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## **Problem 8- Candle Lighting trick**





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## Candle Lighting Trick

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## □ The problem

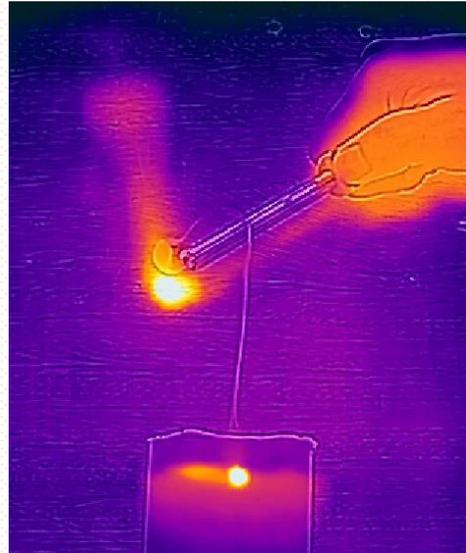
- “It is possible to relight a candle that has just been blown out by lighting the smoke that is created in the process. Indeed, the smoke contains vaporized wax which is the substance that burns in the flame in the first place. **What is the maximum distance** (between the match and the candle) from which one can **relight the candle? Identify the important parameters** and find **how they influence this maximal distance.**”

# □ Physical processes

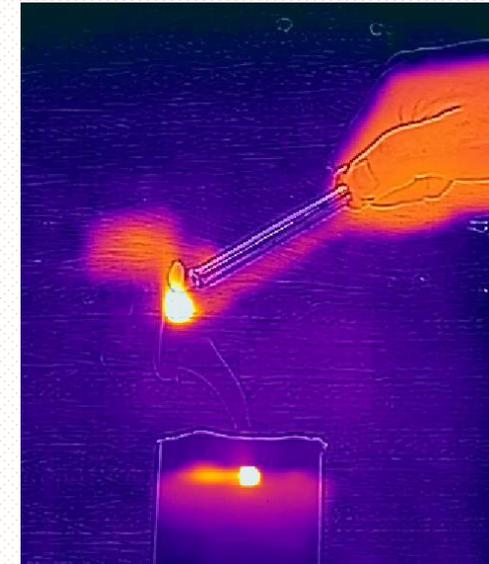
- Smoke **diffusion** in air



- Air **convection** due to temperature gradient



- External **interferences**



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# □ Chemical compounds of candle and smoke



*Candle*

- Candle: paraffin wax and a wick



*Smoke*

- Contains different compounds of paraffin
  - Dynamically important to the problem



*Ember*

350°C~430°C

- Luminous spots keep paraffin wax melt and evaporation
  - Helps in relighting

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# □ How a candle Works- Convection process

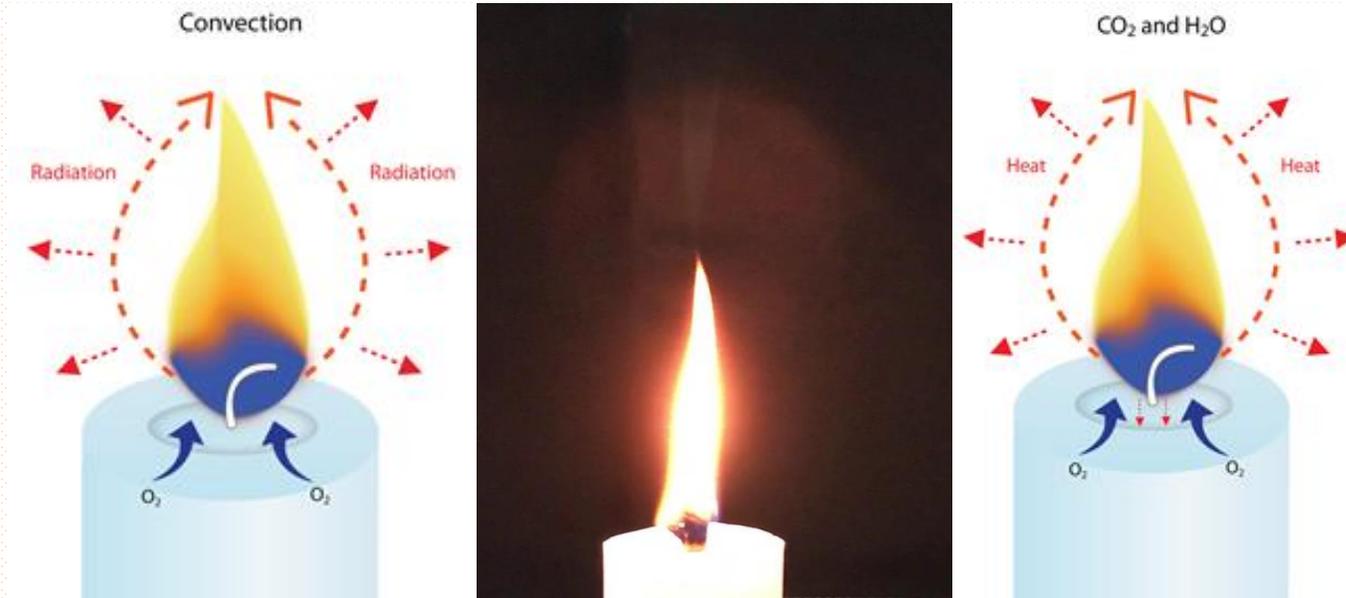
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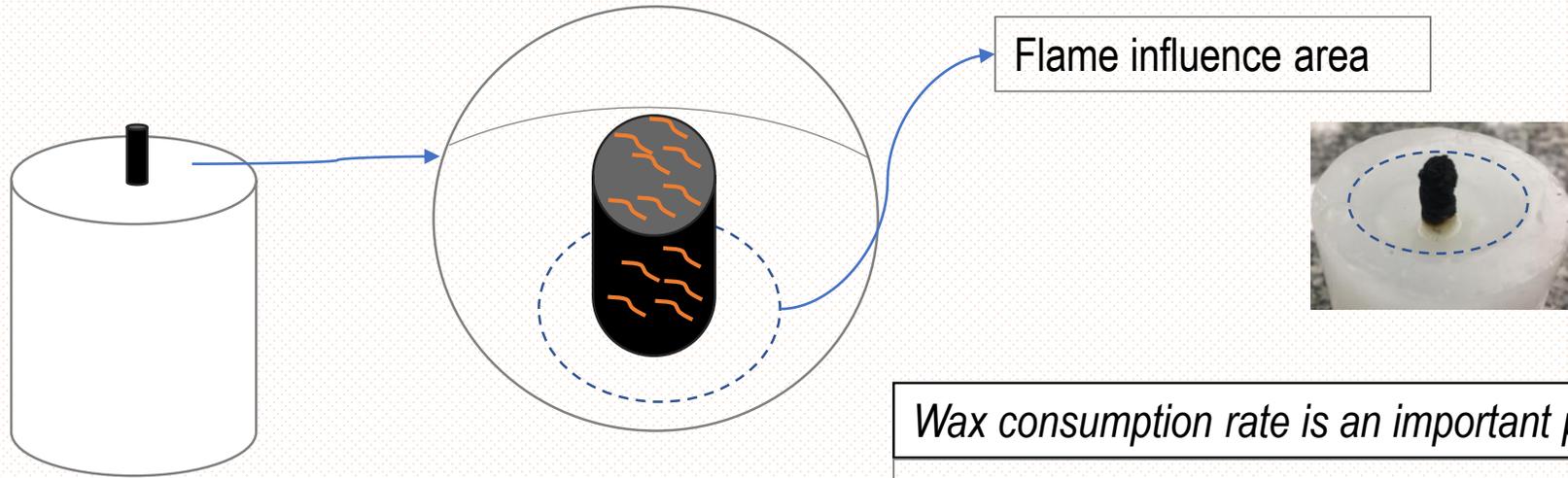
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- Flame heats the air and it starts to rise.
- Cooler air rushes at the bottom to replace it.
- When cooler air is heated, it too rises up and is replaced by cooler air at the base of the flame.
- Continual cycle of upward moving air around the flame (a convection current) giving the flame its elongated or teardrop shape.

# □ Candle characteristics



Flame influence area

*Wax consumption rate is an important parameter!*

*Related directly to candle diameter & wick*

*Heat of the flame melts the wax near the wick. This liquid wax is then drawn up the wick by capillary action. The heat of the flame vaporizes the liquid wax turns it into a hot gas. Molecules are drawn up into the flame and react with oxygen from the air to create heat, light, water vapor ( $H_2O$ ) and carbon dioxide ( $CO_2$ ).*

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# □ Conditions for relighting

- We need a limiar value of wax to relight the candle!
- It is related to time and height, since concentration is a function of time and space
  - A value above a threshold for concentration  $L$  is reached in an optimal instant  $t$
  - Therefore, there is a maximum height value  $z_{max}$



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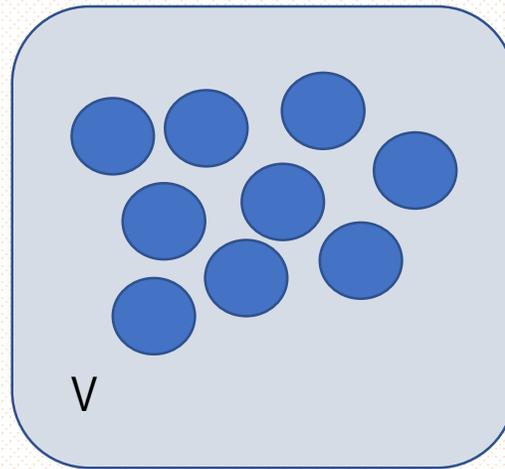
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# Physical modelling for smoke concentration

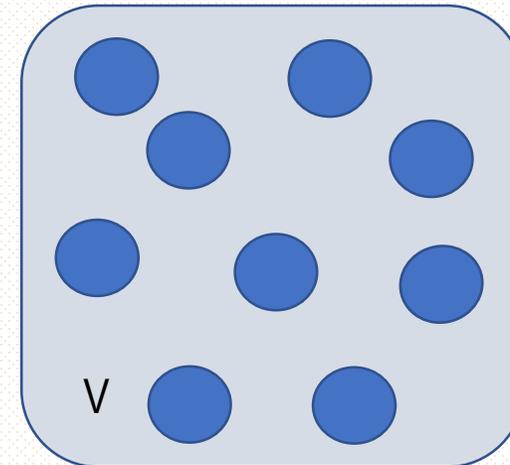
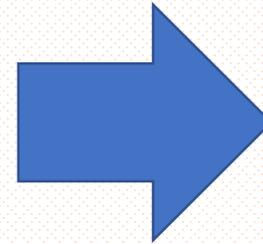
- Air density variation in function of temperature

$$\Delta\rho = -\rho_0\beta\Delta T$$

$\rho$  is the density of the fluid in convection;  $\Delta T$  is the temperature difference between two local points with their respective densities and  $\beta$  is the coefficient of thermal expansion, which for the air equals  $3.43 \times 10^{-3} \text{K}^{-1}$ .



Lower temperatures



Higher temperatures

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# □ Physical modelling for smoke concentration

- Convection process
  - Movement of air masses or particles considering different temperatures on a fluid
- The convection-diffusion process is described by

$$\frac{\partial \phi}{\partial t} = \nabla \cdot (D \nabla \phi) - \nabla (v \phi) + R$$

- Considering  $D > 0$  and  $R > 0$  constants, with  $v = v_z, v_z > 0$ :

$$\frac{\partial \phi}{\partial t} = D \nabla^2 \phi - v_z \frac{\partial \phi}{\partial z} + R$$

$D$  = mass diffusivity coefficient  
 $v$  = relative velocity field of the diffusing quantity  
 $R$  = describes any sources or sinks



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# □ Physical modelling for smoke concentration

- Applying separation of variables, with

$$\phi(x, y, z, t) = \chi(x)\Upsilon(y)\zeta(z)\tau(t)$$

- We reach a solution for the behavior of concentration for each Direction and time

$$\phi(x, y, z, t) = \phi_0 \exp\left(c_1 t - \sqrt{\frac{c_3}{D}}x - \sqrt{\frac{c_2 - c_3}{D}}y\right) \exp\left(-\frac{1}{2}\left(\frac{v_z}{D} - \sqrt{\left(\frac{v_z}{D}\right)^2 - \frac{4(c_2 - c_1)}{D}}\right)z\right)$$

- This solution will be used to **simulate** the phenomenon qualitatively

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# □ Characteristic solution

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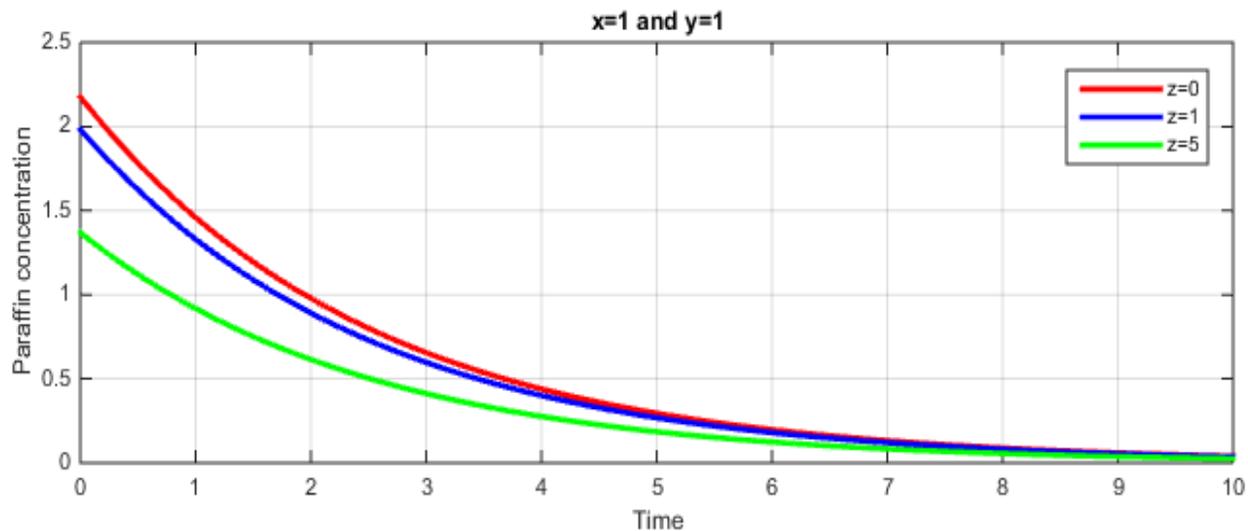
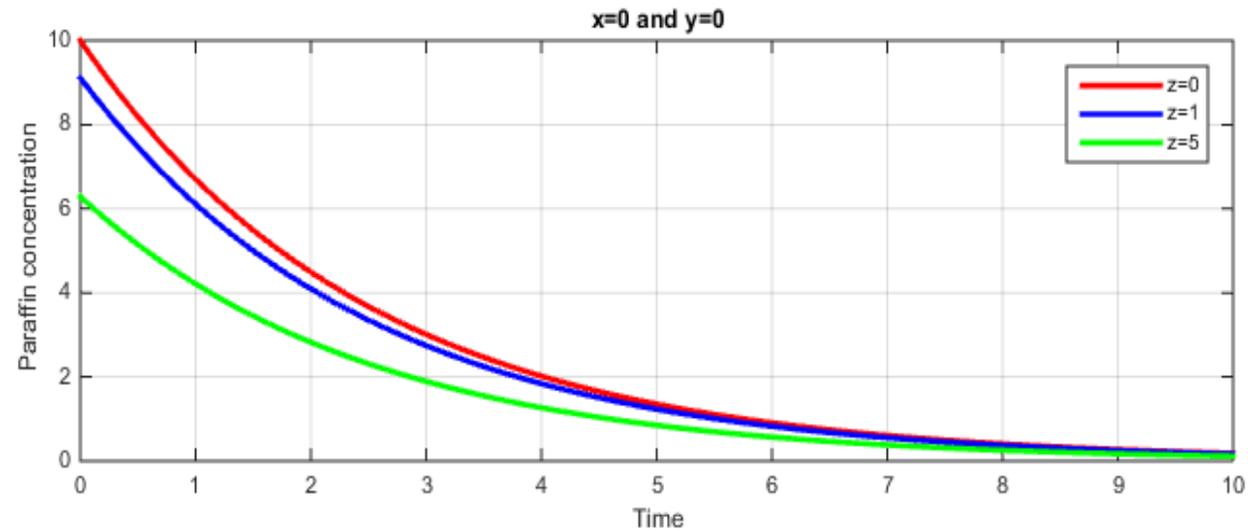
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- Matches qualitative expectations!
- Smoke concentration decreases with time and distance from the source in an exponential rate

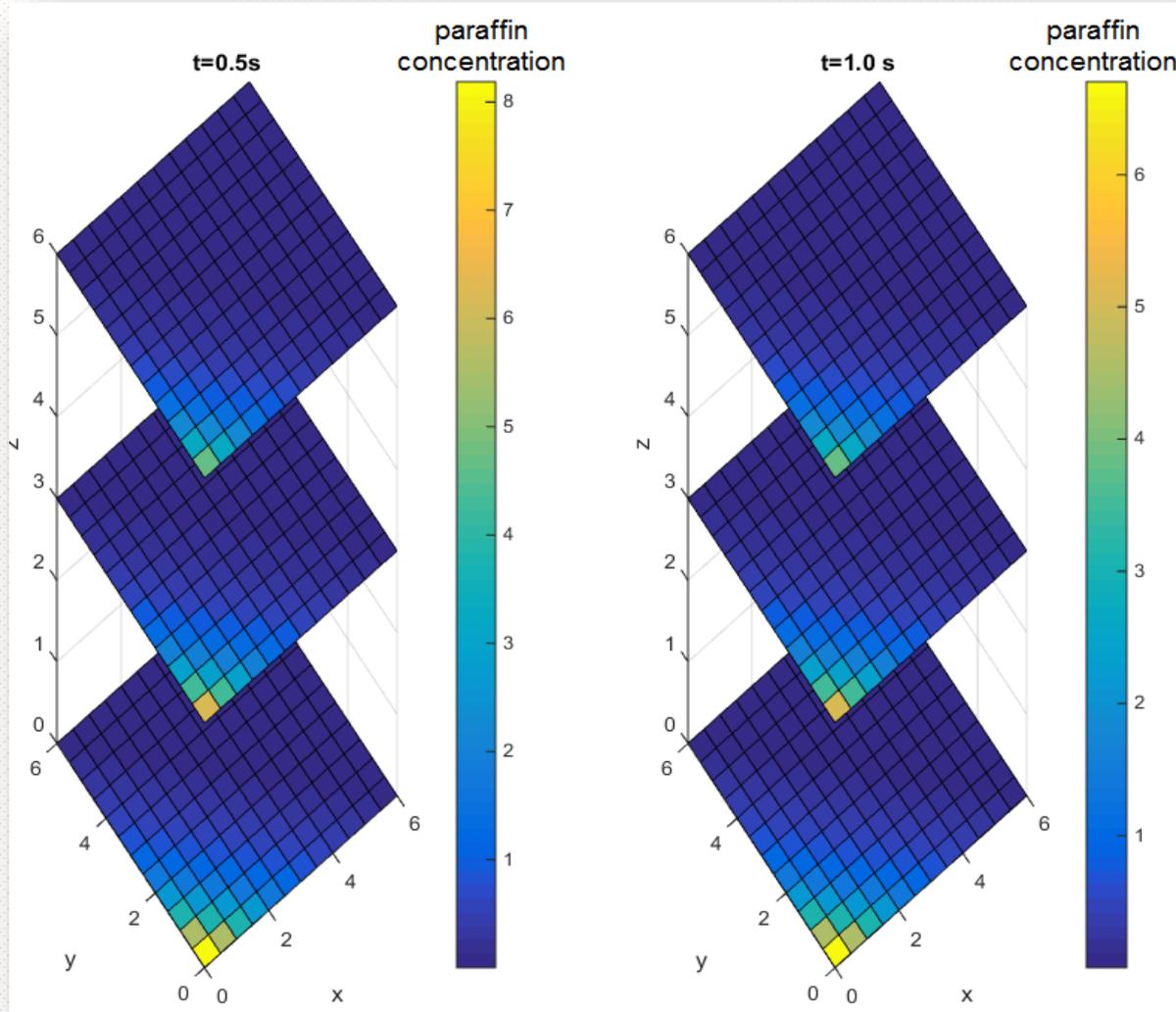


# Simulation for concentration in time

- Different cut planes for different heights and wax concentration in time

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- Diffusion more rapidly in upward direction!
- Behavior is the same for radial directions

Plot!

$$\phi(x, y, z, t) = \phi_0 \exp \left( c_1 t - \sqrt{\frac{c_3}{D}} x - \sqrt{\frac{c_2 - c_3}{D}} y \right) \exp \left( -\frac{1}{2} \left( \frac{v_z}{D} - \sqrt{\left( \frac{v_z}{D} \right)^2 - \frac{4(c_2 - c_1)}{D}} \right) z \right)$$

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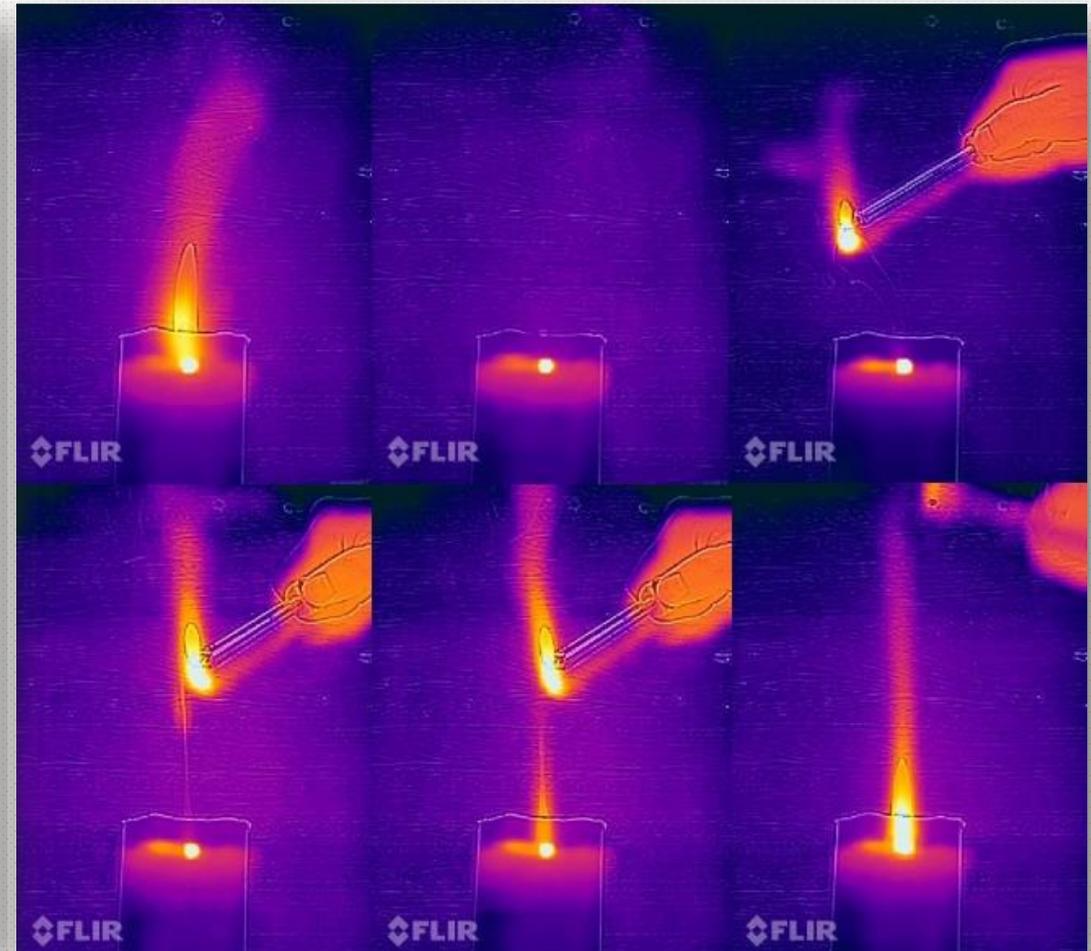
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# ☐ Methodology

Normal footage (SONY Handycam 60 fps)\*



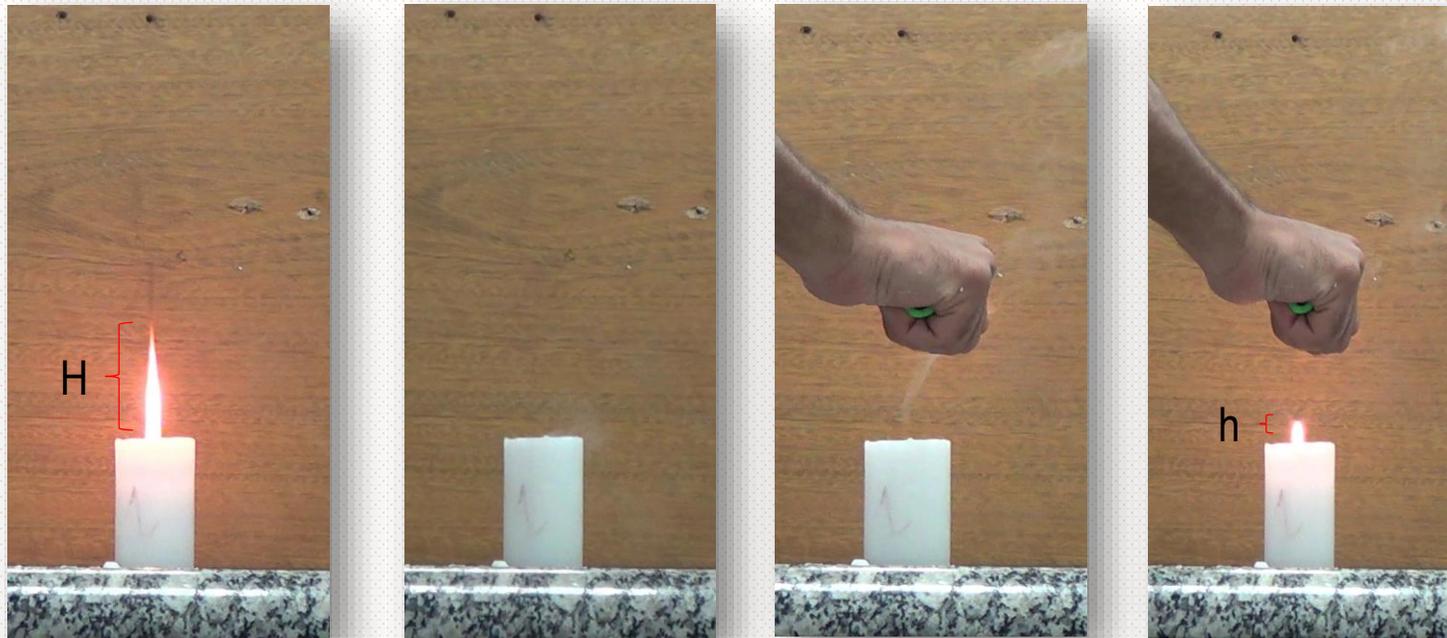
Thermal images used with FLIR sensor



\*Video analysis using the TRACKER software

# □ Methodology

- Obtained experimental data
  - Flame height before and after relighting
  - Time to relight the candle
  - Distance from the source
- All of these for different diameters & wick types



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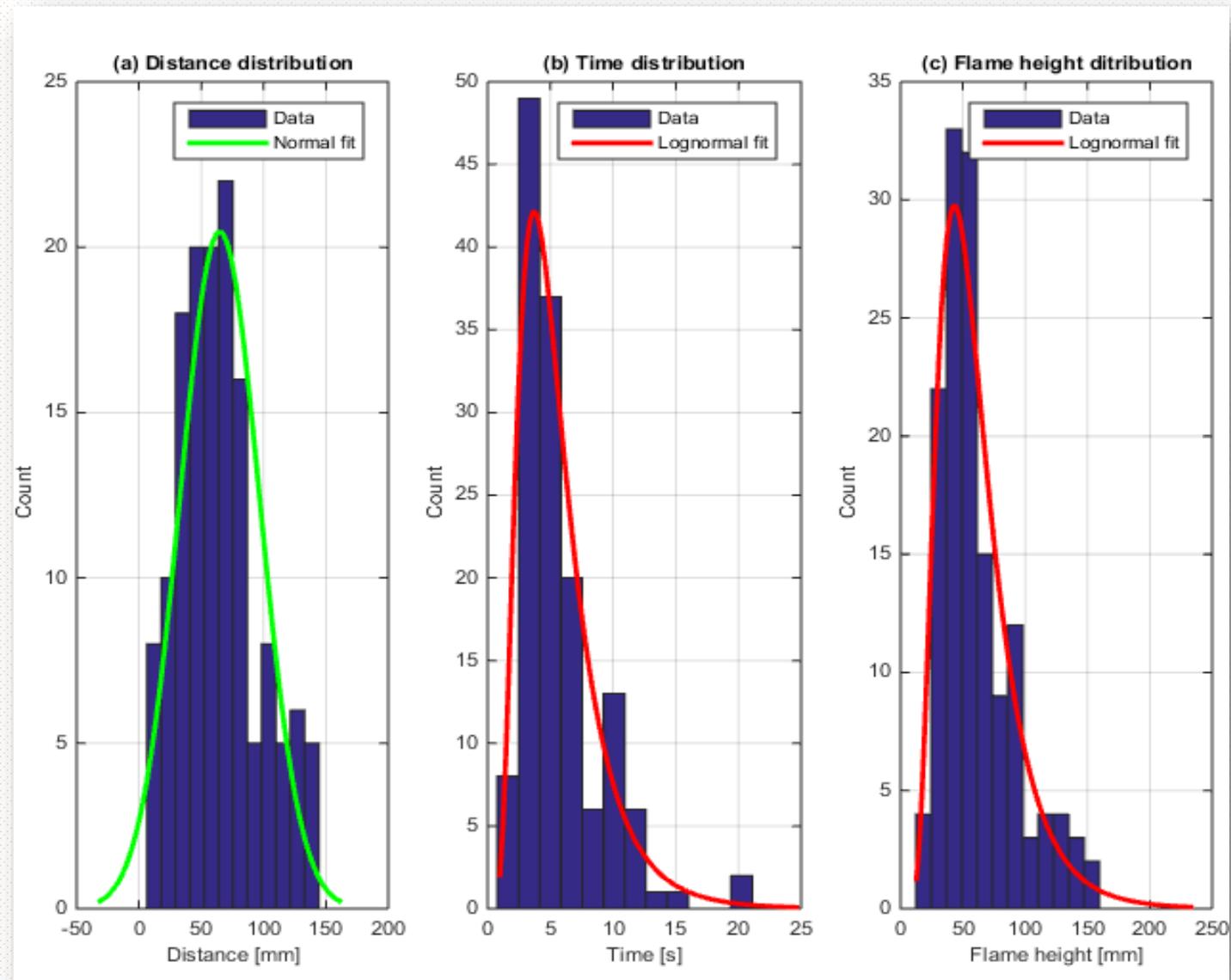
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# Physical considerations

- It is not possible to **perfectly isolate** the system from external interferences!
  - Experimental data will inherit fluctuations on a true value for maximum height ( $h$ )
  - This because we consider interactions between wax molecules and *external* (moving) air
  - Probabilistic approach to the problem



Distribution of all experimental results

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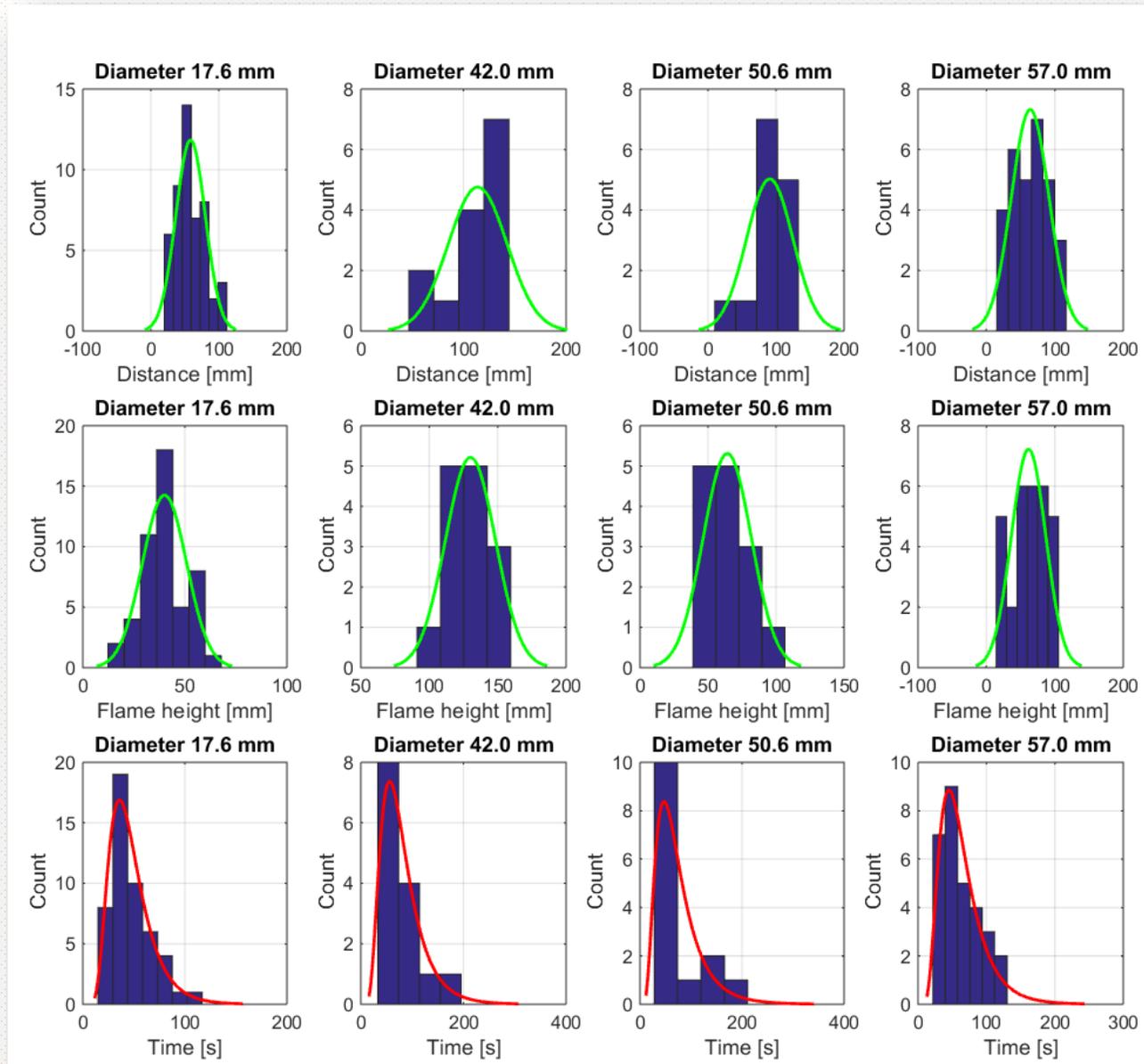
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# □ What can influence the maximum height?

- Candle diameter
  - $(17.5 \pm 0.5)$ mm
  - $(42.0 \pm 0.5)$ mm
  - $(50.06 \pm 0.5)$  mm
  - $(57.0 \pm 0.5)$  mm

Important approximation to rule out external interferences!



Distance

Flame height

Time for relighting

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Results for different candle diameters

# □ Candle diameter

- Diameter and average distance from the source

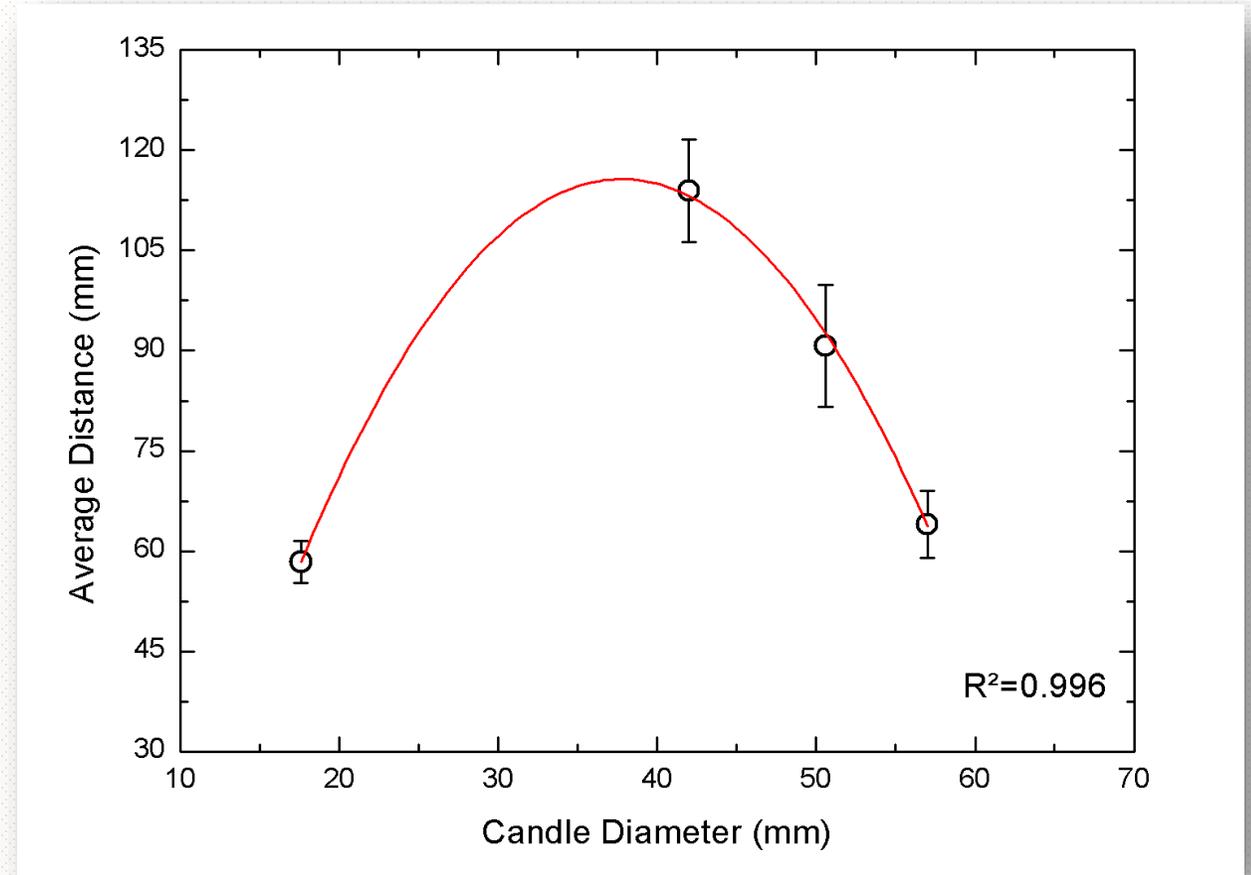
Parabolic fitting function

$$y = ax^2 + bx + c$$

$$a = -(0.14 \pm 0.02)cm$$

$$b = (11 \pm 2)cm$$

$$c = -(85 \pm 22)cm$$



Average behavior for distance in function of diameter

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# □ Candle diameter

- Diameter and average distance from the source

Parabolic fitting function

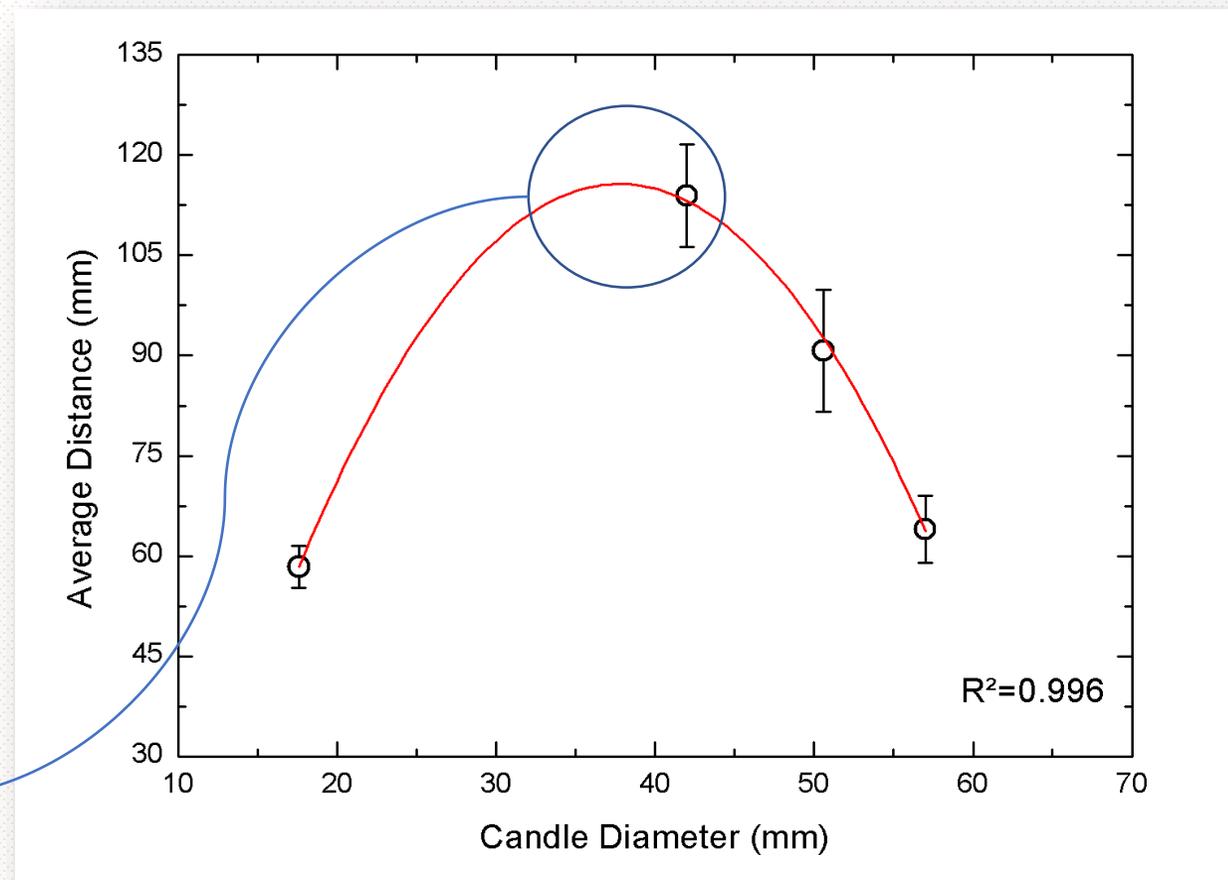
$$y = ax^2 + bx + c$$

$$a = -(0.14 \pm 0.02)cm$$

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$$c = -(85 \pm 22)cm$$

**Diameter as an optimization parameter**



Average behavior for distance in function of diameter

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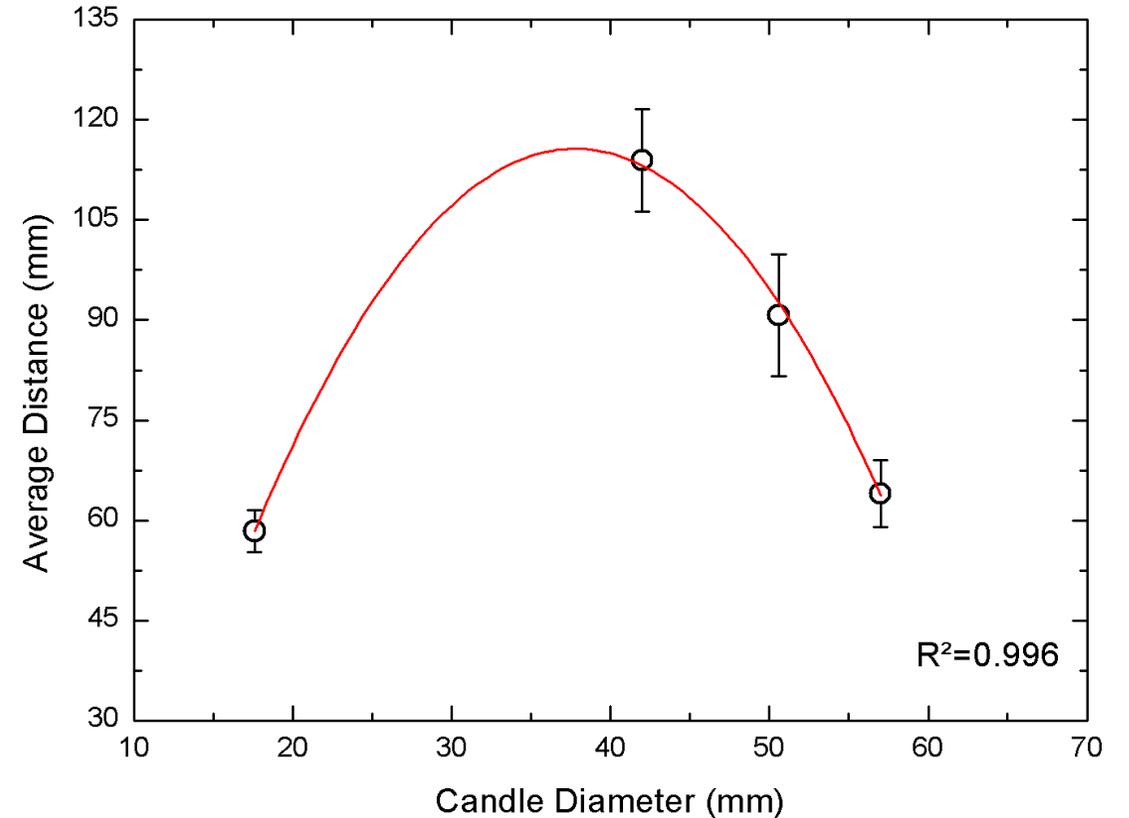
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# □ Candle diameter

- Diameter an average distance from the source

Optimal candle for greater distances  
*Diameter* =  $(38 \pm 8)\text{mm}$   
*Maximum height* =  $(116 \pm 24)\text{mm}$



Average behavior for distance in function of diameter

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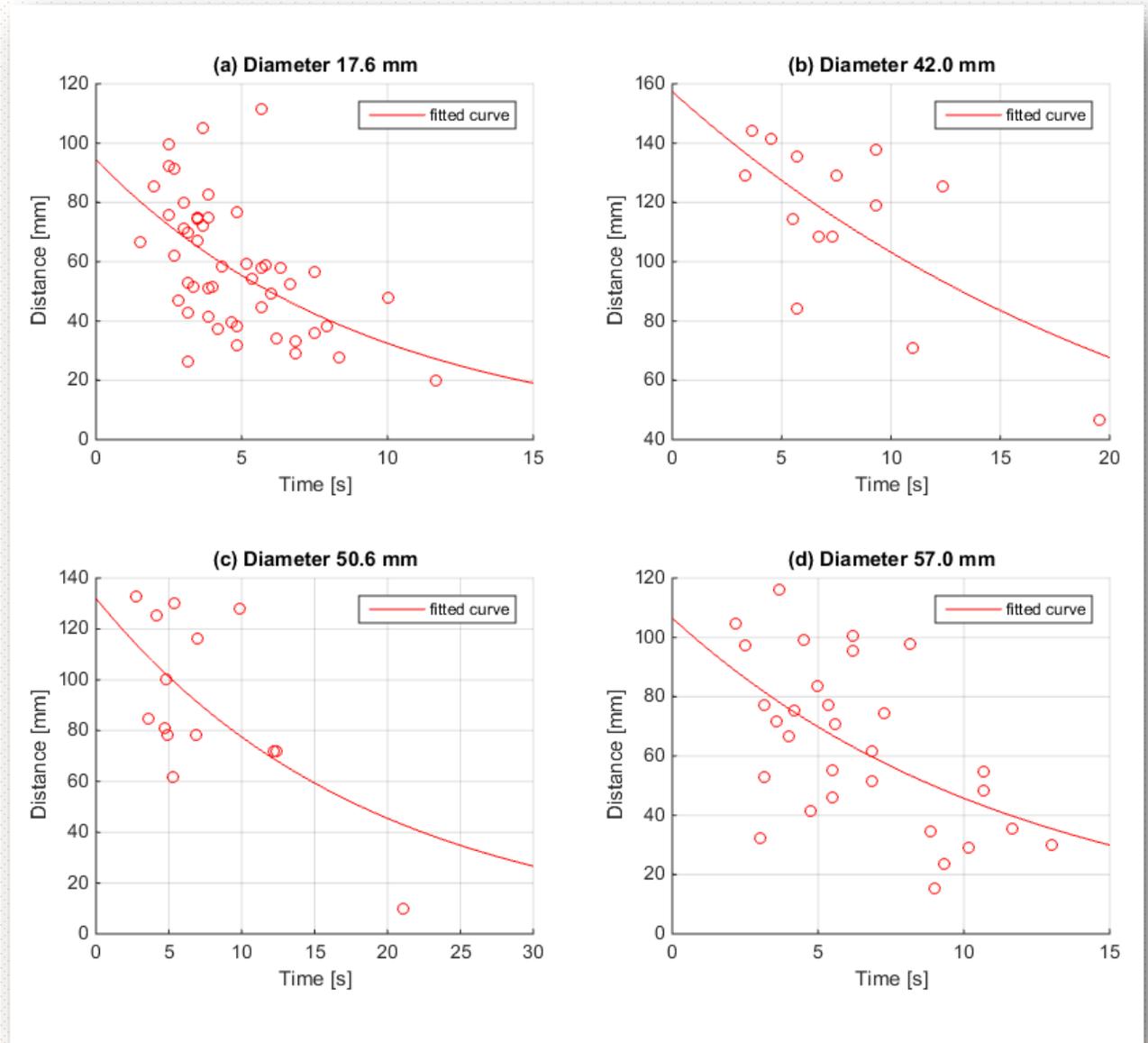
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# □ Candle diameter

- Relation between time and distance
  - Poor correlation coefficient
  - Unrelated!
- Also expected considering randomness of external factors



Dispersion for distance in function of time with different diameters

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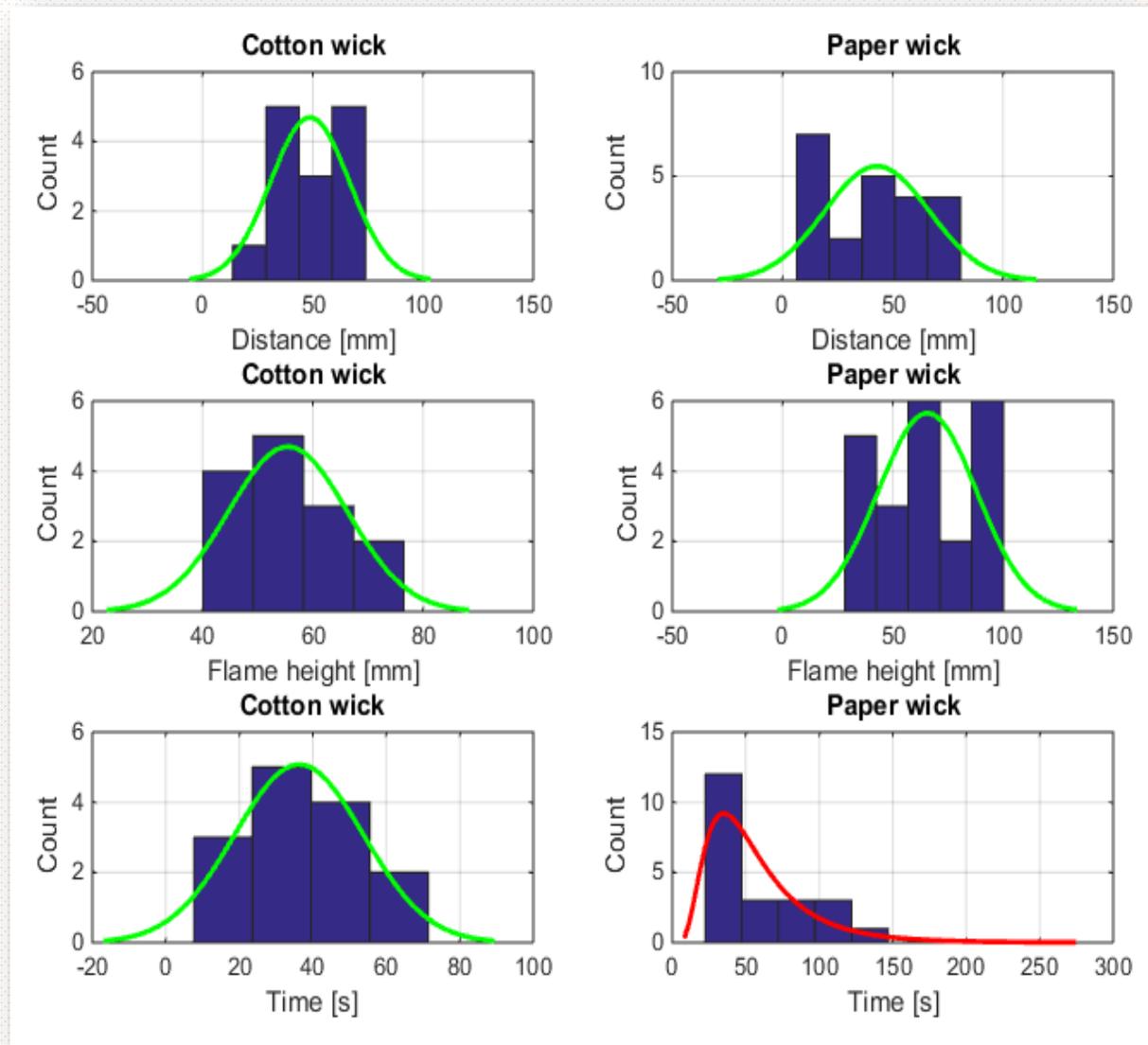
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# □ What can influence the maximum height?

- Type of wick
  - Cotton wick
  - Paper wick



*Distance*

*Flame height*

*Time for relighting*

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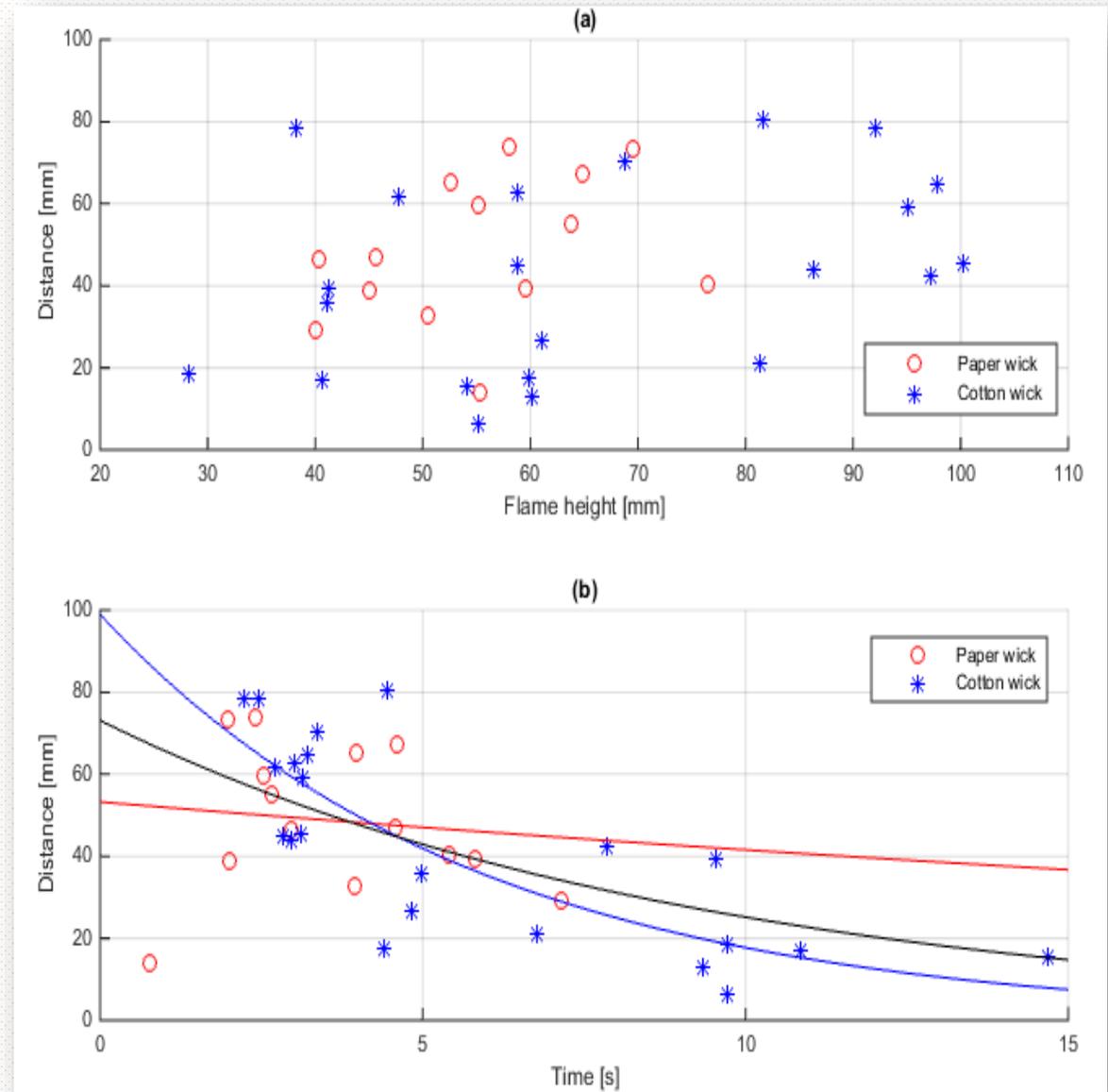
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Distribution for different wick materials (paper and cotton)

# □ Experimental considerations

- We analysed the points dispersion
  - No apparent correlation between flame height or time for distance in any of the different wicks

Dispersion for different wick materials (paper and cotton)



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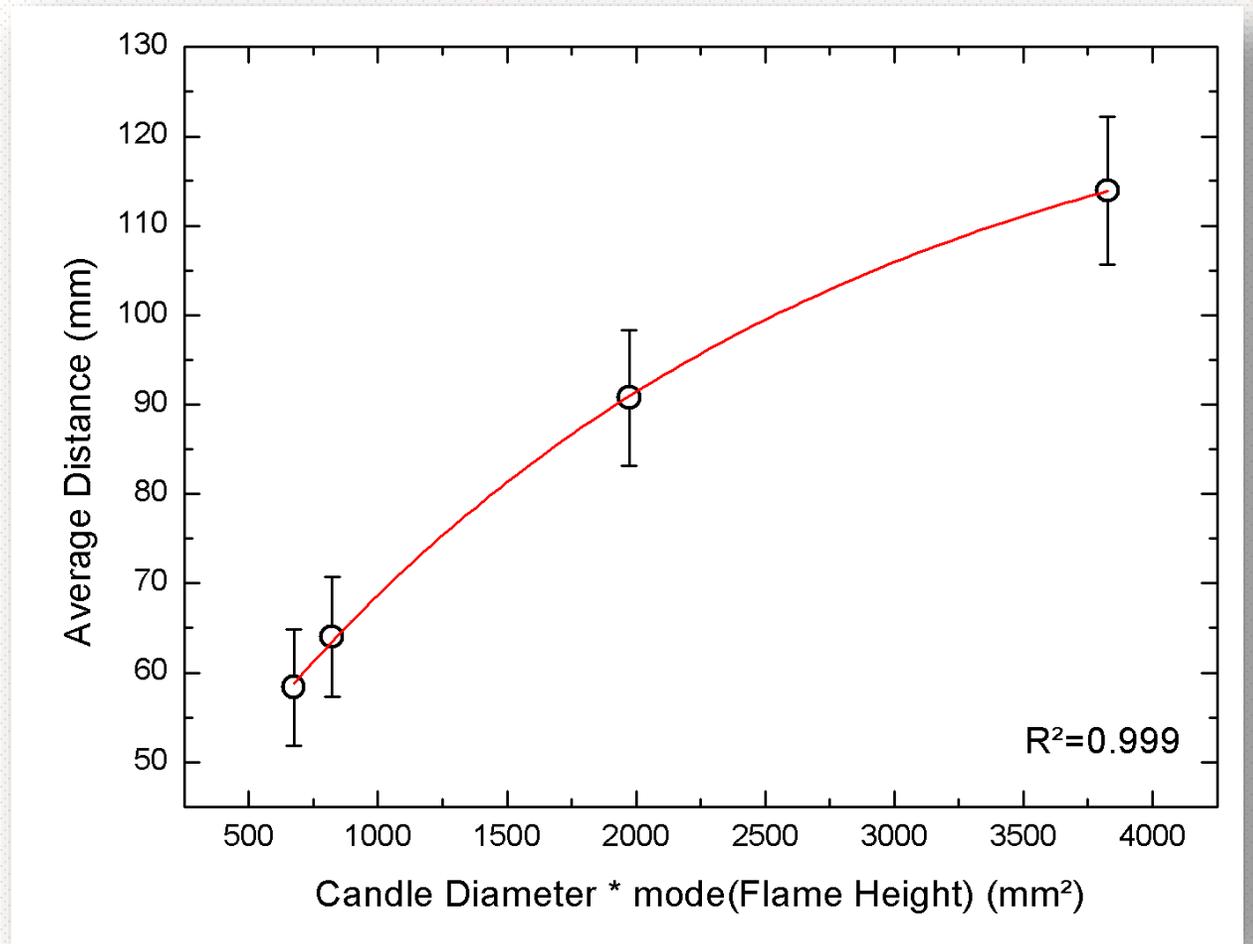
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# □ What can influence the maximum height?

- **Combining parameterss!**
  - Common burn rate produces more wax (size of flame before being put out)
  - Candle diameter related to contact area between flame and solid wax (more vapor wax consumption)
  - Maximum height is proportional to the product between both parameters
- Experimental validation for influence in combined parameters
- Average distance proportional to diameter and flame height (wax consumption)
- **Colocar o que é flame height!**



Relation between average distance and combined parameters

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# Summary

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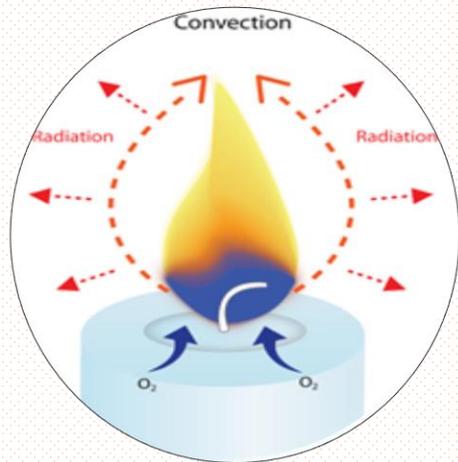
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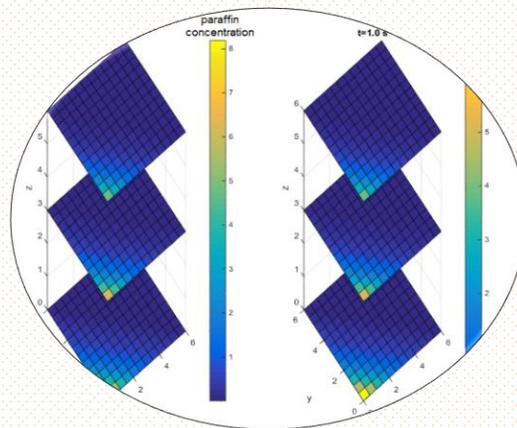
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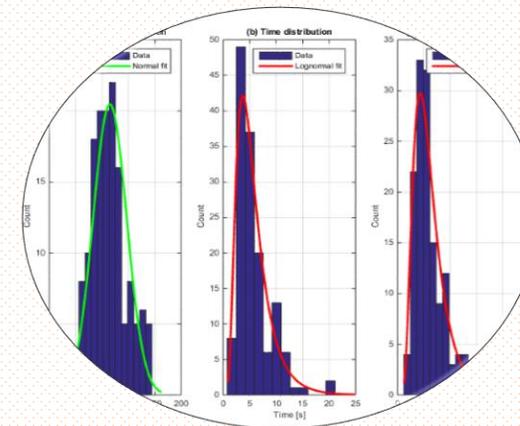
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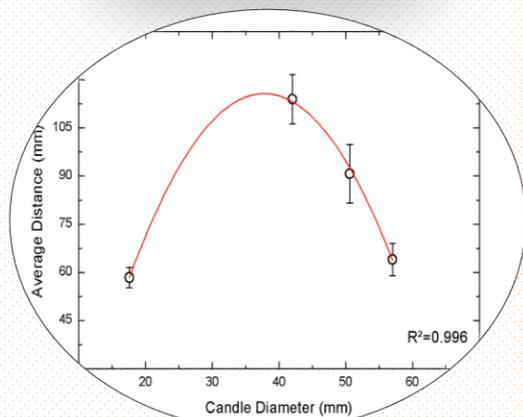
How does a candle work



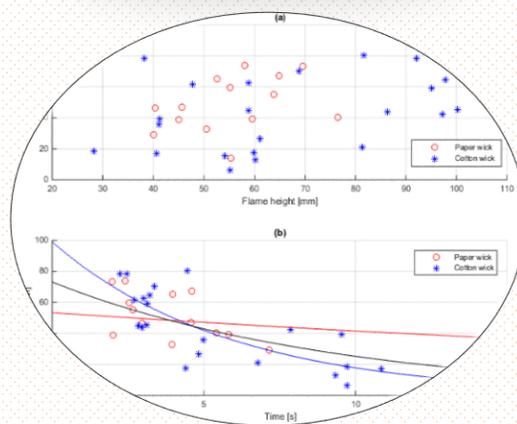
Simulation



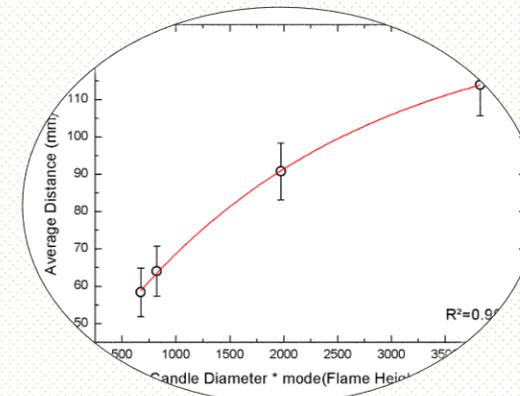
Statistical experimental approach



Optimal candle (diameter)



Different types of wick



Combined parameters



# ☐ References

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### **Candle Lighting Trick**

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# **□ Appendix**

# □ Other experimental setups

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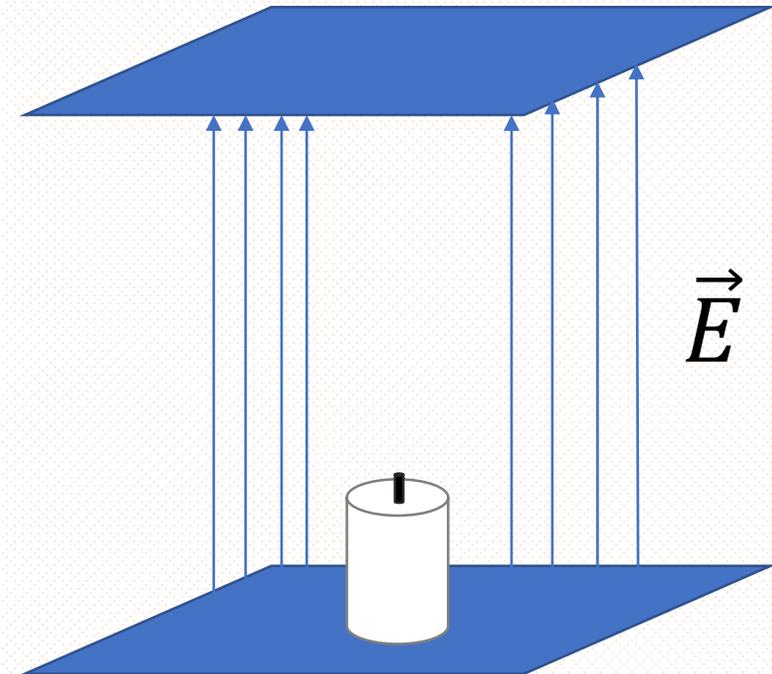
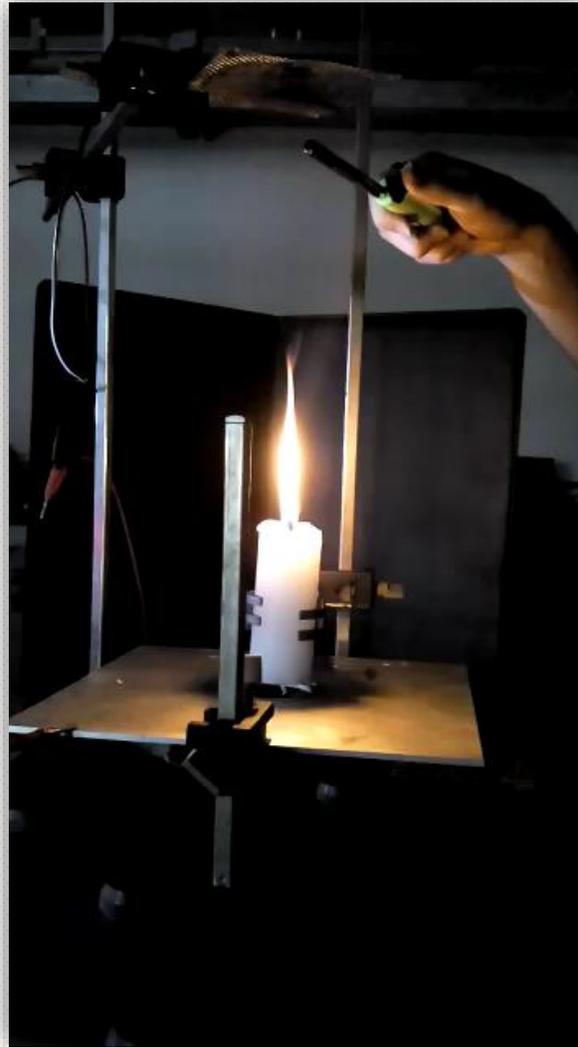
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# ☐ Candles and gravity

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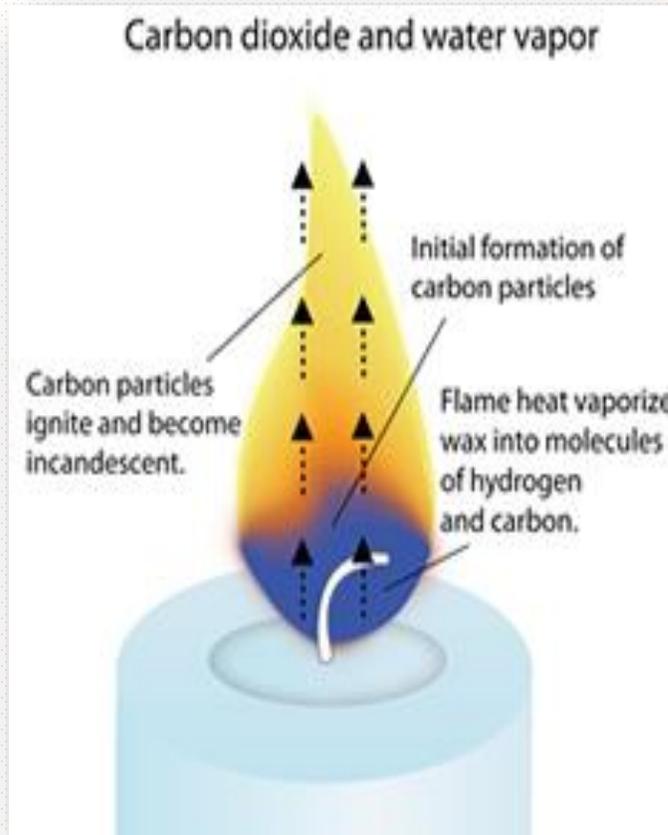


Candle at normal gravity



Candle at microgravity environment

# □ Evolution of the flame & properties



The oxygen-rich blue zone is where the hydrocarbon molecules vaporize and start to break apart into hydrogen and carbon atoms.

The dark or orange/brown region has relatively little oxygen. This is where the various forms of carbon continue to break down and small, hardened carbon particles start to form. As they rise, along with the water vapor and carbon dioxide created in the blue zone, they are heated to approximately 1000 degrees Centigrade.

At the bottom of the yellow zone, the formation of the carbon (soot) particles increases. As they rise, they continue to heat until they ignite to incandescence and emit the full spectrum of visible light. Because the yellow portion of the spectrum is the most dominant when the carbon ignites, the human eye perceives the flame as yellowish. When the soot particles oxidate near the top of the flame's yellow region, the temperature is approximately 1200° C.

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