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REVIEW ON CELLULOSE DERIVATIVES

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ABSTRACT

Cellulose derivatives are most commonly used to modify the release of drugs in tablet and capsule formulations for the purpose of tablet binding, thickening, film formation, water retention, adhesion, and as suspending and emulsifying agents. Annual cellulose synthesis by plants is close to 10^{12} tons. Plants contain approximately 33% cellulose whereas wood contains around 50 % and cotton contains 90%. This fact was the starting point of our research into understanding, designing, synthesising and finding new alternative applications for this well-known but underused biomaterial.

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INTRODUCTION

Cellulose Introduction –

Cellulose is a complex carbohydrate, or polysaccharide consisting of 3000 or more glucose units.

Cellulose + H_3O^+ + heat \rightarrow over 1000 glucose molecules.

Cellulose is a substance found in the cell walls of plants. Although cellulose is not a component of the human body, it is nevertheless the most abundant organic macromolecule on Earth. The scientific community first observed cellulose in 1833 when it was studied in plant cell walls. The chemical structure of cellulose resembles that of starch, but unlike starch, cellulose is extremely rigid. This rigidity imparts great strength to the plant body and protection to the interiors of plant cells.

The most abundant organic compounds on earth.

The basic structure component of plants cell walls

33% Vegetable

90% Cotton

50% Wood

WHAT ARE CELLULOSE DERIVATIVES

- ✕ Pharmaceutically used cellulose derivatives obtained by either mechanical or chemical processing or both.
- ✕ The hydroxyl groups of cellulose can be partially or fully reacted with various reagent to afford derivatives with useful properties.
- ✕ Pure cellulose additional treatment by HCL produced various cellulose derivatives.

Classification of Cellulose

- ✕ Based on solubility:-

Water soluble:-

hydroxypropylmethylcellulose

hydroxyethylcellulose

hydroxypropylcellulose

Water insoluble:-

ethyl cellulose

cellulose acetate phthalate

HPMC phthalate

- ✕ Based on chemical nature:-

cellulose ester:-

cellulose acetate

cellulose triacetate

cellulose propionate

cellulose nitrate (nitrocellulose)

cellulose ether:-

methyl cellulose

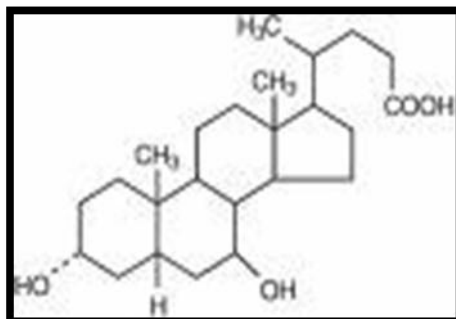
ethyl cellulose

HPC, HPMC

carboxy methyl cellulose

CELLULOSE DERIVATIVES

Microcrystalline cellulose



Synonyms: Cellex, cellulose gel, fibrocel.

Empirical Formula : $(C_6H_{10}O_5)_n$ $n=220$

Molecular Weight : 36000

Function Category : Adsorbent, Suspending agent, table diluent

Application in Pharmaceutical Formulation:

→ Binder (20-90%) → Disintegrant (5-15%)

→ Antiadherent (5-20%) → Lubricant

→ Diluent (20-90%)

Description: White, Odorless, Tasteless, Crystalline power

Typical properties

Density(bulk): 0.337 g/cm^3

Density(tapped): 0.478 g/cm^3

Density(true): $1.512-1.668 \text{ g/cm}^3$

Flow ability: 1.41 g/s

Melting point: $260-270^\circ\text{C}$

Moisture content: < then 5 % w/w

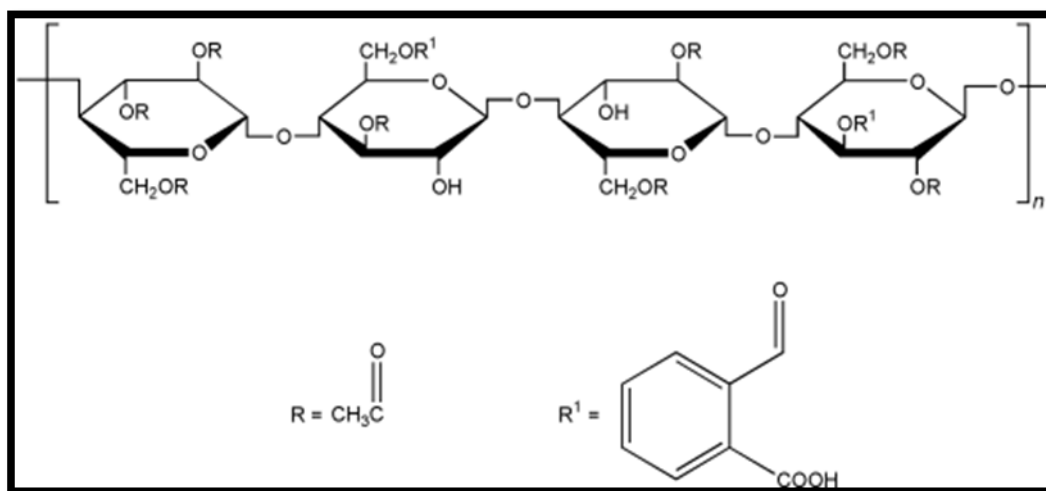
Solubility: Slightly soluble in NaOH

Precaution:

Irritant to eye so recommended eye protection.

Incompatibilities : With strong oxidizing agents.

Cellulose Acetate phthalate



Synonyms: cellacephate

Function Category : coating agent

Application in Pharmaceutical Formulation:

→ used in enteric film coating material (0.5-9%)

→ matrix binder for tablets and capsules

Description:

Hygroscopic, White to off White, Free-flowing powder

Typical properties

Density(bulk): 0.260 g/cm^3

Density(tapped): 0.266 g/cm^3

Melting point: 192°C

Moisture content: 2.2%

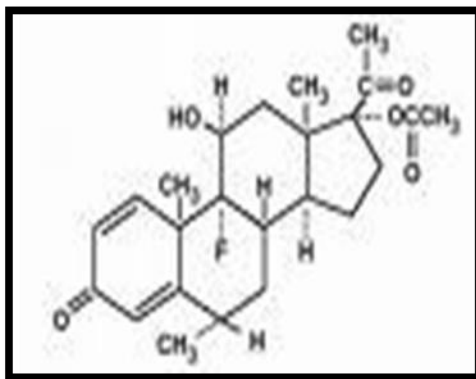
Solubility: <10% w/w

Incompatibilities : With ferrous sulfate, FeCl₂, CaCl₂, HgCl₂, lead acetate, strong oxidizing agent.

Precaution:

Irritant to eye should be handled in a well ventilated environment.

Cellulose Acetate



Synonyms: Acetyl cellulose

Empirical Formula: $(C_6H_{12}O_7)_n$

Molecular Weight: 38000

Function Category : Extended release agent, Diluent

Application in Pharmaceutical Formulation:

→used in sustain release & taste masking

→used in transdermal drug delivery

Description:

Free flowing pellets, tasteless, slightly odour of acetic acid

Typical properties

Density: 1.3 g/cm³

Melting point: 230-300⁰c

Glass transition temperature: 170-190⁰c

Solubility: soluble in acetone-water, dichloromethane-ethanol

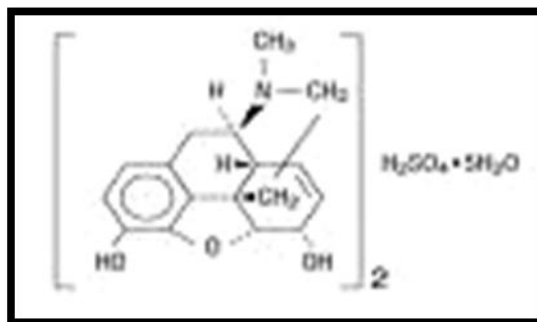
Viscosity: 10-230 mPa s

Incompatibilities : With strongly acidic or alkaline substance

Precaution:

Cellulose acetate irritant to eye so eye protection should be worn

Hydroxy ethyl cellulose



Synonyms: cellulose hydroxy ethyl ether

Molecular Weight: 38000

Function Category : Suspending agent, Binder

Coating agent, Thickening agent

Application in Pharmaceutical Formulation:

→ Ophthalmic & Topical formulation

→ Cosmetic preparation

Description:

Hygroscopy powder, odourless, tasteless, cream to white colour

Typical properties

Density(bulk): 0.35-0.61 g/cm³

Melting point: 135-140°C (softens)

205°C (Decompose)

Moisture content: < 5 % w/w

Solubility: soluble in Hot & Cold water

Insoluble in ethanol, ether, toluene

Viscosity: 2-20000 mPa s

cello size (2-3000)

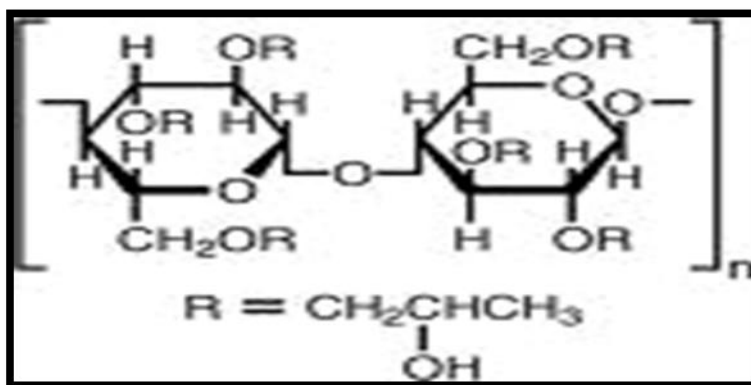
Natrosol (>3000)

Incompatibilities : With fluorescent dyes , quaternary disinfectant.

Precaution:

Irritant to eye so recommended eye protection.

Hydroxy propyl cellulose



Synonyms: Hyprolase

Molecular Weight: 50000-125000

Function Category : Suspending agent, Binder, Coating agent, Emulsifying agent

Application in Pharmaceutical Formulation:

→ Oral & Topical formulation

→ binder in tableting process (2-6% w/w)

→ extended drug release (15-35% w/w)

Description:

odourless, tasteless, white to slightly yellow colour.

Typical properties

Density(bulk): 0.5 g/cm³

Melting point: 130°C (softens)

260-275°C (Decompose)

Moisture content: 4 % w/w (at 50% relative humidity)

12% w/w (at 84% relative humidity)

Solubility: soluble 1 in 10 part dichloromethane

1 in 2.5 part ethanol

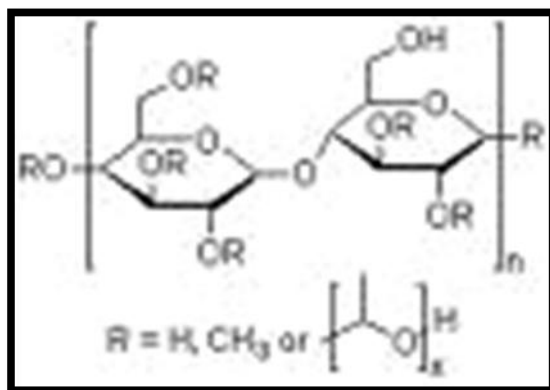
1 in 2 part methanol

Viscosity: 75-3000 mPa s

Incompatibilities : With phenol derivatives such as methylparaben , propylparaben.

Precaution:

Irritant to eye so recommended eye protection.

Hydroxy propyl methyl cellulose

Synonyms: Hypromellose

Empirical Formula : $C_{56}H_{108}O_{30}$

Molecular Weight : 10000-1500000

Function Category : Coating agent, Suspending, agent, tablet binder

Application in Pharmaceutical Formulation:

→Binder (2-5% w/w)

→Thickening agent(0.45-1% w/w)

→Emulsifier, Suspending agent

→Plastic bandage

Description: White, Odorless, Tasteless

Typical properties

Density(bulk):0.341 g/cm³

Density(tapped):0.557g/cm³

Density(true):1.326g/cm³

Melting point: 190-200⁰c

Moisture content: depend on initial moistur content, temperature, relative humidity.

Solubility: soluble in cold water And insolubal in chloroform, ethenol, ethers.

Incompatibilities : With oxidizing agents , metallic, salts or ionic organics.

Precaution: Irritant to eye so recommended eye protection.

Powder cellulose

Synonyms: Arbocel

Empirical Formula: $(C_6H_{10}O_5)_n$ n=500

Molecular Weight : 243000

Function Category : Suspending agent, tablet diluents, disintegrant

Application in Pharmaceutical Formulation:

→Binder (5-25% w/w)

→Disintegrant (5-15-1% w/w)

→Glident(1-2%)

→Plastic bandag

Description: White, Odorless, Tasteless

Typical properties

Density(bulk):0.139-0.391 g/cm³

Density(tapped):0.210-0.481 g/cm³

Density(true):1.5 g/cm³

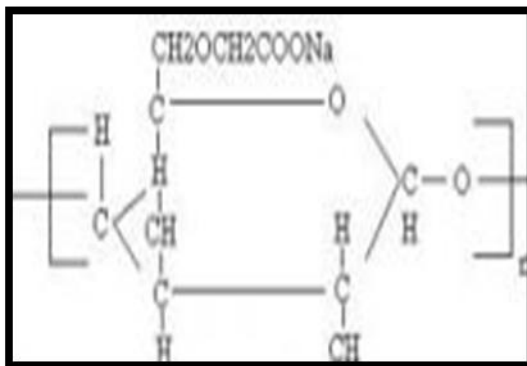
Moisture content: < then 5 % w/w

Solubility: Slightly soluble 5% in NAOH

Incompatibilities : With strong oxidizing agent.

Precaution: Irritant to eye so recommended eye protection.

Carboxyl methyl cellulose



Synonyms: carmellose

Chemical formula: $C_8H_{16}NaO_8$

Molecular formula: 90000-700000

Properties:

- Water soluble
- Stable at pH 5-10, best at 7-9 pH
- At pH 3 CMC is insoluble
- Fine Film Forming Properties
- Resistance to oil Greases & Solvents

Methyl cellulose

- ✖ Synonyms:- Methocel, methyl ether
- ✖ Contain 27.5 to 31.5% of methoxy groups.
- ✖ Description :- White, powder or granules.
- ✖ Solubility:- insoluble in ether, alcohol and chloroform but soluble in glacial acetic acid and in mix. of equal parts of alcohol and chloroform.
- ✖ Uses:- dispersing thickening emulsifying and coating agent.

Ethyl cellulose

- ✖ Synonyms:- Aquacoat, ethocel, surelease.
- ✖ Description :- tasteless, free flowing.
- ✖ Solubility:- insoluble in glycerin, propylene glycol, and water. soluble in chloroform, ethanol.
- ✖ Uses:- Micro encapsulation (10-20%)
Sustained release tablet coating (3-20%)
Tablet coating (1-3%)

Advantages

- ❖ Improve stability of drug.
- ❖ Good Lubrication
- ❖ Good binding properties
- ❖ Rapid disintegration
- ❖ Good flowing properties
- ❖ Reduced friability & weight loss
- ❖ Excellent compression & hardness

Uses of Cellulose

Cellulose has many uses such as the following.

- Anticake agents
- Emulsifier
- Stabiliser
- Dispersing agent
- Thickener
- Gelling agent
- ❖ Most important use is of holding on to water

CONCLUSION

Cellulose and Cellulose derivatives are carbohydrate polymeric system is applicable for pharmaceutical as well as industrial purpose.

REFERENCES

- 1) Raymond C. Rowe, Handbook of pharmaceutical excipients, published by Royal pharmaceutical society of Great Britain, London, 4th edition page no. 108-122, 283-293
- 2) Goran Alderbond, Pharmaceutical powder compaction technology, material for direct compression, page no. 428-440
- 3) Excipient toxicity and safety, Myra L. Weiner (volume 103)
- 4) William J Reilly, Remington's pharmaceutical science, Pharmaceutical Necessities 8th edition page no. 1084
- 5) Y. Nishiyama, P. Langan, H. Chanzy (2002). "Crystal Structure and Hydrogen-Bonding System in Cellulose I β from Synchrotron X-ray and Neutron Fiber Diffraction". 124 (31): 9074–9082.



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