# 1 Critical analyses of nitrous oxide emissions in a full scale activated sludge system<br>2 treating low carbon-to-nitrogen ratio wastewater<br>3 M. Spinelli<sup>1</sup>, A.L. Eusebi<sup>1\*</sup>, V. Vasilaki<sup>2</sup>, E. Katsou<sup>2</sup>, N. Frison<sup>3</sup>, D.Cin 2<br>
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**Critical analyses of nitrous oxide emissions in a full-scale activated sludge system**<br> **Critical analysis of the modified CET**<br> **M. Spinelli<sup>1</sup>, A.L. Eusebi<sup>12</sup>, V. Vasilaki<sup>2</sup>, E. Katsou<sup>2</sup>, N. Frison<sup>3</sup>, D.Cingolani<sup>1</sup>, Critical analyses of nitrous oxide emissions in a full scale activated sludge system**<br> **M. Spinelli<sup>1</sup>, A.L. Eusebi<sup>10</sup>, V. Vasilaki<sup>2</sup>, E. Katsou<sup>2</sup>, N. Frison<sup>3</sup>, D.Cingolani<sup>1</sup>, F. Fatone<sup>1</sup><br>
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<sup>1</sup> Dipartim **Critical analyses of nitrous oxide emissions in a full scale activated sludge system**<br> **Critical parameters of the plant and long-term online monitorial parameters**<br> **Critical Parameters of the plant and long-term online Conduct Conduct Conduct Conduct Conduct Considered** Market Conductions (Fig. 2014)<br> **Consider the analysis considered the effect of off-gas samely P. S. Consider The Section Bureau Binache, 12, 60100 Ancona, IT. a.l.cuscb M. Spinelli<sup>1</sup>, A.L. Eusebi<sup>19</sup>, V. Vasilaki<sup>2</sup>, E. Katsou<sup>2</sup>, N. Frison<sup>3</sup>, D. Cingolani<sup>1</sup>, F. Fatone<sup>1</sup><br><sup>1</sup> Dipartimento SIMAU, Facoltà di Ingegneria, Università Politecnica delle Marche, Via Brecce<br>
Bianche, 12, 6010 M. Spinclli<sup>1</sup>, A.L. Euschi<sup>1</sup><sup>\*</sup>, V. Vasilaki<sup>2</sup>, E. Katsou<sup>2</sup>, N. Frison<sup>3</sup>, D. Cingolani<sup>1</sup>, F. Fatone<sup>1</sup><br>
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<sup>2</sup> Department of Civil Engineering and Environmental Engineering; Instit Furgarmento SIMARC, reacous our miggeneral, conversau Fondeemica dene Marche, Via Brecce<br>Bianche, 12, 60100 Ancona, IT. a.l.cuschiggunivpm.it<br>
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2 Department of Givil Engineering and Environmental Engineering; Ins <sup>2</sup> Department of Civil Engineering and Environmental Engineering; Institute of Environment, Health and Societies, Brunel University London, Uxbridge Campus, Middlesex, UB8 3PH, Uxbridge, UK.<br><sup>3</sup> Department of Biotcehnolog **Example Constraines** and Societies, Brunel University London, Uxbridge Canpus, Middlesex, UB8 3PH, Uxbridge, UK.<br><sup>3</sup> Department of Biotechnology, University of Verona, Strada Le Grazie 15, Verona, IT<br>**Abstract:** A critica <sup>2</sup> and stripped more of the more of the more of the controls, the controls, the controls, the controls, the controls, the controls of the more nitrow carbon Exparement or Biotecnhology, University or Verona, Strada Le Grazie 13, Verona, 11<br>
Abstract: A critical analysis of nitrous oxide emissions in a full-scale modified Ludzack Eitinger<br>
plant treating municipal wastevater w

Gas sampling assessment, Emission Factors

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15 1. Introduction<br>16 Biological processes are significant sources of greenhouse gases (GH<br>17 (CO<sub>2</sub>) methane (CH<sub>4</sub>) and dinitrogen oxide (N<sub>2</sub>O) in wastewater 15 **1. Introduction**<br>16 Biological processes are significant sources of greenhouse gases (GHGs), mainly carbon dioxide<br>17 (CO<sub>2</sub>), methane (CH<sub>4</sub>) and dinitrogen oxide (N<sub>2</sub>O) in wastewater treatment plants (WWTPs)<br>18 (Ka 15 **1. Introduction**<br>
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18 20 **1. Introduction**<br>20 Biological processes are significant sources of greenhouse gases (GHGs), mainly carbon dioxide<br>20 (CO<sub>2</sub>), methane (CH<sub>4</sub>) and dinitrogen oxide (N<sub>2</sub>O) in wastewater treatment plants (WWTPs)<br>20 (Ka 21 **1. Introduction**<br>21 Biological processes are significant sources of greenhouse gases (GHGs), mainly carbon dioxide<br>21 (CO<sub>2</sub>), methane (CH<sub>4</sub>) and dinitrogen oxide (N<sub>2</sub>O) in wastewater treatment plants (WWTPs)<br>20 (Ka 22 **12 I. Introduction**<br>22 methanoceals are significant sources of greenhouse gases (GHGs), mainly earbon dioxide<br>22 (CO2), methane (CH4) and dimitrogen oxide (N<sub>2</sub>O) in wastewater treatment plants (WWTPs)<br>22 (Kampschre 23 1. **Introduction**<br>24 16 Biological processes are significant sources of greenhouse gases (GHGs), mainly carbon dioxide<br>24 (COs), methane (CHs) and dinitrogen oxide (N<sub>2</sub>O) in wastewater treatment plants (WWTPs)<br>23 (Kam 15 **1. Introduction**<br>
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18 (Kampschreur et al., 2008). N<sub>2</sub>O emissions are considered as the most potent (-300 more than CO<sub>2</sub>)<br>
contributor to global 18 (Kampschreur et al., 2008). N<sub>2</sub>O emissions are considered as the most potent (~300 more than CO<sub>2</sub>) contributor to global warming. Numerous studies to date have focussed on real-field N<sub>2</sub>O monitoring and the understa contributor to global warming. Numerous studies to date have focussed on real-field N<sub>2</sub>O monitoring<br>
20 and the understanding of the causes on N<sub>2</sub>O formation (IPPC, 2013). Several studies reveal that<br>
21 characteristics and the understanding of the causes on N<sub>2</sub>O formation (IPPC, 2013). Several studies reveal that<br>
characteristics of the wastewater, operating parameters, configuration, environmental conditions and<br>
microbiological diver characteristies of the wastewater, operating parameters, configuration, environmental conditions and<br>22 microbiological diversity of the biological processes have significant impact on N<sub>5</sub>O generation and<br>23 on operationa 31 needed. 33 on operational carbon footprint of WWTPs. Critical effect was found in relation to the carbon to<br>
324 nitrogen ratio both in lab- or pilot- scale biofilm systems (He et al., 2017; Zhang et al, 2016), for<br>
325 aerobic g mitrogen ratio both in lab- or pilot- scale biofilm systems (He et al., 2017; Zhang et al., 2016), for<br>aerobic granules and suspended activated sludge (Gao et al., 2016; Sun et al., 2014; He et al., 2017;<br>
32 Ge et al., 2 aerobic granules and suspended activated sludge (Gao et al, 2016; Sun et al., 2014; He et al., 2017;<br>
26 Ge et al., 2017). Modelling studies also confirm the phenomena (Jose et al., 2016; Law et al., 2012).<br>
32 To the bes Gic et al., 2017). Modelling studies also confirm the phenomena (Jose et al., 2016; Law et al., 2012).<br>
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been studied. Mo

27 To the best of authors' knowledge the effect low C:N on N<sub>2</sub>O emissions in full scale WWTPs has not<br>28 been studied. Moreover, there is little evidence of extensive analytical studies in the literature<br>29 regarding GHG been studied. Moreover, there is little evidence of extensive analytical studies in the literature<br>regarding GHGs emissions from full scale plants. Therefore, quantifying the N<sub>2</sub>O production and<br>determining an effective regarding GHGs emissions from full scale plants. Therefore, quantifying the N<sub>2</sub>O production and<br>30 determining an effective mitigation approach to employ in existing full scale biological processes is<br>31 needed.<br>33 Uncer

39 emissions (Cavazzuti, M. 2013). The full understanding of the uncertainties in GHG emissions and<br>40 the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al.,<br>2017 and Daelman e 40 the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al., 41 2017 and Daelman et al., 2015).

42 This study addresses in full scale wastewater systems is a pertainties in GHG emissions and<br>40 the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al.,<br>41 2017 and Daclman et 43 emissions (Cavazzuti, M. 2013). The full understanding of the uncertainties in GHG emissions and<br>40 the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al.,<br>41 2017 and Daclm 44 techniques (online gaseous emissions and statistical analysis) to emissions and the biological processes in full scale was exercuater systems is a pertaining challenge (Massara et al., 2017 and Daelman et al., 2015).<br>4 44 ebtological processes in full scale wastewater systems is a pertaining challenge (Massara et al., 2017 and Daelman et al., 2015).<br>41 abtivations and the biological processes in full scale wastewater systems is a pertai 44 the optimal sampling method and to systematically relate the critical role of the systems is a pertaining challenge (Massara et al., 2017 and Dachman et al., 2015).<br>44 the biological processes in full scale wastewater emissions (Cavazzuti, M. 2013). The full understanding of the uncertainties in GHG emissions and<br>the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al.,<br>2017 and Daclman et al., emissions (Cavazzut, M. 2013). Ine ruli understanding of the uncertainties in Orto emissions and<br>the biological processes in full scale wastewater systems is a pertaining challenge (Massara et al.,<br>2017 and Daelman et al. 49 and Dachman et al., 2015).<br>
42 and Dachman et al., 2015).<br>
42 This study addresses this gap of knowledge by analysing the results of N<sub>2</sub>O emissions in a full-scale<br>
43 WWTP by performing classical (liquid ehemical/phys 2017 and Daciman et at, 2015).<br>
42 This study addresses this gap of knowledge by analysing the results of N<sub>2</sub>O emissions in a full-scale<br>
43 WWTP by performing classical (liquid chemical/physical characterization) and new This study addresses this gap of knowledge by analysing the results of<br>
43 WWTP by performing classical (liquid chemical/physical characterize<br>
44 techniques (online gaseous emissions and statistical analysis) to cri-<br>
45 techniques (online gaseous emissions and statistical analysis) to critically examine the related between the monitored variables and N<sub>2</sub>O gaseous emissions. The study attempts both to calibule the optimal sampling method between the monitored variables and N<sub>2</sub>O gaseous emissions. The study attempts both to calibrate<br>the optimal sampling method and to systematically relate the gaseous emissions with the main liquid<br>variables routinely anal the optimal sampling method and to systematically relate the gaseous emissions with the main liquid<br>variables routinely analysed. Moreover, for the first time the critical role of the C:N ratio is discussed<br>calculating the

 $m<sup>3</sup>/d$ . Infiltration from groundwater and marine intrusions cause under-loading influent conditions variables routinely analysed. Moreover, for the first time the critical role of the C:N ratio is discussed<br>
alevantating the emission factors (EF) from the full-scale mass balances. Event-based sensitivity<br>
analysis (Tava calculating the emission factors (EF) from the full-scale mass balances. Event-based sensitivity<br>analysis (Tavakoli et al. 2013 a, b) is applied to identify potential dependencies between the system<br>variables monitored onl analysis (Tavakoli et al. 2013 a, b) is applied to identify potential dependencies between the system<br>variables monitored online and the N<sub>2</sub>O emissions of the biological reactor.<br>2. **Material and methods**<br>2. **2. Naterial** 50 variables monitored online and the N<sub>2</sub>O emissions of the biological reactor.<br>
51 **2. Material and methods**<br>
52 **2.1 Wastewater treatment process**<br>
53 The municipal wastewater treatment plant (WWTP) of Falconara Maritt 51 **2. Material and methods**<br>
52 **2.1 Wastewater treatment process**<br>
53 The municipal wastewater treatment plant (WWTP) of Falconara Marittima (Italy) is fed by low C:N<br>
54 ratio wastewater. It has a design capacity of 80 is  $13,700 \text{ m}^3$ . The aerated compartments are equipped by ceramic fine bubble diffusers; the air supply 22 2.1 Wastewater treatment process<br>
53 The municipal wastewater treatment plant (WWTP) of Falconara Marittima (Italy) is fed by low C:N<br>
54 ratio wastewater. It has a design capacity of 80,000 PE and a design average inf ranges between 1,870 and 9,210  $m^3/h$ . An automatic system controls the four blowers (Robuschi mod. 22 2.1 Wastewater treatment process<br>
53 The municipal wastewater treatment plant (WWTP) of Falconara Marittima (Italy) is fed by low C:N<br>
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RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br> $m^3$ , with surface ar 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>63 different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br>64 m<sup>3</sup>, with su  $m<sup>3</sup>$ , with surface area of 507 m<sup>2</sup>. The nitrification bioreaction volume is 4.900 m<sup>3</sup>, with surface area SS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (the ferent operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8, with surface area of 507 m<sup>2</sup>. entration of the dissolved oxygen (DO) in the aerobic reactor (three<br>
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one of t 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>63 different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br>64 m<sup>3</sup>, with su RBS LP120) based on the concentration of the dissolved oxygen (DO) in t<br>different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bio<br>m<sup>3</sup>, with surface area of 507 m<sup>2</sup>. The nitrification bioreaction volu 62 **RBS LP120)** based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860 m<sup>3</sup>, with surfac 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br>m<sup>3</sup>, with surface 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860 m<sup>3</sup>, with surface 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br>m<sup>3</sup>, with surface 62 RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification biorcaction volume is 8,860 m<sup>3</sup>, with surface RBS LP120) based on the concentration of the dissolved oxygen (DO) in the aerobic reactor (three<br>different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860<br>m<sup>3</sup>, with surface are 71 in the nitrification reactor was  $4.3\pm0.9$  mgO<sub>2</sub>/L and the pH was buffered at  $8.1\pm0.2$  due to underd3 different operating settings: 0.4, 1.0 and 3.0 mg/L). The denitrification bioreaction volume is 8,860 m<sup>3</sup>, with surface area of 500 m<sup>2</sup>. This study analyses one of the two parallel lines of the activated sludge biore 64 m<sup>3</sup>, with surface area of 507 m<sup>2</sup>. The nitrification bioreaction volume is 4,900 m<sup>3</sup>, with surface area<br>65 of 560 m<sup>2</sup>. This study analyses one of the two parallel lines of the activated sludge bioreactor. The<br>66 sy 67 300 m<sup>2</sup>. I mis study analyses one of the two paralier lines of the aerivated sludge bioreactor. The<br>system is continuously monitored by on-line sensors (Dissolved Oxygen – DO-; Temperature – T-;<br>Mixed Liquor Suspended Mixed Liquor Suspended Solid-MLSS- and Oxidation Reduction Potential - ORP) and magnetic flow<br>
meters (influent, effluent, recirculation and waste sludge). The average sludge retention time (SRT)<br>
was 10 days and the slud 76 whereas (influent, effluent, recirculation and waste sludge). The average sludge retention time (SRT)<br>
76 was 10 days and the sludge recycle ratio ( $Q_{\text{chulge~meyclud}}/Q_{\text{mflame}}$ ) was 0.5. The MLVSS (Mixed Liquor<br>
70 Volat 69 was 10 days and the sludge recycle ratio (Q<sub>alidge recycles</sub>/Q<sub>alibon</sub>) was 0.5. The MLVSS/MLSS 0.61). The DO<br>
71 in the nitrification reactor was 4.3±0.9 mgO<sub>2</sub>/L and the pH was buffered at 8.1±0.2 due to under-<br>
12 i

Volatile Suspended Solids) concentration was 3,485±636 mg/L (ratio MLVSS/MLSS 0.61). The DO<br>
in the nitrification reactor was 4.3±0.9 mgO<sub>2</sub>/L and the pH was buffered at 8.1±0.2 due to under-<br>
1020 loading characteristies 80 measured in the filtrate obtained after the filtration of the sample through Whatman 0.45  $\mu$ m 81 membrane filters.  $NO<sub>2</sub>–N$ ,  $NO<sub>3</sub>–N$  were measured by ion chromatography in samples that were first 2.2 Analytical methods and biomass activity tests<br>
22 Analytical riquot grab samples were collected twice per week from the aerobic and pre-anoxic reactors,<br>
23 Mixed-liquor grab samples were taken twice per week from the Mixed-liquor grab samples were collected twice per week from the aerobic and pre-anoxic reactors,<br>whereas 24h composite samples were taken twice per week from the influent and once per week from<br>the effluent. All the sampl Mixed-inquor gran samples were conceted twice per week from the acronic and pre-anoxie reactors,<br>The effluent. All the samples were taken twice per week from the influent and once per week from<br>the effluent. All the sample Whereas 24 composite samples were taken twice per week from the influent and once per week from<br>the effluent. All the samples were analysed in terms of pH, chemical oxygen demand (COD), total<br>Kejdahl nitrogen (TKN), ammon

 $N_2$ O production were not studied by additional dissolved N<sub>2</sub>O in the liquid phase because this<br>87 knowledge has been provided by other bench- or pilot-scale studies (Mannina et al., 2018; Wunderlin<br>88 et al., 2012). Ho  $N_2O$  production were not studied by additional dissolved  $N_2O$  in the liquid phase because this<br>
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87 knowledge has been provided by other bench- or pilot-scale studies (Mannina et al., 2018; Wunderlin<br>
88 et al., 2012). Ho 90 emitted in the atmosphere. N<sub>2</sub>O production were not studied by additional dissolved N<sub>2</sub>O in the liquid phase because this<br>
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88 et al., 2012). Ho 92 86  $\text{N}_2\text{O}$  production were not studied by additional dissolved  $\text{N}_2\text{O}$  in the liquid phase because this<br>87 knowledge has been provided by other bench- or pilot-scale studies (Mannina et al., 2018; Wunderlin

93 N<sub>2</sub>O production were not studied by additional dissolved N<sub>2</sub>O in the liquid phase because this<br>87 knowledge has been provided by other bench- or pilot-scale studies (Mannina et al., 2018; Wunderlin<br>88 et al., 2012). 94 under continuous aeration ( $DO > 4$  mg/L). After 30 min, the biomass was spiked with ammonium 89 So Drootecton were not studied by additional dissolved N<sub>2</sub>O in the iiquid phase because this<br>87 knowledge has been provided by other bench- or pilot-scale studies (Mannina et al., 2018; Wunderlin<br>88 et al., 2012). How 88 et al., 2012). However, a significant gap of knowledge concerns the full scale WWTPs especially considering the real variable influent characteristics (i.e. C:N) and gaseous mass loads directly emitted in the atmospher 98 ct at., 2012). However, a signmeant gap or knowledge concerns the full seate wwitter sepectaty<br>99 considering the real variable influent characteristics (i.e. C:N) and gaseous mass loads directly<br>90 emitted in the atmo 99 considering the real variable influent characteristics (i.e. C:N) and gaseous mass loads directly<br>90 emitted in the atmosphere.<br>91 Moreover, to monitor the stability of the respiratory activity of the microbial communi 99 subspect of 20 °C using the Arrhenius temperature correction and the nitrate unity, mitrification and denitrification kinetics were an analysed by batch tests. To determine the ammonia utilization rate  $\Delta$ UR, 1.5 L of 91 Moreover, to monitor the stability of the respiratory activity of the microbial community, nitrification<br>
92 and denitrification kinetics were an analysed by batch tests. To determine the ammonia utilization<br>
93 rate A 101 with fixed nitrate concentration and with an external carbon source (acetic acid) and the nitrate 93 rate AUR, 1.5 L of mixed liquor was collected from the aerobic reactor and wa<br>
94 under continuous acration ( $DO > 4$  mg/L). After 30 min, the biomass was spike<br>
95 chloride at 40 mgNH<sub>4</sub>-N/L initial concentration and th 24 under continuous acration (*DO*  $>4$  mg/L). After 30 mm, the blomass was spixed with ammonium<br>25 chloride at 40 mgNH<sub>4</sub>-N/L initial concentration and the profiles of ammonium, nitrite and nitrate with<br>26 time were meas 213 N2O sampling and monitoring strategies<br>
26 September 2014 N2O emissions were conducted at room temperature (25 ± 2 °C)<br>
26 and the pH was maintained at 7.4 ± 0.3. The reported activities were normalized to the referen 105 6 time were measured. All batch respirometry tests were conducted at room temperature (25 + 2 °C)<br>
197 and the pH was maintained at 7.4 ± 0.3. The reported activities were normalized to the reference<br>
198 temperature 2061 97 and the pH was maintained at 7.4 + 0.3. The reported activities were normalized to the reference<br>
2068 temperature of 20<sup>6</sup>°C using the Arrhenius temperature correction equation and to the volatile<br>
2068 suspended

107 was performed. The gas flow Arthenius temperature correction equation and to the volatile<br>199 suspended solids (VSS) of the mixture. The nitrate utilization rate (NUR) tests were conducted with<br>107 1.5 L of activated 108 removal and cooled at 4 $\degree$ C. More than one type of sampling chamber was tested to optimize the N<sub>2</sub>O 1.5 L of activated sludge placed in a flask, under mild agitation. Subsequently, the biomass was spiked<br>101 with fixed nitrate concentration and with an external carbon source (acetic acid) and the nitrate<br>102 profiles we

110 floating. The main characteristics of the different gas chambers are shown in Table 1 and in Table 2<br>111 as 3D images, shapes, volumes and configurations. An open tube is located on the surface to allow<br>112 gas suction 110 floating. The main characteristics of the different gas chambers are shown in Table 1 and in Table 2<br>111 as 3D images, shapes, volumes and configurations. An open tube is located on the surface to allow<br>112 gas suction 112 gas suction. The outlet pipe was the same for the different gas hoods (diameter of 10 cm and length 113 of 1 m). <sup>116</sup> The sampling point to measure the N2O emissions was chosen in the aerobic reactor as basin where

# 114 Table 1

# 115 Table 2

110 floating. The main characteristics of the different gas chambers are shown in Table 1 and in Table 2<br>
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112 gas suct 110 floating. The main characteristics of the different gas chambers are shown in Table 1 and in Table 2<br>
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118 gas sucti 111 as 3D images, shapes, volumes and configurations. An open tube is located on the surface to allow<br>
112 gas suction. The outlet pipe was the same for the different gas hoods (diameter of 10 cm and length<br>
113 of 1 m).<br> 112 gas suction. The outlet pipe was the same for the different gas hoods (diameter of 10 cm and length<br>
113 fable 1<br>
114 Table 1<br>
115 Table 2<br>
116 The sampling point to measure the N<sub>2</sub>O emissions was chosen in the aerob 113 of 1 m).<br>
114 Table 2<br>
116 The sampling point to measure the N<sub>2</sub>O emissions was chosen in the aerobic reactor as basin where<br>
117 the gaseous products were mainly stripped and emitted in the atmosphere. The sampling Table 1<br>
115 Table 2<br>
116 The sampling point to measure the N<sub>1</sub>O emissions was chosen in the aerobic reactor as basin where<br>
117 the gaseous products were mainly stripped and emitted in the atmosphere. The sampling point Table 2<br>
116 The sampling point to measure the N<sub>2</sub>O emissions was chosen in the aerobic reactor as basin where<br>
117 the gaseous products were mainly stripped and emitted in the atmosphere. The sampling point was<br>
118 pla 113<br>
116 The sampling point to measure the N<sub>2</sub>O emissions was chosen in the aerobic reactor as basin where<br>
117 the gaseous products were mainly stripped and emitted in the atmosphere. The sampling point was<br>
118 placed a 116 The sampling point to measure the N<sub>1</sub>O emissions was chosen in the aerobic reactor as basin where<br>
117 the gaseous products were mainly stripped and emitted in the atmosphere. The sampling point was<br>
118 placed at th 117 the gaseous products were manny stripped and emitted in the atmosphere. The sampling point was<br>
118 placed at the head of the reactor in the acrobic basin for 46 days and at the end for 7 days. The fixed<br>
119 chambers placed at the head of the reactor in the aerobic basin for 46 days and at the end for *f* days. The fixed<br>chambers were attached to the external wall by steel clamps and the floating chamber was fastened<br>by ropes. The mini 120 by ropes. The minimum monitoring duration for each gas hood was 7 days. High-density polyethylene (HDPE) was used for the fixed gas hoods and polypropylene (PP) was used for the floating hoods.<br>
122 The base of the fi (HDPE) was used for the fixed gas hoods and polypropylene (PP) was used for the floating hoods.<br>
122 The base of the fixed gas chambers was submersed (about 5 cm) to prevent lateral movement and<br>
123 introduction of exter 122 The base of the fixed gas chambers was submersed (about 5 cm) to prevent lateral movement and<br>
123 introduction of external air. Cylindrical fixed chambers were used with volumes equal to 80 L, 141<br>
124 L and 226 L fo

221 introduction of external air. Cylindrical fixed chambers were used with volumes equal to 80 L, 141<br>
226 L for the small, medium and large chamber. The floating gas chambers had a truncated<br>
225 cone structure and volu 124 L and 226 L for the small, medium and large chamber. The floating gas chambers had a truncated<br>
125 cone structure and volumes equal to 64 L (small), 166 L (medium) and 233 L (big) (Table 2).<br>
126 At the end of the co 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O<br>134 data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>135 tube o 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the  $N_2O$ <br>134 data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>135 tube 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O<br>134 data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>135 tube o 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O<br>134 data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>135 tube o in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O<br>data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>tube of the gases a 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O<br>134 data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the<br>135 tube o is in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the tube of the gas 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the tube of the ga 133 in the same relevant period (from hrs 9.30 to hrs 12:30). Each test was carried out acquiring the N<sub>2</sub>O data for 1 hour with one type of chamber. At the end of the acquisition time the connection of the tube of the ga the stable conditions in the liquid phase were monitored and verified by the analysis of the main<br>
issolved nitrogen and organic forms (NH<sub>4</sub>-N, NO<sub>2</sub>-N, NO<sub>2</sub>-N, CODs). Therefore, the liquid<br>
onditions during the differe

$$
S \quad \left(\frac{L}{m^3/h}\right) = \frac{V\left(\frac{H}{g}S\right)}{A\cdot S}
$$
 Equation 1

Where:

Air Supply  $(m^3/h)$  = Inlet air flow to the aerobic reactor

139 conditions during the different short tests were comparable.<br>
140 The N<sub>2</sub>O data was linked with two variables: 1) the air supply and 2) the Sampler Ratio (SR).<br>
141 This second coefficient was set and calculated for 140 The N<sub>2</sub>O data was linked with two variables: 1) the air supply and 2) the Sampler Ratio (SR).<br>
141 This second coefficient was set and calculated for each test according to equation 1:<br>
143 S  $(L/m^3/h) = \frac{V_L}{A} = \frac{H}{S}$ 144 to the inlet airflow was studied to avoid over-estimation of  $N_2O$  concentrations and/or overpressure 145 phenomena.  $S \left( \frac{L}{m^3/h} \right) = \frac{1}{16} \frac{R}{s}$  Equation 1<br>
Where:<br>
Volume Head Space (L) changes for each type of sampling chamber<br>
Air Supply (m<sup>3</sup>/h) = Inlet air flow to the aerobic reactor<br>
142 For the same type of chamber, the S Where:<br>
Volume Head Space (L) changes for each type of sampling chamber<br>
Air Supply (m<sup>3</sup>/h) = Inlet air flow to the aerobic reactor<br>
142 For the same type of chamber, the SR values dynamically varied because of the chang

Volume Head Space (L) changes for each type of sampling chamber<br>
Air Supply (m<sup>3</sup>/h) = Inlet air flow to the acrobic reactor<br>
142 For the same type of chamber, the SR values dynamically varied because of the change of the 149 (APHA, 2005; Tavakoli et al., 2013) along the periods monitored by the different gas chambers 150 (EventiC; Danishvar et al, 2017). This technique enables the identification of patterns (strength of

151 relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>in</sub>) and gas fluxes (N<sub>2</sub>O).<br>152 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>153 events. I 152 Tangible and reasonable changes to the signals of the sensors in the system were translated into 153 events. In order to track events in a sensor signal the standard deviation of the signal fluctuation for 151 relations) between the monitored variables (DO, blowers flow rate, MLSS,  $Q_{in}$ ) and gas fluxes (N<sub>2</sub>O).<br>152 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>153 events. 155 standard deviations of the variables) and presented the results that maximize the Event-based 156 sensitivity analysis coefficients. The thresholds consider as events the following changes in the 151 relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>m</sub>) and gas fluxes (N<sub>2</sub>O).<br>
152 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>
153 cvents. 151 relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>m</sub>) and gas fluxes (N<sub>5</sub>O).<br>
152 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>
153 events. relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>in</sub>) and j<br>Tangible and reasonable changes to the signals of the sensors in the system were<br>vents. In order to track events in a sensor signal the relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>in</sub>) and gas fluxes (N<sub>2</sub>O).<br>
152 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>
153 events. In 161 relations) between the monitored variables (DO, blowers flow rate, MLSS, Q<sub>m</sub>) and gas fluxes (N<sub>2</sub>O).<br>
1620 Tangible and reasonable changes to the signals of the sensors in the system were translated into<br>
1615 event 161 change in the system and the system response and therefore provides insight on which input variables 162 (i.e. ammonia, DO) impact a specific output (i.e.  $N_2O$ ). The un-biased sensitivity analysis detects and 154 all the time period is calculated. Several thresholds were tested (ranging from 5% - 35% of the standard deviations of the variables) and presented the results that maximize the Event-based sensitivity analysis coeffi 164 the algorithm in the data from the different groups the influential variables in a look-up table. A 165 detailed description of the method can be found in the study of Danishvar et al. (2017). variables: 1) DO ->0.2 mg/t (-1.3% of the standard deviation); ii) N<sub>2</sub>O ->1<br>158 deviation); iii) MLSS> 20 (>5% of the standard deviation); iv) Qin>2<br>159 deviation); v) Blowers flow-rate > 110 m<sup>3</sup>/h (>5% of the standard<br> 158 deviation); iii) MLSS> 20 (>5% of the standard deviation); iv) Qin>20 m3/h (>5% deviation); v) Blowers flow-rate > 110 m<sup>3</sup>/h (>5% of the standard deviation).<br>
160 sensitivity analysis enables the identification of ca 159 deviation); v) Blowers flow-rate > 110 m<sup>3</sup>/h (>5% of the standard deviation). The event-base<br>
160 sensitivity analysis enables the identification of cause-effect relationship between the causes of state<br>
161 change i 160 sensitivity analysis enables the identification of cause-effect relationship between the causes of state<br>
161 change in the system and the system response and therefore provides insight on which input variables<br>
162 (

14,210 $\pm$ 4,652 m<sup>3</sup>/d. The TN concentration in the influent is 28.6 $\pm$ 10.5 mg/L, mainly as ammonium 170 nitrogen (25.1±3.2 mg/L). The average effluent mass loads were 2.87±2.00 and 196.50±86.05 kgN/d 162 (i.e. ammonia, DO) impact a specific output (i.e. N<sub>2</sub>O). The un-biased sensitivity analysis detects and<br>
163 defines the most relevant variables (many to one and many to many relationships) by implementing<br>
164 detai 163 defines the most relevant variables (many to one and many to many relatio<br>
164 the algorithm in the data from the different groups the influential variabl<br>
165 detailed description of the method can be found in the st

# 173 Table 3

174 The low TN removal efficiency is related with the low biodegradable carbon to nitrogen ratio that<br>175 limits the denitrification process. On the other hand, complete nitrification was achieved.<br>176 Additionally, the A 174 The low TN removal efficiency is related with the low biodegradable carbon to nitrogen ratio that<br>
175 limits the denitrification process. On the other hand, complete nitrification was achieved.<br>
176 Additionally, the 174 The low TN removal efficiency is related with the low biodegradable carbon to nitrogen ratio that<br>
175 Iimits the denitrification process. On the other hand, complete nitrification was achieved.<br>
176 Additionally, the 177 was 0.057±0.028 kgNOx-N/kgMLVSS/d. 174 The low TN removal efficiency is related with the low biodegradable carbon to nitrogen ratio that<br>
175 limits the denitrification process. On the other hand, complete nitrification was achieved.<br>
180 Additionally, the 174 The low TN removal efficiency is related with the low biodegradable carbon to nitrogen ratio that<br>
1175 limits the denitrification process. On the other hand, complete nitrification was achieved.<br>
1180 Additionally, t

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174 Inc. tow ITV reintractively is related with the for obsologication can be limited and the dentification process. On the other hand, complete intrification was achieved.<br>
176 Intis the AUR was 0.111±0.024 kgNH<sub>4</sub>-N/kgML 173 mmts (act commincaton process). On the other hands, complete intimetaton was actived.<br>
176 Additionally, the AUR was 0.111+0.024 kgNH<sub>4</sub>-N/kgMLVSS/d and the average denitrification rate<br>
183 3.2 N<sub>2</sub>O emissions and gas 217 was 0.057+0.028 kgNO<sub>2</sub>-N/kgMLVSS/d.<br>
184 was 0.057+0.028 kgNO<sub>2</sub>-N/kgMLVSS/d.<br>
184 3.2 N<sub>2</sub>O emissions and gas chamber headspace: optimization of the sampling methodology<br>
181 The N<sub>2</sub>O concentrations obtained during 178<br>
178<br>
179<br>
180 **3.2 N:O emissions and gas chamber headspace: optimization of the sampling methodology<br>
181 The N<sub>2</sub>O concentrations obtained during the calibration tests for the optimization of the sampling<br>
181 The N** 178<br>
180 **3.2 N:O emissions and gas chamber headspace: optimization of the sampling methodology**<br>
181 The N<sub>2</sub>O concentrations obtained during the calibration tests for the optimization of the sampling<br>
182 methodology ar 179<br>
180 3.2 N<sub>2</sub>O emissions and gas chamber headspace: optimization of the sampling methodology<br>
181 The N<sub>2</sub>O concentrations obtained during the calibration tests for the optimization of the sampling<br>
182 methodology ar 3.2 N<sub>2</sub>O emissions and gas chamber headspace: optimization of the sampling methodology<br>The N<sub>2</sub>O concentrations obtained during the calibration tests for the optimization of the sampling<br>methodology are shown in Figure 1 188 both for floating and for fixed chambers. Differently, the Sampler Ratio  $(SR-L/m<sup>3</sup>/h)$  was calculated 3.2 N:20 emissions and gas channer neadspace: optimization of the sampling methodology<br>
181 The N:0 concentrations obtained during the calibration tests for the optimization of the sampling<br>
189 enthodology are shown in F 181 The N<sub>3</sub>O concentrations obtained during the calibration tests for the optimization of the sampling<br>
182 methodology are shown in Figure 1 for different air flows both for the fixed and for the floating gas<br>
183 chamb influent air flux has been found in other works (Ribeiro et al., 2017). In the eurrent work this<br>behaviour has been observed only for the floating chambers (Figure 1-b). Scattered distribution was<br>found for the fixed cham 184 influent air flux has been found in other works (Ribeiro et al., 2017). In the current work this<br>
185 bchaviour has been observed only for the floating chambers (Figure 1-b). Scattered distribution was<br>
186 found for 185 behaviour has been observed only for the floating chambers (Figure 1-b). Scattered distribution was<br>
186 found for the fixed chambers (Figure 1-a). No evident relation was found between the increment of<br>
187 the N<sub>1</sub>O 186 found for the fixed chambers (Figure 1-a). No evident relation was found between the increment of<br>
187 the N<sub>2</sub>O concentrations and the dimension (Small-Medium-Large) of the chamber used for sampling<br>
186 both for flo

191 Figure 1-a-b-c-d

the N<sub>2</sub>O concentrations and the dimension (Small-Medium-Large) of the chamber used for sampling<br>both for floating and for fixed chambers. Differently, the Sampler Ratio (SR-L/m<sup>3</sup>/h) was calculated<br>according to Equation 187 the N<sub>2</sub>O concentrations and the dimension (Small-Medium-Large) of the chamber used for sampling<br>
198 both for floating and for fixed chambers. Differently, the Sampler Ratio (SR-L/m<sup>3</sup>/h) was calculated<br>
199 accordin 188 both for floating and for fixed chambers. Differently, the Sampler Ratio (SR-L/m<sup>3</sup>/h) was calculated<br>
189 according to Equation 1. The results showed that the N<sub>2</sub>O concentrations are linked with the SR value<br>
190 (F

197 influenced by the SR variations (Figure 1-d); the same floating avoids potential over pressure<br>
198 phenomena in the head-space.<br>
199 33 N<sub>2</sub>O emission profiles during continuous monitoring 198 phenomena in the head-space. 197 influenced by the SR variations (Figure 1-d); the same floating avoids potential over<br>
198 phenomena in the head-space.<br>
199 3.3 N<sub>2</sub>O emission profiles during continuous monitoring<br>
199 The N<sub>2</sub>O emissions rate at th

201 Individually the SR variations (Figure 1-d); the same floating avoids potential over pressure<br>
200 B. 3.3 N<sub>2</sub>O emission profiles during continuous monitoring<br>
200 The N<sub>2</sub>O emissions rate at the head of the reactor v 201 influenced by the SR variations (Figure 1-d); the same floating avoids potential over pressure<br>
201 **3.3 N<sub>2</sub>O emission profiles during continuous monitoring**<br>
202 The N<sub>2</sub>O emissions rate at the head of the reactor v 202 influenced by the SR variations (Figure 1-d); the same floating avoids potential over pressure<br>202 phenomena in the head-space.<br>202 3.3 N<sub>2</sub>O emission profiles during continuous monitoring<br>202 The N<sub>2</sub>O emissions rate

# 203 Figure 2

204 The variability of the daily N<sub>2</sub>O emissions rate can be mainly attributed to the actually variable can be mainly attributed to the N<sub>2</sub>O emissions rate at the head of the reactor varied from 66.82 to 4,174.37 mg/h wi 203 Interacts by the one variables (11gare 1-0), the same notaing avoids potential over possure<br>200 Interaction profiles during continuous monitoring<br>201 Interaction ratios (18.2=0.8 °C in Washer Change 1-0.2) that was no 3.3 N:O emission profiles during continuous monitoring<br>
200 The N<sub>2</sub>O emissions rate at the head of the reactor varied from 66.82 to 4,174.37 mg/h with average<br>
201 load equal to 31.99+24.33 gN<sub>2</sub>O/d during the monitoring 207 **3.3 N:O emission profiles during continuous monitoring**<br>
200 The N<sub>2</sub>O emissions rate at the head of the reactor varied from 66.82 to 4,174.37 mg/h with average<br>
202 load equal to 31.99±24.33 gN<sub>2</sub>O/d during the moni 200 The N<sub>2</sub>O emissions rate at the head of the reactor varied from 66.82 to 4,174.37 mg/h with average<br>
201 load equal to 31.99424.33 gN<sub>2</sub>O/d during the monitoring period (Figure 2). The N2O emission profile<br>
202 was no 201 load equal to 31.99±24.33 gN<sub>2</sub>O/d during the monitoring period (Figure 2). The N2O emission profile<br>202 was not affected by temperature variations (18.2–0.8 °C in wastewater).<br>203 Figure 2<br>204 The variability of the 210 sequencing batch reactors; integrated fixed film activated sludge membrane bioreactor, respectively) 203<br>
204 The variability of the daily N<sub>2</sub>O emissions rate can be mainly attributed to the actually variable<br>
205 influent carbon to nitrogen ratios (COD:TN: 1.3 to 5.2) that was always low. The average N<sub>2</sub>O<br>
2016 emissi Figure 2<br>204 The variability of the daily N<sub>2</sub>O emissions rate can be mainly attributed to the actually variable<br>205 influent carbon to nitrogen ratios (COD:TN: 1.3 to 5.2) that was always low. The average N<sub>2</sub>O<br>206 emiss 204 The variability of the daily N<sub>2</sub>O emissions rate can be mainly attributed to the actually variable<br>
205 influent carbon to nitrogen ratios (COD:TN: 1.3 to 5.2) that was always low. The average N<sub>2</sub>O<br>
206 emission rat 205 influent carbon to nitrogen ratios (COD:TN: 1.3 to 5.2) that was always low.<br>
206 emission rate was equal to  $0.856\pm0.905$  gN<sub>2</sub>O/h when the COD:TN was about<br>
207 while it increased to  $1.850\pm0.972$  gN<sub>2</sub>O/h at lowe 200 cmssion rate was equat to 0.856+0.973 gN<sub>2</sub>O*m* when the COD:1N was about 3.2 (1<sup>1-2</sup>20<sup>-2</sup> days),<br>207 while it increased to 1.850±0.972 gN<sub>2</sub>O/h at lower COD:TN ratio:1.9. The latter is in accordance to<br>208 the resul 2016 monitoring were very low (0.174±0.90 gN<sub>2</sub>O/h) at the end of the reactor. The latter is in accordance to the results reported in literature (Quan et al., 2012; Mannina et al., 2017). Similar limiting C:N ratios resul 208 the results reported in literature (Quan et al., 2012; Mannina et al., 2017). Similar limiting C:N ratios<br>209 resulted in N<sub>2</sub>O increase in previous studies applying different processes (aerobic granular sludge<br>3210 a zoy<br>
218 be action N2O increase in previous studies applying different processes (aerobic granuar studes<br>
218 sequencing batch reactors; integrated fixed film activated sludge membrane bioreactor, respectively)<br>
218 at pl

sequencing batch reactors; integrated integral methods sludge membrane bioreactor, respect<br>at pilot scale. Mannina et al. (2018) demonstrated that limiting C:N ratio of 2 gCOD/gTN result<br>5 times increase of N<sub>2</sub>O emission

220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emission 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emission 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emission 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emission 224 (Aboobakar et al, 2013; Daelman et al., 2013; Rodriguez-Caballero et al., 2014). 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>
221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>
222 emissi 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>
221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>
222 emissi 220 Moreover, the N<sub>2</sub>O emissions, the main operative variables and the daily variations were statistically<br>221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emission

# 225 Figure 3

221 analysed to better understand the role of the liquid variables. The boxplots of the hourly N<sub>2</sub>O<br>222 emissions in the nitrification reactor are shown in Figure 3. N<sub>2</sub>O emissions' dynamics are<br>223 characterized by sig emissions in the nitrification reactor are shown in Figure 3. N<sub>2</sub>O emission<br>
characterized by significant daily variability in accordance with the results of<br>
224 (Aboobakar et al., 2013; Daelman et al., 2013; Rodriguezcharacterized by significant daily variability in accordance with the results of previous studies<br>
224 (Aboobakar ct al., 2013; Daclman ct al., 2013; Rodriguez-Caballero ct al., 2014).<br>
225 Figure 3<br>
226 The minimum daily (Aboobakar et al, 2013; Daelman et al., 2013; Rodriguez-Caballero et al., 2014).<br>
225 Figure 3<br>
226 Figure 3<br>
226 Figure 3<br>
228 The minimum daily N<sub>2</sub>O fluxes are observed between 03:00 am and 10:00 am, while a subsequent

233 17.5 and 19.4 mg/L for ammonia nitrogen and between 61.6 and 20.1 mg/L for COD were observed 226 The minimum daily N<sub>2</sub>O fluxes are observed between 03:00 am and 10:00 am, while a subsequent<br>
227 peak occurs between 18:00 pm and 20:00 pm. No specific correlation between the N<sub>2</sub>O emissions<br>
228 and the liquid inf 236 In eminimum daily N<sub>2</sub>O fluxes are observed between 03:00 am and 10:00 am, while a subsequent<br>
227 peak occurs between 18:00 pm and 20:00 pm. No specific correlation between the N<sub>2</sub>O emissions<br>
228 and the liquid inf 236 emissions during the day. 239 Figure 4<br>
239 Figure 4. Hourly variability is shown in Figure 4. Hourly variations between<br>
232 concentrations. A typical example of daily variability is shown in Figure 4. Hourly variations between<br>
233 17.5 and 19.4 232 concentrations. A typical example of daily variability is shown in Figure 4. Hourly variations between<br>
233 17.5 and 19.4 mg/L for ammonia nitrogen and between 61.6 and 20.1 mg/L for COD were observed<br>
234 with peaks

# 237 Figure 4

238 The cumulative emitted N<sub>2</sub>O mass loads  $(LN_2O)$  and the influent TN  $(LTN)$  values are shown in 233 17.5 and 19.4 mg/L for ammonia nitrogen and between 61.6 and 20.1 mg/L for COD were observed<br>
234 with peaks of COD:TN during 12:00-16:00, where the N<sub>2</sub>O distribution showed almost minimum<br>
235 N<sub>2</sub>O emissions. The l with peaks of COD:TN during 12:00-16:00, where the N2O distribution showed almost minimum<br>
N<sub>2</sub>O emissions. The latter reveals that the variation of the COD:TN ratio strongly affects the N<sub>2</sub>O<br>
emissions during the day.<br>

243 emissions are observed when the influent COD:N ratio is higher than 4. The lower was the COD:N<br>244 ratio the higher was the emitted N<sub>2</sub>O: about 5 times higher compared to the periods with higher<br>245 COD:TN ratio (0.0 243 emissions are observed when the influent COD:N ratio is higher than 4. The lower was the COD:N<br>244 ratio the higher was the emitted N<sub>2</sub>O: about 5 times higher compared to the periods with higher<br>245 COD:TN ratio (0.0 245 COD:TN ratio (0.0505 gN2O/kgTN, R2=0.8853). 243 emissions are observed when the influent COD:N ratio is higher than 4. The lower was the COD:N<br>
244 ratio the higher was the emitted N<sub>2</sub>O: about 5 times higher compared to the periods with higher<br>
245 COD:TN ratio (0

247 The biomass-based EF was equal to  $2.11 \pm 0.98$  and  $5.01 \pm 2.09$  mgN<sub>2</sub>O/kgMLVSS/d for the first and 243 emissions are observed when the influent COD:N ratio is higher than 4. The lower was the COD:N<br>
244 ratio the higher was the emitted N<sub>2</sub>O: about 5 times higher compared to the periods with higher<br>
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244 ratio the higher was the emitted N<sub>2</sub>O: about 5 times higher compared to the periods with higher<br>
245 COD:TN ratio (0 252 TN in a plug-flow reactor. 245 COD: TN ratio (0.0505 gN<sub>2</sub>O/kgTN, R2=0.8853).<br>
246 Figure 5<br>
247 The biomass-based EF was equal to 2.11±0.98 and 5.01±2.09 mgN<sub>2</sub>O/k<br>
248 second period with different COD:N ratios. The EF of current study i<br>
249 repo 246 Figure 5<br>
247 The biomass-based EF was equal to 2.11±0.98 and 5.01±2.09 mgN<sub>2</sub>O/kgMLVSS/d for the first and<br>
248 second period with different COD:N ratios. The EF of current study is lower than the EF values<br>
249 repo 247 The biomass-based EF was equal to 2.11±0.98 and 5.01±2.09 mgNzO/kgMLVSS/d for the first and<br>248 second period with different COD:N ratios. The EF of current study is lower than the EF values<br>249 reported in other stud 248 second period with different COD:N ratios. The EF of current study is lower than the EF values<br>
249 reported in other studies that monitor on-line gaseous emissions at full-scale. Yan et al., 2014, found<br>
250 emission reported in other studies that monitor on-line gaseous emissions at full-scale. Yan et al., 2014, found<br>emission factors ranging from 0.04 to 0.1% of the TN influent for an Anaerobie-Anoxie-Oxie system.<br>Similarly, Rodrigu

257 when the floating hood was applied with SR higher than  $0.05$  L/m<sup>3</sup>/h. 259 Constrained States and States relationship was carried out in order to dentify the relationship between the<br>259 An-event based sensitivity analysis was carried out in order to identify the relationship between the<br>259

# 258 Table 4

252 258<br>
261 concentration in the monitored parameters. The results are given Table 4. The gaseous emissions<br>
260 emissions and the monitored parameters. The results are given Table 4. The gaseous emissions<br>
260 from the 253 3.2 Statistical and sensitivity analysis<br>
254 An-event based sensitivity analysis was carried out in order to identify the relationship between the<br>
255 N2O emissions and the monitored parameters. The results are give 263 An-event based sensitivity analysis was carried out in order to identify the relationship between the<br>255 N2O emissions and the monitored parameters. The results are given Table 4. The gaseous emissions<br>256 from the n 255 N2O emissions and the monitored parameters. The results are given Table 4. The gaseous emissions<br>256 from the nitrification reactor have been examined with reference to the variables monitored online<br>257 when the floa 264 2264 226 226 Transmission reactor have been examined with reference to the variables monitored online<br>257 when the floating hood was applied with SR higher than 0.05 L/m<sup>3</sup>/h.<br>269 1. Table 4<br>269 1. Table 4<br>269 1.5 – 2 257 when the floating hood was applied with SR higher than 0.05 L/m<sup>3</sup>/h.<br>
258 Table 4<br>
259 A weak relationship was identified between the N<sub>2</sub>O emissions, the air flow-rate and the DO<br>
260 concentration in the reactor. I

266 According to event-based sensitivity analysis, the blowers' flow-rate affects the N<sub>2</sub>O emission fluxes.<br>267 The typical N<sub>2</sub>O emissions (g/h) profile is shown in Figure 6 with the aeration flow-rate and with the<br>268 266 According to event-based sensitivity analysis, the blowers' flow-rate affects the N<sub>2</sub>O emission fluxes.<br>267 The typical N<sub>2</sub>O emissions (g/h) profile is shown in Figure 6 with the aeration flow-rate and with the resi 266 According to event-based sensitivity analysis, the blowers' flow-rate affects the N<sub>2</sub>O emission fluxes.<br>
267 The typical N<sub>2</sub>O emissions (*g*/h) profile is shown in Figure 6 with the acration flow-rate and with the<br> 266 According to event-based sensitivity analysis, the blowers' flow-rate affects the N<sub>2</sub>O emission fluxes.<br>267 The typical N<sub>2</sub>O emissions (g/h) profile is shown in Figure 6 with the acration flow-rate and with the<br>268 266 According to event-based sensitivity analysis, the blowers' flow-rate affects the N<sub>2</sub>O emission fluxes.<br>
267 The typical N<sub>2</sub>O emissions (g/h) profile is shown in Figure 6 with the acration flow-rate and with the<br>
26 4.1±1.6 mgDO/l during 21st-45th 273 and the constant nitrigate of virth the aeration flow-rate and with the residual DO for two days of monitoring. Low concentrations of residual dissolved oxygen was not a limiting factor 273 days  $4.1 \pm 1.6$  mg/l). 267 The typical N<sub>2</sub>O emissions (g/h) profile is shown in Figure 6 with the aeration flow-rate and with the<br>
268 residual DO for two days of monitoring. Low concentrations of residual dissolved oxygen was not a<br>
270 limit residual DO for two days of monitoring. Low concentrations of residual dissolved oxygen was<br>
limiting factor (Figure 6). The latter supports the obtained results during the experimental camp<br>
considering the low-impact of 268 residual DO for two days of monitoring. Low concentrations of residual dissolved oxygen was not a<br>
269 limiting factor (Figure 6). The latter supports the obtained results during the experimental campaign<br>
270 conside 279 limiting factor (Figure 6). The latter supports the obtained results during the experimental campaign<br>
270 considering the low impact of the DO concentrations (4.610.7 mg DO /l during  $1^{n-20^{th}}$  days and<br>
4.1±1.6 mgD

# 274 **Figure 6** Figure 6

278 increase N2O stripping phenomena and related N2O emissions. 271 4.1±1.6 mgDO/1 during 21<sup>st</sup>-45<sup>th</sup> days) and the constant nitrification rates<br>
272 N/kgMLVSS/d during 1<sup>st</sup>-20<sup>th</sup> days and kn of 0.118±0.031 kgNH<sub>4</sub>-N/k<br>
273 days 4.1±1.6 mg/l).<br>
274 Figure 6<br>
275 The dynamics of th 280 A full scale activated sludge plant treating low carbon:nitrogen ratio municipal wastewater vastes increase N2O stripping phenomena and related N2O emissions.<br>280 A full scale activated sludge plant treating low carbo The dynamics of the variables are different. However, daily peaks of N<sub>2</sub>O emissions occurred when<br>
the aeration flow-rate was higher than 3,500 m<sup>3</sup>/h. Therefore, although higher N2O emissions were<br>
related to the low C:N 284 a) the optimization of the sampling methods was carried out by testing different types of chambers.<br>284 a) the same sample method of the sampling phenomena and related N2O emissions.<br>284 **4 Conclusions**<br>284 A full s

281 continuously analysed for 52 days to study N2O emissions.

282 This long-term continuous critical monitoring led to the following conclusions related to: a) the

278 Incertainties of the N<sub>2</sub>O concentrations were observed when the Sampling Patition and Telectiviste plant treating low carbon:nitrogen ratio municipal wastewater was<br>280 A full scale activated sludge plant treating lo **Conclusions**<br>**A full scale activated sludge plant treating low carbon:nitrogen ratio municipal wastewater was<br>continuously analysed for 52 days to study N2O emissions.<br>This long-term continuous critical monitoring le 4 Conclusions**<br>
286 A full scale activated sludge plant treating low carbon:nitrogen ratio municipal wastewater was<br>
281 continuously analysed for 52 days to study N2O emissions.<br>
282 This long-term continuous critic **4 Conclusions**<br>280 A full scale activated sludge plant treating low carbon:nitrogen ratio municipal<br>281 continuously analysed for 52 days to study N2O emissions.<br>282 This long-term continuous critical monitoring led

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288 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 288 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 288 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 291 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>291 to 292 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>292 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 293 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 294 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>290 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 295 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen ratio<br>299 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to 298 b) the N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to nitrogen rat mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and le to 5-times hi 298 of Me N<sub>2</sub>O load emitted directly from the aeration basin was related to the carbon to introgen ratio<br>289 mainly and to the variability of the influent load. Low COD:N ratio limited the denitrification and led<br>290 to manny and to the variability of the influent toad. Low COD: N ratio immted the dentification and ice<br>
298 to 5-times higher N<sub>2</sub>O emissions. Major differences were observed around the COD: N = 4:<br>
291 0.856±0.905 gN<sub>2</sub>O/h 290 to 3-times ingner NgO emissions. Major direcrences were observed around the COD:TN = 4:<br>
291 0.856+0.905 gN<sub>2</sub>O/h when COD:TN > 4 versus 1.850+0.972 gN<sub>2</sub>O/h when COD:TN < 4. The<br>
31 0.856+0.905 gN<sub>2</sub>O/h when COD:TN > 203 emissions are higher when the COD:N ratio is lower. The sensitivity analysis showed that the N:O<br>
204 dynamics are not significantly affected by DO variations (within the range of  $1.5 - 2$  mg/L). However,<br>
201 daily p

# 300 Acknowledgements

dynamics are not significantly affected by DO variations (within the range of  $1.5 - 2$  mg/L). However,<br>
daily peaks of N<sub>2</sub>O emissions are observed at higher aeration flow-rate that result in higher stripping<br>
of the prod 303 acknowledge the Royal Society for the funding of the current research: Ad-Bio, Advanced 296 of the produced and dissolved N<sub>2</sub>O.<br>
297 Finally, when COD:N ratio was higher than 4, the cumulative emitted N<sub>2</sub>O mass loads (EF) varied<br>
304 from 0.051 gN<sub>2</sub>O/kgTN<sub>mluen</sub> to 0.0089 gN2O/kgTN<sub>mluen</sub>. Therefore, the 297 Finally, when COD:N ratio was higher than 4, the cumulative emitted N<sub>2</sub>O mass loads (EF) varied<br>298 from 0.051 gN<sub>2</sub>O/kgTN<sub>infleent</sub> to 0.0089 gN2O/kgTN<sub>infleen</sub>. Therefore, the equalization of the influent<br>209 can b 306 (Ancona, Italy).

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Table 2: 3D Images, shapes and volumes of the sampling gas chambers



Table 3: Daily influent and effluent characteristics and kinetic rates (Average and Standard Deviation)

tivity analysis algorithm grouping the system parameters in which coincided (dark grey: high impact, light grey: moderate impact).		
	$N_2O$	
	(ppm) 0.52	
$\overline{Qin(m^3/h)}$	0.37	
DO(mg/L)	0.44	
Blowers flow-rate $(m^3/h)$ <b>MLSS</b>	0.39	

Table 4: Event-based sensitivity analysis algorithm grouping the system parameters in which<br>events have systematically coincided (dark grey: high impact, light grey: moderate impact).<br> $N_2O$ <br>(ppm) events have systematically coincided (dark grey: high impact, light grey: moderate impact).

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List of Figures<br>Figure 1: N<sub>2</sub>O emissions during the tests for the calibration of the sampling chambe<br>Figure 2: N<sub>2</sub>O emissions in nitrification reactor<br>interquarile range, whiskers: lines extending from the 5<sup>th</sup> to 95<sup>t</sup> **Figure 3:** Boxplots of the daily variability of N2O emissions and Influent Flow (grey boxes: interquartile range, whiskers: lines extending from the  $5<sup>th</sup>$  to  $95<sup>th</sup>$  percentile, median: line across the **Example 1:** N<sub>2</sub>O emissions during the tests for the calibration of the sampling chambers<br>**Figure 2:** N<sub>2</sub>O emissions in nitrification reactor<br>**Figure 3:** Boxplots of the daily variability of N2O emissions and Influent F box; grey triangles: average liquid influent flow rate) **Example 1:** N<sub>2</sub>O emissions during the tests for the calibration of the sampling chambers<br> **Figure 2:** N<sub>2</sub>O emissions in nitrification reactor<br> **Figure 3:** Boxplots of the daily variability of N2O emissions and Influent

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Figure 6: Profile of the  $N_2O$  emissions, air flow-rate and DO data for the nitrification reactor



Figure 1:  $N_2O$  emissions during the tests for the calibration of the sampling chambers





interquartile range, whiskers: lines extending from the 5th to 95th percentile, median: line across the



flow



Deviations).

