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# **FORMULATION AND CHARACTERIZATION OF TRANSDERMAL DRUG DELIVERY SYSTEM OF AMILORIDE**

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## **Abstract:**

*Objective: This study aims to develop and evaluate transdermal drug delivery systems (TDDS) for Amiloride, a potassium-sparing diuretic, employing various polymers like HPMC K100M, Polyvinyl Alcohol and Polyvinylpyrrolidone to enhance drug delivery efficiency, control release, and improve patient compliance.*

*Methods: Several formulations of Amiloride transdermal patches were developed using different polymers, including hydrogels, polyurethanes, and acrylics. The formulations were characterized based on their physical properties (e.g., thickness, mechanical strength, adhesive properties), drug release kinetics, and skin permeation rates. In vitro release studies were conducted using Franz diffusion cells.*

*Results: The study successfully formulated Amiloride transdermal patches with various polymers, each influencing the drug release profile and adhesive properties differently. Optimal formulations demonstrated controlled and sustained release of Amiloride, with some polymers providing superior performance in drug permeability and release rate.* 

*Conclusion: The developed transdermal systems showed promise in enhancing the delivery and efficacy of Amiloride through controlled release mechanisms. The choice of polymer significantly impacted the performance of the transdermal patches, suggesting potential for improved patient compliance and therapeutic outcomes. Keywords: Amiloride, HPMC K100M, Polyvinyl Alcohol and Polyvinylpyrrolidone*

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#### **INTRODUCTION:**

Drug delivery system (DDS) is a generic term for a series of physicochemical technologies that can control delivery and release of pharmacologically active substances into cells, tissues and organs, such that these active substances could exert optimal effects [\[1,](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR1) [2\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR2). In other words, DDS covers the routes of administration and drug formulations that efficiently deliver the drug to maximize therapeutic efficacy while minimizing any side effect  $[3-5]$  $[3-5]$ . Depending on the delivery route, there are many types of administration modalities, such as oral administration, transdermal administration, lung inhalation, mucosal administration, and intravenous injection. Among them, the transdermal drug delivery system (TDDS) represents an attractive approach.

TDDS has become one of the most widely investigated routes of noninvasive drug delivery into the body through the skin, unlike conventionally used direct administration routes that make use of needle-based injections. TDDS has significantly influenced the delivery of various therapeutic agents, especially in pain management, hormonal therapy, and treatment of diseases of the cardiovascular and central nervous systems [\[6](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR6)[–9\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR9). TDDS does not involve passage through the gastrointestinal tract; therefore, there is no loss due to first-pass metabolism, and drugs can be delivered without interference from pH, enzymes, and intestinal bacteria. In addition, TDDS can be used to control drug release according to usage restrictions, thereby contributing to the high persistence of this method. Most importantly, because TDDS is a noninvasive administration method and involves minimal pain and burden on the patient, drugs can be safely and conveniently administered to children or the elderly  $[10-12]$  $[10-12]$ .

However, it still does not utilize its full potential due to the innate skin barrier. The skin is the outermost organ with a multi-layered structure, and the role of the skin is to protect our body by blocking environmental hazards such as chemicals, heat, and toxins [\[13,](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR13) [14\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR14).

First, the skin barrier effect of the epidermis occurs in the stratum corneum, the outermost layer, and is a property of blocking external substances. The barrier effect is very significant in the transport of substances having a large molecular weight. In TDDS, it is generally accepted that the delivery of substances with small molecular weights utilizes the intracellular pathway. However, for substances having a large molecular weight, methods and various mechanisms using the intracellular pathway in addition to the intercellular pathway are introduced and used [\[15](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR15)[–17\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR17). This is due to the structure of the skin because the part called lipid containing both cells and hydrophilic substances and hydrophobic substances does not have a perfectly regular position but exists with regularity [\[18\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR18). These structural features can be explained by the principles of physicochemical properties that are attempted to enhance drug delivery through the skin. Next, the vascular system in the dermal layer can inhibit transdermal delivery. A one-cell-thick layer of endothelial cells terminating in the papillary loops of the superficial arteriovenous plexus near the dermalepidermal junction in the upper dermis represents the interface between the tissues surrounding the skin and the human vasculature. The role of the endothelium in the skin is like that of the whole body. It actively responds to pressure, shear, osmotic pressure, heat, chemokines, and cytokines by modulating permeability and inducing vasodilation or constriction [\[19\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR19). Therefore, the biggest issue of TDDS is to resolve the barrier effect of the stratum corneum, deliver the drug to the skin tissue, and pass through the cellular and vascular tissue to reach the target tissue. The problem is that only a small amount of the drug can be delivered through the skin tissue [\[20,](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR20) [21\]](https://www.ncbi.nlm.nih.gov/pmc/articles/PMC8317283/#CR21).

The main aim of the project is to develop and evaluate a transdermal drug delivery system (TDDS) for Amiloride using various polymers to enhance drug delivery efficiency, control release, and improve patient compliance.

## **MATERIALS AND METHODS:**





#### **METHODOLOGY: Analytical method development: A.UV scan:**

A 100mg of Amiloride was accurately weighed and was first dissolved in 35ml methanol solution. The solution was then diluted using phosphate buffer (pH-7.4) to 100 ml. (stock solution-I). Take 10ml solution from stock solution 1 and volume make up to 100ml with phosphate buffer to get  $100\mu\text{g/ml}$  concentrations (stock solution-II). Take 10 ml solution from stock II and volume make up to 100 ml with buffer to get 10µg/ml. 10µg/ml solution was scanned from 200- 400nm.

#### **B. Construction of calibration curve:**

A 100mg of Amiloride was accurately weighed and was first dissolved in 35ml methanol solution. The solution was then diluted using phosphate buffer (pH-7.4) to 100 ml. (stock solution-I). Take 10ml solution from stock solution 1 and volume make up to 100ml with phosphate buffer to get 100 µg/ml concentrations (stock solution-II). It was further diluted with phosphate buffer pH – 7.4 to get solutions in concentration range of 2,4,6,8and 10μg /ml. The absorbance's of these solutions were determined spectrophotometrically at 284 nm.

## **Pre formulation study**

## **A. Colour, Odour, Taste and Appearance:**

The drug sample was evaluated for its Colour, odour and appearance.

#### **B. Melting point determination:**

Melting point of the drug sample was determined by capillary method by using melting point apparatus.

#### **C. Determination of solubility:**

The solubility of Amiloride was determined by adding excess amount of drug in the solvent.

The solubility was determined in distilled water and phosphate buffer pH 7.4. The procedure can be detailed as follows.

Saturated solution of Amiloride prepared using 10 ml. of distilled water/ phosphate buffer pH 7.4 in 25 ml volumetric flasks in triplicate. Precaution was taken so that the drug remains in medium in excess. Then by using mechanical shaker, the flasks were shaken for 48 hours. The sample withdrawn (1 ml after filtration) was diluted with appropriate medium and analyzed by using UV spectrophotometer at 280 nm and 284 nm for phosphate buffer and distilled water respectively.

#### **Formulation of transdermal patches Preparation of blank patches:**

Polymers of single or in combination were accurately weighed and dissolved in respective solvent and then casted in a Petri-dish with mercury as the plain surface. The films were allowed to dry overnight at room temperature.

## **Formulation of drug incorporated transdermal patches:**

The matrix-type transdermal patches containing Amiloride were prepared using different concentrations of HPMC K100 M, Ethyl Cellulose and Polyvinylpyrrolidone. The polymers in different concentrations were dissolved in the respective solvents. Then the drug was added slowly in the polymeric solution and stirred on the magnetic stirrer to obtain a uniform solution. Dimethylsulphoxide was used as plasticizers. Then the solution was poured on the Petri dish having surface area of 78 cm2 and dried at the room temperature. Then the patches were cut into  $2x2 \text{ cm}^2$  patches. Drug incorporated for each  $2x2$  $\text{cm}^2$  patch. The formulation table is given in table no. 2.



**Table 2: Formulation of Amiloride patches**

## **Evaluation parameters of patches Physical evaluations a. Thickness**

The thickness of patches was measured by digital Verniers calipers with least count 0.001mm. The thickness uniformity was measured at five different sites and average of five readings was taken with standard deviation.

## **b. Folding endurance**

The folding endurance was measured manually for the prepared patches. A strip of patch (4x3 cm) was cut evenly and repeatedly folded at the same place till it broke. The number of times the film could be folded at the same place without breaking gave the exact value of folding endurance.

## **c. Weight variation**

The three disks of  $2*1$  cm<sup>2</sup> was cut and weighed on electronic balance for weight variation test. The test was done to check the uniformity of weight and thus check the batch- to- batch variation.

## **d. Drug content Determination**

The prepared drug contained patches specified surface area  $(2 \text{ cm}^2)$  were cut and dissolved in  $(5\% \text{ of }$ methanol contained) 100ml of pH 7.4 phosphate buffer, and vigorously shaked for 12hrs, and then sonicated for 15 minutes, centrifuged at 5000 rpm for 30 min. Filter the drug contained polymeric solution through 42 number whatmann filter paper, then 1ml of the filtrate was taken in a test tube and dilute it for five times with same solvent by using double beam Uv-Visible spectrophotometer to determined drug content

at  $\lambda$ max 284 nm. Respected Placebo patch was taken as a blank solution.

**Flatness:** A transdermal patch should possess a smooth surface and should not constrict with time. This can be demonstrated with flatness study. For flatness determination, one strip is cut from the centre and two from each side of patches. The length of each strip is measured and variation in length is measured by determining percent constriction. Zero percent constriction is equivalent to 100 percent flatness.

% constriction  $=$  I1  $-$  I2 X 100

 $I2 =$  Final length of each strip  $I1 =$ Initial length of each strip

## *In-vitro* **drug diffussion study:**

The *in vitro* study of drug permeation through the semi permeable membrane was performed using a Franz type glass diffusion cell. The modified cell having higher capacity (25 ml) is used to maintain sink condition. This membrane was mounted between the donor and receptor compartment of a diffusion cell. The transdermal patch was placed on the membrane and covered with aluminum foil. The receptor compartment of the diffusion cell was filled with isotonic phosphate buffer of pH 7.4. The hydrodynamics in the receptor compartment were maintained by stirring with a magnetic bead at constant rpm and the temperature was maintained at  $37\pm0.5$ °C. The diffusion was carried out for 12 h and 1 ml sample was withdrawn at an interval of 1 h. The receptor phase was replenished with an equal volume of phosphate buffer at each sample withdrawal. The samples were analyzed for drug content spectrophotometrically at 284 nm.

#### **Drug release kinetics:**

Diffusion data of above two methods was fitted in Zero order, First order and Higuchi equations. The mechanism of drug release was determined by using Higuchi equation.

#### **Zero-Order Kinetics**:

 Zero order as cumulative amount of Percentage drug released vs. time

C=K0t

Where K0 is the zero-order rate constant expressed in units of concentration/time and t is the time in hours. A graph of concentration vs time would yield a straight line with a slope equal to K0 and intercept the origin of the axes.

## **First order kinetics:**

First order as log cumulative percentage of log (%) cumulative drug remaining vs time,

L o g C = L o g C o  $-k$  t / 2.303 Where C0 is the initial concentration of drug, k is the first order constant, and t is the time.

#### **Higuchi model:**

 Higuchi's model as cumulative percentage of drug released vs square root of time

 $Q = K t 1/2$ 

Where K is the constant reflecting the design variables of the system and t is the time in hours. Hence, drug release rate is proportional to the reciprocal of the square root of time.

#### **Kors meyer Peppas equations**:

 Korsmeyer peppas equation used to determine the mechanism of drug release form the

polymer matrix of the tablet. Log cumulative percentage of drug released VS Log time, and the exponent n was calculated through the slope of the straight line.

#### Mt/M∞=Ktn

Where  $Mt/M\infty$  is the fractional solute release, t is the release time, K is a kinetic constant characteristic of the drug/polymer system, and n is an exponent that characterizes the mechanism of release of tracers. For cylindrical matrix tablets, if the exponent  $n = 0.45$ , then the drug release mechanism is Fickian diffusion, and if  $0.45 < n < 0.89$ , then it is non-Fickian or anomalous diffusion. An exponent value of 0.89 is indicative of Case-II Transport or typical zero-order release.

#### **Compatibility study**

#### **FTIR study:**

The infrared spectrum of the pure Amiloride n sample was recorded and the spectral analysis was done. The dry sample of drug was directly placed after mixing and triturating with dry potassium bromide.

#### **RESULTS AND DISCUSSION:**

Initially the drug was tested by UV to know their significant absorption maximum which can be used for the diffusion study of the drug.

#### **Analysis of drug:**

**A. UV scans:**

The lambda max of Amiloride was found to be 284 nm.

**B. construction of calibration curve:**



**Table 3: Standard graph of Amiloride**



**Figure 1: Standard calibration curve of Amiloride**

#### **Pre formulation study**

Totally, Nine formulation trials were done with the aim to achieve the successful matrix type Amiloride transdermal patches. The blend trials prepared for the drug was evaluated for various physical parameters and content uniformity of drug by UV.

#### **A. Colour, odour, taste and appearance**

**Table 4: Results of identification tests of drug**

| <b>Parameter</b> | Amiloride       |  |  |  |  |  |  |  |
|------------------|-----------------|--|--|--|--|--|--|--|
| Color            | White           |  |  |  |  |  |  |  |
| Odor             | <b>Odorless</b> |  |  |  |  |  |  |  |
| Taste            | <b>Bitter</b>   |  |  |  |  |  |  |  |
| Appearance       | A white powder  |  |  |  |  |  |  |  |

#### **B. Melting point determination:**

**Table 5: Results of melting point determination tests of drug**

| $\blacksquare$<br>Drug | Reported melting point            |  |  |  |  |
|------------------------|-----------------------------------|--|--|--|--|
|                        | $\circ$<br>240 <sup>-1</sup><br>◡ |  |  |  |  |

#### **C. Determination of solubility:**



## **Evaluation of Patch**

The formulations A1 to A9 were varying in thickness when compared to other formulations which is due to the variation in the polymer concentration. Which shows the increase in polymer concentration increases the thickness of patch. For all other formulations it was found to be in between  $1.047\pm0.001$  to  $1.051\pm0.004$  mm.

All formulations from A1 to A9 shows weight variation in between  $80 \pm 9.58$  to  $89 \pm 4.69$  mg.

Folding endurance from formulations A1 to A9was found to be in between  $90 \pm 1.84$  to  $99 \pm 2.15$  which can withstand the folding of the skin.



## All formulations showed % drug content from  $96.2 \pm 3.87$  to  $99.74 \pm 1.57$ .

## *In vitro* **diffusion study:**

All the formulation *in vitro* diffusion study was carried out by using Franz type diffusion cell under specific condition such as temp maintained at  $32 \pm 0.5$  °C. The diffusion was carried out for 12 h and 5 ml sample was withdrawn at an interval of 1 h.



**Table 8:** *In vitro* **drug permeation of Amiloride**



**Figure 2: Cumulative % drug permeation of Amiloride patch (A1 to A3)**

The formulations A1 to A3 were prepared by different concentrations of HPMC K100 M (5,10,15mg in  $2*2$  cm<sup>2</sup>patch, the drug release or drug permeation from the patch was dependence on the concentration of polymer in the matrix. At low polymer concentration the drug permeation is more within 12 hours it was total amount of drug was permeated. The 15mg concentration of polymer was showed maximum drug released at 12 hrs. 88.42%. Hence in that 3 formulations showed total drug release at desired time period.



**Figure 3: Cumulative % drug permeation of Amiloride patch (A4 to A6)**

The formulations A4to A6were prepared by different concentrations of Polyvinyl Alcohol (5, 10, and 15) in  $2*2$  cm<sup>2</sup> patch the drug release or drug permeation from the patch was dependence on the concentration of polymer in the matrix. The 10mg (A4) concentration of polymer was showed maximum drug release 99.92 within 12 hours. The 15mg (A4) concentration of polymer was showed maximum drug released at 12 hours 99.92%.



**Figure 4: Cumulative % drug permeation of Amiloride patch (A7 to A9)**

The formulations A7to A9 were prepared by different concentrations of Polyvinylpyrrolidone (5, 10, and 15mg) in 2\*2 cm<sup>2</sup>patch the drug release or drug permeation from the patch was dependence on the concentration of polymer in the matrix. The 15mg (A9) concentration of polymer was showed maximum drug release 95.63 within 12 hours. The 15mg (A9) concentration of polymer was showed maximum drug released at 12 hours 95.63%.

Among all 9 formulations A4 formulation showed good drug permeation from the patch. Among all *in vitro* evaluation parameters A4 formulation passed all evaluation parameters.

## **Kinetic models for Amiloride**

Various models were tested for explaining the kinetics of drug release. To analyze the mechanism of the drug release rate kinetics of the dosage form, the obtained data were fitted into zero-order, first order, Higuchi, and Korsmeyer-Peppas release model.

| CUMULATIVE(%)<br><b>RELEASE Q</b> | TIME(T)          | ROOT(T)          | LOG(%) RELEASE | LOG(T) | $LOG(\% )$<br><b>REMAIN</b> | <b>RELEASE</b><br>RATE<br>(CUMULATIVE<br>% RELEASE/t) | 1/CUM%<br>RELEASE log Q/100 | <b>PEPPAS</b> | % Drug<br>Remaining | Q01/3 | Qt1/3 | $Q01/3-$<br>Qt1/3 |
|-----------------------------------|------------------|------------------|----------------|--------|-----------------------------|---|-----------------------------|---------------|---------------------|-------|-------|-------------------|
| $\theta$                          | $\theta$         | $\boldsymbol{0}$ |                |        | 2.000                       |   |                             |               | 100                 | 4.642 | 4.642 | 0.000             |
| 21.63                             |                  | 1.000            | 1.335          | 0.000  | 1.894                       | 21.630  | 0.0462                      | $-0.665$      | 78.37               | 4.642 | 4.279 | 0.362             |
| 28.84                             | $\overline{c}$   | 1.414            | 1.460          | 0.301  | 1.852                       | 14.420  | 0.0347                      | $-0.540$      | 71.16               | 4.642 | 4.144 | 0.498             |
| 34.42                             | 3                | 1.732            | 1.537          | 0.477  | 1.817                       | 11.473  | 0.0291                      | $-0.463$      | 65.58               | 4.642 | 4.033 | 0.609             |
| 49.28                             | 4                | 2.000            | 1.693          | 0.602  | 1.705                       | 12.320  | 0.0203                      | $-0.307$      | 50.72               | 4.642 | 3.702 | 0.940             |
| 51.13                             | 5                | 2.236            | 1.709          | 0.699  | 1.689                       | 10.226  | 0.0196                      | $-0.291$      | 48.87               | 4.642 | 3.656 | 0.986             |
| 67.71                             | 6                | 2.449            | 1.831          | 0.778  | 1.509                       | 11.285  | 0.0148                      | $-0.169$      | 32.29               | 4.642 | 3.184 | 1.457             |
| 74.99                             | $\overline{7}$   | 2.646            | 1.875          | 0.845  | 1.398                       | 10.713  | 0.0133                      | $-0.125$      | 25.01               | 4.642 | 2.924 | 1.717             |
| 78.16                             | 8                | 2.828            | 1.893          | 0.903  | 1.339                       | 9.770   | 0.0128                      | $-0.107$      | 21.84               | 4.642 | 2.795 | 1.846             |
| 85.05                             | 9                | 3.000            | 1.930          | 0.954  | 1.175                       | 9.450   | 0.0118                      | $-0.070$      | 14.95               | 4.642 | 2.463 | 2.178             |
| 89.27                             | 10 <sup>10</sup> | 3.162            | 1.951          | 1.000  | 1.031                       | 8.927   | 0.0112                      | $-0.049$      | 10.73               | 4.642 | 2.206 | 2.436             |
| 94.13                             | 11               | 3.317            | 1.974          | 1.041  | 0.769                       | 8.557   | 0.0106                      | $-0.026$      | 5.87                | 4.642 | 1.804 | 2.838             |
| 99.92                             | 12               | 3.464            | 2.000          | 1.079  | $-1.097$                    | 8.327   | 0.0100                      | 0.000         | 0.08                | 4.642 | 0.431 | 4.211             |

**Table 9: Kinetics data of A4 Amiloride patch** 



**Figure 7: Graph of peppas release kinetics**





## **Compatibility studies: IR SPECTROSCOPY:**







## **Figure 10: FTIR of Optimized formulations**

The compatibility studies of the drug with excipients indicate no characteristic visual changes and no additional peaks were observed during FT-IR studies.

## **CONCLUSION:**

The formulation and characterization of a transdermal drug delivery system (TDDS) for Amiloride utilizing various polymers have demonstrated promising outcomes with respect to enhancing drug delivery efficiency and patient compliance. The key findings and conclusions are as follows:

- 1. **Successful Formulation:**
- o Various polymers were successfully to develop Amiloride transdermal patches, each contributing differently to the overall properties of the patches. The formulations were designed to optimize drug release and skin adhesion properties.
- 2. **Characterization Results:**
- o **Physical Properties:** The formulated patches exhibited acceptable physical characteristics including uniform thickness, adequate mechanical strength, and satisfactory adhesive properties. The polymer choice significantly influenced these properties.
- **Drug Release:** In vitro release studies indicated that the selected polymers controlled the release of Amiloride in a sustained manner, with some formulations showing a more favorable release profile, suggesting potential for enhanced therapeutic efficacy.

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