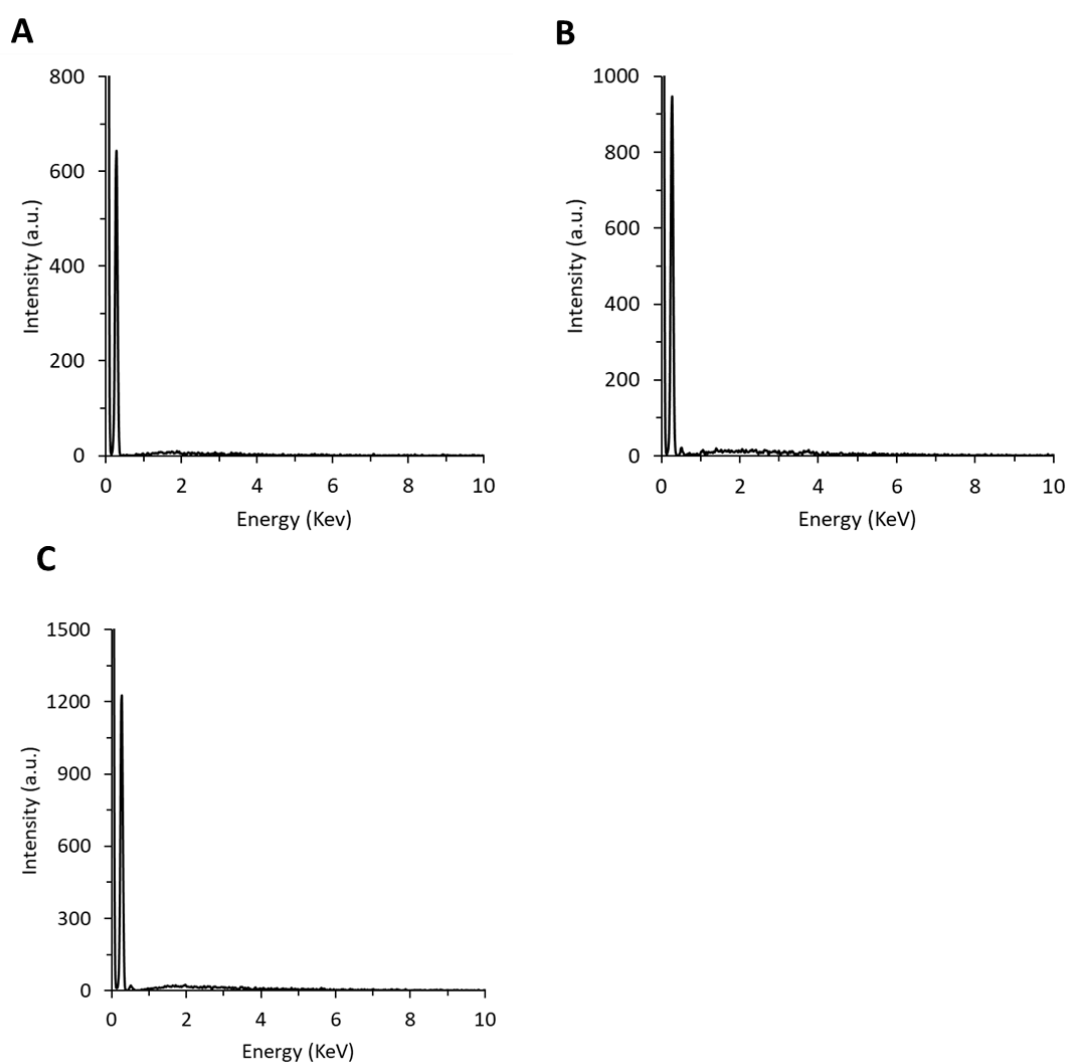


Figure S55. EDX spectra of the glassy carbon electrodes after a 30 minute electrolysis in phosphate buffer pH 11.6 at 1.6 V vs. NHE. Code: (A) $[(L4)Cu]^{2+}$, (B) $[(L7)Cu]^{2+}$, (C) $[(L8)Cu]^{2+}$. There is no appreciable presence of copper on the surface of the electrode (Cu expected peaks, $K\alpha = 8.04$ KeV and $L\alpha = 0.930$ KeV).

SUPPORTING INFORMATION

Computational studies

Computational Details

All calculations were carried out with the Gaussian09 program package¹³ using DFT methodology. We used B3LYP as the functional, including D3 empirical dispersion correction developed by Grimme (B3LYP-D3).^{14,15,14,15} The basis set was split into 6-31+G(d) for C, N, S, O and H^{16,17,18} and LANL2TZ(f) for Cu.^{19,20,21,22,23} Implicit solvation was introduced through the SMD model,²⁴ with water as the solvent. All geometry optimizations were computed in solution without symmetry restrictions. The nature of all computed stationary points as minima or transition states was confirmed through vibrational frequency calculations. Free energy corrections were calculated at 298.15 K and 105 Pa pressure, including zero point energy corrections (ZPE). In addition, a correction term of 1.89 kcal/mol (at 298 K) was added when necessary to account for the standard state concentration of 1 M, except for water, whose concentration was considered to be 55.6 M and its correction term 4.3 kcal/mol. Unless otherwise mentioned, all reported energy values are free energies in solution. In addition, stability of the wave function was checked for the calculations (stable option in G16).

The reaction energy barriers of the Minimum Energy Crossing Points (MECP) were estimated from potential energy relaxed scan from the crossing point of the quartet and doublet potential energy surfaces, along the O-O internal reaction coordinate, when the transition states could be found (or do not exist), applying entropic corrections from the minima to compute an estimated free energy change.

In the transformation from free energies to electrochemical magnitudes the values of 4.28 V for the absolute potential of the standard hydrogen electrode²⁵ and -11.72 eV for the free energy of the proton in aqueous solution at pH=0 were taken from the literature.²⁶ The value for the free energy of the proton was translated to the experimental pH value by adding a correction term of $-0.059 \cdot \text{pH}$, following the same procedure described elsewhere.²⁷

The functional for the DFT calculations was B3LYP-D3 based on the calibration carried out in a previous work on related systems,^{6,28} where its performance was compared with that of M06, M06-D3, M06L, M06-2X, ω B97xD and B97D. In order to validate this DFT methodology, the calculated optimized structures were compared to the X-Ray ones. Table S4 summarizes all the main metrics for the coordination environment of the copper metal center. In addition, we have recalculated as single points all the species involved in Figure 3 using a larger basis set (6-311++G(3d,2p)) for all the atoms except Cu/LANL2TZ(f) for Cu) and no significant differences were found (see Figure S65).

Table S4. Comparison of the main metrics for the X-Ray structure and the DFT optimized structure of complexes $[(L4)Cu]^{2-}$ and $[(L5)Cu]^{2-}$.

COMPLEX	METRIC	X-RAY (SOLID)	DFT OPTIMIZED (WATER)
$[(L4)Cu]^{2-}$	Cu-N _{bpy}	1.95 Å, 1.95 Å	2.00 Å, 2.00 Å
	Cu-N _{amide}	2.00 Å, 2.00 Å	2.04 Å, 2.05 Å
	N _{bpy} -Cu-N _{bpy}	78.27°	77.55°
	N _{amide} -Cu-N _{amide}	119.69°	123.15°
$[(L5)Cu]^{2-}$	Cu-N _{bpy}	1.95 Å, 1.95 Å	1.99 Å, 1.99 Å
	Cu-N _{amide}	2.00 Å, 2.00 Å	2.02 Å, 2.03 Å
	N _{bpy} -Cu-N _{bpy}	78.49°	78.02°
	N _{amide} -Cu-N _{amide}	120.78°	123.01°

Table S5. Comparison of the experimental and calculated redox potential for the Cu(III)/Cu(II) couples in complexes $[(L4)Cu]^{2-}$ and $[(L5)Cu]^{2-}$.

COMPLEX	$E_{1/2}^{\circ}$ (EXP)	$E_{1/2}^{\circ}$ (CALC)
$[(L4)Cu]^{2-}$	1.37 V	1.27 V
$[(L5)Cu]^{2-}$	1.27 V	1.14 V

SUPPORTING INFORMATION

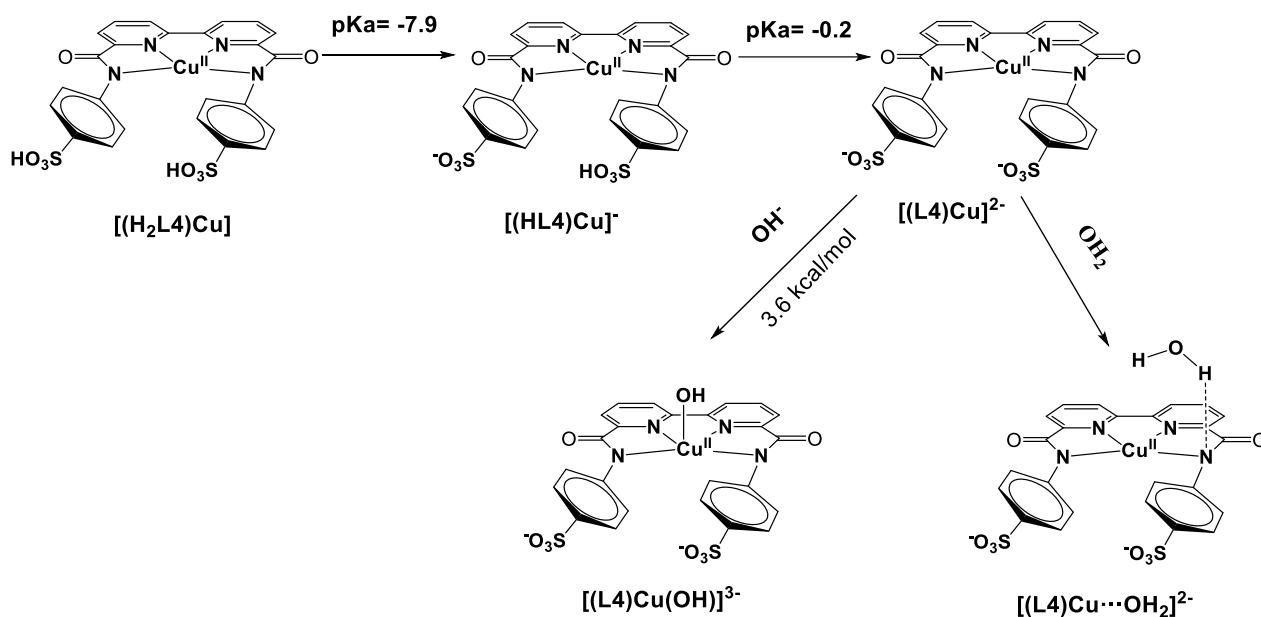
Speciation for complex $[(L4)Cu]^{2-}$ in water

Figure S56. Deprotonation processes in complex $[(L4)Cu]^{2-}$ and apical coordination of hydroxo or water molecules. The corresponding pK_a values or the free energy changes for each process are indicated above the arrows.

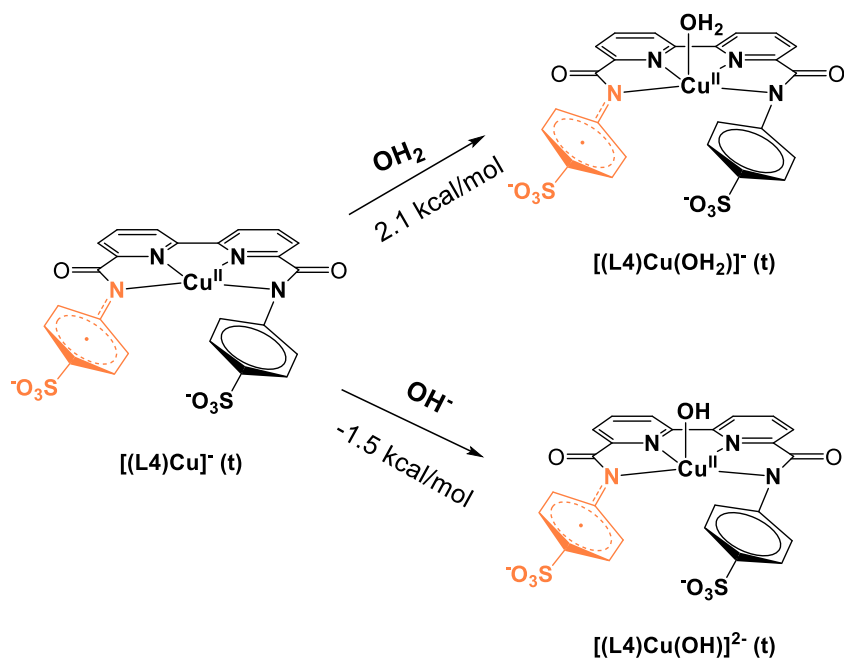


Figure S57. Apical coordination of hydroxo or water molecules in the one electron oxidized complex $[(L4)Cu]^{2-}$. The corresponding values of the free energy changes for each process are indicated above the arrows.

SUPPORTING INFORMATION

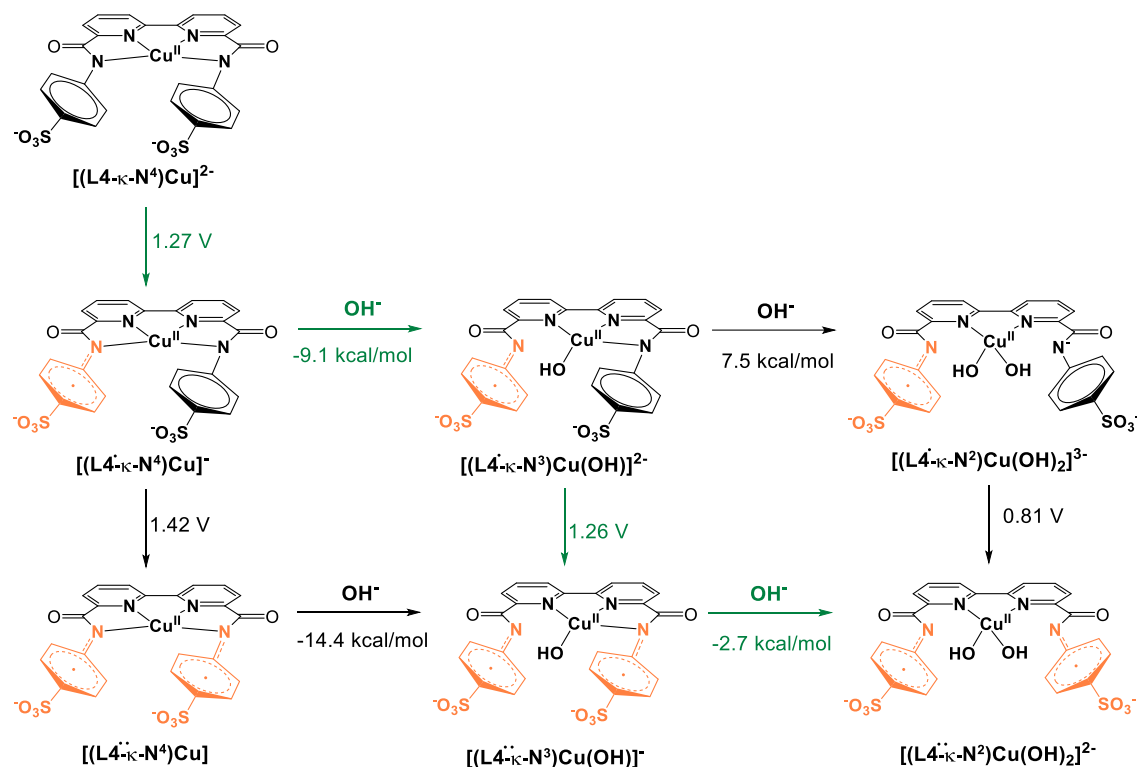
Electrochemical activation for complex $[(L4)Cu]^{2-}$ 

Figure S58. Possible pathways for the oxidation of catalyst $[(L4)Cu]^{2-}$ to generate an active species for the O-O bond formation step.

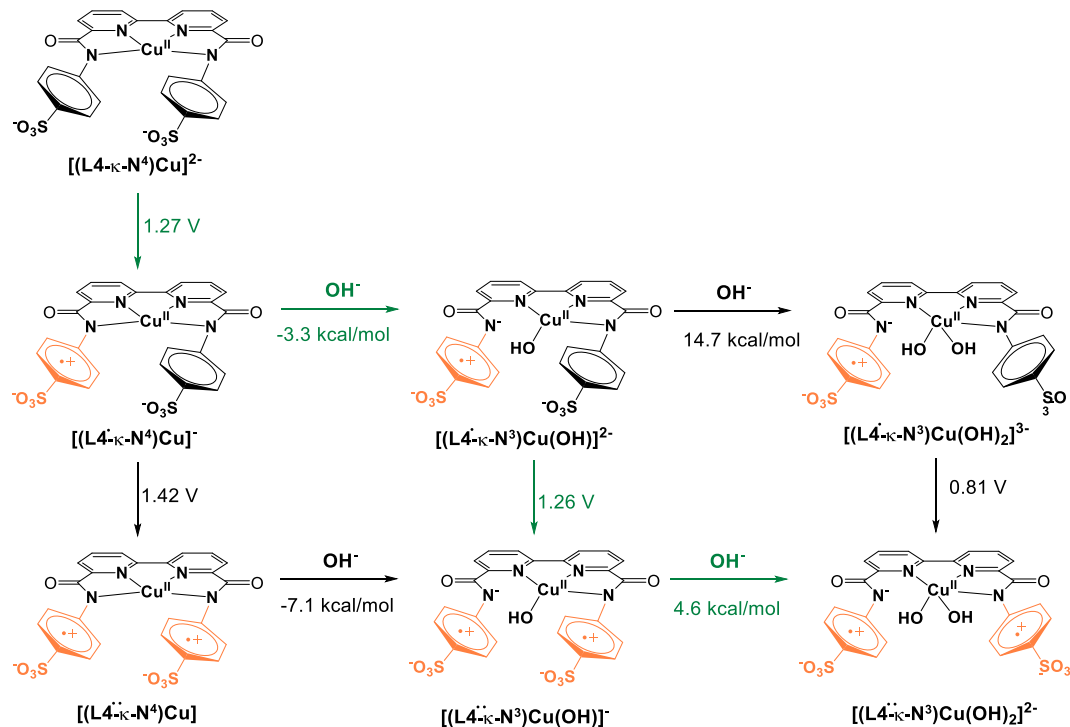


Figure S59. Possible pathways for the oxidation of catalyst $[(L4)Cu]^{2-}$ to generate an active species for the O-O bond formation step using the experimental value for the energy of OH^- solvation at standard state ($-102.8 \text{ kcal}\cdot\text{mol}^{-1}$).²⁹ This results suggest that the energy of solvation for OH^- might be underestimated by SMD calculations, leading to less favorable coordination of the OH^- to the Cu center. However, the calculated speciation in the different oxidation states is not substantially altered from that proposed using SMD, with the exception of formation of complex $[(L4-\kappa-N^3)Cu(OH)_2]^{2-}$ that is now slightly uphill but still accessible at room temperature towards formation of O_2 . Therefore, the proposed reaction pathways are still supported by DFT but the calculated barrier for the O-O bond formation step might be overestimated when using the SMD model.

SUPPORTING INFORMATION

Potential energy relaxed scan for $[(L4)Cu(OH)]^-$ and $[(L4)Cu(OH)_2]$

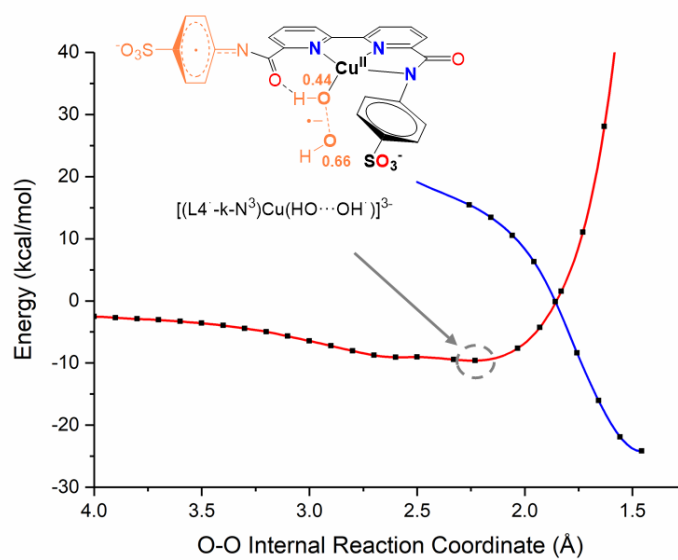


Figure S60. Potential energy relaxed scan for catalyst $[(L4)Cu]^{2-}$ of the O-O reaction coordinate considering a monohydroxylation pathway. Red color represents the quartet state while blue color indicates the doublet state. The inset represents the optimized structure of the 2c-3e⁻ intermediate.

SUPPORTING INFORMATION

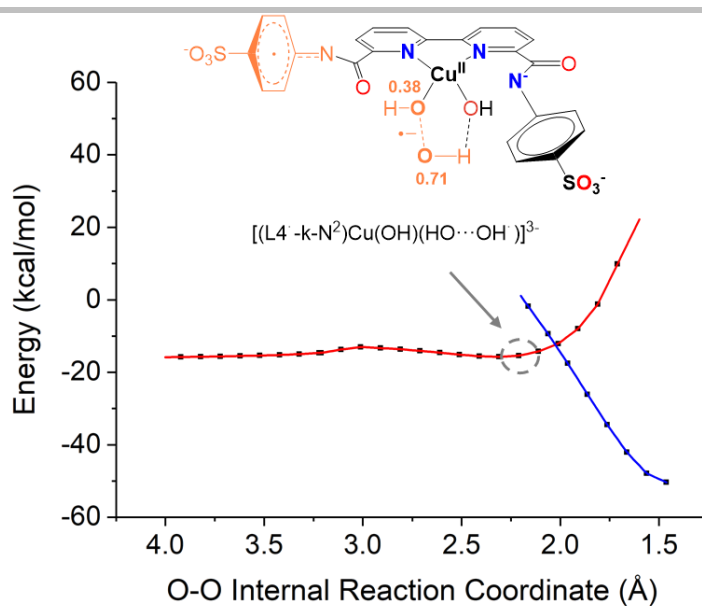


Figure S61. Potential energy relaxed scan for catalyst of $[(L4)Cu]^{2+}$ of the O-O reaction coordinate considering a dihydroxylation pathway. Red color represents the quartet state while blue color indicates the doublet state. The inset represents the optimized structure of the 2c-3e⁻ intermediate.

Structure and spin density distribution of 2c-3e⁻ intermediates

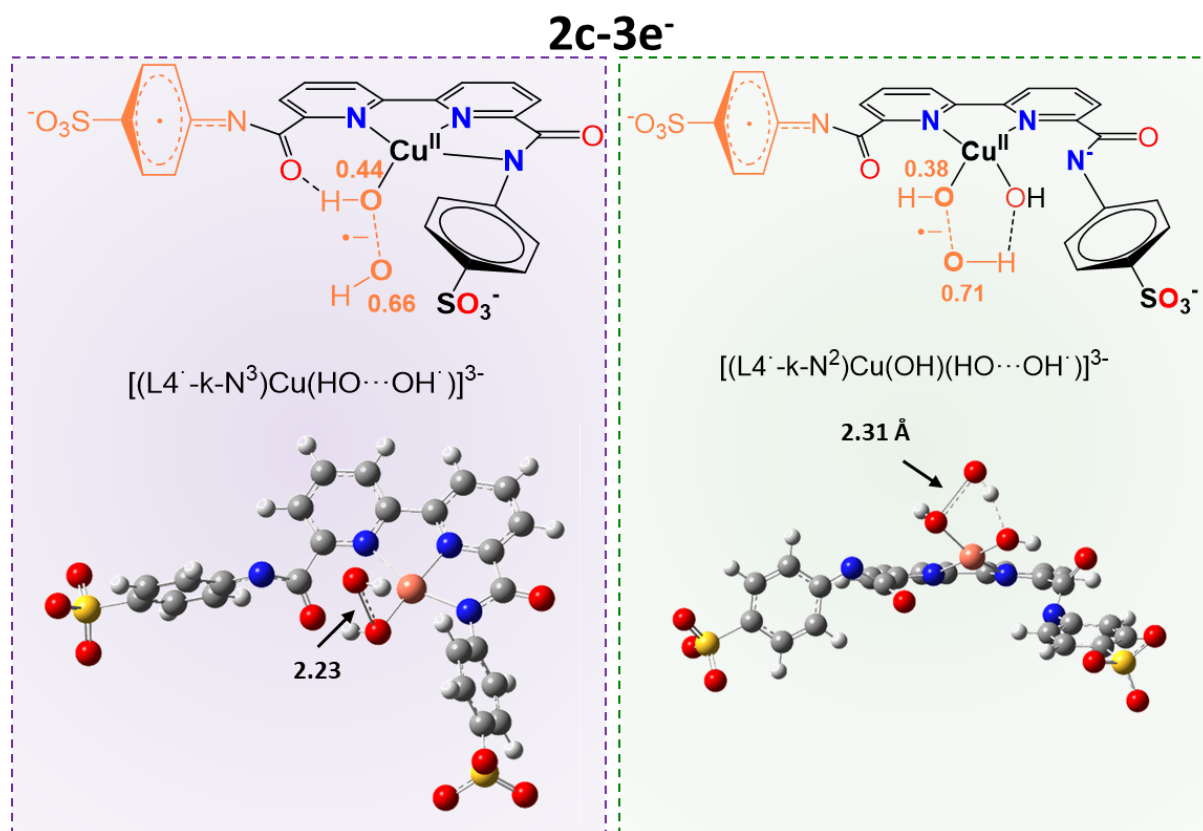


Figure S62. Schematic drawing with the spin distribution of each oxygen atom in orange (top) and optimized structure (bottom) of the 2c-3e⁻ intermediates for $[(L4^{\cdot-}-\kappa-N^3)Cu(OH)]^{2+}$ (left) and $[(L4^{\cdot-}-\kappa-N^2)Cu(OH)]^{2+}$ (right) catalysts.

SUPPORTING INFORMATION

Intramolecular relaxed scan for $[(L4-\kappa-N^2)Cu(OH)_2]^{2-}$

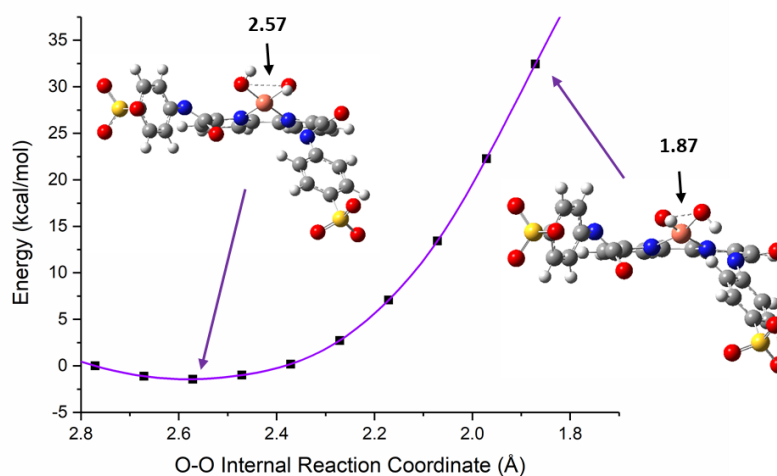


Figure S63. Potential energy relaxed scan of the intramolecular HO...OH coupling using catalyst $[(L4-\kappa-N^2)Cu(OH)_2]^{2-}$.

Structure of complex $[(L5)Cu]^{2-}$ and its oxidized species

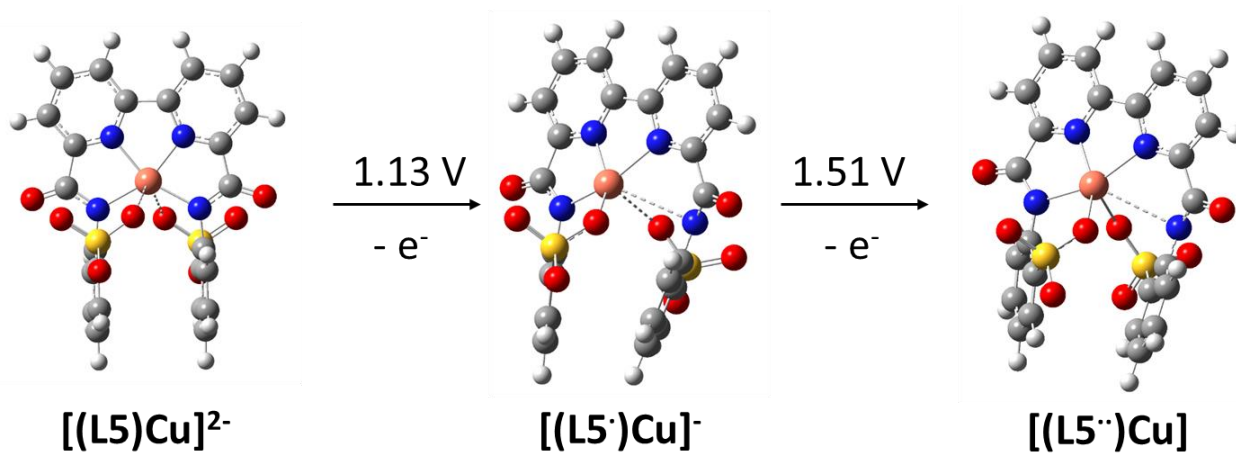


Figure S64. Optimized structures of complex $[(L5)Cu]^{2-}$ and its oxidized species with the calculated redox potential above the arrows. All the structures feature coordination of the copper center to at least one of the sulfonate groups.

SUPPORTING INFORMATION

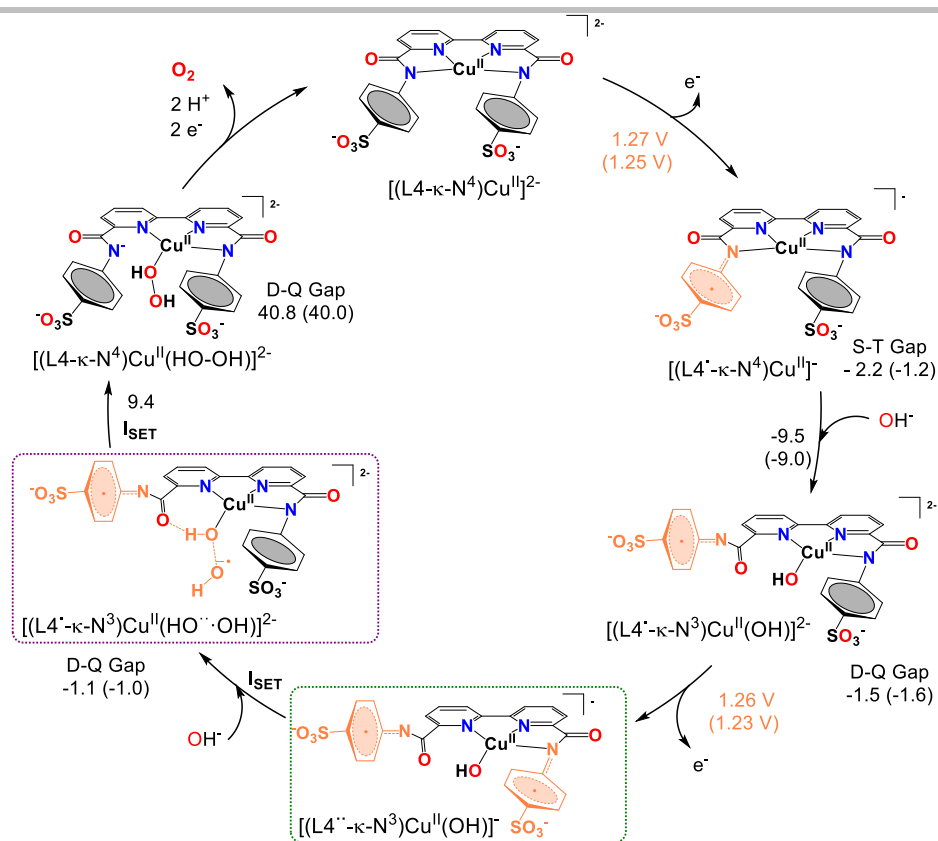


Figure S65. Influence of the basis set size in the catalytic cycle depicted in Figure 3 of the main text. Small basis set (6-31+G(d)/LANL2TZ(f)) and large basis set (6-311++G(3d,2p)) shown in brackets, calculated energies are very similar and no influence is observed in oxidation potentials, OH coordination or spin state gap. Energies in kcal/mol.

SUPPORTING INFORMATION

Effect of explicit solvation in spin state and reaction profile for the SET-WNA mechanism

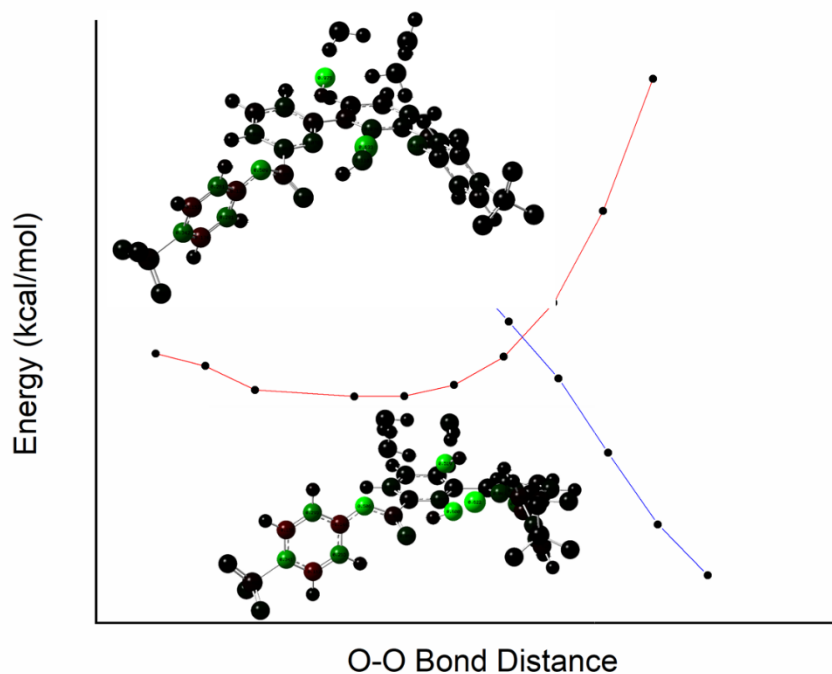


Figure S66. Potential energy relaxed scan of the intramolecular $(\text{H}_2\text{O})_3\text{HO}\cdots\text{OH}$ SET-WNA using catalyst for $[(\text{L}4-\kappa\text{-N}^3)\text{Cu}(\text{OH})]^{2-}$ in the quartet state.

Water Nucleophilic Attack potential energy surface

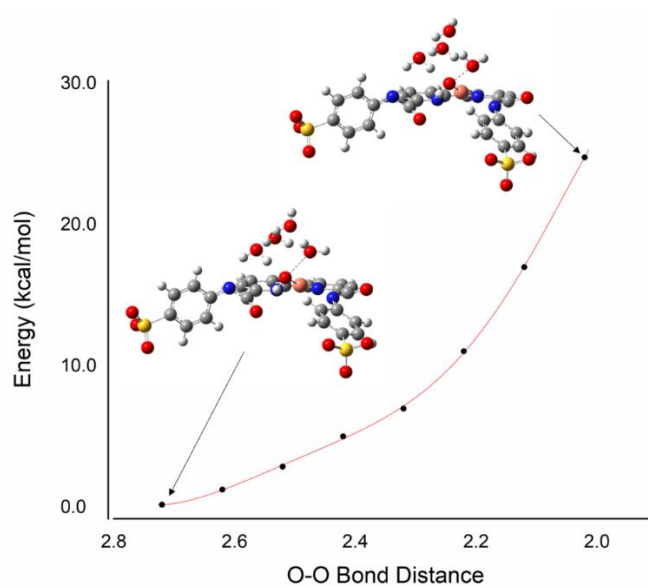


Figure S67. Potential energy relaxed scan of the intramolecular $(\text{H}_2\text{O})_3\text{H}_2\text{O}\cdots\text{OH}$ Water Nucleophilic Attack using catalyst for $[(\text{L}4-\kappa\text{-N}^3)\text{Cu}(\text{OH})]^{2-}$ in the quartet state.

SUPPORTING INFORMATION

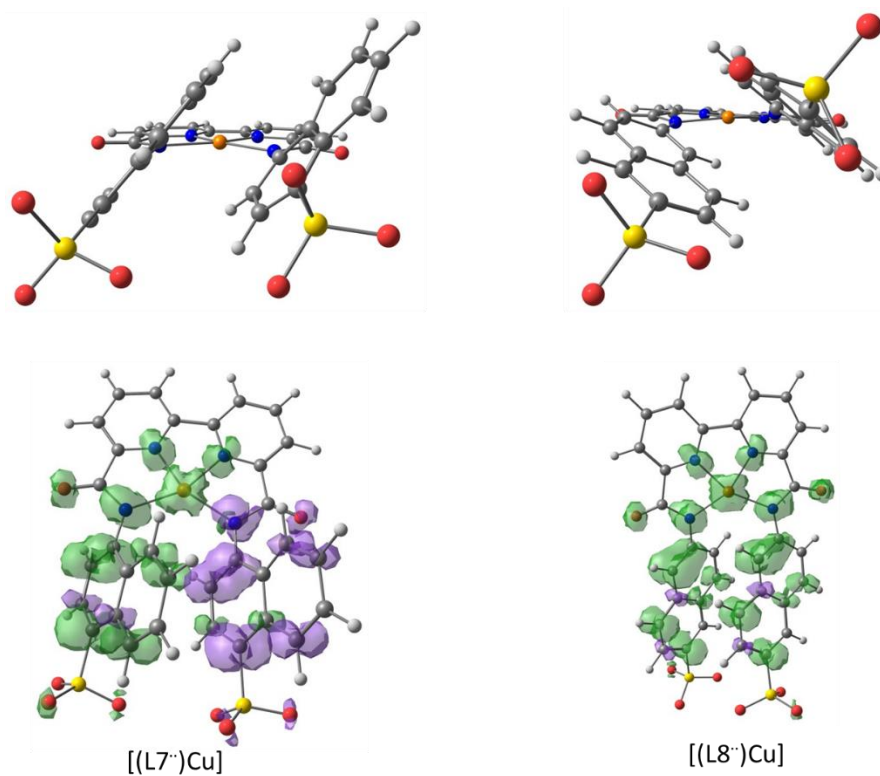


Figure S68. DFT optimized structures with representation of the spin distribution for the species [(L7[•])Cu] and [(L8[•])Cu] showing the delocalization of the radicals over the naphthyl-amide moiety.

SUPPORTING INFORMATION

Table S6. Summary of energy data

Complex (multiplicity)	Ep (H)	ZPVCs (H)	G (H)
OH ⁻	-75.947690	-0.00776	-75.95545
[(L4)Cu] ²⁻ (d)	-2736.833258	0.290922	-2736.542336
[(H ₂ L4)Cu] (d)	-2737.702827	0.316735	-2737.386092
[(HL4)Cu] ⁻ (d)	-2737.275645	0.303058	-2736.972587
{[(L4)Cu]...H ₂ O} ²⁻ (d)	-2813.280029	0.310897	-2812.969132
[(L4)Cu(OH) _{apical}] ³⁻ (d)	-2812.786660	0.297645	-2812.489015
[(L4)Cu] ⁻ (s)	-2736.628191	0.293373	-2736.334818
[(L4)Cu] ⁻ (t)	-2736.629034	0.290744	-2736.33829
[(L4)Cu(OH ₂) _{apical}] ⁻ (s)	-2813.074690	0.315265	-2812.759425
[(L4)Cu(OH ₂) _{apical}] ⁻ (t)	-2813.071800	0.310034	-2812.761766
[(L4)Cu(OH) _{apical}] ²⁻ (s)	-2812.585820	0.301714	-2812.284106
[(L4)Cu(OH) _{apical}] ²⁻ (t)	-2812.592184	0.299074	-2812.29311
[(L4)Cu(OH)] ³⁻ (d)	-2812.787172	0.296464	-2812.490708
[(L4)Cu(OH)] ²⁻ (s)	-2812.585856	0.299376	-2812.28648
[(L4)Cu(OH)] ²⁻ (t)	-2812.602386	0.294792	-2812.307594
[(L4)Cu(OH ₂)] ⁻ (s)	-2813.036962	0.312095	-2812.724867
[(L4)Cu(OH ₂)] ⁻ (t)	-2813.069746	0.309626	-2812.76012
[(L4)Cu(OH)] ⁻ (d)	-2812.397681	0.295942	-2812.101739
[(L4)Cu(OH)] ⁻ (q)	-2812.400366	0.296262	-2812.104105
[(L4)Cu(OH ₂)] (d)	-2812.847401	0.310589	-2812.536812
[(L4)Cu(OH ₂)] (q)	-2812.858131	0.308672	-2812.549459
[(L4)Cu(O)] ²⁻ (d)	-2811.927027	0.285831	-2811.641196
[(L4)Cu(O)] ²⁻ (q)	-2811.926833	0.28338	-2811.643453
[(L4)Cu(HO...OH)] ²⁻ (d)	-2888.362321	0.304879	-2888.057442
[(L4)Cu(HO...OH)] ²⁻ (q)	-2888.363391	0.304178	-2888.059213
[(L4)Cu(HO-OH)] ²⁻ (d)	-2888.386577	0.311244	-2888.075334
[(L4)Cu(HO-OH)] ²⁻ (q)	-2888.316567	0.306258	-2888.01031
[(L4)Cu(OH) ₂] ⁴⁺ (d)	-2888.732095	0.304238	-2888.427857
[(L4)Cu(OH) ₂] ³⁻ (s)	-2888.530842	0.307416	-2888.223427
[(L4)Cu(OH) ₂] ³⁻ (t)	-2888.548855	0.300718	-2888.248137
[(L4)Cu(OH) ₂] ²⁻ (dos)	-2888.350618	0.308289	-2888.042329
[(L4)Cu(OH) ₂] ²⁻ (q)	-2888.364446	0.303684	-2888.060762

SUPPORTING INFORMATION

$[(L4)Cu(OH)(O)]^{3-}$ (d)	-2887.887073	0.297118	-2887.589955
$[(L4)Cu(OH)(O)]^{3-}$ (q)	-2887.874387	0.291281	-2887.583106
$[(L4)Cu(OH)(HO...OH)]^{3-}$ (d)	-2964.313943	0.313294	-2964.000649
$[(L4)Cu(OH)(HO...OH)]^{3-}$ (q)	-2964.314770	0.313757	-2964.001013
$[(L4)Cu(OH)(HO-OH)]^{3-}$ (d)	-2964.379201	0.32112	-2964.058081
$[(L4)Cu(OH)(HO-OH)]^{3-}$ (q)	-2964.298809	0.317063	-2963.981746
$[(L5)Cu]^{2-}$ (d)	-2736.83333164	0.292585	-2736.5407
$[(L5)Cu(OH)]^{3-}$ (d)	-2812.783247	0.296405	-2812.4868
$[(L5)Cu]^{-}$ (s)	-2736.633847	0.294614	-2736.3392
$[(L5)Cu]^{-}$ (t)	-2736.632465	0.290809	-2736.3417
$[(L5)Cu(OH)]^{2-}$ (s)	-2812.585527	0.301584	-2812.2839
$[(L5)Cu(OH)]^{2-}$ (t)	-2812.590105	0.299515	-2812.2906
$[(L5)Cu]$ (d)	-2736.422273	0.293613	-2736.1287
$[(L5)Cu]$ (q)	-2736.4213	0.292433	-2736.1289
$[(L7)Cu]$ (dos)	-3043.74310150	0.381056	-3043.362046
$[(L8)Cu]$ (q)	-3043.74912970	0.379864	-3043.369266

SUPPORTING INFORMATION

Cartesian coordinates (Å) and calculated potential energies (atomic units)

OH ⁻	E= -75.947690 H			S	9.04134800	11.38747200	10.47216700
O	-1.77132600	0.29368800	0.00000000	S	13.18105500	12.15055900	10.54984000
H	-2.09552100	1.20918600	0.00000000	N	9.42014900	10.61566600	16.36417900
				N	9.99772400	10.37382100	18.87769200
				N	12.35073900	11.22157200	18.96968400
				N	12.86424000	11.63668700	16.47288500
				O	7.26590600	9.78472100	16.71940600
				O	15.03040200	12.35536000	16.97461400
				O	7.68642400	11.11410900	9.98214000
				O	9.94220500	10.20608300	9.84042000
				O	9.65344600	12.69125800	10.12070400
				O	12.27592000	13.43762200	10.18968000
				O	12.56592500	10.95225100	9.93115300
				O	14.53309600	12.52401900	10.12180900
				C	9.26672300	10.78812500	14.97958000
				C	9.64521400	12.01654100	14.40893800
				C	9.58825100	12.21464500	13.03279400
				C	9.14555200	11.16316600	12.22439000
				C	8.75737500	9.92916900	12.76391000
				C	8.81984900	9.74736700	14.13940700
				C	8.41455500	10.11007200	17.10618100
				C	8.75590400	9.99959100	18.57740200
				C	7.88803500	9.58043300	19.58566700
				C	8.36487800	9.57983000	20.90204600
				C	9.67087200	9.99308600	21.19242800
				C	10.48266300	10.39723000	20.13189700
				C	11.87961200	10.89848500	20.18603000
				C	12.69653800	11.06414300	21.30544200
				C	13.98845500	11.56809000	21.10973200
				C	14.44692800	11.89137400	19.82648300
				C	13.57646200	11.69401400	18.75493800
				C	13.88890800	11.94663500	17.29498500
				C	12.99801600	11.77199600	15.08203000
				C	13.46420000	12.95920800	14.48113200
				C	13.51329200	13.07560200	13.09786000
				C	13.09119900	11.99563300	12.31031400
				C	12.62504200	10.80622800	12.87876000
				C	12.58150400	10.70554100	14.26551700
				H	14.64229700	11.70687500	21.96518700
				H	15.44584400	12.27836700	19.66055600
				H	12.33813400	10.80826400	22.29659700
				H	10.04189400	10.00676600	22.21159000
				H	7.71333000	9.26242900	21.71035500
				H	6.87578800	9.27405600	19.34768000
				H	8.54253800	8.79144100	14.56908300
				H	10.01013200	12.80878200	15.05467200
				H	9.90018500	13.15838300	12.59929200
				H	8.42385300	9.11905100	12.12258400
				H	12.19689100	9.80444700	14.73256800
				H	13.76806400	13.79319400	15.10352500
				H	13.86467100	13.99661200	12.64270500
				H	12.28462200	9.98587500	12.25665900
				H	10.91538000	10.41899800	9.92455200
				H	11.30410300	13.20659300	10.22691600
				[(HL4)Cu] ⁻ (d)	E= -2737.275645 H		
				Cu	11.15877700	10.98593500	17.38530900
				S	8.94947500	11.53170300	10.49633600
				S	13.30102800	12.04483500	10.49896300
				N	9.42244800	10.56864300	16.36691800
				N	9.99960200	10.35588900	18.88169000
				N	12.33640500	11.24487200	18.96775400
				N	12.84179900	11.64138600	16.46661200
				O	7.27385600	9.72087300	16.72918000
				O	14.99020300	12.41774700	16.95862800
				[(H ₂ L4)Cu] (d)	E= -2737.702827 H		
				Cu	11.16912300	10.98999100	17.38722000

SUPPORTING INFORMATION

[(L4)Cu(OH)] ²⁻ (s)			E= -2812.585856 H				
				O	2.73219300	5.82247300	14.15096500
				O	3.51664700	5.81974000	11.78047800
Cu	11.64512900	10.77920800	17.32850500	O	15.83100700	14.88432500	10.72375800
S	6.34180200	8.04066800	10.65608900	O	16.51366800	12.49669000	10.42530400
S	13.87408800	12.16024800	10.64357500	O	17.96979200	14.06532700	11.70869500
N	8.61978400	8.79448400	16.17495700	C	7.43681200	7.91534100	14.55703900
N	10.13527100	10.26508600	18.60755000	C	7.28079400	6.71390900	13.79151700
N	12.45364500	11.32258200	18.94431000	C	6.03923400	6.34470600	13.31126400
N	13.28386700	11.49077500	16.56678300	C	4.92185400	7.15843300	13.57561400
O	7.71801400	10.91835900	16.64968200	C	5.04212500	8.34160400	14.33032500
O	15.15808500	12.70268200	17.17604100	C	6.27670000	8.72188400	14.81709100
O	11.13392700	10.19298400	15.68909900	C	8.97599600	9.39392600	15.64409900
O	5.52986400	9.26538700	10.34174700	C	9.28329900	9.29637300	17.11209400
O	5.48165700	6.80841500	10.74857300	C	8.47182600	8.49223700	17.91666000
O	7.50367800	7.85759800	9.72507900	C	8.66962500	8.51842600	19.29648700
O	12.46871400	12.33598400	10.14595900	C	9.66645300	9.33559200	19.81744000
O	14.52705200	10.92025800	10.11756900	C	10.45452500	10.09616700	18.94158700
O	14.70854700	13.39123200	10.44695500	C	11.54266000	10.98047500	19.42692000
C	8.06375800	8.68500600	14.88806300	C	11.76742700	11.29544200	20.77277700
C	8.88051500	8.20672100	13.84488700	C	12.82420900	12.14495700	21.09861100
C	8.37583500	8.02380800	12.55686000	C	13.64024400	12.65959800	20.08936400
C	7.03268000	8.30567600	12.29403900	C	13.36251900	12.30463100	18.77229300
C	6.19468700	8.75986600	13.32121500	C	14.18770900	12.77960300	17.60158400
C	6.70557600	8.95037500	14.60214500	C	14.44803200	12.62650100	15.24354900
C	8.35762900	9.85414600	16.91470600	C	14.50728100	13.96271400	14.80871900
C	8.90566100	9.80351200	18.32845800	C	15.13195600	14.28975800	13.60646700
C	8.04292800	9.37698700	19.35000500	C	15.69877200	13.27725300	12.82350400
C	8.45610600	9.44614600	20.67197700	C	15.64636000	11.94396700	13.24055400
C	9.71310500	9.98415200	20.96384800	C	15.01924600	11.62342000	14.44473500
C	10.52693400	10.39308700	19.91686500	H	11.03242900	10.48538600	14.53038000
C	11.84005300	11.02657000	20.10120700	H	13.00947900	12.40241200	22.13684400
C	12.47461800	11.36387000	21.30207500	H	14.47265500	13.31822400	20.30745100
C	13.71787400	11.99746200	21.25168500	H	11.13069200	10.89629300	21.55299100
C	14.32413600	12.28614900	20.02250800	H	9.83668900	9.37037200	20.88665600
C	13.64865700	11.91418600	18.86974900	H	8.05323000	7.91442200	19.95463600
C	14.11398100	12.09631000	17.45901800	H	7.69457700	7.87990700	17.47302700
C	13.43761000	11.64957400	15.18169700	H	6.37615200	9.63063100	15.40210200
C	13.14885500	12.88679100	14.57446900	H	8.15784300	6.10424400	13.59850000
C	13.30247300	13.03899200	13.20224400	H	5.92675100	5.43753700	12.72745200
C	13.72940300	11.94696200	12.43162900	H	4.16803200	8.95391400	14.52564800
C	14.00432600	10.71046300	13.01974300	H	14.96115900	10.58961800	14.77091800
C	13.85252100	10.56124000	14.39815500	H	14.05344600	14.74086600	15.41485500
H	10.36110500	9.60123300	15.81748800	H	15.16726500	15.32360100	13.27742800
H	14.21749600	12.26723400	22.17658300	H	16.08212400	11.16010900	12.62945300
H	15.28900700	12.77583800	19.95378200				
H	12.00915400	11.13919100	22.25437200				
H	10.04655500	10.08611300	21.98999300				
H	7.80745300	9.10755200	21.47366600				
H	7.06003800	9.00319400	19.08261800				
H	6.05086100	9.30050400	15.39311600				
H	9.92435400	7.98497500	14.05090200				
H	9.02553200	7.66792300	11.76392900				
H	5.14753800	8.97063900	13.12301000				
H	14.05693900	9.60869000	14.87571000				
H	12.81402300	13.71638600	15.18972500				
H	13.08902500	13.99532900	12.73486400				
H	14.32965800	9.87379800	12.41147000				
[(L4)Cu(OH)] ²⁻ (t)			E= -2812.602386 H				
Cu	12.07759000	11.10599500	16.53725000				
S	3.28686500	6.63162400	13.01567300				
S	16.56449900	13.71113300	11.30541600				
N	8.65641400	8.21291500	15.02048300				
N	10.26729500	10.06883500	17.60469800				
N	12.33469200	11.49448200	18.47058100				
N	13.78464700	12.26847400	16.43595500				
O	9.06979500	10.46393700	15.03416600				
O	15.16329800	13.55018700	17.81245700				
O	11.96245800	10.70473600	14.70055900				
O	2.51274200	7.88727600	12.76588100				
[(L4)Cu(OH ₂)] ⁻ (s)			E= -2813.036962 H				
Cu				Cu	11.68746300	10.91979500	17.15037000
S				S	6.30062900	7.43151700	10.80583500
S				S	13.98052400	12.93670700	10.71737000
N				N	8.21209600	8.55696400	16.39606300
N				N	10.12420100	10.40168800	18.49512400
N				N	12.58303200	11.16882700	18.81289600
N				N	13.52096300	11.27138800	16.43672900
O				O	8.45699800	10.86809400	16.08718100
O				O	15.51670200	12.16986200	17.16259400
O				O	10.91690200	10.80312400	15.35970700
O				O	5.50893500	8.62152200	10.34704600
O				O	5.44353400	6.20194100	10.94487300
O				O	7.51778900	7.18199400	9.96419500
O				O	12.96727400	14.02456700	10.54336400
O				O	13.70086600	11.72504800	9.88907700
O				O	15.39032200	13.42991700	10.58348600
C				C	7.77893200	8.35597300	15.06982800
C				C	8.54552100	7.55755300	14.20304600
C				C	8.11313800	7.29634400	12.90223800
C				C	6.89846400	7.82586300	12.45595100
C				C	6.11333800	8.60968000	13.30989200
C				C	6.55456600	8.87516200	14.60468900
C				C	8.47350600	9.77411600	16.77837700
C				C	8.90237600	9.91521300	18.22086200

SUPPORTING INFORMATION

H	8.07403700	6.29581200	12.19621400	N	12.33495000	11.46571400	18.51587600
H	4.51134800	8.02153500	13.93843200	N	13.79799800	12.19829700	16.49778500
H	13.43853800	10.18713300	14.06065000	O	9.37497500	10.44303400	14.92478000
H	14.72772300	13.89925100	15.89209000	O	15.13509600	13.54020400	17.85162400
H	15.28275100	14.80239800	13.66495500	O	12.10326100	10.43824200	14.67404000
H	14.02948200	11.09724200	11.81751600	O	2.73332000	7.92084000	12.57859200
[(L4)Cu(OH)]⁺ (q)				E= -2812.400366 H			
Cu	12.34102300	10.45297100	16.68810100	O	2.79257800	6.30321200	14.48645500
S	3.65335300	6.64766700	12.67674600	O	3.50877400	5.57239700	12.20225200
S	15.32798600	14.58617300	11.30451800	O	15.45794300	14.68961500	10.63072000
N	8.86485700	8.02824500	15.18158300	O	16.45247400	12.40230900	10.45653100
N	10.42766300	10.00392800	17.65378900	O	17.70542800	14.22867300	11.60799500
N	12.58846700	11.22281300	18.51198000	C	7.55298500	7.99732900	14.47650400
N	14.07576600	11.80143600	16.41842900	C	7.37432000	6.75331900	13.72446500
O	9.59041100	10.18587000	14.89851200	C	6.12366900	6.37263800	13.32409000
O	15.82019500	12.46227700	17.81495600	C	5.02549000	7.21361200	13.62806400
O	12.54833400	9.36195000	15.17610500	C	5.15525300	8.44423900	14.33513800
O	3.60665600	7.46003700	11.41705100	C	6.38807900	8.84306600	14.75596000
O	2.61210600	7.06354100	13.67024600	C	9.13831500	9.42280100	15.56041100
O	3.67585700	5.17407100	12.42695100	C	9.34749000	9.29138400	17.03408500
O	14.52129900	15.84655800	11.38124300	C	8.48640600	8.50266000	17.79934600
O	14.89586300	13.67287800	10.20493100	C	8.62482200	8.52437200	19.18818700
O	16.80358000	14.83594400	11.32673400	C	9.60577500	9.32704100	19.75347600
C	7.68719300	7.76330400	14.60643900	C	10.44975800	10.97245700	18.91500500
C	7.48539300	6.45658100	14.05460100	C	11.51949800	10.94864400	19.44749700
C	6.27457100	6.11556600	13.48009400	C	11.71117700	11.26598800	20.79653800
C	5.23574100	7.06085000	13.44249100	C	12.76002800	12.12039700	21.14127500
C	5.40462300	8.35605000	13.97590300	C	13.59906300	12.63732400	20.15116500
C	6.60698100	8.70979000	14.55082600	C	13.35318800	12.27813200	18.82912700
C	9.25643800	9.26046800	15.64105500	C	14.18216400	12.74584100	17.65931400
C	9.35009500	9.38771800	17.13948800	C	14.43438900	12.56562300	15.29179300
C	8.28500600	8.95040700	17.93081700	C	14.43525900	13.89798700	14.83938800
C	8.32898100	9.19561100	19.30231100	C	15.01041700	14.22471100	13.61425000
C	9.43164500	9.85511000	19.83687500	C	15.58092400	13.21828000	12.82393500
C	10.47225700	10.24119300	18.98485000	C	15.58898700	11.89188800	13.25912500
C	11.68479000	10.94542300	19.46548100	C	15.01777800	11.56968600	14.49249800
C	11.92567900	11.32060300	20.79406600	H	11.20759400	10.59117100	14.31338500
C	13.11788000	11.97129500	21.10884600	H	12.92112100	12.38097800	22.18258000
C	14.05035100	12.24033900	20.10440900	H	14.42291400	13.30035300	20.38817200
C	13.74120000	11.84444000	18.80637400	H	11.05768800	10.86611400	21.56250200
C	14.65771100	12.09521600	17.65187100	H	9.72607200	9.36630900	20.82931800
C	14.41292600	12.45764500	15.28858200	H	7.96930900	7.92954600	19.81547900
C	15.14225600	13.69796100	15.26376800	H	7.71414800	7.90253000	17.33272100
C	15.40912000	14.31270600	14.06125600	H	6.52579600	9.77292200	15.29864000
C	14.97474300	13.72333100	12.85286000	H	8.25430900	6.15293000	13.51741900
C	14.24704200	12.52175500	12.84943300	H	5.96381900	5.45208100	12.77488600
C	13.95480900	11.90141200	14.04838900	H	4.27766500	9.05068200	14.53002900
H	13.31822200	12.26667800	22.13365000	H	15.01798500	10.54071600	14.84124800
H	14.98756700	12.74368900	20.31154700	H	13.97434800	14.66879700	15.44919300
H	11.20250000	11.10852300	21.57190600	H	15.00036800	15.25423800	13.26985100
H	9.48133500	10.06103000	20.89913400	H	16.03080800	11.11436500	12.64496800
H	7.51292300	8.88112000	19.94483100	H	12.71679400	10.94193100	14.10665300
H	7.43702300	8.44831000	17.47857600	[(L4)Cu(OH)₂] (q)			
H	6.74165000	9.70095500	14.97262100	E= -2812.858131 H			
H	8.30121600	5.74199000	14.09925100	Cu	12.04526300	11.11923600	16.58068500
H	6.12337300	5.12493200	13.06585400	S	3.38303900	6.60588700	12.83153300
H	4.58871500	9.07127500	13.94277700	S	16.55370700	13.55517100	11.26740500
H	13.38643900	10.97664500	14.08673000	N	8.65813200	8.08936900	15.13269600
H	15.45480000	14.16391100	16.18891100	N	10.33454600	10.03665400	17.63356900
H	15.94779800	15.25469200	14.04555700	N	12.35277400	11.55304000	18.47932200
H	13.91405500	12.08964600	11.91245400	N	13.81174300	12.33117400	16.41763200
[(L4)Cu(OH)₂] (d)				O	9.07143700	10.33482700	15.05704000
E= -2812.847401 H				O	15.16491800	13.60484900	17.83034100
Cu	12.11523300	11.04334900	16.60515200	O	11.66232000	10.83679000	14.63179000
S	3.34888400	6.69648000	13.15744900	O	2.63022800	7.87507900	12.58869200
S	16.35626200	13.66491900	11.25941700	O	2.77092900	5.76151800	13.90950400
N	8.75273500	8.26897100	14.87817800	O	3.67000500	5.83573400	11.58196700
N	10.33112900	10.03876600	17.57152300	O	16.09160400	14.91774100	10.86267000
				O	16.15227000	12.47832600	10.31531200
				O	18.00471400	13.51332600	11.63720400
				C	7.46262100	7.81202500	14.60144900

SUPPORTING INFORMATION

H	11.42448000	10.66067000	21.73640300
H	9.84201300	9.35651400	21.01264300
H	7.92855900	8.10995800	20.02494200
H	7.57432600	8.19941100	17.54369500
H	6.38875300	9.69906500	15.40230600
H	8.54029200	6.42234400	13.53345800
H	6.41418000	5.62490000	12.51309200
H	4.29254300	8.89541400	14.38312100
H	14.16152600	10.62559500	14.43329300
H	14.45745300	14.69420400	15.79667500
H	15.25745600	15.39110300	13.56727500
H	14.95690100	11.31810300	12.18388300

[(L4)Cu(HO...OH)]²⁺ (d) E= -2888.362321 H

Cu	12.14961800	11.07702500	16.64937800
S	3.40297900	6.74463500	12.89334400
S	16.23704700	13.86433700	11.26794000
N	8.72938200	8.14991600	15.12976700
N	10.38626400	9.97605300	17.71053700
N	12.48173800	11.35685800	18.57992400
N	13.85191800	12.21560100	16.53089200
O	9.15980600	10.39674700	15.14573400
O	15.28757300	13.43472400	17.90279400
O	11.93084700	10.84400100	14.76188000
O	2.96972200	7.93806700	12.09805900
O	2.54328800	6.51129500	14.09914200
O	3.59623900	5.51803500	12.06109500
O	15.17374700	13.50988100	10.26471300
O	17.48578900	13.05357700	11.07505000
O	16.50147400	15.33934100	11.32338500
C	7.52046000	7.87981500	14.62496500
C	7.37002300	6.69207000	13.83874200
C	6.13838100	6.35052000	13.30950900
C	5.02835100	7.17632700	13.55243200
C	5.14328200	8.34848600	14.32985000
C	6.36492300	8.70155300	14.86202700
C	9.06884800	9.32262700	15.75221500
C	9.39562800	9.21303800	17.21489200
C	8.59161800	8.39584800	18.01393800
C	8.80528100	8.39243500	19.39119900
C	9.81758700	9.18824100	19.91477200
C	10.59920600	9.96154900	19.04472600
C	11.71308200	10.80911500	19.53578200
C	11.98875500	11.05803800	20.88599500
C	13.06425400	11.88305800	21.21342000
C	13.84943600	12.44029100	20.20172900
C	13.52476200	12.14736600	18.88074700
C	14.30480600	12.67308700	17.70111300
C	14.45635300	12.63163100	15.32484700
C	14.44896500	13.98001400	14.93128200
C	15.00791400	14.36528200	13.71239800
C	15.57960200	13.39903400	12.87908700
C	15.59184400	12.05132800	13.25612900
C	15.03142600	11.67308100	14.47476300
H	10.99778900	10.59371700	14.63222500
H	13.28780300	12.08942000	22.25549100
H	14.69342700	13.08329300	20.42195700
H	11.37676200	10.62706300	21.66884900
H	10.00547400	9.19589400	20.98147100
H	8.19263400	7.77958700	20.04460400
H	7.80717900	7.79535100	17.56667300
H	6.45992300	9.59898100	15.46485900
H	8.24297600	6.07051900	13.66617200
H	6.02764300	5.45398000	12.70997800
H	4.27333800	8.97172600	14.51198100
H	15.03549900	10.63005400	14.77645500
H	13.99850000	14.72338100	15.58208600
H	14.99811300	15.40889900	13.41565800
H	16.03629000	11.30189800	12.60841600
O	12.47125500	8.66518500	14.89034300

H	13.34763100	9.06329500	15.04723200
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[(L4)Cu(HO...OH)]²⁺ (q) E= -2888.363391 H

Cu	12.15960800	11.06851400	16.63283600
S	3.41191400	6.69312500	12.91624900
S	16.20786000	13.91731800	11.25708400
N	8.73797900	8.14067500	15.12619400
N	10.37887500	9.98199500	17.70239000
N	12.48733800	11.34863300	18.56736300
N	13.86039600	12.20768400	16.51681700
O	9.17241600	10.38631900	15.12787600
O	15.30402100	13.41352200	17.89148300
O	11.94084300	10.83873500	14.74744500
O	2.96600200	7.88115600	12.11983400
O	2.55936700	6.45670600	14.12654500
O	3.61027900	5.46568800	12.08644800
O	15.02971500	13.89495700	10.32196000
O	17.25426900	12.91120500	10.87833300
O	16.76358300	15.30030300	11.43457400
C	7.52912600	7.86104100	14.62656700
C	7.38169000	6.66613500	13.85069800
C	6.15004000	6.31507200	13.32776400
C	5.03715800	7.13825800	13.56658200
C	5.14918700	8.31742800	14.33372500
C	6.37080400	8.68000900	14.85960400
C	9.07251900	9.31789400	15.74304000
C	9.38662900	9.22014800	17.20906400
C	8.57525400	8.41282800	18.01072800
C	8.78493200	8.41801900	19.38867600
C	9.80174100	9.20987600	19.90957900
C	10.59009400	9.97327200	19.03652100
C	11.71040500	10.81456600	19.52444700
C	11.98252600	11.07121300	20.87404900
C	13.06318900	11.88977600	21.20028800
C	13.85558400	12.43488800	20.18773300
C	13.53324200	12.13589000	18.86748800
C	14.31655800	12.65803100	17.68836900
C	14.46319200	12.63069700	15.31232500
C	14.45383100	13.98317800	14.92906700
C	15.00786500	14.37722700	13.71242500
C	15.57662300	13.41684500	12.86800200
C	15.59104000	12.06756300	13.23391400
C	15.03628900	11.68002900	14.45392500
H	11.00743100	10.58243600	14.62702600
H	13.28414300	12.10179600	22.24178300
H	14.70240900	13.07465400	20.40650700
H	11.36321000	10.65155900	21.65734700
H	9.98807300	9.22163900	20.97650700
H	8.16677700	7.81338400	20.04455400
H	7.78933400	7.81293800	17.56527200
H	6.46376100	9.58294700	15.45454200
H	8.25670500	6.04661800	13.68129800
H	6.04141100	5.41303500	12.73614400
H	4.27719700	8.93868200	14.51281200
H	15.04414300	10.63494700	14.74840400
H	14.00541800	14.72098800	15.58756600
H	14.99645200	15.42396700	13.42532800
H	16.03326700	11.32518400	12.57729400
O	12.47888100	8.70594200	15.11597300
H	13.38186200	9.06903300	15.05827100

[(L4)Cu(HO-OH)]²⁺ (d)E= -2888.386577 H

Cu	11.69255900	11.00398900	17.20650100
S	5.84031600	7.67243800	10.95600800
S	14.39134600	12.96226200	10.74119300
N	8.31722600	8.28683200	16.40086000
N	10.30356700	10.03964900	18.57670300
N	12.64383600	11.15979200	18.92284200
N	13.40497900	11.87382700	16.56806400

SUPPORTING INFORMATION

O	8.50126200	10.62261300	16.34022500	C	5.64533300	6.57935200	13.59867500
O	15.40987900	12.74232400	17.36987000	C	4.88137200	7.65119500	14.08596500
O	10.78552100	11.09568900	15.38770800	C	5.48525700	8.71931800	14.77832200
O	5.47993900	9.04885100	10.47334100	C	6.85087800	8.71923500	14.98565900
O	4.61994900	6.82161100	11.18274400	C	9.67543700	8.62736300	15.30219600
O	6.85355300	6.99651400	10.08037400	C	9.93645000	8.57511500	16.70448200
O	13.05323900	12.72198000	10.09954600	C	9.30645600	7.54869100	17.47410100
O	15.42942100	11.98371000	10.27870900	C	9.35296100	7.56731200	18.84663500
O	14.83635300	14.38966100	10.61700200	C	10.03087400	8.62877700	19.48925000
C	7.75702400	8.20918100	15.11370800	C	10.68960800	9.56924300	18.71419400
C	8.34348000	7.35577200	14.16239100	C	11.46751200	10.65776300	19.34531600
C	7.78266800	7.20022100	12.89329100	C	11.46862200	11.01860000	20.70221500
C	6.61442900	7.89206300	12.56402600	C	12.27226200	12.08072100	21.11364300
C	6.00333100	8.73325000	13.50353100	C	13.07367400	12.76844300	20.19019800
C	6.57230000	8.89205200	14.76391700	C	13.02918000	12.35987200	18.86481800
C	8.60189400	9.45743700	16.90194600	C	13.83970400	12.95555900	17.74142600
C	9.12556500	9.44125000	18.31764700	C	14.32846000	12.74670900	15.41784000
C	8.36041000	8.84908700	19.33043000	C	13.93331600	13.90931000	14.73633100
C	8.82144500	8.88744300	20.64150300	C	14.56475600	14.28165900	13.55081200
C	10.04573800	9.49797800	20.91085000	C	15.59760600	13.48775900	13.03802900
C	10.76601200	10.05858000	19.85670100	C	16.00356700	12.32941800	13.70561100
C	12.08389000	10.70609700	20.05163200	C	15.37246600	11.96615900	14.89742700
C	12.75938700	10.86438700	21.26837700	H	11.34383300	10.12537500	14.47743000
C	14.00587100	11.49221400	21.26398800	H	12.27695300	12.37524300	22.15860000
C	14.56560300	11.94954900	20.06550900	H	13.71095100	13.59295800	20.48860000
C	13.83842300	11.76388000	18.89507100	H	10.85052100	10.49427900	21.42100400
C	14.29261800	12.18373900	17.51854700	H	10.05864900	8.68383500	20.57060900
C	13.69422300	12.15820400	15.21332900	H	8.86510300	6.79254900	19.42992300
C	13.50410200	13.44543200	14.68936400	H	8.76594000	6.76227300	16.95982300
C	13.73938100	13.69776700	13.33761100	H	7.31975200	9.53809900	15.52201600
C	14.16324900	12.65765000	12.50344400	H	7.62306100	5.75016000	13.43154700
C	14.35370000	11.36906600	13.01093200	H	5.16368500	5.76677800	13.06601700
C	14.12354100	11.12423100	14.36582400	H	4.88073800	9.54192700	15.14806100
H	9.79849000	10.91904400	15.63791300	H	15.68875800	11.07428800	15.43164000
H	14.54331800	11.62303900	22.19809200	H	13.12433300	14.51196700	15.13888800
H	15.53346600	12.43644900	20.03592000	H	14.25270700	15.18225400	13.03089000
H	12.32883600	10.51003600	22.19719800	H	16.80871900	11.72031200	13.30814100
H	10.42997600	9.53424200	21.92320100	O	13.22464700	9.58706500	14.25196800
H	8.23983200	8.44781300	21.44594900	H	13.99429500	10.14907500	14.02922400
H	7.41527600	8.38207100	19.07819300				
H	6.09931300	9.54485500	15.49096700				
H	9.25097000	6.81740900	14.42205800				
H	8.25389600	6.54594200	12.16717000				
H	5.09188300	9.26824400	13.25236700				
H	14.27513500	10.12910300	14.77463900				
H	13.16919000	14.24341200	15.34560200				
H	13.59435300	14.69666700	12.93854100				
H	14.68572800	10.56483100	12.36235600				
O	11.11298700	9.90560500	14.61316200				
H	11.88823300	10.23583700	14.11471300				

[(L4)Cu(OH-OH)]²⁺ (q)E= -2888.316567 H

Cu	12.21782100	10.88513900	16.59602300
S	3.08955700	7.66878100	13.87610500
S	16.35475900	13.93422100	11.46497200
N	8.99030500	7.55038200	14.69820100
N	10.70481700	9.54658300	17.34120800
N	12.23570400	11.34475000	18.49012900
N	13.64694300	12.32007800	16.58024600
O	10.02685900	9.56164900	14.48319300
O	14.60963800	13.92655100	17.96523000
O	12.25722700	10.62121100	14.59302700
O	2.76308700	9.01508600	13.30287300
O	2.52828800	7.48155000	15.25563900
O	2.75748800	6.53912800	12.95278800
O	15.43190900	13.41847300	10.39562600
O	17.69533800	13.26425000	11.43510800
O	16.44789500	15.43242500	11.45820700
C	7.66032700	7.63600800	14.50546500
C	7.01575600	6.57059300	13.80173400

[(L4)Cu(OH)₂]⁴⁺ (d) E= -2888.732095 H

Cu	11.10198700	10.48303400	16.69061100
S	5.41432900	6.51523000	11.21341700
S	16.12547900	16.35132500	11.76030200
N	7.95998600	8.03970100	16.44272900
N	9.88080800	10.06375700	18.32922500
N	12.19141400	11.43247300	18.36488800
N	13.77479500	13.83128300	16.68663700
O	7.85266900	10.37656600	16.20891500
O	14.35276300	11.59490900	16.25813600
O	11.04124500	8.71346000	15.84385100
O	4.64212900	7.69998100	10.70744500
O	4.50962600	5.36422300	11.56665600
O	6.51762500	6.09944900	10.28389700
O	16.85491600	17.56108800	12.28240000
O	14.98533500	16.72538900	10.85831200
O	17.06394400	15.36063300	11.13180100
C	7.37456400	7.76914900	15.19843700
C	8.00436700	6.84585600	14.33900400
C	7.42971000	6.48296500	13.12147000
C	6.20304000	7.03761200	12.74049100
C	5.55129300	7.94749100	13.58240300
C	6.13448100	8.31250400	14.79381800
C	8.10307200	9.29200500	16.83313000
C	6.88571500	9.45147300	18.22130700
C	7.96849400	9.04313100	19.34838800
C	8.50136700	9.29417300	20.61034400
C	9.72543400	9.95149800	20.71692200
C	10.39901800	10.32968900	19.55038700
C	11.68681800	11.07128300	19.57016900

SUPPORTING INFORMATION

H	12.45674500	12.70750000	13.69707200
H	15.63550300	14.11743800	16.29711800
H	15.98101800	16.05449100	14.81210200
H	12.82673400	14.66310500	12.20347900
O	13.62049600	8.93365600	16.45287000
H	14.28198000	9.15396300	17.12788000

[[L4]Cu(OH)₂]²⁻ (dos) E= -2888.350618 H

Cu	11.61907400	9.73420700	17.20824900
S	5.74976600	7.07951000	11.05778500
S	15.01869500	16.56538500	12.02514300
N	8.36072300	8.65928500	16.23878400
N	10.08443400	10.10199400	18.42596800
N	12.52301200	11.02839800	18.40496100
N	13.45558300	12.39256300	15.97078100
O	7.43991100	10.79643200	16.54532800
O	15.47713800	11.46766100	16.62508300
O	11.00114200	8.21156900	16.43265200
O	5.66404900	8.32604900	10.21817400
O	4.39017100	6.54814600	11.41275200
O	6.63741800	6.03944100	10.43960600
O	15.27448600	17.80586600	12.82512000
O	13.87854600	16.70264400	11.06902600
O	16.26176500	16.01956900	11.39016500
C	7.72840800	8.37064300	15.01953200
C	8.47516900	7.71227900	14.02218100
C	7.88381400	7.31381700	12.82386800
C	6.52354200	7.55422800	12.60941100
C	5.75997100	8.20118800	13.58922100
C	6.35639800	8.60630500	14.78092200
C	8.13205900	9.78900800	16.88036300
C	8.76285300	9.91204600	18.25816500
C	7.89091600	9.96189500	19.35528800
C	8.39511200	10.20353300	20.62525200
C	9.75322300	10.49592700	20.77282000
C	10.57137800	10.46997600	19.64982300
C	11.96080700	10.96158700	19.63775300
C	12.64698700	11.42289400	20.76406400
C	13.91810200	11.96706300	20.60313900
C	14.46405400	12.06971800	19.32412400
C	13.73275400	11.59049800	18.23677800
C	14.31069800	11.80197000	16.86039700
C	13.85278100	13.34533500	15.11044200
C	15.06849100	14.09962700	15.24824000
C	15.39228500	15.06632800	14.32048300
C	14.52754400	15.31622900	13.23388500
C	13.32208300	14.61143300	13.08433300
C	12.97926500	13.64829300	14.01659200
H	10.00277500	8.32332800	16.37378800
H	14.47046900	12.32556100	21.46575100
H	15.43270800	12.52622000	19.15365100
H	12.19617000	11.36002900	21.74690500
H	10.15366500	10.77083100	21.74093200
H	7.73815000	10.21117000	21.48925200
H	6.83018000	9.80986400	19.18656400
H	5.75597400	9.09146500	15.54282700
H	9.52791600	7.50882500	14.19702100
H	8.47573200	6.80714400	12.06844200
H	4.70053500	8.37963900	13.42991700
H	12.05240300	13.09138500	13.92411500
H	15.72379400	13.91297900	16.09085300
H	16.30943800	15.63650200	14.42991500
H	12.66726100	14.82444600	12.24676600
O	13.15938700	9.42075600	16.30484000
H	12.93409600	8.66063000	15.73741300

[[L4]Cu(OH)₂]²⁻ (q) E= -2888.364446 H

Cu	12.65833500	9.44445500	16.88617900
S	3.67170200	7.34682700	12.53725600
S	14.94986600	17.39960700	12.57841700
N	8.88564500	7.53531800	15.38741400
N	10.69449300	9.73529500	17.56343500
N	12.78443900	11.23404400	18.14092300
N	14.09333700	12.61869900	16.01467600
O	10.32257000	9.22305500	14.80954700
O	15.99041300	12.31884900	17.29098700
O	12.43957200	7.52273000	16.60733400
O	3.75802400	8.34066700	11.42086400
O	2.67963400	7.74459000	13.58981700
O	3.49248300	5.93927800	12.06447100
O	14.19501700	18.49260000	13.27363400
O	14.39475700	17.06450300	11.23159700
O	16.42914400	17.63631100	12.56725500
C	7.72862700	7.53530300	14.71695700
C	7.13134100	6.27120600	14.40322800
C	5.92178500	6.21149600	13.73587400
C	5.27797000	7.40334500	13.36126600
C	5.83914200	8.66274200	13.65793900
C	7.04486200	8.73685300	14.32356500
C	9.62166400	8.66551600	15.65641400
C	9.58286900	9.15066000	17.08239700
C	8.39885600	9.09025600	17.81742700
C	8.36847200	9.68339500	19.08065000
C	9.50596300	10.32522800	19.55973200
C	10.66450000	10.33353600	18.77097200
C	11.90276000	11.06318800	19.14918100
C	12.13077500	11.59740300	20.42288800
C	13.29065700	12.33780900	20.64357100
C	14.17089300	12.56146600	19.58536700
C	13.87203000	11.99557000	18.34379000
C	14.75740800	12.32973900	17.17608200
C	14.33810500	13.70457400	15.26709600
C	15.07247700	14.84800000	15.73466300
C	15.24949600	15.94121500	14.91317700
C	14.70969000	15.93592400	13.60955600
C	13.97969100	14.83739600	13.12778200
C	13.78670000	13.73822600	13.94567500
H	11.57265100	7.23467700	16.93686400
H	13.49491300	12.75267100	21.62557400
H	15.06051700	13.17038000	19.70246400
H	11.42814500	11.43513000	21.23147800
H	9.48443100	10.82307700	20.52150100
H	7.46057800	9.66263900	19.67486400
H	7.51854500	8.61217600	17.40345500
H	7.48332700	9.70226800	14.55627800
H	7.64886800	5.36494000	14.70174300
H	5.47319900	5.25351100	13.49726400
H	5.32615600	9.57231000	13.36184000
H	13.22349400	12.87858300	13.59674800
H	15.47593400	14.85110700	16.74123600
H	15.80179800	16.80419700	15.27186600
H	13.57124600	14.85398400	12.12355800
O	14.48845000	9.36132900	16.28898400
H	15.07684100	9.74145800	16.96083500

[[L4]Cu(OH)(O)]³⁻ (d) E= -2887.952624 H

Cu	11.47604200	10.02067900	16.42655100
S	5.08513200	5.60568400	11.85292000
S	15.84554200	17.50531100	14.44828800
N	8.03985700	8.36158200	16.31117600
N	9.99277700	10.45162200	17.78839900
N	12.46460500	11.13865900	17.59390500
N	13.47716900	12.04448700	14.90510300
O	7.59994800	10.59374200	15.70551600
O	15.23079900	10.55414500	15.49892300
O	10.76749200	8.81393600	15.24477900

SUPPORTING INFORMATION

O	4.35144300	6.66410800	11.07908400	C	6.26477800	8.72012500	14.65819300
O	4.13877300	4.65601600	12.53757700	C	8.33406000	9.65469600	16.61795200
O	6.11113000	4.88790700	11.02623600	C	8.85082000	9.89797700	18.01941900
O	15.70217500	18.04053100	15.84429700	C	7.95315300	9.87817900	19.09359600
O	15.01438000	18.25020700	13.45098600	C	8.40897400	10.22537000	20.36028300
O	17.28033800	17.37055100	14.03211800	C	9.74163400	10.60492200	20.52316800
C	7.34554500	7.78507700	15.23414900	C	10.58596900	10.60276700	19.41123800
C	7.96210400	6.74825400	14.50692100	C	11.99652600	11.06704900	19.47561500
C	7.29271600	6.09479800	13.47168500	C	12.61980300	11.48747800	20.65933100
C	5.98331500	6.46395200	13.15161400	C	13.93055800	11.95195500	20.60901100
C	5.34119700	7.47784700	13.87498900	C	14.58597900	12.02125700	19.37968000
C	6.01682600	8.13242000	14.90137800	C	13.90259400	11.59820000	18.23957200
C	8.04530200	9.67226900	16.45925900	C	14.57660400	11.78124500	16.90746000
C	8.68089200	10.16725000	17.74299800	C	14.10660000	13.30919500	15.13503700
C	7.84271700	10.41941700	18.83848500	C	15.18783100	14.21018100	15.42546800
C	8.37317200	10.98081300	19.99200500	C	15.44208800	15.27857900	14.59199900
C	9.72885300	11.31715600	20.02242400	C	14.63901500	15.48656400	13.45063100
C	10.51154200	11.05142200	18.90574500	C	13.57051300	14.62850900	13.14506100
C	11.92840400	11.42699900	18.79558600	C	13.30028800	13.55685500	13.97772000
C	12.72992300	12.03620300	19.76040700	H	10.16805000	8.45204800	15.73000100
C	14.06483200	12.30665300	19.44165200	H	14.43111300	12.27220100	21.51736100
C	14.57661100	11.99844500	18.17783600	H	15.59502100	12.40819700	19.29256900
C	13.72576900	11.40509700	17.24899900	H	12.09778200	11.45217400	21.60708300
C	14.00783300	10.97205500	15.78960900	H	10.10070700	10.90377000	21.49999800
C	14.12228700	13.22449900	14.79030100	H	7.73639000	10.21742300	21.21276200
C	15.39970500	13.56831300	15.35020700	H	6.91820400	9.60262500	18.92055200
C	15.91118600	14.84600600	15.21720700	H	5.84090200	9.53325000	15.23942100
C	15.18762000	15.82847000	14.51540100	H	8.85420100	6.50957200	14.69746100
C	13.95781400	15.52056700	13.91645900	H	7.75913000	5.79240400	12.59290600
C	13.44310700	14.23932000	14.03907800	H	4.73829900	8.81863900	13.14980300
H	9.91183000	8.53085800	15.62893800	H	12.47767100	12.88162100	13.76518000
H	14.70913900	12.76070800	20.18830700	H	15.79560500	14.05170800	16.30915700
H	15.61238900	12.19544000	17.92940800	H	16.25661700	15.95976100	14.81760700
H	12.33302200	12.28104000	20.73847700	H	12.96273500	14.80785500	12.26523600
H	10.16437900	11.78740700	20.89633300	O	13.52612600	9.12716700	16.44221400
H	7.74291900	11.17323400	20.85462500				
H	6.78890000	10.17235700	18.76042600				
H	5.51547400	8.91654500	15.45964100				
H	8.97948500	6.45945400	14.75687700				
H	7.78991600	5.30715400	12.91485200				
H	4.32050300	7.76188700	13.63523000				
H	12.48975100	13.98947100	13.58351300				
H	15.98411100	12.81255900	15.85592100				
H	16.87682000	15.08671500	15.65203400				
H	13.41515400	16.27931600	13.36296800				
O	13.12677800	9.82638400	15.59985400				

[[L4]Cu(OH)(HO...OH)]³⁺ (d) E=-2964.313943 H

Cu	12.36614300	10.13490200	16.72694500
S	3.51221600	6.99501800	12.81703300
S	15.97506700	16.41137700	12.04423800
N	8.93562200	7.87869400	15.08736800
N	10.52353800	9.82461800	17.63767600
N	12.48474500	11.39406300	18.45037300
N	13.54894200	13.56811300	16.75299500
O	9.83056200	9.97916000	14.89491500
O	15.50781700	12.60874900	17.63375600
O	12.29826700	8.33934000	15.84566200
O	3.16011500	8.23653700	12.05756700
O	2.65301700	6.79405500	14.03028400
O	3.59850800	5.77692700	11.95312300
O	15.50029700	17.83425900	12.17882700
O	15.44381800	15.74940100	10.80694100
O	17.47016000	16.31030400	12.16104900
C	7.71714100	7.73342500	14.55676100
C	7.38766700	6.47058700	13.96423000
C	6.13277500	6.25182400	13.42660500
C	5.17491100	7.27991300	13.46169500
C	5.46821900	8.53293700	14.03798300
C	6.71618600	8.76455800	14.57781000
C	9.42679800	9.05469600	15.60353800
C	9.50096400	9.12907700	17.10649400
C	8.48323300	8.59178100	17.89318800
C	8.52906200	8.80244500	19.27328400
C	9.56812100	9.54926200	19.81619100
C	10.56119300	10.05694100	18.96540800
C	11.67110600	10.92659900	19.42959500
C	11.83982900	11.30694100	20.76134000
C	12.87024100	12.19367700	21.08204300
C	13.67397300	12.70004900	20.06654900
C	13.43778500	12.29214600	18.74811600

[[L4]Cu(OH)(O)]³⁺ (q) E=-2887.874387 H

Cu	11.79698800	9.83972100	16.88437800
S	5.34385800	6.73498600	11.23307000
S	15.00328900	16.90889500	12.39926400
N	8.04324400	8.39767000	16.34455800
N	10.14305000	10.23119700	18.18239300
N	12.64868600	11.11014300	18.29459300
N	13.77054100	12.26981700	15.91340900
O	8.24371200	10.69506000	15.88538200
O	15.77265900	11.49807300	16.76249600
O	11.05881300	8.74681200	15.47201500
O	5.16166900	7.97008000	10.39481400
O	4.02296100	6.15492400	11.66157600
O	6.22979300	5.71700300	10.57733800
O	14.79158700	18.10688300	13.27545700
O	14.04917300	16.84387900	11.25040300
O	16.43470500	16.74183600	11.98790900
C	7.43819000	8.07995600	15.11944400
C	7.95609200	7.01675900	14.35500700
C	7.33985400	6.60971900	13.17043200
C	6.18256400	7.25927100	12.73287300
C	5.64139100	8.31383800	13.48211100

SUPPORTING INFORMATION

C	14.25454900	12.87729700	17.61964800	H	14.47560300	13.40349900	20.27291000
C	14.18753400	14.18694200	15.67026900	H	11.19700700	10.92124700	21.54009200
C	15.27412400	15.07814300	15.82122200	H	9.59889400	9.74606300	20.88022800
C	15.81916100	15.73650400	14.72181200	H	7.74722700	8.41087700	19.91400700
C	15.29372400	15.51158400	13.44209300	H	7.66517900	8.04614500	17.43901600
C	14.21655100	14.63767300	13.26862100	H	6.94942200	9.72552800	15.02208400
C	13.67066800	13.98594300	14.37507100	H	8.14505400	5.69011700	13.94938100
H	11.91163700	7.70500300	16.47388200	H	5.89006900	5.29436600	12.97568800
H	13.02499400	12.49749700	22.11292600	H	4.71784300	9.31408500	14.05292000
H	14.46529000	13.41330100	20.27203900	H	12.83456000	13.30543800	14.24142300
H	11.18758200	10.93016800	21.53960500	H	15.67894600	15.26000100	16.81138700
H	9.60236000	9.73578800	20.88231400	H	16.64889400	16.42462900	14.86069700
H	7.75267500	8.39856000	19.91525600	H	13.80748200	14.46436300	12.27780900
H	7.66819700	8.03961200	17.43954700	O	14.07358100	10.37103200	15.85725800
H	6.94826700	9.72686500	15.02310100	H	14.62485800	11.02796400	16.31286600
H	8.14397400	5.69206100	13.94821300	O	14.40452800	7.71022700	16.52791700
H	5.88840500	5.29620000	12.97587600	H	14.44601500	8.67817700	16.29887800
H	4.71611500	9.31530300	14.05535700				
H	12.83251100	13.30705700	14.24252900				
H	15.68058100	15.25687800	16.81198700				
H	16.65256600	16.41941800	14.86099800				
H	13.80747800	14.46381400	12.27863200				
O	14.07538000	10.37269300	15.86420200				
H	14.63009300	11.02419800	16.32355300				
O	14.43891100	7.71379000	16.48896300				
H	14.45336100	8.68726800	16.27333100				

[(L4)Cu(OH)(HO-OH)]³⁻ (d) E=-2964.369951 H[(L4)Cu(OH)(HO...OH)]³⁻ (q) E=-2964.314770 H

Cu	12.36723500	10.13364800	16.72295700	Cu	10.76396200	10.64121400	17.08006400
S	3.51419700	6.99340800	12.81501500	S	5.39426900	7.27558700	11.07044400
S	15.97165200	16.41581800	12.04391500	S	15.75148900	15.88261600	11.78384300
N	8.93643600	7.87715900	15.08812200	N	7.70413200	7.90402500	16.58225900
N	10.52498000	9.82415500	17.63679700	N	10.04856200	9.63098700	18.61403400
N	12.48458500	11.39688200	18.44849700	N	12.12592400	11.20942200	18.69375400
N	13.55010200	13.56763600	16.75199300	N	13.42924600	13.52651100	16.80658800
O	9.83250400	9.97677000	14.89321100	O	9.34685100	9.50833000	16.12261500
O	15.51024400	12.61127600	17.63308700	O	14.52738600	11.45129500	16.75636600
O	12.29047000	8.33881600	15.84014900	O	13.29968800	9.39253100	15.49707500
O	3.16228400	8.23490400	12.05542800	O	5.86627400	8.42250700	10.22521000
O	2.65457700	6.79236200	14.02795500	O	3.90736600	7.31534800	11.30161400
O	3.60090400	5.77532500	11.95113500	O	5.84657500	5.94216300	10.55029100
O	15.49609700	17.83831300	12.17980800	O	15.73971500	17.36619300	12.03889000
O	15.44006300	15.75435500	10.80648600	O	14.84118400	15.50225300	10.65110600
O	17.46690600	16.31563300	12.15969300	O	17.14395900	15.34769600	11.61640700
C	7.71824100	7.73181200	14.55685000	C	7.24496400	7.80861000	15.26220000
C	7.38889600	6.46880400	13.96462200	C	6.93422400	6.52718800	14.76265700
C	6.13432300	6.25010700	13.42623300	C	6.40643100	6.35961000	13.48390900
C	5.17662300	7.27839300	13.46032700	C	6.16393400	7.48031600	12.68236900
C	5.46981700	8.53158000	14.03629700	C	6.44490200	8.76408900	13.16413800
C	6.71747600	8.76315000	14.57686700	C	6.98756900	8.92601900	14.43741100
C	9.42789100	9.05354400	15.60302500	C	8.64872700	8.74280800	16.91084000
C	9.50102700	9.13046700	17.10595300	C	9.01382900	8.81396600	18.36817900
C	8.48099000	8.59734100	17.89245500	C	8.38683500	8.13742200	19.41563000
C	8.52558700	8.81112400	19.27214300	C	8.86428900	8.33976300	20.71190200
C	9.56585200	9.55656300	19.81459100	C	9.94575600	9.19460800	20.94310300
C	10.56157700	10.05939900	18.96397200	C	10.53104700	9.83998500	19.85051800
C	11.67349900	10.92660300	19.42829200	C	11.69473800	10.75643400	19.90146600
C	11.84691800	11.30082500	20.76122500	C	12.32889800	11.12003900	21.08859500
C	12.87993800	12.18434500	21.08249000	C	13.43740200	11.96712200	21.02776600
C	13.68189300	12.69310500	20.06674700	C	13.89237700	12.40331800	19.78791700
C	13.44078200	12.29116300	18.74733100	C	13.21272000	11.99411000	18.63367600
C	14.25654700	12.87786000	17.61893400	C	13.77293500	12.35040600	17.27554600
C	14.18788100	14.18748400	15.66945400	C	14.00423300	13.99340800	15.61171300
C	15.27288700	15.08063800	15.82057100	C	15.39626400	13.99942300	15.38316300
C	15.81676500	15.74019400	14.72130900	C	15.92940300	14.55517300	14.22103900
C	15.29177700	15.51460100	13.44151400	C	15.07710000	15.10554900	13.25766700
C	14.21623600	14.63872800	13.26784100	C	13.69295300	15.11495600	13.46594600
C	13.67152100	13.98576800	14.37414400	C	13.16674700	14.56943800	14.63511000
H	11.91285700	7.70478000	16.47443000	H	13.77376500	10.10969800	16.00211900
H	13.03852700	12.48334200	22.11420500	H	13.94260800	12.27111600	21.93961000
				H	14.76315300	13.04456800	19.69726000
				H	11.97124500	10.75203900	22.04306500
				H	10.31726600	9.34792200	21.94932100
				H	8.39475400	7.83024300	21.54778600
				H	7.55015700	7.47903900	19.21690300
				H	7.20272100	9.92340100	14.80443700
				H	7.11554300	5.65896500	15.39037700
				H	6.18613500	5.36279900	13.11477700
				H	6.24931000	9.63529800	12.54648500

SUPPORTING INFORMATION

H	12.09277900	14.58293500	14.80141600
H	16.06086200	13.57312000	16.12766100
H	17.00323600	14.55403100	14.06314200
H	13.02733100	15.54368400	12.72254200
O	11.12756700	11.71881300	15.54937800
H	11.71679500	12.44211400	15.82740500
O	12.79708600	10.13080200	14.33947400
H	12.11095100	10.75393900	14.77792000

[(L4)Cu(OH)(HO-OH)]³⁻ (q) E= -2964.298809 H

Cu	11.51086800	10.16462100	16.45172900
S	3.52490500	7.81695300	12.98580300
S	16.48200800	16.24210100	12.22930900
N	8.99906300	7.04150200	15.20580100
N	10.33340600	9.30848400	17.73494900
N	11.96995100	11.25471500	18.29659400
N	13.28587800	13.56240200	16.56560700
O	10.53869300	8.77794700	15.17829500
O	14.92874600	12.08204100	17.35955500
O	14.78480400	9.42675600	16.83209700
O	3.59520300	9.04189800	12.12352400
O	2.65348800	8.01240000	14.19149600
O	3.19482100	6.57625900	12.21794100
O	16.81829300	17.60631500	12.76634300
O	15.60329800	16.30244500	11.01432700
O	17.71560900	15.41056300	12.00852800
C	7.77249700	7.26929000	14.70674100
C	7.11349100	6.18998900	14.03361100
C	5.84274900	6.34988800	13.50823500
C	5.19296800	7.58817800	13.63486500
C	5.81322700	8.67286300	14.28814400
C	7.07990800	8.52411200	14.81554300
C	9.67940200	8.07815700	15.86130900
C	9.51819300	8.30753200	17.25055500
C	8.61372100	7.65783800	18.13251300
C	8.57312600	8.05168100	19.45430400
C	9.41364700	9.09696900	19.92262400
C	10.28037100	9.70591200	19.02172900
C	11.21393700	10.81154700	19.34068200
C	11.33065200	11.37016700	20.61420100
C	12.25250700	12.39567100	20.82358400
C	13.04024700	12.82613100	19.76034300
C	12.87034900	12.23021800	18.50543600
C	13.77999400	12.65541400	17.37545900
C	14.08629300	14.12823900	15.55955100
C	15.39542800	14.60962700	15.78952400
C	16.10979200	15.24796300	14.78008100
C	15.53392400	15.41458600	13.51281300
C	14.24085100	14.94891200	13.26308500
C	13.52454900	14.31982700	14.28248800
H	14.85198900	10.38194900	17.11433800
H	12.35639800	12.84535000	21.80658500
H	13.77874700	13.61172200	19.88476800
H	10.71622400	11.01336300	21.43230900
H	9.36875900	9.40602700	20.95952700
H	7.88958200	7.56760100	20.14558400
H	7.96668900	6.87135400	17.75780800
H	7.56094000	9.35659600	15.31933300
H	7.63295100	5.24075500	13.94500600
H	5.35057900	5.52756000	13.00087600
H	5.29791000	9.62458500	14.37511700
H	12.51423100	13.96728200	14.09214100
H	15.84317200	14.48976300	16.77045800
H	17.11213700	15.61567900	14.98196700
H	13.79172700	15.08099700	12.28441900
O	12.56758000	10.85719600	15.02136000
H	12.82039300	11.76463900	15.26093300
O	14.78507400	9.54963100	15.37424200

H	13.90218600	10.04679800	15.20800900
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Complex [(L5)Cu]²⁻ and related species:

[(L5)Cu] ²⁻ (d)	E= -2736.83333164 H		
Cu	9.00221100	14.67040800	0.97793100
C	9.62349300	11.46738400	2.74352500
C	9.86844100	10.09983600	2.58627800
H	10.81957600	9.68606400	2.90219500
C	8.89403300	9.27287700	2.02284600
H	9.09567500	8.21357600	1.89190700
C	7.66833700	9.81640000	1.62731000
H	6.90450400	9.18063100	1.18764400
C	7.42760900	11.18279200	1.77494200
H	6.49064900	11.61867000	1.44346600
C	8.40067400	12.03092300	2.32351600
C	7.46764100	13.99227600	3.32339400
C	7.43171500	15.50610300	3.21901400
C	6.77756200	16.38067000	4.08607600
H	6.20967400	16.00063000	4.92776600
C	6.88332500	17.75435000	3.83301600
H	6.38682200	18.46128900	4.49073200
C	7.62236900	18.23160500	7.42887700
H	7.70753100	19.29452900	2.54421600
C	8.24815800	17.29726100	1.91573600
C	9.07757300	17.54488200	0.70552100
C	9.42559300	18.76834100	0.13105400
H	9.08802500	19.70276000	0.56646600
C	10.21873100	18.75181100	-1.02323400
H	10.50113500	19.69099700	-1.48914900
C	10.64830400	17.54269000	-1.58509800
H	11.25956900	17.51820000	-2.48013500
C	10.26130800	16.35872100	-0.95775800
C	10.59746900	14.94948700	-1.41010300
C	10.24606800	12.64255900	-0.91698800
C	11.43367800	11.99362300	-0.54855400
H	12.22204200	12.58256300	-0.09047700
C	11.58947000	10.62037800	-0.73931900
H	12.51451800	10.13717000	-0.43625100
C	10.55312900	9.86823900	-1.30063100
H	10.66191900	8.79618300	-1.43763800
C	9.36811600	10.50000400	-1.68478300
H	8.56029900	9.92656500	-2.12539800
C	9.21521300	11.87771200	-1.50057700
N	8.19439600	13.42590500	2.35536400
N	8.12588700	15.98743600	2.18984700
N	9.50963900	16.40438000	0.14035800
N	10.06474800	14.00960800	-0.62320600
O	6.86492400	13.41950300	4.26938600
O	11.29807300	14.77976800	-2.44331000
O	11.30887600	13.54394400	2.54961500
O	10.23731600	13.08986000	4.75501600
O	12.01050800	11.56778000	3.88676300
O	7.03485600	13.30700700	-0.87721400
O	6.85104400	11.57594900	-2.65423200
O	8.10486700	13.68593900	-3.09728900
S	10.88567700	12.49808600	3.53733000
S	7.68943200	12.67447800	-2.06867400

[(L5)Cu(OH)] ³⁻ (d)	E= -2812.783247 H		
Cu	7.77402000	12.64565100	1.83947300
C	9.02465500	9.12933000	1.75208800
C	9.63861200	8.14622200	0.96977900
H	10.61959900	7.77701100	1.24722900
C	8.99120700	7.64630800	-0.16236700
H	9.47796400	6.89130500	-0.77304400
C	7.72174100	8.12374900	-0.50251300
H	7.20973800	7.74047600	-1.38112600
C	7.11355400	9.11057500	0.27381900
H	6.14104000	9.50818800	0.00150000
C	7.75716400	9.63761900	1.40333200

SUPPORTING INFORMATION

C	7.15094400	15.51356200	3.17994800	C	11.06166100	15.21252900	-1.56184000
C	6.35563000	16.44562700	3.83690800	C	10.66038100	12.85140100	-1.40858200
H	5.66841400	16.12599200	4.61166400	C	12.03645600	12.44255300	-1.35651400
C	6.48225600	17.78697800	3.46539900	H	12.80881000	13.20181900	-1.36593600
H	5.87860000	18.54611100	3.95210100	C	12.37140000	11.10305300	-1.30797300
C	7.39130200	18.15651900	2.47457100	H	13.41596500	10.80830200	-1.28202100
H	7.49405700	19.19827800	2.19663500	C	11.36636700	10.12195300	-1.26710700
C	8.16037300	17.16380200	1.85077600	H	11.63237600	9.07094100	-1.21159800
C	9.16603800	17.38989900	0.78641000	C	10.00633700	10.49128200	-1.26157700
C	9.52828800	18.67400000	0.36812500	H	9.23980100	9.72922300	-1.18092400
H	9.09219500	19.55028300	0.83162100	C	9.65209800	11.82280500	-1.33300500
C	10.46550100	18.81371000	-0.64995300	N	8.07723000	13.33629600	2.77103900
H	10.76999300	19.80016700	-0.98452400	N	7.99888300	15.88874700	2.20220100
C	10.98320100	17.66975600	-1.25032100	N	9.70135300	16.28271800	0.21950700
H	11.68472100	17.73202200	-2.07339100	N	10.25142000	14.11920700	-1.47284100
C	10.56803000	16.41485400	-0.79735600	O	6.43192700	13.60035100	4.39613100
C	11.05872200	15.20105800	-1.55382000	O	12.03162400	15.31670600	-2.32607500
C	10.64564600	12.84138500	-1.39648500	O	10.75164100	13.26917600	1.48395600
C	12.01910600	12.42397400	-1.34470200	O	10.93197600	12.58411300	3.88036800
H	12.79622900	13.17827000	-1.36007900	O	11.87044700	11.14126300	2.06408600
C	12.34508000	11.08263000	-1.28914000	O	7.76179300	13.21358300	-0.08841600
H	13.38767400	10.78100400	-1.26364100	O	7.15990400	11.00141000	-1.03586600
C	11.33359300	10.10840200	-1.24100900	O	7.57490100	12.92496000	-2.56563400
H	11.59280000	9.05599900	-1.17999100	S	10.83062500	12.11325500	2.47296100
C	9.97588700	10.48657700	-1.23567900	S	7.90076800	12.28010900	-1.25546300
H	9.20449500	9.73002100	-1.14971100				
C	9.63053100	11.81990500	-1.31438500				
N	8.12013000	13.34084200	2.81081600	[(L7)Cu] (dos)	E= -3043.74310150 H		
N	8.00241400	15.88277700	2.21159900	Cu	11.15256800	11.17616200	17.33940200
N	9.70312700	16.27577000	0.22872200	N	9.33021100	10.80249600	16.29087900
N	10.24419200	14.11102900	-1.46537900	N	9.96955300	10.49936500	18.80116500
O	6.42464400	13.58037800	4.38031400	N	12.41283900	11.13208900	18.89538700
O	12.03260900	15.30168300	-2.31323700	N	12.96852100	11.88198500	16.44857200
O	10.75911000	13.27299300	1.49092500	O	7.17774300	10.01714900	16.71113400
O	10.95922000	12.62110500	3.89805300	O	15.17612300	12.30597700	17.05938400
O	11.90449800	11.16589500	2.09645300	C	8.30273100	10.35945300	17.09542000
O	7.75180700	13.22250200	-0.06525600	C	8.70028700	10.19340600	18.53229600
O	7.13247300	11.01703700	-1.01723600	C	7.84469100	9.74027700	19.53444800
O	7.55909100	12.94097400	-2.54229700	C	8.36064700	9.61445900	20.82762300
S	10.85649700	12.13355200	2.49656800	C	9.69787600	9.92798000	21.08891500
S	7.88261100	12.28998100	-1.23508700	C	10.49517600	10.37737100	20.03414000
				C	11.92838900	10.74201100	20.08815900
				C	12.76764200	10.711101400	21.20363000
[(L5)Cu] (q)	E= -2736.633847 H			C	14.10409200	11.08949100	21.04313300
Cu	9.13908200	14.51404500	1.37495800	C	14.58020100	11.49124400	19.79166100
C	9.23637800	11.27575200	2.33849100	C	13.68233200	11.49943500	18.72539600
C	9.20648700	9.92857200	2.06512800	C	14.03438300	11.91058400	17.32755300
H	10.12487200	9.39718900	1.84221200	H	14.77359800	11.07143600	21.89718800
C	7.97609600	9.23335200	2.08891800	H	15.61089100	11.79067200	19.64118900
H	7.96030300	8.17540300	1.84742400	H	12.39024600	10.40036400	22.17139000
C	6.78666900	9.89955000	2.42593200	H	10.10934500	9.82143400	22.08629300
H	5.84669800	9.35743300	2.43253300	H	7.72317700	9.26623000	21.63399800
C	6.80019200	11.25044100	2.71645900	H	6.81333600	9.49944900	19.30389000
H	5.87619700	11.77946000	2.91039300	C	9.13344400	11.12361200	14.98576600
C	8.02632200	11.99497000	2.65162200	C	8.05891000	11.96700100	14.50085200
C	7.13675200	14.06380200	3.50256000	C	10.06909000	10.59916200	14.05294100
C	7.14147100	15.52036000	3.16573400	C	7.28438100	12.74170700	15.38174900
C	6.34360300	16.45331600	3.81912800	C	7.85238800	12.09494800	13.08639100
H	5.65552900	16.13511900	4.59366400	C	9.85758900	10.71309900	12.69014800
C	6.47033900	17.79416900	3.44599500	H	10.91991200	10.04569200	14.43646200
H	5.86285300	18.55330500	3.92789500	C	6.28980100	13.58981000	14.90241800
C	7.38427100	18.16275200	2.45950600	H	7.48363100	12.70864400	16.44669500
H	7.48430400	19.20343200	2.17705200	C	6.81967000	12.94951600	12.62831300
C	8.15898000	17.16939700	1.84313800	C	8.74693500	11.41167100	12.20117600
C	9.17058000	17.39605500	0.78374100	H	10.54768100	10.24772400	11.99659600
C	9.54514500	18.68112400	0.37852100	C	6.04939000	13.68108600	13.52450300
H	9.11464600	19.55717100	0.84793700	H	5.70763800	14.18475400	15.59944900
C	10.48957700	18.82205600	-0.63272600	H	6.63827900	13.05873500	11.56710900
H	10.80522300	19.80882800	-0.95586400	H	5.27181800	14.33917500	13.14887300
C	11.00115200	17.67921000	-1.24031200	S	8.49210200	11.43048600	10.39562800
H	11.70939400	17.74287900	-2.05749300	O	7.09657400	10.92307800	10.21208200
C	10.57313600	16.42390000	-0.80032400	O	8.65919100	12.85740000	9.98275500

SUPPORTING INFORMATION

O	9.52975400	10.52275700	9.83004500	C	12.32529900	9.68540000	12.28324500
C	13.20701600	12.01986300	15.11512100	C	13.38762000	11.91392200	12.28141500
C	12.29383000	12.77527200	14.28843500	H	14.26762200	13.89070800	12.47263300
C	14.31707600	11.37346200	14.49744300	C	12.42296900	9.63426400	10.90177400
C	11.22854900	13.49806000	14.86015600	H	11.86627800	8.86983700	12.83349900
C	12.49285000	12.82551400	12.86774700	C	13.48602800	11.84179300	10.88722200
C	14.49313000	11.41326600	13.12466000	C	13.00969900	10.71024000	10.21552900
H	14.98957800	10.78035200	15.10437300	H	12.04289100	8.77917700	10.35429800
C	10.35411700	14.23719400	14.06539600	H	13.92560800	12.66589900	10.33364400
H	11.08382600	13.47270300	15.93496300	S	8.46062600	11.04375300	8.08407300
C	11.57542400	13.56756100	12.08508800	S	13.24408400	10.60134300	8.42470500
C	13.60566000	12.12179100	12.30861100	O	7.01951500	11.46068800	8.04682100
H	15.31923200	10.87369600	12.67659800	O	8.73411700	9.82571900	7.26279700
C	10.52378100	14.26119800	12.67712900	O	9.40057100	12.17432600	7.81732900
H	9.52888500	14.77046200	14.52595500	O	12.22656800	9.62352200	7.93570700
H	11.68531500	13.60268600	11.00944600	O	13.04417600	11.98873800	7.90243200
H	9.82764800	14.81045400	12.05187000	O	14.65400500	10.11492000	8.25794200
S	13.90915600	12.10066300	10.51357700				
O	12.68366300	11.47651000	9.92913200				
O	14.08177100	13.53551900	10.12755300				
O	15.14236300	11.28426000	10.31490500				

[(L8)Cu] (q) E= -3043.74912970 H

Cu	11.23093400	11.01393500	17.27515800
N	9.62035300	10.23780000	16.15356800
N	10.18197200	10.07302200	18.67932700
N	12.17730300	11.59533700	18.92011000
N	12.86086000	12.03484500	16.45370100
O	7.88423000	8.68795500	16.39077500
O	14.57573200	13.47381100	17.14868000
C	8.83256300	9.33585200	16.84662100
C	9.16216300	9.29404100	18.31618900
C	8.45356400	8.56463300	19.27004000
C	8.84012000	8.69124300	20.60874800
C	9.90193000	9.52844600	20.96827700
C	10.57321100	10.21946100	19.95767800
C	11.72624000	11.14345400	20.10283400
C	12.35245300	11.55566700	21.28133500
C	13.43436500	12.43822900	21.18407800
C	13.88423700	12.88281700	19.93592100
C	13.21756100	12.42047900	18.80199200
C	13.61460100	12.73496800	17.38235300
H	13.93340500	12.77429500	22.08743700
H	14.72870900	13.55570500	19.84074300
H	12.00597600	11.19985400	22.24534800
H	10.20155700	9.63672200	22.00502500
H	8.30884500	8.14067900	21.37853900
H	7.62611000	7.93042900	18.97341200
C	9.44827900	10.50080900	14.83097900
C	9.03124300	9.52848800	13.87045400
C	9.78515800	11.81982100	14.38159500
C	8.94453000	9.86216100	12.50929600
H	8.84772800	8.50863700	14.18451500
C	9.64876700	12.16646500	13.06711200
H	10.11967600	12.53904400	15.12183700
C	8.58557500	8.88739800	11.53498700
C	9.23247400	11.20386200	12.08580500
H	9.86822100	13.17900900	12.74081700
C	8.48150000	9.23281700	10.19778200
H	8.38807300	7.86878600	11.85594800
C	9.12361000	11.53451400	10.73119300
C	8.73863700	10.55869000	9.80624500
H	8.20517700	8.49008600	9.45760400
H	9.34247600	12.54599700	10.40379100
C	13.10362300	12.06844800	15.11443100
C	12.66227400	10.91897500	14.38553300
C	13.72420600	13.13745800	14.38948200
C	12.80277100	10.81897800	12.99578200
H	12.22552400	10.09633100	14.94225500
C	13.83817100	13.05921300	13.02511100
H	14.05539000	14.02149900	14.91579800

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