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# Study of Ag doped GeSbTe thin films using X-ray photoelectron spectroscopy



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# Content



## Introduction

Phase change materials (PCMs), based on structural transition, are potential candidate for data storage, thermal storage, drug delivery, image recording.

Chalcogenide (ChG) based **PCMs exhibit reversible** structural phase transition from amorphous to crystalline, accompanied by a drastic change in optical and electrical properties, on the nanosecond timescale.

In the modern era, materials having different structure under different conditions have unique importance from application point of view.

**Crystalline Phase** 

**Amorphous Phase** 

Alloys on GeTe-Sb<sub>2</sub>Te<sub>3</sub> tie line and AgInSbTe are potential ChG based PCMs used for various technological applications.

Nat. Photonics 11, 465-476 (2017), Adv. Opt. Mater. 5, 1700261 (2017), Nat. Commun. 8, 1446 (2017), Adv. Funct. Mater. 1705563 (2018), Sci. Reports 7, 42712 (2017).



Among them, Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub> (GST) is a potential candidate for phase transition technology due to its capability of phase transition on annealing and by applying current/voltage pulse and pressure owing to its versatile properties like fast crystallization speed, good thermal stability and better data retention.

Nat. Mater. 6, 824 (2007), Sci. Rep. 5, 8050 (2015).

## Phase change mechanism



Schematic of phase transition mechanism. State I is the crystalline phase having high reflectivity and low electrical resistivity. State I can be converted to State III (amorphous phase), having low reflectivity and high electrical resistivity, by applying short and intense laser/electrical pulse and with quenching of liquid (intermediate state II). Amorphous structure (state III) can be reconverted to crystalline structure (state I) by applying long and moderate laser/electrical pulse.

Nat. Mater. 6, 824 (2007).

## **Properties of Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>**

Face centered cubic (fcc) phase is achieved by pa annealing thin films above d temperature of re 150 °C. in te

Hexagonal close packed phase is achieved by annealing thin films above temperature of 250 °C.

20% contrast in optical reflectivity with phase transition from amorphous to

fcc.



Three order of











Nat. Mater. 6, 824 (2007), Chem. Rev. 110, 240 (2010), Sci. Rep. 5, 8050 (2015).

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As-deposited thin films are amorphous in nature.



*Nat. Mater.* **6**, 824 (2007), Mater. Today **11**, 20 (2008), *Chem. Rev.* **110**, 240 (2010), Nature 511, 206 (2014), *Sci. Rep.* **5**, 8050 (2015), *Nat. Photonics* **11**, 465-476 (2017), Appl. Phys. Lett. **111**, 261102 (2017).

#### **REVIEW ARTICLES** INSIGHT

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REVIEW

Phase Change Materials and Their Application to Nonvolatile Memories

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## Phase-change materials for rewriteable data storage

Phase-change materials are some of the most promising materials for data-storage applications. They are already used in rewriteable optical data storage and offer great potential as an emerging non-volatile electronic memory. This review looks at the unique property combination that characterizes phase-change materials. The crystalline state often shows an octahedral-like atomic amangement, thequently accompanied by pronounced latics distortions and huge vacancy concentrations. The can be attributed to the chemical bonding in phase-change alloys, which is promoted by p-orbitals. From this insight, phase-change alloys with desired properties can be designed. This is demonstrated for the optical properties of phase-change alloys, in particular the contrast between the amorphous and crystalline states. The origin of the fast crystallization kinotics is also discussed.

MATTHIAS WUTTIG<sup>1+</sup> AND NOBORU YAMADA<sup>2</sup> <sup>19</sup>hyskialente terlist, WITH Actore Extensis, 5385 Actor, Genzary <sup>40</sup> Car Technology Development Check, Matanita Katori, Katoriani Congang, 3-1-1 Tagane-Taska-mach, Bielgach, Back 379-500, Apar <sup>4</sup>-mail: wellightyski.rdt-actor.dt

attractive features<sup>14</sup>, and we consider the potential of thes strange concept as well as the unique material properties required. In applications as a non-volatile memory, the pronounced difference in electrical notatively us such. The amorphone state has a high resistance. Applying a long voltage police (set poles) locally board the amorphone motion and loads to nervarialization. Amorphics

Nat. Mater. 6, 824 (2007).

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www.advmat.de

www.MaterialsViews.con

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Design Rules for Phase-Change Materials in Data Storage Applications

Dominic Lencer, Martin Salinga, and Matthias Wuttig\*

Adv. Mater. 23, 2030 (2011).



**REVIEW ARTICLE** 

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## Phase-change materials for non-volatile photonic applications

M. Wuttig<sup>1,2,3\*</sup>, H. Bhaskaran<sup>4</sup> and T. Taubner<sup>1,5</sup>

## **Experimental details**



≻ As-deposited thin films were annealed in vacuum at 160 °C and 260 °C.

Singh et. al ; Appl. Phys. Lett. 111, 261102 (2017) and Semicond. Sci. Technol. 32, 045015 (2017).

## **Results and discussions**

X-ray diffractometer (X'Pert PRO PANalytical) with radiation of Cu  $K_{\alpha 1}$  ( $\lambda = 1.540$  60 Å) was used for X-ray diffraction study.

As-deposited thin films of  $(Ge_2Sb_2Te_5)_{100-x}Ag_x$  (x = 0, 1, 3, 5 and 10) are amorphous in nature.



X-ray diffraction patterns of as-deposited thin films of  $(Ge_2Sb_2Te_5)_{100-x}Ag_x$  (x = 0, 1, 3, 5 and 10).

- Phase transition is achieved with annealing.
- (Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>)<sub>100-x</sub>Ag<sub>x</sub> (x = 0, 1 and 3) thin films have fcc phase annealed at 160 °C and hcp phase at 260 °C.
- (Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>)<sub>95</sub>Ag<sub>5</sub> more fcc and minor hcp phase, on the other hand, (Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>)<sub>90</sub>Ag<sub>10</sub> has more hcp phase but minor fcc at 160 °C.
- Amorphous thin films are highly transparent in the NIR region; transmission sharply decreases with phase transition and it is negligible in the hcp phase.
- Transmission spectra of (Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>)<sub>90</sub>Ag<sub>10</sub> annealed at 160 °C and 260 °C almost overlaps





The transmission contrast can be used to fabricate transmission window in nearinfrared (NIR) region.



(a-b) Transmittance contrast between amorphous-FCC and amorphous HCP phases in  $(Ge_2Sb_2Te_5)_{100-x}Ag_x$  (x = 0 and 10) as-deposited and annealed thin films. (c-d) Schematic of reversible NIR-window.

## **Raman spectroscopy**

- Raman scattering spectra were measured at room temperature by a Micro-Raman spectrophotometer (LabRAM HR800, JY).
- The Raman band at ~124 cm<sup>-1</sup> with A<sub>1</sub> mode is assigned to GeTe<sub>4-n</sub>Ge<sub>n</sub> (n=1, 2) corner sharing tetrahedral and band at ~152 cm<sup>-1</sup> to be associated with Sb-Te vibrations in SbTe<sub>3</sub> unit or from defective octahedral coordination of Sb atoms.
- With Ag doping upto 3% the intensity of these bands improves but with 10% Ag the intensity of these bands starts decreasing, which confirms the structure modification and distortion in the system.

Mater. Chem. Phys., 136, 935 (2012).



Raman spectra of as-deposited ( $Ge_2Sb_2Te_5$ )<sub>100-x</sub> $Ag_x$  (x = 0, 3 and 10) thin films.

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## X-ray photoelectron spectroscopy

Omicron Multiprobe Surface Analysis (Scienta Omicron, Germany). A Mg  $K_{\alpha}$  radiation source (1253.6 eV) along with a seven channel detector was employed for XPS data acquisition. The surface contaminants were removed by means of the mild sputtering method using 500 eV Ar<sup>+</sup> ions for 10 min. **Valance band and core level spectra was measured**.



Silver metal  $(3d_{5/2}) = 368.2 \text{ eV}$ Ag (3d) region has well separated spin-orbit components ( $\Delta_{\text{metal}} = 6.0 \text{eV}$ )

### Core level of Sb (3d):



Sb metal  $(3d_{5/2}) = 528.3 \text{ eV}$ Sb (3d) region has well separated spin-orbit components ( $\Delta_{\text{metal}} = 9.3 \text{ eV}$ )

### Core level of Te (3d):



Te metal  $(3d_{5/2}) = 573 \text{ eV}$ Te (3d) region has well separated spin-orbit components ( $\Delta_{\text{metal}} = 10.4 \text{ eV}$ )

### Valance band spectra:

- The Valence band spectra (VBS) can be divided into two main parts, Region A contains the *p*-bands of Ge 4*p*, Sb 5*p*, and Te 5*p*, while region B is due to the Ge 4*s*, Sb 5*s*, and Te 5*s*.
- The VBS of GST and GST:3Ag has same shape but an additional feature around 2 eV was appeared with Ag addition because of increase in metallic character.
- The drastic change in density of valence state is observed with 10% Ag content. This may be due to the distortion in the host lattice



Valence band spectra of  $(Ge_2Sb_2Te_5)_{100-x}Ag_x$  (x = 0, 3 and 10) thin films: (a) as-deposited, (b) annealed at 160 °C and (c) annealed at 260 °C.



(Ge<sub>2</sub>Sb<sub>2</sub>Te<sub>5</sub>)<sub>90</sub>Ag<sub>10</sub> is the potential candidate for reversible near infrared window.

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