

# **A Breakthrough in Plating for Solar Cell Metallization**

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# Agenda / Outline / Overview

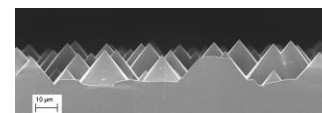
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- **Introduction**
  - Solar cell and cost pressure
- **History and State of the art**
  - **Plating Techniques**
  - **Technology Key Issues of Plating in Solar Cell**
  - **Localized Plating**
- **Our Approach** (A Breakthrough in Plating for Solar Cell Metallization)
  - **DLD/DLM**
    - Main results achieved
    - On going work (H2020 AMPERE Project)
- **Acknowledgement**
- **Conclusions**

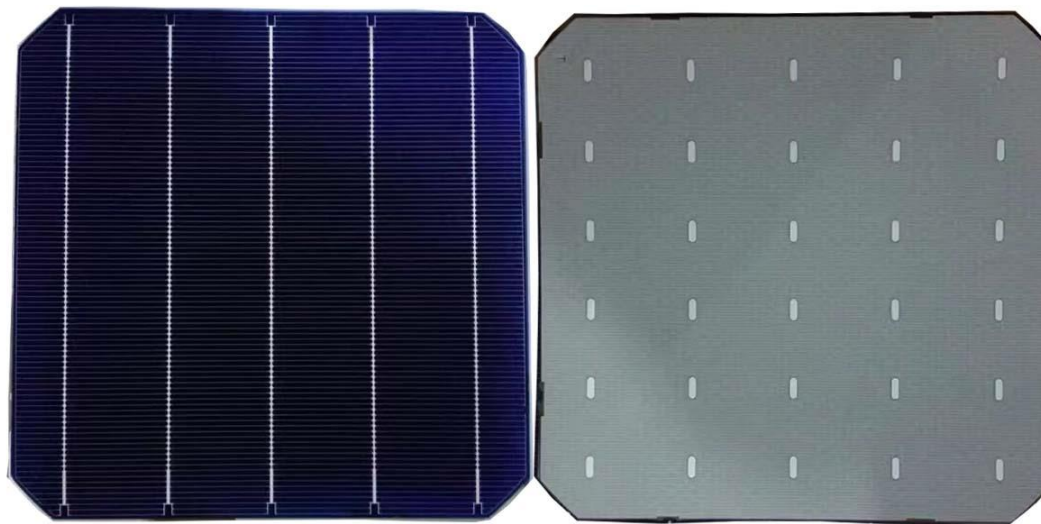
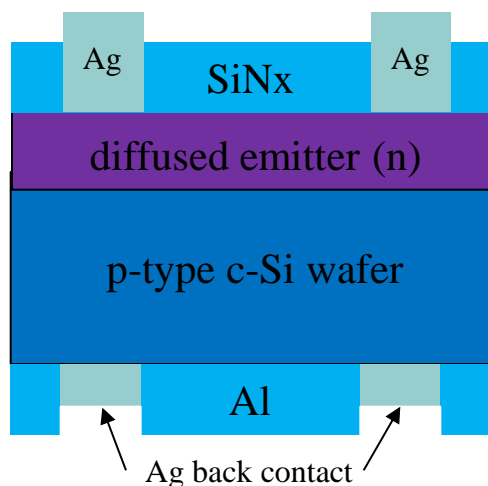
# Introduction

## ● Solar Cell

World wide production capacity > 110GW



A solar cell, or photovoltaic cell, is an electrical device that converts the energy of light directly into electricity by the photovoltaic effect, which is a physical and chemical phenomenon. It is a form of photoelectric cell, defined as a device whose electrical characteristics, such as current, voltage, or resistance, vary when exposed to light.



ITRPV 2018

Al-BSF

75%

$16.0 < \eta < 20.0\%$

- (2028 ~ 7%)

Market Share

Efficiency

trend for such technologies

PERC/PERL/PERT

22%

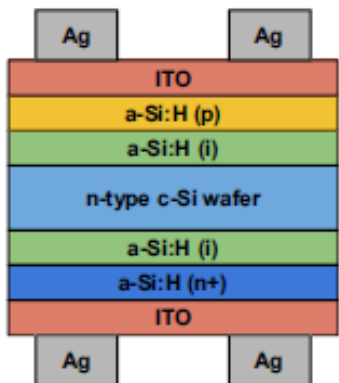
$19.0 < \eta < 22.5\%$

+ (2028 ~ 60%)

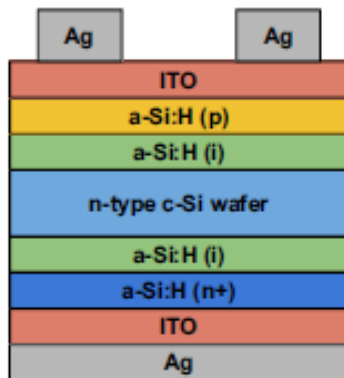
# Introduction

## ● Silicon HeteroJunction (SHJ) Solar Cell $\eta > 22.0\%$

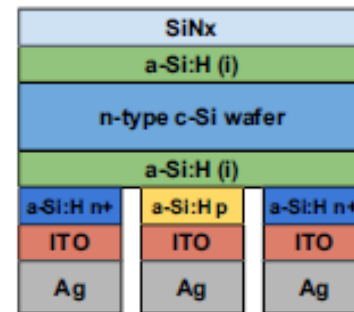
Ref. SHJ  
 $\eta \geq 22.4\%$



PVD SHJ  
 $\eta \geq 22.7\%$



IBC SHJ  
 $\eta \geq 23.3\%$

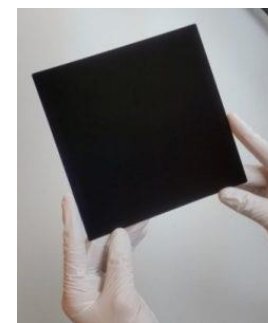


$\eta_{record} = 26.63\%$   
size (180 cm<sup>2</sup>)



April 2020  $\eta = 22.5\%$  bifacial cell  
at industrial level full scaled automated cell line 100MW

Look to solutions for  $\eta = 23.5\%$

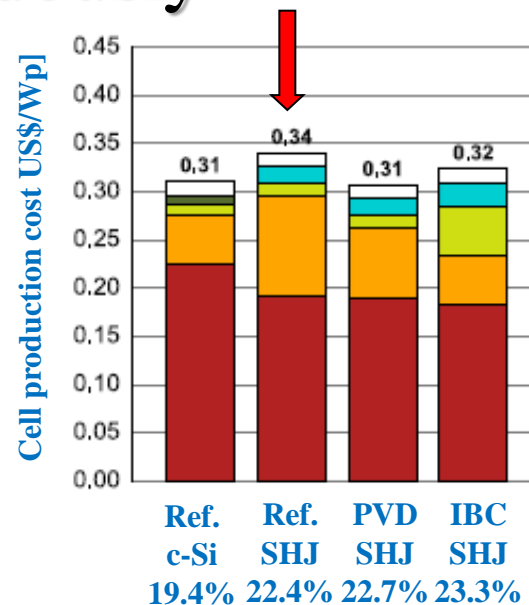
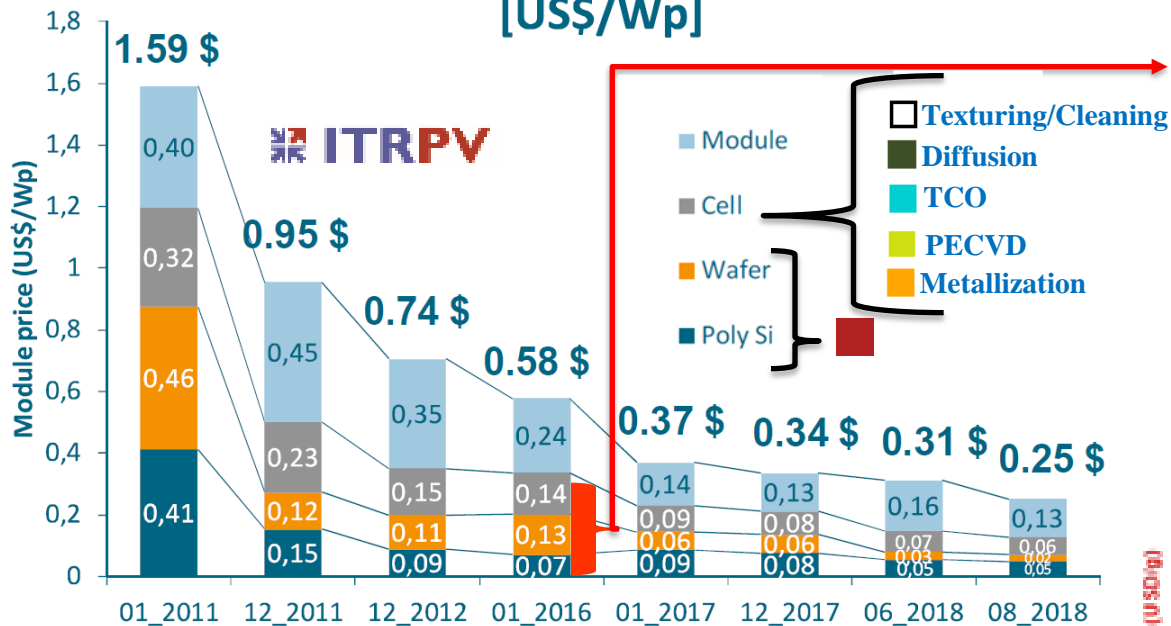


Kaneka Corporation

# Introduction

## ● Cost Pressure Increasing Tremendously in SC

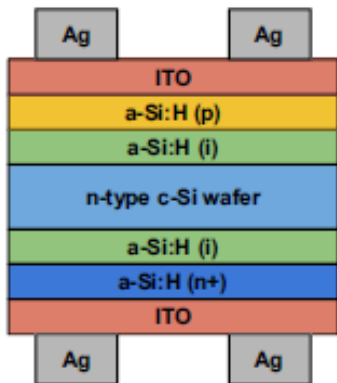
Module price break down [US\$/Wp]



15% < Metallization influence cost (\$/Wp) < 30% on Solar cell production cost

# Introduction

- Ag \$/g influences Tremendously solar cell cost



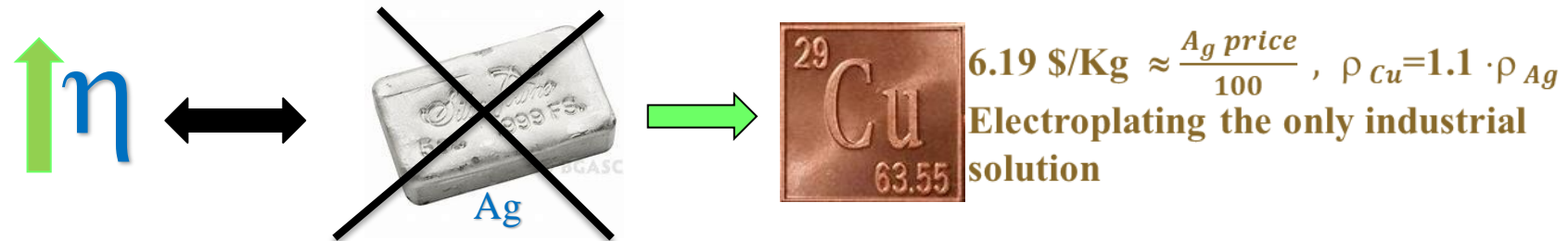
Low temperature Ag paste  $\approx 0.7\$/g$   
 Sheet resistance post curing is not bulk Ag: (Temp  $< 200^\circ \text{C}$ )

$$5 \frac{m\Omega}{\square} < R_{sq} @ 25\mu m < 25 \frac{m\Omega}{\square}$$

$$8.5 \cdot \rho_{Ag \text{ bulk}} \qquad \qquad \qquad 42.5 \cdot \rho_{Ag \text{ bulk}}$$

Specific contact resistance  $1m\Omega \cdot cm^2$  to  $4m\Omega \cdot cm^2$

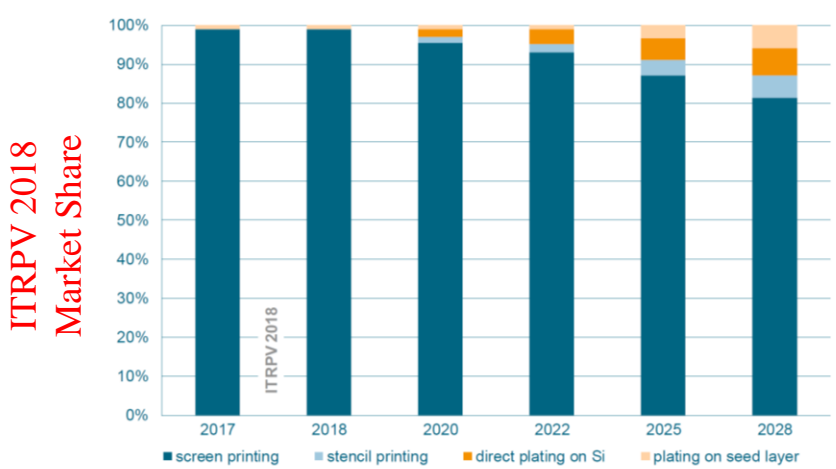
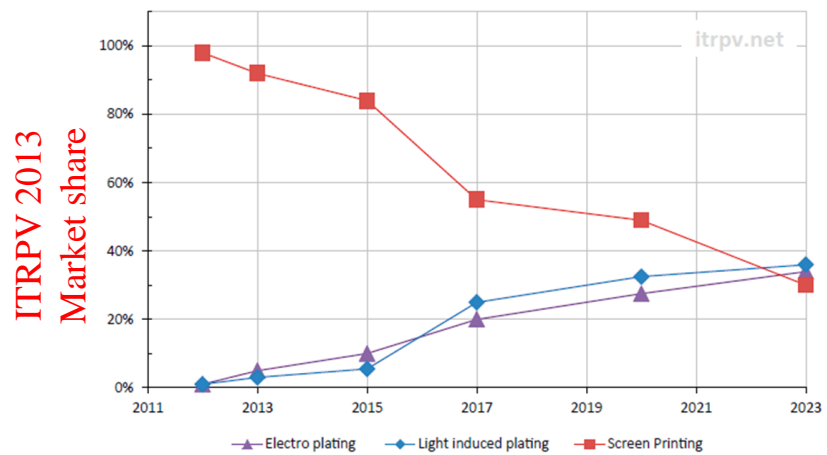
Bi-facial SHJ Ag/cell (5BB)  $\approx 0.2g \rightarrow 0.14\$/cell \xrightarrow{22.5\%} 0.073 \$/W_p \rightarrow 0.34\$/W_p \rightarrow 21.4\%$



# History and State of the art

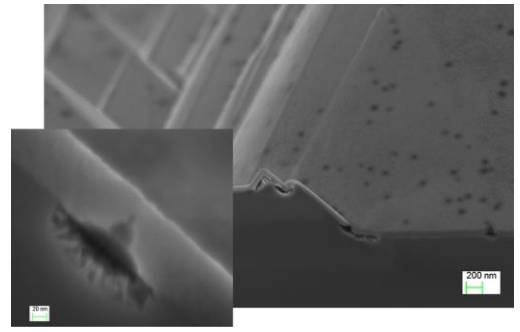
## ● Plating Technique in Solar Cell

Before the introduction of alternative metallization techniques, technical issues in reliability and adhesion have to be solved. Appropriate equipment also needs to be available. **(ITRPV)**



## ● Plating key issues

- Copper not directly in contact with silicon!!!! → Need Barrier Layer
- Pin-holes and Scratches
- Adhesion (> 1 N/mm)
- Speed of Plating → Throughput, Space floor and Chemical quantities
- Drag-out → Reduce additive consumption





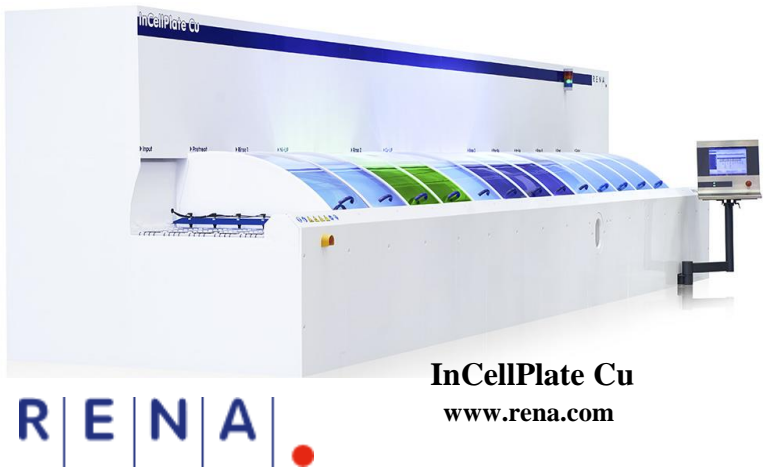
# State of the art

- Industrial Plating for Solar Cell
  - LIP or LAP

BE Semiconductor Industries N.V.



**Meco Direct Plating Line (DPL)**  
([www.besi.com/products-technology/productgroup/plating/com](http://www.besi.com/products-technology/productgroup/plating/com))



**InCellPlate Cu**  
[www.rena.com](http://www.rena.com)





# State of the art

## ● Industrial Plating for Solar Cell

- Plating seed layer



Meco CPL - Plating on seed layer  
<https://www.besi.com/products-technology/product-details/product/meco-cpl/>

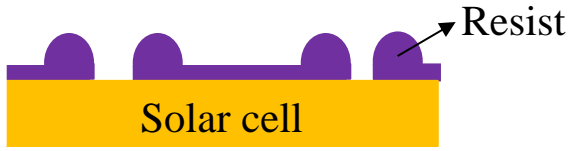
Whatever plating industrial technique you use is necessary to use Protective MASK



PIXDRO JETx P

Inkjet printing equipment for solar cell fabrication

Alternative to printing:  
Apply a film and open it by laser



# State of the art

## ● Plating Issues

- Mask + Plating



≈ Ag

- Adhesion ....to silicon .....to TCO
- Speed of plating.....

Industrial line for 100MWp is:

≈ 30 – 45 m long !!! HUGE FLOOR SPACE!!!!!!

Solution make-up is >1500 liter

!!!!!!

!!!!

Is there a possible solution that could solve all these plating issues



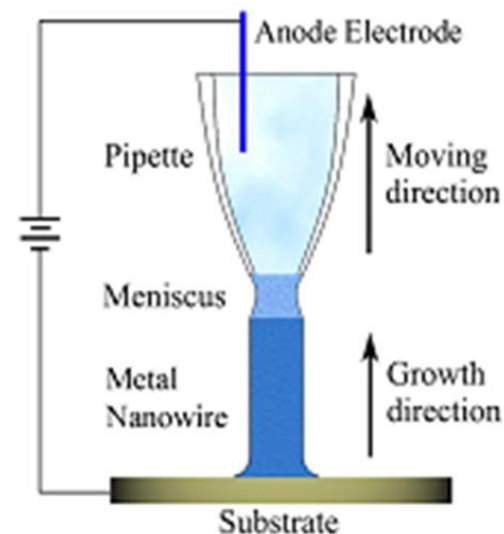
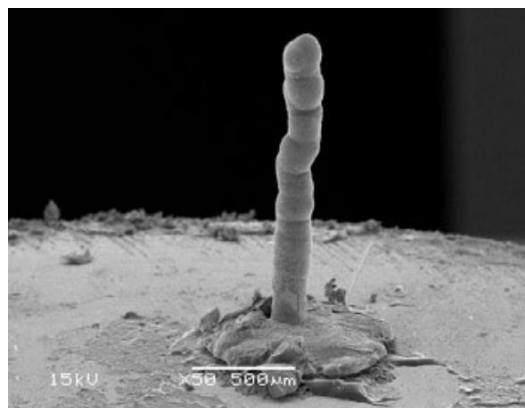
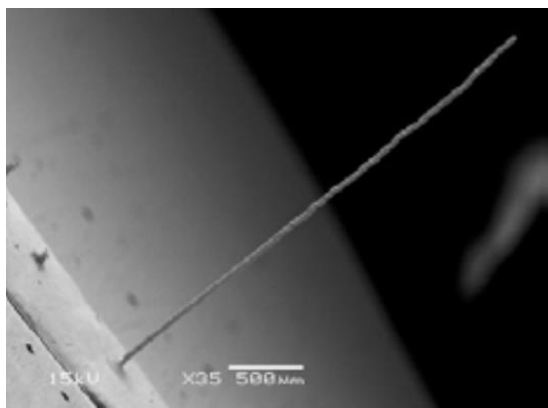
**Selective Plating Process and a Fast process**

**Without using RESIT or PHOTORESIST**

# State of the Art: Localized Plating

## Localized electrochemical deposition (LECD)

### Static Meniscus



### Main Issues:

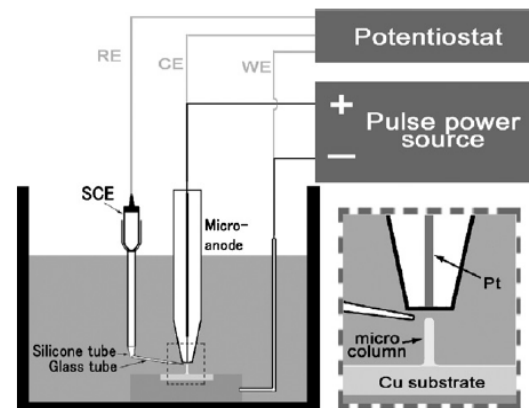
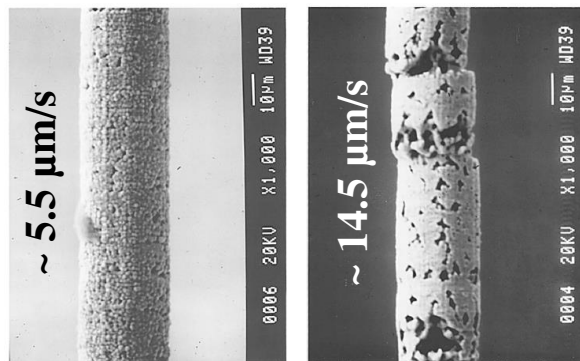
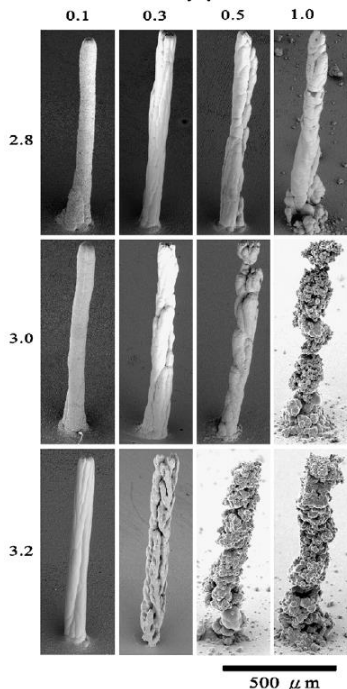
- **No Refreshment of solution**
- **High current density: strong bubbling**
- **Base footprint dimension**

[R. A. Said Nanotechnology 14 (2003) pp. 523–53]

# State of the Art: Localized Plating

## Localized electrochemical deposition (LECD)

### 1,2 Microanode guided Electroplating (MAGE)



Cu deposition rate up to 867 μm/min

#### Main Issues:

- (1) Electrode must be quite close to cathode (i.e.  $< 50 \mu\text{m}$ )
- (2) Very difficult to keep confined the plated area: reduced only for (1)  $< 5 \mu\text{m}$
- (3) Necessary to move the micro-anode
- (4) Porous structure due to very low convection contribution

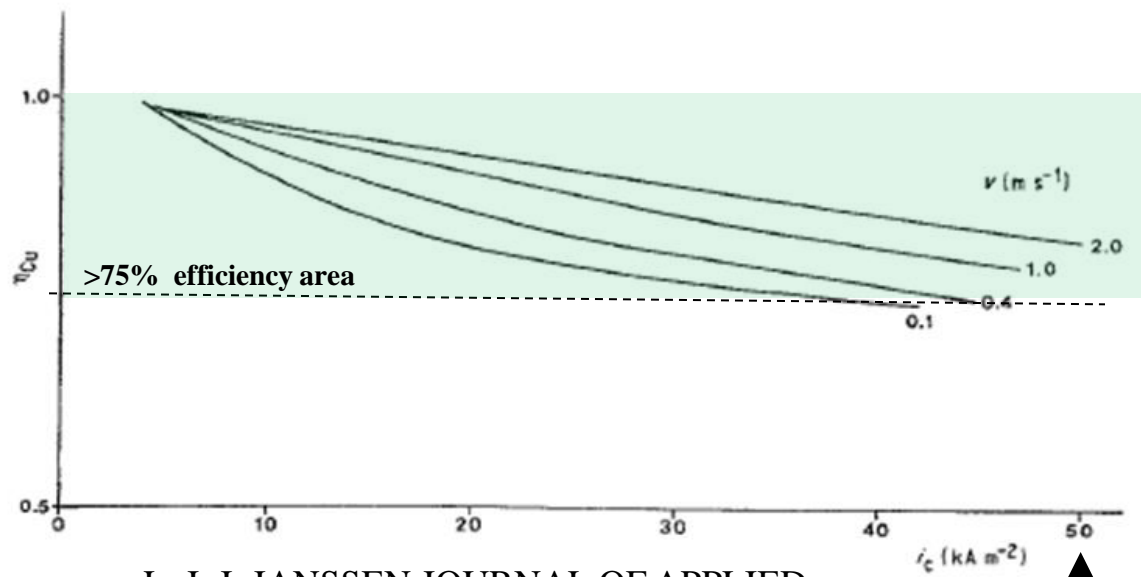
1] E. M. El-Giar et al. Journal of The Electrochemical Society, 147 (2) 586-591 (2000)

2] J.C. Lin et al. Electrochimica Acta 55 (2010) 1888–1894]

# State of the Art: Localized Plating

Mass transport of copper ions in the electroplating process is theoretically governed by the Nernst–Planck equation:

$$F_{ion} = -D\nabla c - \frac{nF}{RT} Dc\nabla\phi + cv \quad [\text{mol s}^{-1} \text{m}^{-2}]$$



L. J. J. JANSSEN JOURNAL OF APPLIED ELECTROCHEMISTRY 18 (1988) 339-346

IBM by Jet plating (1)  
Record of Cu deposition  
**50 μm s<sup>-1</sup>**  
**J=150 A cm<sup>-2</sup>**

**Not porous and very nice morphology**

(1) R. J. von Gutfeld and D. R. Vigliotti, High-speed electroplating of copper using the laser-jet technique, Appl. Phys. Lett. 46, (1985), pp. 1003-1005

↑  
**~2 μm s<sup>-1</sup> Cu deposition rate**

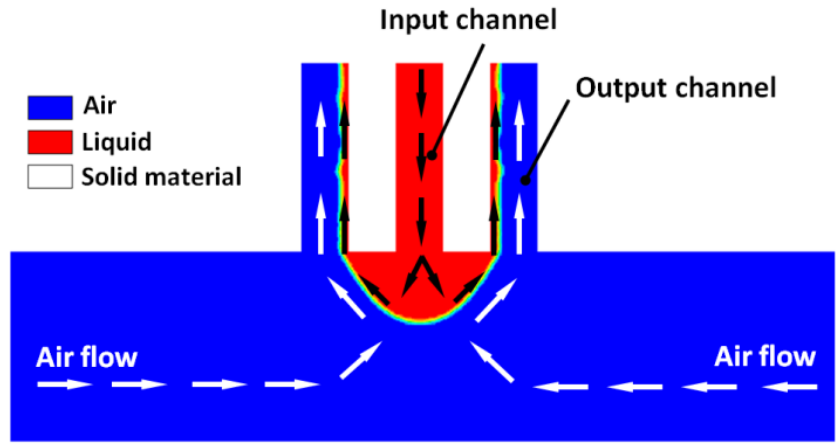
# WHAT WE LOOK FOR

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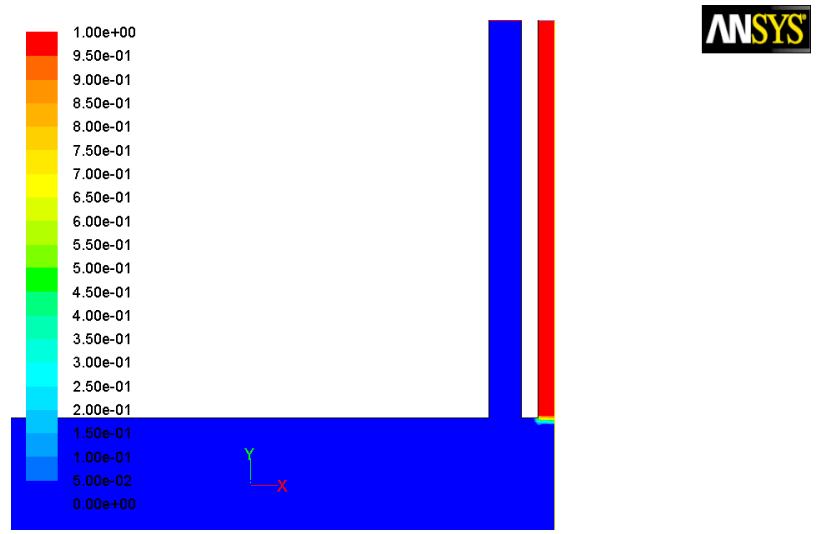
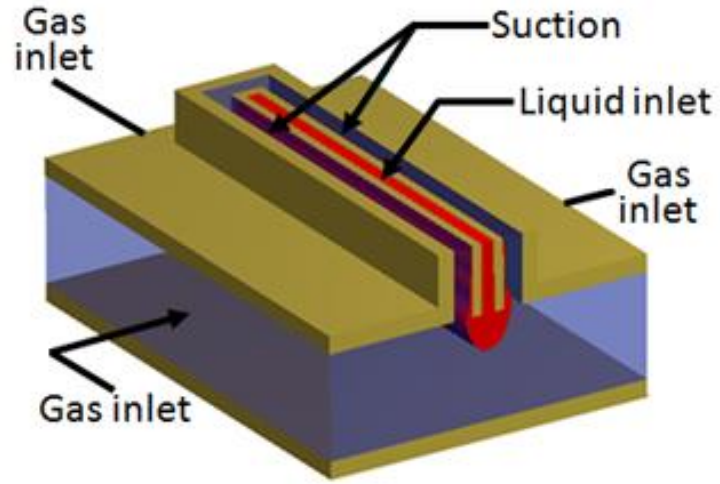
- Don't want to use resist or photoresist
  - Confined plating
- Want to have fast plating regimes
  - Non-porous structure ...as low as possible !!!
  - Low chemical consumption, small footprint and reasonable make-up (<300 litre)
  - Find a process to guarantee adhesion



# Our Approach: DLD



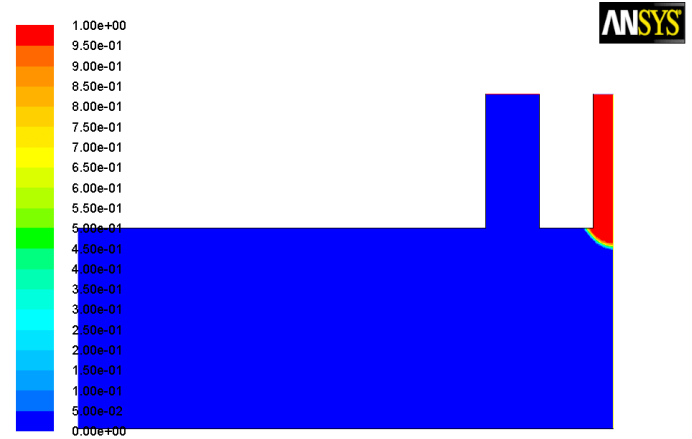
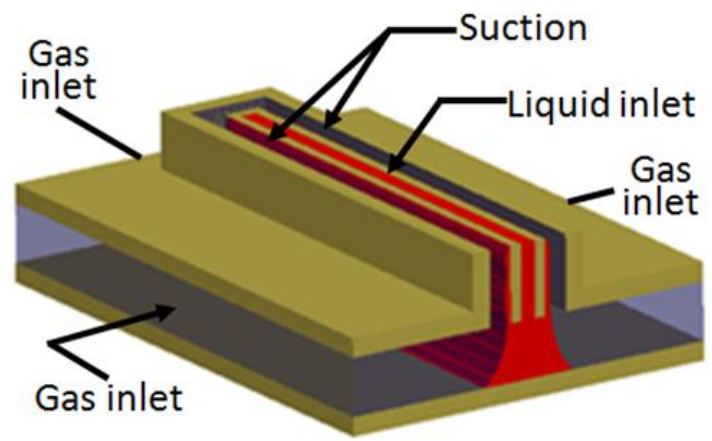
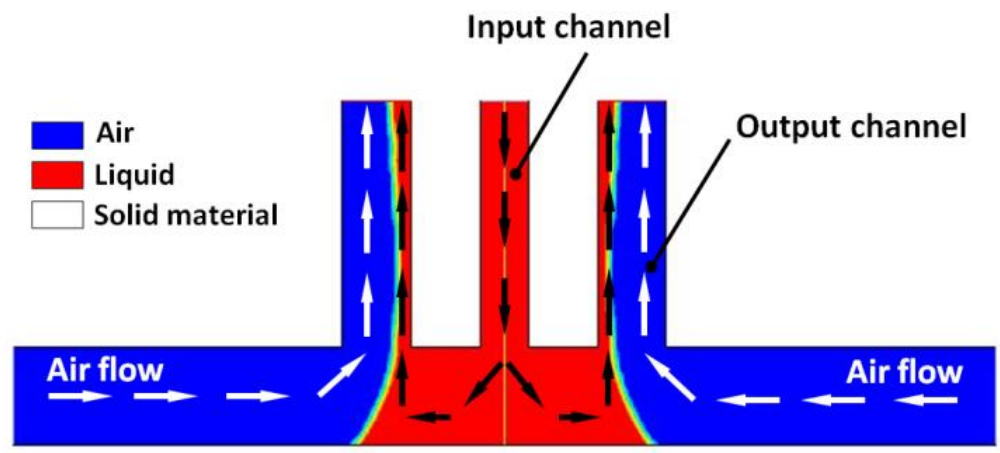
## Dynamic Liquid Drop DLD



Contours of Volume fraction (water) (Time=5.0000e-04)  
ANSYS FLUENT

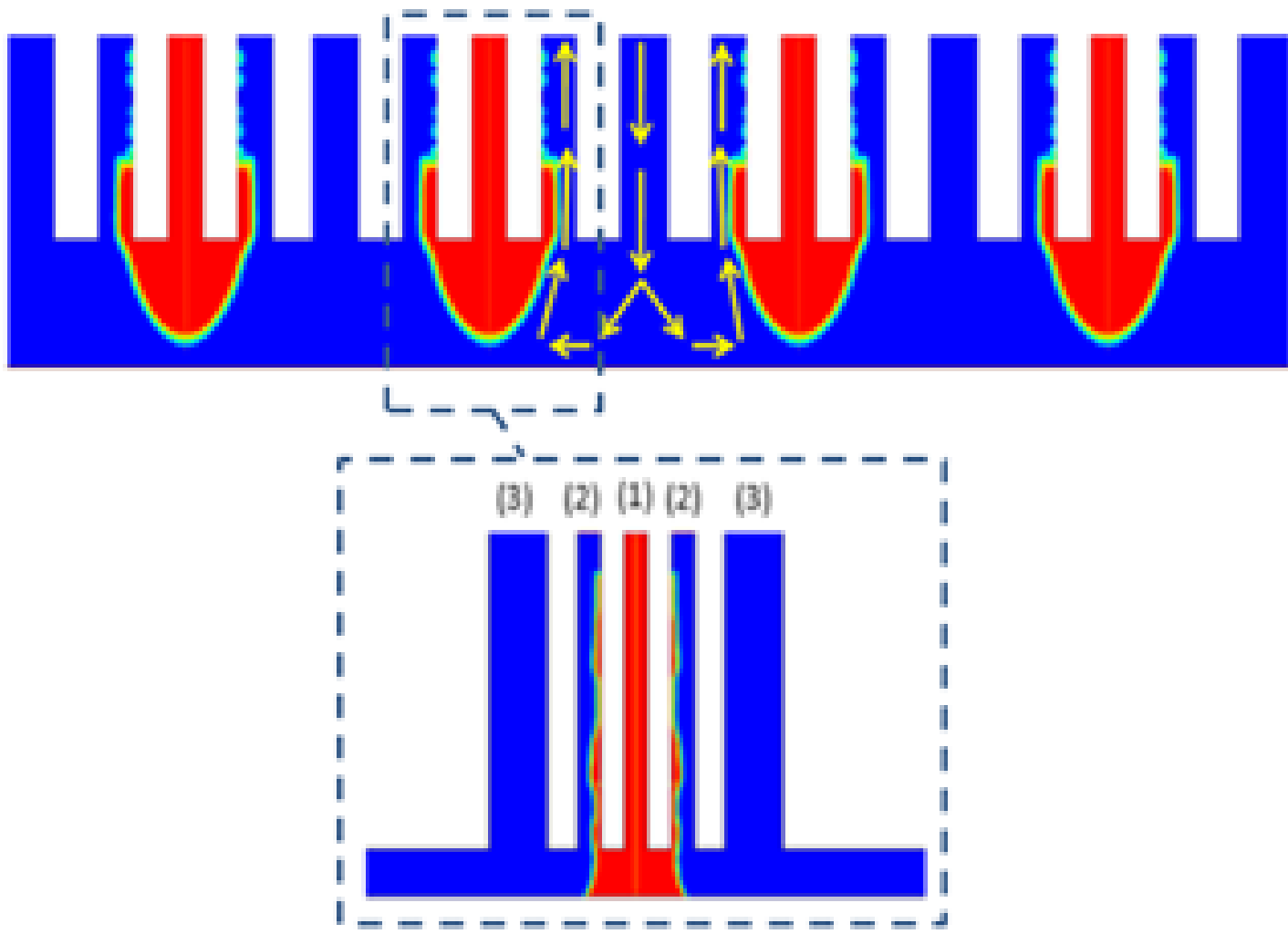
# Our Approach: DLM

## Dynamic Liquid Meniscus DLM



Contours of Volume fraction (water) (Time=5.0000e-06) Feb 14, 2011  
ANSYS FLUENT 12.1 (2d, dp, pbns, vof, lam, transient)

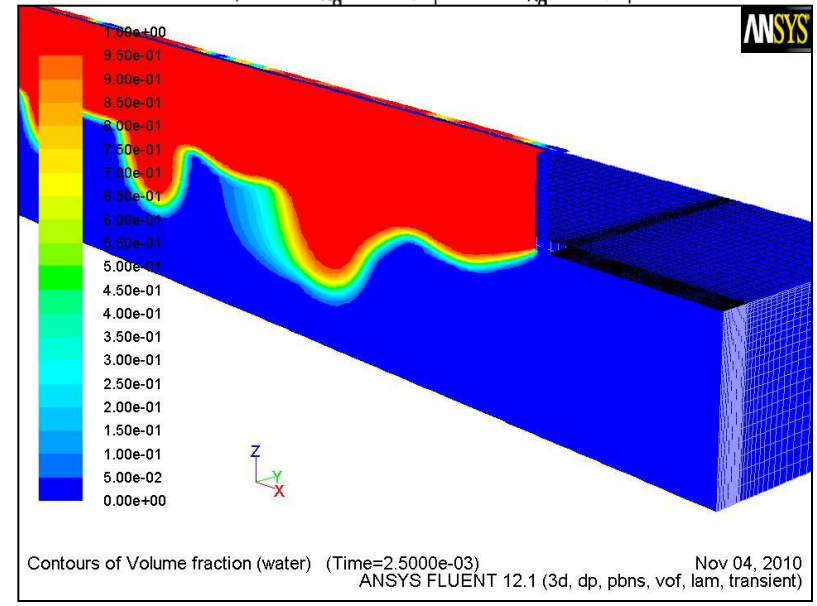
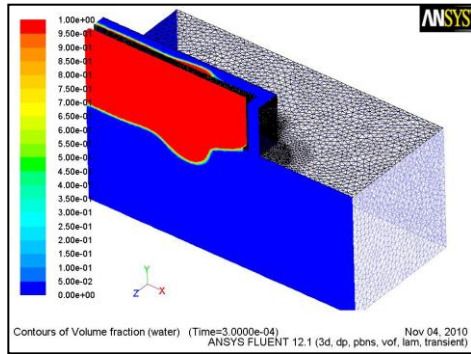
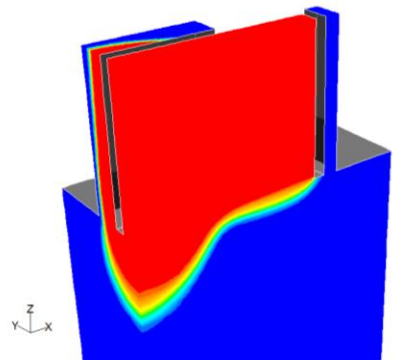
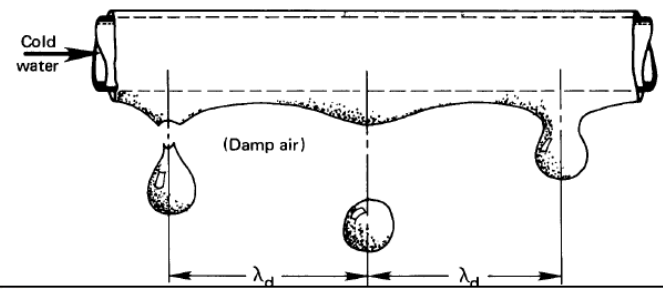
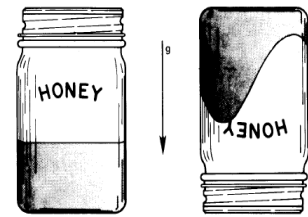
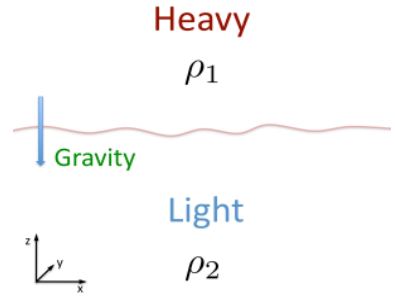
# Dynamic Liquid Multi Drop/Meniscus



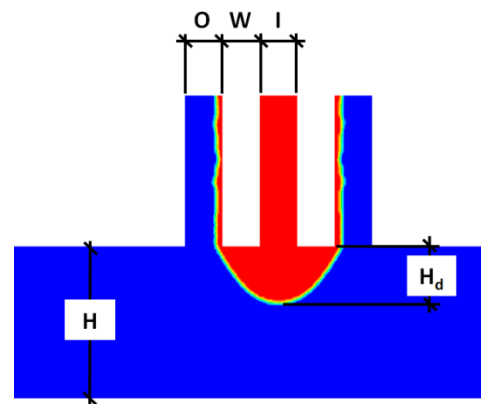
# DLD/DLM: Problem to solve

## Rayleigh Taylor (RT) instability:

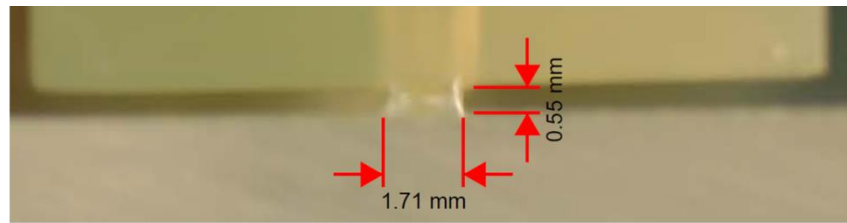
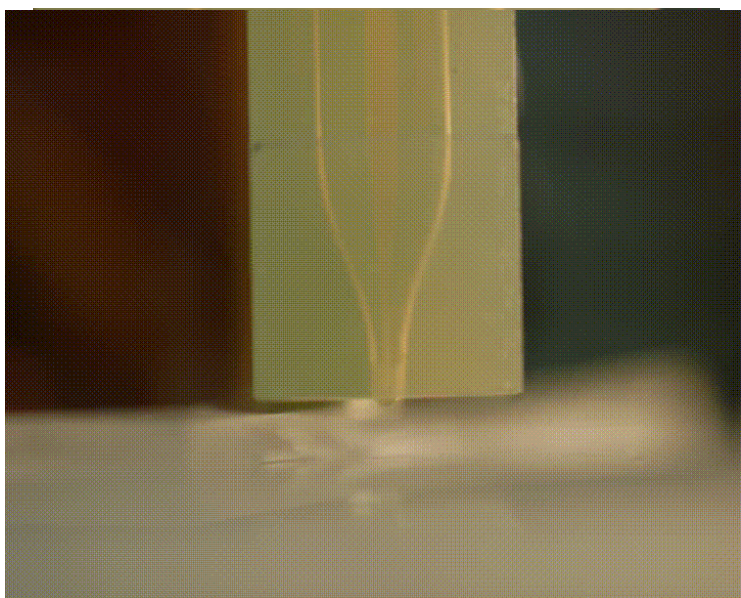
When a heavy fluid is supported by a light fluid, the system is RT unstable



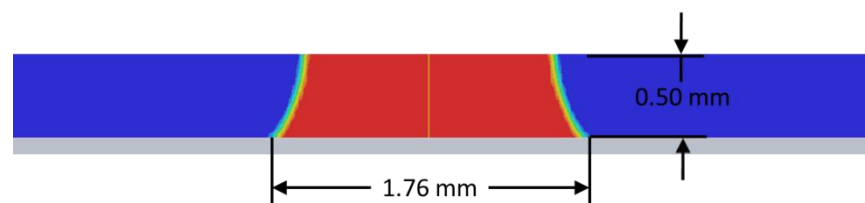
# Simulation and Experimental Results



$I$ ( $\mu\text{m}$ )	$W$ ( $\mu\text{m}$ )	$O$ ( $\mu\text{m}$ )	$H$ ( $\mu\text{m}$ )	$V_{\text{inlet}}$ ( $\text{ms}^{-1}$ )	$\Delta p$ (Pa)	$Re_{\text{Liquid}}$	$H_d$ ( $\mu\text{m}$ )
500	500	500	2000	0,45	1000	437	733
300	300	300	1000	0,5	1500	291	573
100	100	100	600	0,8	2000	155	213
70	70	70	600	0,9	2500	122	187
50	50	50	300	1	3000	97	118
30	30	30	200	1,25	3500	73	73
15	20	20	75	1,5	4000	44	32



Experimental result

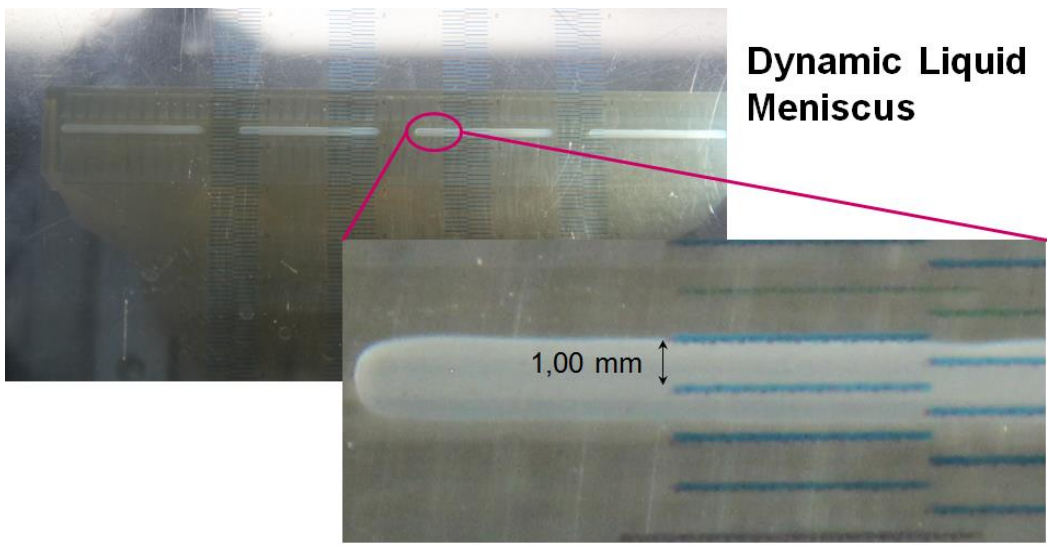
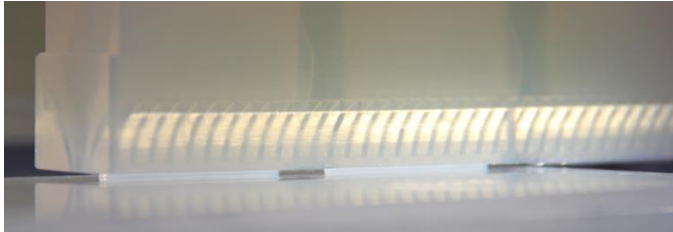
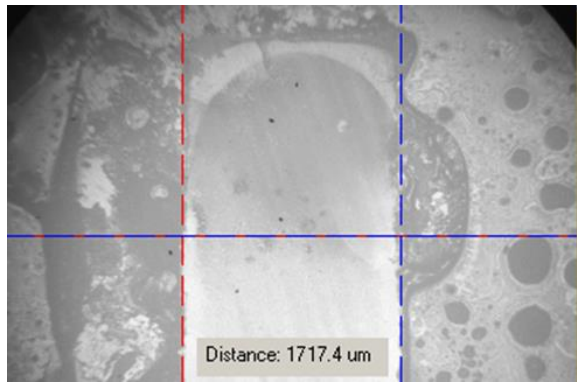


CFD result



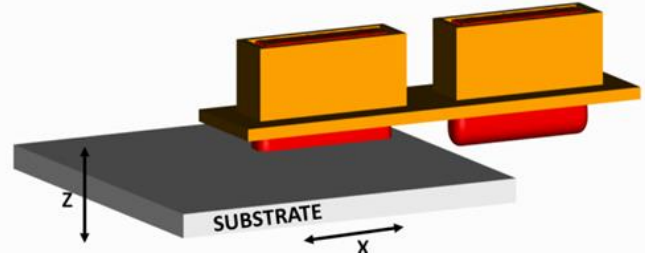
# Simulation and Experimental Results

	I ( $\mu\text{m}$ )	W ( $\mu\text{m}$ )	O ( $\mu\text{m}$ )	H ( $\mu\text{m}$ )	V <sub>inlet</sub> (ms <sup>-1</sup> )	$\Delta p$ (Pa)
<b>Nozzles operating conditions</b>	500	500	500	2000	0,4	2000
	<b>TEST</b>			<b>CFD</b>		
<b>Dynamic drop height (<math>\mu\text{m}</math>)</b>	570			590		
<b>Dynamic meniscus footprint (<math>\mu\text{m}</math>)</b>	1710			1760		
<b>Air volume flow rate (l/min)</b>	60,5			63,9		



**Dynamic Liquid Meniscus**

**PV and Cu Pillar Application: Continuous or Stop-Go**



# Equipment Evolution in Time



2011



2012



2013

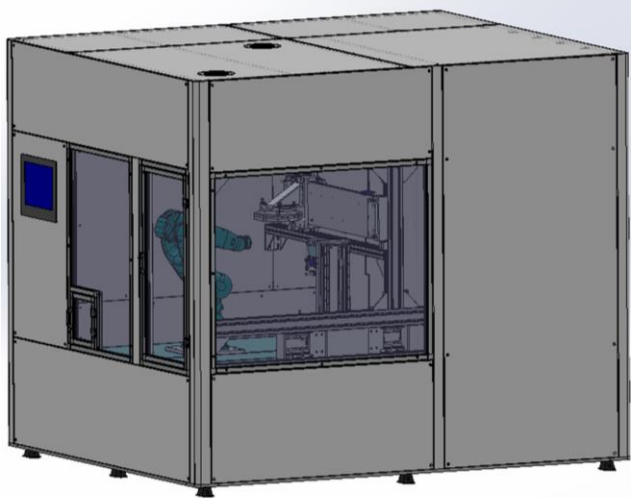


2014

January 2019



2016



<https://www.youtube.com/watch?v=xDIsubIi5EM>

Search in you tube “meniscuspad”



# Micro Nozzles

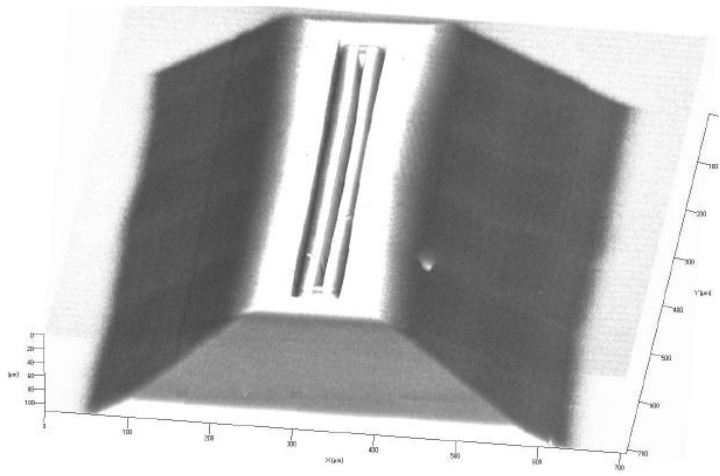
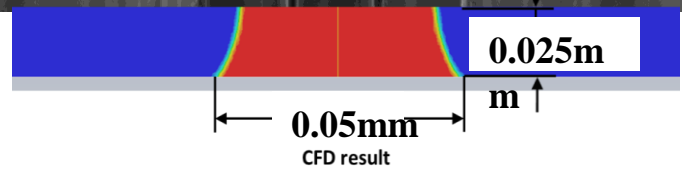
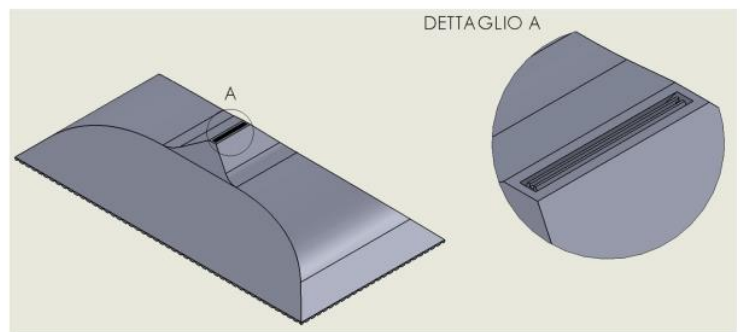


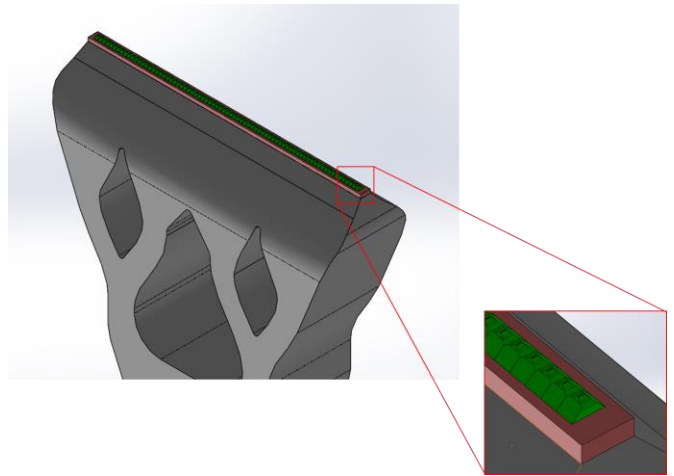
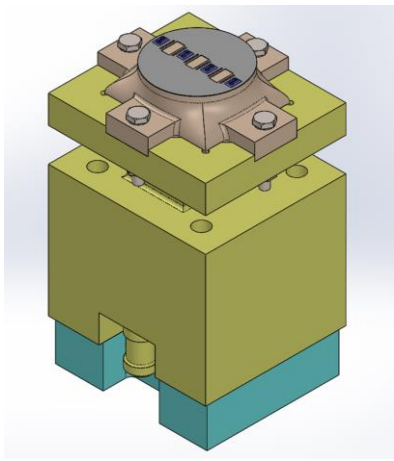
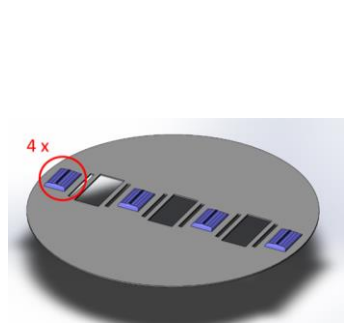
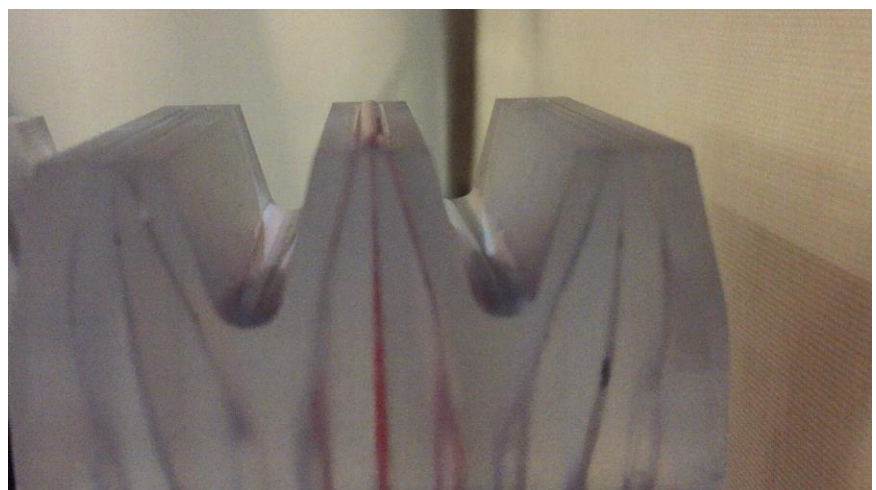
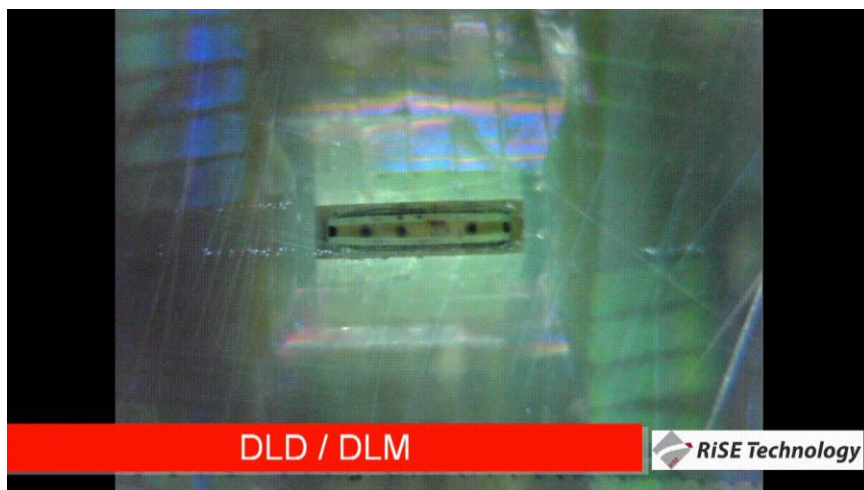
Figura 154: Immagine del micro-ugello finale fatta al microscopio confocale

Microugello MN\_12.5\_7\_15\_V02.2



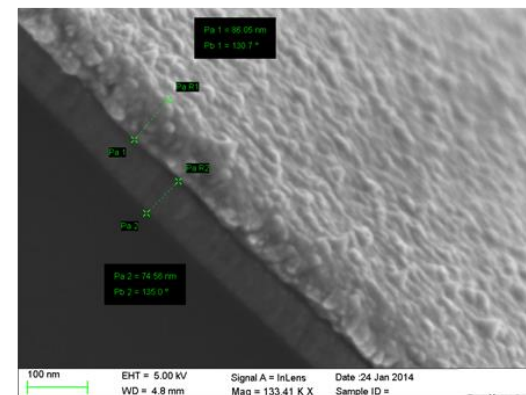
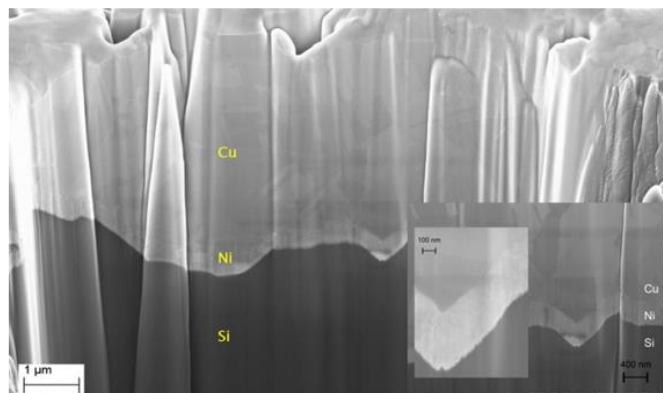
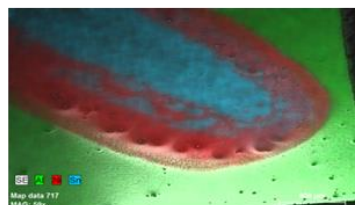
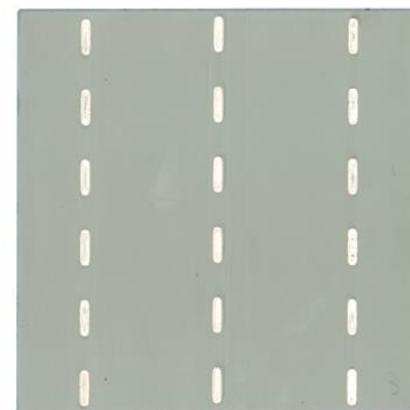
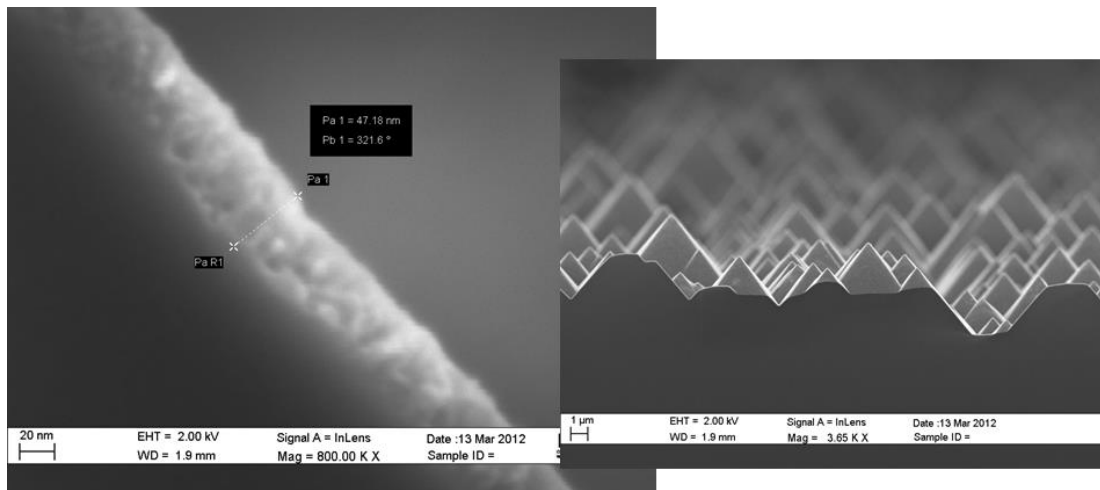
# Micro Nozzles

*Speed of liquid is in the range of 1m/s*



# Main results achieved

- Front and back side of c-Si



# On Going Work – H2020 AMPERE

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## ● Cu on ITO

### • *Main Tasks:*

- Adhesion > 1N/mm
- No damage of SHJ solar cell after ITO processing
- Quality of metal better than low temp. Ag-paste
- FF and BB plating without mask
- Speed of plating → Prove industrial feasibility of technique
  - Productivity > 2500 w/h
  - Make-up solution quantity
  - Floor space



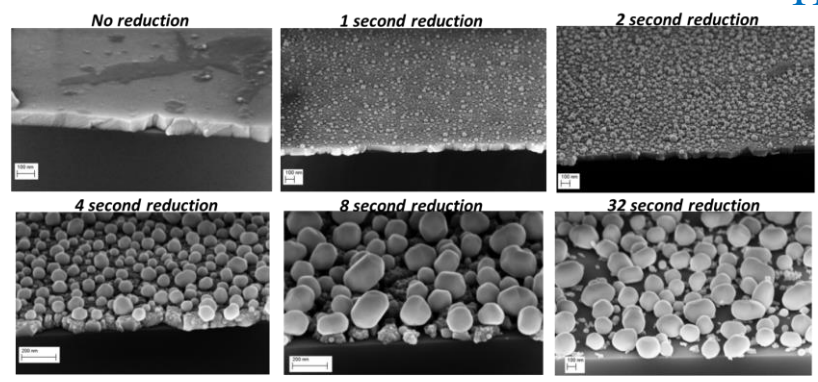
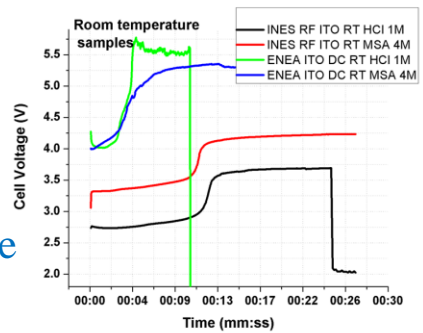
# On Going Work – H2020 AMPERE

## ● Cu on ITO



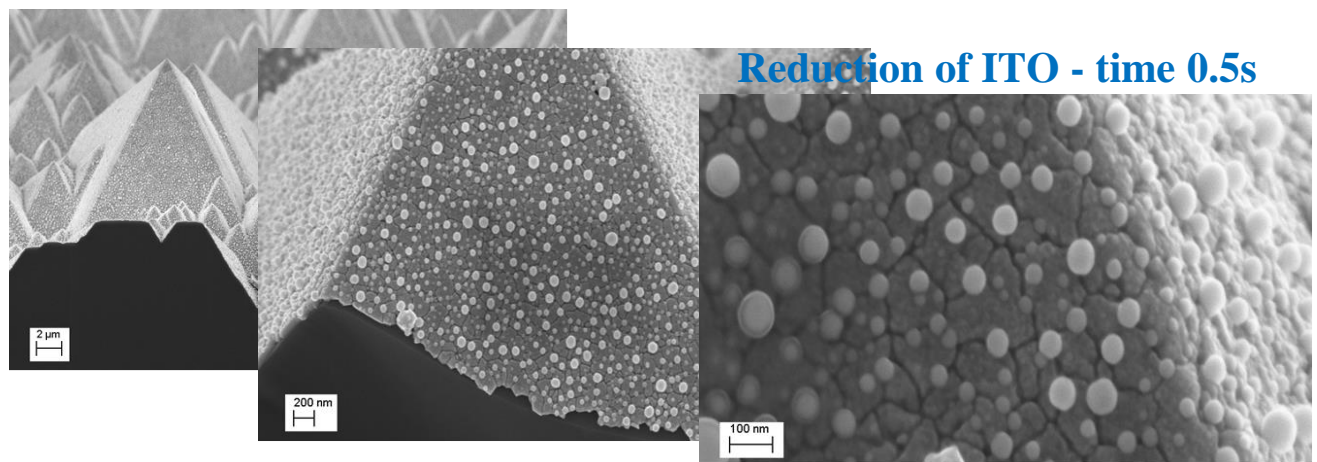
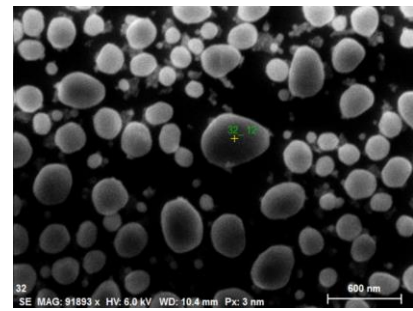
- Adhesion solved by ITO reduction

ITO reduction in galvanic regime  
( $10\text{mA}/\text{cm}^2 \leq J_0 \leq 320\text{mA}/\text{cm}^2$ )

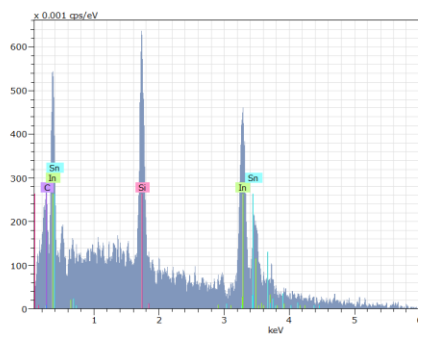


Spectrum: 32\_12

El	AN	Series	unn. C	norm. C	Atom. C	Error (1 Sigma)
			[wt.%]	[wt.%]	[at.%]	[wt.%]
C	6	K-series	1.12	1.22	7.66	0.44
Si	14	K-series	12.44	13.60	36.53	0.74
In	49	L-series	71.30	77.94	51.20	4.33
Sn	50	L-series	6.63	7.24	4.60	0.80
Total:			91.49	100.00	100.00	

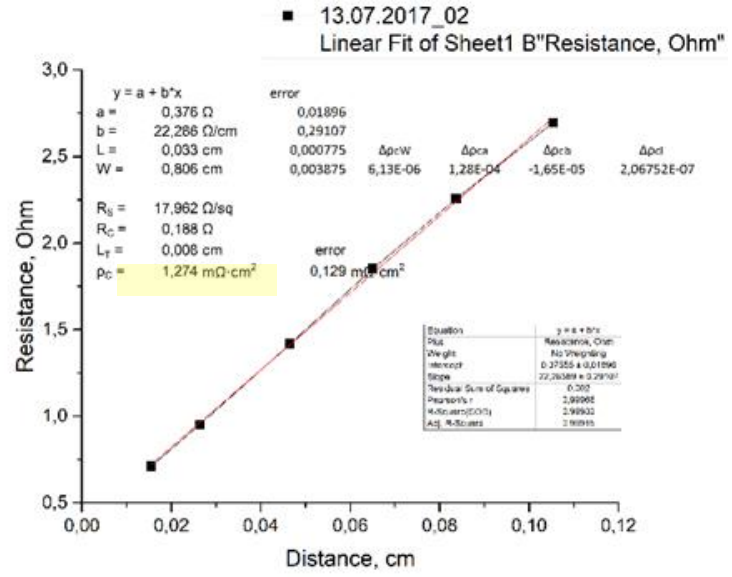
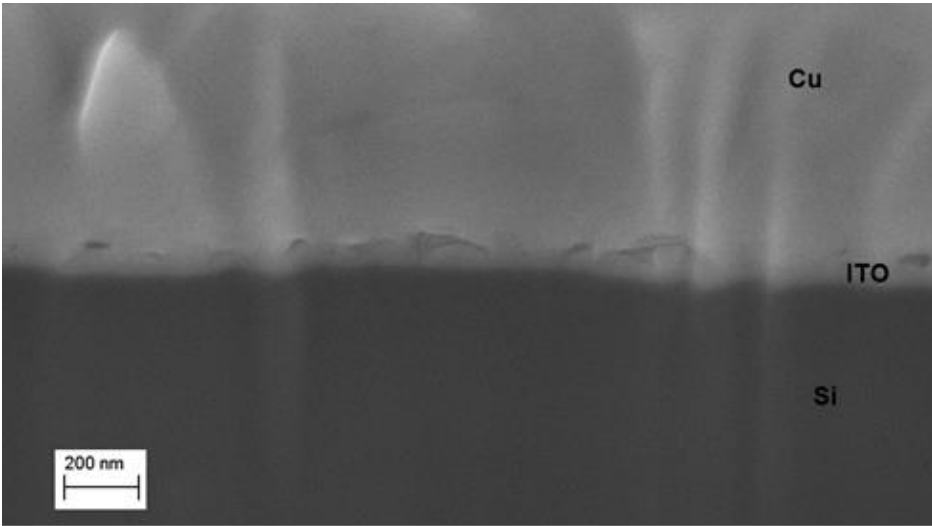


Reduction of ITO - time 0.5s



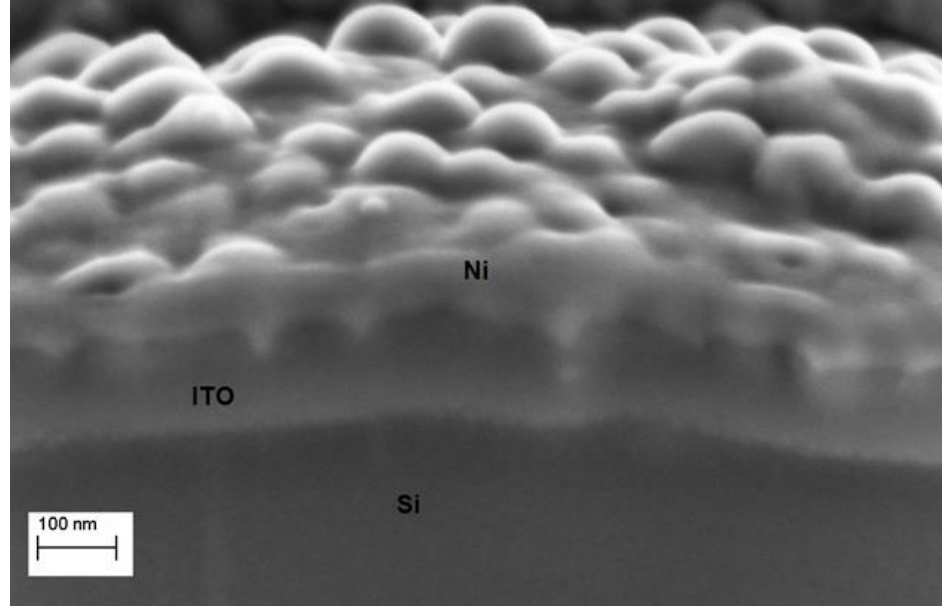
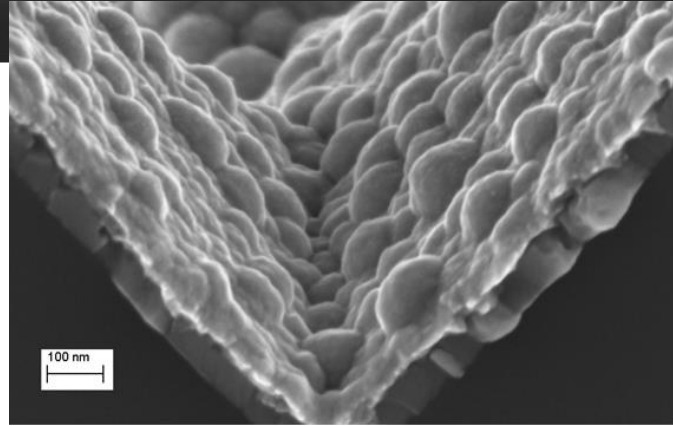
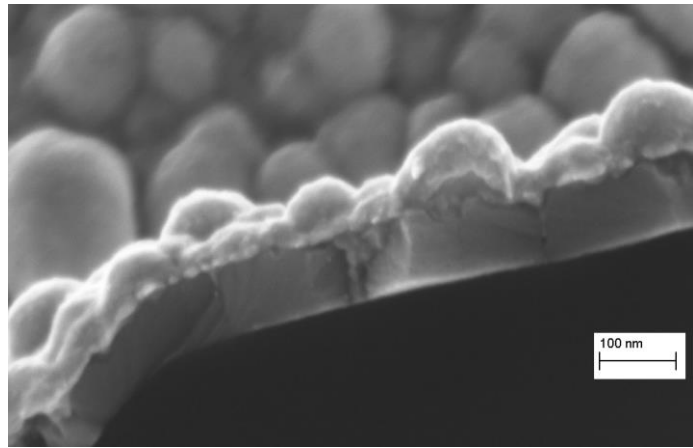
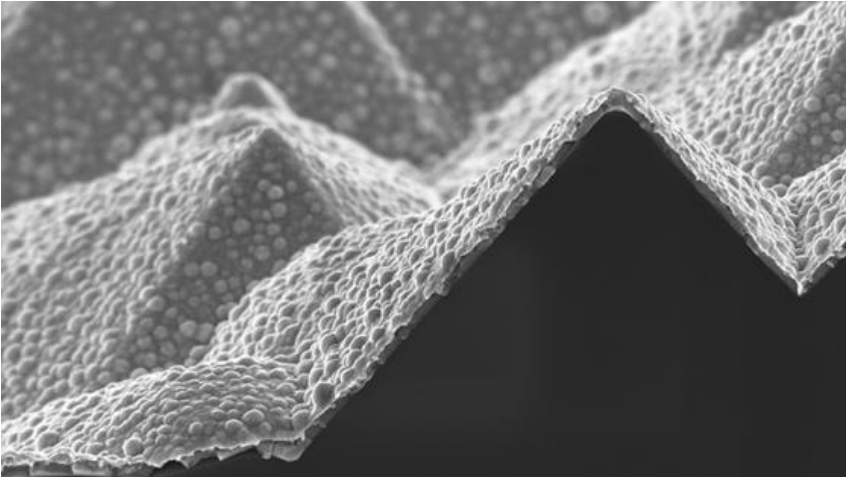
# On Going Work – H2020 AMPERE

- Cu on ITO directly
  - *Adhesion Issue*



# On Going Work – H2020 AMPERE

- Cu on ITO after flash of Ni

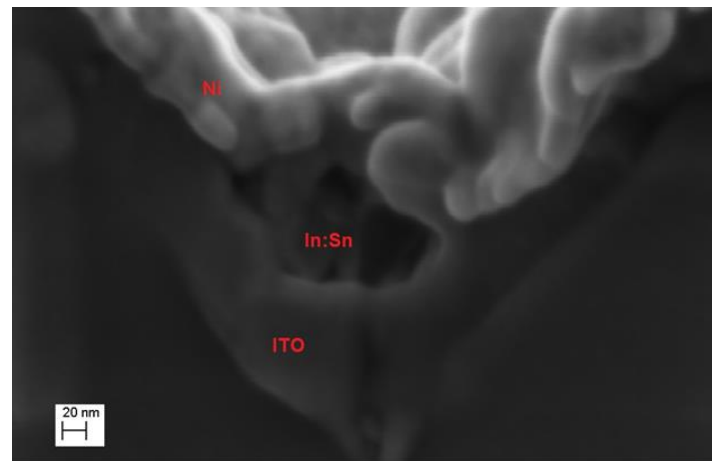




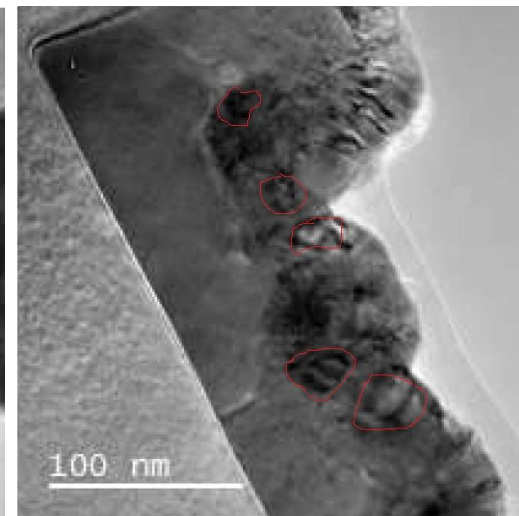
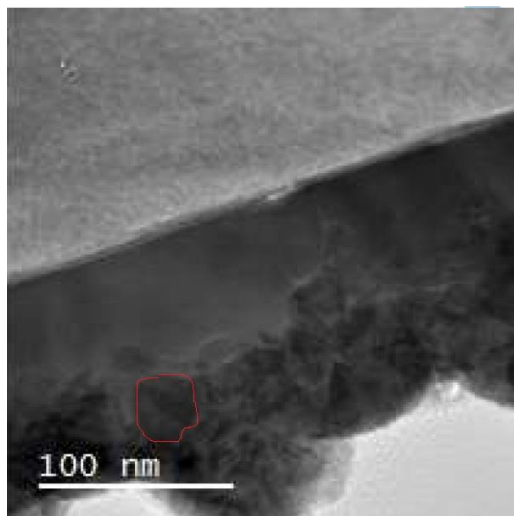
# On Going Work – H2020 AMPERE

- Cu on ITO after flash of Ni

After 2 min @ 200° C

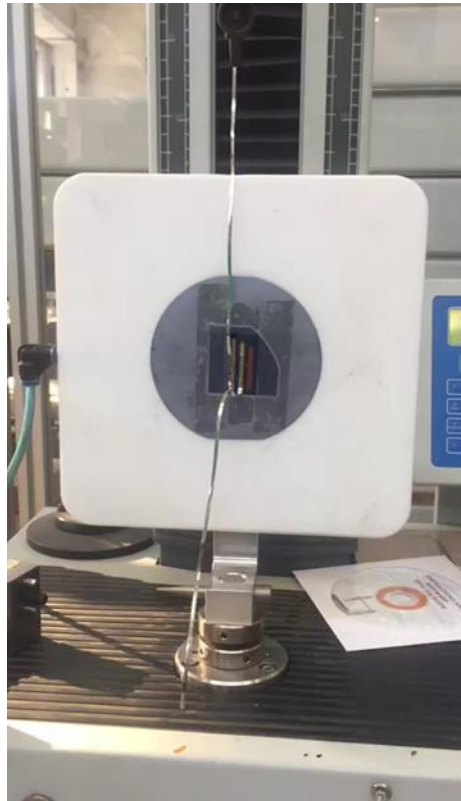
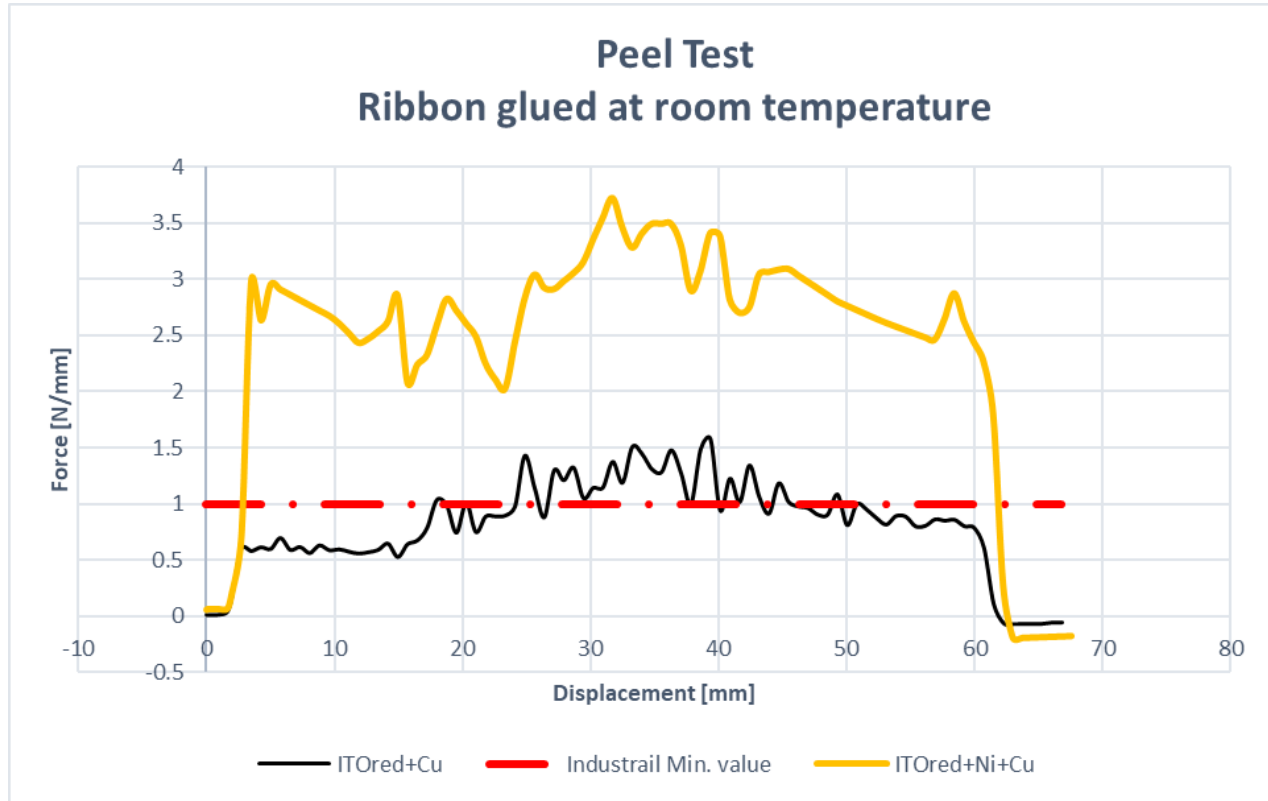


Plating post annealing	Specific contact resistance [mΩcm <sup>2</sup> ]
NO	0.61 ± 0.07
200°C - 120 s	0.58 ± 0.02
150°C - 600 s	0.43 ± 0.1
NO	0.49 ± 0.05
150°C - 600 s	0.3 ± 0.1



# On Going Work – H2020 AMPERE

- Cu (10 $\mu$ m) on ITO after flash of Ni



# On Going Work – H2020 AMPERE

## ● Evaluation of ITO damage:

### Recombination lifetime

Before ITO reduction	After ITO reduction	Reduction in MSA
$\tau = 1054 \mu\text{s}$	$\tau = 1079 \mu\text{s}$	p-side 320 mA 0.1 sec.
$\tau = 749 \mu\text{s}$	$\tau = 803 \mu\text{s}$	p-side 200 mA 0.1 sec.
$\tau = 1291 \mu\text{s}$	$\tau = 1227 \mu\text{s}$	p-side 100 mA 0.1 sec.

ITO reduction in MSA solution doesn't affect recombination lifetime leaving the passivation layer intact.

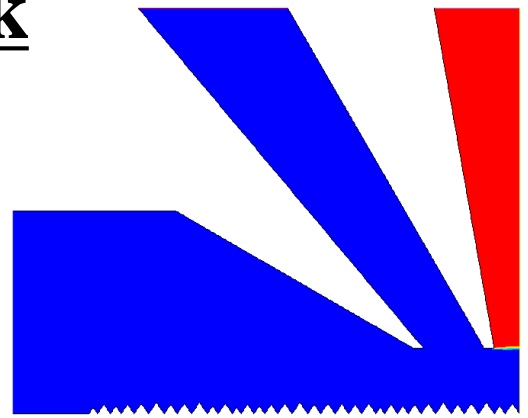
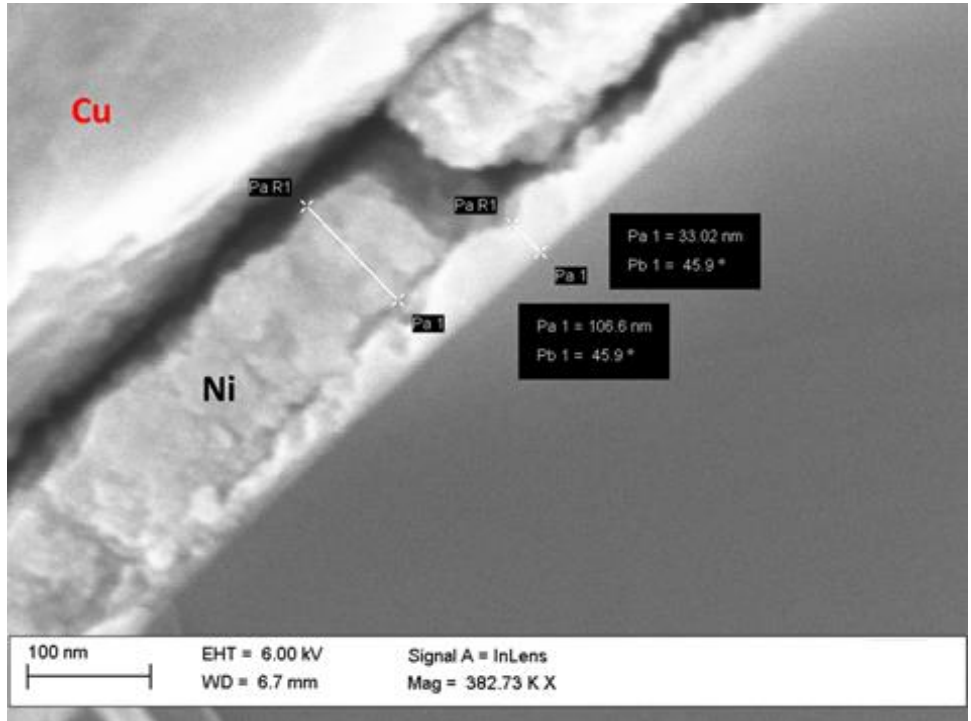
INES (RF) ITO Reduction current		HCl (1 M, 82 gr HCl 37%/L)	MSA (4 M, 250 g pure/L)
80 mA	0.5 s	427 $\Omega$	263.8 $\Omega$
80 mA	0.4 s	313.9 $\Omega$	229 $\Omega$
110 mA	0.3 s	342.1 $\Omega$	245.5 $\Omega$
160 mA	0.2 s	339.7 $\Omega$	227.2 $\Omega$
320 mA	0.1 s	386.6 $\Omega$	239.1 $\Omega$
100 mA	0.2 s	239.2 $\Omega$	204.5 $\Omega$

### HCl-MSA comparison

MSA is capable of reducing ITO inducing less damage and limiting the corrosion of the formed In-Sn phase

# On Going Work – H2020 AMPERE

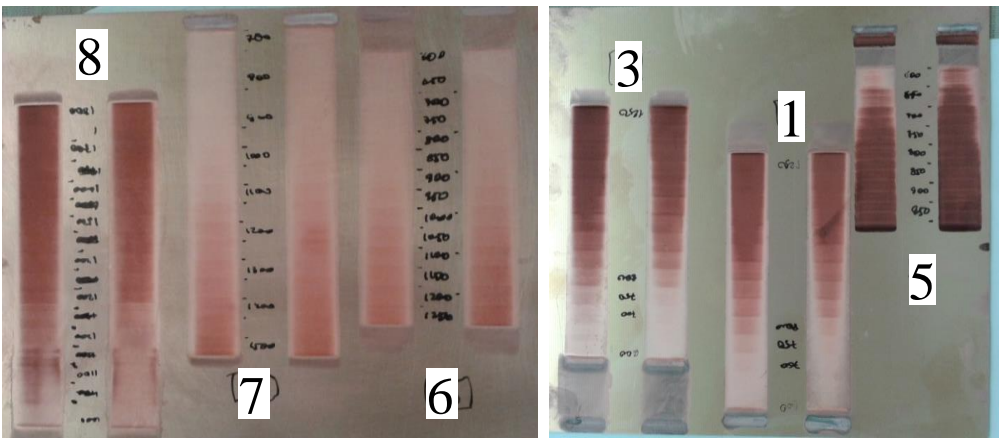
- FF and BB plating without mask



Moving down to 50µm FF

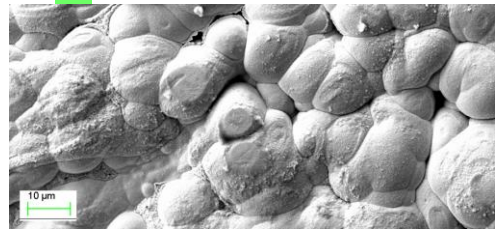
# On Going Work – H2020 AMPERE

## ● Cu Speed of plating

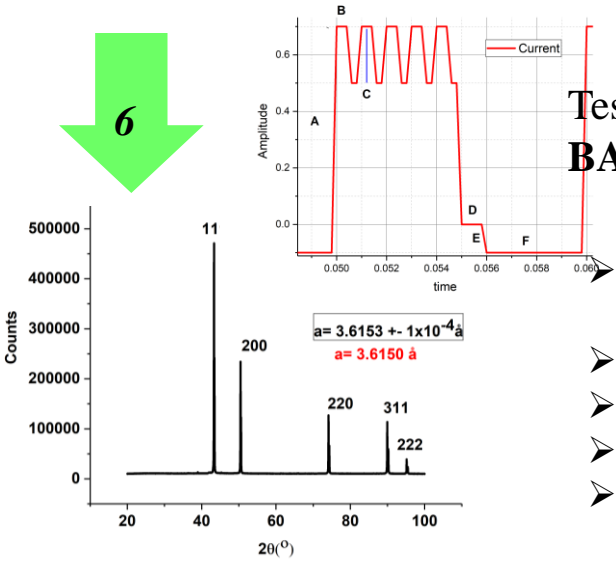


N.	$I_{min}$ [ASD] DR [ $\mu\text{m/s}$ ]	$I_{max}$ [ASD] DR [ $\mu\text{m/s}$ ]	Fluid Velocity [m/s]	Optimal $I$ [ASD] DR [ $\mu\text{m/s}$ ]	Cu [g/l]
5	200 0.74	300 1.10	0.17	-	17
3	200 0.74	413 1.53	0.87	200 0.74	17
1	166 0.61	433 1.59	1.04	233 0.86	17
6	200 0.74	300 1.10	1.04	300 1.10	38
7	233 0.86	500 1.84	1.04	366 1.35	45
8	333 1.23	600 2.21	1.04	366 1.35	45

$\rho_{Cu} \leq 3.0 \mu\Omega\cdot\text{cm}$



Testing PRP average  $J$  up to 1500 ASD  
**BASF additive for dendrite suppression**

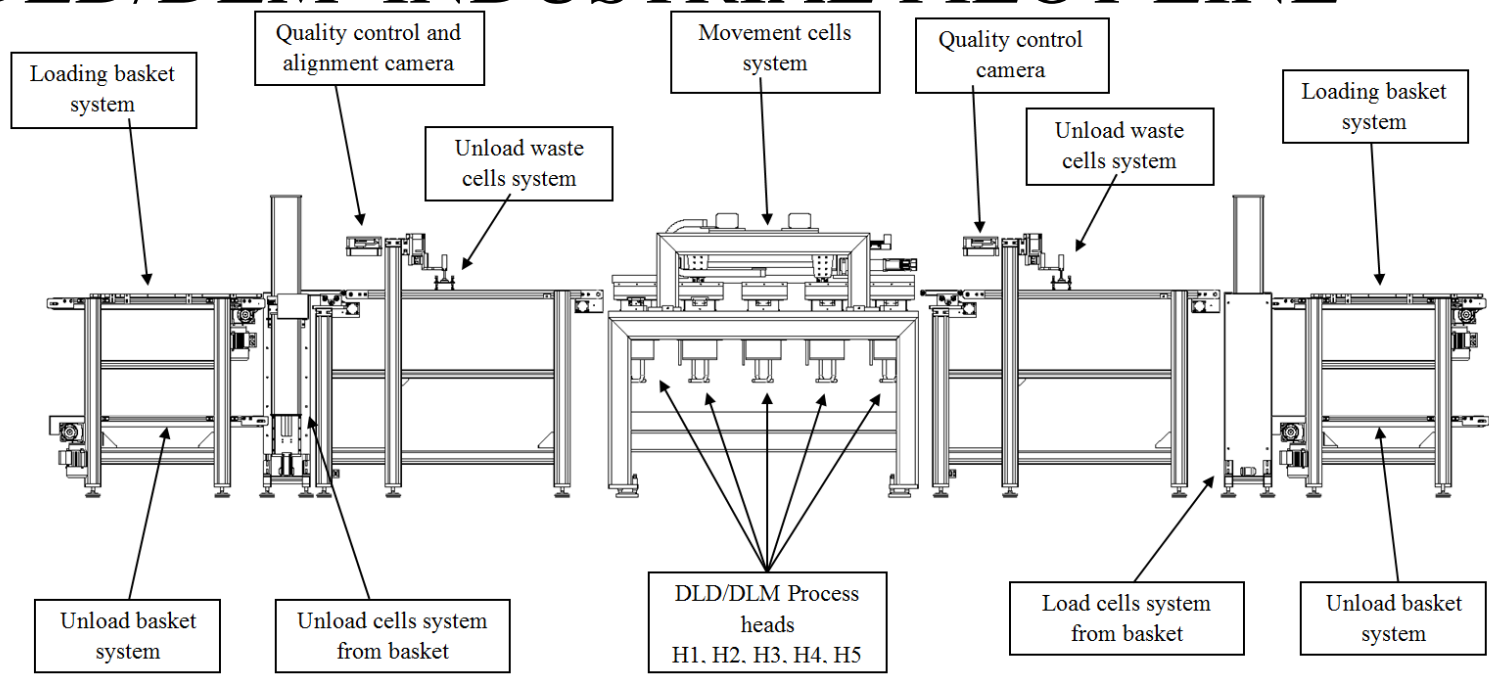


- High crystallinity (low background, sharp and intense peaks,  $k_{02}$  visible even at high  $\theta$ ).
- High purity (no detectable  $\text{Cu}_2\text{O}$ ,  $\text{Cu}(\text{OH})_2$ ,  $\text{CuO}$  or  $\text{CuCO}_3$  phases).
- Large crystallites (over the instrumental resolution  $\approx 150 \text{ nm}$ )
- High density ( $8.93 \text{ g/cm}^3$ ), substrate (brass) contribution is immaterial.
- Williamson-Hall Plots and Warren-Averbach shows strain in the range 100-200MPa

# Industrial Feasibility

## ● DLD/DLM INDUSTRIAL PILOT LINE

PIPELINE CONFIGURATION



Speed of Plating	1um/s	2um/s	5um/s
Length of Equipment for 100MW @20um Cu [liter make-up]	12 m x 1 m= 12 m <sup>2</sup> [300]	6 m x 1 m= 6 m <sup>2</sup> [200]	2.4 m x 1 m= 2.4 m <sup>2</sup> [150]



# Acknowledgement



Italian National Agency for New Technologies,  
Energy and Sustainable Economic Development

Dr. Mario Tucci and Dr. Massimo Izzi  
samples, measurement and discussion



Dr. Jean-François Lerat and Dr. Delfina Munoz  
ITO samples and discussion



H2020 Project Coordinator



Dr. Salvatore Lombardo  
TEM and discussion



# Conclusion

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- Adhesion > 1N/mm
- No damage of SHJ solar cell after ITO processing
- Quality of Cu better than low temp. Ag-paste
- ITO red., Ni and Cu plating - **NO MASK**
- Speed of plating → **1 μm/s** → **2 μm/s** → **10 μm/s**
  - Productivity > 2500 w/h
  - Make-up solution ≤ 300 liter
  - Floor space at least 5 times less than state of the art

## Breakthrough in Plating for Solar Cell Metallization