

Charge Carriers Are Not Affected by the Relatively Slow-Rotating Methylammonium Cations in Lead Halide Perovskite Thin Films

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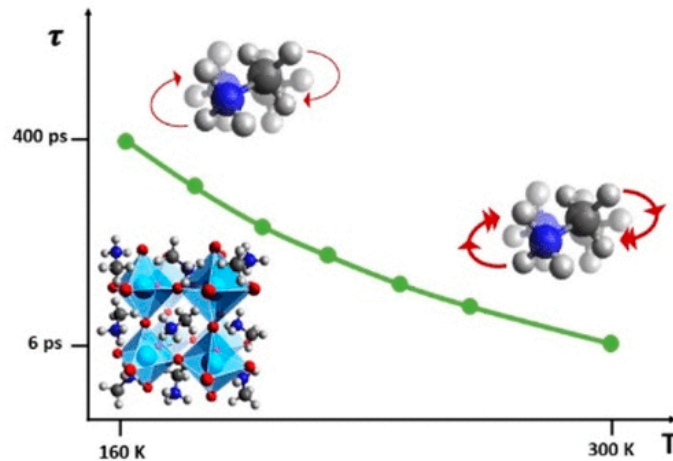
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Abstract: Recently, several studies have investigated dielectric properties as a possible origin of the exceptional optoelectronic properties of metal halide perovskites (MHPs). In this study we investigated the temperature-dependent dielectric behavior of different MHP films at different frequencies. In the gigahertz regime, dielectric losses in methylammonium-based samples are dominated by the rotational dynamics of the organic cation. Upon increasing the temperature from 160 to 300 K, the rotational relaxation time, τ , decreases from 400 (200) to 6 (1) ps for MAPb-I₃ (-Br₃). By contrast, we found negligible temperature-dependent variations in τ for a mixed cation/mixed halide FA_{0.85}MA_{0.15}Pb(I_{0.85}Br_{0.15})₃. From temperature-dependent time-resolved microwave conductance measurements we conclude that the dipolar reorientation of the MA cation does not affect charge carrier mobility and lifetime in MHPs. Therefore, charge carriers do not feel the relatively slow-moving MA cations, despite their great impact on the dielectric constants.



In the past few years, solar cells based on metal halide perovskite (MHP) semiconductors, with general structure ABX_3 , have emerged as promising low-cost alternatives to established semiconductors like silicon and CIGS. In addition, other applications such as light-emitting diodes, lasers and photodetectors were recently reported.⁽¹⁾ The success of MHP-based solar cells is due to their exceptional optoelectronic properties, such as high absorption coefficients, low exciton binding energies, and long charge carrier lifetimes and diffusion lengths.^(2–5) Part of these properties can be attributed to the low densities of trap states and background charges.^(5,6) In addition, it has been suggested that the large dipole moment of the methylammonium group (MA^+) in $MAPbI_3$ can result in the formation of ferroelectric domains by the collective motion of the MA^+ dipoles.^(7,8) This, although still under debate, might lead to enhanced exciton dissociation, polaron formation,⁽⁹⁾ reduced charge carrier recombination, and defect screening.^(10–12) The rotational dynamics of the MA cation have been theoretically and experimentally studied.^(11,13–21) First studies on the rotational dynamics were carried out using IR⁽¹⁹⁾ and dielectric⁽²²⁾ techniques. IR measurements on powders revealed a very rapid reorientation of the MA cation around the C–N axis with activation energies of 2.6 and 2.0 kJ/mol and relaxation times 1 and 0.7 ps for $MAPbI_3$ for $MAPbBr_3$, respectively.⁽¹⁹⁾ On the other hand, from dielectric measurements at 90 GHz, Poglitsch and Weber determined relaxation times of 5.37 and 2.73 ps for $MAPbI_3$ and $MAPbBr_3$, respectively.⁽²²⁾ As extensively discussed by Gallop et al.,⁽¹²⁾ these values can be attributed to two different motions, namely, a rapid wobbling and a slower reorientation of the MA^+ group. Raman spectroscopy has been more recently applied to study the dynamic (dis)order in perovskite materials. While the link between the (dis)order of the MA cation and the orthorhombic–tetragonal phase transition in MA^+ -based MHPs appears to be clear,^(11,20) its influence on the lattice dynamics is still under debate.^(12,23) Furthermore, the impact of MA^+ dipole on the formation of ferroelectric domains is still under discussion, although recent dielectric⁽¹¹⁾ and impedance spectroscopy (IS)⁽²¹⁾ studies reported no indication of formation of ferroelectric domains in these compounds. Theoretical modeling confirmed the limited stability of a possible ferroelectric $MAPbI_3$ phase at room temperature.^(24,25)

Despite the above-mentioned studies on dipolar motion in MHPs, knowledge of dielectric properties in the gigahertz (GHz) regime is still limited. Poglitsch and Weber⁽²²⁾ and Anusca et al.⁽¹¹⁾ have shown the great impact of temperature on dielectric properties of these materials, but both studies are missing information around 10 GHz. In this range, substantial changes in the temperature-dependent dielectric behavior related to the presence of the MA^+ dipolar reorientational motion can be expected. Moreover, to the best of our knowledge, no such study has been devoted to MHPs with different cations.

In this study we investigate the relationship between the perovskite constituents, temperature, and the dielectric properties by means of microwave conductance (MC) measurements in the GHz regime. All these measurements are performed in the dark in contrast to standard time-resolved microwave conductance (TRMC) measurements in which light-induced changes in the real and imaginary parts of the dielectric constant of many materials have been studied, revealing the time-dependent formation of, for example, charge-transfer states,⁽²⁶⁾ excitons,⁽²⁷⁾ and mobile charges.⁽⁶⁾ The high quality factor of our cavity with resonance frequencies around 10 GHz ensures that even small changes in the dielectric properties by, for example, changing the constituents of the MHPs can be determined. Next, we prepared devices from the corresponding layers by using indium tin oxide (ITO) and Au as contact layers and recorded the real and imaginary dielectric properties by IS between 20 Hz and 2 MHz in the dark.

We performed MC measurements on the following spin-coated MHP layers: MAPbI₃, MAPbBr₃, FAPbI₃, CsPbI₃, and FA_{0.85}MA_{0.15}Pb(I_{0.85}Br_{0.15})₃ (FAMA). Interestingly, MA-based MHPs exhibited a strong temperature-dependent dielectric loss in the GHz regime, which is absent for the FA- and Cs-based MHPs. With the low-frequency IS measurements, no appreciable temperature-dependent changes were observed for both MAPbI₃ and FAPbI₃, meaning that the dielectric loss in the GHz regime is not related to background charges. Therefore, we attribute the changes in dielectric losses to the rotational motion of the MA⁺ dipoles. Activation energies for the rotational motion and corresponding temperature-dependent relaxation times for MAPbI₃, MAPbBr₃, and FAMA were obtained by applying the Cole–Cole model⁽²⁸⁾ to our results. Despite the great impact of the dipolar orientation on dielectric properties in the GHz regime, no major implications for the charge carrier dynamics were found.

MHP samples were fabricated on quartz or on patterned ITO by the two step spin-coating procedure followed by a 5 min annealing step at 100 °C. For optical and structural characterization see [Figure S1](#) of the Supporting Information. For the MC technique an MHP layer on quartz was mounted in a temperature-controlled microwave cavity without any exposure to air. Apart from its superior sensitivity, the MC technique does not require electrical contacts. First, a scan of the reflected microwave power is recorded by sweeping the frequency, ν , over a certain microwave range revealing a dip, which can be attributed to the formation of a standing wave pattern within the loaded cavity. Next, the microwave scans are normalized to a scan recorded by replacing the cavity with a fully reflecting end plate. In [Figure 1](#) the normalized scans of MAPbI₃, MAPbBr₃, FAPbI₃, and FAMA are shown in a temperature range varying from 120 to 300 K. The gradual shift to lower frequencies upon heating combined with the increase in microwave absorption can be explained by changes of the cavity, i.e. expanding of the cavity length and enhanced losses in the metallic walls, respectively (see the [Supporting Information](#) for more details). Most importantly, the scans of the MA-based samples are substantially different from those of FAPbI₃ and CsPbI₃. That is, up to 220 K for MAPbI₃ and up to 200 K for MAPbBr₃, we observe an increase of microwave absorption, which then decreases again at higher temperatures. For FAPbI₃ and CsPbI₃, on the other hand, the scans are very similar to those observed for a bare quartz substrate (see [Figure S2b](#)).

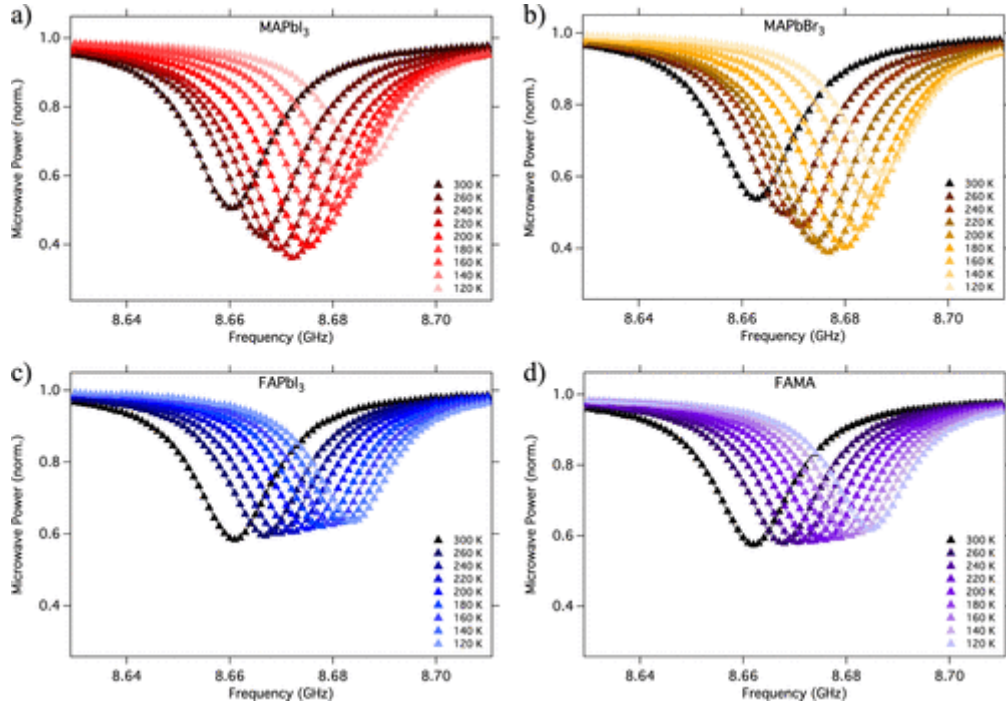


Figure 1. Normalized microwave power as a function of frequency recorded at different temperatures for (a) MAPbI₃, (b) MAPbBr₃, (c) FAPbI₃, and (d) FA_{0.85}MA_{0.15}Pb(I_{0.85}Br_{0.15})₃ (FAMA). The solid lines are the results of the fits to the experimental data points (triangles).

The power scans in Figure 1 were modeled using a custom-built computer program, which numerically solves the Maxwell equations using the resonance characteristics of the cavity and the dielectric properties and dimensions of all media inside the cell as input.⁽²⁹⁾ The results of the fits are shown as solid lines in Figure 1, and the temperature-dependent conductivities are summarized in Table S1 in the Supporting Information. The values of the imaginary dielectric constant, ϵ'' , are calculated from the real conductivities, σ_{Re} , by

$$\sigma_{\text{Re}} = \omega \epsilon_0 \epsilon'' \quad (1)$$

where ω is the angular frequency and ϵ_0 the vacuum permittivity (Figure 2a). For both MA-based MHPs a clear maximum is observed, while FAPbI₃ and FAMA show almost no temperature dependence.

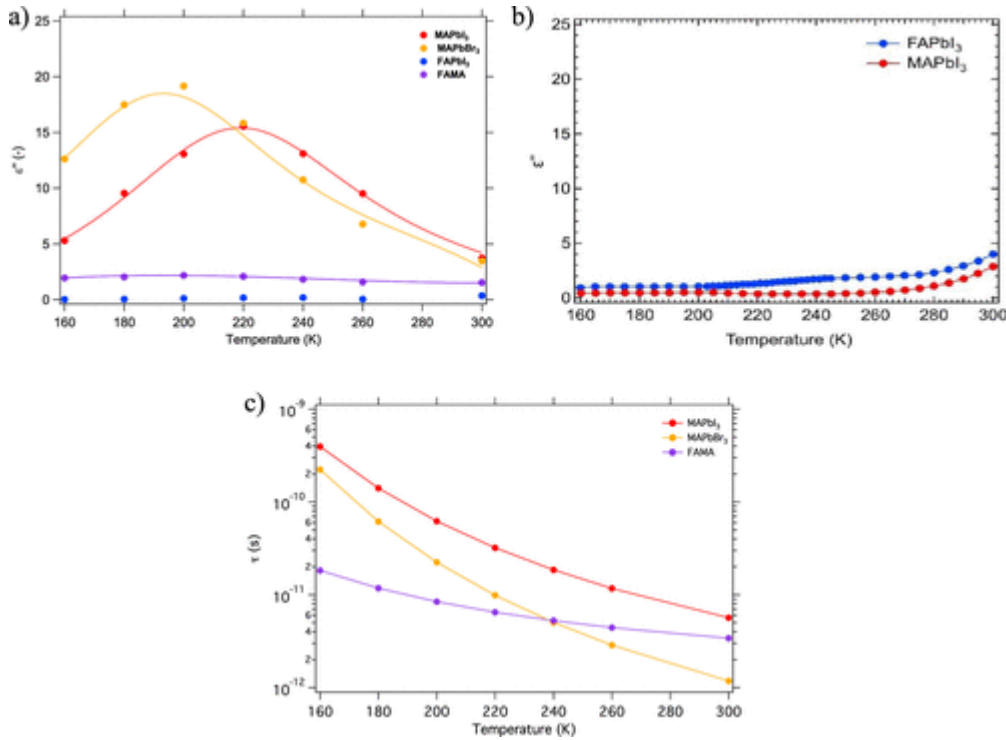


Figure 2. Imaginary part of the complex dielectric constant (a) from MC measurements at 8.6 GHz and (b) at 1 kHz obtained by impedance spectroscopy. (c) Relaxation times for the MA⁺ containing MHPs.

We note that absorption of microwaves by MHPs can have multiple origins, including the presence of mobile charge carriers, of mobile ions, and/or dipolar reorientation-related losses. For this reason, we conducted additional IS measurements on MAPbI₃ and FAPbI₃ over a wide frequency range, from 20 Hz to 2 MHz. At frequencies of 10 kHz and above, ϵ'' is already influenced by series resistance resulting from the electrical contacts ITO and Au (see Figure S7). Therefore, the resulting relative value of ϵ'' as a function of temperature at 1 kHz is shown for FAPbI₃ and MAPbI₃ in Figure 2b. In contrast to the MC results, the temperature dependence of ϵ'' determined by IS for MAPbI₃ and FAPbI₃ are similar and the specific maximum observed at 220 K in the GHz regime for MAPbI₃ (Figure 2a) is not present at kilohertz (kHz) frequencies (Figure 2b). In addition, we observe an increase of ϵ'' with temperature for both devices in the kHz regime, which we ascribe to temperature-activated ionic conductivity^(30,31) and which is not observed in MC. These dissimilarities between the MC and IS measurements provide evidence against the hypothesis that temperature-dependent dielectric losses in the GHz range are due to mobile species such as charges or mobile ions. The observation that only the MA-based MHPs show a dielectric behavior that is considerably different than that of a quartz reference substrate implies that the dipolar reorientation of the MA dipolar cation is responsible for the distinctive temperature dependence of ϵ'' shown in Figure 2a.

Previously the dielectric behavior of the MA⁺-based MHPs has been modeled successfully to extract relaxation times and activation energies of MA reorientation using the Cole–Cole model,⁽²⁸⁾ in which the complex dielectric constant ϵ^* is defined as

$$\varepsilon^* = \varepsilon_\infty + \frac{\varepsilon_S - \varepsilon_\infty}{1 + (i\omega\tau)^{(1-\alpha)}} \quad (2)$$

where ε_∞ and ε_S represent the high-frequency and static dielectric constants, respectively; τ represents the specific relaxation time, and α is a parameter reflecting the width of the relaxation time distribution. The specific expressions for the real and imaginary parts are given in the [Supporting Information](#). The temperature dependence of the static and high-frequency dielectric constants are derived from the IS (at 1 kHz) and MC (fitting in the range 8.6–8.7 GHz) measurements, respectively. Values are collected in [Table S1](#) in the Supporting Information. The relaxation time, τ , is defined as

$$\tau = \tau_0 \exp\left(\frac{E_A}{k_B(T - T_0)}\right) \quad (3)$$

in which τ_0 is the relaxation time at high temperatures, E_A the activation energy, and T_0 the temperature at which the relaxation time could be considered as “infinitely slow”.[\(32,33\)](#)

The experimental ε'' data points are fitted with the Cole–Cole equation, and the resulting fits are added in [Figure 2a](#), showing excellent overlap for both MAPbI₃ and MAPbBr₃.

We notice that for MAPbBr₃ the model is able to fit the results even in the cubic phase ($T > 235$ K) with minimal discrepancies. When a prefactor of 0.41 for the FAMA sample is included, a proper match between experimental data points and fit is observed. This experimentally obtained number accounts for different factors, i.e., the reduced MA⁺ concentration in the sample, MA⁺–FA⁺ interactions, and the influence of the mixed halide. Proper quantification of each of these factors in such a complex system would require analyses and computational studies that go beyond the scope of this work. Despite the fact that τ of the FA cations should be in the same range as those of MA cations,[\(12\)](#) our analysis could not be applied to the FAPbI₃ results because of the low magnitude of the dielectric losses. The difference with MA⁺-based MHPs can be explained on the basis of the reduced dipolar character of FA compared to MA cations, which implies a significantly reduced response to the applied electric field in the former system.

The temperature-dependent relaxation times for the MA⁺-based MHPs are shown in [Figure 2c](#). Values at room temperature vary from 1 ps for MAPbBr₃ to 6 ps for MAPbI₃ and are comparable to those previously determined on compressed powders by neutron scattering[\(17\)](#) and dielectric measurements.[\(11,22\)](#) Activation energies of 114, 135, and 37 meV were determined for the rotational motion of MAPbI₃, MAPbBr₃, and FAMA, respectively. The increase in E_A from MAPbI₃ to MAPbBr₃ can be related to the increased hydrogen bonding strength and reduction of the octahedral cavity size,[\(12\)](#) as confirmed by XRD measurements (see [Figure S1b](#)). DFT calculations (see computational details section in the [Supporting Information](#)) also confirm the stronger hydrogen bond between MA cations and the inorganic framework in MAPbBr₃ than in MAPbI₃ (0.06 eV difference in MA binding energy in favor of MAPbBr₃). Notably, FA experiences a higher rotational freedom compared to MA, as evidenced by ab initio molecular dynamics simulations,[\(34\)](#) consistent with the reduced activation energy for cation reorientation observed here. Despite such trends, the relaxation times are shorter at all temperatures for MAPbBr₃ in comparison with MAPbI₃. This has

been previously reported by Selig et al.,⁽³⁵⁾ and has been attributed to a higher probability of large angle jumps combined with tilting of the inorganic cage.^(12,35) In contrast to previous findings,⁽³⁵⁾ for the mixed cation/halide, FAMA, we observe a much lower activation energy and relaxation times comparable to MAPbI₃ at room temperature. Furthermore, the relaxation times for FAMA do not show an evident temperature dependence, while, as expected, for both MAPbI₃ and MAPbBr₃, τ is considerably slowed down upon cooling. We attribute the FAMA results to a combined effect of the rotational freedom in a larger crystal structure induced by the predominant presence of FA⁺ cations and a reduced dipolar interaction between the MA ions.

Next, we can use the experimentally found relaxation times, to calculate the frequency-dependent values of ϵ'' using the Cole–Cole equation (see Figures S5 for ϵ' versus frequency). For MAPbI₃ and MAPbBr₃, these extrapolations are shown as solid lines for different temperatures in Figure 3, crossing our experimental values at 8.6 GHz. Interestingly, the maximum in dielectric loss shifts with temperature from frequencies much lower than 8.6 GHz to values much higher than 8.6 GHz. This behavior explains our observed peaks for the dielectric loss at 220 and 200 K for MAPbI₃ and MAPbBr₃, respectively. The present interpretation agrees with the IS measurements; that is, ϵ'' continuously decreases with lower frequencies and drops below 2 at 4 kHz for all investigated temperatures (see Figure S8). Furthermore, at these frequencies we observe a huge difference in ϵ' between MAPbI₃ and FAPbI₃ (see Figure S7). Thus, we conclude that large, temperature-dependent variations in the real and imaginary parts of the dielectric constant are observed specifically in the GHz regime for MA-based MHPs.

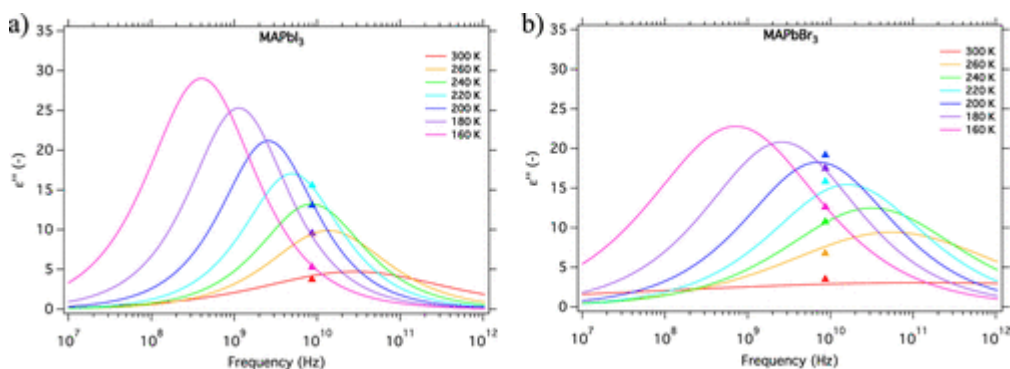


Figure 3. Calculated values of the dielectric losses at different temperatures versus frequency for (a) MAPbI₃ and (b) MAPbBr₃ by applying the Cole–Cole equation. The triangles indicate the experimental data points.

As previously mentioned, the motion of the MA cation has been subject of research to understand the exceptional optoelectronic properties of MAPbI₃. In particular, the possibility of enhanced charge carrier diffusion lengths as a result of weaker trapping^(7,11) and slower recombination due to defect and electronic screening^(11,36) are still debated. Over the years we have investigated the charge carrier decay kinetics of several MHPs, differing in composition and fabrication method.^(37–40) On the basis of the results obtained, some of us previously concluded that the organic cation does not play a major role in the charge carrier kinetics.^(38,41) In agreement with these findings, charge/lattice response due to formation of large polarons was found to have a similar subpicosecond time scale in both MA- and Cs-PbBr₃ perovskites.⁽⁴²⁾

To verify if the rotational relaxation times affect the charge carrier dynamics, time-resolved microwave photoconductance (TRMC) traces are recorded at different temperatures and shown in Figure 4. Generally, all the MHP layers analyzed above (CsPbI₃ in Supporting Information) show similar trends on increasing the temperature from 160 to 300 K: (i) lower charge carrier mobilities

and (ii) shorter charge carrier lifetimes. A detailed discussion of these findings can be found in previously published papers.[\(43,44\)](#)

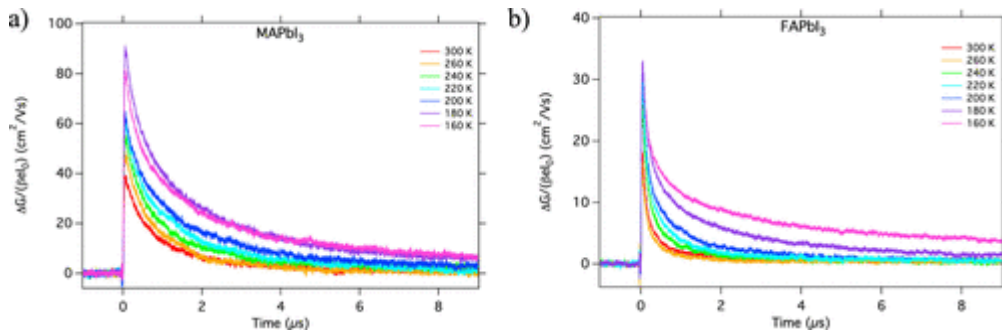


Figure 4. TRMC traces recorded on laser pulses with intensities of ca. $1\text{--}3 \times 10^9$ photons/cm² at 650 nm for (a) MAPbI₃ and (b) FAPbI₃.

With respect to the above discussion on the dielectric losses we can conclude the following: the temperature-dependent TRMC traces confirm that the changes in the dielectric losses observed with MC for the MA-based MHPs are not due to mobile charge carriers. If the changes in dielectric loss were due to mobile electrons and/or holes, the decay kinetics of MA-based samples should have been greatly affected, especially around 200–220 K. However, such behavior is not observed for MAPbI₃, as shown in [Figure 4](#). More importantly, the results reported here support the conclusion of a negligible impact of the nature of the A-site cation on the decay kinetics. In fact, despite the large change in dielectric loss, the temperature-dependent trend in mobility and lifetime of excess charge carriers in MA- and FA-based MHPs is very similar. On the other hand, no direct similarities can be found between MAPbI₃ and MAPbBr₃,[\(40,43\)](#) for which charge carrier mobility and recombination are substantially different ([Figures 5a and S9a](#), respectively). Hence, from the present data no indications of reduced charge carrier recombination and/or defect screening by the MA are observed. A similar effect was previously discussed in the context of exciton screening, finding that the low-energy MA rotational dynamics barely affects the exciton binding energy in MAPbI₃, which is instead significantly affected by phonons in the terahertz region.[\(45\)](#)

In this work, we have investigated the temperature-dependent microwave conductivity of several metal halide perovskite materials in the dark. Interestingly, all MA-based samples showed high dark conductivities in the GHz regime with distinctive temperature variations. In contrast, no appreciable dark conductivity can be observed at frequencies of 1 kHz using IS measurements, which excludes ionic and electronic contributions. Therefore, we attribute the conductivities observed in the GHz regime to dielectric losses related to the rotation dynamics of the MA cation. Using the Cole–Cole model, we determined the activation energies and temperature-dependent relaxation times for different MA-based samples. On comparison of these results with the temperature-dependent photoconductivity, no correlation between dipolar rotational dynamics and light-induced charge carrier dynamics in MHPs was found. These results show that the huge changes in the dielectric constant at GHz frequencies do not impact the carrier dynamics due to the different time scales involved. Hence, we conclude that the fast carriers do not feel the slow-moving MA cations, despite the fact that the latter contribute to the overall value of the dielectric constant.

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References

- (1) Sun, J.; Wu, J.; Tong, X.; Lin, F.; Wang, Y.; Wang, Z. M. Organic/Inorganic Metal Halide Perovskite Optoelectronic Devices beyond Solar Cells. *Adv. Sci.* 2018, 5 (5), 1700780.
- (2) De Wolf, S.; Holovsky, J.; Moon, S. J.; Löper, P.; Niesen, B.; Ledinsky, M.; Haug, F. J.; Yum, J. H.; Ballif, C. Organometallic Halide Perovskites: Sharp Optical Absorption Edge and Its Relation to Photovoltaic Performance. *J. Phys. Chem. Lett.* 2014, 5 (6), 1035–1039.
- (3) Savenije, T. J.; Ponseca, C. S.; Kunneman, L.; Abdellah, M.; Zheng, K.; Tian, Y.; Zhu, Q.; Canton, S. E.; Scheblykin, I. G.; Pullerits, T.; et al. Thermally Activated Exciton Dissociation and Recombination Control the Carrier Dynamics in Organometal Halide Perovskite. *J. Phys. Chem. Lett.* 2014, 5 (13), 2189–2194.
- (4) Umari, P.; Mosconi, E.; De Angelis, F. Infrared Dielectric Screening Determines the Low Exciton Binding Energy of MetalHalide Perovskites. *J. Phys. Chem. Lett.* 2018, 9 (3), 620–627.
- (5) Bi, Y.; Hutter, E. M.; Fang, Y.; Dong, Q.; Huang, J.; Savenije, T. J. Charge Carrier Lifetimes Exceeding 15 Ms in Methylammonium Lead Iodide Single Crystals. *J. Phys. Chem. Lett.* 2016, 7 (5), 923–928.
- (6) Hutter, E. M.; Eperon, G. E.; Stranks, S. D.; Savenije, T. J. Charge Carriers in Planar and Meso-Structured Organic-Inorganic Perovskites: Mobilities, Lifetimes, and Concentrations of Trap States. *J. Phys. Chem. Lett.* 2015, 6 (15), 3082–3090.
- (7) Frost, J. M.; Butler, K. T.; Walsh, A. Molecular Ferroelectric Contributions to Anomalous Hysteresis in Hybrid Perovskite Solar Cells. *APL Mater.* 2014, 2 (8), 081506.
- (8) Frost, J. M.; Butler, K. T.; Brivio, F.; Hendon, C. H.; van Schilfgaarde, M.; Walsh, A. Atomistic Origins of High-Performance in Hybrid Halide Perovskite Solar Cells. *Nano Lett.* 2014, 14 (5), 2584–2590.
- (9) Herz, L. M. How Lattice Dynamics Moderate the Electronic Properties of Metal-Halide Perovskites. *J. Phys. Chem. Lett.* 2018, 9 (23), 6853–6863.
- (10) Ma, J.; Wang, L. W. Nanoscale Charge Localization Induced by Random Orientations of Organic Molecules in Hybrid Perovskite $\text{CH}_3\text{NH}_3\text{PbI}_3$. *Nano Lett.* 2015, 15 (1), 248–253.
- (11) Anusca, I.; Balciū̄nas, S.; Gemeiner, P.; Svirskas, Š.; Sanlialp, M.; Lackner, G.; Fettkenhauer, C.; Belovickis, J.; Samulionis, V.; Ivanov, M.; et al. Dielectric Response: Answer to Many Questions in the Methylammonium Lead Halide Solar Cell Absorbers. *Adv. Energy Mater.* 2017, 7 (19), 1700600.
- (12) Gallop, N. P.; Selig, O.; Giubertoni, G.; Bakker, H. J.; Rezus, Y. L. A.; Frost, J. M.; Jansen, T. L. C.; Lovrincic, R.; Bakulin, A. A. Rotational Cation Dynamics in Metal Halide Perovskites: Effect on Phonons and Material Properties. *J. Phys. Chem. Lett.* 2018, 9 (20), 5987–5997.
- (13) Mattoni, A.; Filippetti, A.; Saba, M. I.; Delugas, P. Methylammonium Rotational Dynamics in Lead Halide Perovskite by Classical Molecular Dynamics: The Role of Temperature. *J. Phys. Chem. C* 2015, 119 (30), 17421–17428.
- (14) Kanno, S.; Imamura, Y.; Hada, M. Theoretical Study on Rotational Controllability of Organic Cations in Organic-Inorganic Hybrid Perovskites: Hydrogen Bonds and Halogen Substitution. *J. Phys. Chem. C* 2017, 121 (47), 26188–26195.

- (15) Kanno, S.; Imamura, Y.; Saeki, A.; Hada, M. Rotational Energy Barriers and Relaxation Times of the Organic Cation in Cubic Methylammonium Lead/Tin Halide Perovskites from First Principles. *J. Phys. Chem. C* 2017, 121 (26), 14051–14059.
- (16) Govinda, S.; Kore, B. P.; Bokdam, M.; Mahale, P.; Kumar, A.; Pal, S.; Bhattacharyya, B.; Lahnsteiner, J.; Kresse, G.; Franchini, C. Behavior of Methylammonium Dipoles in MAPbX₃ (X = Br and I). *J. Phys. Chem. Lett.* 2017, 8, 4113.
- (17) Chen, T.; Foley, B. J.; Ipek, B.; Tyagi, M.; Copley, J. R. D.; Brown, C. M.; Choi, J. J.; Lee, S. H. Rotational Dynamics of Organic Cations in the CH₃NH₃PbI₃ Perovskite. *Phys. Chem. Chem. Phys.* 2015, 17 (46), 31278–31286.
- (18) Li, J.; Bouchard, M.; Reiss, P.; Aldakov, D.; Pouget, S.; Demadrille, R.; Aumaitre, C.; Frick, B.; Djurado, D.; Rossi, M.; et al. Activation Energy of Organic Cation Rotation in CH₃NH₃PbI₃ and CD₃NH₃PbI₃: Quasi-Elastic Neutron Scattering Measurements and First-Principles Analysis Including Nuclear Quantum Effects. *J. Phys. Chem. Lett.* 2018, 9 (14), 3969–3977.
- (19) Onoda-Yamamuro, N.; Matsuo, T.; Suga, H. Calorimetric and IR Spectroscopic Studies of Phase Transitions in Methylammonium Trihalogenoplumbates (II). *J. Phys. Chem. Solids* 1990, 51 (12), 1383–1395.
- (20) Quarti, C.; Grancini, G.; Mosconi, E.; Bruno, P.; Ball, J. M.; Lee, M. M.; Snaith, H. J.; Petrozza, A.; De Angelis, F. The Raman Spectrum of the CH₃NH₃PbI₃ Hybrid Perovskite: Interplay of Theory and Experiment. *J. Phys. Chem. Lett.* 2014, 5 (2), 279–284.
- (21) Hoque, M. N. F.; Yang, M.; Li, Z.; Islam, N.; Pan, X.; Zhu, K.; Fan, Z. Polarization and Dielectric Study of Methylammonium Lead Iodide Thin Film to Reveal Its Nonferroelectric Nature under Solar Cell Operating Conditions. *ACS Energy Lett.* 2016, 1 (1), 142–149.
- (22) Poglitsch, A.; Weber, D. Dynamic Disorder in Methylammoniumtrihalogenoplumbates (II) Observed by Millimeter-Wave Spectroscopy. *J. Chem. Phys.* 1987, 87 (11), 6373–6378.
- (23) Yaffe, O.; Guo, Y.; Tan, L. Z.; Egger, D. A.; Hull, T.; Stoumpos, C. C.; Zheng, F.; Heinz, T. F.; Kronik, L.; Kanatzidis, M. G.; et al. Local Polar Fluctuations in Lead Halide Perovskite Crystals. *Phys. Rev. Lett.* 2017, 118 (13), 1–6.
- (24) Stroppa, A.; Quarti, C.; De Angelis, F.; Picozzi, S. Ferroelectric Polarization of CH₃NH₃PbI₃: A Detailed Study Based on Density Functional Theory and Symmetry Mode Analysis. *J. Phys. Chem. Lett.* 2015, 6 (12), 2223–2231.
- (25) Quarti, C.; Mosconi, E.; De Angelis, F. Interplay of Orientational Order and Electronic Structure in Methylammonium Lead Iodide: Implications for Solar Cell Operation. *Chem. Mater.* 2014, 26 (22), 6557–6569.
- (26) Piet, J. J.; Schuddeboom, W.; Wegewijs, B. R.; Grozema, F. C.; Warman, J. M. Symmetry Breaking in the Relaxed S₁ Excited State of Biantthryl Derivatives in Weakly Polar Solvents. *J. Am. Chem. Soc.* 2001, 123 (22), 5337–5347.
- (27) Fravventura, M. C.; Deligiannis, D.; Schins, J. M.; Siebbeles, L. D. A.; Savenije, T. J. What Limits Photoconductance in Anatase TiO₂ Nanostructures? A Real and Imaginary Microwave Conductance Study. *J. Phys. Chem. C* 2013, 117 (16), 8032–8040.
- (28) Cole, K. S.; Cole, R. H. Dispersion and Absorption in Dielectrics. *J. Chem. Phys.* 1941, 9 (1913), 341–351.
- (29) Schins, J. M.; Talgorn, E. Conductive Response of a PhotoExcited Sample in a Radio-Frequent Driven Resonance Cavity. *Rev. Sci. Instrum.* 2011, 82 (6), 064703.
- (30) Azpiroz, J. M.; Mosconi, E.; Bisquert, J.; De Angelis, F. Defect Migration in Methylammonium Lead Iodide and Its Role in Perovskite Solar Cell Operation. *Energy Environ. Sci.* 2015, 8 (7), 2118–2127.
- (31) Frost, J. M.; Walsh, A. What Is Moving in Hybrid Halide Perovskite Solar Cells? *Acc. Chem. Res.* 2016, 49 (3), 528–535.
- (32) Küçükcelebi, H.; Durmus, H.; Deryal, A.; Tas, M.; Karakaya, N. Activation Energy of Polarization Due to Electrical Conductivity and Dipole Rotation in Purified Ca-Bentonite. *Appl. Clay Sci.* 2012, 62–63 (March 2019), 70–79.

- (33) Davidson, D. W.; Cole, R. H. Dielectric Relaxation in Glycerol, Propylene Glycol, and n-Propanol. *J. Chem. Phys.* 1951, 19 (12), 1484.
- (34) Quarti, C.; Mosconi, E.; De Angelis, F. Structural and Electronic Properties of Organo-Halide Hybrid Perovskites from Ab Initio Molecular Dynamics. *Phys. Chem. Chem. Phys.* 2015, 17 (14), 9394–9409.
- (35) Selig, O.; Sadhanala, A.; Müller, C.; Lovrincic, R.; Chen, Z.; Rezus, Y. L. A.; Frost, J. M.; Jansen, T. L. C.; Bakulin, A. A. Organic Cation Rotation and Immobilization in Pure and Mixed Methylammonium Lead-Halide Perovskites. *J. Am. Chem. Soc.* 2017, 139 (11), 4068–4074.
- (36) Milot, R. L.; Eperon, G. E.; Snaith, H. J.; Johnston, M. B.; Herz, L. M. Temperature-Dependent Charge-Carrier Dynamics in CH₃NH₃PbI₃ Perovskite Thin Films. *Adv. Funct. Mater.* 2015, 25 (39), 6218–6227.
- (37) Hutter, E. M.; Sutton, R. J.; Chandrashekar, S.; Abdi-Jalebi, M.; Stranks, S. D.; Snaith, H. J.; Savenije, T. J. Vapour-Deposited Cesium Lead Iodide Perovskites: Microsecond Charge Carrier Lifetimes and Enhanced Photovoltaic Performance. *ACS Energy Lett.* 2017, 2, 1901–1908.
- (38) Hu, Y.; Hutter, E. M.; Rieder, P.; Grill, I.; Hanisch, J.; Aygüler, M. F.; Hufnagel, A. G.; Handloser, M.; Bein, T.; Hartschuh, A.; et al. Understanding the Role of Cesium and Rubidium Additives in Perovskite Solar Cells : Trap States, Charge Transport, and Recombination. *Adv. Energy Mater.* 2018, 8, 1703057.
- (39) Abdi-Jalebi, M.; Andaji-Garmaroudi, Z.; Cacovich, S.; Stavrakas, C.; Philippe, B.; Richter, J. M.; Alsari, M.; Booker, E. P.; Hutter, E. M.; Pearson, A. J.; et al. Maximizing and Stabilizing Luminescence from Halide Perovskites with Potassium Passivation. *Nature* 2018, 555, 497.
- (40) Guo, D.; Bartesaghi, D.; Wei, H.; Hutter, E. M.; Huang, J.; Savenije, T. J. Photoluminescence from Radiative Surface States and Excitons in Methylammonium Lead Bromide Perovskites. *J. Phys. Chem. Lett.* 2017, 8 (17), 4258–4263.
- (41) Hutter, E. M.; Savenije, T. J. Thermally Activated SecondOrder Recombination Hints toward Indirect Recombination in Fully Inorganic CsPbI₃ Perovskites. *ACS Energy Lett.* 2018, 3, 2068.
- (42) Trinh, M. T.; Miyata, K.; Joshi, P. P.; Jones, S. C.; De Angelis, F.; Mosconi, E.; Zhu, X.-Y.; Meggiolaro, D. Large Polarons in Lead Halide Perovskites. *Sci. Adv.* 2017, 3 (8), e1701217.
- (43) Hutter, E. M.; Gelvez-Rueda, M. C.; Osherov, A.; Bulovic, V.; Grozema, F. C.; Stranks, S. D.; Savenije, T. J. Direct-Indirect Character of the Bandgap in Methylammonium Lead Iodide Perovskite. *Nat. Mater.* 2017, 16 (1), 115–120.
- (44) Osherov, A.; Hutter, E. M.; Galkowski, K.; Brenes, R.; Maude, D. K.; Nicholas, R. J.; Plochocka, P.; Bulovic, V.; Savenije, T. J.; Stranks, S. D. The Impact of Phase Retention on the Structural and Optoelectronic Properties of Metal Halide Perovskites. *Adv. Mater.* 2016, 28 (48), 10757–10763.
- (45) Umari, P.; Mosconi, E.; De Angelis, F. Infrared Dielectric Screening Determines the Low Exciton Binding Energy of MetalHalide Perovskites. *J. Phys. Chem. Lett.* 2018, 9 (3), 620–627.