

1 **CAN OCCURRENCE OF PESTICIDE METABOLITES DETECTED IN CROPS PROVIDE THE**  
2 **EVIDENCE ON ILLEGAL PRACTICES IN ORGANIC FARMING?**

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11 **Abstract**

12 Modern pesticides rapidly degrade after their application due to both physicochemical  
13 factors and through biotransformation. Consequently, pesticide residues in samples might  
14 be either undetectable or detected at low concentrations ( $\leq 10 \mu\text{g}/\text{kg}$ ). Under such  
15 conditions, a monitoring of pesticide metabolites in samples might be a conceivable solution  
16 enabling the documentation of earlier pesticide use. Analysis of metabolites might pose  
17 analytical challenges, because pesticide degradation leads to the production of a number of  
18 metabolites, differing somewhat in their structure and polarity. This study was focused on  
19 the determination of pesticide residues and their metabolites in samples of grapevine and  
20 wine, using ultra-high-performance liquid chromatography coupled with high-resolution  
21 mass spectrometry, with the objective of supporting the possibility of the verification of the  
22 method of farming. It documents the identification of pesticide metabolites commonly used  
23 in conventional farming and provides a characterization of pesticide degradation during  
24 grapevine growth, maturation and during the wine-making process.

25 **Keywords**

26 pesticide metabolites, pesticide residues, grapevine, organic farming, UHPLC-HRMS(/MS)

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28           **1. INTRODUCTION**

29   The market of organic products is continuously growing in all EU member states, including  
30   the Czech Republic. Besides the concern about the impact of conventional farming practices  
31   on the environment, consumers' purchase of organic foods is motivated by various beliefs  
32   including that they are safer, healthier and taste better. Although it is rather difficult to  
33   provide clear evidence of the last two features, for some consumers the major expectation is  
34   the absence of synthetic pesticides, which are perceived as a serious health risk. It is worth  
35   noting that the incidence of pesticide residues in organic food is generally low. For instance,  
36   the recent European Food Safety Authority (EFSA) report shows that 83.1 % of organic food  
37   produced in 2016 in EU countries was free of quantifiable residues, while it was only 53.3 %  
38   in the case of conventional foods.<sup>1</sup> Only a few substances (listed in Annex II of the  
39   Commission Regulation (EC) No 889/2008)<sup>2</sup> are permitted, but the use of synthetic  
40   pesticides is one of the restrictions applied in organic farming (Council Regulation (EC) No  
41   834/2007).<sup>3</sup> Nevertheless, in some cases, contamination can be detected by current highly  
42   sensitive methods. In some cases, when applied in a conventional field, pesticides can drift  
43   through the air to a neighboring organic farm, leaving traces on the food crops. Similarly,  
44   some contamination may occur during transportation or storage of organic foods, if they are  
45   not carefully separated from conventional food items. The common practice of food  
46   inspection authorities or certification bodies is to tolerate residues at or below the level  
47   0.01 mg/kg. However, when a rapidly degrading pesticide is employed in the pre-harvest  
48   period, no or very low residues can be detected at harvest time. Therefore, the  
49   unambiguous identification of such an illegal practice is fairly complicated. Laboratory  
50   analysis aimed at the control of compliance with legislation typically targets only compounds  
51   that are included in the definitions of maximum residue level (MRL)<sup>4</sup>. These mostly involve

52 merely the parent pesticides and, only in limited cases, certain toxicologically relevant  
53 metabolites. However, dissipation of modern pesticides after application leaves a number of  
54 various (nontoxic) metabolites in treated plants. These metabolites can be considered, in  
55 some respect, as 'markers' of earlier unauthorized pesticide usage. Therefore,  
56 documentation of their presence at higher concentrations in a product labeled as 'organic'  
57 may support identification of fraud even in the case when a parent pesticide is not  
58 detectable. (Bio)transformation of pesticides in plants has been a subject of many studies  
59 and has to be described in documentation submitted on the occasion of pesticide  
60 preparation registration.<sup>5</sup> In general terms, in Phase I, oxidation and hydrolysis and  
61 sometimes also reduction of the parent molecules take place. In Phase II, these products  
62 undergo conjugation with polar molecules such as sugars, amino acids or glutathione. In  
63 some cases, such conjugates occur directly with parent compounds. In Phase III, conversion  
64 of Phase II metabolites into (nontoxic) 2<sup>o</sup> conjugates takes place; metabolites move to the  
65 vacuole for storage or are incorporated into the cell wall.<sup>6</sup>

66 Analysis of pesticide metabolites is fairly challenging for several reasons: (i) their  
67 concentration might be very low, thus difficult to detect; as several metabolites originate  
68 from a parent pesticide; (ii) the dynamics of the origin of individual metabolites in specific  
69 crops are unknown, thus multiple metabolites should be always targeted; (iii) most of these  
70 metabolites are somewhat more polar compared to the parent compound, thus cannot be  
71 directly incorporated into common multiresidue methods, so new analytical procedures  
72 have to be developed and implemented; (iv) analytical standards are commercially available  
73 for only a few metabolites, thus accurate quantification of most of them is practically  
74 impossible.

75 As mentioned above, the popularity of organic products is rapidly growing. This also applies  
76 in the case of 'organic wines' offered on the Czech market. Like other organic farm products,  
77 these wines are also regularly controlled for quality and authenticity. While the procedures  
78 for the identification of a geographic origin, based on a combined analysis of elemental and  
79 isotopic composition, are available in control laboratories, the methodology is still missing  
80 for the reliable authentication of farming practice. To investigate the possibility of using  
81 pesticide metabolites as markers of unauthorized pesticide usage, we performed a pilot  
82 study within which a number of samples was collected from an experimental vineyard where  
83 grapevine plants were treated with common fungicides according to an agreed treatment  
84 plan. Using ultra-high-performance liquid chromatography coupled with high-resolution  
85 mass spectrometry (UHPLC-HRMS(/MS)), the dynamics of parent compounds' dissipation  
86 and the origin of the relevant metabolites were monitored. The transfer of these compounds  
87 into wine was also studied. Several studies realized earlier by other authors documented the  
88 fate of pesticides during wine making process, the impact of various procedures such as  
89 filtration, clarification and maceration was described in detail. [7-8](#) Optimization of the  
90 analytical strategy for the extraction of metabolites formed an integral part of this study.

## 91 **2. MATERIALS AND METHODS**

### 92 **2.1. Materials**

93 Certified standards of pesticides (dimethomorph, fenhexamid, iprovalicarb, metrafenone,  
94 pyraclostrobin, quinoxyfen, spiroxamine, tebuconazole and triadimenol) were purchased  
95 from Dr. Ehrenstorfer GmbH (Augsburg, Germany), Honeywell Fluka<sup>TM</sup> or Honeywell Riedel-  
96 de Haen<sup>TM</sup> (both Seelze, Germany). The purity of standards was in the range of 98-99.9 %.  
97 The internal standards for triphenyl phosphate (TPP) and nicarbazin were obtained from  
98 Sigma-Aldrich (St. Louis, Missouri, USA). Individual pesticides' stock solutions were prepared

99 in pure methanol, acetonitrile or acetone containing 1 % formic acid (v/v), depending on the  
100 solubility of the specific pesticide. A composite stock standard in acetonitrile was prepared  
101 at 50 000 ng·mL<sup>-1</sup> from stock solutions and was stored at -18 °C. The working standard  
102 mixtures (20-2000 ng·mL<sup>-1</sup>) used for matrix-matched calibration were prepared from stock  
103 solution by further dilution with acetonitrile.

104 HPLC-grade acetonitrile, LC-MS-grade formic acid, ammonium formate and ammonium  
105 acetate were obtained from Sigma-Aldrich. Methanol was obtained from Merck (Darmstadt,  
106 Germany). Acetone and sodium chloride were obtained from Penta (Chrudim, Czech  
107 Republic). Anhydrous magnesium sulfate was obtained from Honeywell Fluka™. Deionised  
108 water (18 MΩ) was produced, using a Millipore Milli-Q system (Bedford, USA).

## 109 **2.2. Samples**

110 All samples tested in this study were collected from the vineyard located at the Oblekovice  
111 experimental station in the Czech Republic (South Moravian Region). The Ryzlink rynsky  
112 (Riesling) grapevine cultivar, treated with various fungicides was used for the experiments.

113 The schedule of treatment and a list of applied Plant Protection Products (PPP) and other  
114 preparations, together with the sampling intervals of tested materials, are documented in

115 **Table 1**. Treatments were performed at the doses recommended by the manufacturers.

116 Vine leaves were sampled from May to October (20-week period), wine grapes were  
117 sampled during the ripening period from September to October (5 weeks). Both types of  
118 samples were collected every 6-8 days (see **Table 1**).

119 Grapes were harvested after the pre-harvest interval (given on a label of PPP) of the  
120 particular pesticide had elapsed. Thereafter, the grapes were processed into a white wine.

121 The wine-making process involved the following steps: (i) crushing and pressing of grapes  
122 into a juice; (ii) clarification of juice (separation of lees); (iii) fermentation of must;

123 (iv) racking; (v) fining of young wine; (vi) filtration and (vii) bottling of wine. During the wine  
124 production, sampling was done after crushing and pressing of the grapes (juice), after  
125 racking (clarified juice), during the alcoholic fermentation (musts), after racking (young wine  
126 and lees), and after bottling (white wine). During the alcoholic fermentation, samples of  
127 musts were collected at 6- to 14-day intervals. The amount of an individual sample was 100-  
128 200 g and 200-250 mL respectively. All samples were stored at -18 °C in polyethylene bags or  
129 in plastic bottles. The list of samples is shown in **Table 1**.

### 130 **2.3. Sample preparation**

131 Prior to the analysis, solid samples (vine leaves and wine grapes) were homogenized, using a  
132 laboratory blender. Liquid samples (musts and wines) were mixed thoroughly. Parent  
133 pesticides were determined, using an ISO 17025 accredited method routinely used in our  
134 laboratory. A new extraction/detection method had to be implemented for the analysis of  
135 metabolites.

#### 136 **2.3.1. Extraction of parent pesticide residues**

137 The extraction procedure was based on the QuEChERS method. 2.5 g of homogenized vine  
138 leaves were weighed into a 50-mL centrifugation tube, followed by the addition of 10 mL of  
139 water containing 1 % (v/v) of formic acid. The matrix was allowed to soak for 20 min. In the  
140 case of wine grapes/wine, 10 g of previously homogenized samples was weighed into a 50-  
141 mL plastic centrifuge tube, without water addition. Subsequently, 10 mL of acetonitrile was  
142 added and the tube was vigorously shaken for 2 min. In the next step, 1 g of NaCl and 4 g of  
143 MgSO<sub>4</sub> were added and the shaking process was repeated for 1 min. Then 100 µL of mixture  
144 of TPP and nicarbazin (5 µg·mL<sup>-1</sup>) as an internal standard was added and the tubes were  
145 centrifuged for 5 min at 11 200 rcf. An aliquot of the supernatant was transferred into a vial.

146 In the case of wine samples, the volume of the extract (top organic layer) was affected by  
147 the ethanol naturally present in wines (11-15 vol.%). To compensate for this effect, an  
148 addition of internal standards was used.

### 149 **2.3.2. Extraction of pesticide metabolites**

150 Four alternative sample preparation procedures were tested for the extraction of pesticide  
151 metabolites from the experimental samples.

#### 152 *Procedure A*

153 The same extraction procedure based on the modified QuEChERS method described above  
154 for the extraction of parent pesticides (see Section **2.3.1**) was also tested for the extraction  
155 of pesticide metabolites.

#### 156 *Procedure B*

157 A sample of homogenized vine leaves (2.5 g) or wine grapes/wine (10 g) was weighed into  
158 a 50-mL centrifugation tube, followed by the addition of 10 mL of a mixture of  
159 methanol:water (80:20, v/v) containing 1 % (v/v) of formic acid. The tubes were closed and  
160 shaken vigorously by a mechanical shaker for 20 min. The tubes were centrifuged for 5 min  
161 at 11 200 rcf. An aliquot of the supernatant was filtered through a 0.22 µm PTFE filter and  
162 transferred into a vial.

#### 163 *Procedure C*

164 A sample of homogenized vine leaves (2.5 g) or wine grapes/wine (10 g) was weighed into  
165 a 50-mL centrifugation tube, followed by the addition of 10 mL of mixture of  
166 methanol:water (50:50, v/v) containing 1 % (v/v) of formic acid. The tubes were closed and  
167 shaken vigorously by a mechanical shaker for 20 min. The tubes were centrifuged for 5 min  
168 at 11 200 rcf. An aliquot of the supernatant was filtered through a 0.22 µm PTFE filter and  
169 transferred into a vial.



170 *Procedure D*

171 10 g of wine grapes were weighed into a 50-mL centrifugation tube. Thereafter, the tubes  
172 were centrifuged for 5 min at 11 200 rcf. An aliquot of the filtered juice was transferred into  
173 a vial.

#### 174 **2.4. Identification and quantification of pesticide residues**

175 Identification of pesticide residues in the samples was based on a comparison of retention  
176 time, accurate mass ( $m/z$ ) of the (de)protonated molecule, isotopic pattern matching and  
177 accurate mass of MS/MS fragments, to those obtained for pesticide reference standards.  
178 The acceptable mass error of potential elemental composition for (de)protonated molecule  
179 was  $\pm 5$  ppm. The identification criteria were in accordance with the requirements in the  
180 European Commission's guideline SANTE/11813/2017.<sup>9</sup>

181 Quantification was performed by using a calibration curve based on matrix-matching  
182 calibration standards. To obtain matrix-matched standards corresponding to concentration  
183 level 1, 2, 5, 10, 20, 50, and 100 ng·mL<sup>-1</sup>, 50  $\mu$ L of a specific working standard mixture and  
184 50  $\mu$ L of internal standard (1  $\mu$ g·mL<sup>-1</sup>) were added to 900  $\mu$ L of the blank extract (blank  
185 extract diluted with acetonitrile in ratios of 1:9 and 1:99)

#### 186 **2.5. LC-MS parameters**

187 The LC–HRMS(/MS) analyses of fungicide residues and their metabolites were performed  
188 using an Agilent Infinity 1290 LC system (Agilent Technologies, USA), equipped with an  
189 Acquity UPLC HSS T3 analytical column (100 mm  $\times$  2.1 mm, 1.8  $\mu$ m particle size, Waters,  
190 USA). Mass spectrometry detection was performed using Quadrupole-Time of Flight  
191 spectrometry (Agilent Ion-Mobility Q-TOF 6560, USA) in positive and negative ESI modes.

192 The column temperature was maintained at 40 °C. The injected sample volume was 4  $\mu$ L.

193 The mobile phases were different for analyses in electrospray positive (ESI+) and negative

194 (ESI-) ionization modes. For compounds detected in the ESI+, mobile phases were (A) water  
195 with 5 mM ammonium formate and 0.1 % (v/v) formic acid and (B) methanol, respectively.  
196 For compounds detected in the ESI- , mobile phases were (A) water with 5 mM ammonium  
197 acetate and (B) pure methanol. The gradient was the same in both polarities: the starting  
198 mobile phase composition was 5 % of the organic phase (B) with flow 0.2 mL·min<sup>-1</sup> and  
199 linearly changed to 99 % (B) with flow 0.3 mL·min<sup>-1</sup> in 10 min. This mobile phase composition  
200 was held for 2 min simultaneously with the flow rate being changed from 0.3 to  
201 0.4 mL·min<sup>-1</sup>. The column was reconditioned for 2 min in the starting composition of 5 % (B)  
202 (flow rate 0.4 mL·min<sup>-1</sup>). The autosampler temperature was maintained at 5 °C.  
203 The MS source conditions were as follows: capillary voltage (V<sub>Cap</sub>) was 4 kV (-4kV in ESI-);  
204 nozzle voltage was 1 kV; gas temperature and sheath gas temperature were 210 °C and  
205 380 °C respectively; drying gas flow and sheath gas flow were 10 L·min<sup>-1</sup> and 12 L·min<sup>-1</sup>  
206 respectively, and nebulizer pressure was 342.6 kPa (35 psig) in both acquisition modes.  
207 Collision induced dissociation was performed, using nitrogen and the collision energy was  
208 fixed (20 V). An Agilent MassHunter Workstation Software (version B.07.00; Agilent  
209 Technologies, USA) was used for data acquisition and data analysis.

## 210 **2.6. Strategy for detection and identification of pesticide metabolites**

211 Detection and identification of pesticide metabolites in sample extracts (see Section **2.3.1**)  
212 were based on calculated accurate mass ( $m/z$ ), isotopic pattern matching and accurate mass  
213 of MS/MS fragments. The acceptable mass error of potential elemental composition for  
214 (de)protonated molecule was  $\pm 5$  ppm.

215 At first, HRMS mass spectra in a full-scan technique (without fragmentation – MS<sup>1</sup>) were  
216 acquired across the entire chromatographic run, using a mass range of  $m/z$  100-1100. The  
217 obtained data were searched against the database of elemental composition of metabolites

218 (csv format of file), created manually based on a survey in the available literature on  
219 pesticide metabolism in plants (see **Table 2**), and consideration of common metabolic  
220 reactions (*e.g.* oxidation, dealkylation).

221 In the next step, the identity confirmation of metabolites detected in MS<sup>1</sup> was based on data  
222 acquired in the MS/MS run. 3 categories of fragments were searched:

- 223 (i) diagnostic ions, known for some groups of fungicides [10](#);
- 224 (ii) common fragments detected in the MS/MS spectrum of the parent pesticide  
225 as well as its metabolite;
- 226 (iii) fragments characterizing a part of molecule with metabolic modification, not  
227 detected in the MS/MS spectrum of the parent pesticide.

228 For identification of the conjugates of the parent pesticide and/or its metabolite, a search  
229 for neutral losses (*e.g.* hexoses) in fragmentation mass spectra was performed.

## 230 **2.7. Method validation**

231 Performance characteristics (recovery, repeatability, within-laboratory reproducibility and  
232 limit of quantification) were determined for pesticide residues (parent compounds) in vine  
233 leaves, grapes and wine. Validation studies were performed on spiked blank samples. Two  
234 spiking levels (0.002 mg·kg<sup>-1</sup> and 0.02 mg·kg<sup>-1</sup> in grapes and wine or 0.008 mg·kg<sup>-1</sup> and  
235 0.08 mg·kg<sup>-1</sup> in vine leaves) were used and analyzed in 6 replicates. Within-laboratory  
236 reproducibility (RSD<sub>R</sub>) was determined from on-going QC-data in routine analyses. Samples  
237 were extracted by the extraction procedure described in Section **2.3.1**.

238 As standards of pesticide metabolites were not available, recovery experiments could not be  
239 performed. The precision (repeatability) of the method was determined by an analysis of  
240 samples containing incurred pesticide metabolites, extracted by *Procedure A* (see Section  
241 **2.3.2**) in 6 replicates.

### 242 3. RESULTS AND DISCUSSION

243 Like other crops, grapevine can also be invaded by pests. In specific cases, fungal infections  
244 are common. In conventional vineyards, synthetic fungicides are commonly applied for  
245 prevention and treatment throughout the growing season, as well as after the harvest.  
246 However, these compounds are not permitted in organic grape production. The only  
247 chemicals conceivable for this purpose are *e.g.* sulfur, lime sulfur or some copper  
248 compounds.<sup>2</sup> The analysis of residues of synthetic pesticides might seem to be an efficient  
249 tool to control compliance with the restrictions in organic farming. Nevertheless, as  
250 mentioned in the Introduction, the absence of detectable pesticide residues, or the presence  
251 of their negligible traces, do not necessarily document that unauthorized use of banned  
252 plant protection products has occurred. To introduce some more reliable solution, we  
253 decided to investigate the possibility of screening the pesticide metabolites that are not  
254 involved in routine MRLs control (as they are not of toxicological concern), as treatment  
255 'markers'. The implementation of the relevant screening method was the key task in the first  
256 phase of our research. In order to understand the dynamics of parent pesticides' dissipation  
257 and to have the relevant experimental matrices with incurred residues available, a number  
258 of fungicidal treatments were performed by our partners from the Central Institute for  
259 Supervising and Testing in Agriculture in their experimental vineyards. The outcome of our  
260 pilot study is described in the paragraphs below.

#### 261 3.1. Development of method for pesticide metabolites

262 With the exception of pesticide metabolites that are involved in residue definition<sup>11</sup>, in  
263 scientific literature, information on the occurrence of these compounds originated through  
264 plant metabolism and a description of their analysis is very limited. Since nontoxic pesticide  
265 metabolites are practically unavailable as analytical standards, the only applicable screening

266 method for them is high-resolution mass spectrometry, which allows for the detection of  
267 their (de)protonated molecules, supposing that the presumable elemental formula of the  
268 relevant compound is known. On this account, as described in the Section **2.6** and  
269 summarized in **Table 2**, based on the available documentation on existing pesticide  
270 metabolites, we established a database into which the exact masses of metabolites'  
271 molecular ions calculated were inserted (see **Table 2**). To implement the method enabling  
272 screening of these analytes, we had to perform simultaneous testing of an extraction  
273 procedure and their HRMS detection in the respective plant extracts prepared from treated  
274 grape wine samples.

### 275 **3.1.1. Optimization of sample preparation procedure**

276 As pesticide metabolites are generally more polar than parent compounds (the increase of  
277 polarity depends on the type of metabolic transformation), we were not sure whether a  
278 common QuEChERS extraction method would provide the highest (apparent) recovery.

279 In total, four different procedures for the extraction of pesticide metabolites from pooled  
280 wine grapes with incurred residues were tested (the grapes used for this purpose were not  
281 treated by metrafenone). In addition to QuEChERS, also two methanol–water mixtures  
282 (80:20 and 50:50, v/v) were used as more polar extraction solvents. The last sample  
283 processing method, aimed at the illustration of metabolite transfer into juice, was realized  
284 by a simple separation of solids by centrifugation of the grapes' homogenate.

285 With the exception of the QuEChERS extraction, in which phases partition results in  
286 discrimination of the most polar matrix components (they are not transferred into an  
287 acetonitrile layer), no clean-up step was employed in the above procedures, to prevent  
288 losses of potentially occurring metabolites assumed to be more polar than parent pesticides.

289 **Figure 1** shows the apparent extraction efficiencies, expressed as signal intensities as

290 measured by the UHPLC-HRMS method. The ions of targeted metabolites were extracted  
291 using their calculated masses (**Table 3**). Considering the QuEChERS extraction as a reference,  
292 we can see that this sample preparation procedure provided, on average, the highest  
293 analyte signals compared to the other procedures. It is worth noting that these results do  
294 not necessarily mean that the highest concentrations of target analytes were contained in  
295 QuEChERS extracts, since matrix effects (signal suppression) in samples prepared by other  
296 procedures might be more severe. This assumption supports observation of more intensive  
297 background signal in total ion chromatogram (TIC). It should be noted that, while the most  
298 polar components such as sugars or hydroxy-carboxylic acids, remain in the aqueous phase  
299 during the partition step in QuEChERS, these compounds elute in front part of the  
300 chromatogram, where the polar metabolites also mostly elute. Considering these facts, the  
301 QuEChERS method was employed in all follow-up experiments focused on pesticide  
302 metabolite screening.

### 303 **3.1.2. Confirmation of metabolites' identity**

304 Altogether 18 metabolites originating from 7 pesticides were tentatively identified in the  
305 experimental samples, with their quantity depending on the sampling time and processing  
306 step. The list of parent compounds and detected metabolites, with exact mass of the  
307 (de)protonated molecule and their fragments, ion type and retention time is shown in **Table**  
308 **3**. The identity of pesticide metabolites tentatively identified by screening of their calculated  
309 (de)protonated molecules (mass error tolerance  $\pm 5$  ppm) in both ionization polarity modes  
310 and assessment of the isotopic pattern match were further confirmed by a critical analysis of  
311 their fragmentation spectra. To illustrate this generic approach, an example of the  
312 identification of three fenhexamid metabolites is given in the following paragraph.

### 313 **3.1.3. Identification of metabolites of fenhexamid**

314 Based on data from field trials<sup>12</sup>, residues of fenhexamid were not extensively metabolized in  
315 grapevine. Two metabolic pathways were described: (i) hydroxylation on the cyclohexyl  
316 group of parent molecule at 2- or 4-position, followed by formation of sugar conjugates, and  
317 (ii) glycosylation on the phenolic hydroxyl group of fenhexamid. The mass spectra acquired  
318 during analysis were investigated to confirm the presence of these metabolites in the tested  
319 samples.

320 A peak with a possible elemental composition  $C_{14}H_{17}Cl_2NO_3$  ( $m/z$  318.0658,  $[M+H]^+$ ) that  
321 might correspond to hydroxylated metabolite (Fen-OH) was detected in the chromatogram  
322 at 8.56 min. Its final tentative identification was based on the presence of fragment ions  
323  $m/z$  97.1006 and  $m/z$  55.0527, which were the most abundant ions in the fragmentation  
324 mass spectrum and were also detected in the MS/MS spectrum of the parent fenhexamid.  
325 Fragment  $m/z$  113.0960 in the spectrum of Fen-OH characterized the position of the  
326 hydroxyl group cyclohexyl ring in the metabolite molecule. Finally, the fragment  
327  $m/z$  300.0553 in the MS/MS spectrum corresponded to a loss of water molecule  
328 ( $\Delta m/z$  18.0156 Da) from the metabolite Fen-OH (**Figure 2**).

329 A similar approach was used for the identification of two sugar conjugates, glycoside of  
330 hydroxylated metabolite (Fen-OH-glycoside) and glycoside of fenhexamid (Fen-glycoside).  
331 Peaks with possible elemental composition  $C_{20}H_{27}Cl_2NO_8$  ( $m/z$  480.1187,  $[M+H]^+$ ) and  
332  $C_{20}H_{27}Cl_2NO_7$  ( $m/z$  464.1237,  $[M+H]^+$ ) were detected in the chromatogram at 8.6 min and at  
333 7.45 min, respectively. The identity of both conjugates was confirmed by matching of the  
334 fragment ions with fenhexamid or Fen-OH respectively, and by a neutral loss of hexose ( $[M-$   
335  $C_6H_9O_5]^+$ ;  $\Delta m/z$  162.0535 Da) in the fragmentation mass spectra.

336 As expected, in reversed phase chromatography all the tentatively identified metabolites of  
337 fenhexamid showed shorter retention times, compared to the parent pesticide, due to their  
338 more polar nature. (**Figure 3**).

### 339 **3.2. Validation**

340 Performance characteristics obtained within the validation method of parent pesticide  
341 compound in various matrices are summarized in **Table 4**. Recoveries were in the range of  
342 81-98 % and repeatabilities, expressed as relative standard deviation (RSD; %), were  $\leq 16$  % in  
343 all tested matrices. Within-laboratory reproducibilities were in the range of 5-18 %. Limits of  
344 quantification (LOQ) were in the range of 0.001-0.1 mg kg<sup>-1</sup> for grapes and wine, and in the  
345 range of 0.004-0.4 mg kg<sup>-1</sup> for vine leaves.<sup>9</sup>

346 In the case of pesticide metabolites, the recovery values could not be determined, since, as  
347 mentioned earlier, pure standards are not commercially available. Therefore, only  
348 repeatabilities of measurements (repeatabilities of metabolites' signal intensities) were  
349 calculated as relative standard deviations of 6 replicate injections of QuEChERS extracts  
350 prepared from samples containing incurred metabolites. The obtained results of metabolites  
351 in vine leaves and grapes are summarized in **Table 5**.

### 352 **3.3. Changes in pesticide residues and levels of their metabolites**

353 The key objective of this study was to obtain further knowledge on the changes of pesticide  
354 residues and the origin of their metabolite levels in various parts of grapevine, including  
355 grapes, and during the wine-making process. With regard to the impossibility of quantifying  
356 metabolites' concentration, 'Response', the ratio between the area of detected metabolite  
357 and area of internal standard (TPP in positive ionization mode and nicarbazin in negative  
358 ionization mode) was used for illustration of the concentration trends.

#### 359 **3.3.1. Vine leaves and wine grapes**



360 Following field treatment (a detailed description is in **Table 1**), 9 out of 13 LC-MS amenable  
361 pesticide residues, together with their conceivable metabolites, were monitored in the  
362 collected samples. Degradation of particular pesticide residues in vine leaves was monitored  
363 during 8-18 weeks, based on their application in the vineyard. Sampling was performed  
364 always on the 1st to 4th day after each pesticide treatment. The last five weeks of a  
365 sampling period in the vineyard, and samples of wine grapes were collected, together with  
366 samples of vine leaves. The complete results of pesticide residue analysis in vine leaves and  
367 wine grapes are summarized in the **Supporting information – Figure S1**.

368 Overall, concentrations of pesticide residues in vine leaves significantly dropped in the first  
369 week after specific treatment; the decrease of residues of dimethomorph, spiroxamin,  
370 tebuconazole and triadimenol was in the range of 27-40 %. In the case of fenhexamid,  
371 iprovalicarb, metrafenone, meptyldinocap, pyraclostrobin and quinoxyfen residues, it was  
372 even higher, 60-69 %. In subsequent weeks, the decline was somewhat slower.  
373 Nevertheless, with the exception of triadimenol, residues of applied fungicide were still  
374 detectable in the vine leaves.

375 At the wine grape harvest, the last sampling was performed in the vineyard. In these  
376 samples of vine leaves and wine grapes, all residues were detected except for triadimenol.  
377 All residue levels decreased by 84-100 %.

378 Pesticide metabolites were already detectable in vine leaves collected within the first  
379 sampling after the treatment. The spectrum of metabolites found in vine leaves and wine  
380 grapes was similar in the case of fenhexamid, iprovalicarb and spiroxamine. Metabolites of  
381 tebuconazole were detected in vine leaves only. In the case of dimethomorph, the pattern of  
382 its metabolites differed between grapes and vine leaves. However, dimethomorph-demethyl  
383 was found in both these matrices and, moreover, in vine leaves another Phase I metabolite,

384 dimethomorph-hydroxy, was also present. Regarding wine grapes, in addition to  
385 dimethomorph-demethyl, a Phase II metabolite, dimethomorph-demethyl glycoside, was  
386 also detected.

387 The obtained results for all tentatively identified metabolites are summarized in **Supporting**  
388 **information – Figures S3-S8**. To compare the changes in levels of metabolites and parent  
389 pesticides, the degradation dynamic of the parent pesticide is also displayed.

390 As an example, the data for metabolites of fenhexamid in both types of samples are shown  
391 in **Figure 4**. Fen-OH was the metabolite of fenhexamid with the highest signal intensity.  
392 Nevertheless, the response of fenhexamid was higher than the response of its metabolites in  
393 all tested samples. For instance, in samples collected at harvest time, the response of  
394 fenhexamid was 5 times (in wine grapes) and 10 times (in vine leaves) higher than the  
395 response of Fen-OH. In samples collected 8 weeks after the treatment, the response of Fen-  
396 OH and Fen-OH glycoside in vine leaves slightly increased. On the other hand, the response  
397 of the third metabolite, Fen-glycoside, decreased (it is noteworthy that the background of  
398 co-eluting compounds did not change, thus the analyte decrease is not due to stronger  
399 matrix effects). As the biotransformation of pesticides in plants is a complex of enzymatic  
400 reactions, Fen-glycoside might be an intermediate product of a metabolic pathway, resulting  
401 in the formation of the Phase III metabolites.

### 402 **3.3.2. Wine-making process**

403 We also attempted to find whether, and to what extent, pesticide residues and their  
404 metabolites are transferred from grapes into wine and could be used as markers of illegal  
405 practices in organic vineyards. For this purpose, samples were collected across the wine-  
406 making process. The obtained data are summarized in **Supporting information – Figure S2**.  
407 Residues of some pesticides, such as fenhexamid, iprovalicarb, pyraclostrobin in the juice

408 obtained after the crushing and pressing of grapes, did not significantly decrease when  
409 compared with their levels in processed grapes. On the other hand, a remarkable reduction  
410 of dimethomorph occurred. After clarification of must, concentrations of dimethomorph,  
411 fenhexamid and iprovalicarb decreased by 59-67 %. No detectable residues of metrafenone,  
412 pyraclostrobin, quinoxyfen, spiroxamine and tebuconazole were present in clarified must.  
413 The elimination of residues corresponds to the physicochemical properties of the respective  
414 pesticide. In general terms, less water soluble pesticides with a low octanol-water partition  
415 coefficient (*K<sub>ow</sub>*) are not significantly transferred into wine, as they are mainly absorbed in  
416 solid waste (seeds and skins).

417 During fermentation of must, the concentrations of three pesticide residues (dimethomorph,  
418 fenhexamid and iprovalicarb) did not show any significant changes. In the final product –  
419 bottled white wine – pesticide residues were detected in the same quantities as in  
420 fermenting must. None of the processes following clarification resulted in considerable  
421 changes in the residues of parent pesticides.

422 As described above, crushing, juicing of wine grapes and clarification of must led to reduced  
423 concentrations of parent residues. These processes had a significant influence on the  
424 metabolites of metrafenone, pyraclostrobin and spiroxamine; they were separated from  
425 must as well as the parent compounds.

426 In samples collected during the wine-making process, metabolites of fenhexamid and  
427 iprovalicarb were detected. Changes in the levels of parent compounds and their  
428 metabolites are documented in **Figure 5**. In the case of fenhexamid, the response of Fen-OH  
429 significantly increased and the response of glycosylated metabolites declined. These changes  
430 were probably caused by hydrolysis of glycosidic bonds in the molecules of Fen-glycoside  
431 and Fen-OH glycoside by yeast enzymes during the fermentation of must. Analysis of the

432 metabolites of iprovalicarb showed different results. Levels of both tentatively identified  
433 metabolites showed no significant changes during the wine-making process. Based on the  
434 results shown in **Figure 5**, clarification of must resulted in a decrease of levels of parent  
435 pesticides, but not their metabolites. All five metabolites were also detected in the final  
436 product – white wine.

437 In conclusion, detection of pesticide metabolites in matrices with very low or non-detectable  
438 residues of parent pesticides may indicate illegal practices in organic farming. Nevertheless,  
439 for an unbiased conclusion whether contamination is not due to a drift of pesticides,  
440 quantification of the respective metabolites might be helpful, as it would enable an  
441 estimation of the earlier pesticide burden. However, for this purpose, the availability of pure  
442 analytical standards of pesticide metabolites would be needed.

443 The strategy of pesticide metabolite screening could generically be employed for various  
444 other plant matrices and products thereof, including *e.g.* baby food where high standards  
445 regarding the quality of raw materials are required.

446 Also worthy of note is the fact that the knowledge on the occurrence of pesticide  
447 metabolites in the human diet should be taken into consideration in biomonitoring studies,  
448 where such compounds occurring in biological fluids are considered as exposure markers to  
449 respective pesticides. The exposure to parent compounds might be overestimated,  
450 supposing that the product of biotransformation of the parent compound in the human  
451 body and that associated with (nontoxic) metabolites in the diet are identical.

452

453 **Abbreviations used:**

454 Fen-OH        fenhexamid-hydroxy

455 PPP            plant protection product

456 **Associated contents**

457 *Supporting information*

458 Complete results of pesticide residue analysis in vine leaves, wine grapes and musts/wines  
459 **(Figure S1 – S2)** and changes in levels of specific pesticide metabolites in vine leaves and  
460 wine grapes **(Figures S3 – S8)**.

461

462 **Acknowledgements**

463 This work was supported by “Complex strategies for effective detection of food fraud in the  
464 chain production – consumer“ (QJ1530272) from the Ministry of Agriculture. This work was  
465 supported by the “Operational Programme Prague – Competitiveness”  
466 (CZ.2.16/3.1.00/21537 and CZ.2.16/3.1.00/24503) and the “National Programme of  
467 Sustainability I” - NPU I (LO1601 - No.: MSMT-43760/2015), as well as by financial support  
468 from specific university research (MSMT No 21-SVV/2018).

469

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529



530 **FIGURE CAPTIONS**

531 **Figure 1** Comparison of HRMS signal intensities (peak area, mass window +/- 5ppm) for the  
532 target pesticide metabolites in wine grape extracts prepared by various procedures  
533 (reference: QuEChERS value = 100 %), n = 3. "Y error bars" represent repeatability of the  
534 respective procedure in %.

535 **Figure 2** MS/MS fragmentation mass spectra of fenhexamid and its metabolites (collision  
536 energy 20 V)

537 **Figure 3** Overlaid extracted ion chromatograms of fenhexamid (*m/z* 302.0709) and  
538 metabolites Fen-OH (*m/z* 318.0658), Fen-glycoside (*m/z* 464.1237) and Fen-OH-glycoside  
539 (*m/z* 480.1187)

540 **Figure 4** Changes in concentration of fenhexamid and levels of its metabolites in (A) vine  
541 leaves and (B) wine grapes

542 **Figure 5** Changes in concentrations of fenhexamid, iprovalicarb and levels of their  
543 metabolites during the wine-making process

544

## 545 TABLES

546 Table 1 Vineyard treatment by pesticide preparations and sampling of tested materials

Day of experiment	Type of action	Type of sample*	Characterization of pesticide treatment		
			Trade name of PPP	Common name of active ingredients (activity**)	Doses of applied a.i.
day 1 (4/16/2015)	pesticide treatment	-	Masai	tebufenpyrad (S)	0.20 kg·ha <sup>-1</sup>
36	pesticide treatment	-	Kumulus WG	sulfur (C)	3.00 kg·ha <sup>-1</sup>
36	sampling	VL	Polyram WG	metiram (C)	1.10 kg·ha <sup>-1</sup>
43	sampling	VL	X		
48	pesticide treatment	-	Cabrio Top	metiram (C), pyraclostrobin (S)	1.00 kg·ha <sup>-1</sup>
50	sampling	VL	X		
57	sampling	VL	X		
60	pesticide treatment	-	Melody Comby 65.3 WG	iprovalicarb (S), folpet (C)	2.00 kg·ha <sup>-1</sup>
64	sampling	VL	Vivando	metrafenone (S)	0.32 L·ha <sup>-1</sup>
64	sampling	VL	X		
69	pesticide treatment	-	Acrobat MZ	mancozeb (C), dimethomorph (QS)	2.50 kg·ha <sup>-1</sup>
71	sampling	VL	Falcon 460 EC	tebuconazole (S), triadimenol (S), spiroxamine (S)	0.50 L·ha <sup>-1</sup>
71	sampling	VL	X		
78	pesticide treatment	-	Karathane NEW	meptyldinocap (C)	0.50 L·ha <sup>-1</sup>
78	sampling	VL	Melody Comby 65.3 WG	iprovalicarb (S), folpet (C)	2.00 kg·ha <sup>-1</sup>
78	sampling	VL	X		
85	pesticide treatment	-	IQ-Crystal	quinoxifen (S)	0.15 L·ha <sup>-1</sup>
85	sampling	VL	Kumulus WG	sulfur (C)	3.00 kg·ha <sup>-1</sup>
85	sampling	VL	Flowbrix	copper oxychloride (C)	3.00 L·ha <sup>-1</sup>
92	sampling	VL	X		
97	pesticide treatment	-	Cabrio Top	metiram (C), pyraclostrobin (S)	2.00 kg·ha <sup>-1</sup>
99	sampling	VL	X		
105	sampling	VL	X		
106	pesticide treatment	-	Karathane NEW	meptyldinocap (C)	0.50 L·ha <sup>-1</sup>
113	sampling	VL	Melody Comby 65.3 WG	iprovalicarb (S), folpet (C)	2.00 kg·ha <sup>-1</sup>
113	sampling	VL	X		
119	pesticide treatment	-	IQ-Crystal	quinoxifen (S)	0.15 L·ha <sup>-1</sup>
119	pesticide treatment	-	Teldor 500 SC	fenhexamid (QS)	1.00 L·ha <sup>-1</sup>
119	pesticide treatment	-	Alliette Bordeaux	copper oxychloride (C), fosetyl-Al (S)	4.0 kg·ha <sup>-1</sup>
120	sampling	VL	X		
127	sampling	VL			
134	sampling	VL			
141	sampling	VL			
148	sampling	VL+WG			
155	sampling	VL+WG			
162	sampling	VL+WG			
169	sampling	VL+WG			
176	sampling	VL+WG			
180	sampling	J			
187	sampling	M			
194	sampling	M			
201	sampling	M			
208	sampling	M			
216	sampling	M			
222	sampling	M			
236	sampling	M			
249	sampling	M			
292	sampling	YW			
412	sampling	W			

547 \* VL – vine leaves; WG – wine grapes; J – grape juice; M – must; YW – young wine; W – wine

548 \*\* C – contact; S – systemic; QS – quasi-systemic; X – not relevant

549 **Table 2** Library of screened pesticide residues and their metabolites

#	analyte (parent pesticide and its metabolite)	elemental composition	ref.	#	analyte (parent pesticide and its metabolite)	elemental composition	ref.	
1	<b>DIMETHOMORPH</b>	<b>C<sub>21</sub>H<sub>22</sub>ClNO<sub>4</sub></b>	13	6	<b>QUINOXYFEN</b>	<b>C<sub>15</sub>H<sub>8</sub>Cl<sub>2</sub>FNO</b>	14	
1a	dimethomorph-demethyl	C <sub>20</sub> H <sub>20</sub> ClNO <sub>4</sub>		6a	3-hydroxy-quinoxifen	C <sub>15</sub> H <sub>8</sub> Cl <sub>2</sub> FNO <sub>2</sub>		
1b	dimethomorph-demethyl glycoside	C <sub>26</sub> H <sub>30</sub> ClNO <sub>9</sub>		6b	CFBPQ	C <sub>15</sub> H <sub>7</sub> ClFNO		
1c	dimethomorph-Z7	C <sub>15</sub> H <sub>13</sub> ClNO <sub>3</sub>		7	<b>SPIROXAMINE</b>	<b>C<sub>18</sub>H<sub>35</sub>NO<sub>2</sub></b>	15	
1d	dimethomorph-Z37	C <sub>21</sub> H <sub>20</sub> ClNO <sub>5</sub>		7a	spiroxamine-N-oxide	C <sub>18</sub> H <sub>35</sub> NO <sub>3</sub>		
1e	dimethomorph-hydroxy	C <sub>21</sub> H <sub>22</sub> ClNO <sub>5</sub>		7b	spiroxamine-N-desethyl	C <sub>16</sub> H <sub>31</sub> NO <sub>2</sub>		
2	<b>FENHEXAMID</b>	<b>C<sub>14</sub>H<sub>17</sub>Cl<sub>2</sub>NO<sub>2</sub></b>	12	7c	spiroxamine-N-despropyl	C <sub>15</sub> H <sub>29</sub> NO <sub>2</sub>		
2a	fenhexamid-glycoside	C <sub>20</sub> H <sub>27</sub> Cl <sub>2</sub> NO <sub>7</sub>		7d	spiroxamine-cyclohexanol	C <sub>10</sub> H <sub>20</sub> O		
2b	fenhexamid-hydroxy	C <sub>14</sub> H <sub>17</sub> Cl <sub>2</sub> NO <sub>3</sub>		7e	spiroxamine-cyclohexanol glycoside	C <sub>16</sub> H <sub>30</sub> O <sub>6</sub>		
2c	fenhexamid-hydroxy glycoside	C <sub>20</sub> H <sub>27</sub> Cl <sub>2</sub> NO <sub>8</sub>		7f	spiroxamine-diol	C <sub>10</sub> H <sub>20</sub> O		
3	<b>IPROVALICARB</b>	<b>C<sub>18</sub>H<sub>28</sub>N<sub>2</sub>O<sub>3</sub></b>		16	7g	spiroxamine-diol glycoside	C <sub>16</sub> H <sub>30</sub> O <sub>6</sub>	
3a	iprovalicarb-hydroxy	C <sub>18</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub>	8		<b>TEBUCONAZOLE</b>	<b>C<sub>16</sub>H<sub>22</sub>ClN<sub>3</sub>O</b>	17	
3b	iprovalicarb-hydroxy glycoside	C <sub>24</sub> H <sub>38</sub> N <sub>2</sub> O <sub>9</sub>	18	8a	tebuconazole-hydroxy	C <sub>16</sub> H <sub>22</sub> ClN <sub>3</sub> O <sub>2</sub>		
4	<b>METRAFENONE</b>	<b>C<sub>19</sub>H<sub>21</sub>BrO<sub>5</sub></b>		8b	tebuconazole-hydroxy glycoside	C <sub>22</sub> H <sub>32</sub> ClN <sub>3</sub> O <sub>7</sub>		
4a	metrafenone CL 1500836	C <sub>19</sub> H <sub>20</sub> O <sub>6</sub>		9	<b>TRIADIMENOL</b>	<b>C<sub>14</sub>H<sub>18</sub>ClN<sub>3</sub>O<sub>2</sub></b>	19	
4b	metrafenone CL 3000402	C <sub>19</sub> H <sub>19</sub> BrO <sub>6</sub>			9a	triadimenol glycoside		C <sub>20</sub> H <sub>28</sub> ClN <sub>3</sub> O <sub>7</sub>
4c	metrafenone CL 379395	C <sub>19</sub> H <sub>19</sub> BrO <sub>6</sub>			9b	triadimenol-hydroxy		C <sub>14</sub> H <sub>18</sub> ClN <sub>3</sub> O <sub>3</sub>
4d	metrafenone CL 197675	C <sub>19</sub> H <sub>19</sub> BrO <sub>7</sub>	9c		triadimenol-hydroxy glycoside	C <sub>20</sub> H <sub>28</sub> ClN <sub>3</sub> O <sub>8</sub>		
5	<b>PYRACLOSTROBIN</b>	<b>C<sub>19</sub>H<sub>18</sub>ClN<sub>3</sub>O<sub>4</sub></b>	20					
5a	pyraclostrobin-hydroxy	C <sub>19</sub> H <sub>18</sub> ClN <sub>3</sub> O <sub>5</sub>						
5b	pyraclostrobin-desmethoxy	C <sub>18</sub> H <sub>16</sub> ClN <sub>3</sub> O <sub>3</sub>						
5c	pyraclostrobin-hydroxy glycoside	C <sub>25</sub> H <sub>28</sub> ClN <sub>3</sub> O <sub>10</sub>						

550 **Table 3** List of targeted pesticide residues and their tentatively identified metabolites in

551 tested samples (\* WG – wine grapes; VL – vine leaves; M – must; W – wine)

#	analyte (parent pesticide and its metabolite)	elemental composition	ion type	detected ion (MS <sup>1</sup> )	measured <i>m/z</i> of fragments (MS <sup>2</sup> )	retention time [min]	sample type*
1	<b>DIMETHOMORPH</b>	<b>C<sub>21</sub>H<sub>22</sub>ClNO<sub>4</sub></b>	[M+H] <sup>+</sup>	388.1310	301.0650 (C <sub>17</sub> H <sub>14</sub> ClO <sub>3</sub> ) 273.0655 (C <sub>16</sub> H <sub>14</sub> ClO <sub>2</sub> ) 165.0541 (C <sub>9</sub> H <sub>9</sub> O <sub>3</sub> ) 114.0546 (C <sub>5</sub> H <sub>8</sub> NO <sub>2</sub> ) 70.0271 (C <sub>3</sub> H <sub>4</sub> NO)	8.9; 9.1	WG; VL; M; W
1a	dimethomorph-demethyl	C <sub>20</sub> H <sub>20</sub> ClNO <sub>4</sub>	[M+H] <sup>+</sup>	374.1154	287.0480 (C <sub>16</sub> H <sub>12</sub> ClO <sub>3</sub> ) 151.0378 (C <sub>8</sub> H <sub>7</sub> O <sub>3</sub> ) 114.0546 (C <sub>5</sub> H <sub>8</sub> NO <sub>2</sub> ) 70.0271 (C <sub>3</sub> H <sub>4</sub> NO)	8.5; 8.7	WG; VL
1b	dimethomorph-demethyl glycoside	C <sub>26</sub> H <sub>30</sub> ClNO <sub>9</sub>	[M+H] <sup>+</sup>	536.1682	374.1154 (C <sub>20</sub> H <sub>21</sub> ClNO <sub>4</sub> ) 287.0480 (C <sub>16</sub> H <sub>12</sub> ClO <sub>3</sub> ) 151.0378 (C <sub>8</sub> H <sub>7</sub> O <sub>3</sub> ) 70.0271 (C <sub>3</sub> H <sub>4</sub> NO)	7.7	WG; VL
1e	dimethomorph-hydroxy	C <sub>21</sub> H <sub>22</sub> ClNO <sub>5</sub>	[M+H] <sup>+</sup>	404.1259	386.1135 (C <sub>21</sub> H <sub>21</sub> ClNO <sub>4</sub> ) 317.0558 (C <sub>17</sub> H <sub>14</sub> ClO <sub>4</sub> ) 289.0614 (C <sub>16</sub> H <sub>14</sub> ClO <sub>3</sub> ) 165.0541 (C <sub>9</sub> H <sub>9</sub> O <sub>3</sub> ) 114.0546 (C <sub>5</sub> H <sub>8</sub> NO <sub>2</sub> ) 70.0271 (C <sub>3</sub> H <sub>4</sub> NO)	8.1; 8.3; 8.5	VL
2	<b>FENHEXAMID</b>	<b>C<sub>14</sub>H<sub>17</sub>Cl<sub>2</sub>NO<sub>2</sub></b>	[M+H] <sup>+</sup>	302.0709	177.9817 (C <sub>6</sub> H <sub>6</sub> Cl <sub>2</sub> NO) 143.0124 (C <sub>6</sub> H <sub>6</sub> ClNO) 97.1008 (C <sub>7</sub> H <sub>13</sub> ) 55.0525 (C <sub>4</sub> H <sub>7</sub> )	9.4	WG; VL; M; W
			[M-H] <sup>-</sup>	300.0564	264.0796 (C <sub>14</sub> H <sub>15</sub> ClNO <sub>2</sub> ) 249.0558 (C <sub>13</sub> H <sub>12</sub> ClNO <sub>2</sub> ) 221.0241 (C <sub>11</sub> H <sub>8</sub> ClNO <sub>2</sub> )	9.3	

#	analyte (parent pesticide and its metabolite)	elemental composition	ion type	detected ion (MS <sup>1</sup> )	measured m/z of fragments (MS <sup>2</sup> )	retention time [min]	sample type*
2a	fenhexamid-glycoside	C <sub>20</sub> H <sub>27</sub> Cl <sub>2</sub> NO <sub>7</sub>	[M+H] <sup>+</sup>	464.1237	302.0702 (C <sub>14</sub> H <sub>18</sub> Cl <sub>2</sub> NO <sub>2</sub> ) 177.9817 (C <sub>6</sub> H <sub>6</sub> Cl <sub>2</sub> NO) 143.0124 (C <sub>6</sub> H <sub>6</sub> ClNO) 97.1008 (C <sub>7</sub> H <sub>13</sub> ) 55.0525 (C <sub>4</sub> H <sub>7</sub> )	8.7	WG; VL; M; W
2b	fenhexamid-hydroxy	C <sub>14</sub> H <sub>17</sub> Cl <sub>2</sub> NO <sub>3</sub>	[M+H] <sup>+</sup>	318.0658	300.0549 (C <sub>14</sub> H <sub>16</sub> Cl <sub>2</sub> NO <sub>2</sub> ) 175.9651 (C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> NO) 113.0961(C <sub>7</sub> H <sub>13</sub> O) 97.1008 (C <sub>7</sub> H <sub>13</sub> ) 55.0525 (C <sub>4</sub> H <sub>7</sub> )	8.6	WG; VL; M; W
			[M-H] <sup>-</sup>	316.0513	280.0730 (C <sub>14</sub> H <sub>15</sub> ClNO <sub>3</sub> ) 237.0710 (C <sub>11</sub> H <sub>8</sub> ClNO <sub>3</sub> )	8.5	
2c	fenhexamid-hydroxy glycoside	C <sub>20</sub> H <sub>27</sub> Cl <sub>2</sub> NO <sub>8</sub>	[M+H] <sup>+</sup>	480.1187	318.0646 (C <sub>14</sub> H <sub>18</sub> Cl <sub>2</sub> NO <sub>3</sub> ) 300.0549 (C <sub>14</sub> H <sub>16</sub> Cl <sub>2</sub> NO <sub>2</sub> ) 175.9651 (C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> NO) 113.0961(C <sub>7</sub> H <sub>13</sub> O) 97.1008 (C <sub>7</sub> H <sub>13</sub> ) 55.0525 (C <sub>4</sub> H <sub>7</sub> )	7.6	WG; VL; M; W
3	IPROVALICARB	C <sub>18</sub> H <sub>28</sub> N <sub>2</sub> O <sub>3</sub>	[M+H] <sup>+</sup>	321.2173	144.0644 (C <sub>6</sub> H <sub>10</sub> NO <sub>3</sub> ) 119.0852 (C <sub>9</sub> H <sub>11</sub> ) 116.0700 (C <sub>5</sub> H <sub>10</sub> NO <sub>2</sub> ) 98.0591 (C <sub>5</sub> H <sub>8</sub> NO) 91.0533 (C <sub>7</sub> H <sub>7</sub> ) 72.0797 (C <sub>4</sub> H <sub>10</sub> N)	9.4	WG; VL; M; W
			[M+CH <sub>3</sub> COO] <sup>-</sup>	319.2027	259.1470 (C <sub>15</sub> H <sub>19</sub> N <sub>2</sub> O <sub>2</sub> ) 216.0911 (C <sub>12</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> ) 97.0040 (C <sub>3</sub> HN <sub>2</sub> O <sub>2</sub> ) 59.0128 (C <sub>2</sub> H <sub>3</sub> O <sub>2</sub> )	9.3	
3a	iprovalicarb-hydroxy	C <sub>18</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub>	[M+H] <sup>+</sup>	337.2122	319.1811 (C <sub>18</sub> H <sub>27</sub> N <sub>2</sub> O <sub>3</sub> ) 144.0644 (C <sub>6</sub> H <sub>10</sub> NO <sub>3</sub> ) 135.0800 (C <sub>9</sub> H <sub>11</sub> O) 116.0696 (C <sub>5</sub> H <sub>10</sub> NO <sub>2</sub> ) 98.0591 (C <sub>5</sub> H <sub>8</sub> NO) 72.0797 (C <sub>4</sub> H <sub>10</sub> N)	7.9; 8.1	WG; VL; M; W
3b	iprovalicarb-hydroxy glycoside	C <sub>24</sub> H <sub>38</sub> N <sub>2</sub> O <sub>9</sub>	[M+H] <sup>+</sup>	499.2650	337.2099 (C <sub>18</sub> H <sub>29</sub> N <sub>2</sub> O <sub>4</sub> ) 319.1811 (C <sub>18</sub> H <sub>27</sub> N <sub>2</sub> O <sub>3</sub> ) 144.0644 (C <sub>6</sub> H <sub>10</sub> NO <sub>3</sub> ) 135.0800 (C <sub>9</sub> H <sub>11</sub> O) 116.0696 (C <sub>5</sub> H <sub>10</sub> NO <sub>2</sub> ) 98.0591 (C <sub>5</sub> H <sub>8</sub> NO) 72.0797 (C <sub>4</sub> H <sub>10</sub> N)	7.3; 7.5	WG; VL; M; W
4	METRAFENONE	C <sub>19</sub> H <sub>21</sub> BrO <sub>5</sub>	[M+H] <sup>+</sup>	409.0645	226.9706 (C <sub>9</sub> H <sub>8</sub> BrO <sub>2</sub> ) 209.0808 (C <sub>11</sub> H <sub>13</sub> O <sub>4</sub> ) 194.0563 (C <sub>10</sub> H <sub>10</sub> O <sub>4</sub> ) 166.0626 (C <sub>6</sub> H <sub>10</sub> O <sub>3</sub> )	10.1	WG; VL
4a	metrafenone CL 1500836	C <sub>19</sub> H <sub>20</sub> O <sub>6</sub>	[M+H] <sup>+</sup>	345.1333	253.0837 (C <sub>16</sub> H <sub>13</sub> O <sub>3</sub> ) 181.0849 (C <sub>10</sub> H <sub>13</sub> O <sub>3</sub> ) 165.0545 (C <sub>9</sub> H <sub>9</sub> O <sub>3</sub> ) 163.0387 (C <sub>9</sub> H <sub>7</sub> O <sub>3</sub> )	8.5	VL
4b	metrafenone CL 3000402	C <sub>19</sub> H <sub>19</sub> BrO <sub>6</sub>	[M+H] <sup>+</sup>	423.0438	393.0310 (C <sub>18</sub> H <sub>18</sub> BrO <sub>5</sub> ) 268.1079 (C <sub>17</sub> H <sub>16</sub> O <sub>3</sub> ) 242.9640 (C <sub>9</sub> H <sub>8</sub> BrO <sub>3</sub> ) 240.9500 (C <sub>9</sub> H <sub>6</sub> BrO <sub>3</sub> ) 212.9530 (C <sub>8</sub> H <sub>6</sub> BrO <sub>2</sub> )	9.7	WG; VL
4c	metrafenone CL 379395	C <sub>19</sub> H <sub>19</sub> BrO <sub>6</sub>	[M+H] <sup>+</sup>	423.0438	226.9674 (C <sub>9</sub> H <sub>8</sub> BrO <sub>2</sub> ) 225.0758 (C <sub>11</sub> H <sub>13</sub> O <sub>5</sub> ) 223.0596 (C <sub>11</sub> H <sub>11</sub> O <sub>5</sub> ) 212.9909 (C <sub>9</sub> H <sub>10</sub> BrO)	8.9	WG; VL

#	analyte (parent pesticide and its metabolite)	elemental composition	ion type	detected ion (MS <sup>1</sup> )	measured <i>m/z</i> of fragments (MS <sup>2</sup> )	retention time [min]	sample type*
					195.0648 (C <sub>10</sub> H <sub>11</sub> O <sub>4</sub> )		
5	PYRACLOSTROBIN	C <sub>19</sub> H <sub>18</sub> ClN <sub>3</sub> O <sub>4</sub>	[M+H] <sup>+</sup>	388.1059	324.0523 (C <sub>17</sub> H <sub>11</sub> ClN <sub>3</sub> O <sub>2</sub> ) 296.0585 (C <sub>16</sub> H <sub>11</sub> ClN <sub>3</sub> O) 194.0811 (C <sub>10</sub> H <sub>12</sub> NO <sub>3</sub> ) 163.0628 (C <sub>9</sub> H <sub>9</sub> NO <sub>2</sub> ) 149.0468 (C <sub>8</sub> H <sub>7</sub> NO <sub>2</sub> ) 133.0517 (C <sub>8</sub> H <sub>7</sub> NO)	9.9	WG; VL
5a	pyraclostrobin-hydroxy	C <sub>19</sub> H <sub>18</sub> ClN <sub>3</sub> O <sub>5</sub>	[M+H] <sup>+</sup>	404.1008	312.0469 (C <sub>16</sub> H <sub>11</sub> ClN <sub>3</sub> O <sub>2</sub> ) 194.0811 (C <sub>10</sub> H <sub>12</sub> NO <sub>3</sub> ) 163.0628 (C <sub>9</sub> H <sub>9</sub> NO <sub>2</sub> ) 149.0468 (C <sub>8</sub> H <sub>7</sub> NO <sub>2</sub> ) 133.0517 (C <sub>8</sub> H <sub>7</sub> NO)	9.9	WG; VL
			[M-H] <sup>-</sup>	402.0862	208.0045 (C <sub>9</sub> H <sub>5</sub> ClN <sub>2</sub> O <sub>2</sub> ) 164.0134 (C <sub>11</sub> H <sub>12</sub> NO) 157.0006 (C <sub>6</sub> H <sub>4</sub> ClNO <sub>2</sub> )	9.9	
5b	pyraclostrobin-desmethoxy	C <sub>18</sub> H <sub>16</sub> ClN <sub>3</sub> O <sub>3</sub>	[M+H] <sup>+</sup>	358.0953	326.0677 (C <sub>17</sub> H <sub>13</sub> ClN <sub>3</sub> O <sub>2</sub> ) 298.0585 (C <sub>16</sub> H <sub>13</sub> ClN <sub>3</sub> O) 164.0704 (C <sub>9</sub> H <sub>10</sub> NO <sub>2</sub> ) 132.0434 (C <sub>8</sub> H <sub>6</sub> NO)	9.9	WG; VL
5c	pyraclostrobin-hydroxy glycoside	C <sub>25</sub> H <sub>28</sub> ClN <sub>3</sub> O <sub>10</sub>	[M+H] <sup>+</sup>	566.1536	404.1008 C <sub>19</sub> H <sub>19</sub> ClN <sub>3</sub> O <sub>5</sub> 312.0469 (C <sub>16</sub> H <sub>11</sub> ClN <sub>3</sub> O <sub>2</sub> ) 194.0811 (C <sub>10</sub> H <sub>12</sub> NO <sub>3</sub> ) 163.0628 (C <sub>9</sub> H <sub>9</sub> NO <sub>2</sub> ) 149.0468 (C <sub>8</sub> H <sub>7</sub> NO <sub>2</sub> ) 133.0517 (C <sub>8</sub> H <sub>7</sub> NO)	8.6	VL
6	SPIROXAMINE	C <sub>18</sub> H <sub>35</sub> NO <sub>2</sub>	[M+H] <sup>+</sup>	298.2741	144.1345 (C <sub>8</sub> H <sub>18</sub> NO) 100.1083 (C <sub>6</sub> H <sub>14</sub> N) 72.0795 (C <sub>4</sub> H <sub>10</sub> N)	8.9	WG; VL
6a	spiroxamine- <i>N</i> -oxide	C <sub>18</sub> H <sub>35</sub> NO <sub>3</sub>	[M+H] <sup>+</sup>	314.2690	160.1328 (C <sub>8</sub> H <sub>18</sub> NO <sub>2</sub> ) 130.1218 (C <sub>7</sub> H <sub>16</sub> NO) 100.1112 (C <sub>6</sub> H <sub>14</sub> N) 88.0750 (C <sub>4</sub> H <sub>10</sub> NO)	9.0; 9.2; 9.3	WG; VL
6b	spiroxamine- <i>N</i> -desethyl	C <sub>16</sub> H <sub>31</sub> NO <sub>2</sub>	[M+H] <sup>+</sup>	270.2428	116.1066 (C <sub>6</sub> H <sub>14</sub> NO) 72.0810 (C <sub>4</sub> H <sub>10</sub> N)	8.7	WG; VL
6c	spiroxamine- <i>N</i> -despropyl	C <sub>15</sub> H <sub>29</sub> NO <sub>2</sub>	[M+H] <sup>+</sup>	256.2271	102.0909 (C <sub>5</sub> H <sub>12</sub> NO) 84.0797 (C <sub>5</sub> H <sub>10</sub> N) 58.0639 (C <sub>3</sub> H <sub>8</sub> N)	8.4	WG; VL
7	TEBUCONAZOLE	C <sub>16</sub> H <sub>22</sub> ClN <sub>3</sub> O	[M+H] <sup>+</sup>	308.1524	151.0312 (C <sub>9</sub> H <sub>8</sub> Cl) 139.0285 (C <sub>8</sub> H <sub>8</sub> Cl) 125.0147 (C <sub>7</sub> H <sub>6</sub> Cl) 70.0390 (C <sub>2</sub> H <sub>4</sub> N <sub>3</sub> ) 57.0704 (C <sub>4</sub> H <sub>5</sub> )	9.8	WG; VL
			[M-H] <sup>-</sup>	306.1379	223.0911 (C <sub>13</sub> H <sub>18</sub> ClO) 82.0407 (C <sub>3</sub> H <sub>4</sub> N <sub>3</sub> ) 68.0255 (C <sub>2</sub> H <sub>2</sub> N <sub>3</sub> )	9.7	
7a	tebuconazole-hydroxy	C <sub>16</sub> H <sub>22</sub> ClN <sub>3</sub> O <sub>2</sub>	[M+H] <sup>+</sup>	324.1473	141.0078 (C <sub>7</sub> H <sub>6</sub> ClO) 125.0147 (C <sub>7</sub> H <sub>6</sub> Cl) 70.0390 (C <sub>2</sub> H <sub>4</sub> N <sub>3</sub> )	8.9, 9.4; 9.7	WG; VL
			[M-H] <sup>-</sup>	322.1334	239.0838 (C <sub>13</sub> H <sub>18</sub> ClO <sub>2</sub> ) 223.0911 (C <sub>13</sub> H <sub>18</sub> ClO) 68.0255 (C <sub>2</sub> H <sub>2</sub> N <sub>3</sub> )	9.2; 9.5	
7b	tebuconazole-hydroxy glycoside	C <sub>22</sub> H <sub>32</sub> ClN <sub>3</sub> O <sub>7</sub>	[M+H] <sup>+</sup>	486.2002	324.1481 (C <sub>16</sub> H <sub>23</sub> ClN <sub>3</sub> O <sub>2</sub> ) 141.0078 (C <sub>7</sub> H <sub>6</sub> ClO) 125.0147 (C <sub>7</sub> H <sub>6</sub> Cl) 70.0390 (C <sub>2</sub> H <sub>4</sub> N <sub>3</sub> )	8.4	VL

553 **Table 4** Recoveries (REC), LOQs, RSD<sub>R</sub> and repeatabilities (n=6) in grapes, wine and vine  
554 leaves

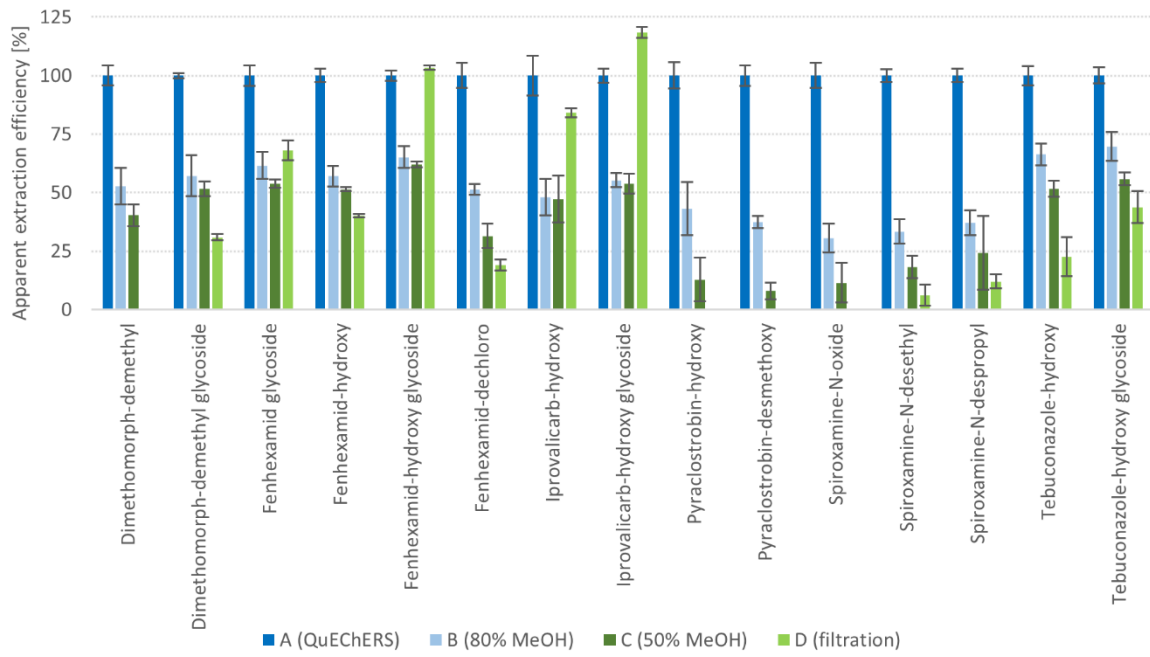
analyte	GRAPES						
	LOQ	0.002 mg kg <sup>-1</sup>			0.02 mg kg <sup>-1</sup>		
		REC	RSD	RSD <sub>R</sub>	REC	RSD	RSD <sub>R</sub>
[mg kg <sup>-1</sup> ]	[%]	[%]	[%]	[%]	[%]	[%]	
dimethomorph	0.001	93	2	7	92	1	6
fenhexamid	0.001	88	3	10	89	4	5
iprovalicarb	0.001	90	2	13	94	5	8
metrafenone	0.001	84	6	7	97	3	5
pyraclostrobin	0.001	90	2	5	89	1	7
quinoxifen	0.001	83	4	11	85	2	9
spiroxamine	0.001	94	1	12	90	2	9
tebuconazole	0.001	87	3	9	92	2	5
triadimenol	0.01	<LOQ	-	-	92	4	16
analyte	WINE						
	LOQ	0.002 mg kg <sup>-1</sup>			0.02 mg kg <sup>-1</sup>		
		REC	RSD	RSD <sub>R</sub>	REC	RSD	RSD <sub>R</sub>
[mg kg <sup>-1</sup> ]	[%]	[%]	[%]	[%]	[%]	[%]	
dimethomorph	0.001	96	2	8	91	4	5
fenhexamid	0.001	98	3	9	94	2	5
iprovalicarb	0.001	94	16	11	93	2	9
metrafenone	0.001	94	2	8	95	1	7
pyraclostrobin	0.001	89	2	6	88	1	6
quinoxifen	0.001	88	3	9	90	1	5
spiroxamine	0.001	91	1	10	84	9	7
tebuconazole	0.001	92	2	11	95	1	5
triadimenol	0.01	<LOQ	-	-	96	5	15
analyte	VINE LEAVES						
	LOQ	0.008 mg kg <sup>-1</sup>			0.08 mg kg <sup>-1</sup>		
		REC	RSD	RSD <sub>R</sub>	REC	RSD	RSD <sub>R</sub>
[mg kg <sup>-1</sup> ]	[%]	[%]	[%]	[%]	[%]	[%]	
dimethomorph	0.004	95	3	9	93	4	8
fenhexamid	0.008	89	3	11	88	2	10
iprovalicarb	0.004	90	11	17	90	2	15
metrafenone	0.004	88	2	9	93	2	9
pyraclostrobin	0.004	89	1	10	88	1	8
quinoxifen	0.004	81	4	8	82	3	10
spiroxamine	0.004	92	2	12	93	3	10
tebuconazole	0.004	91	2	7	88	2	6
triadimenol	0.04	<LOQ	-	-	83	7	18

555 **Table 5** Method repeatability (RSD, n=6) for analysis of metabolites in vine leaves and grapes

	vine leaves	grapes
metabolite of pesticide	RSD [%]	RSD [%]
dimethomorph-demethyl	13	4
fenhexamid glycoside	8	4
fenhexamid-hydroxy	12	3
fenhexamid-hydroxy glycoside	5	2
iprovalicarb-hydroxy	14	8
iprovalicarb-hydroxy glycoside	13	3
metrafenone-CL 1500836	6	-
metrafenone-CL 379395	6	-
metrafenone-CL 3000402	4	-
pyraclostrobin-desmethoxy	6	4
pyraclostrobin-hydroxy	3	6
spiroxamine-N-desethyl	14	3
spiroxamine-N-despropyl	10	3
spiroxamine-N-oxide	2	5
tebuconazole-hydroxy	9	4
tebuconazole-hydroxy glycoside	8	4

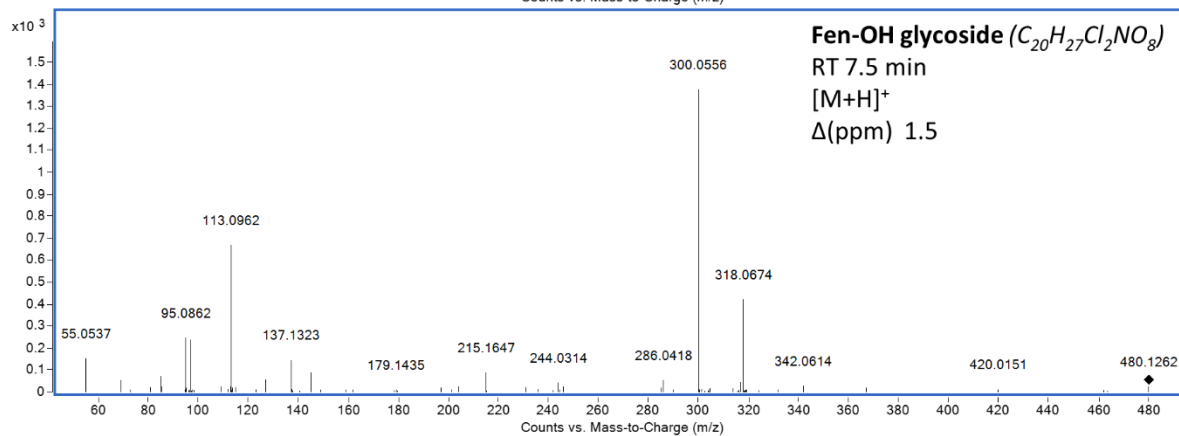
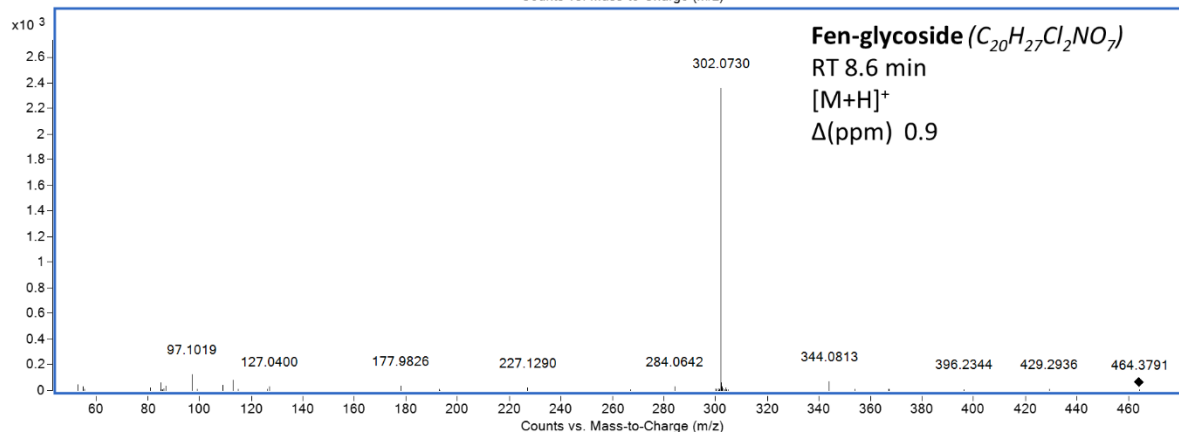
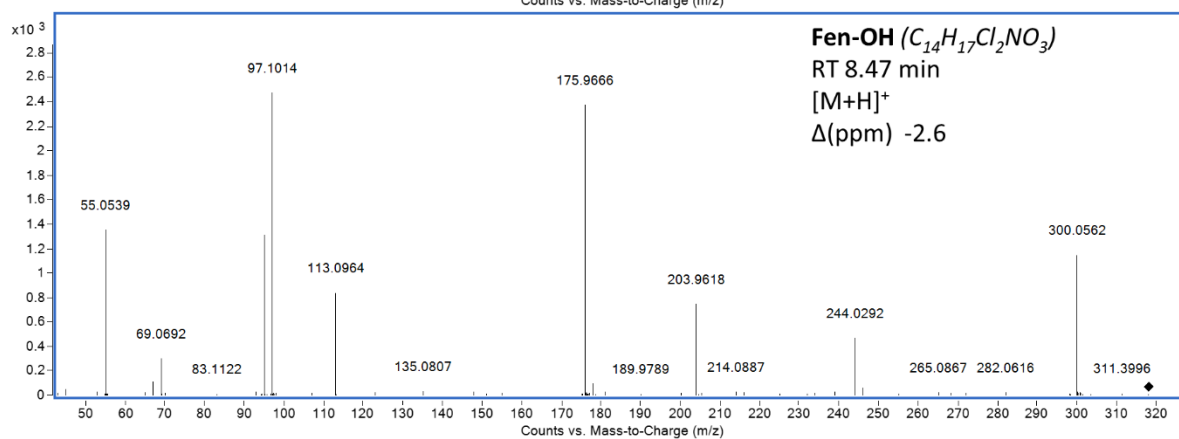
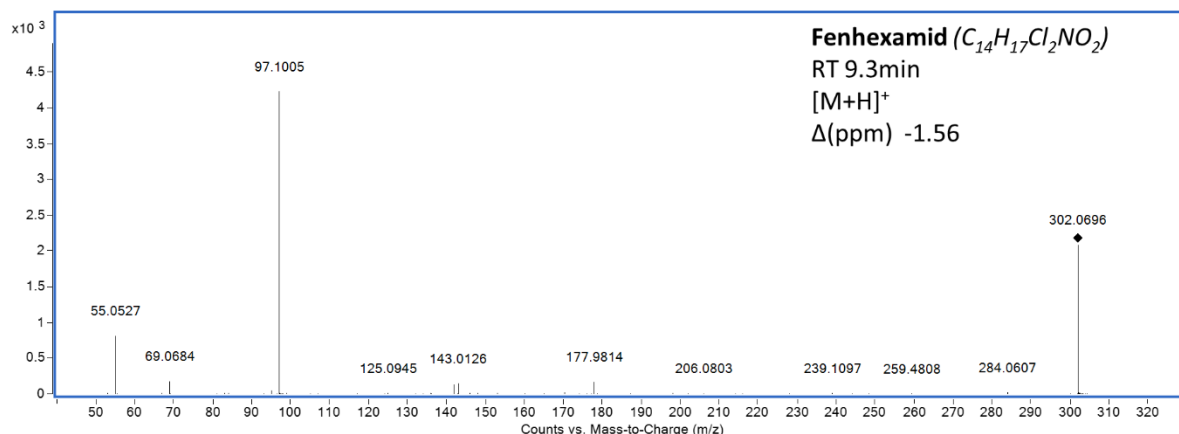
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557 **FIGURES & GRAPHICS**



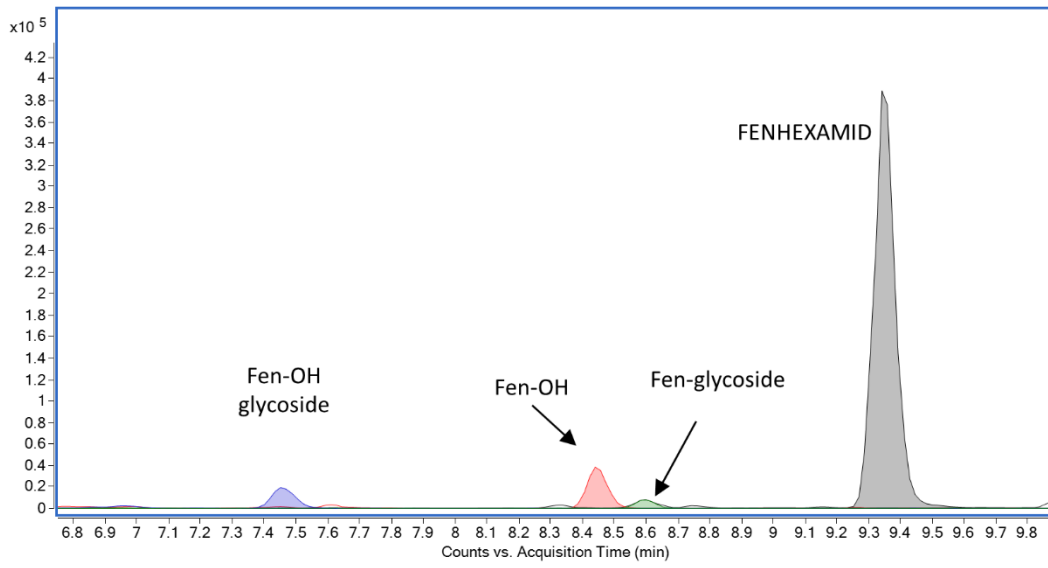
558  
559 **Figure 1**

560



561  
 562 **Figure 2**

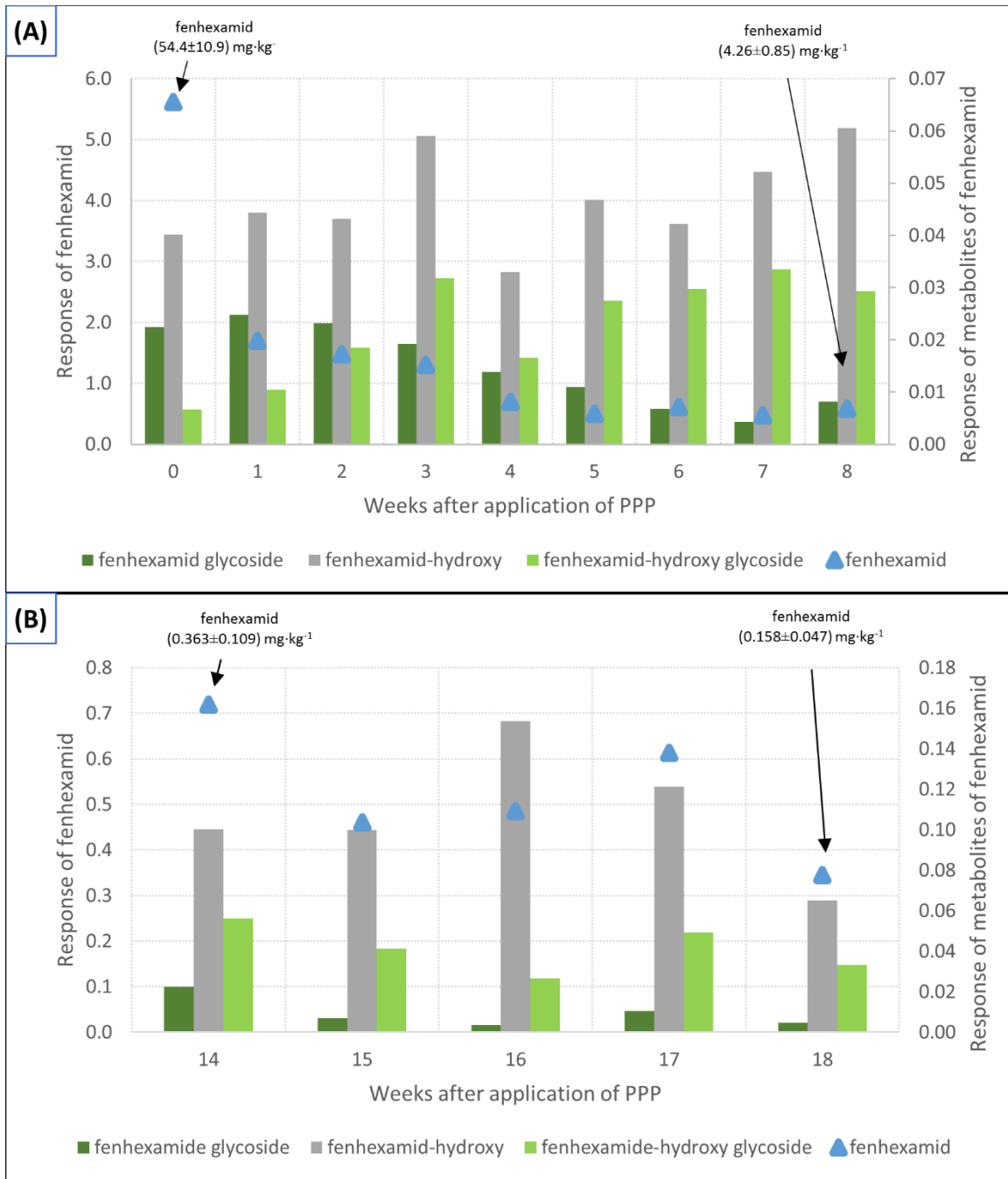




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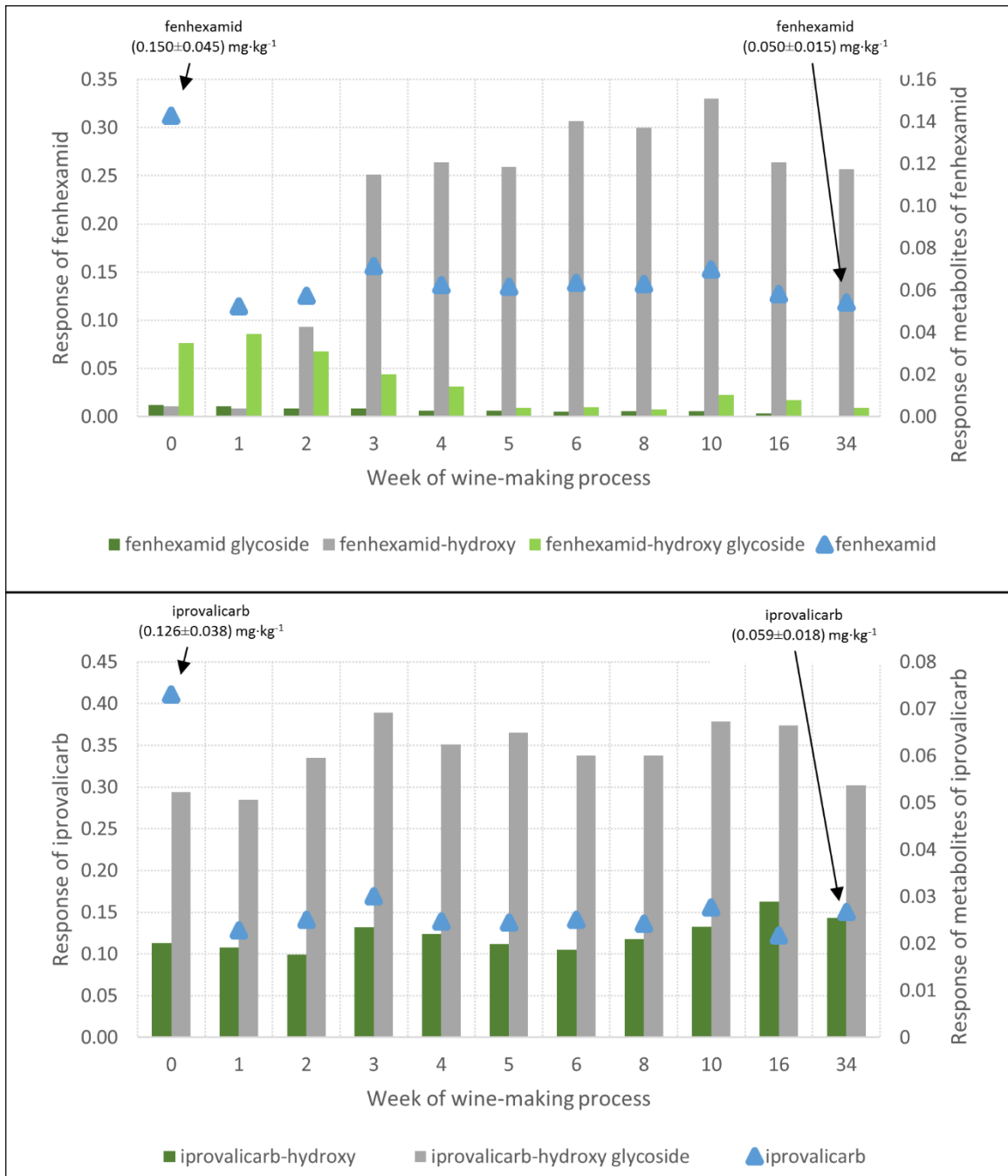
564 **Figure 3**

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566

567 **Figure 4**

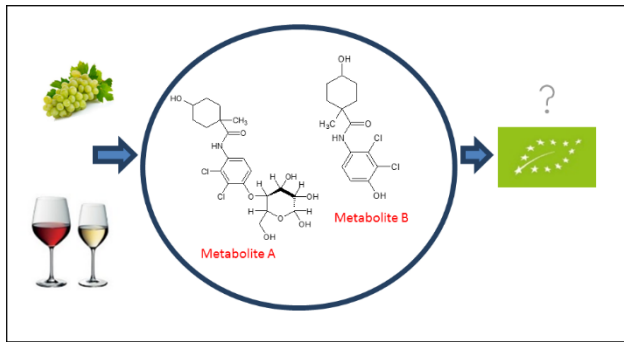


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569 **Figure 5**

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571 For Table of Contents only



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