



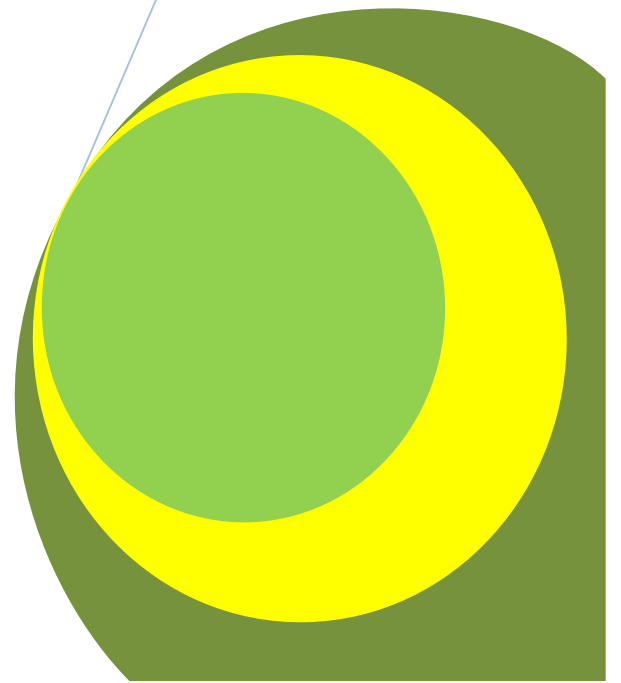
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The Effect of Soil and Foliar Fertilization with Iron on Yield and Leaf Chemical Composition of Four Spotted Bean Cultivars in a Calcareous Soil

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Research Article

The Effect of Soil and Foliar Fertilization with Iron on Yield and Leaf Chemical Composition of Four Spotted Bean Cultivars in a Calcareous Soil

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ABSTRACT

Legumes are one of the main protein sources for human beings and animals. Bean by 20- 25% protein, 55 -56% carbohydrates and 1-5% lipid plays a basic rule in the nutrition of poor nations. So the increase of yield and quality of bean can be an efficient way for reduction of protein loss in the developing countries. One of the efficient factors in reduction of bean yield and quality in the calcareous soils is lack of iron. High reaction and lime amount and extra bicarbonate of calcareous soils cause leaves chlorosis which reduces growth, yield and quality of bean. In the calcareous soils, absorption of less application elements especially iron in the majority of agricultural products in the bean farming which is sensitive to iron lack encountered with difficulty and its lack is clearly observable. According to farming observations, various type of bean have different potentials from the yield view spotted and there is probability that they have different reaction towards the iron lack, so it is necessary to study in this ground in order to determine the reaction of these figures to iron in various fertilizers. In this regard and in order to improve, bean products quality and increase the yield level of this product to the most probable level, it is necessary to investigate the effect of sources fertilizer methods (soil consume- sparing) on the yield and the quality of this product. In order to inspect the number, sources and various methods of iron fertilizer consuming on micro elements and chemical compound, variants kinds of spotted bean in a limy soil, a green hones trial by 44 treatments and 3 iterates it erases in the factorial form and in the form of complete incidental sign in the zanjan agriculture university of Iran were implemented. In this testing, the consumption of sulphoric acid in watering (2 g/L),4 g/L), soil application of Ferrous sulfate (25 mg/ in each kilogram soli), 50 mg/ in each kilogram soli), soil application of Fe- EDTA(3 mg/ in each kilogram soli), 6 mg/ in each kilogram soli), spraying Ferrous sulfate by density of 2 and 4 g/ L and spraying Fe- EDTA by density of 1 and 2 g/ L are considered as iron treatments. By the way an evidence cure is also performed on four types of the spotted bean including Talash, khomein, COS 16 and Local. The result of analysis of the data variance Clemtion started that the types Talash and khomein were superior to other kinds from the yield of the seed weight of 100 seeds, numbers of the seeds in the sheath viewpoint. The Local and Talash beans were preferred to other kinds. The treatment of sulphoric acid application in irrigation water (4 g/L) has the higher level of average compared to other treatments of iron. The treatment of sulphoric acid application in irrigation water causes the increase of manganese, zinc and copper of the leaves in the spotted bean types.

Keywords: Spotted bean types, Iron fertilizers, sulphoric acid, Yield, chemical composition.

INTRODUCTION

Grains are from the legumes (papilionaceae) family including 750 genus and 20000 species. Legumes are the most important agriculture products after wheat and rice which are consumed by the people of the world and supply a major part of their requirement proteins. In many countries situated in arid and semi-arid regions such as Iran, soils are mostly lime-rich, light- to medium-textured and of low organic matter concentration, resulting in an inadequate nutrient supply (Shadkami-Til, 2013c; Javanmiri Pour et al., 2013). Calcareous soils are defined as soils containing sufficient CaCO_3 and MgCO_3 to effervesce visibly when treated with a strong acid (Soil Science Society of America, 1997). Alkaline and hard irrigation waters are difficult to manage due to its high contents of bicarbonate, Ca and Mg which easily create insoluble compounds causing clogging of the irrigation system. Insoluble Ca and Mg compounds will not be formed if pH of irrigation water is kept continuously on the acidic side. Therefore farmers normally add acids to irrigation water in order to reduce water pH. This involves additional treatment with hazardous liquid acids which are difficult to handle and store. In addition, the bicarbonate ion can be toxic to plants, but more importantly, it interferes with other nutrients and makes them less available to plants (Patricia et al., 2009). Soil acidification sometimes is necessary for optimum plant growth. Soil acidification is best performed prior to planting; it is much more difficult in the established planting of the solubility

of metallic micronutrients such as iron(Fe), zinc(Zn), and manganese(Mn) in soil solution because it decreases rapidly as soil pH increases, making these elements less available to plants (Horneck and Abak, 2004; Emadi et al., 2012). Soil application of H_2SO_4 has been shown to increase Fe availability, theory eliminating chlorosis in sorghum and to increase P availability in some calcareous soils. Available information suggests that application of H_2SO_4 in irrigation water is relatively ineffective for evoking a nutritional response in crops. Iron plays essential roles in the metabolism of chlorophylls. External application of Fe increased photosynthesis, net assimilation and relative growth in seawater-stressed rice (Sultana *et al.*, 2001). This is especially true for soils of high pH where equilibrium conditions favor the oxidation of plant-available Fe^{+2} to unavailable Fe^{+3} . Plant yield on many soils is, therefore, limited by poor Fe availability, rather than a low Fe content in the soil. Also Fe leaching is the main pathway for Fe loss in coarse-textured soil with high pH, while excessive Fe uptake was the main pathway for Fe loss in clay-textured and acid soil. Application of Zn or Fe has been reported to have significant positive effects, in most cases, on growth measurements and chemical composition of safflower (Lewis and McFarlane, 1986), wheat (Lu *et al.*, 2004), common bean (Fernandez et al., 2007) and rice (Wissuwa *et al.*, 2008). Iron deficiency is quite commonly observed in calcareous soils in all the important field crops. Deficiency of Fe manifest into yellowish inter-venal paling of younger leaves (commonly referred as iron chlorosis). In general, the plants are prone to iron deficiency in alkaline, calcareous, coarse textured, eroded and low organic matter containing cold region soils. Recent research has shown that a small amount of nutrients, particularly Zn, Fe and Mn applied by foliar spraying increases significantly the yield of crops (Sarkar *et al.*, 2007; Wissuwa *et al.*, 2008). Among the micronutrients, Fe nutrition can affect the susceptibility of plants to drought stress (Sultana *et al.*, 2001; Khan *et al.*, 2003; Cakmak, 2008). Further, application of iron salts to soils recovers plants from iron deficiency and compensates for the loss in the yield (Singh *et al.*, 1995). Seed coating with Fe-EDDHA has also been successfully used and shown to increase seed yields (Wiersma, 2005; Penas *et al.*, 1990; Karkosh *et al.*, 1988; Shadkami-Til, 2013a; Shadkami-Til, 2013b). In contrast, Goos and Johnson (2000) did not observe yield increase with Fe-EDDHA seed treatment. However, to the best of our knowledge, information regarding application method efficiency of iron on the growth and development of bean is not available. Therefore, the purpose of this study is to understand whether application of Fe as foliar application is better than soil application.

MATERIALS AND METHODS

The chosen soil for complementing this study showed the low level of iron absorption. So the samples were provided from the Zanjan University by composed and from the ground level of the soil (0-20 centimeters). The chosen soil passed through a millimeter sieve. The results of the applied soil in this test are shown in table 1.

Cu	Zn	Mn	Fe	K	P	Mg/kg
1.04	2.02	1.22	0.68	490	2.5	

Soil texture	Organic carbon percentage	pH	($\frac{ds}{m}$)	saturation percentage
loam	1.14	7.67	1.27	41.28

In order to study the various methods for removing the iron loss in bean and the effect these methods has on the bean operation on calcareous soil trail 44 treatment and 3 iterate in factorial form and in quietly incidental frame at the green house were implemented. In this test sulphoric acid in irrigation water (2 and 4 g/L), soil application of Ferrous sulfate (75 and 150 kg/ h), soil application of Fe-EDDHA (3 and 6 mg/ kg) Ferrous sulfate, spray by density (2 and 4 g/ L) and Fe- EDTA , spray with (1 and 2 g/ L) are included. There was also an evidence treatment (zero level). This test had 44 treatment which were implemented in the 3 iterates and there were 132 trill Crete totally. Exertion of thial treatments was performed in the plastic pots soil and there was 3 kg soil for each pot. The exertion of sulphoric acid treatment was included in the watering stage and it was used by the water. pH of watering reduced from 7.3 to 1.701 and 1.90 respectively with (2 and 4 g/ L) density. Ferrous sulfate cure and Fe-EDDHA dissolved in distilled water and equally sprayed and added in the pots soil. Iron spraying with Fe- EDTA and Ferrous sulfate tried primarily out of the trail design area on the bean leaves and after assuring the luck of related crates dry matter in two stages, it exerted for the trail beans in 18 days intervals before their blooming. The first stage of spraying performed at 10th of April so that the plant is moisture completely and the second stage was implemented after 15 days. Spraying performed at the evening, one hour before sunset in the weather without precipitation and wind. Beans seeds contained four types of khomein; Talash, Local and COS 16 supplied from Khodabandeh Township and cultivated on March 6, 1999. In each pot 3 bean seeds are cultured in 5 cm depth to the soil level and they are watered after culturing. The consumed amount of water used at the limit can reach the soil moisture up to the field capacity. Watering pH reduced from 7.3 to 1.70 and 1.90 after adding sulphoric acid with (2 and 4 g/ L) density, determination of yield and its

operation (contain weight of 100 seeds ,number of sheaths in one bush , and the number of seed , in a sheath) is accomplished. Elements with less application in clouding zinc, iron, manganese and copper were measured by using of atomic absorption device.

RESULTS AND DISCUSSION

Yield

Results obtained from the data variance analysis table demonstrates that the effect of types, sources and various ways of iron fertilizer consumption on the yield property, weight of 100 seeds, numbers of the seeds in the bush and the numbers of seeds in the sheath by %1 probability level is meaningful (Table 2).

Table 2: Results of variance analysis of sources effect and various ways of iron fertilizer consumption on the yield and operation elements on the spotted beans types.

Changes sources	Deliver grade	The auarage of squares			
		yield (gram in the bush unit)	Weight of 100 seeds (gram)	Numbers of sheet in a bush	Numbers of the seeds in the sheath
Type	3	2044/42499**	1102/862889**	76/85000**	68/46481**
Fertilizer	14	1064/84405**	28/882008 ^{ns}	34/25556**	41/45794**
Treatment	42	52/02815**	20/239437 ^{ns}	5/33810 ^{ns}	27/64735 ^{ns}
Type * Fertilizer Treatment	120				
Error					
C.V (percent)		7/63	9/04	4/38	4/40

** / * / ns respectively the yare meaningful at %5 level, one percent and non meaningful

Comparison of the yield averages of spotted beans types shows that there is meaningful difference between types from this property viewpoint. Talash and Local types have the highest yield and difference of these two types with other types was meaningful (figure 1).

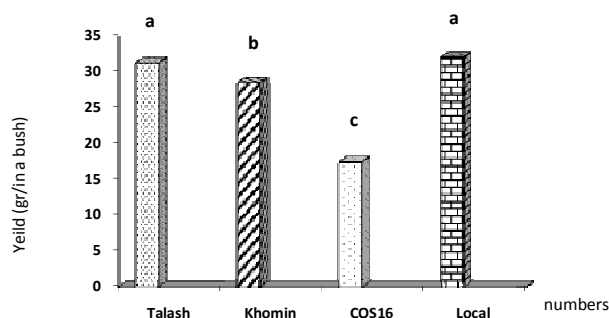


Figure 1: The effect of spotted bean types on the seed yield.

It seems that exceeding the yield of Talash and Local beans type is because of the mote Production of sheath numbers and the seeds in the sheath in comparison to khomein and COS 16. Khomein and COS 16 are among the types that are sensitive to iron lack in calcareous soils condition, so they show the lowest yield. The some results are proved by Godsey *et al.* (2003). The effect of sources and various kinds of iron fertilizer consumption on the seed yield amount was meaningful (figure 2).

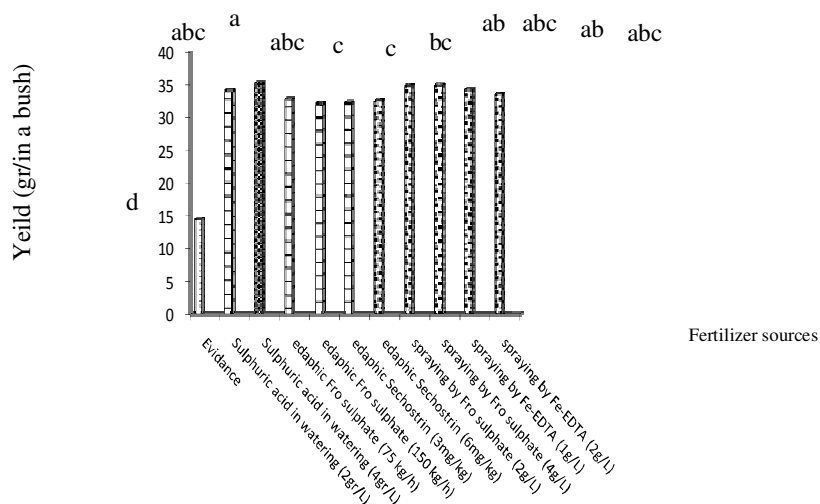


Figure 2: The effect of various fertilizers on the seed yield.

The most amount of the seed yield relates to treatment of **sulphuric** acid in irrigation water (4 g/L) and the lowest amount relates to evidence treatment. In calcareous soils, elements accessibility increases by the reduction of soil pH- Acidizing of the watering is a suitable way for reducing soil pH (Donald and Billore, 2007). Acidifying the irrigation water improves the uniformity of pH in field and enhances the early establishment and growth of the plants (Patricia *et al.*, 2009). Patricia *et al.* (2009) reported that the yield of agriculture products increases by reduction of soil pH in the calcareous soils on the biotypes of chickpea which causes increase in yield. Kari *et al.* (2004) studied the edaphic and phyline effect of Fe- EDTA on the barley in calcareous soils and the result of their study showed that edaphic and phyline application of Fe- EDTA Khomein and Local type beans have higher average range than the other types from the numbers of 100 seeds (figure 3).

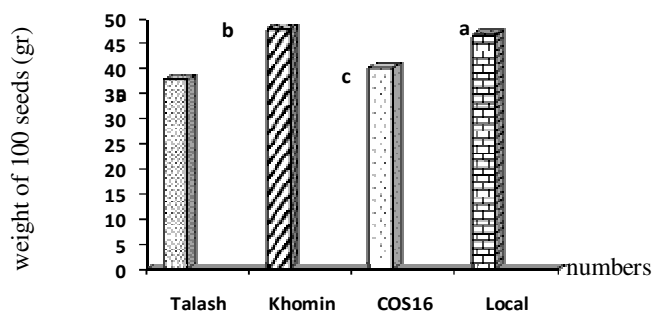


Figure 3: The main effect of pinto bean on the weight of 100 seeds.

Weight of 100 seeds is one of the efficient factors on the Grains seeds and relates to efficiency of photosynthesis organs and carbon oscillation during the filling of the considered seed. Morgan *et al.* (2002) reported that biotypes of bean differ greatly with each other regarding to weight of 100 seeds. There was not meaningful difference between various fertilizer sources and the evidence, for the weight of 100 seeds. Infact, application of various fertilizer sources doesn't have effect on the weight of 100 seeds and it seems this property is under the influence of genetic factors of the plant. Weight of 100 seeds showed positive and meaningful conjunction. Positive conjunction between the weight of 100 seeds and seed yield reveals that the seeds yield increases by increasing the weight of 100 seeds. Talash type possesses the highest average from the number of sheath in a bush and number of the seeds in the sheath viewpoint, and it has a meaningful difference regarding to these properties with other types (figure4).

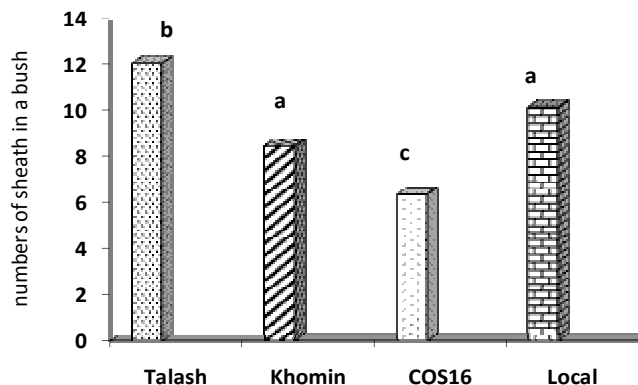


Figure 4: The main effect of spotted bean on numbers sheath in a bush.

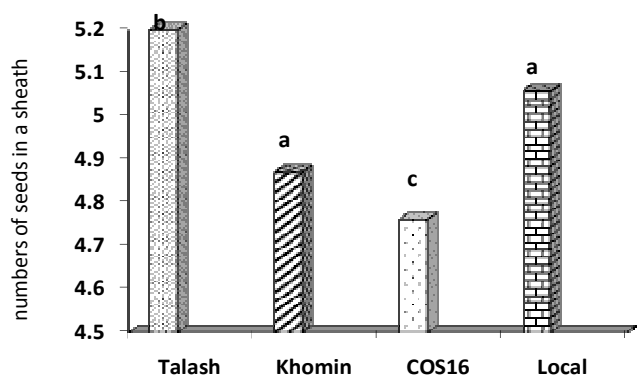


Figure 5: The main effect of spotted bean on numbers of seed in a sheath.

Grain number in each spike is one of the most important yield components; it has the (Ghorbani and Hamid, 2009) most effect on yield (Mahmoudi et al., 2013). From the property of numbers of the sheath in a bush, treatment of **sulphoric** acid in watering by the amount of 4 g/l has higher average than the other fertilizer sources (figure 6).

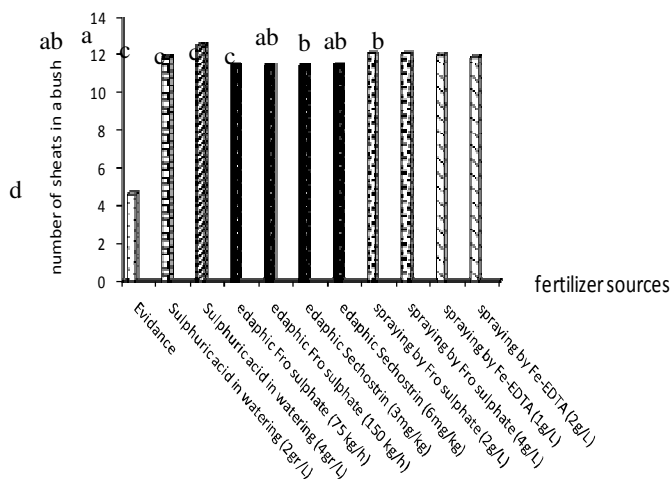


Figure 6: The effect of various fertilizer sources on the numbers of sheaths in a bush.

In the study which is accomplished on the effect of watering iterates and application of acid in watering for the bean, it is observed that negative specifications (Dry weight, height, number of the branches) and non-negative (numbers of in a bush), increased by the extra iterates of watering and acid application in watering (Jendy et al, 1995). Treatments of spraying by sulphate and Fe-EDTA had no meaningful difference regarding to the average of the sheath numbers in a bush and both treatments have caused the increase in the sheath numbers but they have lower average in comparison to the treatment of acid consumption in the watering. Lingenfelter et al (2005) in a research by removing the chlorosis due to the iron back and spraying Ferrous sulfate reported the yield increase with the increase of seed amount. Treatments of Ferrous sulfate and Fe-EDDHA have lower average than the spraying treatment but both of them cause increase in the number of sheath in a bush toward the evidence ratio. Wiersma (2003) reported the increase of soybean yield due to the increase in the seed numbers in a bush, using Fe-EDDHA in various amounts.

sulphoric Acid in irrigation water by the ratio of (4 g/L) has a higher level of average than the other sources from the property of numbers of seed in the sheath, and there is a meaningful difference between the various fertilizer source (figure 7).

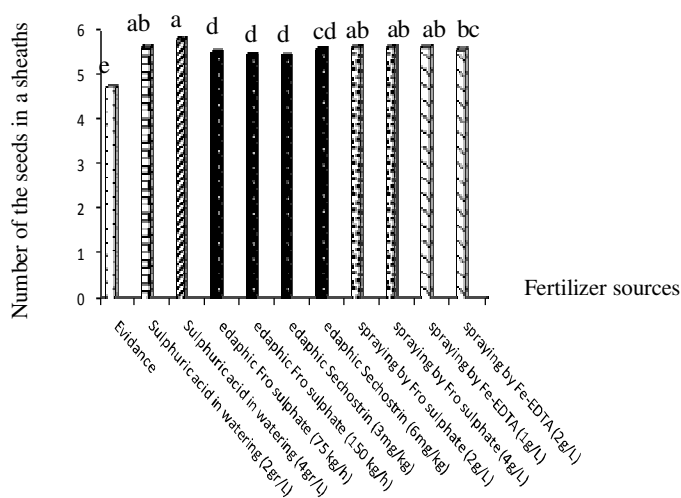


Figure 7: The effect of various fertilizer sources on the numbers of seeds in a sheath.

By considering that the numbers of seeds in a sheath is one of the related factors to the growth, so any factor which increases the yield, in effects increases the growth. Ferrous sulfate and Fe-EDTA treatments in spraying form have lower average than the acid consumption in watering treatment but they caused increase in the seeds numbers of a sheath comparing to evidence. Hegazy et al. (1993) observe that maximum yield of the bean seed is obtained by spaying ferrous sulfate, zinc, and manganese with %2 densities after plant blooming.

Chemical Element of the leave

The effect of various types and ways of iron fertilizer consumption were also meaningful on the iron, manganese, zinc and copper elements of the leave in %1 probability level (table 2).

Table 2 continuation: Results of variance analysis of the iron various treatments ' effect in the spotted bean and on the chemical compound of the leave

Changes sources	Deliver grade	Leaf's iron	Leaf's manganese	Leaf's zinc	Leaf's copper
Type	3	5835/368**	0/509259 ^{ns}	8/95062 ^{ns}	7/710908 ^{ns}
Fertilizer Treatment	14	587192/942**	556/166448**	1282/80475**	434/226364**
Type * Fertilizer Treatment	42	2723/032**	2/294974**	15/12346**	4/558350**
Error	120				
C.V (percent)		0/78	7/75	4/94	6/07

*** n s is meaningful respectively in %5, %1 and non meaningful setting.

There was meaningful difference between the numbers compared to the amount of leaf iron and Talash type has the higher- level average than the others (figure 8).

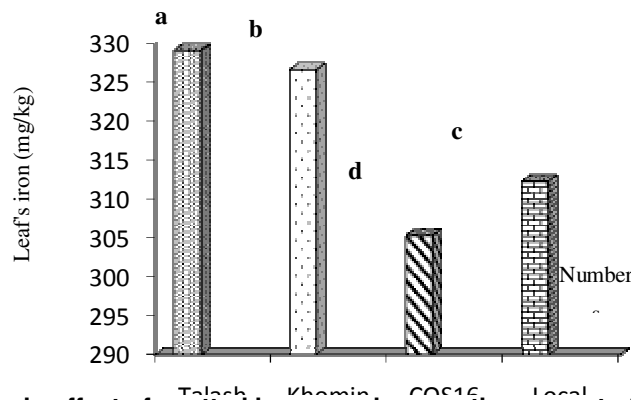


Figure 8: The main effect of spotted bean numbers on the amount of leaf's iron.

Raising the iron amount of the leave in Talash Type compared to other types shows the capability and ability of reduction of Iron (3) to Iron (2) in its roots. The results coincided with Godsey et al. (2003) report that edaphic factors and radical system variety are effective for absorption of the elements by the biotypes. The same results are also proved by Vempati et al. (1988). The effect of applied fertilizer sources caused iron absorption. Spraying with Fe- EDTA up to (1 g/L) had the highest average of leaves iron, and there was meaningful differences between the fertilizer sources in this regard (figure 9).

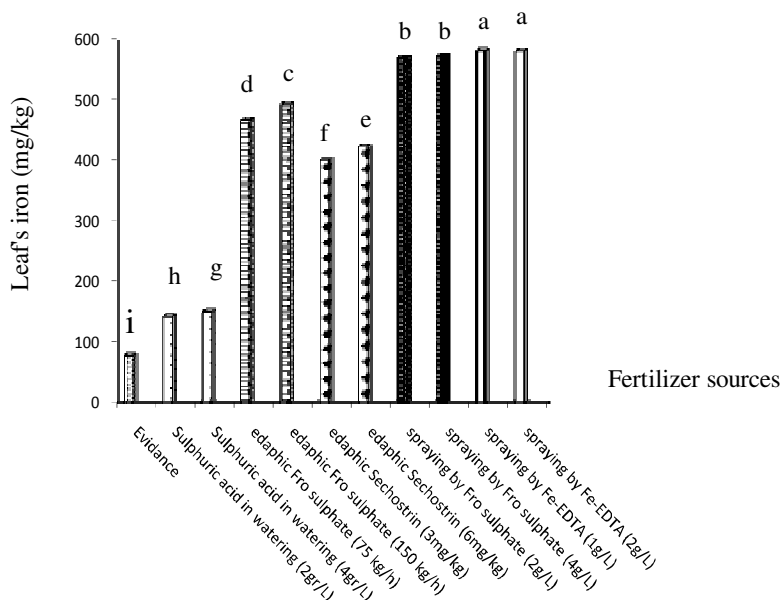


Figure 9: The effect of various fertilizer sources on the leaf's iron.

The application of Fe- EDTA increased young plant leaf iron concentrations by 10% when compared to three-year- old plants treated with and without iron chelates (Natt, 1992). Spraying iron fertilizers create higher density of iron compared to its edaphic usage. Because of the present of calcium carbonate in the soil, capability of nutrient elements specially iron reduces and causes chlorosis on the leaves. Application of iron fertilizer sources by spraying is more effective for rapid iron chlorosis and its absorption through leaves than its edaphic application (Tisdale et al., 1984) iron sulfate added to calcareous soils which quickly reacts with CaCO₃ to form Fe oxides that are less available for plant uptake (Godsey et al., 2003). Lack of less application element especially iron is

harmful for the function and quality of agriculture products (Ghorbani and Hamid, 2009) although soil application of iron sulphate is not recommended, especially in calcareous soils; the data revealed that it could be used successfully as a foliar spray (Modaihsh et al., 1996). Han *et al.* (1994) Reports foliar application of Fe- chelates was shown to be more efficient than soil application because of the direct uptake of Fe by the plant through cuticle pores from the leaf surface. Foliar application can correct Fe chlorosis in bean, sorghum, peanut, rice and other fruit crops. Using optimal foliar Fe fertilization application in a field experiment, Fe content of rice was significantly increased by 37.1%, compared with no foliar application, without affecting grain yields, protein and ash content of rice produced. Study of in a limy soil, investigated the effect of iron fertilizer sources on the granular broom corn at field conditions. The results demonstrated that spraying Fe-EDTA and iron sulphate caused increase in the leaves iron. Erdal (2004) by the application of Fe- EDTA and FeSO₄ in spraying from the strawberry at three stages revealed that both iron sources caused increase in the iron density at the leaves. Roomizadeh and karimian (1996) reported that iron application in Fe- EDDHA from, in all tested calcareous soil meaningful increased the iron density in soybean. Lowering the pH of the soil often improves availability on iron, and other micronutrients on calcareous soils. Acidifying water also helps to improve water penetration and alleviates some of the stress of soda soils. Horneck et al. (2004) by acidizing the soils found that dissolubility and absorption of less application elements such as iron reduce by reduction of pH. There was no meaningful between the numbers regarding to the amounts of leaves geneses, zinc and copper. Treatment of sulphuric acid in irrigation water had higher level of leaf manganese average compared to evidence cure and the other fertilizer sources. Results showed that **sulphuric** acid application in watering caused increase in the density of leaves manganese while the application of iron sources whether in soil or spray form caused reduction in the leaves manganese density (figure 10).

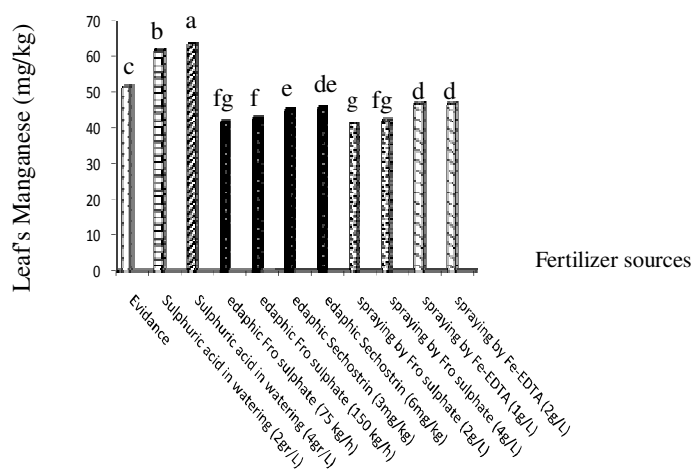


Figure10: The effect of various fertilizer sources on the leaf's manganese.

In calcareous soils dissolubility and access to less application elements reduced. One of the reduction ways of pH is applying acid in watering which causes dissolubility of less application elements (Stevens and Poorte, 2006). Factors which influence the manganese dissolubility, oxidation and restoration conditions of soil, determine the manganese absorbance capability for plants. Soil pH is among the most important factors. Reduction of soil pH causes increase in manganese compounds dissolubility which leads to the increase in its absorbance capability. Donald and Billore (2007) reported that acidizing the water was a costly and time consuming process with soil pH reduction and less application element absorption increase. Kalbasi *et al.* (1988) by the study of sulphur application in manganese absorption at three plant in clouding maize, surgme and soybean in the same argillaceous soil with 40% lime and in the field test showed that sulphur consumption caused a meaningful reduction pH and Bicarbonate density of soil and increased the absorption manganese of the plant. Reduction in manganese absorption due to the iron fertilizer application may be contributed to the same places of iron and manganese absorbance on the root level (Schenkedeld, 2008). Probably iron and manganese compete in the absorption processes and its transfer from the root to the air organs (Malakouti, 2000) and this competition causes the reduction of manganese density in all treatments that contain iron fertilizer except for the **sulphuric** acid in watering cure, compared to the evidence. Ghasemi et al. (2008) by using the various iron sources on the soybean found that iron application in soil form caused reduction of manganese in the root and leaves and which of course the result showed that the percent of manganese reduction differs between the varieties. Karaman et al. (1997) through their researches suggested that application of various iron sources

caused manganese reduction in the leaves. Luccena et al. (1992) demonstrated that, although iron schostrin is the most suitable iron fertilizer in calcareous soil, but it should be noted that much application of this application could lead to incidence of manganese loss in soybean. Roomizadeh and karimian (1996) also reported a reduction of manganese density due to the iron application. The highest level of density average on the leaf related to sulphuric acid in watering (2 g/L) which caused increase of zinc absorption on the plant compared to evidence, but other fertilizer source had a reduction of zinc absorption regarding to evidence (figure 11).

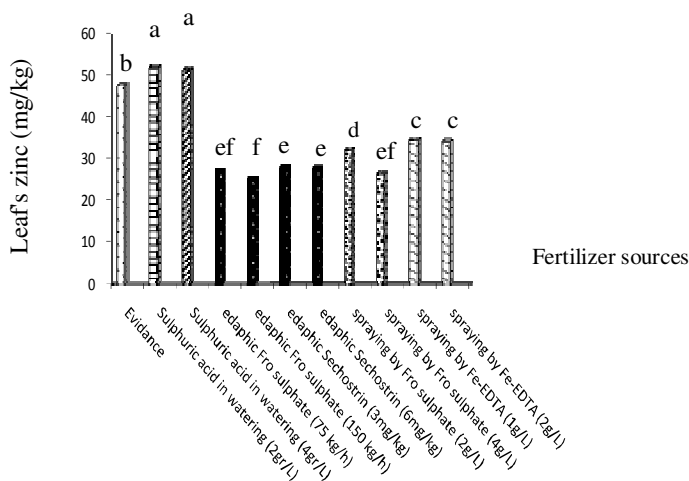


figure11: The effect of various fertilizer sources on the leaf's zinc.

Presence of sulphuric acid in soil caused the reduction of soil pH and the soil absorbance of zinc to increase (William and Keatinge, 2006). Bioavailability of all micronutrients is significantly affected by soil pH, decreasing with increasing soil pH. The activity of Fe, Mn, Cu and Zn decreases for each unit increase in soil pH. As illustrated in (figure 11) it is observed that various iron sources caused zinc reduction on the leaves. The reason of zinc density reduction due to the various iron sources consumption could be the result of iron and zinc competition in absorption by the plant. Negative interaction between iron and zinc could be a reason for zinc density reduction due to the iron consumption. Negative interaction between iron and zinc is reported by many researchers including Malakouti et al. (2000). Antagonism between Fe and Zn is well known. Previous studies have shown that Zn interfered with Fe uptake and translocation, whereas Fe interfered with Zn translocation, but only when Zn concentrations were high (Alloway, 2008). There are three possible mechanisms for this antagonism. First, there could be competition between Zn^{2+} and Fe^{2+} during uptake. Second, there could be interference in the chelation process during Fe uptake and translocation (Kabata-Pendias, 2001). Third, there could be competitive inhibition between Zn and Fe during unloading in the xylem (Alloway, 2008). The highest level average of leaf's copper density related to **sulphuric** acid consumption cure in watering up to 2 g/L. **sulphuric** acid treatment in irrigation caused increase of copper amount of leaves while other fertilizer sources caused the reduction of the copper amount of leaves as compared to evidence (figure 12).

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