



RCSI

Leading the world
to better health

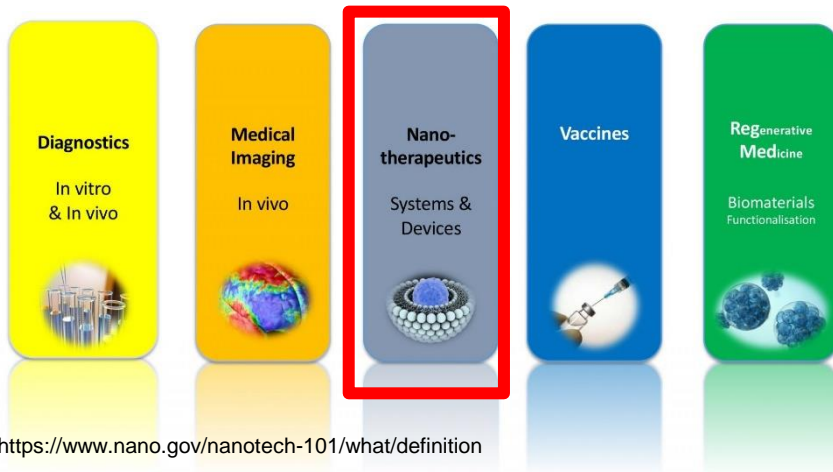
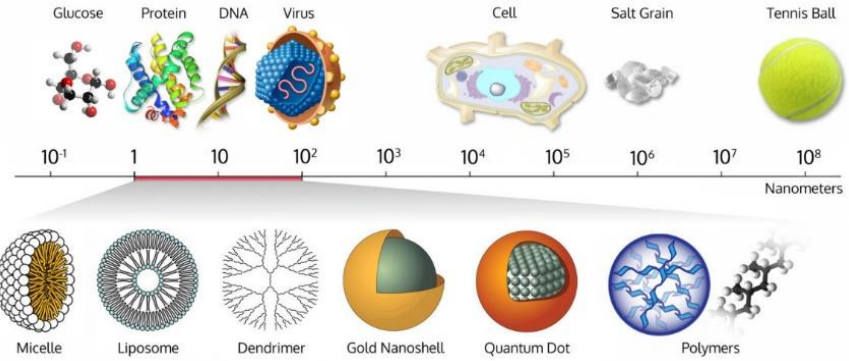
Nanoparticles *via* SET-LR-PISA

D. V. Tomasino
Supervisor A. Heise



Nanomedicine

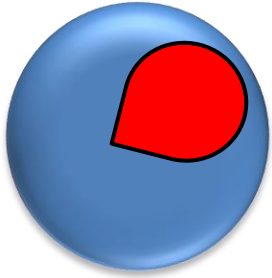
“Nanotechnology is the science conducted at the nanoscale, about 1 to 100 nm”



“Nanomedicine is the application of nanotechnology for medical purposes”

NPs in biological fluid

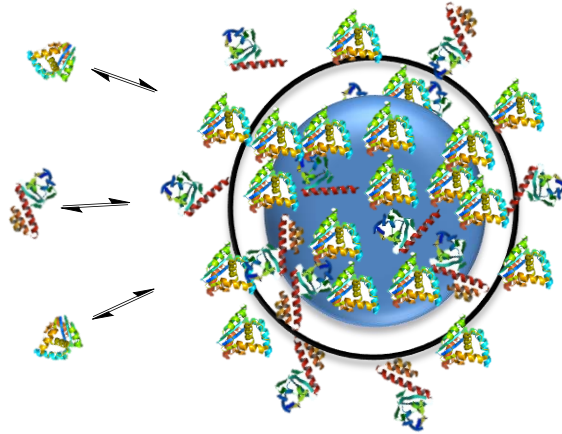
Polymeric NPs



Classified in terms of

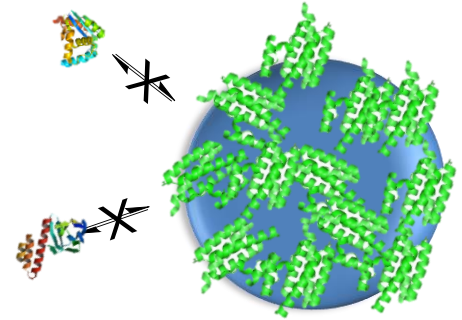
- Type of polymers
- Surface functionalisation
- morphology

Biological environment



- Formation of protein corona layer.
- New biological identity.
- Poor biodistribution.

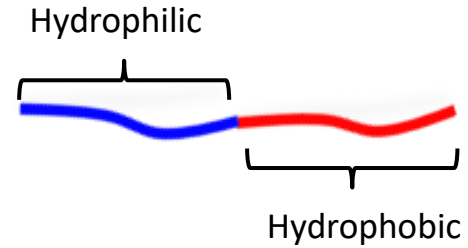
Surface modification



- Creation of artificial corona layer.
- **Glycopolymers.**

Amphiphilic block copolymer

Amphiphilic block copolymers comprise of discrete immiscible hydrophobic and hydrophilic blocks which undergo phase separation in aqueous conditions.



Microphase separation promoted by;

Hydrophilic repulsion \longrightarrow increase hydrophilic interactions with water

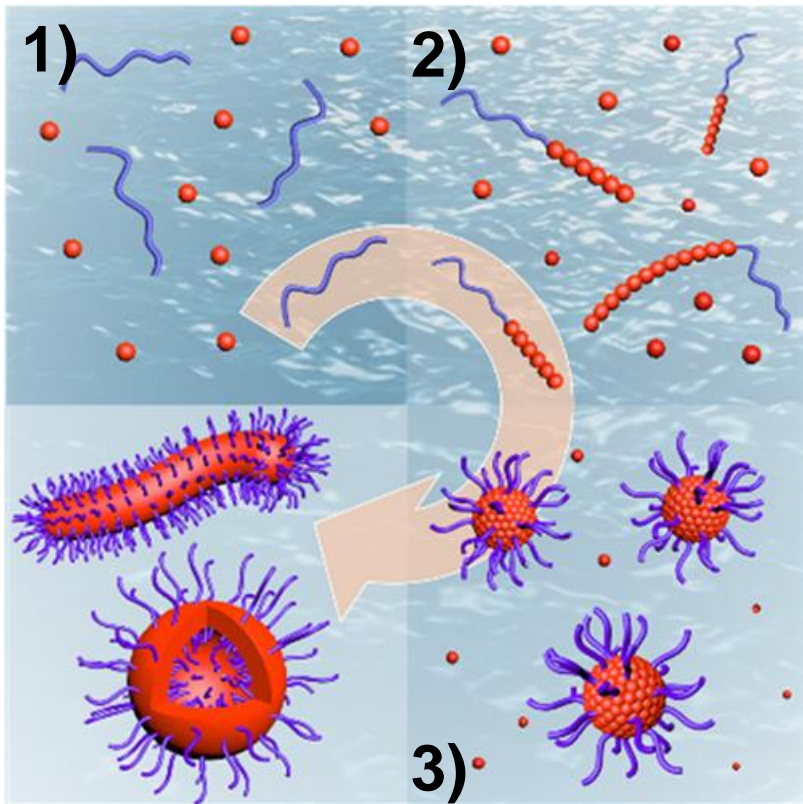
Hydrophobic attractions \longrightarrow decrease hydrocarbon water interface



f_{red}

NPs synthesis by Polymerisation-induced self-assembly (PISA)

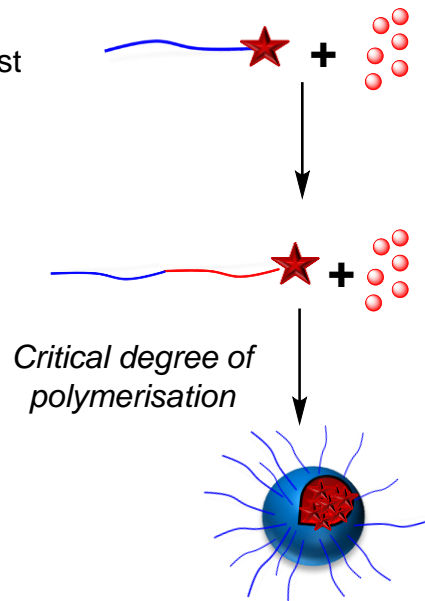
Carried out *via* controlled living radical polymerisation (*i.e.* RAFT, ATRP).



1) Synthesis of macro initiator (first block).

2) Chain extension.

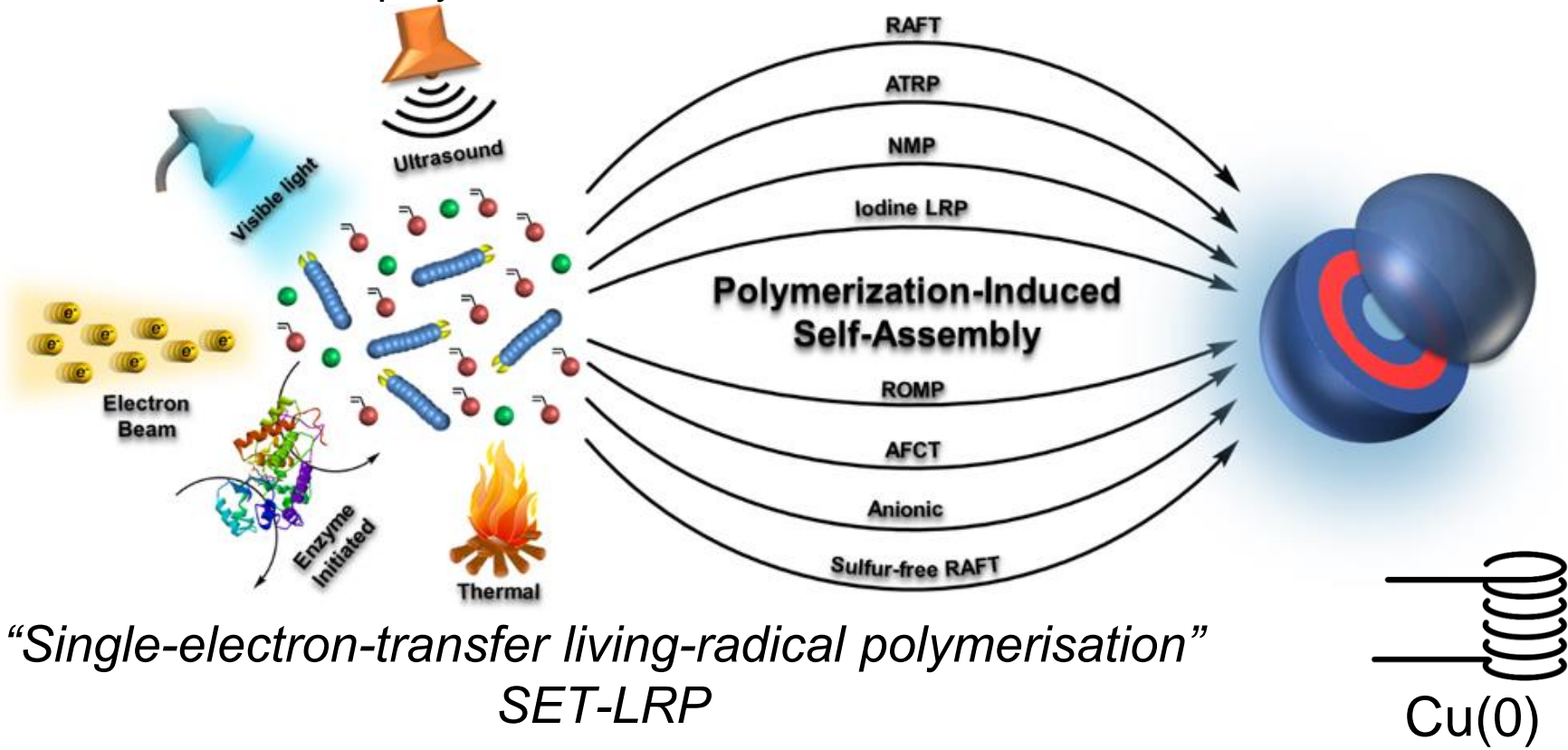
3) *In situ* self-assemble.



- ✓ Up to 50 w/w% solid concentration.
- ✓ Good candidate for industrial scale up.
- ✓ Nano-object of controlled size and morphology without further processing steps.

Polymerisation methods for PISA

All controlled radical polymerization methods suitable



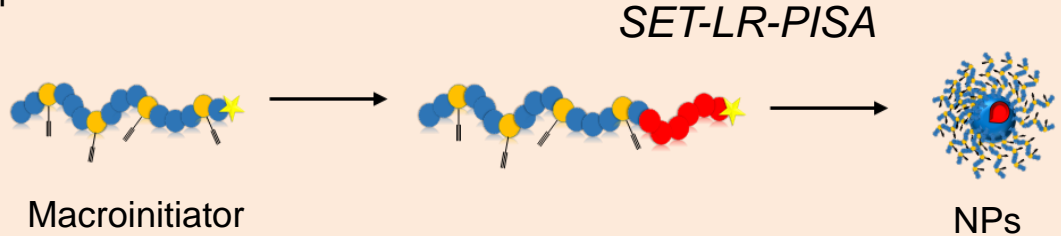
“Single-electron-transfer living-radical polymerisation”
SET-LRP

Project stages

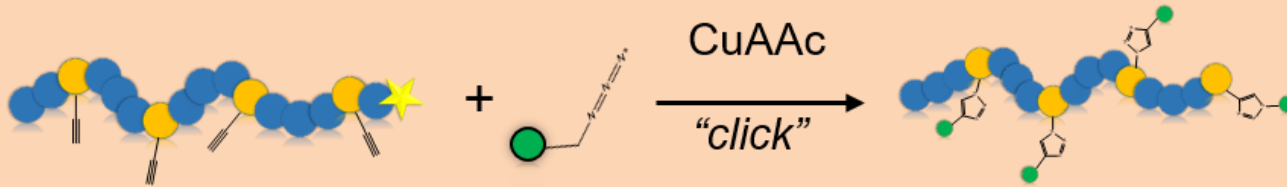
A) Preparation of NPs *via* aqueous dispersion SET-LR-PISA

Morphology evaluation:

- Core forming monomer concentration $[M]_0$
- Solvent

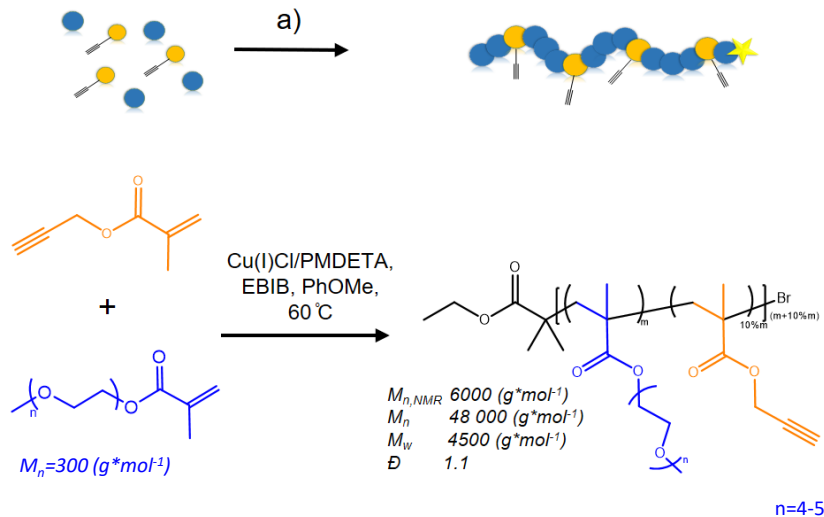


B) Macroinitiator functionalisation with mannosides via alkyne-azide copper catalysed cycloaddition:



Optimisation of NPs via aqueous dispersion SET-LR-PISA

SYNTHESIS MACROINITIATOR

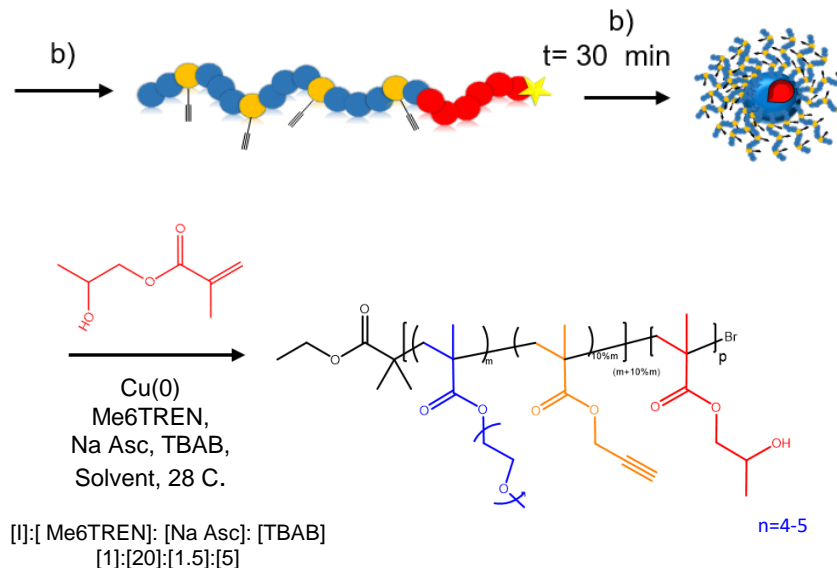


a) Conventional ATRP

- $DP_{PEGMA} = 20$
- $DP_{PgMA} = 5$



SET-LR-PISA



b) SET-LR-PISA

- $DP_{HPMA} = 150; 300$
- Solid content (SC)=10%

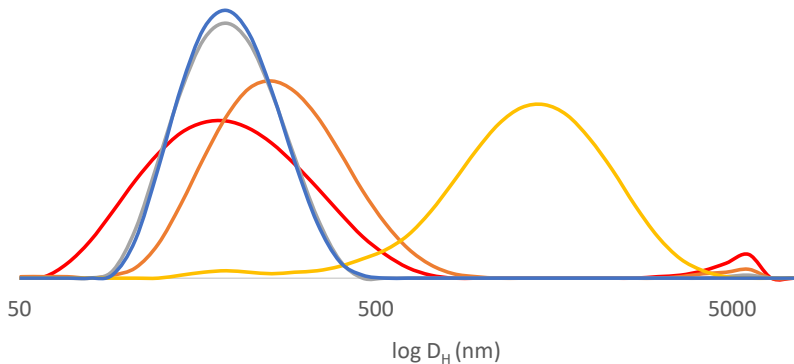
Followed by: DLS, TEM

Parameters:

- Solvent (H₂O vs. PBS buffer)
- Monomer concentration

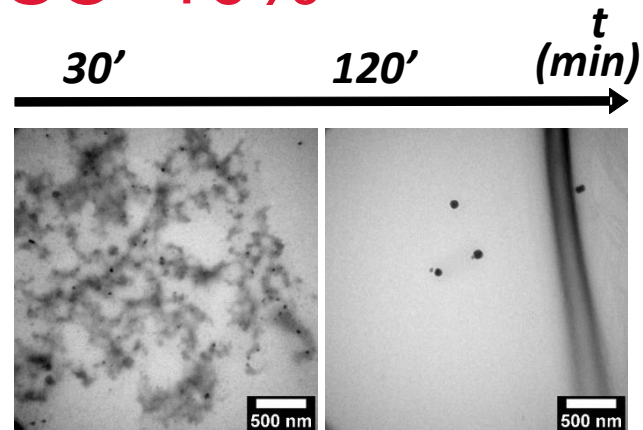
Influence of solvent: $DP_{HPMA}=150$ $SC=10\%$

• H₂O

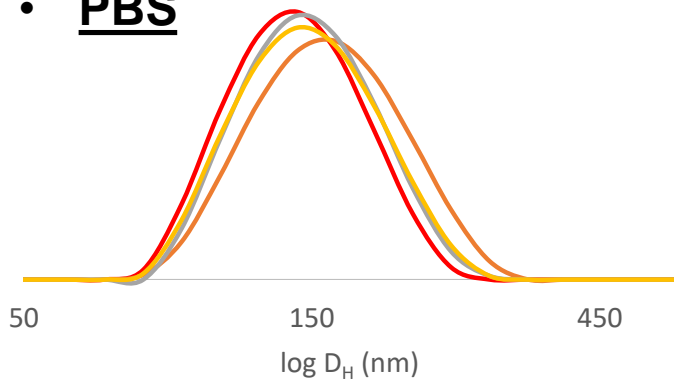


— 30 min — 60 min — 90 min — 120 min — 150 min

| Time (min) | D _H (nm) | PDI |
|------------|---------------------|------|
| 30 | 186.6 | 0.33 |
| 60 | 254.2 | 0.20 |
| 90 | 185.6 | 0.11 |
| 120 | 1100 | 0.27 |
| 150 | 188.8 | 0.14 |

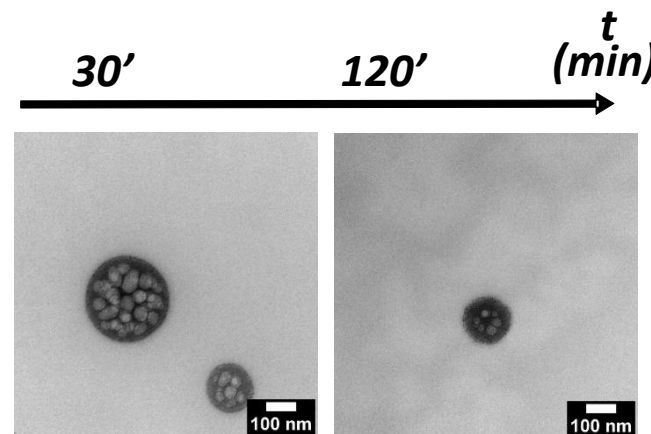


• PBS



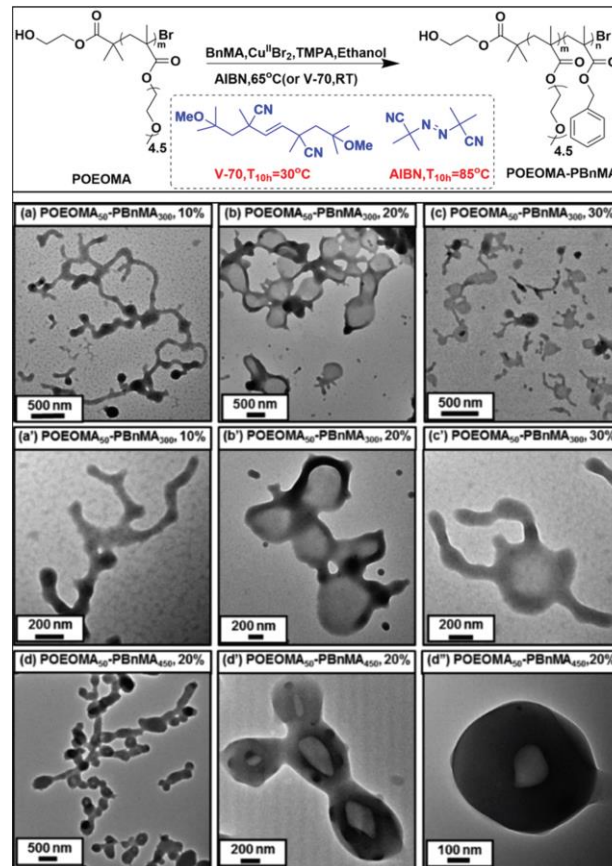
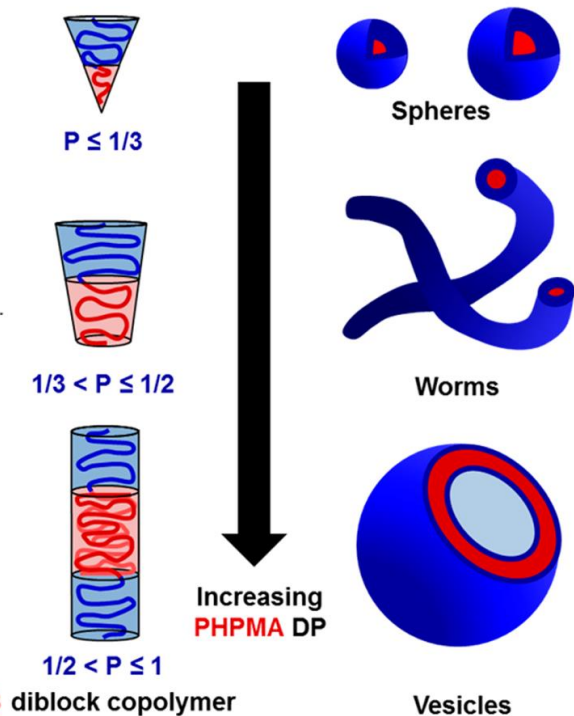
— 30 min — 60 min — 90 min — 120 min

| Time (min) | D _H (nm) | PDI |
|------------|---------------------|------|
| 30 | 136.9 | 0.04 |
| 60 | 153.2 | 0.06 |
| 90 | 143.1 | 0.08 |
| 120 | 142.3 | 0.06 |



Influence of monomer concentration

- Traditional morphology evolution



Warren, N. J.; Armes, S. P., Polymerization-induced self-assembly of block copolymer nano-objects via RAFT aqueous dispersion polymerization. *J Am Chem Soc* **2014**, *136* (29), 10174-85.

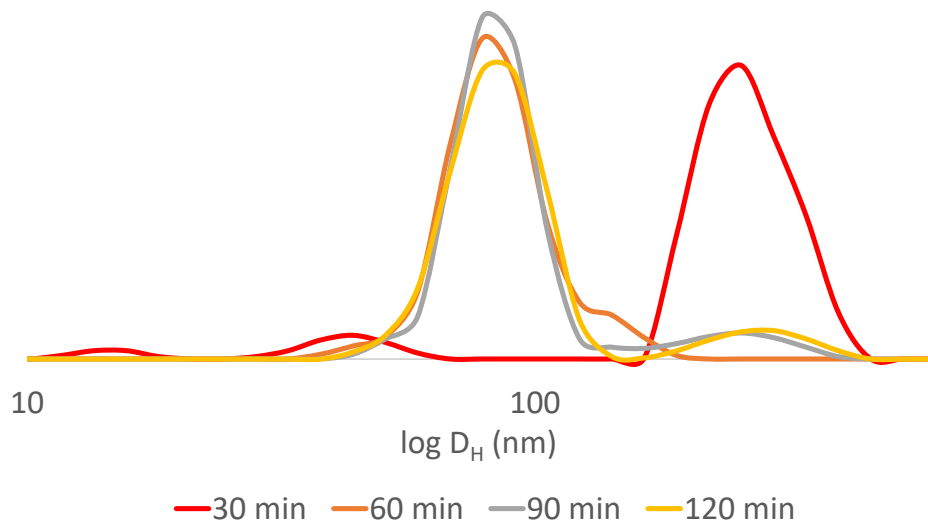
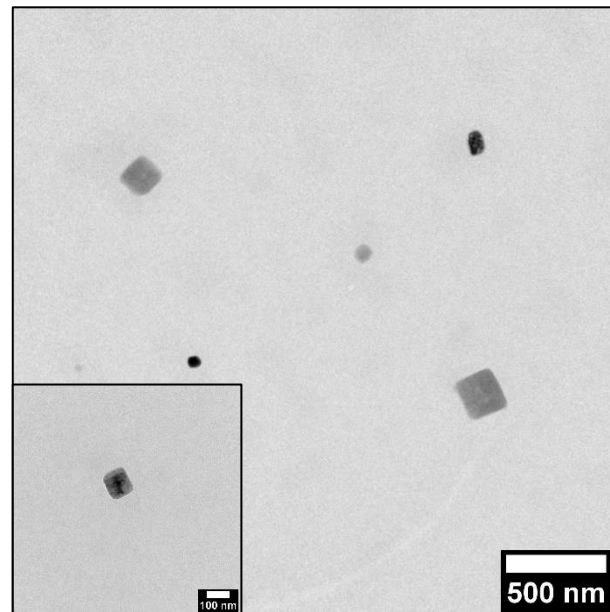
Wang, G.; Schmitt, M.; Wang, Z.; Lee, B.; Pan, X.; Fu, L.; Yan, J.; Li, S.; Xie, G.; Bockstaller, M. R.; Matyjaszewski, K., Polymerization-Induced Self-Assembly (PISA) Using ICAR ATRP at Low Catalyst Concentration. *Macromolecules* **2016**, *49* (22), 8605-8615.

Influence of monomer concentration

- **H₂O**
- **DP_{HPMA} = 300**
- **SC = 10%**

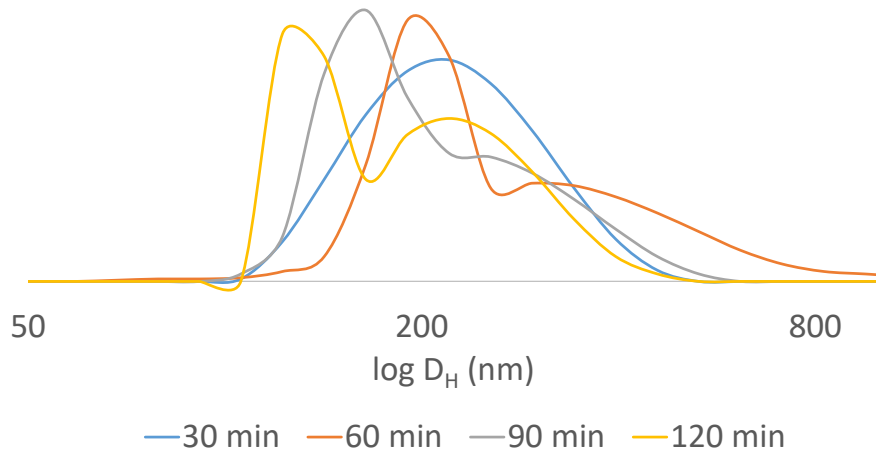
| Time (min) | D _H (nm) | PDI |
|------------|---------------------|------|
| 30 | 565.1 | 0.53 |
| 60 | 398.4 | 0.50 |
| 90 | 400.7 | 0.51 |
| 120 | 409.2 | 0.45 |

- 120 min

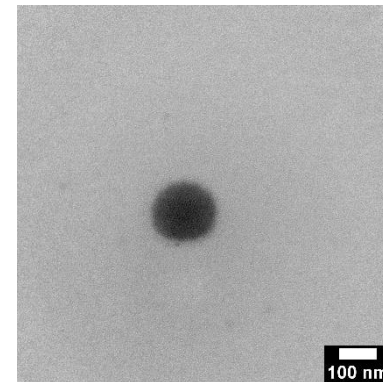
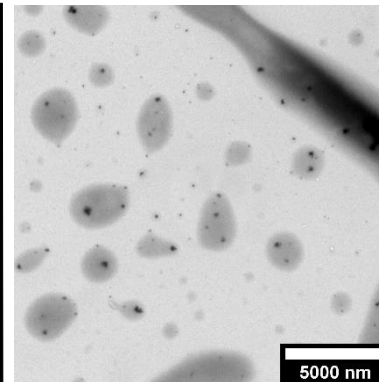


Influence of monomer concentration

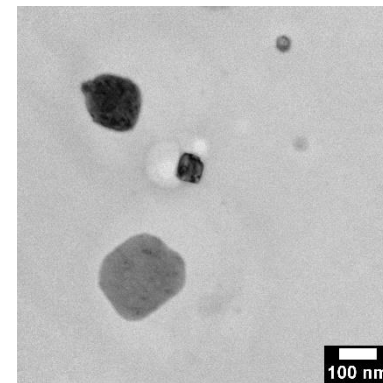
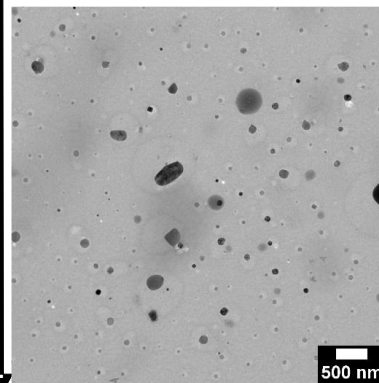
- **PBS**
- $DP_{\text{HPMA}} = 300$
- **SC=10%**



30'



120'



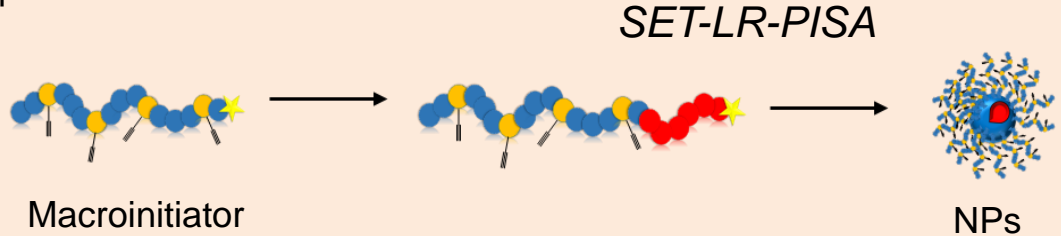
t
(min)

Project stages

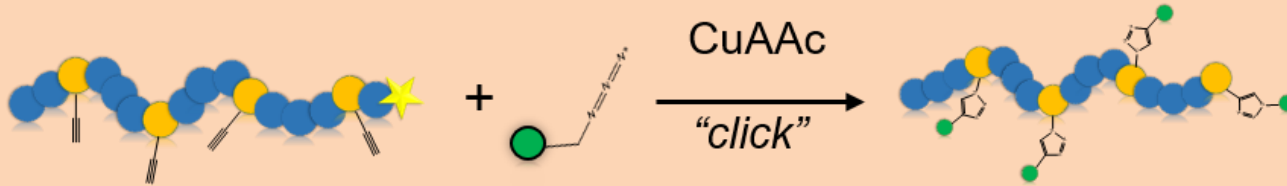
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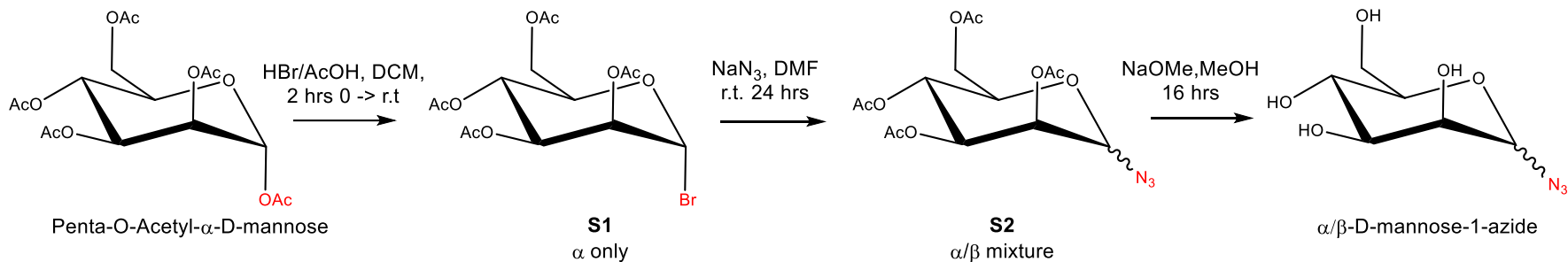


B) Macroinitiator functionalisation with mannosides via alkyne-azide copper catalysed cycloaddition:

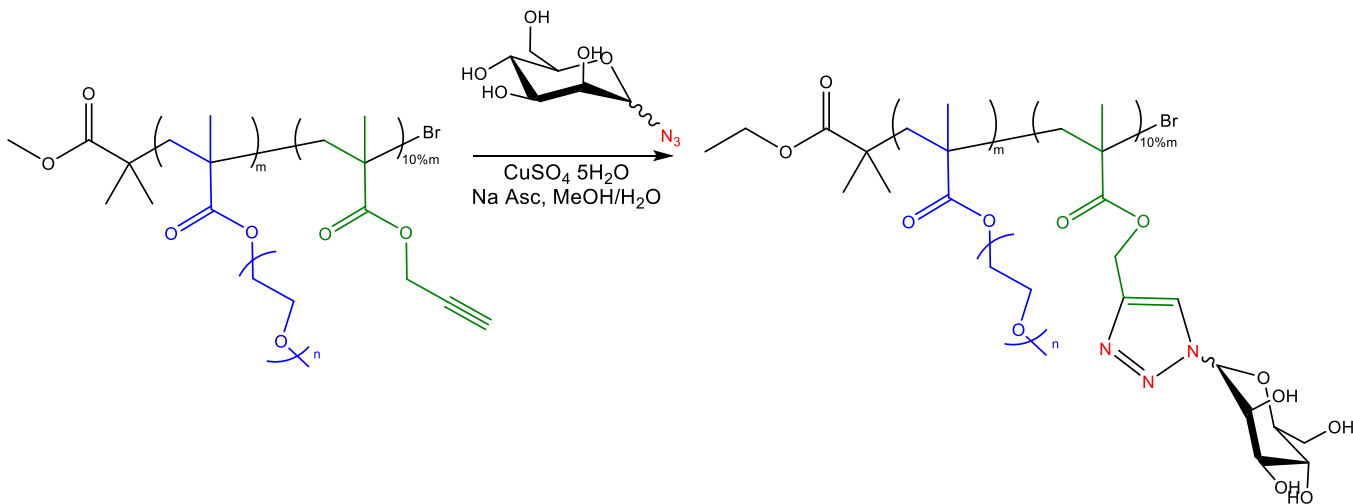


Glycosylated macroinitiator for SET-LR-PISA system

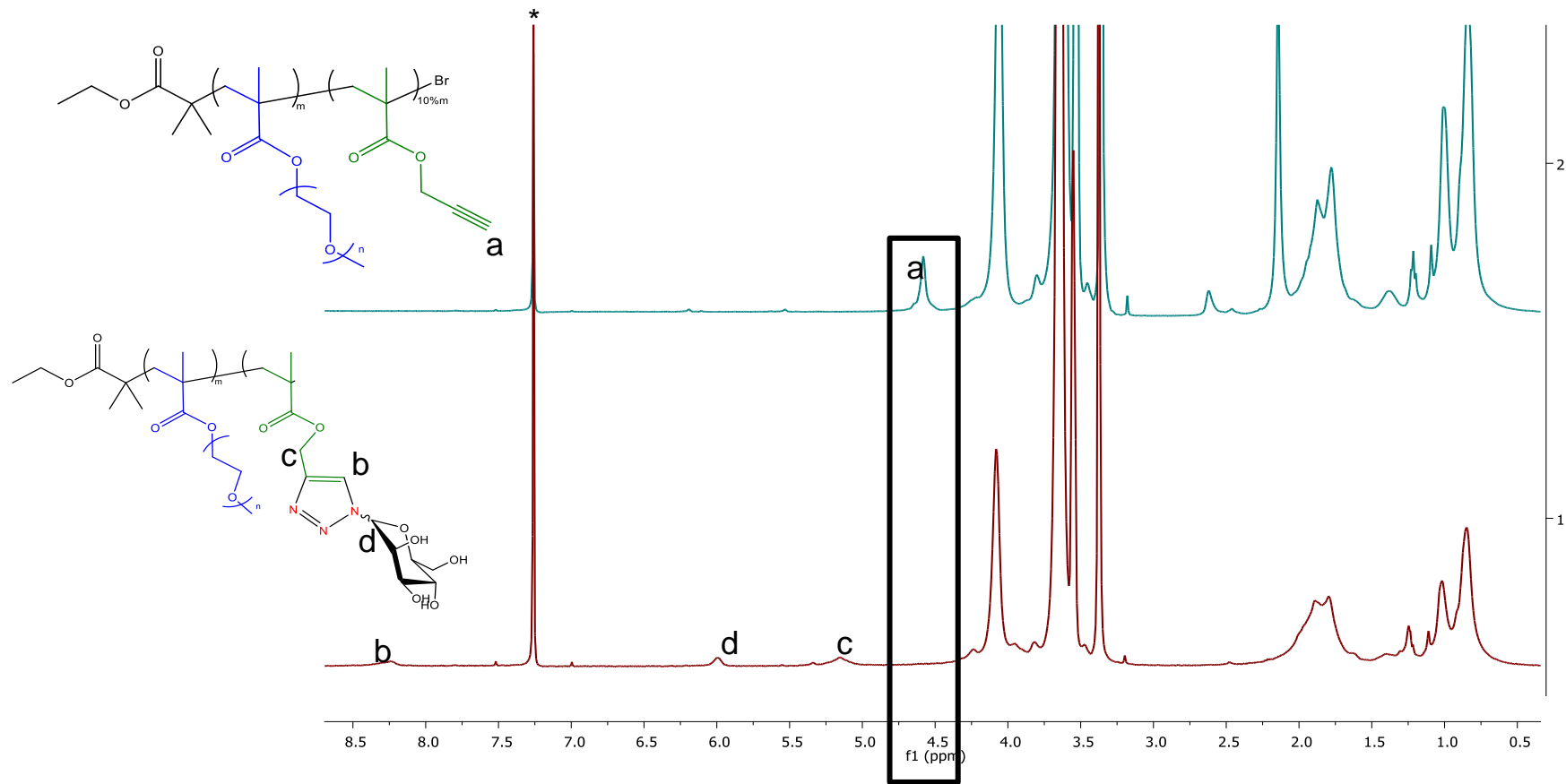
1)



2)

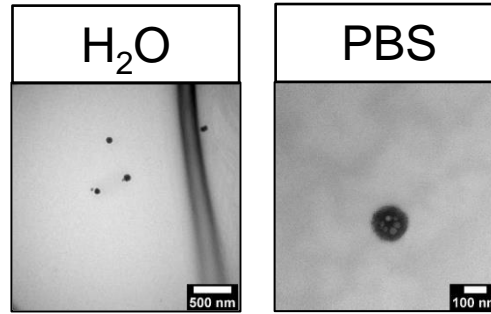


Glycosylated macroinitiator: racemic mixture



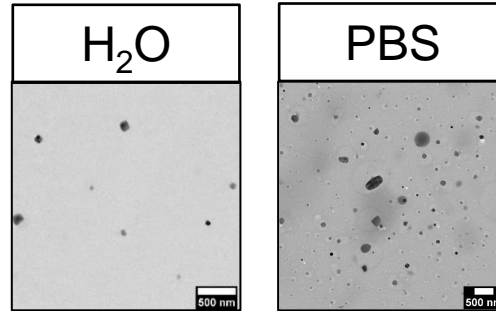
Conclusion

✓ Spherical morphology obtained when targeting $DP_{\text{HPMA}}=150$ after 120 min



✓ PBS provides more stable system

✓ Mixture of cubosomes and spherical NPs obtained when targeting $DP_{\text{HPMA}}=300$



✓ Macroinitiator suitable for glycosylation with mannosides



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Dr Shona O'Brien

Dr Rafał Bielas

Smiljana Stefanovic

Nicola Judge

**Thank you
for your
kind
attention**



This project has received funding from European Union's Horizon 2020 research and innovation programme under the Marie Skłodowska-Curie grant agreement No 814236.

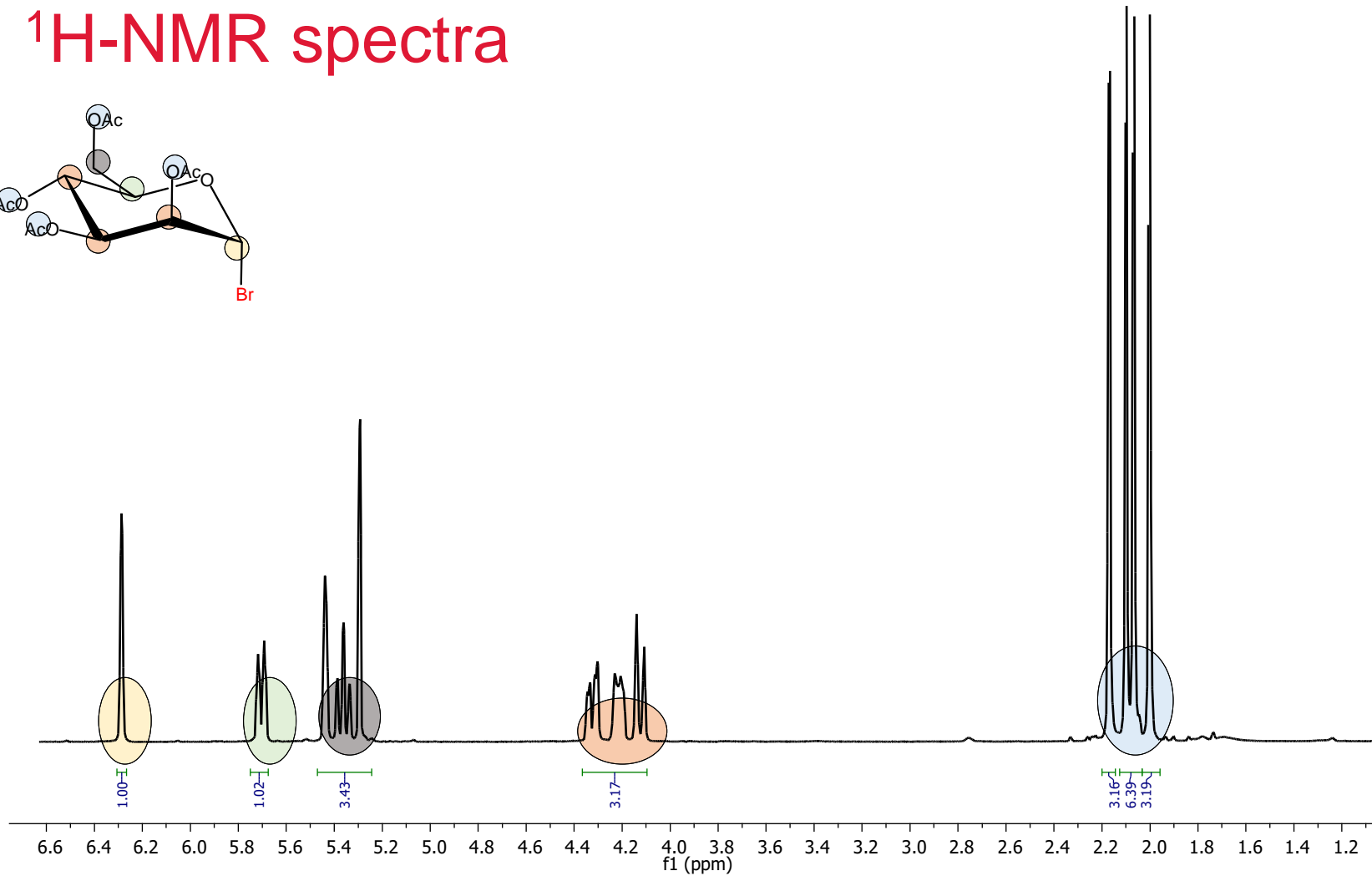
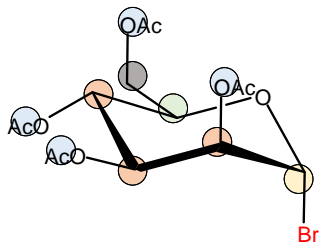




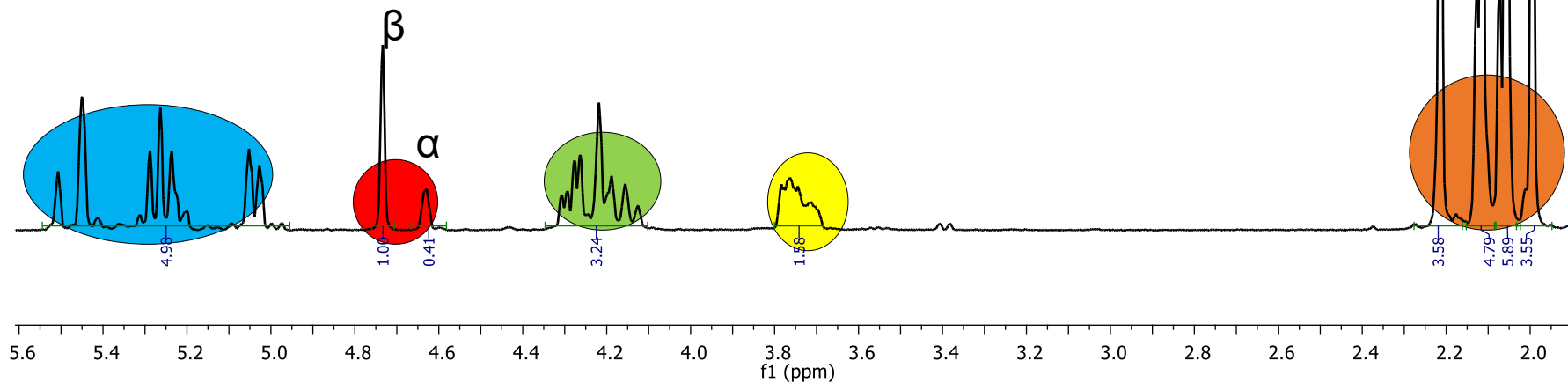
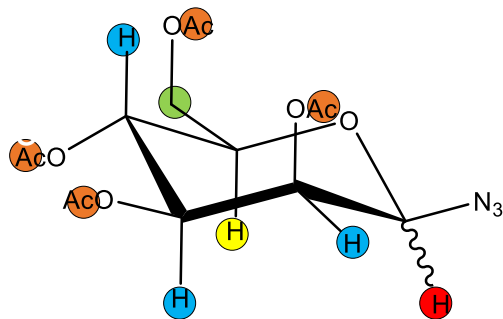
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Thank you

S1 $^1\text{H-NMR}$ spectra

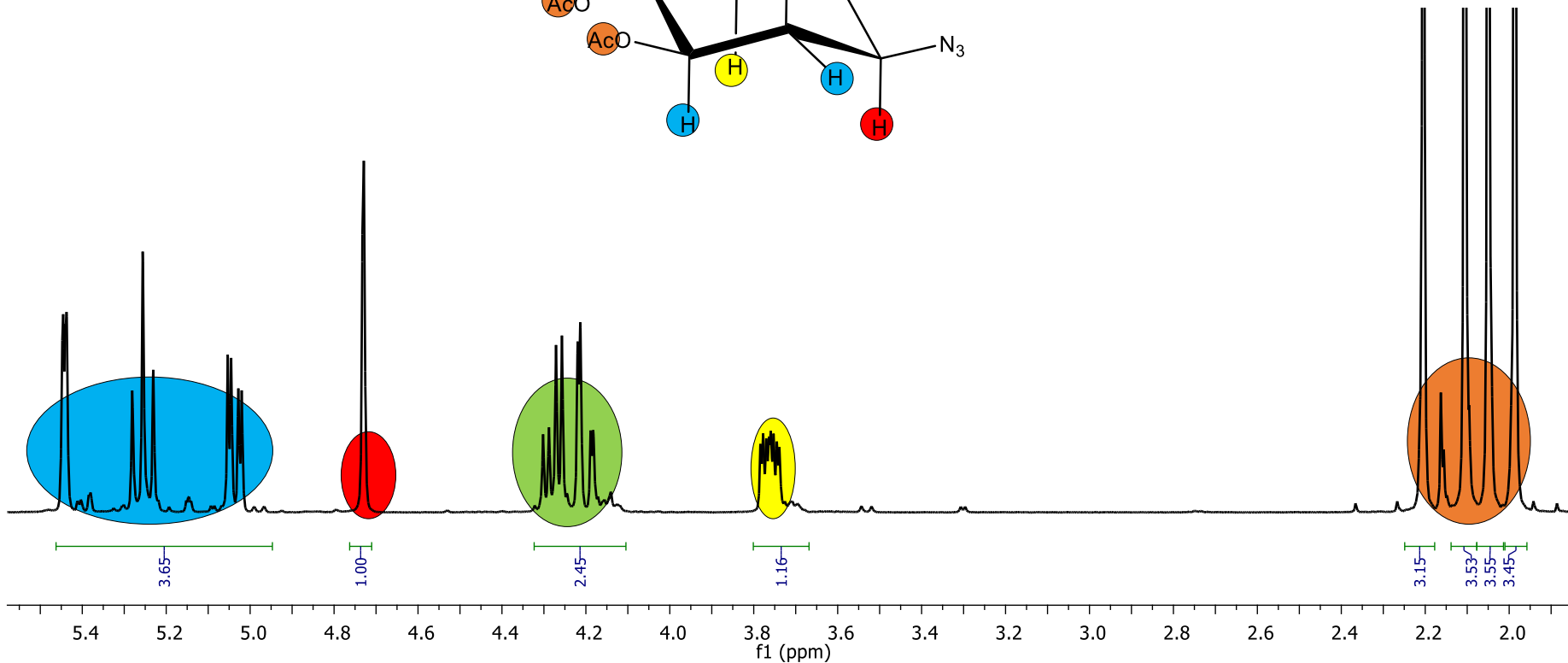
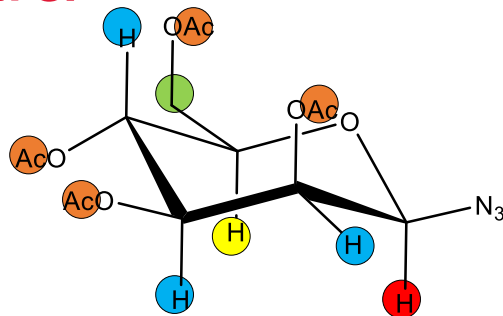


S2 ^1H -NMR spectra



S2 $^1\text{H-NMR}$ spectra

Reaction carried out at 70 °C



$^1\text{H-NMR}$ spectra $\beta\text{-D-mannose-1-azide}$

