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5 **OPTIMIZATION OF SHORT CHAIN VOLATILE FATTY ACIDS PRODUCTION FROM**
6 **HOUSEHOLD FOOD WASTE FOR BIOREFINERY APPLICATIONS**

7 *Giuseppe Strazzera^a, Federico Battista^{a, *}, Barbara Tonanzi^b,*

8 *Simona Rossetti^b, David Bolzonella^a*

9 ^a University of Verona, Department of Biotechnology, Strada Le Grazie 15, 37134, Verona, Italy

10 ^b Water Research Institute, IRSA-CNR, Via Salaria km 29,300, 00015 Monterotondo, Italy

11 *Corresponding Author: Dr. Federico Battista, e-mail contact: federico.battista@univr.it

12
13 **ABSTRACT**

14 Household Food Wastes (HFW) are the most abundant organic wastes at urban level with a worldwide
15 annual production of about 2 billion tons. This material can be the right feedstock for a carboxylate
16 biorefinery platform. This work investigated the influence of different operational parameters (pH,
17 temperature, Organic Loading Rate) on the Volatile Fatty Acids production and on their composition.
18 It was demonstrated that, keeping constant the fermentation duration at 6 days, neutral pH,
19 thermophilic temperature (55°C) and an organic loading rate of 11 gTVS per Liter per day maximized
20 the VFA production with a yield of 0.38 gVFA per gTVS fed. Under these conditions, the main VFAs
21 were butyric and caproic acids with percentages around 60 and 20%, respectively.3. The analysis of
22 the mixed microbial community performed over the reactor operation showed the dominance of

23 members of *Firmicutes* phylum mainly affiliated to *Clostridium* and *Lactobacillus* genera. The
24 process was mostly driven by *Clostridium* species in mesophilic runs characterized by stable VFAs
25 production and highest yields whereas *Lactobacillus* was enriched under thermophilic operating
26 conditions.

27

28 **Keywords:** Volatile Fatty Acids; Food wastes; Acidogenic fermentation; Microbial community
29 structure; process optimization; Bioeconomy; Biorefinery

30 **1. Introduction**

31 Decarbonization of modern society passes through a change in our economic models (WEF, 2018):
32 bioeconomy, where organic waste, by-products and residues rich in carbon can be used as feedstock
33 to obtain high added value bio-based molecules, will be an important part of this new paradigm (EEA,
34 2018). These residual streams can be the new oil feeding the chemical sector, now depending on fossil
35 oil and gas (JRC, 2019).

36 The worldwide production of solid wastes has reached the alarming level of 17 billion tons and it has
37 been estimated to be 27 billion tons within 2050 as consequence of the human population growth
38 (Laurent et al., 2013), with half of the world population living in cities, causing 70% of the total
39 emissions of carbon dioxide (Angeli et al., 2018). With an annual production of approximately 2
40 billion metric tons on a dry matter basis, household food wastes (HFW) are the most abundant organic
41 wastes (substrates) produced in the urban context (Battista et al., 2020). HFW can be defined as the
42 sorted organic fraction of wastes collected at domestic level. The abundance and presence of these
43 wastes, and the necessity to produce high added value molecules beside biogas and compost
44 (European Commission, 2008) make HFW an optimal candidate as renewable carbon sources for
45 feeding a multi-purpose and multi-product biorefinery (Battista et al., 2020).

46 In particular, HFW can be adopted as feeding materials for the production of volatile fatty acids
47 (VFAs) within the so called “carboxylate platform” by hydrolysis and fermentation carried out by
48 mixed microbial cultures (MMCs) (Monti et al., 2005). VFAs have several interesting applications,
49 as they are considered biological precursors for different biofuels and high added value compounds,
50 in particular bioplastics (Moretto et al., 2020). VFA are short-chain aliphatic mono-carboxylate
51 compounds with two (acetic acid) to six (caproic acid) carbon atoms, deriving from acidogenic
52 fermentation of organic wastes (Bhatia and Yang, 2017).

53 HFW include very heterogeneous types of fruit and vegetable, meat, fish and other carbohydrates
54 wastes, which influence the choice of best operating parameters to increase the VFAs production. In
55 particular, the variations of the Hydraulic Retention Time (HRT), the Organic Loading Rate (OLR),
56 the pH and the temperature determine different VFA yields and compositions. In particular, the
57 operational conditions, optimizing the VFA yield, seemed to be: slightly controlled pH (6.0–7.0),
58 short HRT (3-7 days depending on the substrates complexity), thermophilic temperatures and high
59 OLR (Strazzera et al., 2018, Battista et al. 2020). The combination of these parameters gives different
60 VFA yields and requires to be optimized for complex and heterogenous substrates, such as HFW.
61 The VFAs profile is also strongly influenced by HFWs chemical nature. Generally, lipids in HFWs
62 are less prone to fermentation than carbohydrates and proteins because of their low solubility and
63 slower biodegradation kinetics (Alibardi and Cossu, 2016). Furthermore, the hydrolysis of lipids
64 produces glycerol and long chain fatty acids (LCFAs) (Alibardi and Cossu, 2016; Shen et al., 2017).
65 On the contrary, carbohydrates are immediately available for glycolysis and fermentation into VFAs
66 (Moretto et al., 2019; Valentino et al., 2019). In particular, glucose conversion privileges the
67 formation of acetic acid, which is immediately used for the production of H₂ or for microbial growth
68 by microorganisms, resulting in a low final concentration of the mentioned acid. More complex
69 carbohydrates typically lead to a total VFA production where acetic acid, butyric and propionic acids
70 as most abundant VFAs. Proteins are generally characterized by lower kinetics due to their tertiary
71 and quaternary structures. Anyway, also in this case, acetic, propionic and butyric acids are the most

72 relevant products of protein fermentation, with a dominant presence of acetic acid, which usually
73 accounts for 70% of the total VFAs produced (Strazzera et al., 2018; Yin et al. 2016).

74 The main aim of this work is the individuation of the best operational parameters to exploit the huge
75 potential of HFW for the creation of a robust continuous process which can maximize the VFAs
76 production by MMC's acidogenic fermentation. In particular, different OLR, pH ranges and
77 mesophilic/thermophilic temperature conditions were tested for the conversion of HFW into VFAs
78 in a lab-scale reactor working in a continuous configuration. At the best of the authors knowledge,
79 the variations of the operational conditions were not previously performed in the same reactor, along
80 an experimental campaign of almost 450 days using the same batch of food waste. This aspect also
81 allowed the investigation of an element which is very important in an industrial full-scale process:
82 the ability of the system to react at eventual working stresses or changes. Finally, the influence of the
83 different operational parameters on microbial community dynamics and VFAs profile was evaluated.

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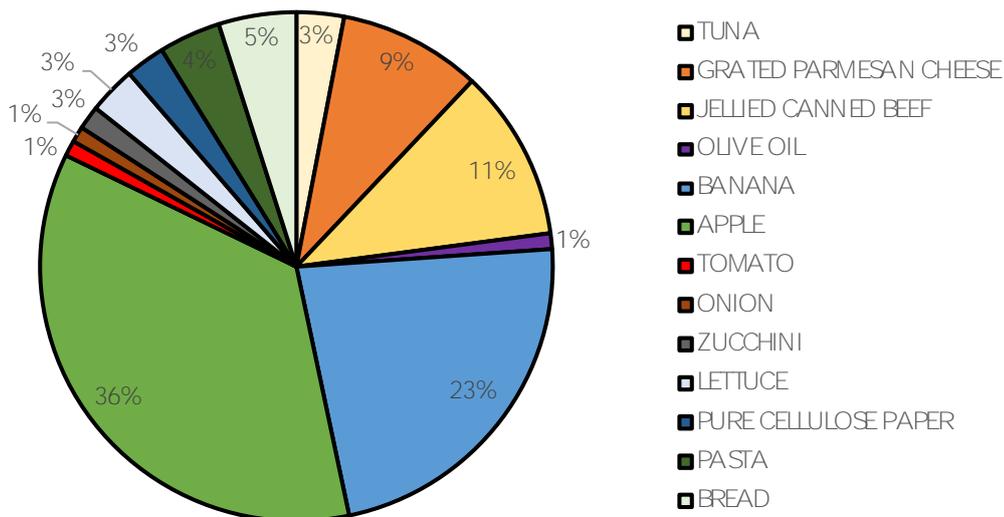
85 **2. Materials and Methods**

86 **2.1 Substrates and inoculum characterization**

87 HFW used in this study was appositely prepared to carry out the entire experimental work with the
88 same substrate, thus reducing the external variability of real HFW. The chemical compositions of
89 the different fresh food fractions considered to assemble the synthetic HFW, were taken from the
90 United States Department of Agriculture database (<https://www.usda.gov>).

91 In particular, to produce this synthetic HFW, the most common fresh foods in the Mediterranean area
92 and paper materials were mixed, according to the proportions reported in Figure 1, then they were cut
93 and homogenized by means of a professional blender (Cutter Mixer K25 produced by DitoSama).
94 The obtained synthetic HFW was stored at $-20\pm 1^{\circ}\text{C}$, until its usage.

95



96

97 Figure 1. The proportion (% w/w) of the single fresh food and paper materials used to assemble the
 98 synthetic HFW.

99

100 The synthetic HFW was formulated considering the most typical carbohydrates, lipids, proteins and
 101 fibers concentrations' ranges of real HFWs collected in Countries of the Mediterranean Area,
 102 considering data from the scientific literature (Garcia et al., 2005; Matsakas et al., 2014; Alibardi and
 103 Cossu, 2016). With reference to data reported in Table 1, this material is clearly equivalent to typical
 104 real HFW with a solid content around 25% and a COD content of some 290 g/kg. Nitrogen and
 105 phosphorus were at 9.9 and 1.4 g/kg dry matter with a C/N value of 29. Proteins, carbohydrates, and
 106 lipids were all well represented with average values around 23.9%, 21.1%, and 15.6%, respectively.

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115 Table 1. Physic-chemical characteristics and macromolecular composition of synthetic HFW

Parameter	Value	Parameter	Value (%TS)
pH	5.66 ± 0.18	Proteins	53.9 ± 1.0
TS (g/kg)	257.26 ± 13.92	Carbohydrates	21.1 ± 0.6
TVS (g/kg)	250.87 ± 13.03	Lipids	15.6 ± 1.0
COD* (g/kg)	292.42 ± 47.93	Fibers	31.6 ± 0.5
sCOD (g/L)	51.68 ± 8.82	Cellulose	10.7 ± 0.6
TKN* (g/kgTS)	9.90 ± 1.95	Hemicellulose	10.4 ± 1.0
TP* (g/kgTS)	1.45 ± 0.51	Lignin	10.5 ± 1.0
C/N	29	Sugars	30.2 ± 1.0

116 *on dry basis

117

118 The inoculum for the fermentation tests was represented by the endogenous microorganisms, already
 119 present in the food wastes. Along the fermentation, the most adapted microorganisms to the
 120 operational conditions of the different experimental runs survived and grew, influencing the VFA
 121 amount and profile.

122

123 2.2 The optimization of the VFA production from HFW

124 The study aimed at optimizing the operational parameters of the fermenter to maximize the VFAs
 125 production from acidogenic fermentation of the synthetic HFWs in a semi-continuous mode. The
 126 HRT was set up and kept constant at 6 days along the operation, as suggested by the good
 127 performances obtained by a previous research work conducted by Moretto et al. (2019), while OLR
 128 changed along the experimentation.

129 The effects of different OLR, pH and temperatures' ranges on the VFAs production were investigated
 130 along five experimental runs (Table 2). In particular, during the first run, acidogenic fermentation

131 was tested at uncontrolled pH, high OLR (22 gTS/Ld) and mesophilic temperature (37°C). Then the
 132 pH was set at 5.5 during the second run, keeping constant the OLR at 22gTS/Ld and the temperature
 133 at the mesophilic value of 37°C. These conditions did not allow for reaching steady state conditions
 134 of the system. Thus, the OLR was reduced at 11 gTS/Ld keeping constant the pH at 5.5 and the
 135 operative temperature (37°C), in the third experimental run. Then, the pH was shifted to neutral value
 136 in the fourth experimental run. Finally, temperature was increased to thermophilic temperature
 137 (T=55°C), while keeping constant the OLR and pH at 11 gTS/Ld and 7, respectively, during the last
 138 experimental run. The passage from one experimental run to the next one was conducted after the
 139 reaching of steady stable condition in term of VFAs concentration. In particular, the steady state of
 140 the reactor was considered reached when the VFAs concentration was stable at least for 3 HRTs, that
 141 means at least 18 days. With reference to the passage from run 4 to run 5, it is important to remark
 142 that the reactor's feeding was stopped for 45 days. This passage corresponded to the change from a
 143 mesophilic temperature (37°C) to a thermophilic one (55°C), which comported a deep change of the
 144 microbial communities too. In order to assure the complete acclimatation at the new operational
 145 condition to the thermophilic microorganisms without any other external interference, the feeding
 146 was interrupted for 45 days.

147 Table 2. Operational parameters during each experimental run

	T (°C)	HRT (d)	OLR (gTS/Ld)	pH
RUN 1	37 ± 1	6	22	Not controlled
RUN 2	37 ± 1	6	22	5.5
RUN 3	37 ± 1	6	11	5.5
RUN 4	37 ± 1	6	11	7
RUN 5	55 ± 1	6	11	7

148
 149 Each test was carried out in a hermetically sealed laboratory scale CSTR, with a working volume of
 150 4.2 L and a headspace of 0.8 L for a total volume of 5 L. The reactor was fed daily diluting the
 151 synthetic HFW with a proper volume of tap water (for a total amount of the feeding mixture of 700

152 mL) to reach the designed OLR, specific for each experimental run. pH was monitored twice a day and
153 adjusted manually, according to the specific condition of the experimental run, using a NaOH solution
154 (30% w/v).

155 At the same time, an equal amount of reaction medium was discharged from the reactor to keep
156 constant the reactor volume and collected for the analysis. In particular, the feeding and discharging
157 operations were performed manually through two different valves at the top and at bottoms of the
158 reactor, respectively. The reactor was continuously stirred at 16.5 rpm, and volumetric biogas
159 production was measured by a Milligascounter (RITTER, Germany). Temperature was controlled by
160 water flowing through an external jacket.

161 **2.3 Analytical methods**

162 Total Solids (TS), Total Volatile Solids, Total Kjeldahl Nitrogen (TKN), Total Phosphorous (TP),
163 Chemical Oxygen Demand (COD) and soluble COD (sCOD) were measured according to Standard
164 Methods (APHA-AWWA-WPCF, 2005). pH was determined using a portable probe (Eutech pH
165 700). N-NH₄⁺ concentrations were measured by an ion selective electrode (Orion 9512). VFAs
166 concentration was determined by ion chromatography system (Dionex ICS 1100 with AS23 column).
167 Lignocellulosic composition was determined in terms of neutral detergent fiber (NDF), acid detergent
168 fiber (ADF), acid detergent lignin (ADL), and crude fiber (CF) according to Van Soest and Wine
169 (1967). Lipids analysis and crude protein were determined by the standard procedure of the
170 Association of Official Analytical Chemists (AOAC, 1990). Sugars were determined by mass
171 spectrometry (MS), previous sample filtration at 0.20 μm. Lactic acid was measured using a
172 commercial kit (Megazyme, Bray, Ireland). Grams of VFAs and lactic acids are following reported
173 as grams of COD.

174 **2.4 Parameters for the evaluation of the process performances**

175 Two main parameters were used for the evaluation of the performances of the tests. The first is
176 represented by the specific VFAs yield, determined by the ratio between the daily VFAs production
177 rate and the grams of TS fed per day, as follows:

$$178 \text{Yield}_{\text{totVFAs}} = \frac{\text{gVFAs/d}}{\text{gTS fed/d}} \quad \text{/Equation 1/}$$

179 The second parameter is the solubilization rate of the substrates, useful to reveal a potential inhibition
180 of the process. It was calculated as the ratio between the variation of the soluble COD concentration
181 in the reaction medium (sCOD_t) and the soluble COD concentration in the feeding mixture (sCOD₀),
182 and the particulate COD (pCOD) daily fed, as follows:

$$183 \text{Solubilization} = \frac{\text{sCOD}_t - \text{sCOD}_0 \text{ (g/d)}}{\text{pCOD (g/d)}} \quad \text{/Equation 2/}$$

184

185 **2.5 Microbial community structure: high-throughput 16 s rRNA gene sequencing**

186 The analysis of the bacterial populations involved in the acidogenic mono-fermentation of HFW was
187 performed by high-throughput 16 s rRNA gene sequencing. Two samples were analyzed for each run,
188 one at the beginning of steady state and one at the end for run 1 (40.90 – 80.87 days), 3 (240.83 and
189 257.80 days), 4 (289.82 and 324.84 days), and 5 (419.08 and 440.12 days). Due to instability of the
190 system, a sample taken at the first maximum (104.13 day) and at the first minimum (131.07) of total
191 VFAs production were analysed within run 2. Genomic DNA was extracted using about 0.25 g of dry
192 weight of each sample. The extraction was performed with PowerSoil DNA Isolation kit (MoBio,
193 Italy), according to the manufacturer's instructions. DNA was eluted in 50 µL of sterile water and the
194 concentration and purity were determined by NanoDrop (2000c) spectrophotometer (Thermo
195 Scientific, USA). Aliquots were stored at –20 °C for a few days and then used for high-throughput
196 16 S rRNA gene sequencing. PCR reactions were set up in 25 µL volumes containing 12.5 ng of DNA,
197 0.5 µM primers V1-3 (27 F AGAGTTTGATCCTGGCTCAG and 534 R
198 ATTACCGCGGCTGCTGG) and 1× Phusion High-Fidelity PCR Master Mix (Thermo Fisher

199 Scientific, Waltham, MA USA). All PCR reactions were run in duplicate and pooled afterwards. The
200 amplicon libraries were purified using the Agencourt® AMPure XP bead protocol (Beckmann
201 Coulter, USA). Library concentration was measured with Qubit 3.0 Fluorometer (Thermo Fisher
202 Scientific, Waltham, MA USA). The purified sequencing libraries were mixed in equimolar
203 concentrations and diluted to a final concentration of 4 nM. The samples were paired end sequenced
204 (2x301bp) on a MiSeq (Illumina) using a MiSeq Reagent kit v3, 600 cycles (Illumina, USA)
205 following the standard guidelines for preparing and loading samples on the MiSeq. 10% Phix control
206 library was spiked in to overcome low complexity issue often observed with amplicon samples.
207 Sequences were processed using QIIME2 version 2018.2. To denoise paired-end sequences,
208 dereplicate them and filter chimeras with the “consensus” method, DADA2 software package was
209 employed (Callahan et al., 2016) to resolve amplicon sequence variants (ASVs) differing by as little
210 as one nucleotide and Silva 132 database were used to assign the taxonomy (Kehrmann et al., 2017).

211

212 **3 Results and discussion**

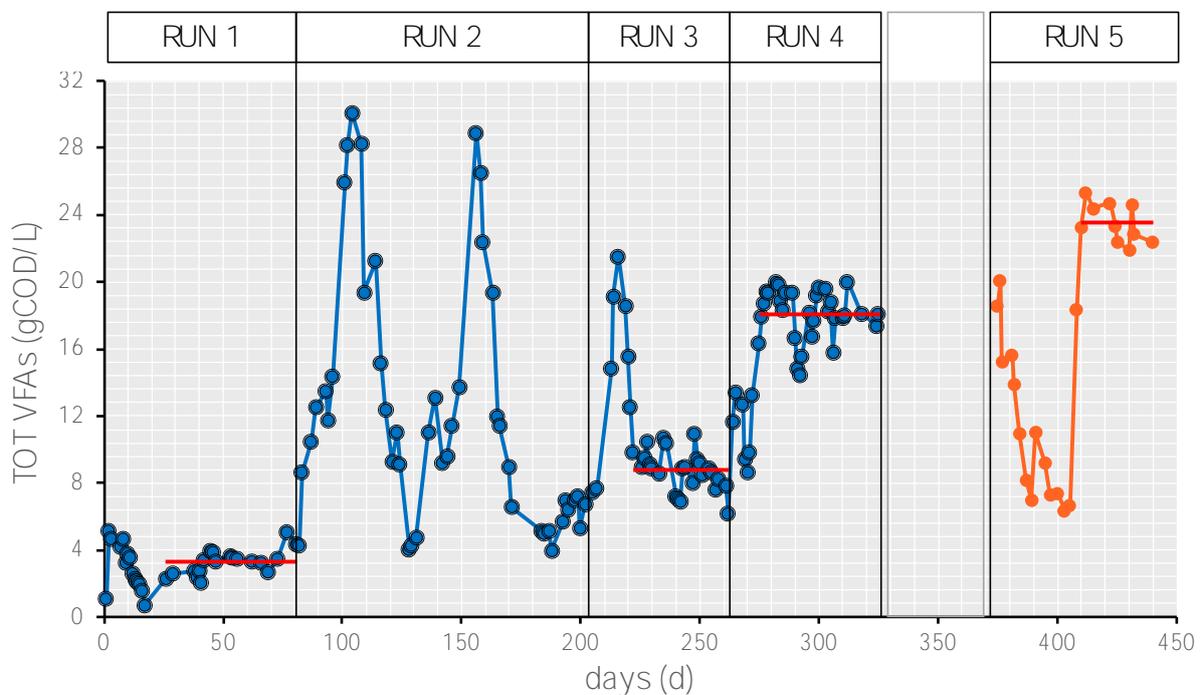
213 **3.1.1 Effects of different operational parameters on VFA production**

214 The effects of different operational conditions are clearly reported in Figure 2 and Figure 3 which
215 show the total VFAs production and the VFA profile from HFW fermentation along the different
216 experimental runs, respectively. At the beginning, high OLR (22 gTS/Ld), mesophilic condition, and
217 uncontrolled pH were tested. The high OLR immediately caused a quick accumulation of VFAs in
218 the reaction medium, with the consequent decline of pH, which was not corrected, as scheduled for
219 this experimental run (see Table 2). When the pH declined to very acidic values (3.6 ± 0.2), the
220 microbial activity was inhibited, and the system reached a low performant steady state condition of
221 VFA production rate (3.5 g/Ld). Acetic acid was almost the unique fermentation product (Figure 3)
222 accounting for 90% of total VFAs produced. The predominance of acetic acid when VFA production
223 is low, was already reported by a recent work (Luo et al. 2021). As consequence, a very low VFAs
224 yield of 0.03 gVFA/gTVS fed was achieved along this first experimental run. Low VFA production

225 yields were also reported in previous studies carried out in similar conditions (Bolzonella et al., 2005;
226 Kong et al., 2016). Although the low VFA yield, this first experimental run showed a lactic acid
227 production of about 9.40 g/Ld. As discussed, high OLR led to the acidification of the systems,
228 inhibiting the VFA production process. The acidification occurs more frequently with food waste, as
229 this substrate is characterized by a high C/N ratio and very fast hydrolysis process (Kong et al., 2016;
230 Bolzonella et al., 2018). Moreover, in line with the results obtained in this experimental run, Kong et
231 al. (2016) detected acetic acid as the main VFA in the reaction medium. At low pH ($\text{pH} < 4$), VFAs
232 are present in the medium under their undissociated form. This condition makes VFA more
233 liposoluble and increases their ability to diffuse in the reaction medium and to penetrate in the
234 bacterial cells. This causes a decrease of the intracellular pH, compromising the cellular activities of
235 the microorganisms and, consequently, the capability of producing VFAs (Palmqvist and Hahn-
236 Hagerdal, 2000). But the lactic acid producing microorganisms have a lower inhibition as effect of
237 the lower pKa of the lactic acid (3.10). It means that at the pH condition of the first experimental run
238 ($\text{pH} 3.6$), lactic acid is not still in the undissociated form and was not able to penetrate across the
239 cellular membrane of the lactic acid producing microorganisms, inhibiting their biological activities
240 (Gottardo et al., 2017). Another research work confirms that low pH has a strong inhibiting effect on
241 the growth of most microbes in acidogenic fermentation, with the only exception of some acidic
242 tolerant ones, such as the ones responsible of the lactic acid production (Luo et al., 2020). The
243 inhibition was also confirmed by the COD conversion into VFA. Even if the solubilization rate was
244 high, with observed values equivalent to 0.30 g soluble COD per gTVS fed to the system per day,
245 only 5% of the influent COD was indeed bio-converted into VFAs.

246 A first strategy to avoid the acidification of the system, keeping the feeding at high OLR, considered
247 a pH correction. High solids content, in fact, are preferable as they allow to operate with reduced
248 reactor volumes and, consequently, capital and operational costs. Thus, a continuous correction of
249 the pH at slightly acidic value (5.5) was operated, while OLR and temperature were kept constant.
250 Although the test lasted for a long time (more than 100 days), these conditions did not permit to reach

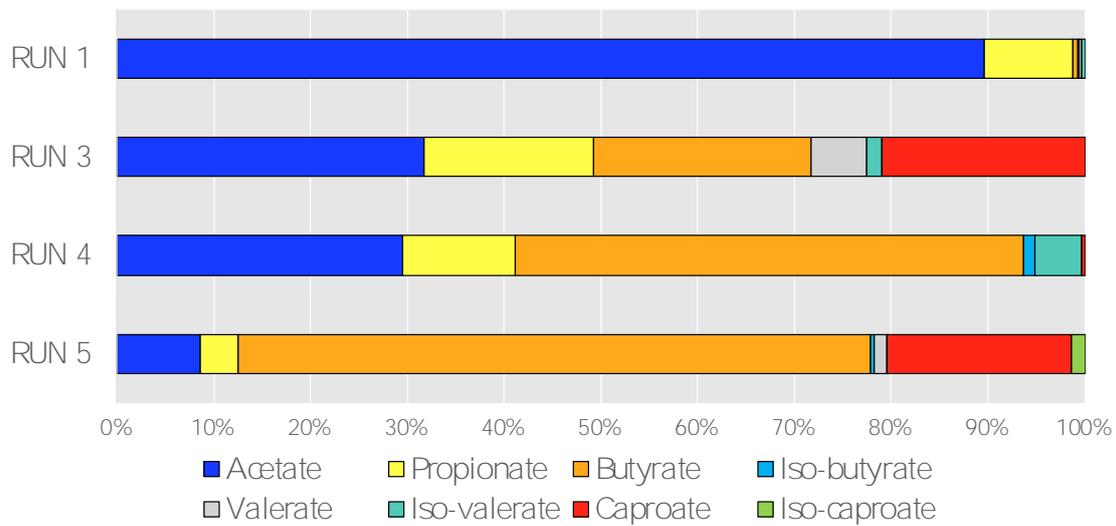
251 a stable process and VFAs production rate followed a symmetric oscillation of concentration between
 252 maximum and minimal valued of 30 g/L and 4 g/L, respectively. This phenomenon was already
 253 described by other authors working on anaerobic digestion of food wastes, and it was ascribed to a
 254 detrimental effect of substrate overloading on system stability (Nagao et al., 2012; Cheng et al., 2016).
 255 Recently, Jia et al. (2019) investigated the hydrogen production from HFW at high OLR and observed
 256 that high OLR caused a pH dropping. The authors buffered the system to repriminate the initial pH
 257 value, but these continuous pH oscillations had a negative effect on the VFA production without the
 258 reaching of a steady state condition, as in the case of this work. Probably, this situation was the
 259 combined effect of the high OLR and the pH correction at 5.5: the high OLR led to a very quick VFA
 260 accumulation causing the inhibition of the VFA producing microorganisms, as already happened in
 261 the run 1. Contrary to run 1, in this second run the pH was corrected at 5.5, allowing to
 262 microorganisms to react against the inhibition. It favoured the VFA production's recovery, until the
 263 reaching of the VFA inhibition's level again.
 264



265
 266 Figure 2. Total VFAs production trend. Blue curve is referred to mesophilic fermentation of
 267 HFW; orange curve is referred to thermophilic fermentation of HFW. Red lines show the average

268 productivity at each steady-state; grey rectangle show the period of time during which the reactor
269 was not fed to avoid washing-out phenomena.

270



271

272 Figure 3. Average distribution pattern among VFAs for steady state at each experimental run. Run 2
273 is not included, due to the instability of the system.

274

275 Considering the controversial performances during Run 2 due to a too high organic matter content in
276 the feed, the OLR was halved, from 22 to 11 gTS/Ld, in the third experimental run (Run 3). pH was
277 kept constant at 5.5. This new OLR condition, allowed the achievement of a steady state condition
278 after about 15 days of operation: a stable VFAs concentration around 8.5 g/L was observed,
279 corresponding to a yield of 0.14 gVFA/gTVS fed. VFAs showed a wider distribution, as shown in
280 Figure 3: acetic acid was still the main product, accounting for 32% of total, but also butyric, caproic,
281 propionic, and valeric acids were produced, accounting for 23%, 21%, 17%, and 6 % of total VFAs,
282 respectively. An average lactic acid concentration around 7 g/L was achieved. The positive effect of
283 lower OLR was also discussed by Jiang et al. (2013), who explained that high substrates'
284 concentration inhibited the microorganisms' metabolism. Moreover Battista et al. (2020) found that
285 high solids concentration reduced also the efficiency of the mass and heat diffusions in the CSTR,
286 causing a loss of the process performances.

287 During the fourth experimental run, the influence of neutral pH values was also investigated
288 increasing pH from 5.5 to 7. As a consequence, the VFAs concentration increased to stable values

289 around 18 g/L, after 11 days of operation, corresponding to a yield of 0.29 gVFA/gTVS fed. This
290 value is in line with the best yields reported in literature. A similar yield, 0.27 gVFA/gTS fed, was
291 achieved by Zhang et al. (2016) in tests of acidogenic fermentation of kitchen waste under similar
292 operating conditions. The improvement of acidogenic fermentation performance due to neutral pH
293 condition was due to an increase of the hydrolytic rate. The optimal pH range of the VFA producing
294 microorganisms is around the neutrality, which optimize their kinetics and their metabolisms (Detman
295 et al., 2019). Along the third and fourth experimental runs, the average sCOD concentrations were 30
296 g/L during run 3 and 50 g/L during run 4, corresponding to solubilization rates of 0.27 and 0.59 g
297 soluble COD per gram VS fed to the system, respectively. Also, the acidification efficiency increased
298 reaching 30% and 38% of sCOD under the third and fourth experimental runs, respectively. Similar
299 results were also obtained by Jiang et al. (2013) who studied the influence of different pH values on
300 VFAs production from synthetic kitchen waste. They found out that a pH value between 6 and 7
301 brought to an increase of around 20% of the hydrolysis rate, achieving a sCOD concentration of 82
302 g/L. At lower uncontrolled pH conditions, the sCOD concentration decreased to 60 g/L. The high
303 hydrolysis performances allowed for the increase of the following VFAs production which doubled
304 the concentration. Also, other fermentation products showed increased concentrations compared to
305 the run with uncontrolled pH (run 1).

306 Regarding the VFA profile, the concentrations of acetate and propionate were similar along the third
307 and fourth phases (Figure 3), while caproate and valerate disappeared during the passage from the
308 third to the fourth experimental run. On the other hand, butyrate became the main component of VFAs
309 with a concentration around 50% of the total VFAs. In addition, a strong decrease of the lactic acid
310 production was observed along the fourth run (concentrations decreased to 1.09 g/L), in line with the
311 results reported by Kim et al. (2009), Gottardo et al. (2017) and Al Dhahi et al. (2020). According to
312 the results reported in these research works, the lactic acid production has optimized for pH values in
313 the range 5.0-5.5. In particular, Al Dhahi et al. (2020), who optimized lactic acid production from

314 cofermentation of food wastes and municipal sludge, achieved the maximum lactic acid production
315 after 48 h at pH 5.5, while it decreased at higher pH ranges (6.0–7.0).

316 In the last experimental run, the working temperature was increased to 55°C, while the other
317 parameters remained constant (Table 2). Passing from mesophilic to thermophilic conditions
318 determined the concentration decline of the mesophilic microorganisms, and a consequent growing
319 of the thermophilic microbial communities. To avoid any disturbance to microorganisms during this
320 transient condition, the reactor feeding was stopped for 45 days. A stable total VFAs concentration
321 around 23.5 g/L was achieved after 30 days of operation. Butyric acid was the main product,
322 accounting for around 65% of total VFAs, while caproic, acetic, and propionic acids constituted only
323 the 19%, 9%, and 4% of total VFAs, respectively. Furthermore, a yield of 0.38 gVFA/gTVS fed was
324 achieved. The good performances of the system were probably due to the acidogenic fermentation,
325 rather than a higher hydrolytic activity, since a solubilization rate of 0.63 g sCOD per g VS was
326 achieved during run 5, comparable to the one observed during run 4 (0.59 g sCOD per g VS). A
327 higher conversion of soluble COD into VFAs was also observed: 45% versus 38% observed during
328 run 4. The importance of the temperature choice in the organic matter hydrolysis and, consequently,
329 on its conversion into VFA has been recently confirmed by Soomro et al. (2020), who remarked the
330 positive effect of higher temperatures on the performances of acidogenic fermentation of Organic
331 Fraction of Municipal Solid Wastes (OFMSW). Finally, an increase of lactic acid concentration was
332 observed compared to the previous run: lactic acid passed from 1.09 g/L to 8.37 g/L, along the fourth
333 and fifth experimental run, respectively.

334 As discussed above, the variations of the operational conditions were conducted along the same
335 experimental campaign to observe the system's kinetics in the reaching of a new steady-state
336 conditions. Except for run 2, when a new steady state was not reached, it was demonstrated that the
337 OLR and pH changes (runs 3 and 4) were followed by a lag phase period in the range of 15-25 days.
338 In a full-scale prospective, this time is economically more advantageous than the discharge of the
339 reactor and the starting of a new process at different operational conditions. The lag phase was longer

340 (around 40 days, without considering the period when reactor was not fed) after the change of the
341 temperature (from mesophilic to thermophilic conditions). With the temperature shift, as better
342 explained below, was highlighted a switch in the microbial population inhabiting the reactor. This
343 rearrangement was probably the cause of a longer lag phase previously described.

344 Regarding the biogas production, it increased in correspondence with the run characterized by better
345 VFA yields, reaching an average value of about 1.19 L/Ld along run 5. Hydrogen was the main
346 component, accounting for the 55% v/v of the total produced biogas. Methane was absent in the first
347 four experimental runs, appearing just in the last one, even if its concentration was lower than 1%
348 v/v. It is a further demonstration the reactor was set at optimal operational conditions for VFA
349 production, without the risk to shift towards methanogenic phase.

350 **3.2 Microbial community structure and dynamics**

351 The main bacterial genera found over the operation are reported in Figure 4. As expected, the lowest
352 microbial diversity was observed in Run 1 where selective operating conditions were adopted (high
353 OLR and not controlled pH). Indeed, despite the complex composition of the food waste, only
354 lactobacilli populations were enriched during run 1 (up to 99% of total reads were affiliated to
355 *Lactobacillus* genus). Remarkably, *Lactobacillum gasseri* (ASV1 in Table 1 Supplementary
356 information), an obligate homofermentative bacterium typically retrieved in the gastrointestinal tracts
357 of humans and animals, was found highly abundant in both samples taken during stable VFAs
358 production in Run 1 (82.1% and 42.2% in 1A and 1B samples respectively). This finding is in line
359 with the average yield of 0.07g of lactic acid per gram of TS fed obtained during run 1, twice the
360 value of VFAs concentration. The higher lactic acid yield may be explained by the known acid
361 tolerance of lactobacilli that are also specialized lactic acid producers (Corcoran et al., 2005).
362 Nonetheless, a certain production of VFAs was ensured by heterofermentative *Lactobacillus* spp. in
363 the reactor such as *L. pontis*, *L. vaginalis*, *L. buchneri*, and *L. farraginis* (Filya et al., 2003; Endo and

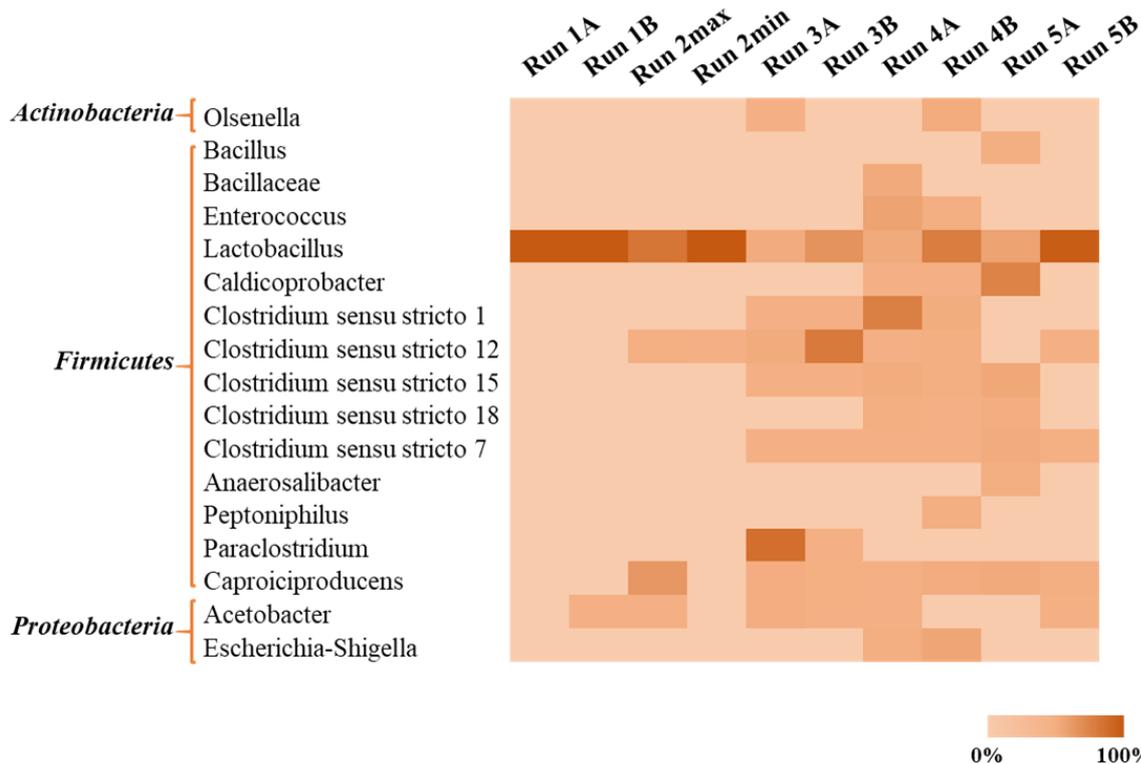
364 Okada, 2007; Ibrahim, 2016) that represented 5.1% and 44% of total reads in sample 1A and 1B,
365 respectively.

366 The analysis of the sample “2 max”, taken at the first maximum of total VFA production (day 104)
367 and “sample 2 min”, taken at the first minimum of total VFA production (day 131) showed the
368 dominance of *Lactobacillus* species. In the first case, most of the sequences were attributed to the
369 genera *Lactobacillus* (68.7%) and *Caproiciproducens* (29.5%) whereas sample “2 min” was almost
370 entirely composed by *Lactobacillus* species. Sample 2 max was mainly composed by the
371 heterofermentative *L. rhamnosus* with the dominance of *L. rhamnosus* ASV4 (32.5% of total reads)
372 whereas *Lactobacillus* ASV3 was highly abundant (84.1%) in sample 2min (Table S1 supplementary
373 information).

374 *Caproiciproducens* ASV2 was found only in sample “2 max” (Table S1 supplementary information).

375 Members of the genus *Caproiciproducens* are strictly anaerobic bacteria able to produce caproic acid
376 along with acetic and butyric acids (Kim et al., 2015; Bengelsdorf et al., 2019). Some
377 *Caproiciproducens* species are capable of making a lactate-based chain elongation. The pyruvate
378 produced during lactate oxidation is further oxidized to acetyl-CoA and CO₂ with electrons released
379 in form of reduced ferredoxin. The derived acetyl-CoA is used for butyrate and caproate formation
380 through a reverse β -oxidation pathway (Spirito et al., 2014).

381



382

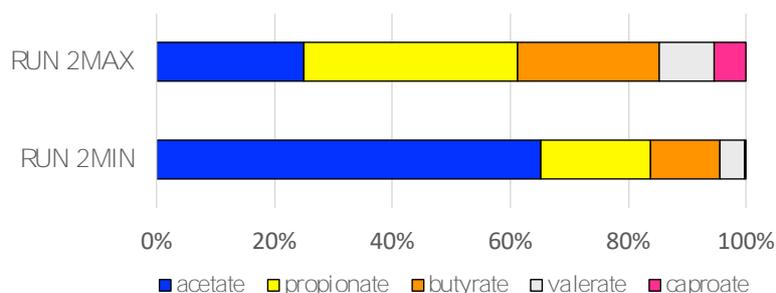
383 Figure 4. Frequency heat-map of microbial communities at taxonomical genera levels (only taxonomy group
 384 $\geq 2\%$ of abundance in at least one sample are shown). The colour intensity in each cell shows the relative
 385 abundance.

386

387 This pathway allows the conversion of the available acetate (C2) to butyrate (C4) by the condensation
 388 of two acetyl-CoA, and, if other molecules of acetyl-CoA are available, further elongation to caproic
 389 acid (C6) (Kucek et al., 2016). Figure 5 shows the punctual distribution among VFAs produced at
 390 time of sampling during run 2. As it can be noticed, when *Caproiciproducens* ASV2 was present in
 391 the reactor along with *Lactobacillus rhamnosus* ASV4 during the first maximum of total VFA
 392 production, a greater amount of butyric and caproic acids and a lower amount of acetic acid were
 393 produced, probably as consequence of a chain elongation process mediated by *Caproiciproducens*,
 394 using lactic acid produced by lactobacilli as electrons donor (de Leeuw et al., 2019).

395 In line with this finding, at the first minimum production of total VFA, apart from the extremely
 396 lower VFAs production rate, acetic acid was the main fermentation product mirrored by the
 397 establishment of lactic acid bacteria and the decline of *Caproiciproducens*. The correlation among
 398 lactobacilli and *Caproiciproducens* spp. was observed by Contreras-Dávila et al. (2020) during an

399 experiment on caproic acid production by reverse β -oxidation pathway using food waste as feedstock.
 400 They observed that the process took place by two separate phases: an acidification phase towards
 401 lactate as main product, followed by a chain elongation phase. The oscillation between these two
 402 phases required an increase of OLR, as in the case of the second run.
 403



404
 405 Figure 5. Pattern of distribution at time of sampling during the maximum and minimum of total VFAs
 406 production
 407
 408

409 During the operation at lower OLR (Run 3, 4 and 5), the production of more heterogeneous VFAs
 410 mixtures is mirrored by the higher microbial diversity associated to known fermentative bacterial
 411 genera (Figure. 4). The analysis of Run 3 revealed the establishment of a variety of putative
 412 fermenters mostly belonging to *Paraclostridium*, *Clostridium sensu stricto* 12, *Lactobacillus* and
 413 *Caproiciproducens*. *Paraclostridium* ASV5 (38% in sample 3A), *Clostridium sensu stricto* 12 ASV6
 414 (55.5% in sample 3B) and *Lactobacillus* ASV3 (Table S1) represented the main taxa even though
 415 marked fluctuations of their relative abundance were observed over the operation. Nearly all members
 416 of the taxon *Clostridium sensu stricto* form butyrate as main fermentation product, but also other
 417 compounds, including various organic acids and alcohols, are produced. Members of *Clostridium*
 418 *sensu stricto* have a wide physiological versatility and this may explain the greater distribution among
 419 VFAs achieved during this experimental run. Consistently with the occurrence of lactobacilli, the
 420 lactic acid production rate reached a stable value of 7.43 g/L.

421 In Run 4, the increase of pH toward neutrality led to an initial enrichment of bacteria belonging to
 422 the class *Clostridia*. Consistent with the VFA distribution pattern observed in this run (Figure 3),

423 *Clostridium butyricum* ASV7, known butyric acid producing microorganism (Wiegel, 2015),
424 dominated in sample 5A (46.8%; Table S1) together with *Lactobacillus* ASV3 found at lower
425 abundance (6.4%). At the time of sampling, the lactic acid production rate was about 1.79 g/Ld, one
426 of the lowest measured during the whole experimental campaign. This low production can be partially
427 explained either by the relative low abundance of *Lactobacillus* spp., and, likely by a metabolic cross-
428 feeding with lactate-utilizing, butyrate-producing bacteria. It is known that some bacteria, such as *C.*
429 *butyricum*, are able to convert lactate and acetate to butyrate (Detman et al., 2019).

430 *Lactobacillus* ASV3 and ASV8 were the main taxa retrieved in sample 4B (31.7% and 16.5%
431 respectively). This result can explain the heterogeneous distribution of fermentation products
432 obtained within this run, which was however characterized by butyric acid as main fermentation
433 product. Other conversion pathways of lactic acid to other products, such as caproic acid, most likely
434 took place in the system, as revealed by the occurrence of *Caproiciproduces* spp., and this may
435 explain the observed low lactate production rate.

436 Members of *Caldicoprobacter* genus represented one of the main microbial components under
437 thermophilic operating conditions (52.4% in sample 5A). The peculiarity of species belonging to this
438 genus is the capability of growing on xylan, and degrading hemicellulose (Yokoyama et al., 2020).

439 *Lactobacillus* species were also enriched during operation (up to 95.8%) in line the observed increase
440 of lactic acid concentration compared to previous mesophilic run.

441 **4. Conclusions**

442 The optimization of the VFA production from HFW in continuous mode was investigated in the
443 present research work. It was demonstrated that different pH, temperature and OLR values influenced
444 the VFA productivity and VFA profile. The conditions which allowed the best VFA yield were: OLR
445 around 11 gVS/Ld, pH above 7 and thermophilic temperature. Under these conditions, the VFA yield
446 was of 0.38 gVFA/gTVS, corresponding to a solubilization rate of 0.63 sCOD/gTVS. The VFA
447 productivity of almost 0.40 gVFA/gTS is one of the highest present in the scientific literature. A very

448 recent work (Jon Jones et al., 2021) achieved a similar result of 0.35gVFA/gVS, but after a
449 combination of filtration and electro dialysis on food wastes.

450 In view of a practical application of the process, it is fundamental to remark that, after the optimization
451 of the VFA production, the next challenge is their recovery and purification, that means the finding
452 of economically and efficient techniques for their separation from the fermentation broth and then,
453 for their concentration till 70-100 g/L.

454 Moreover, other interesting observations for the economy of the process can be done from the
455 discussion of the lag phase, which is around of 15-25 days when OLR and pH were gradually
456 modified, while the passage from mesophilic to thermophilic temperature required longer time. These
457 considerations must be considered in an eventual scale-up of the process.

458 It was also demonstrated that different operational parameters influence the VFA profile and the
459 microbial community structure. Regarding the VFA composition, acetic acid was the main product
460 when the process was not optimized and the VFA productivity was low. Instead, under the optimal
461 conditions, butyric acid was the predominant VFA (more than the 60% of the total acids) followed
462 by a significant amount (almost the 20%) by caproic acid. It demonstrated that the system can be
463 adapted if the production of specific VFA is desired through the variation of the operational
464 conditions.

465 As for microbiological aspects, the runs with the best performances during HFW mesophilic
466 treatment showed the enrichment of bacteria belonging to the class *Clostridia*, while *Lactobacillus*
467 species dominated under acidic conditions or when the process was performed at high temperature.

468

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