

The Beauty of Imperfection (and it's x-ray spectroscopic signatures)

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Department of Chemistry & Department of Physics,
University of Warwick



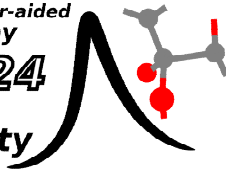
www.warwick.ac.uk/maurergroup

*Recent Advances in Computer-aided
X-ray Spectroscopy*

RACXS 2024

17-20 June

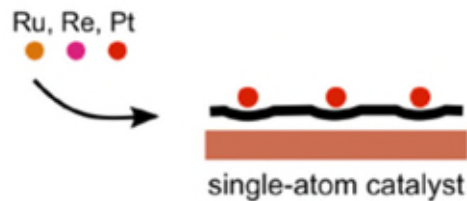
**Aalto University
Finland**



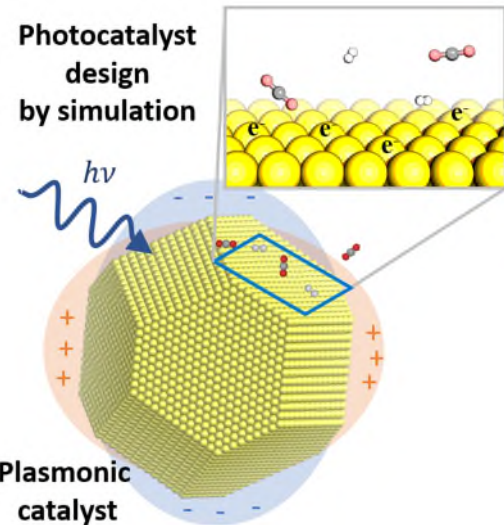
Computational Surface Science



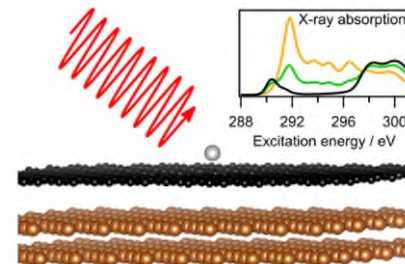
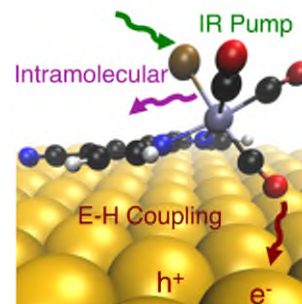
Design of functional nanostructures



Photocatalyst design by simulation



Ultrafast energy transfer & surface spectroscopy



Sam Hall
(now HZB,
Germany)



Benedikt Klein
(now KBSI,
Korea)



David Duncan
(Diamond/
Nottingham)



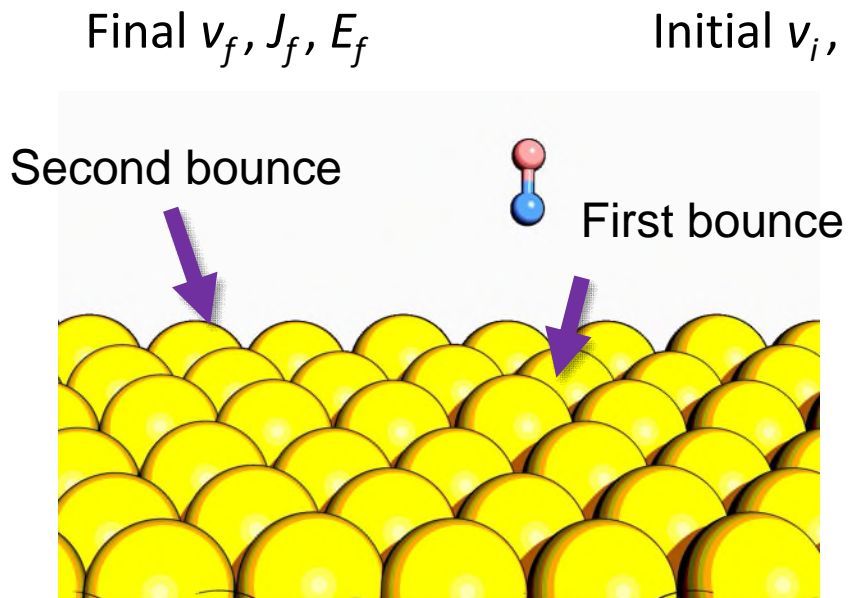
Michael Gottfried
(U Marburg)



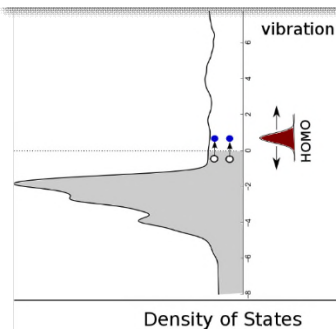
Alex Saywell
(Nottingham)

Measuring signatures of surface chemistry I

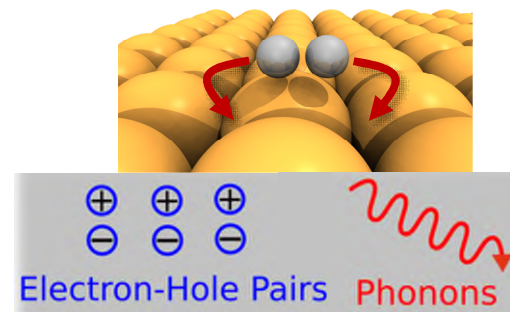
Throwing things at surfaces



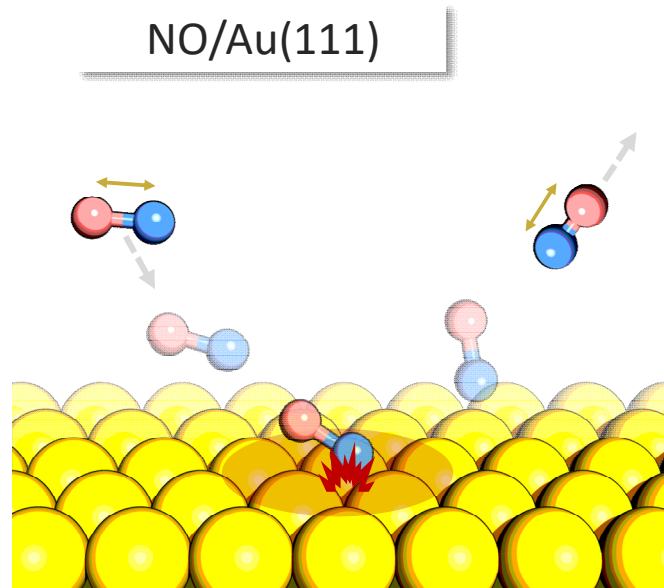
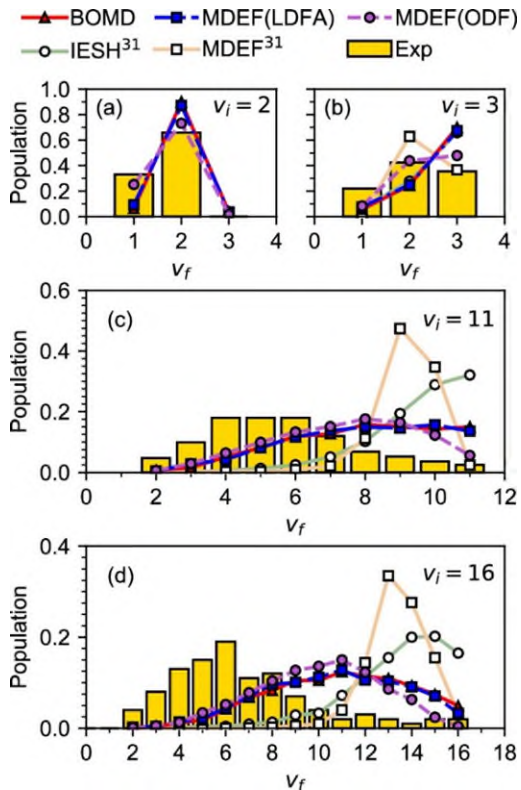
Electronic structure



Energy dissipation



Direct simulation of quantum state-to-state scattering



Box et al., JACS Au 1, 164-173 (2021)

Nonadiabatic effects lead to energy dissipation into “Hot Electrons”

Molecular Beam Scattering experiments:

Huang, Rettner, Auerbach, Wodtke, Science 290, 111-114 (2000); Int. Rev. Phys. Chem. 23 513 (2004)

Our computational approach



Electronic Structure Theory

Machine Learning
of Electronic
Structure

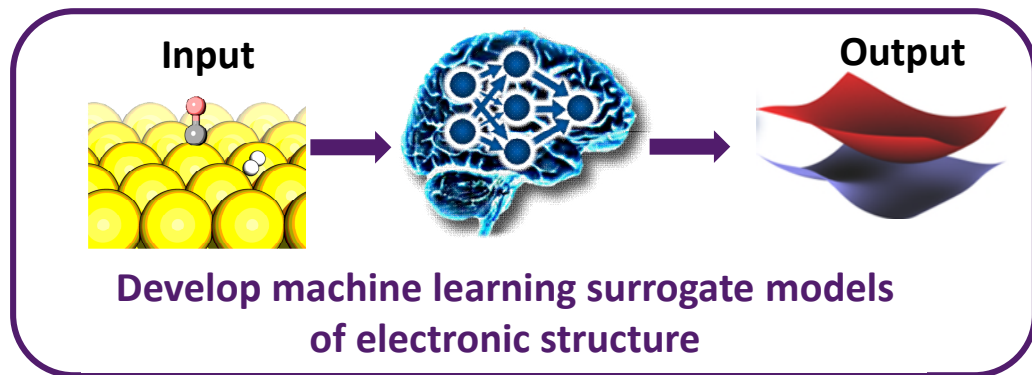
Coupling of Light
and Electrons



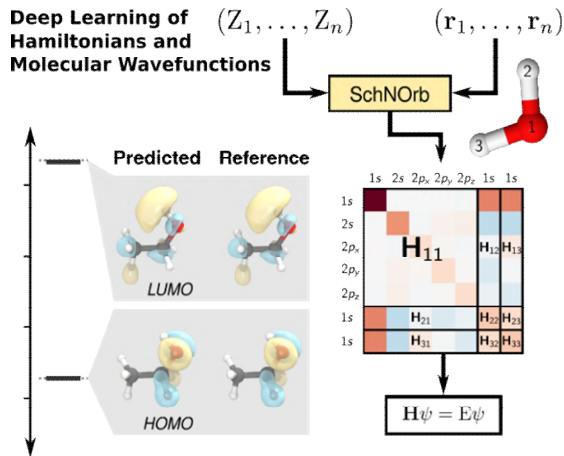
Nonadiabatic
Molecular
Dynamics



NQCDynamics.jl



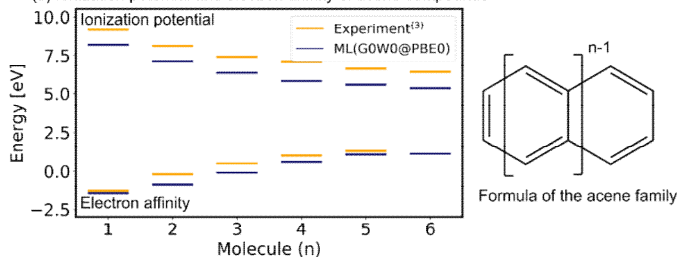
Deep Learning of Hamiltonians and Molecular Wavefunctions



Nature Commun 10, 5024 (2019)

Deep learning of molecular spectroscopy

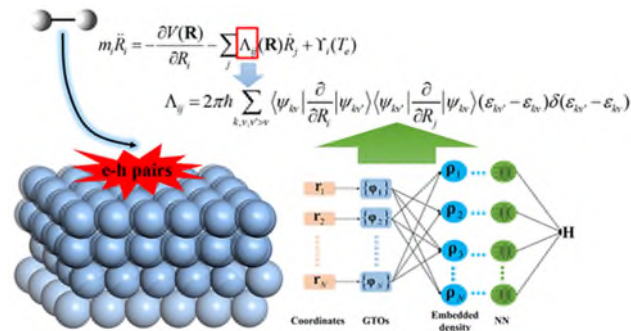
(e) Ionization potential and electron affinity of acene compounds



Chem. Sci. 12, 10755-10764 (2021)

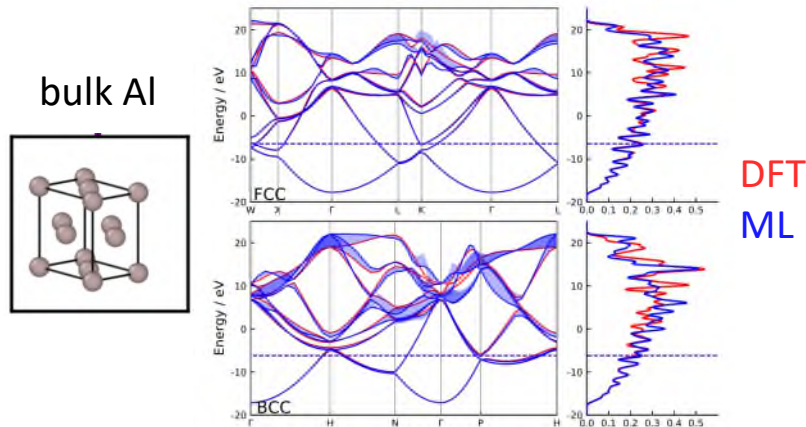
Machine Learning of Electronic Structure

Machine Learning of nonadiabatic response



J. Phys. Chem. C 124, 186–195 (2020)

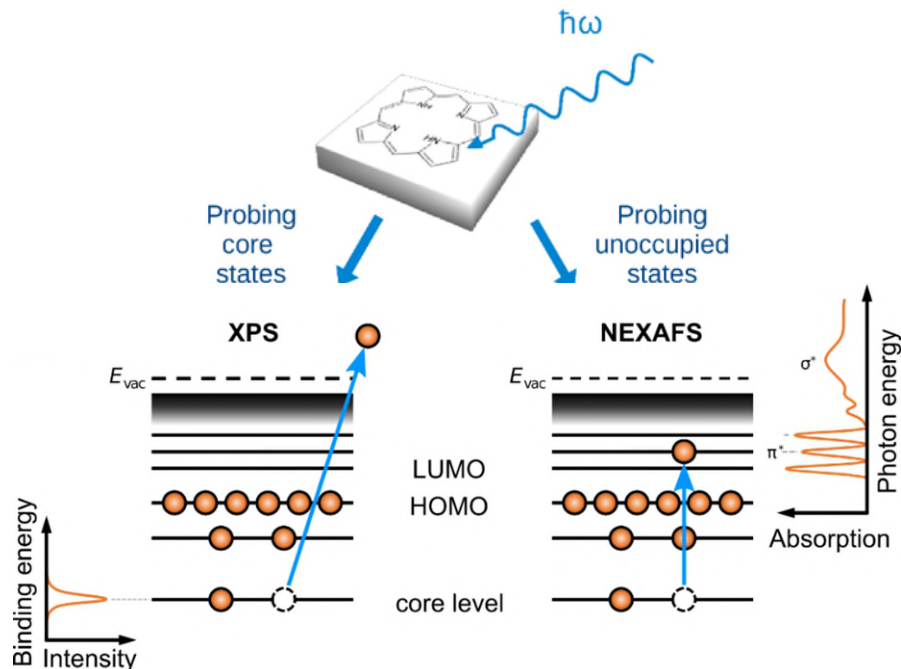
Machine learning of materials electronic structure



npj Computational Materials (2022)

Measuring signatures of surface chemistry II

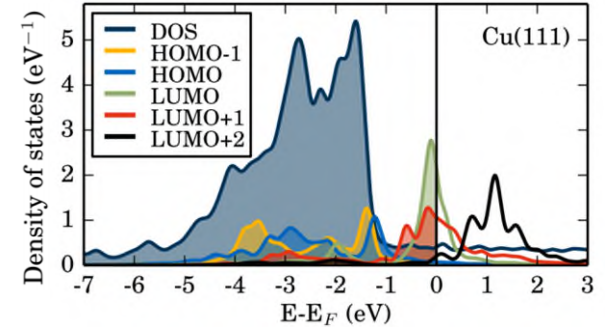
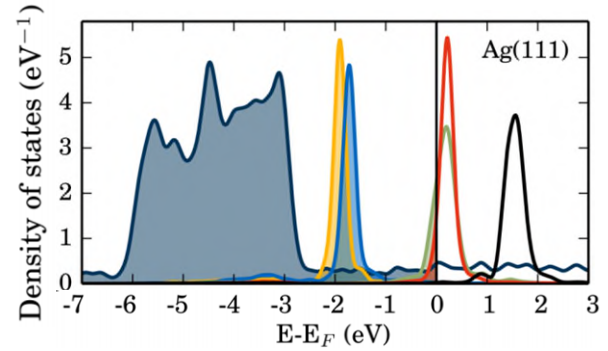
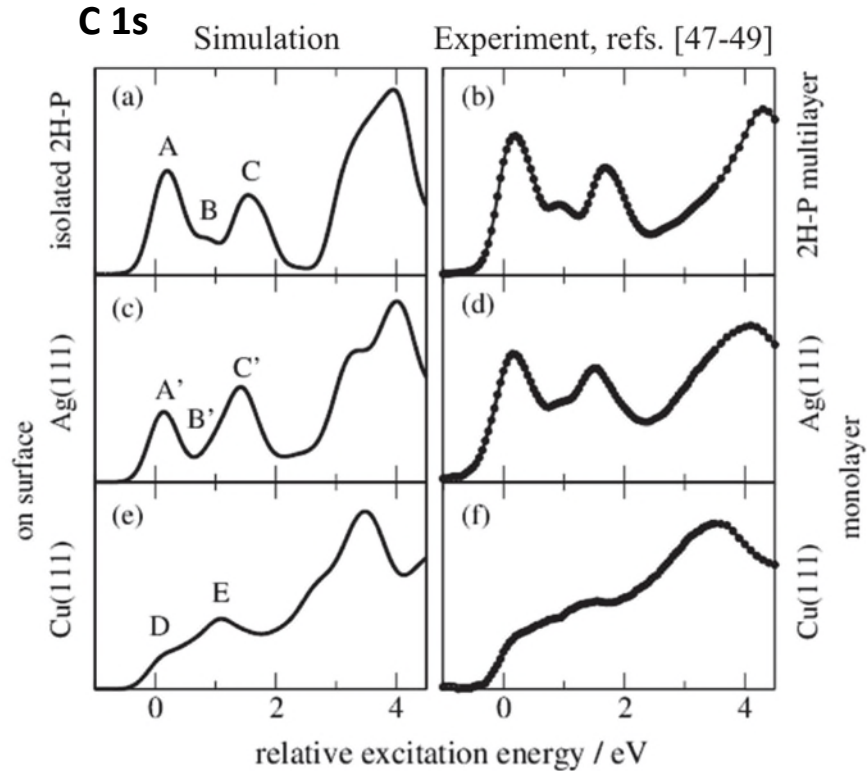
Blasting surfaces with X-rays



Klein et al., J. Phys.: Condens. Matter 33, 154005 (2021)

Diller et al., J. Chem. Phys. 146, 214701 (2017)

Surface Chemical Bonding – Quo Vadis?



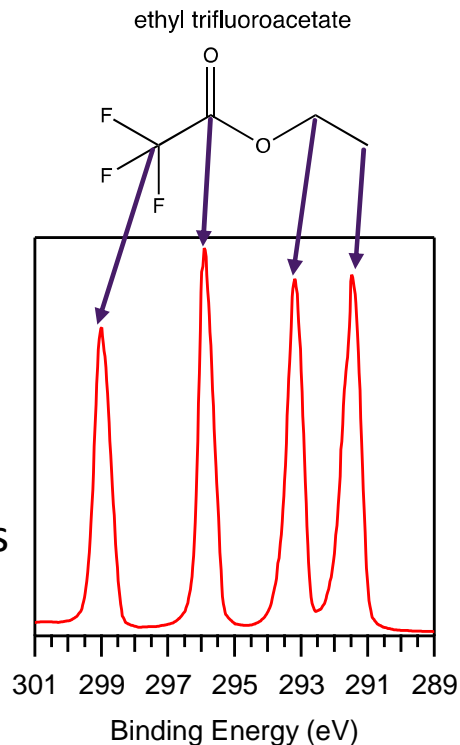
Hybridisation and charge transfer

Müller, Diller, Maurer, Reuter, *J. Chem. Phys.* **144**, 024701 (2016)

Diller et al., *J. Chem. Phys.* **146**, 214701 (2017)

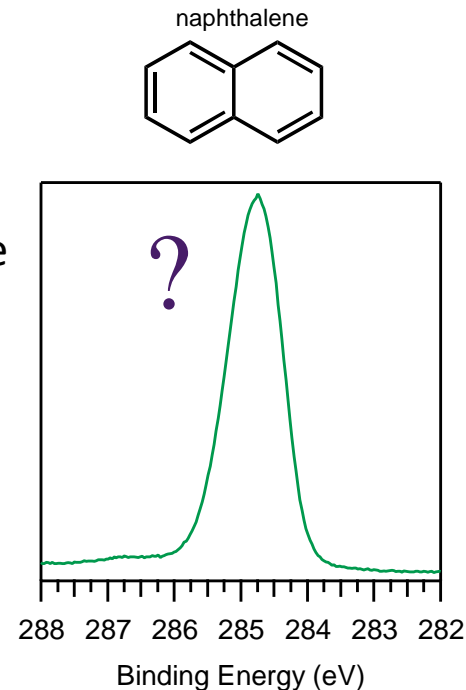
Problem 1: Relative Peak assignment

- Multiple chemical environments
- Clearly resolved peaks



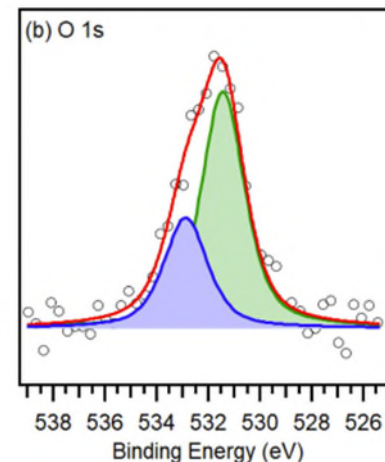
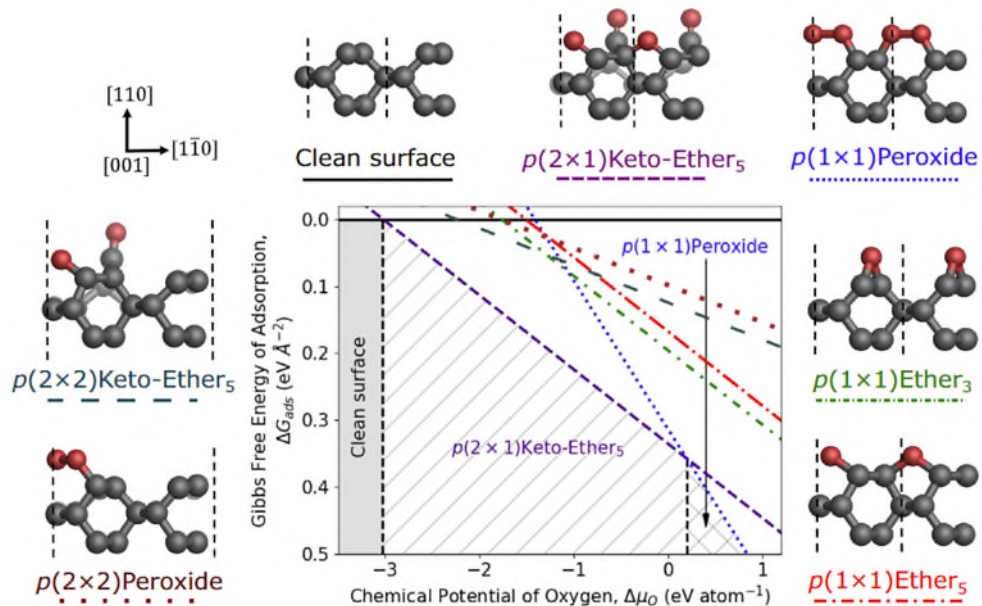
C1s XPS

- π -conjugated molecule
- All carbons have similar chemical environment
- Unable to distinguish due to experimental resolution



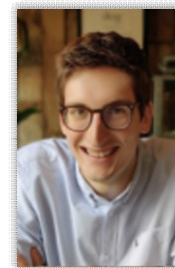
Problem 2: Absolute referencing of binding energies

Example: Oxygen surface termination of diamond (110)



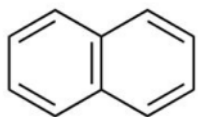
Absolute binding energy calculations

See talk by
Dylan Morgan

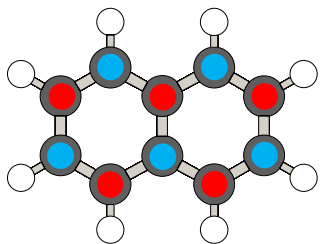


FHI-aims
The ab initio materials
simulation package

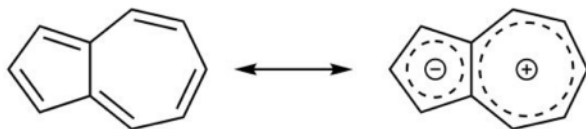
Topology and Spectroscopy



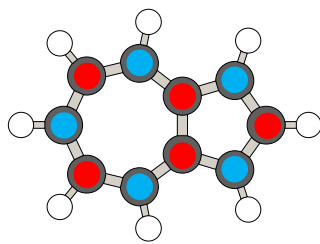
Naphthalene (Np)



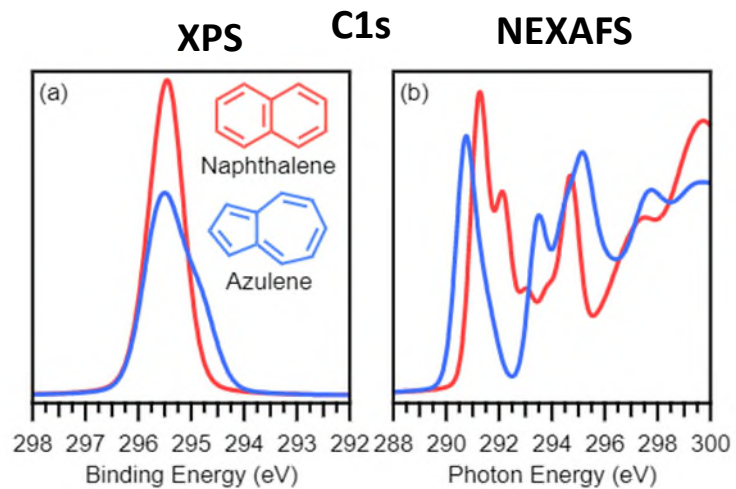
alternant



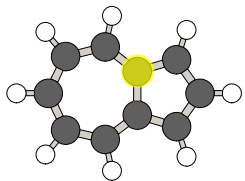
Azulene (Az)



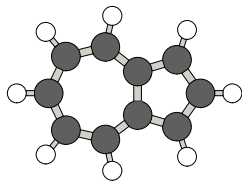
non-alternant



Core-level Simulation of XPS & NEXAFS

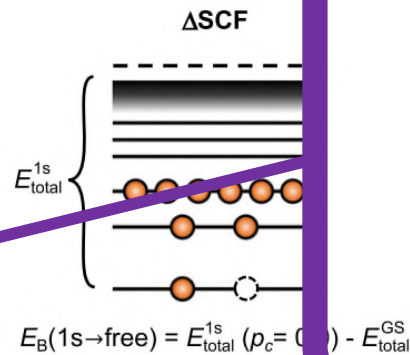
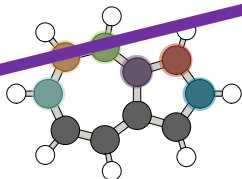
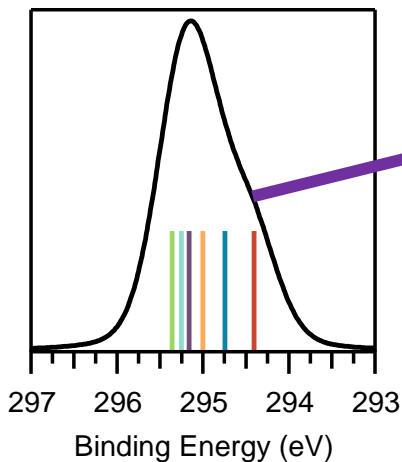


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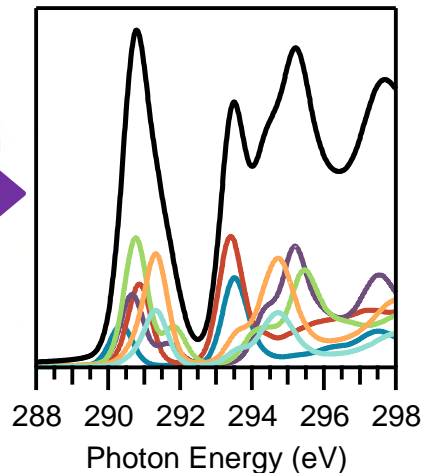
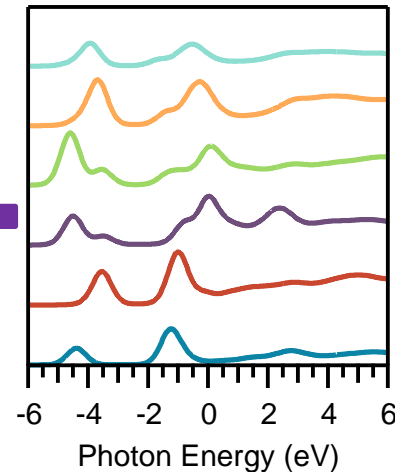
=

XPS BE



Solve KS equations with 1 electron removed from core-state for N atomic species to get total energies

N calculations
(+ 1 ground state calculation)

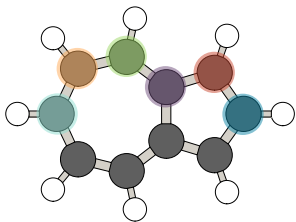


T. Mizoguchi, I. Tanaka, S. Gao and C. J. Pickard, *J. Phys. Condens. Matter* **21**, 104204 (2009)

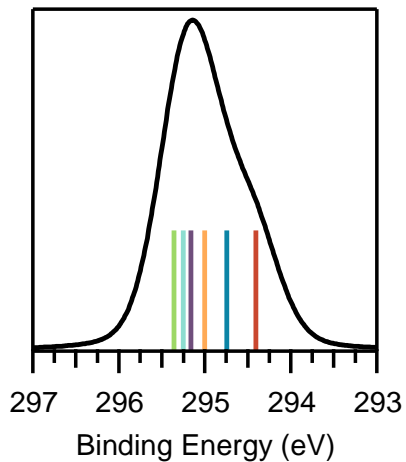
B. P. Klein, S. J. Hall and R. J. Maurer, *J. Phys. Condens. Matter* **33**, 154005 (2021)



Unpicking spectral contributions

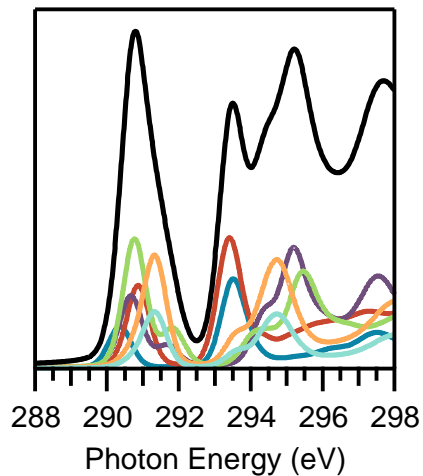


XPS



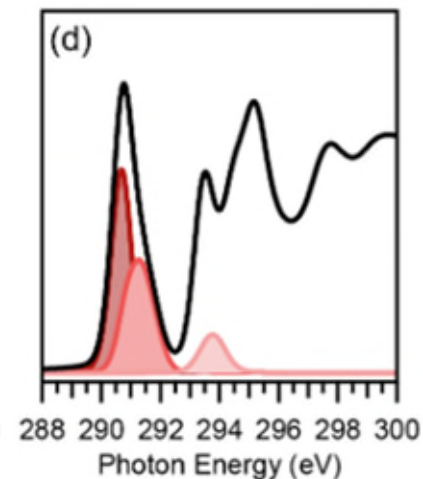
Hall et al., *J. Phys. Chem. C* **127**, 1870–1880 (2023)

Klein et al., *J. Phys. Condens. Matter* **33**, 154005 (2021)

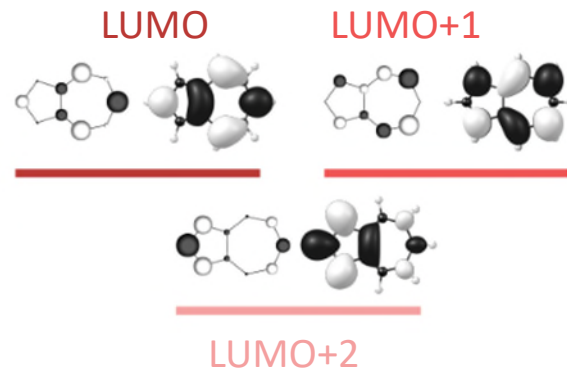


NEXAFS

Initial states



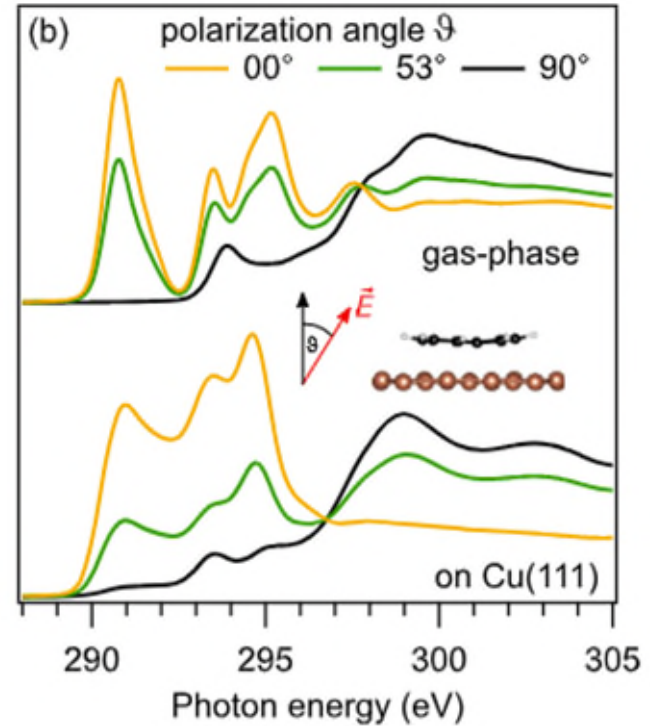
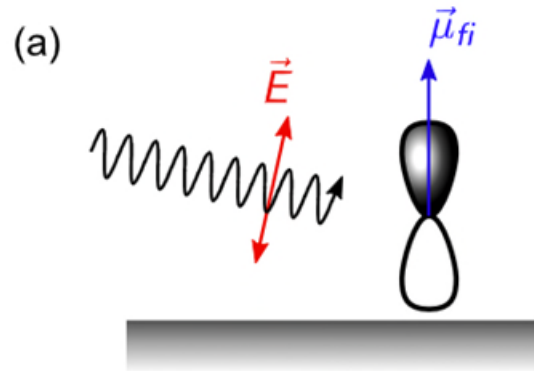
Final states



Orientalional/polarization dependence

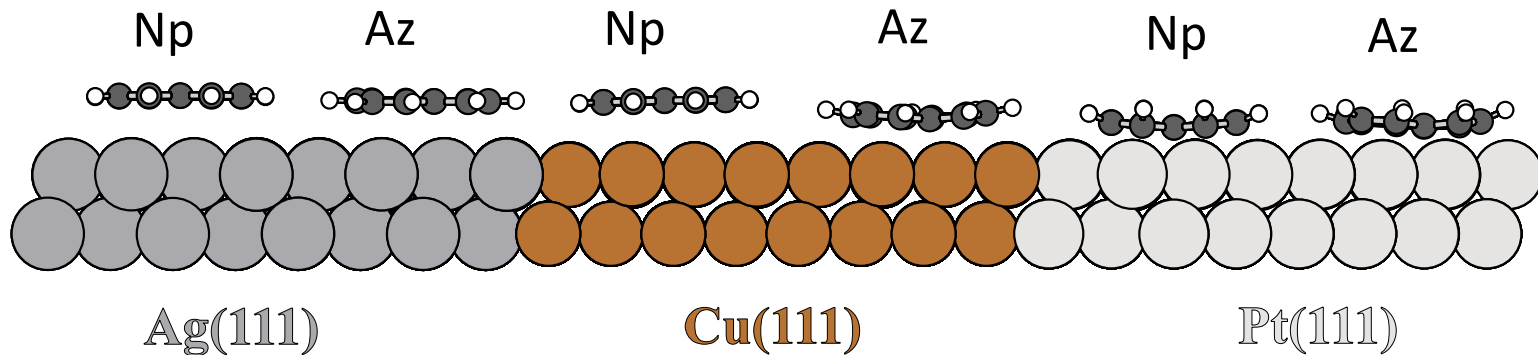
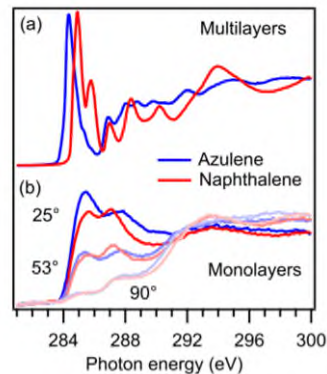
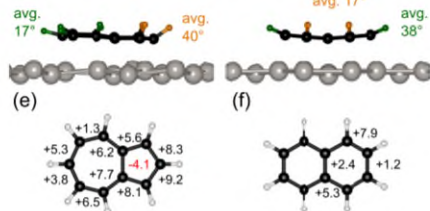
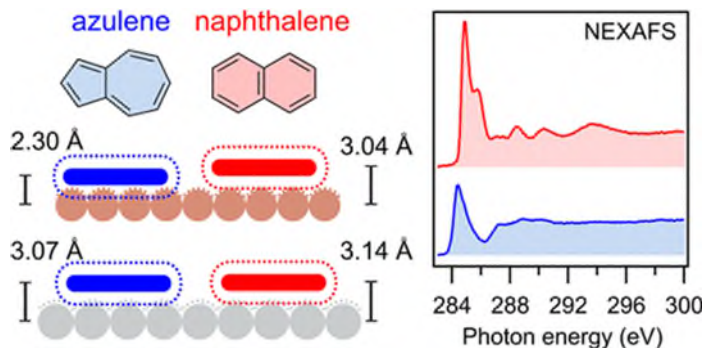
$$\sigma_i(h\nu) \propto \sum_f |\langle \psi_f | \mathbf{e} \cdot \mathbf{p} | \psi_i \rangle|^2 \delta(h\nu - \underbrace{(E_f - E_i)}_{\Delta E_{fi}})$$

$$\mu_{fi} = |\langle \psi_f | \mathbf{e} \cdot \mathbf{p} | \psi_i \rangle|^2 \approx |\mathbf{e} \cdot \underbrace{\langle \psi_f | \mathbf{p} | \psi_i \rangle}_{\mathbf{D}_{fi}}|^2$$



Naphthalene

Azulene



Detailed experiment/theory characterisation: XPS, NEXAFS, XSW, UPS, TPD, LEED, STM, AFM, DFT, ...



Michael Gottfried
(Marburg)

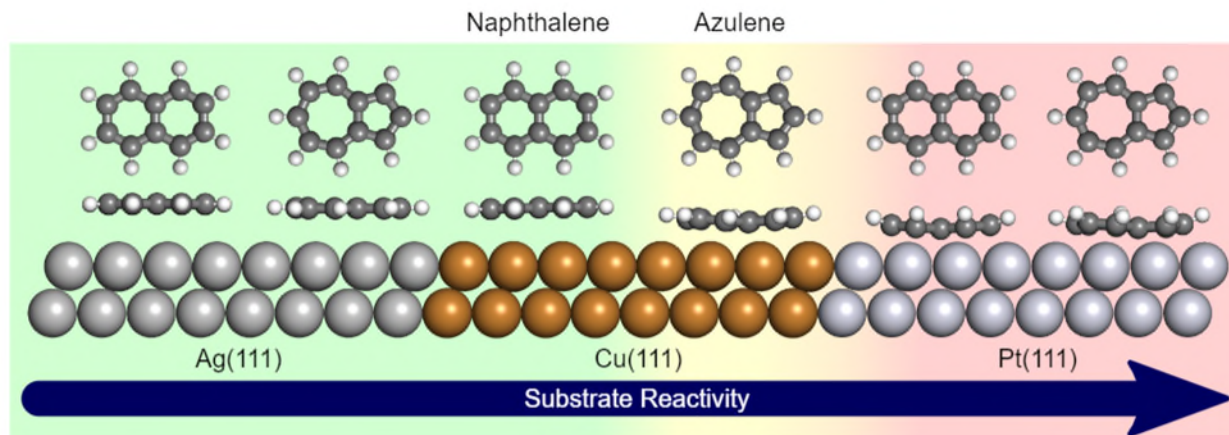


Benedikt Klein



Sam Hall

Naphthalene & Azulene: a playground to understand surface chemistry



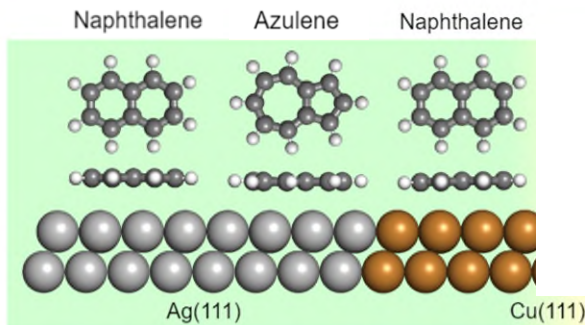
Adsorption Energy (kJ/mol)	118	128	135	173	298	335
Adsorption Height (Å)	3.10	2.94	2.96	2.33	2.09	2.10
Net Charge Transfer (e)	0.06	0.01	0.06	-0.49	0.03	0.02
Structural Deformation	None	None	None	Weak	Strong	Strong
Molecule-Metal Hybridisation	Weak	Weak	Weak	Intermediate	Strong	Strong

Category 1

Category 2

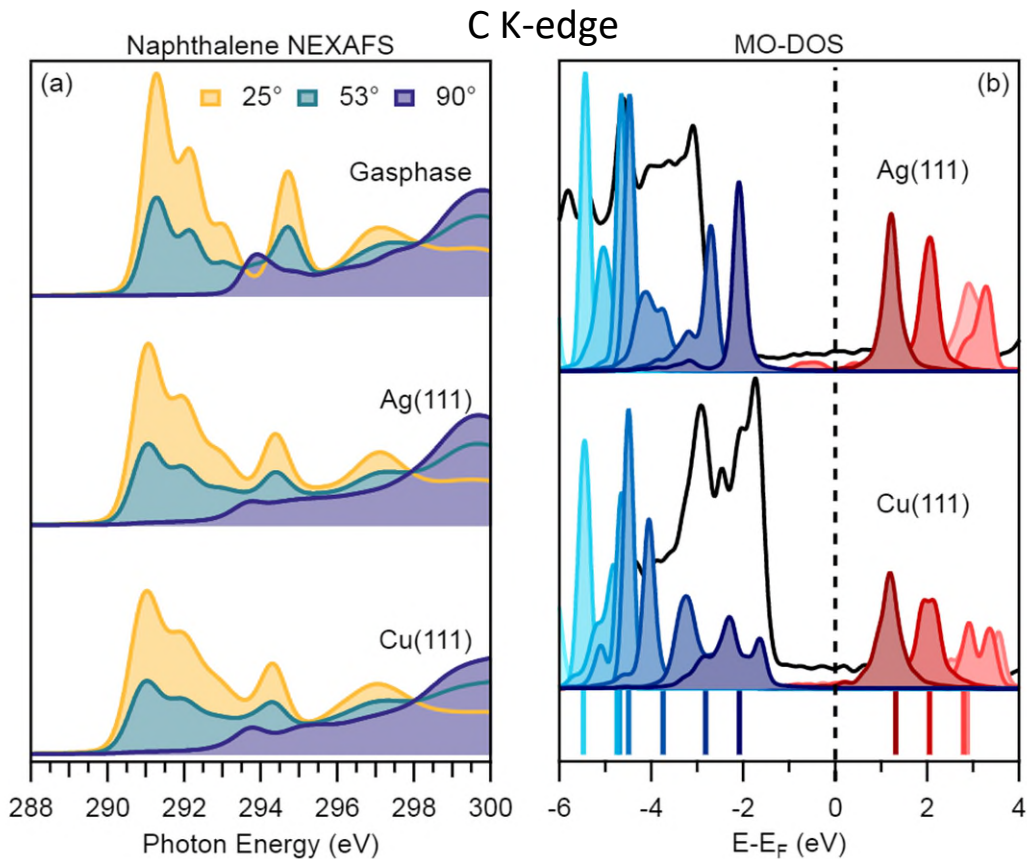
Category 3

Category 1

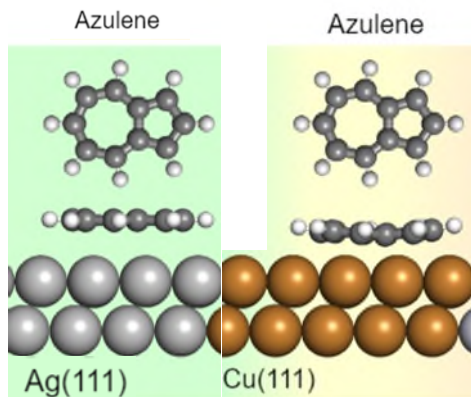


- No significant charge transfer
- Little hybridisation
- Spectra only slightly broadened
- Dichroism unaffected

“Physisorption”



Category 2



- Hybridisation for both Ag and Cu
- Only Az/Cu shows charge transfer
- Intensity of first peak diminished
- Dichroism also affected
- Neither clear case of “physisorption” nor “chemisorption”

Charge Transfer

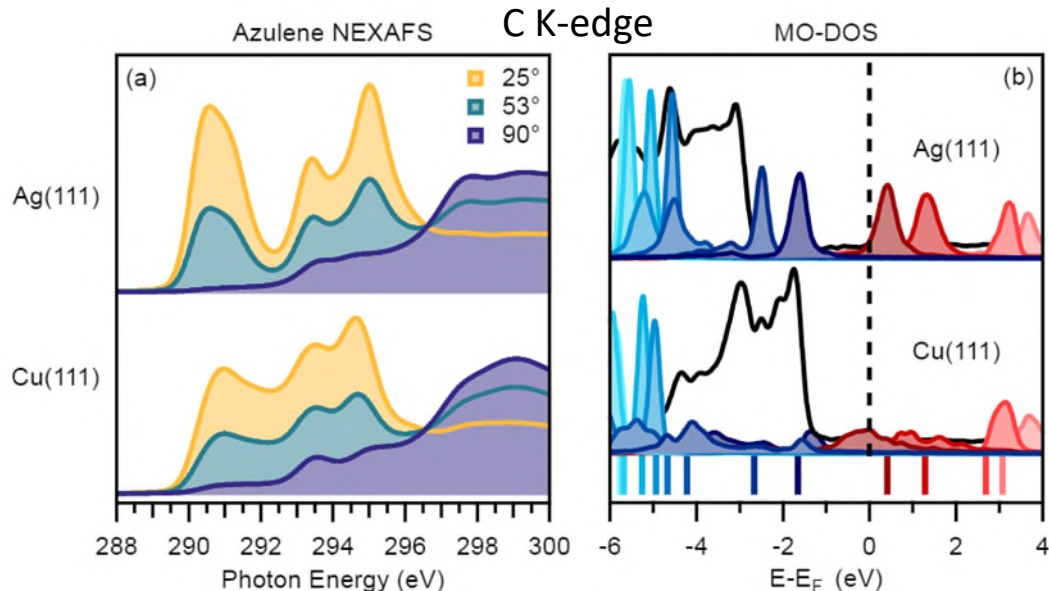


Table 1: Relative intensities of the first 3 peaks seen in the NEXAFS spectra of azulene on Ag(111) and Cu(111) with respect to the corresponding peaks of azulene gas-phase spectrum.

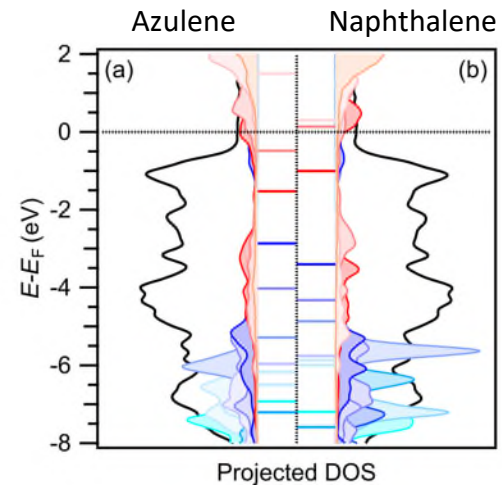
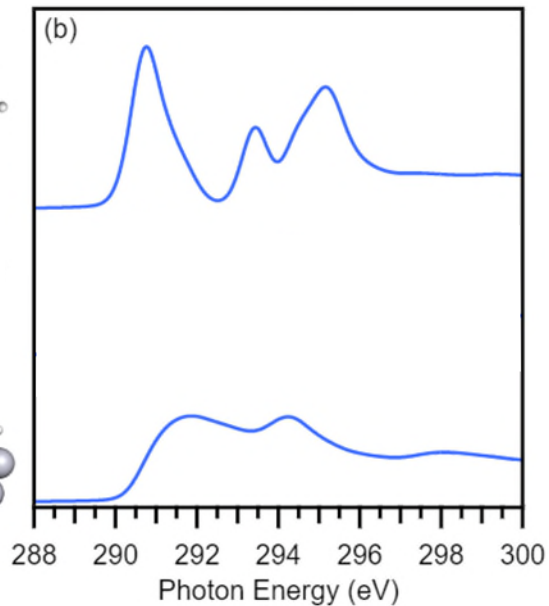
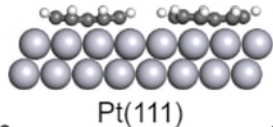
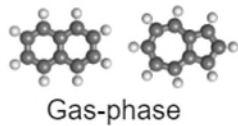
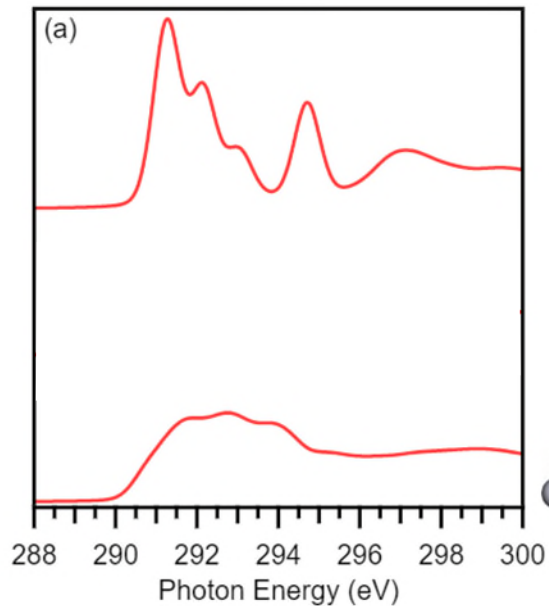
Peak	Relative Intensity	
	Az/Ag	Az/Cu
1	0.65	0.44
2	0.93	1.04
3	0.96	0.82

Category 3: "Chemisorption"

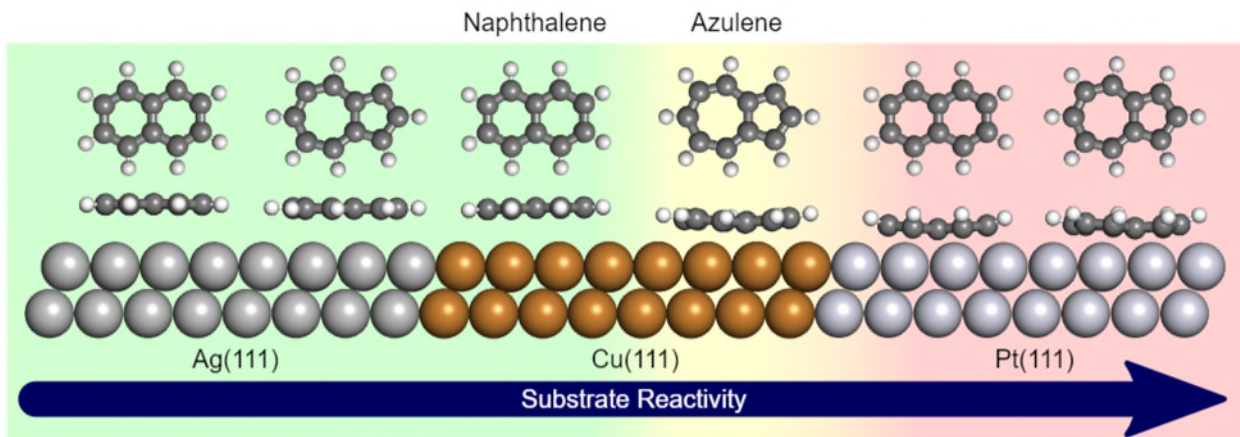
Naphthalene

C K-edge

Azulene



A playground to understand surface chemistry



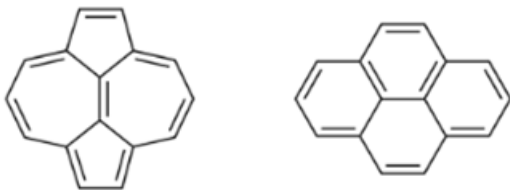
Adsorption Energy (kJ/mol)	118	128	135	173	298	335
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Molecule-Metal Hybridisation	Weak	Weak	Weak	Intermediate	Strong	Strong

Category 1
"Physisorption"

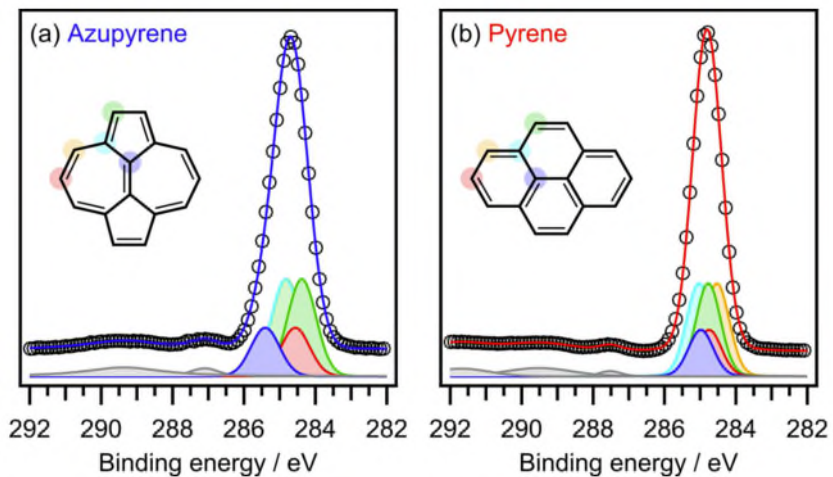
Category 2
Charge Transfer

Category 3
"Chemisorption"

Azupyrene and Pyrene



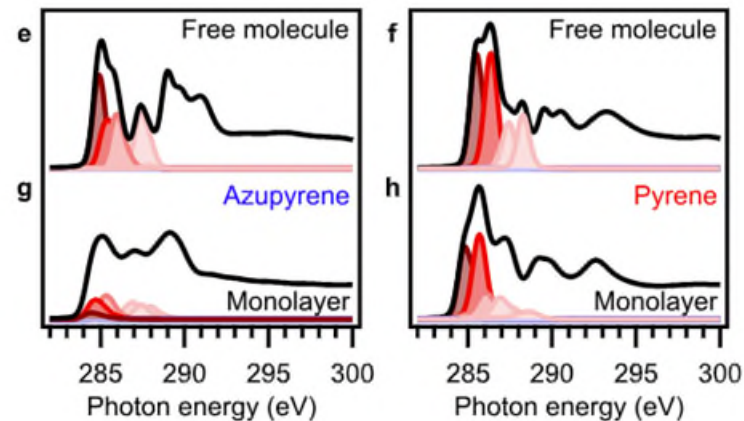
XPS



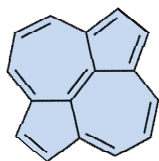
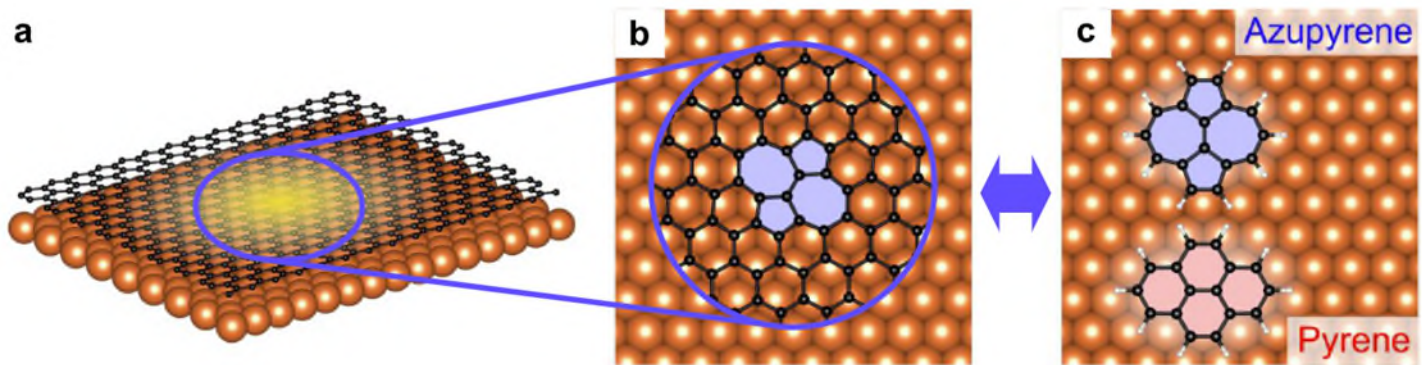
@Cu(111)



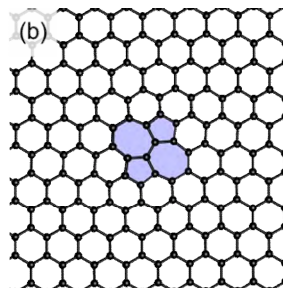
NEXAFS



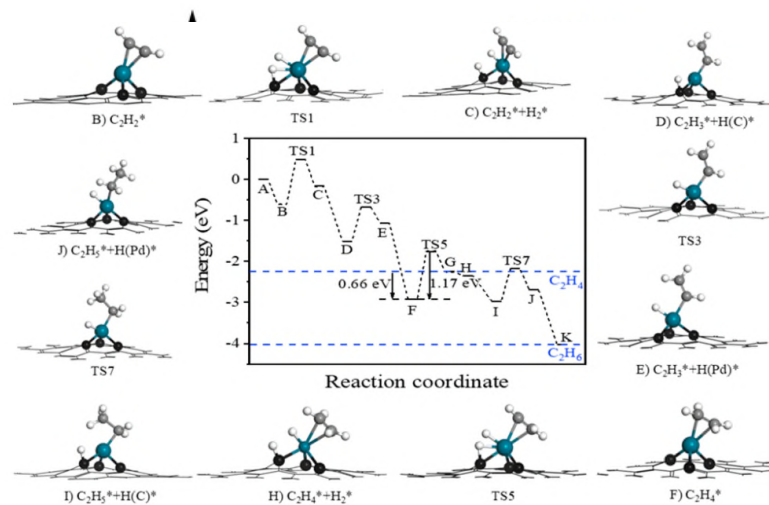
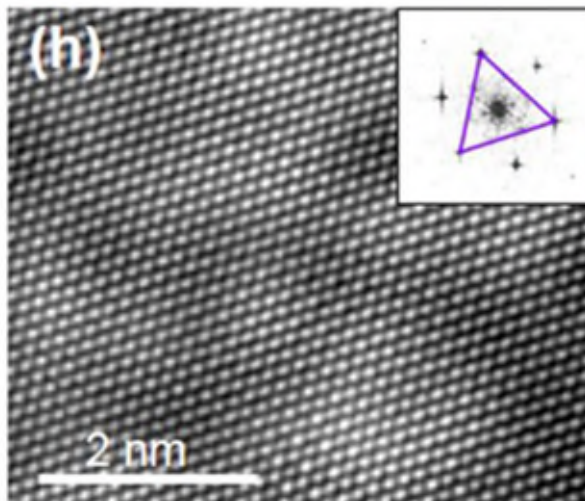
Azupyrene – a molecular model of a Stone-Wales defect



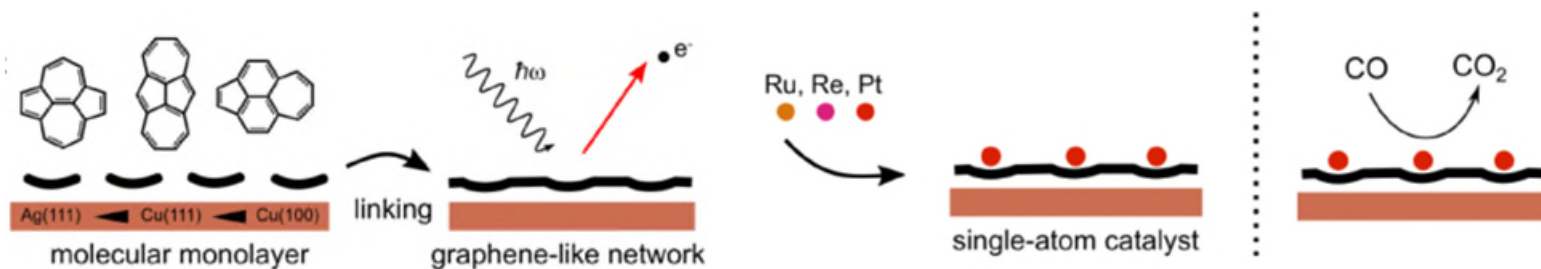
??



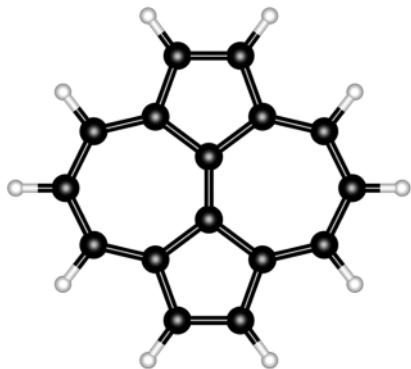
The Beauty of Imperfection: Topological Design of Defective Graphene



F. Huang, et al., *J. Am. Chem. Soc.*, 2018, 140, 13142-13146



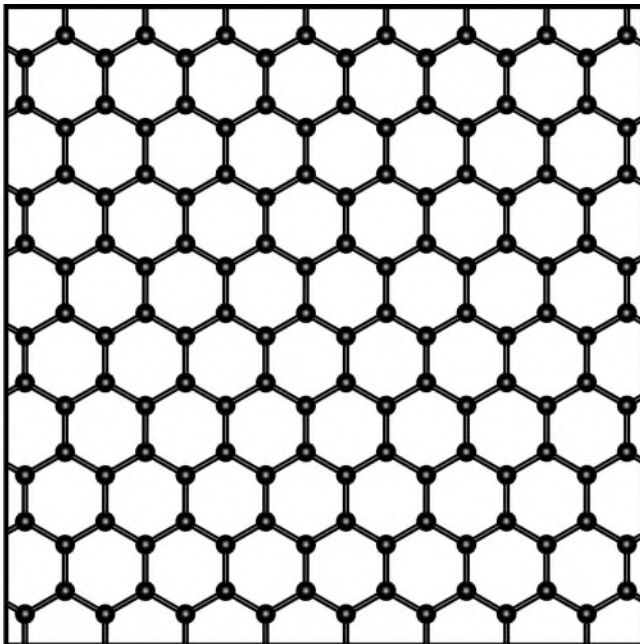
Growing defective graphene from azupyrene?



Azupyrene



High T
deposition
on Cu(111)



Benedikt Klein



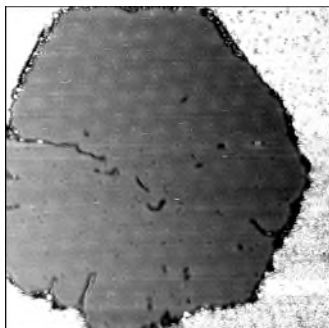
Matt Stoodley



David Duncan

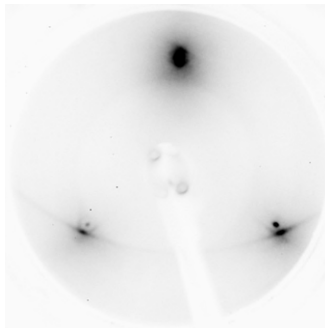
Proving graphene synthesis using azupryrene

1. Moire pattern visible in STM



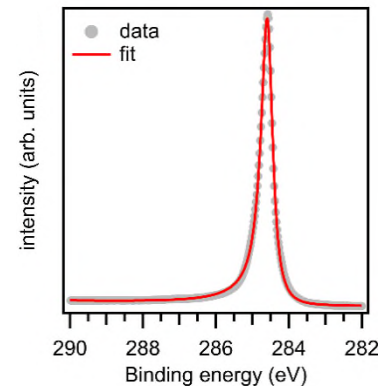
90x90 nm²

2. Rings in LEED

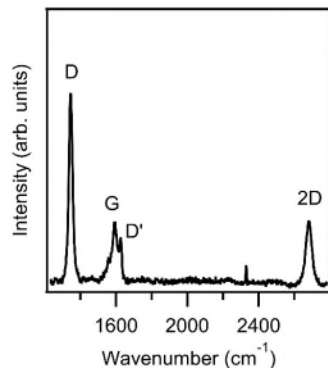


63 eV

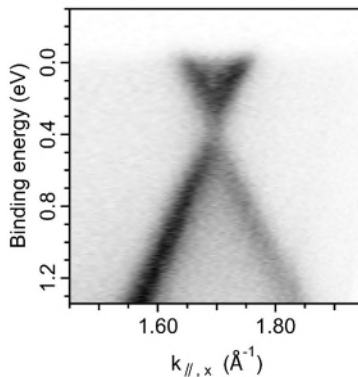
3. Sharp peak in C 1s XPS



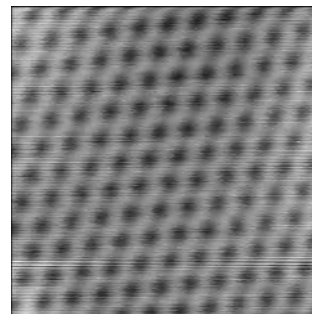
4. Specific Raman peaks



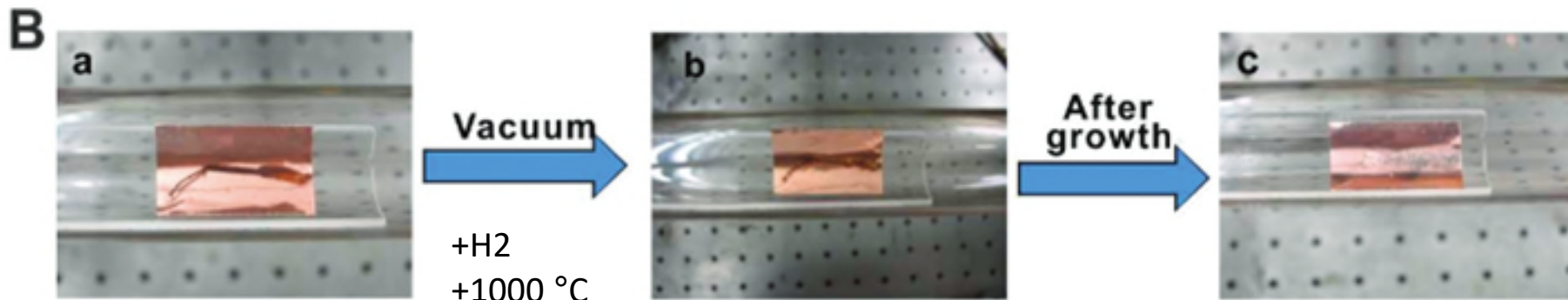
5. Dirac cone in ARPES



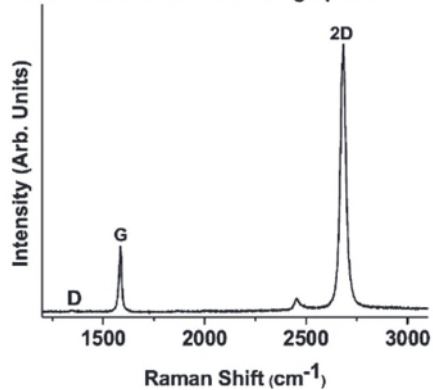
6. Atomic resolution, LT-STM



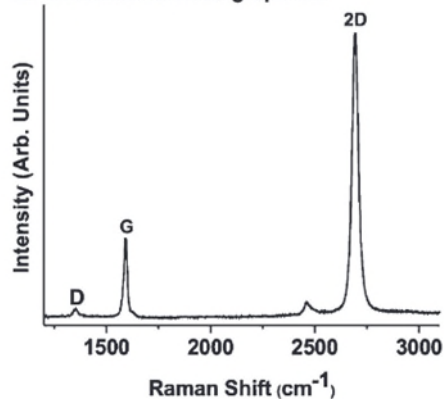
Is there anything that doesn't convert to graphene on copper?



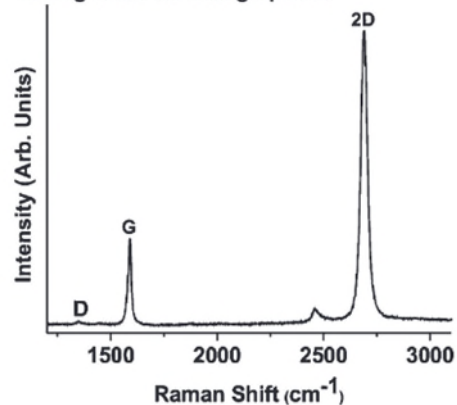
A. Girl Scout cookie-derived graphene



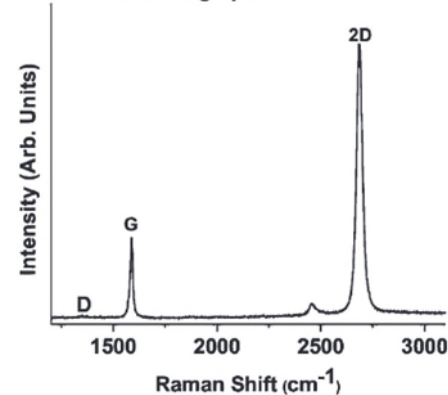
B. Chocolate-derived graphene



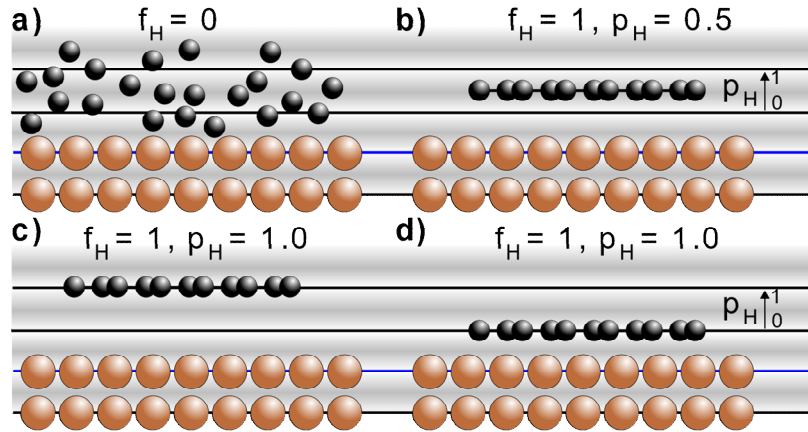
E. Dog feces-derived graphene



F. Roach-derived graphene



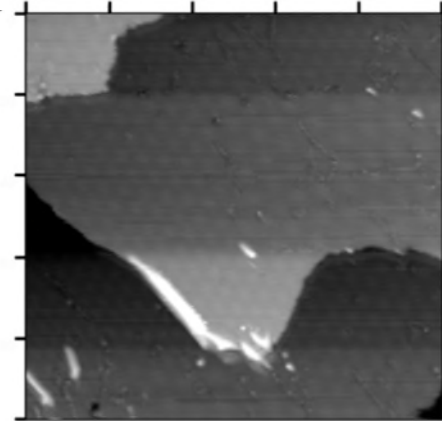
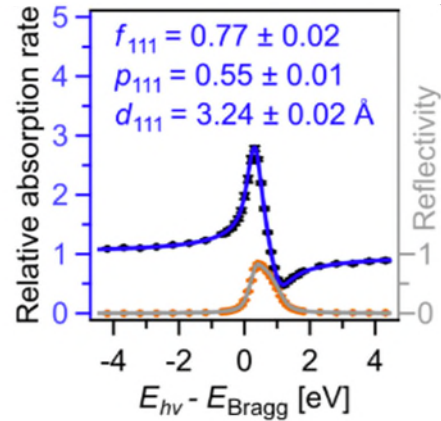
Normal Incidence X-ray standing waves

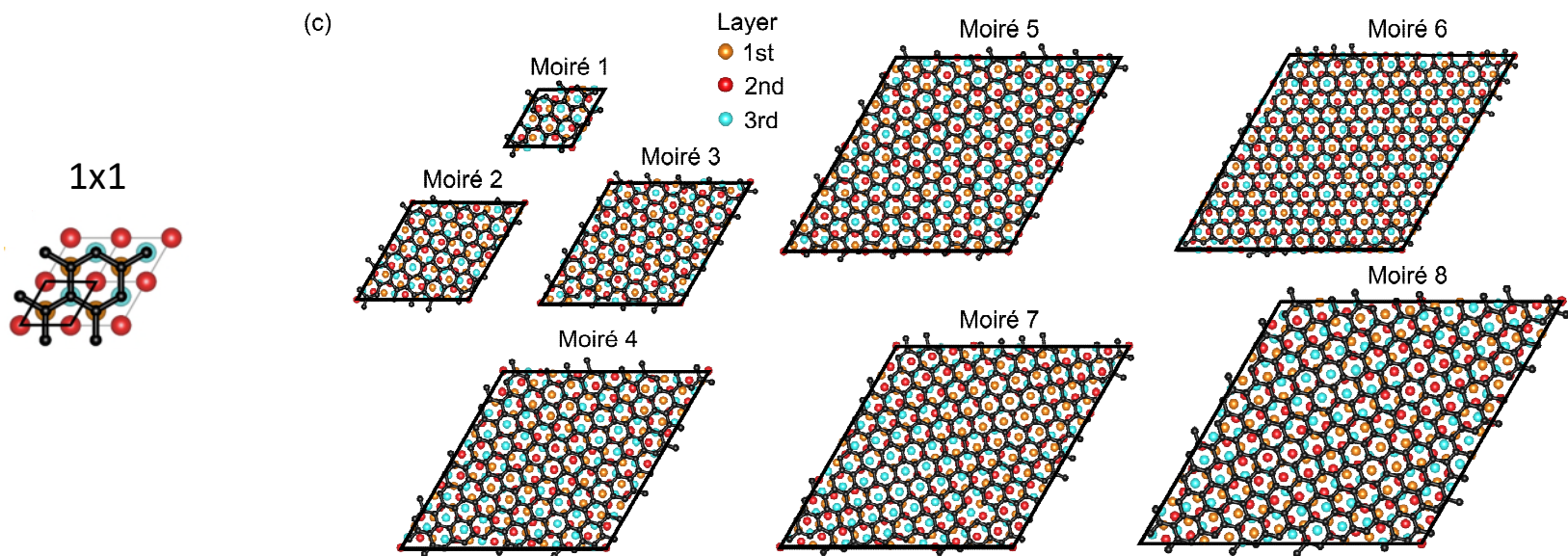
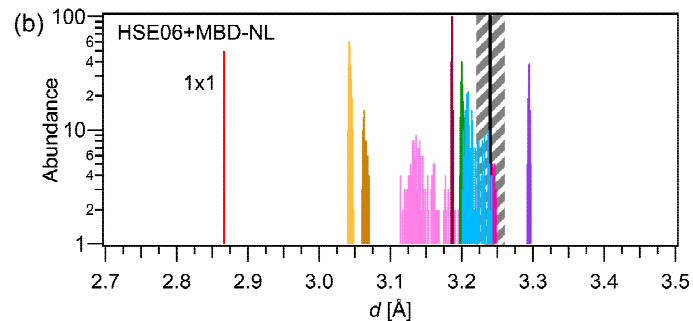
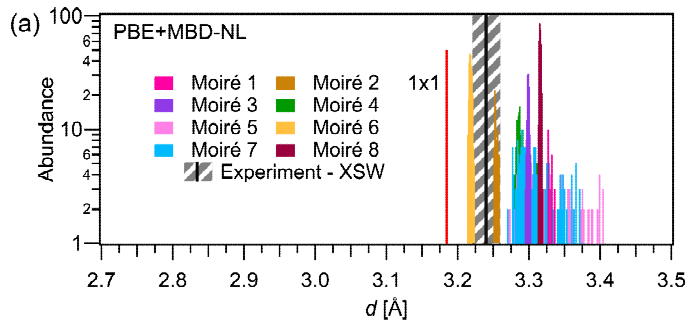


f_H = Coherent fraction

p_H = Coherent position

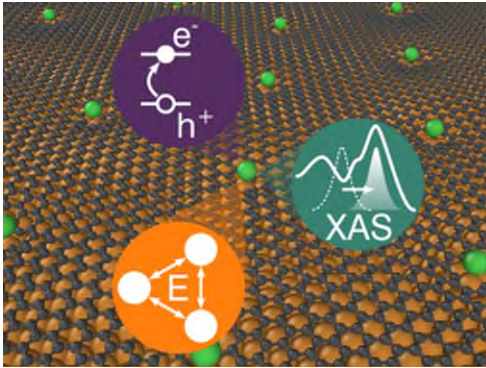
CVD grown graphene on Cu(111)





Deliberate imperfection

Topological Defect Design in 2D materials



Thank you for your attention!

Theory



Reinhard Maurer
(Warwick)

Spectroscopy



David Duncan
(Diamond/
Nottingham)

Imaging



Alex Saywell
(Nottingham)

Synthesis



Christian Nielsen (QMUL)