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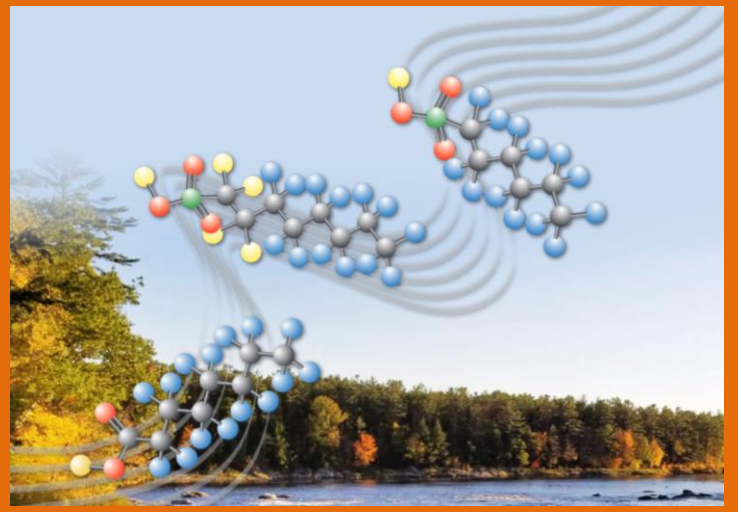
Introduction to Chemistry Library Research for Juniors : An Overview | 25 September 2023



Emily C. Wild

Chemistry, Geosciences and Environmental Studies Librarian

https://pubs.usgs.gov/publication/pp1802K

https://pubs.usgs.gov/publication/cir1490

PRINCETON UNIVERSITY

My Bio

Emily C. Wild

ewild@princeton.edu

https://library.princeton.edu/staff/ewild

Princeton University Library, 2018-Present
Chemistry, Geosciences and Environmental Studies Librarian
Subject Specialist & Selector for Chemistry, Geosciences, Environment, Energy (contact me directly)
LinkedIn: https://www.linkedin.com/in/emilycwild/
ORCID: https://orcid.org/0000-0001-6157-7629

U.S. Geological Survey employee 1995/1996 to 2018:
https://web.archive.org/web/20170716130801/https://www.usgs.gov/staff-profiles/emily-wild

Denver, Colorado : 2008-2018 - Librarian (Physical Scientist) for Water, Geology, Geography, Biology

NH-VT & MA-RI: 1996-2008 - Hydrologist: Water Use, Surface Water, Groundwater, Water Quality, Coastal Waters, Hydrogeology, Bibliographic Databases

[Emily C. Wild (0000-0001-6157-7629) - ORCID]

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ewild@princeton.edu

Websites & social links
Princeton University Library
Chemistry Databases and Library Guide
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Environment Databases
Energy Databases
LinkedIn
Threads

Other IDs
Scopus Author ID: 7005382800
ResearcherID: AAA-6920-2019

Biography

Emily C. Wild is the Chemistry, Geosciences and Environmental Studies Librarian in the Princeton University Library. In 2022, Emily received the Geosciences Information Society's (GSIS) 2022 Mary B. Ansari Distinguished Service Award for Geosciences Librarianship. Emily has a Bachelor of Arts in Geology from Hartwick College, and a Master of Library and Information Studies from the University of Rhode Island.

From 2008 to 2018, Emily was a Librarian (Physical Scientist) at the U.S. Geological Survey Library, where she helped library users find and use science and legislative materials, provided science and government outreach information, bibliographic and map instruction, as well as develop and present online and in-person training sessions on topics such as chemical and physical properties of the atmosphere, rock, sediment, and water; crustal geochemistry and geophysics, organic and inorganic chemistry, and trends in use and availability of mineral, energy, and water resources.

Emily was a U.S. Geological Survey Hydrologist, from 1996 to July 2008, researching and publishing in the subject areas of water use, groundwater, surface water, coastal waters, saltwater intrusion and water quality. She performed water quantity and quality assessments of surface water and groundwater, fieldwork, modeling, report writing, database management, project management and STEM outreach.

Her past field work experiences include hydrologic and hydrogeologic research in Vermont, New Hampshire, Rhode Island, Massachusetts, and Connecticut, as well as geologic mapping in the Grand Canyon, Arizona ; O'ahu, Maui, and Island of Hawai'i, Hawai'i ; San Salvador Island, Bahamas ; Adirondacks, Catskills, Finger Lakes, and Hudson River, New York ; Whaleback Anticline, Pennsylvania ; and Green Mountains, Vermont.

Activities Collapse all

Employment (3) Sort

Princeton University: Princeton, New Jersey, US

https://libguides.princeton.edu/ld.php?content_id=73624938

Emily C Wild - Google Scholar

Rhode Island Water Use and Availability, 1995-99 = 7 publications

Emily Wild Rhode Island Water Use and Availability, 1995-99 = 6 publications

Current:

Estimated water use and availability in the South Coastal Drainage Basin, southern Rhode Island, 1995-99

Emily C Wild, Mark T Nimiroski

US Geological Survey (2004-5288), 2005

Edited:

Estimated water use and availability in the South Coastal Drainage Basin, southern Rhode Island, 1995-99

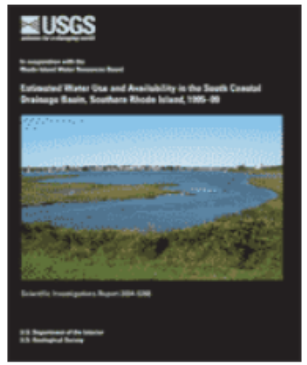
Emily C Wild

US Department of the Interior, US Geological Survey

4(4), 2005

USGS

science for a changing world



Estimated Water Use and Availability in the South Coastal Drainage Basin, Southern Rhode Island, 1995-99

By Emily C. Wild and Mark T. Nimiroski

Prepared in cooperation with the Rhode Island Water Resources Board

Scientific Investigations Report 2004-5288

https://pubs.usgs.gov/sir/2004/5288/

Generative AI, Text – Chemistry

FAQ: Should I trust AI or Chemistry Librarians?

[Turning ChatGPT into a 'chemistry assistant' - American Chemical Society (acs.org)]

[ChatGPT Chemistry Assistant for Text Mining and the Prediction of MOF Synthesis | Journal of the American Chemical Society (acs.org)]

FAQ: Is there an environmental impact on water resources from AI?

[Microsoft's water usage surges by thousands of gallons after the launch of ChatGPT: Study – BusinessToday]

Every time you talk to ChatGPT it drinks 500ml of water; here's why

OpenAI's AI chatbot ChatGPT gulps 500ml of water for every 5-50 prompts it answers

Princeton University Library

Peter B. Lewis Library - Lewis Science Library

https://library.princeton.edu/

2008 - https://facilities.princeton.edu/projects/peter-b-lewis-library-2008

ABOUT RECAP

The Research Collections and Preservation Consortium (ReCAP) was created in 2000 to support its members' goals of preserving their library and archival collections, and making them available to researchers. The ReCAP facility consists of a preservation repository and resource sharing services, and is located on Princeton University's Forrestal Campus. ReCAP is jointly owned and operated by Columbia University, Harvard University, The New York Public Library and Princeton University

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Materials Collection

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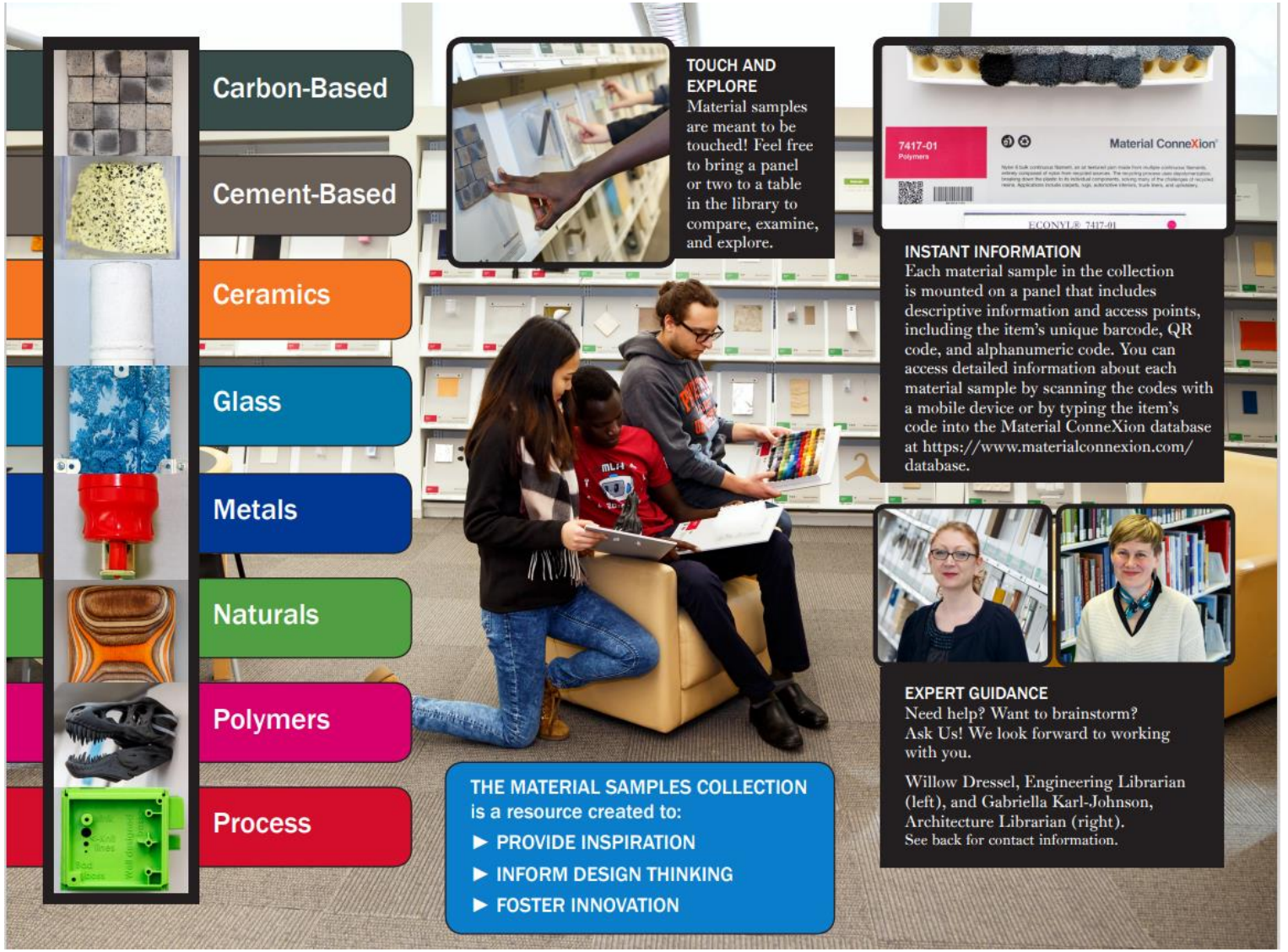
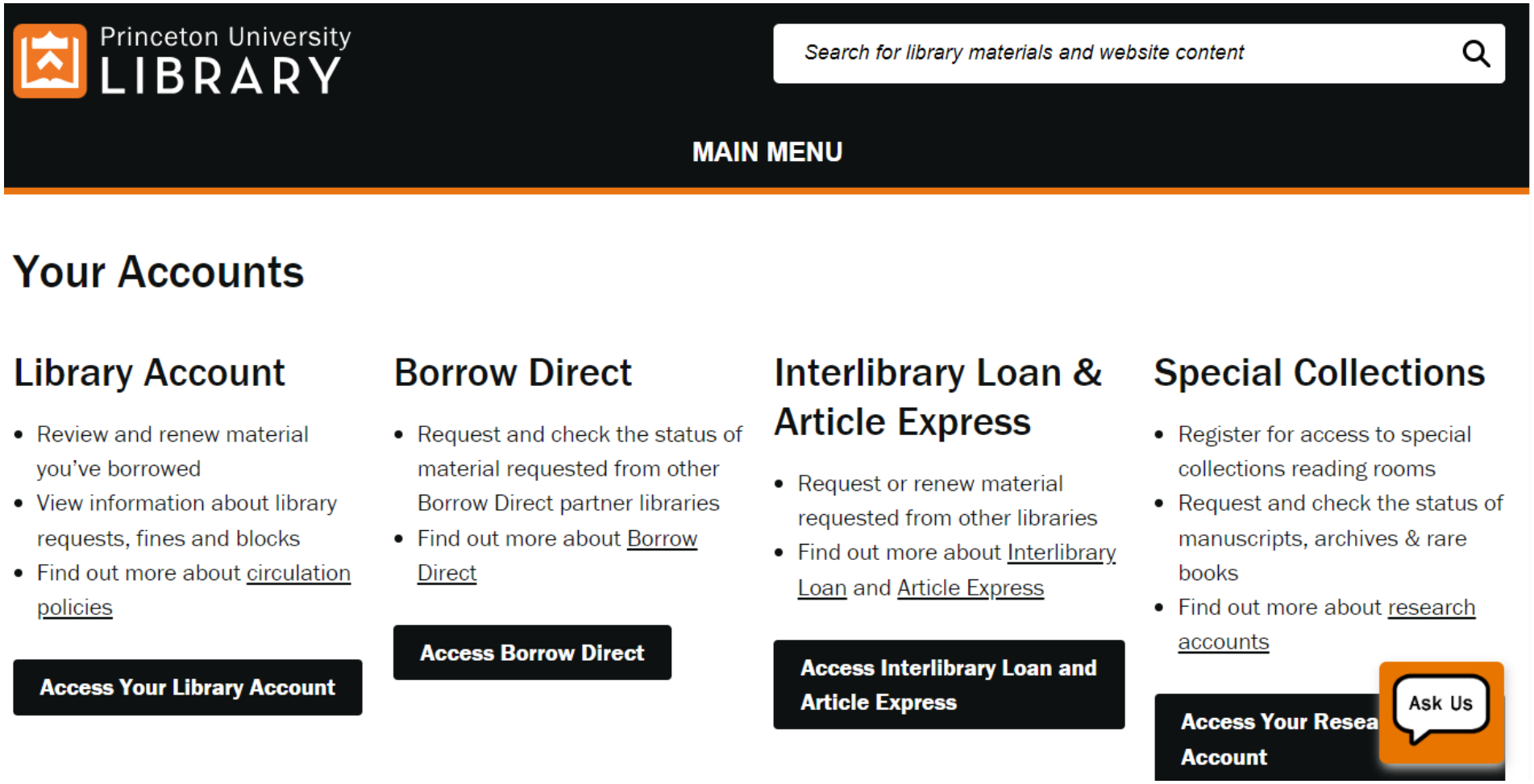
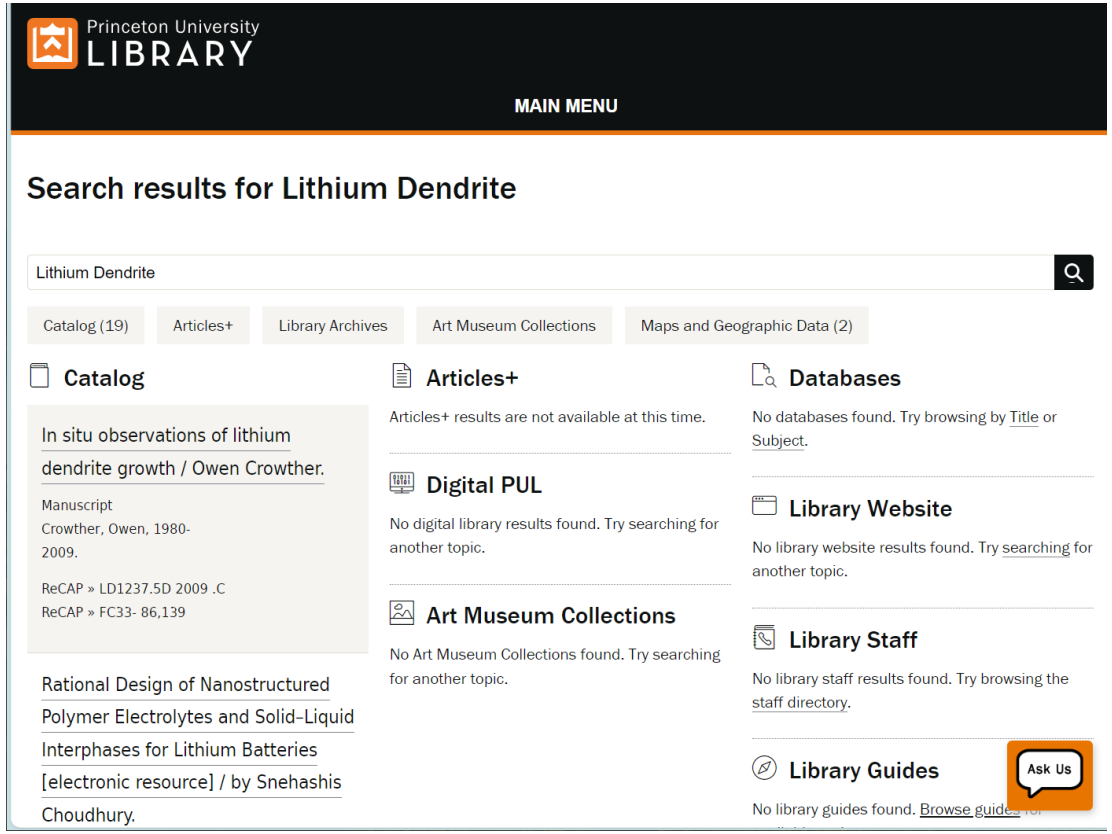


[Your Accounts | Princeton University Library]

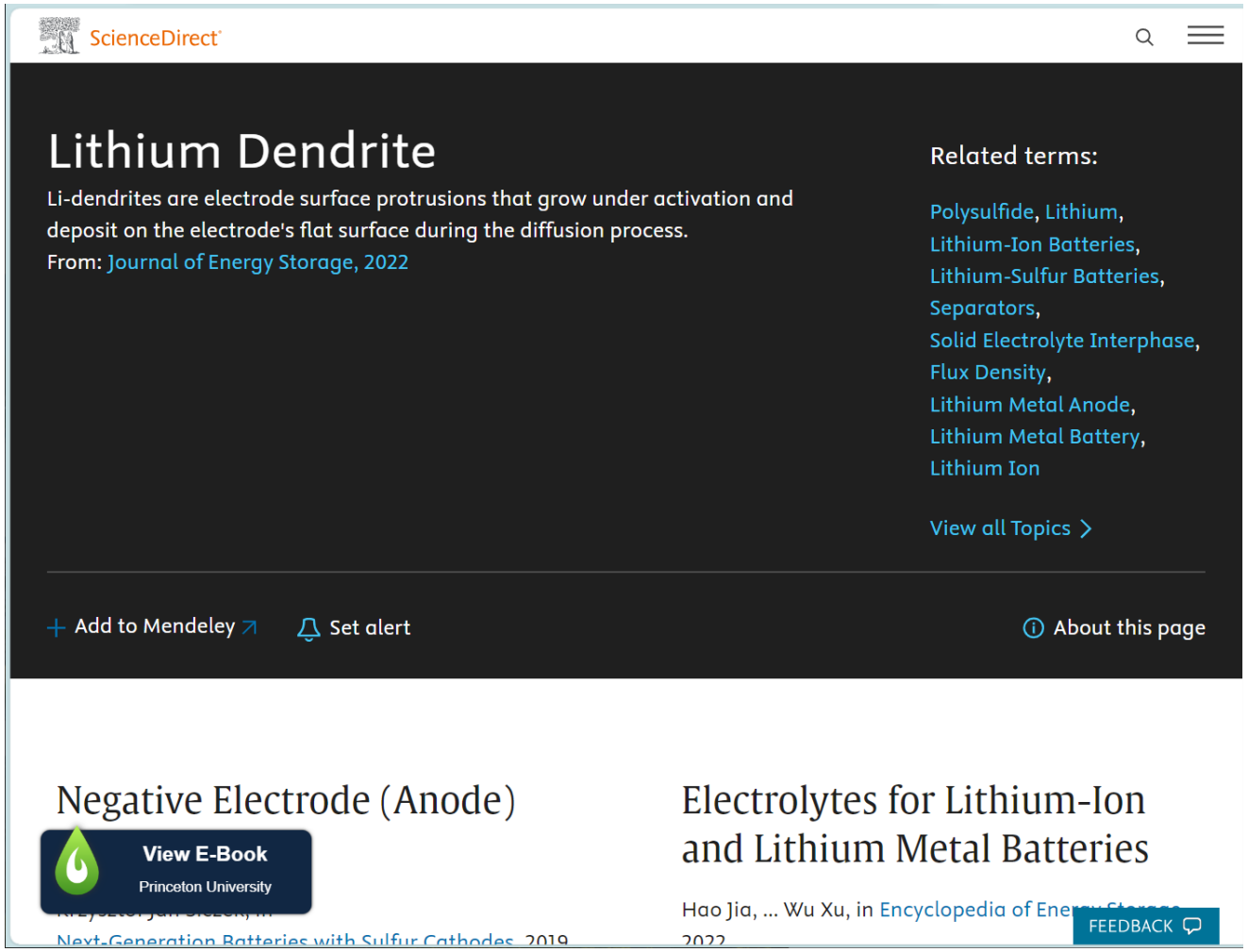


Library Catalog v. Web Search

Search results for Lithium Dendrite | Princeton University Library

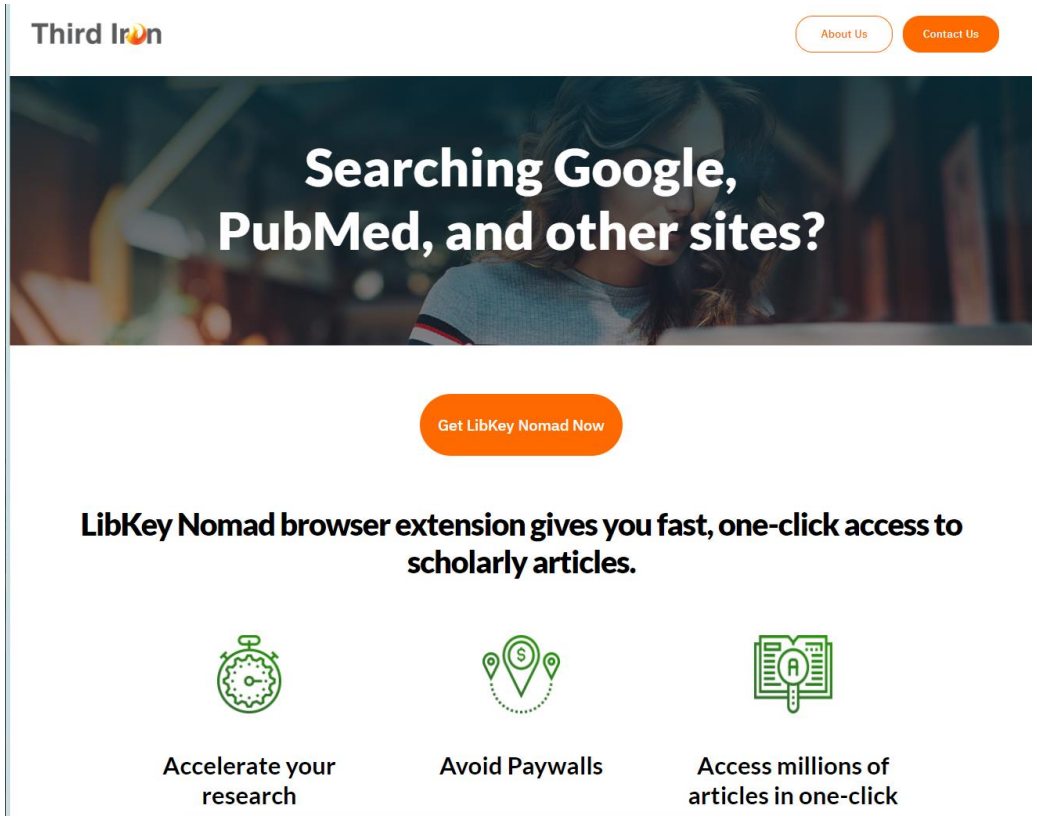


lithium dendrites at DuckDuckGo

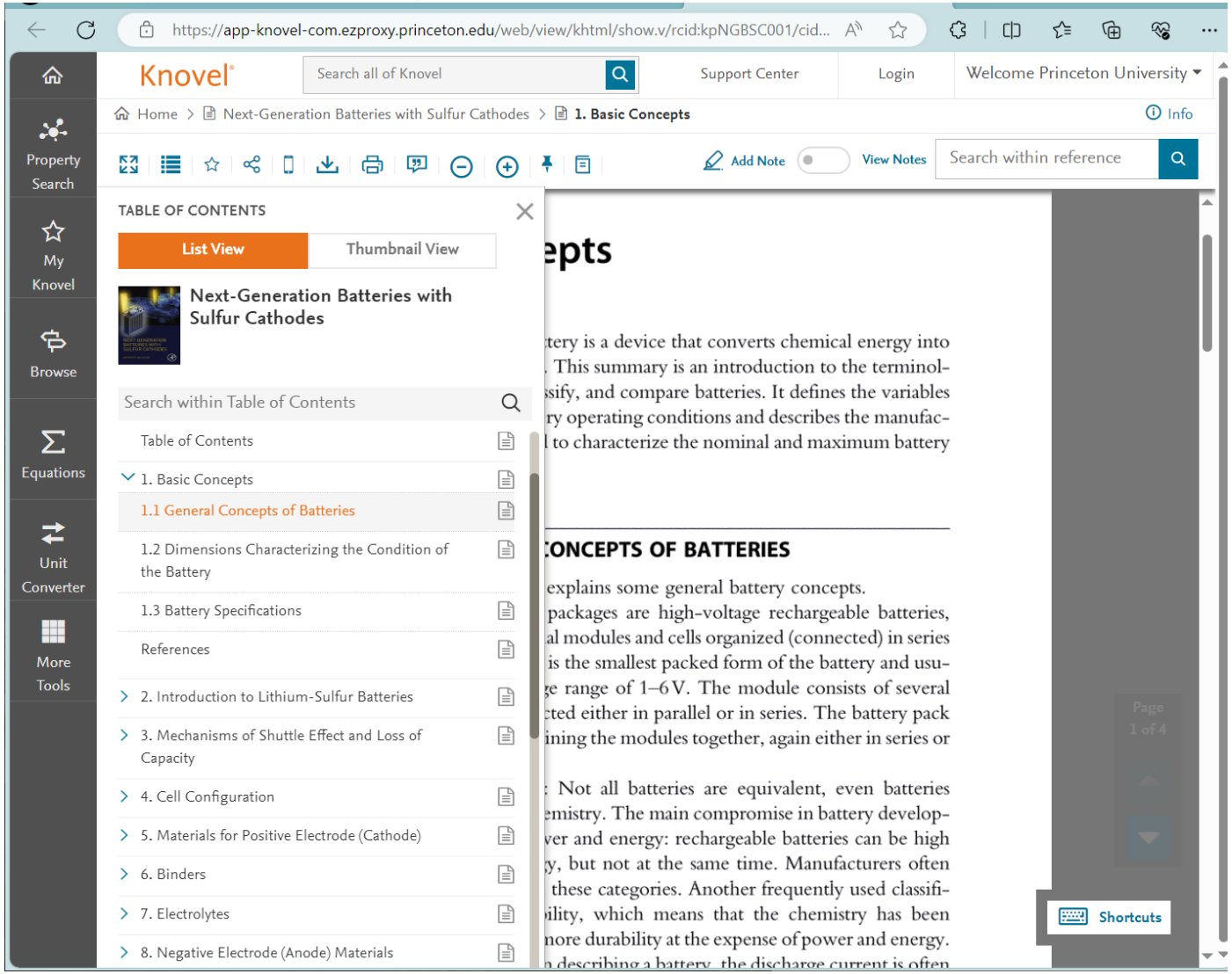


LibKey Nomad

[Download LibKey Nomad – Third Iron]

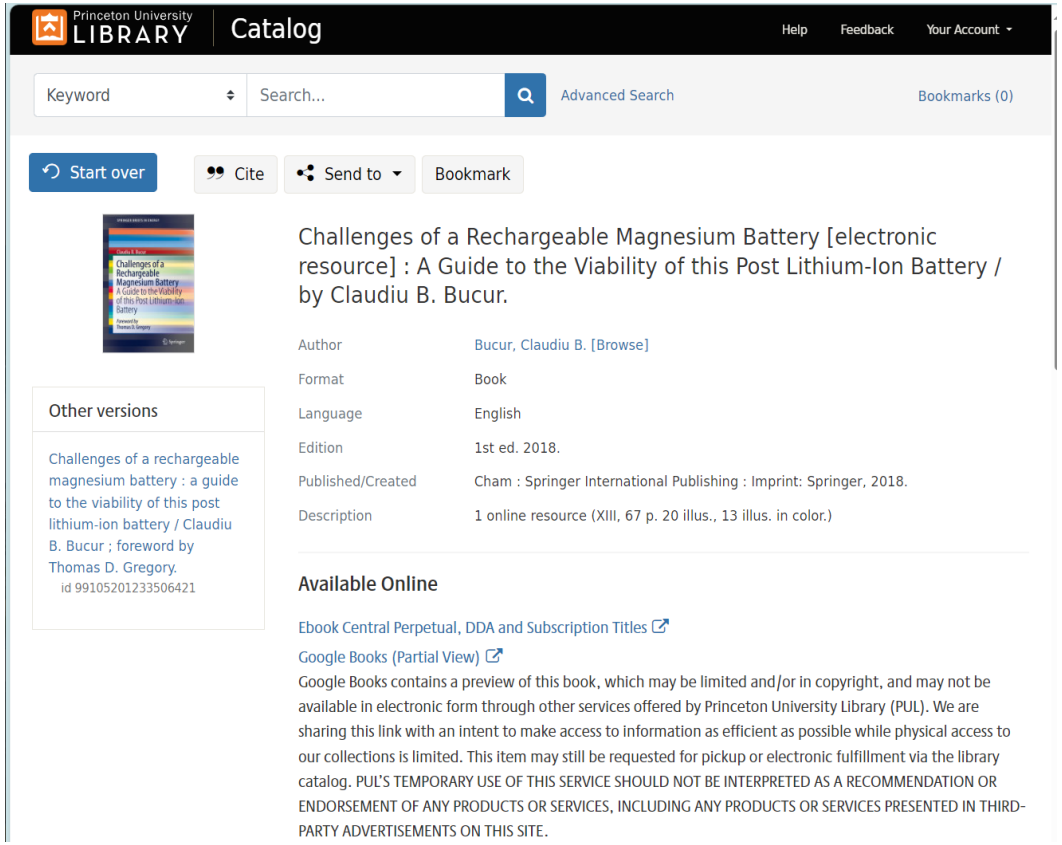


[1. Basic Concepts - Knovel (princeton.edu)]

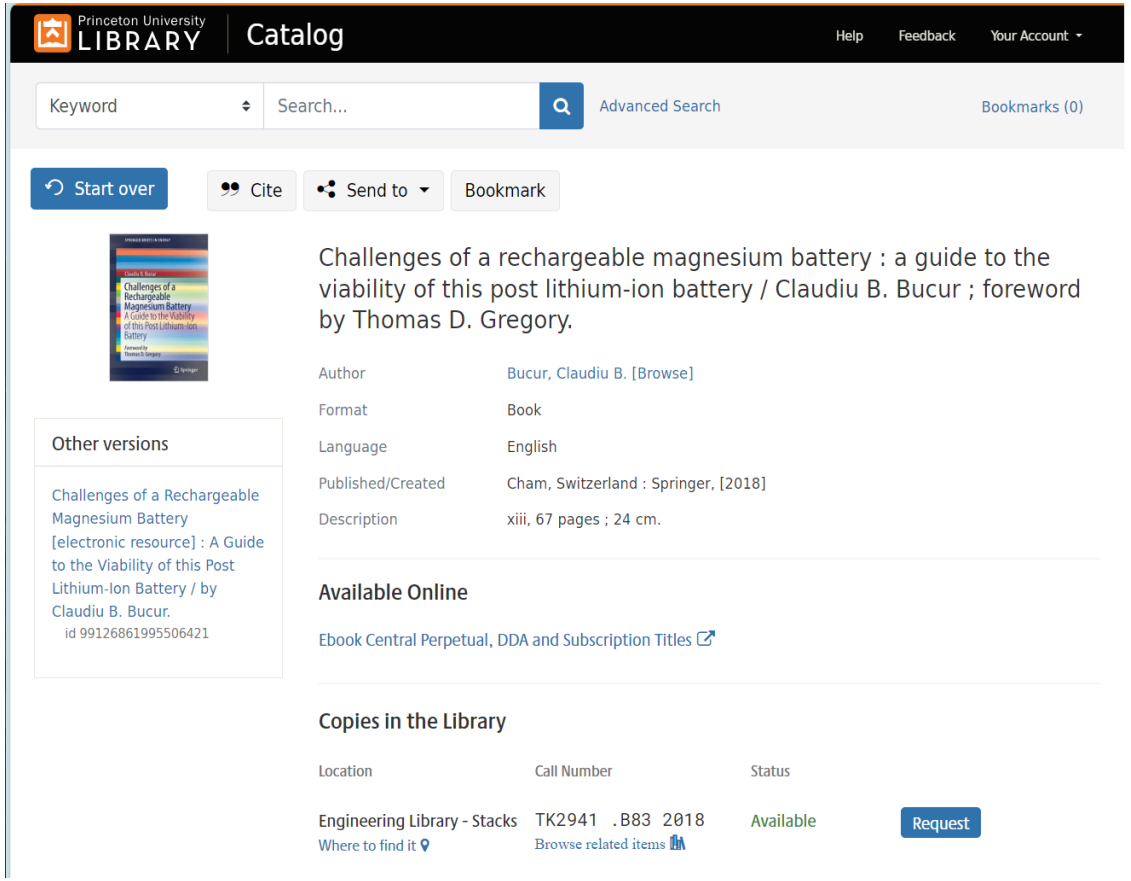


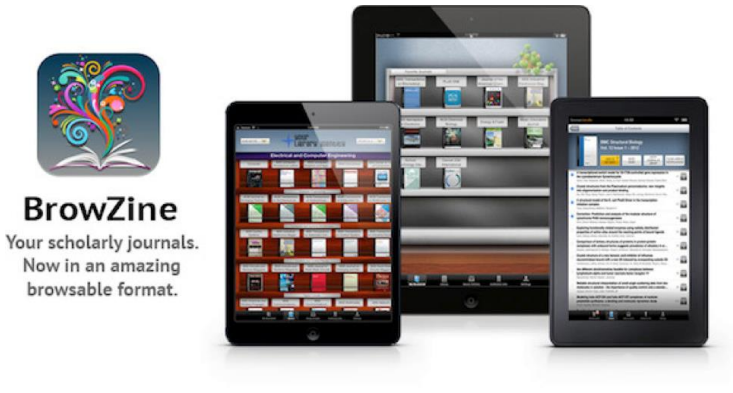
Library Catalog

Challenges of a Rechargeable Magnesium Battery [electronic resource] : A Guide to the Viability of this Post Lithium-Ion Battery / by Claudiu B. Bucur. - Princeton University Library Catalog



Challenges of a rechargeable magnesium battery : a guide to the viability of this post lithium-ion battery / Claudiu B. Bucur ; foreword by Thomas D. Gregory. - Princeton University Library Catalog





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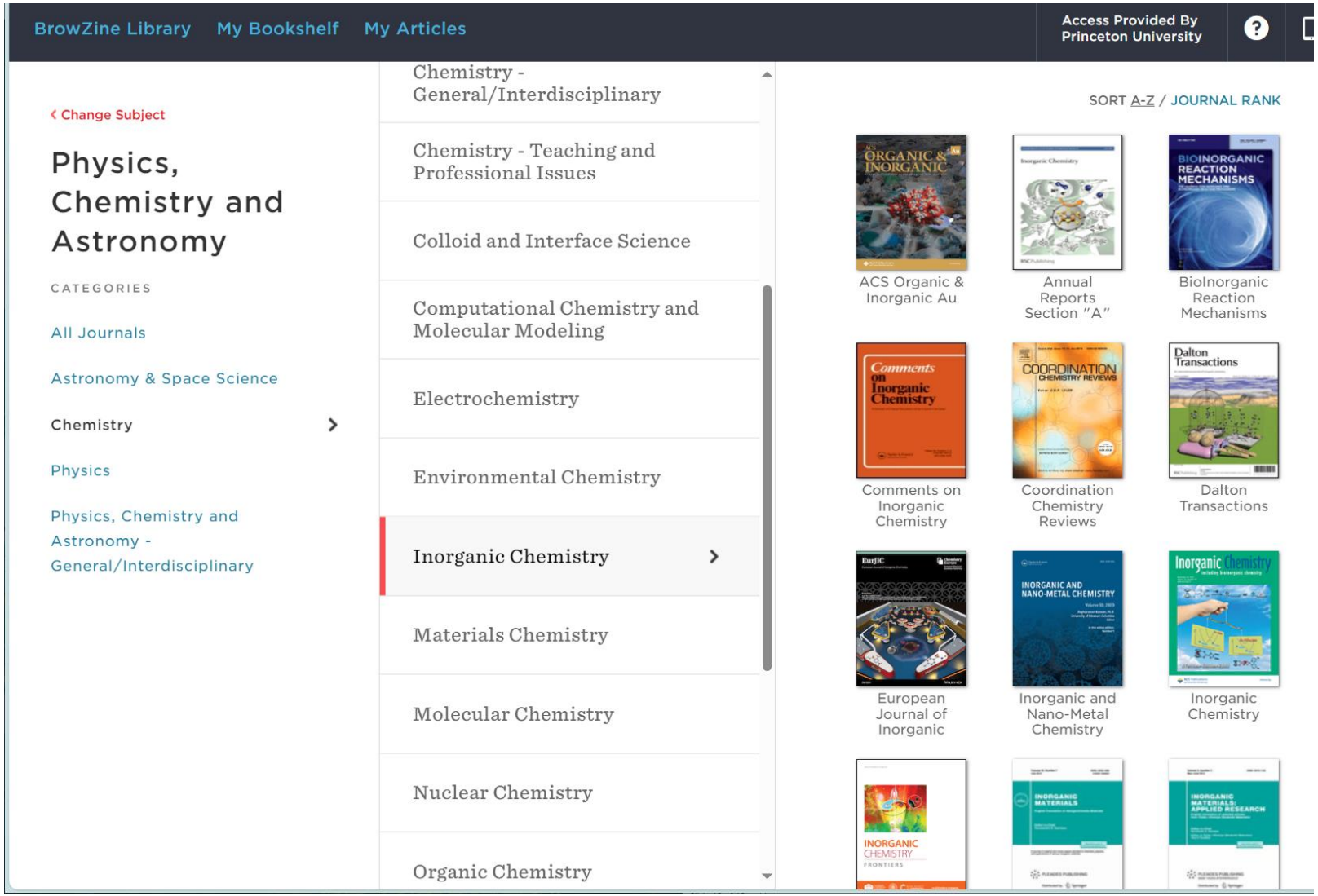
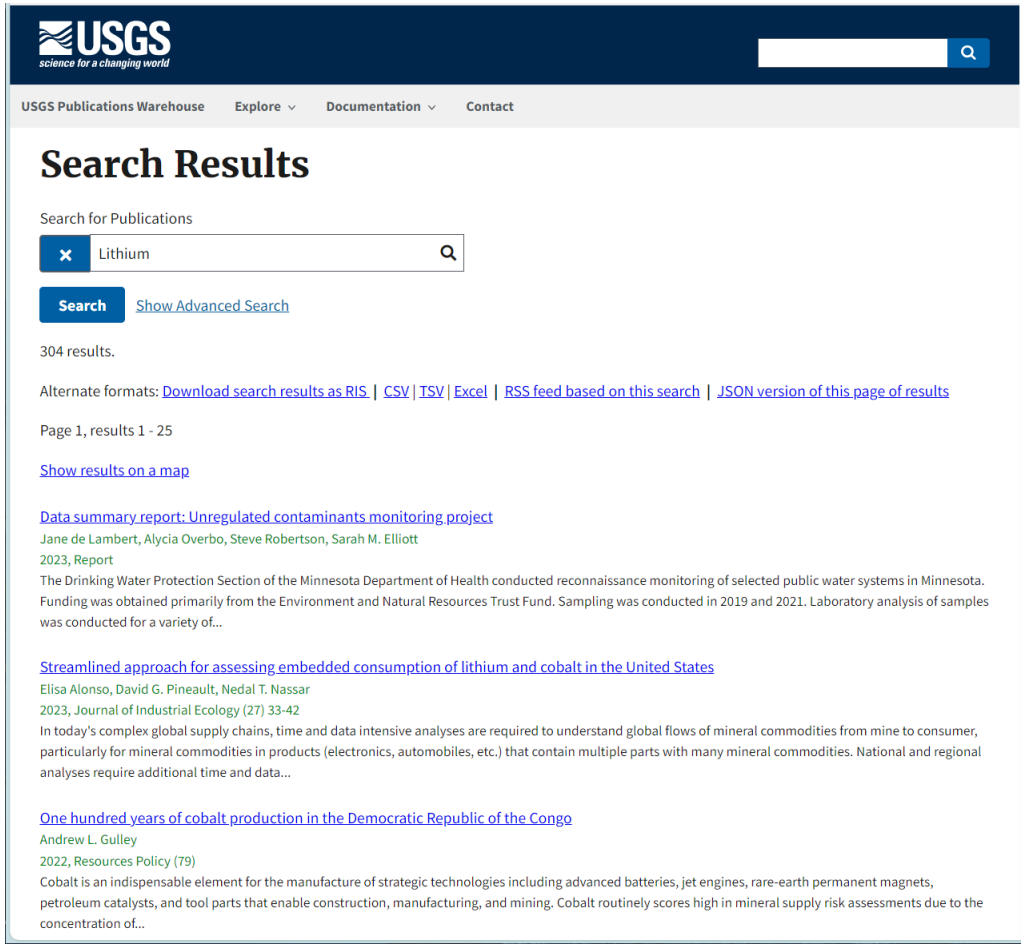
Inorganic Chemistry — Chemistry — Princeton University — BrowZine



USGS Lithium

[Search Results - USGS Publications Warehouse]

[Lithium Statistics and Information | U.S. Geological Survey (usgs.gov)]



USGS science for a changing world

USGS Publications Warehouse | Explore | Documentation | Contact

Search Results

Search for Publications

Lithium

Search | Show Advanced Search

304 results.

Alternate formats: Download search results as RIS | CSV | TSV | Excel | RSS feed based on this search | JSON version of this page of results

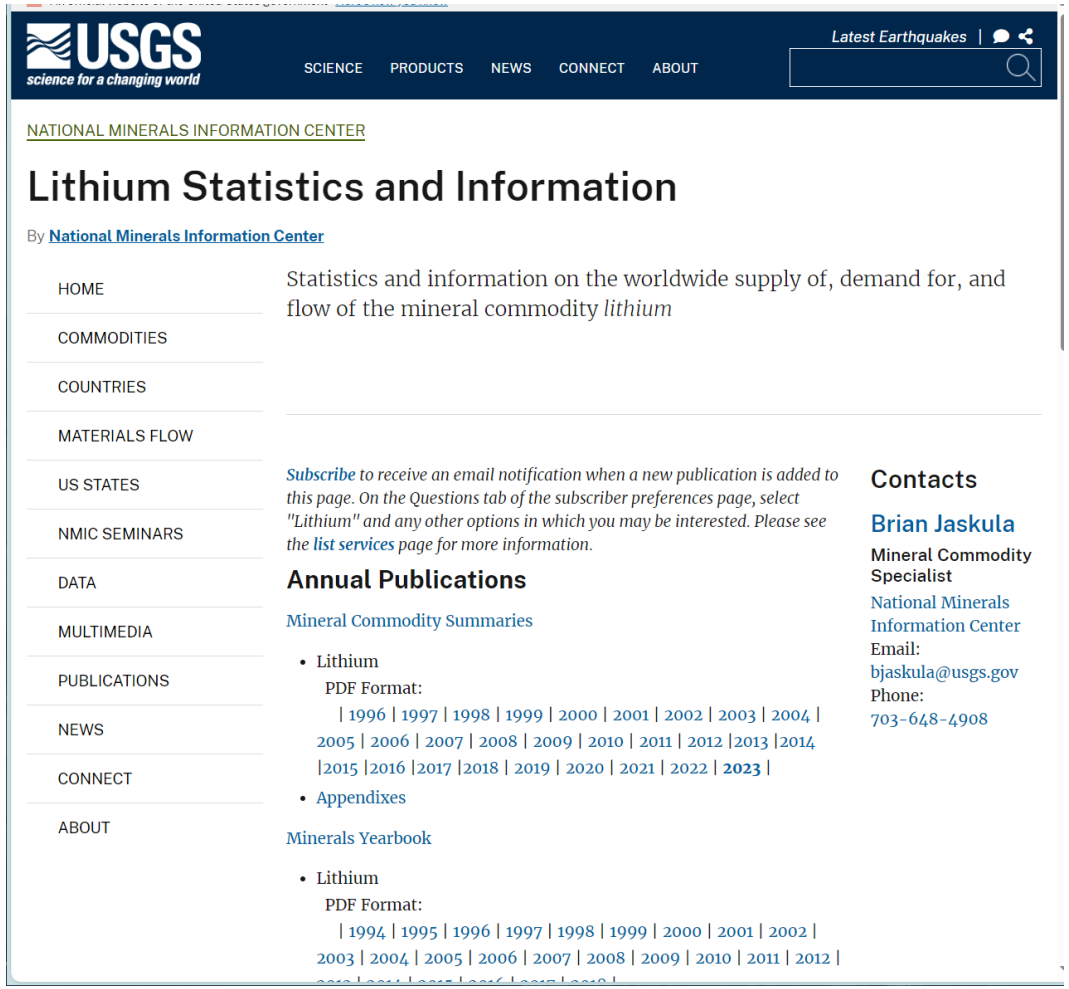
Page 1, results 1 - 25

Show results on a map

Data summary report: Unregulated contaminants monitoring project
Jane de Lambert, Alycia Overbo, Steve Robertson, Sarah M. Elliott
2023, Report
The Drinking Water Protection Section of the Minnesota Department of Health conducted reconnaissance monitoring of selected public water systems in Minnesota. Funding was obtained primarily from the Environment and Natural Resources Trust Fund. Sampling was conducted in 2019 and 2021. Laboratory analysis of samples was conducted for a variety of...

Streamlined approach for assessing embedded consumption of lithium and cobalt in the United States
Elisa Alonso, David G. Pineault, Nedal T. Nassar
2023, Journal of Industrial Ecology (27) 33-42
In today's complex global supply chains, time and data intensive analyses are required to understand global flows of mineral commodities from mine to consumer, particularly for mineral commodities in products (electronics, automobiles, etc.) that contain multiple parts with many mineral commodities. National and regional analyses require additional time and data...

One hundred years of cobalt production in the Democratic Republic of the Congo
Andrew L. Gulley
2022, Resources Policy (79)
Cobalt is an indispensable element for the manufacture of strategic technologies including advanced batteries, jet engines, rare-earth permanent magnets, petroleum catalysts, and tool parts that enable construction, manufacturing, and mining. Cobalt routinely scores high in mineral supply risk assessments due to the concentration of...



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NATIONAL MINERALS INFORMATION CENTER

Lithium Statistics and Information

By National Minerals Information Center

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Statistics and information on the worldwide supply of, demand for, and flow of the mineral commodity *lithium*

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Annual Publications

Mineral Commodity Summaries

- Lithium
  PDF Format:
  | 1996 | 1997 | 1998 | 1999 | 2000 | 2001 | 2002 | 2003 | 2004 | 2005 | 2006 | 2007 | 2008 | 2009 | 2010 | 2011 | 2012 | 2013 | 2014 | 2015 | 2016 | 2017 | 2018 | 2019 | 2020 | 2021 | 2022 | **2023** |
- Appendixes

Minerals Yearbook

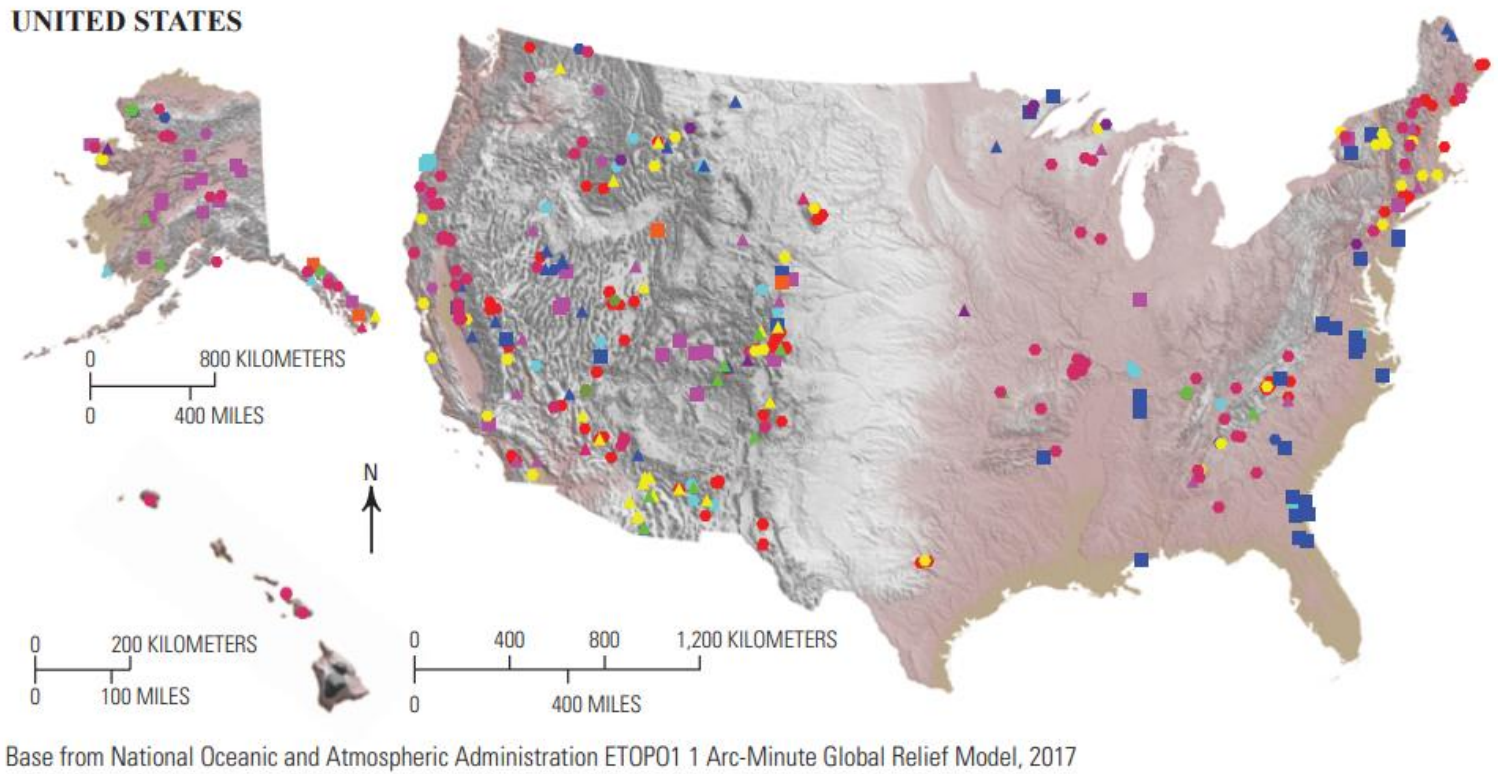
- Lithium
  PDF Format:
  | 1994 | 1995 | 1996 | 1997 | 1998 | 1999 | 2000 | 2001 | 2002 | 2003 | 2004 | 2005 | 2006 | 2007 | 2008 | 2009 | 2010 | 2011 | 2012 |

Contacts

Brian Jaskula
Mineral Commodity Specialist
National Minerals Information Center
Email: bjaskula@usgs.gov
Phone: 703-648-4908

Critical Minerals - Lithium

[Australia, Canada and US Unify Critical Minerals Data | U.S. Geological Survey (usgs.gov)]



Base from National Oceanic and Atmospheric Administration ETOPO1 1 Arc-Minute Global Relief Model, 2017

Figure 1. Critical mineral resources in Australia, Canada, and the United States (Labay and others, 2017). Critical minerals support a broad range of industrial sectors and a diversity of high-tech industries important to global economies (see Primary Uses of Critical Minerals sidebar).

Chemistry Databases

https://libguides.princeton.edu/chemistry

Databases | Guides

Showing 46 Databases

Starting Points

CHEMnetBASE Starting Point

Consists of several chemical data handbooks from Chapman & Hall/CRC.

Encyclopedia of Polymer Science and Technology Starting Point

Comprehensive coverage of the entire field of polymer science and technology. Includes over 400 articles covering fundamental and practical aspects of polymer science including materials, properties, methods, applications, processing, and analytical methods.

KnowItAll U spectra database Starting Point

Offers access to the KnowItAll U reference collection of over 1.3 million spectra including structures and chemical property information; includes IR, UV/Vis, NMR, Raman and mass spectra. ATTENTION KnowItALL Desktop Software Users: Please update your license at...

more...

Pharmaceutical Substances Starting Point

Authoritative information resource for literature and data associated with the syntheses, patents, and applications for active pharmaceutical ingredients (APIs).

PubChem Starting Point

PubChem is the world's largest collection of freely accessible chemical information. Search chemicals by name, molecular formula, structure, and other identifiers. Find chemical and physical properties, biological activities, safety and toxicity information, patents, literature citations and more.

Reaxys Starting Point

Detailed index covering organic and inorganic chemistry. Includes patents. Provides access to chemical reactions and physical, chemical and bioactivity data 1772+

Science of Synthesis Starting Point

Considers and critically reviews methods from journals, books, and patent literature and presents important synthetic methods for all classes of organic compounds. Web version covers former Houben-Weyl treatise (1952-2003) and its continuation title, Science of Synthesis (2001+). Critically reviews chemical reaction literature with emphasis on functional group conversions.

SciFinder Web (Chemical Abstracts) Starting Point

Comprehensive coverage of all aspects of chemistry. Combines several Chemical Abstracts Service (CAS) databases. 1907+

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Chemistry Experts

Chemistry, Geosciences and Environmental Studies Librarian

Emily Wild

Email Me

ewild@princeton.edu

Contact:

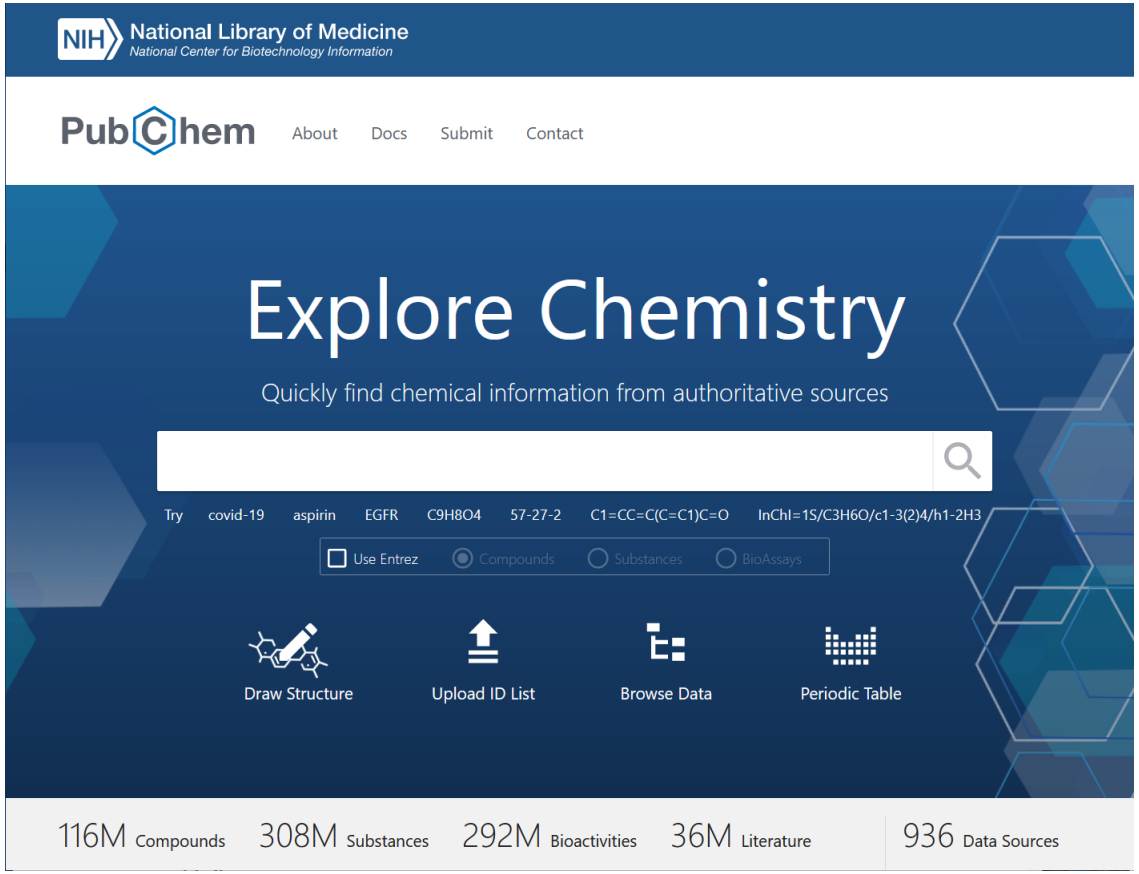
212 Lewis Science Library

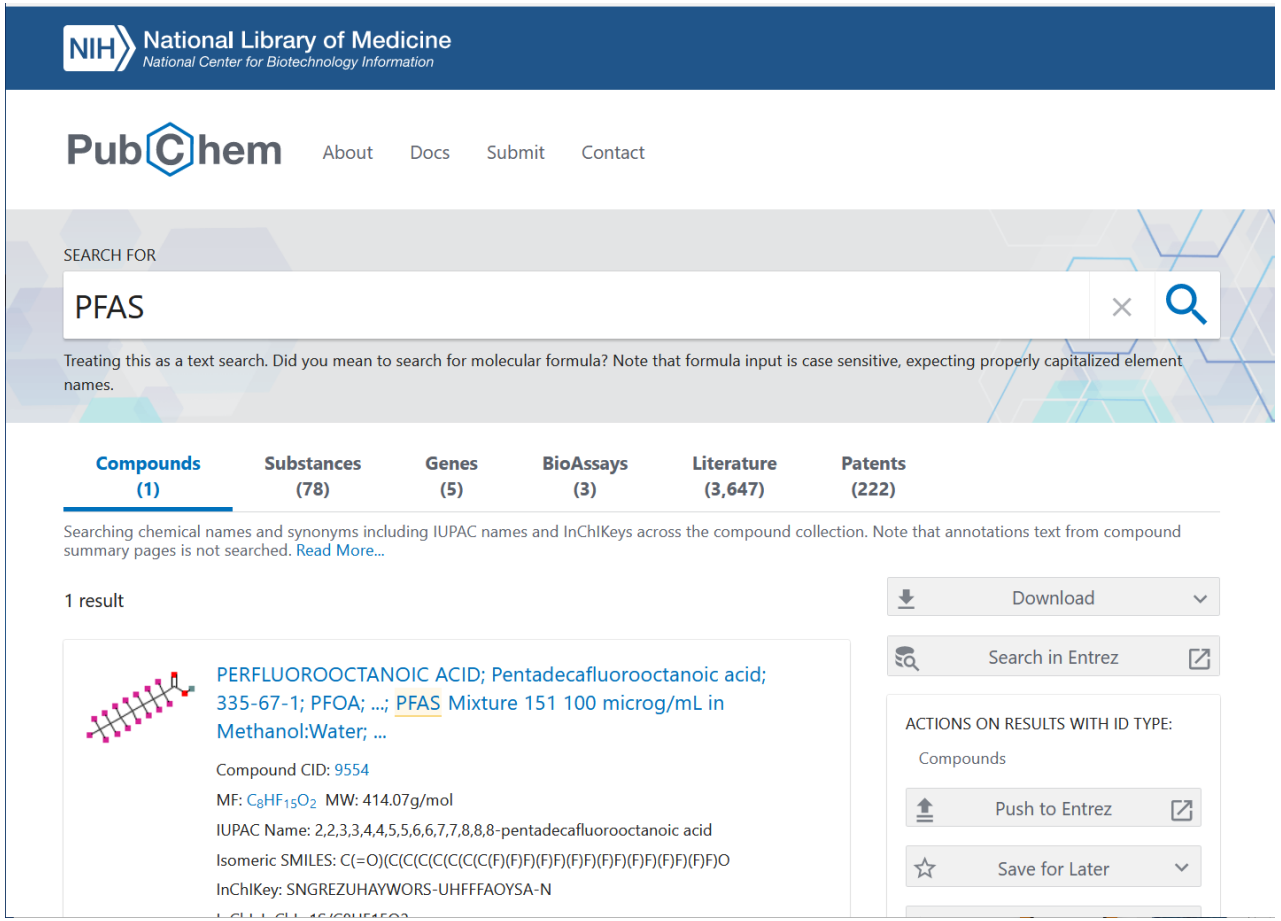
609-258-5484

PubChem

https://pubchem.ncbi.nlm.nih.gov/

"per- and polyfluorinated alkyl substances"





Pharmaceutical Substances

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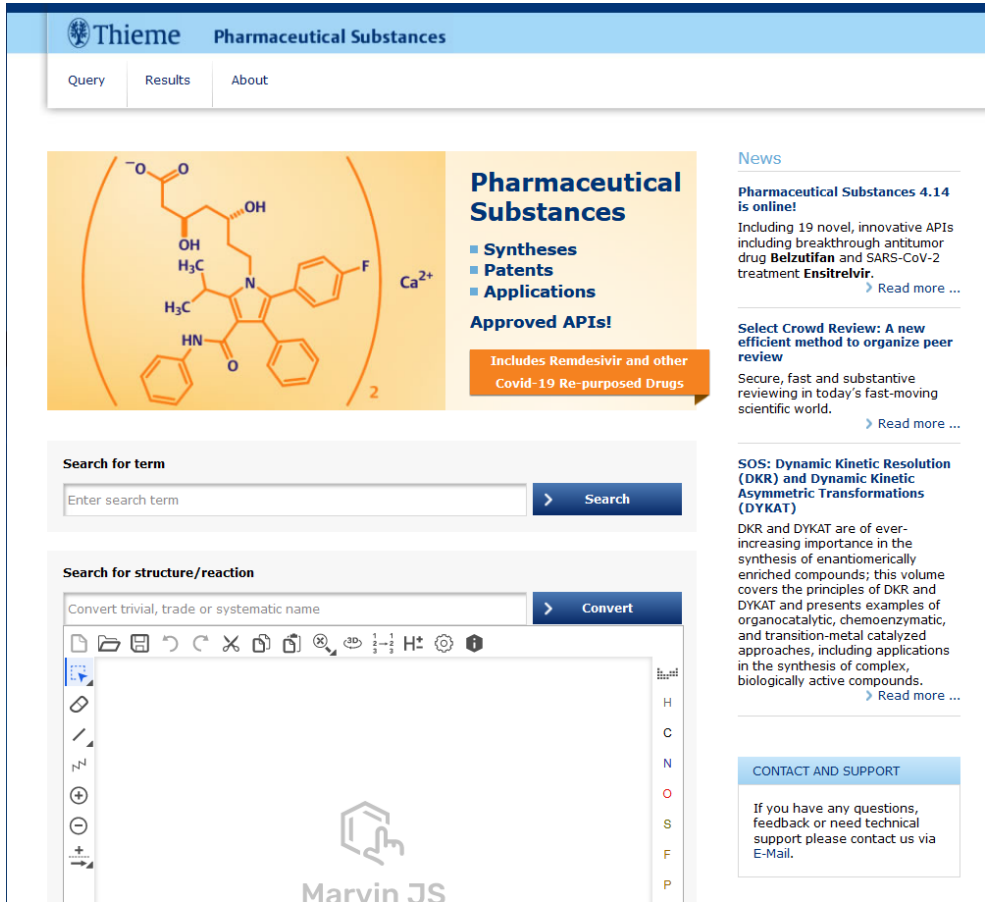
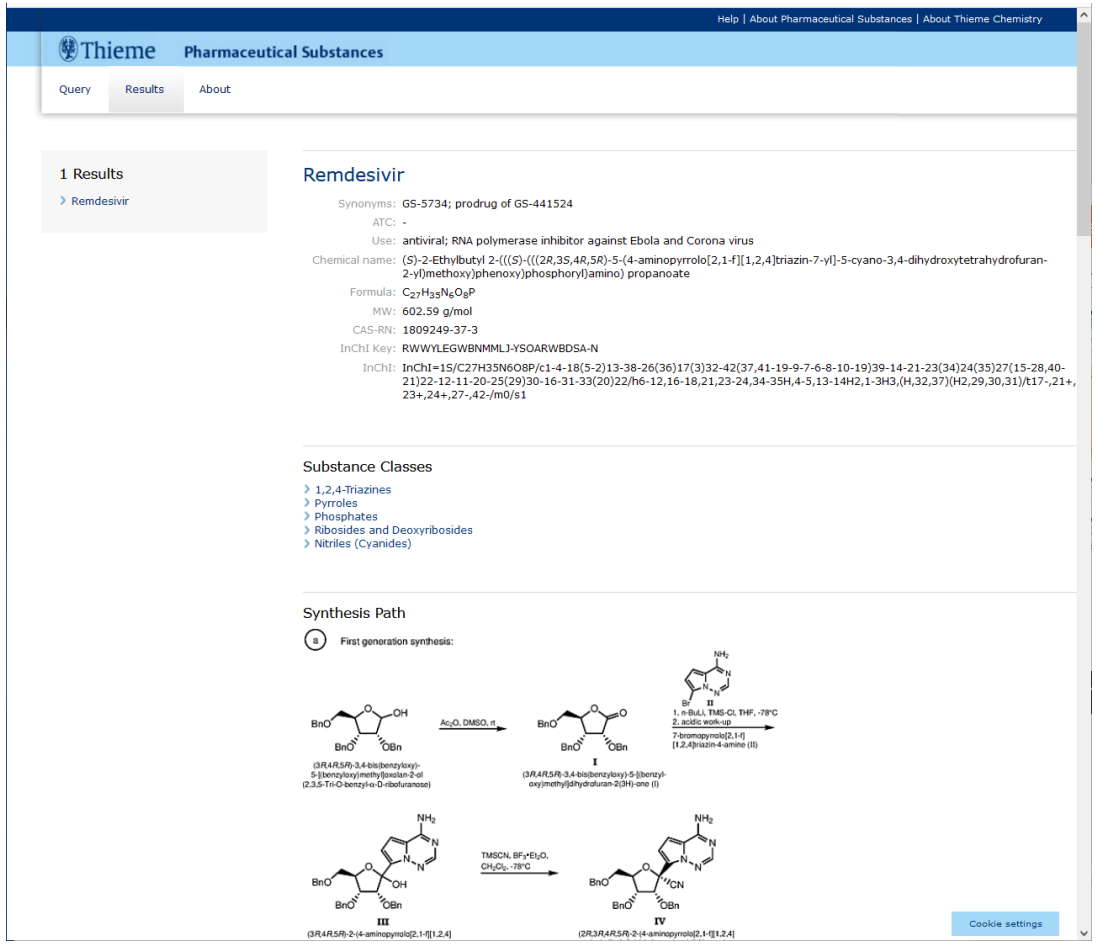
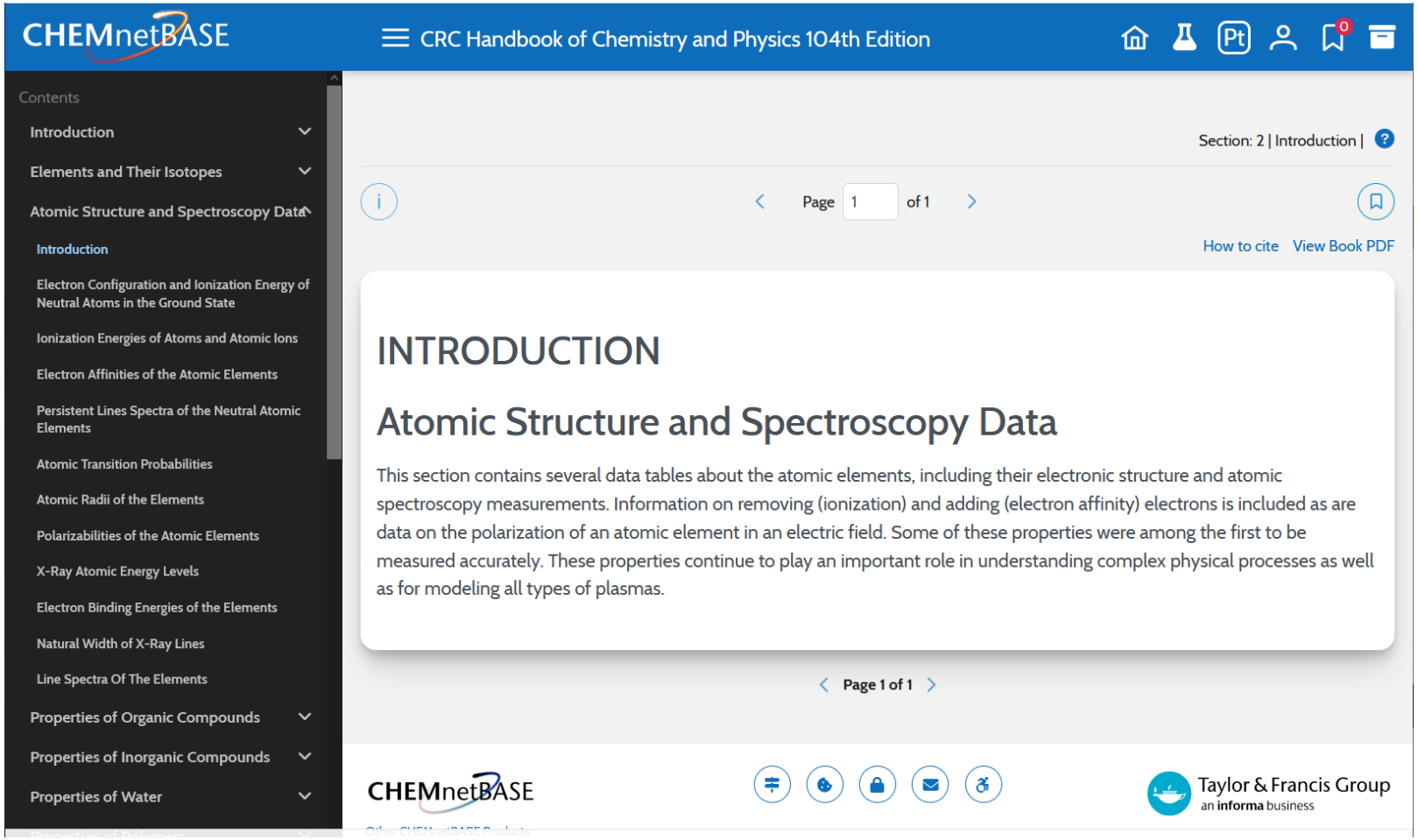




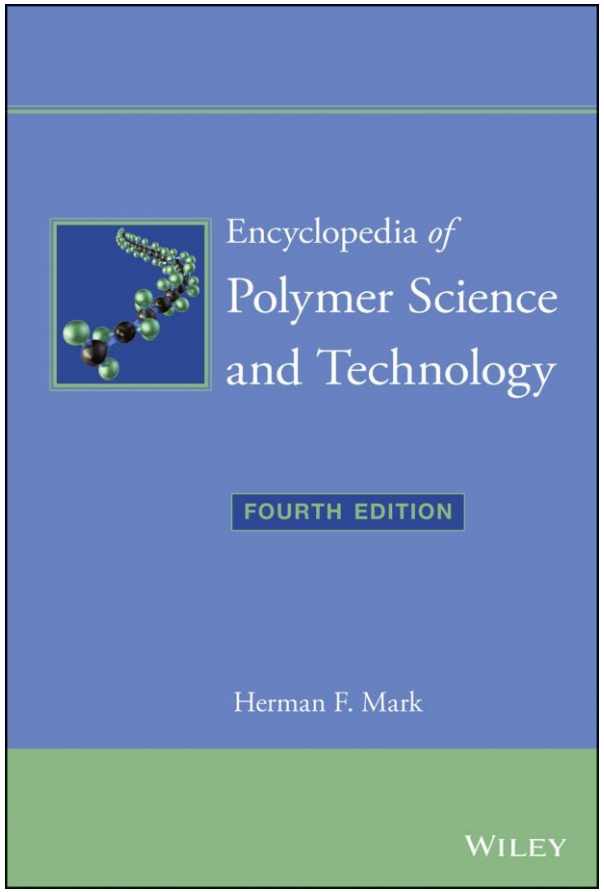
Handbook of Chemistry and Physics (CRC)

https://hbcp-chemnetbase-com.ezproxy.princeton.edu/documents/02_01/02_01_0001.xhtml?dswid=-6552



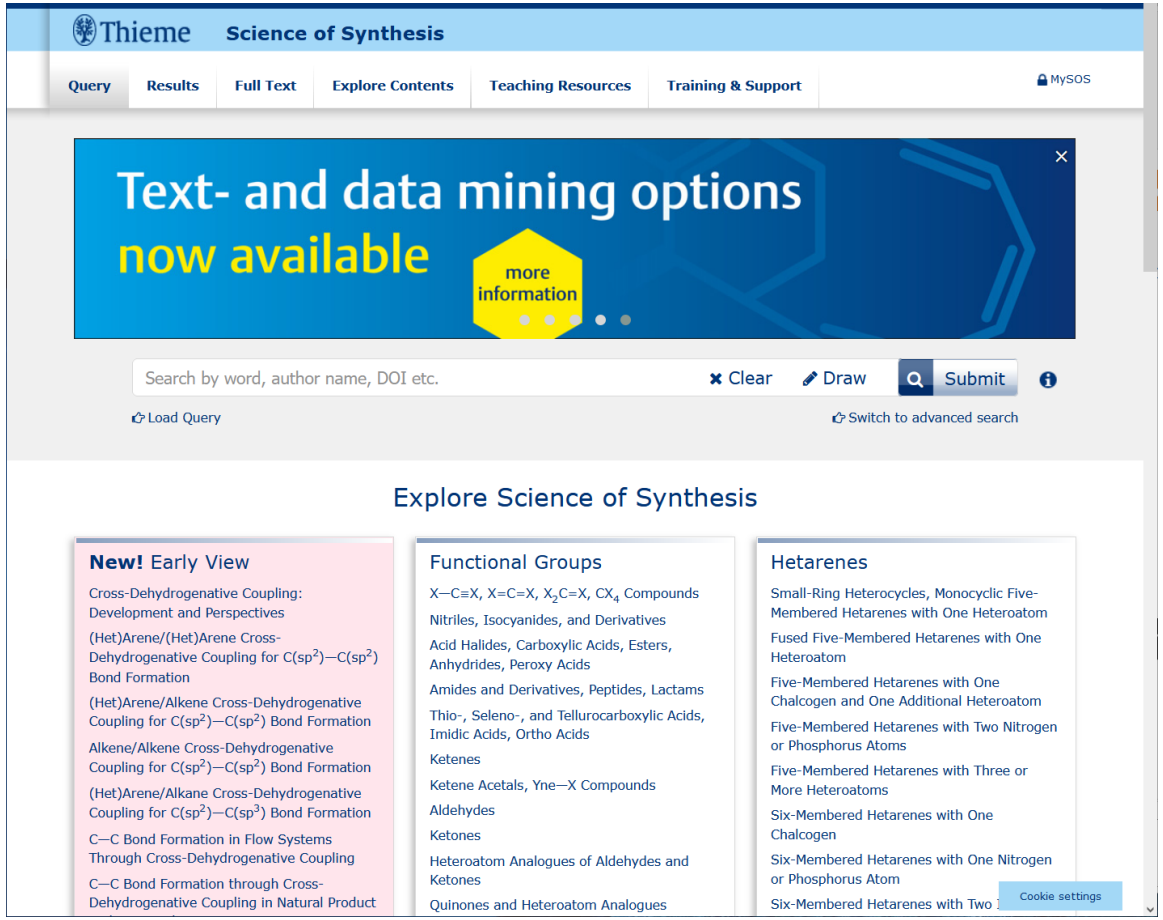
Encyclopedia of Polymer Science and Technology

https://onlinelibrary-wiley-com.ezproxy.princeton.edu/doi/book/10.1002/0471440264



Science of Synthesis

https://science-of-synthesis-thieme-com.ezproxy.princeton.edu/



SciFinder-n

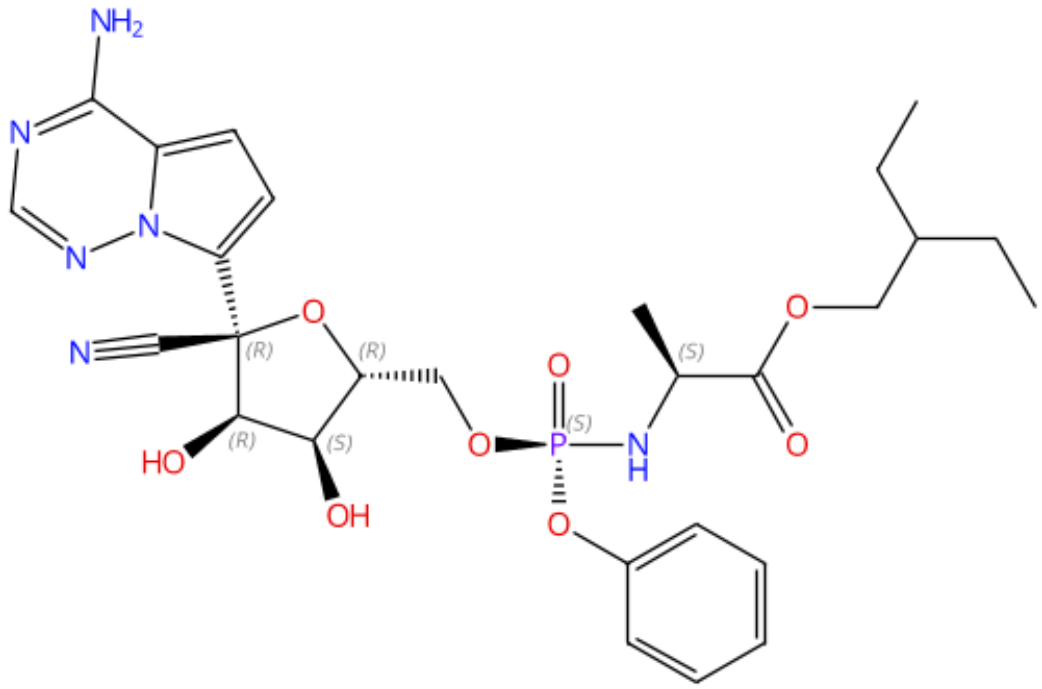
Remdesivir

← Return to Results

← Prev (1 of 1) Next →

CAS Registry Number: 1809249-37-3

References (4,967) | Reactions (747) | Suppliers (65) | Save



Absolute stereochemistry shown.

CAS SciFinder^n | Substances | remdesivir | Draw

$C_{27}H_{35}N_6O_8P$

L-Alanine, *N*-[(*S*)-hydroxyphenoxyphosphinyl]-, 2-ethylbutyl ester, 6-ester with 2-*C*-(4-aminopyrrolo[2,1-*f*][1,2,4]triazin-7-yl)-2,5-anhydro-D-altrononitrile (ACI)

Key Physical Properties	Value	Condition
Molecular Weight	602.58	-
Melting Point (Experimental)	89.4-90.4 °C	-
Density (Predicted)	1.47±0.1 g/cm^3	Temp: 20 °C; Press: 760 Torr
pKa (Predicted)	12.00±0.70	Most Acidic Temp: 25 °C

Experimental Properties | Spectra

Expand All | Collapse All

Other Names and Identifiers

Canonical SMILES

N#CC1(OC(COP(=O)(OC=2C=CC=CC2)NC(C(=O)OCC(CC)CC)C)C(O)C1O)C3=CC=C4C(=NC=NN43)N

Isomeric SMILES

C(#N)[C@]1(O[C@H](COP(OC2=CC=CC=C2)(N[C@H](C(OCC(CC)CC)=O)C)=O)[C@@H](O)[C@H]1O)C=3N4C(=CC3)C(N)=NC=N4

InChI

InChI=1S/C27H35N6O8P/c1-4-18(5-2)13-38-26(36)17(3)32-42(37,41-19-9-7-6-8-10-19)39-14-21-23(34)24(35)27(15-28,40-21)22-12-11-20-25(29)30-16-31-33(20)22/h6-12,16-18,21,23-24,34-35H,4-5,13-14H2,1-3H3,(H,32,37)(H2,29,30,31)/t17-,21+,23+,24+,27-,42-/m0/s1

InChI Key

RWWYLEGWBNMMLJ-YSOARWBDSA-N

10 Other Names for this Substance

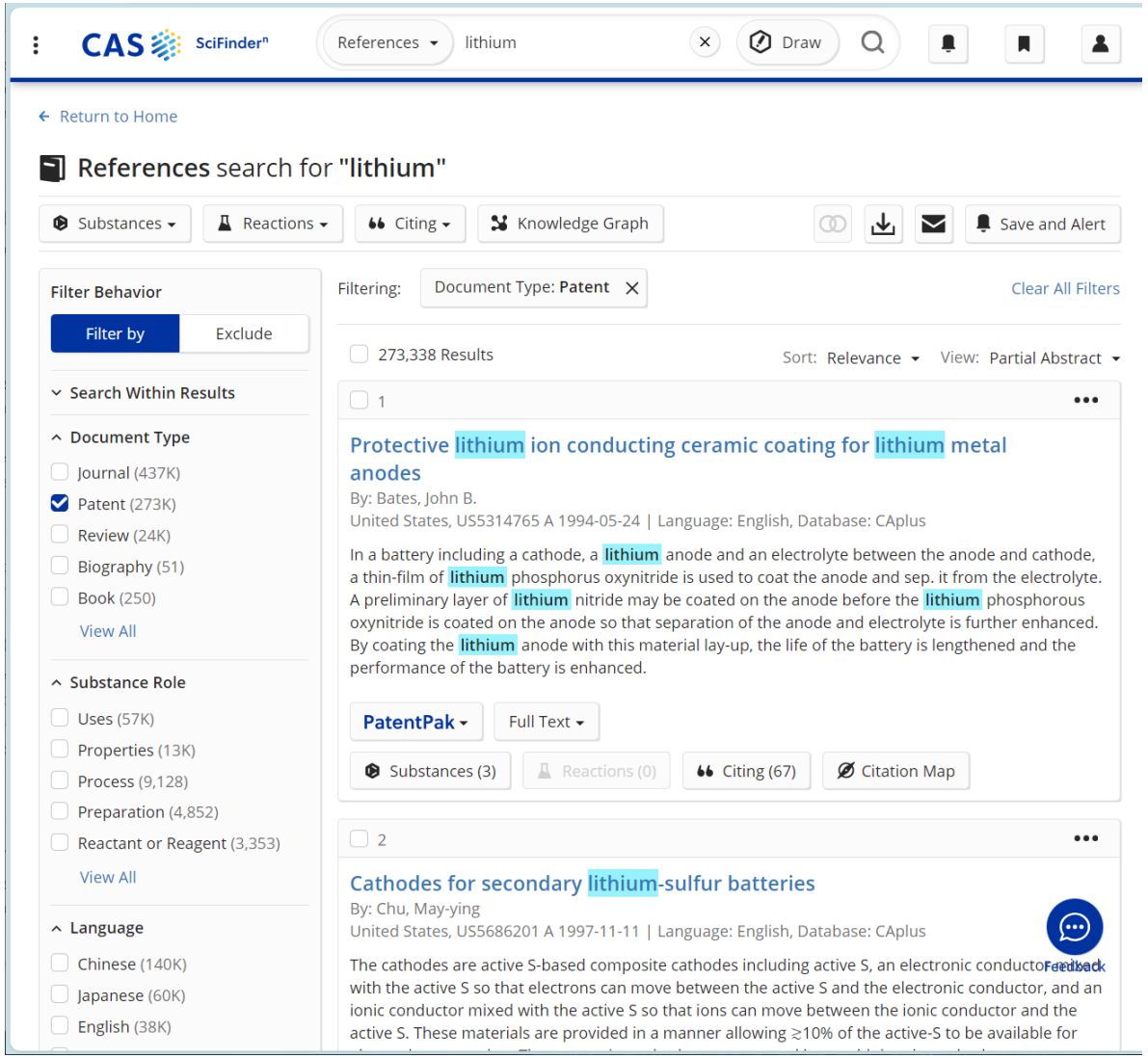
Remdesivir

(*S*)-2-Ethylbutyl 2-(((*S*)-(((2R,3S,4R,5R)-5-(4-aminopyrrolo[2,1-*f*][1,2,4]triazin-7-yl)-5-cyano-3,4-dihydroxytetrahydrofuran-2-yl)methoxy)(phenoxy)phosphoryl)amino)propanoate

2-Ethylbutyl (2*S*)-2-[[(*S*)-[[(2*R*,3*S*,4*R*,5*R*)-5-(4-aminopyrrolo(2,1-*f*)(1,2,4)triazin-7-yl)-5-cyano-3,4-dihydroxytetrahydrofuran-2-

Feedback

SciFinder-n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References
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References search for "lithium"
Substances
Reactions
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Patent (273K)
Review (24K)
Biography (51)
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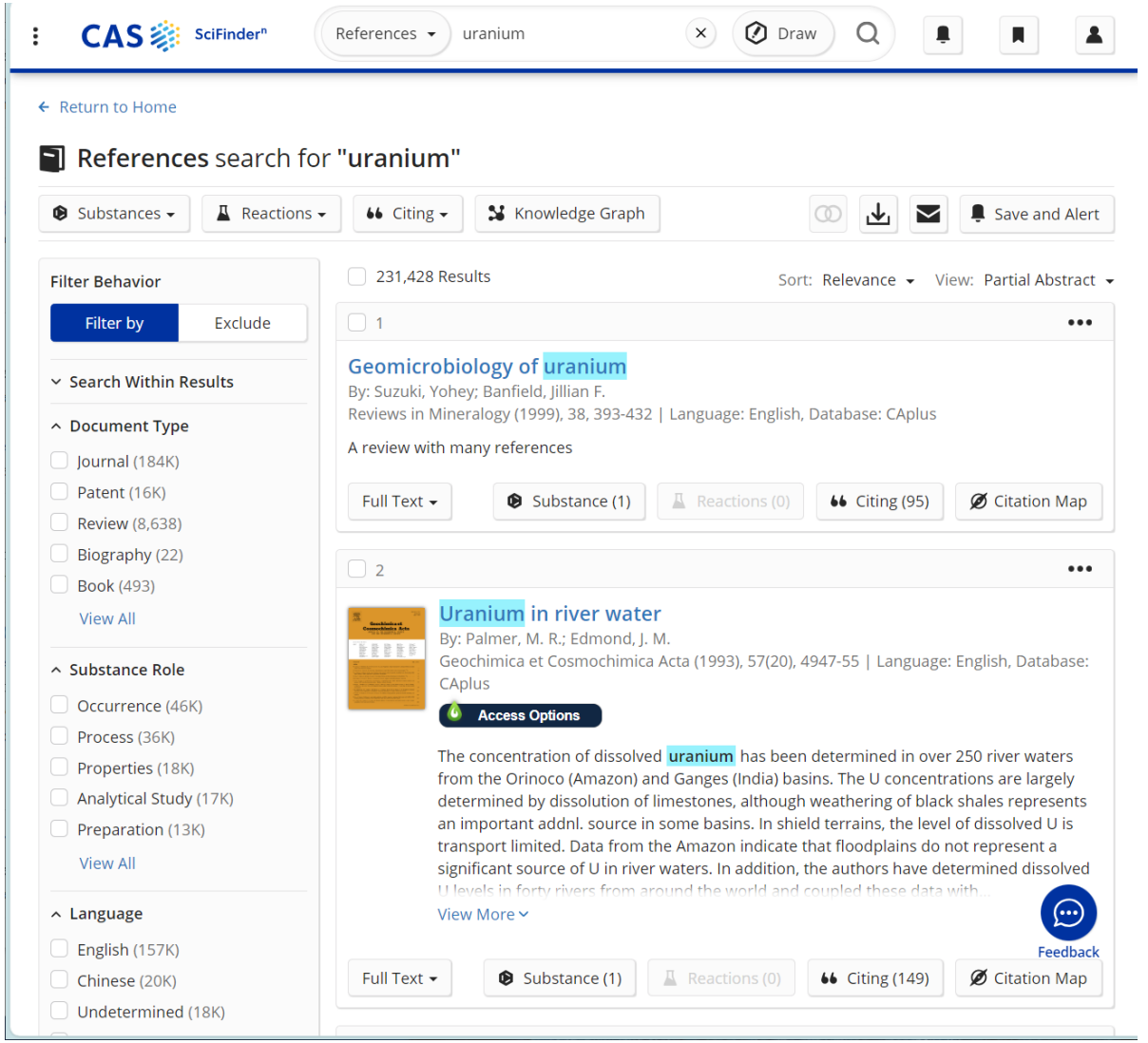
Substance Role
Uses (57K)
Properties (13K)
Process (9,128)
Preparation (4,852)
Reactant or Reagent (3,353)
View All
Language
Chinese (140K)
Japanese (60K)
English (38K)
Filtering:
Document Type: Patent
Clear All Filters
273,338 Results
Sort: Relevance
View: Partial Abstract
1
Protective lithium ion conducting ceramic coating for lithium metal anodes
By: Bates, John B.
United States, US5314765 A 1994-05-24 | Language: English, Database: CAplus
In a battery including a cathode, a lithium anode and an electrolyte between the anode and cathode, a thin-film of lithium phosphorus oxynitride is used to coat the anode and sep. it from the electrolyte. A preliminary layer of lithium nitride may be coated on the anode before the lithium phosphorous oxynitride is coated on the anode so that separation of the anode and electrolyte is further enhanced. By coating the lithium anode with this material lay-up, the life of the battery is lengthened and the performance of the battery is enhanced.
PatentPak
Full Text
Substances (3)
Reactions (0)
Citing (67)
Citation Map
2
Cathodes for secondary lithium-sulfur batteries
By: Chu, May-ying
United States, US5686201 A 1997-11-11 | Language: English, Database: CAplus
The cathodes are active S-based composite cathodes including active S, an electronic conductor mixed with the active S so that electrons can move between the active S and the electronic conductor, and an ionic conductor mixed with the active S so that ions can move between the ionic conductor and the active S. These materials are provided in a manner allowing ≥10% of the active-S to be available for
Feedback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References
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References search for "uranium"
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Geomicrobiology of uranium
By: Suzuki, Yohey; Banfield, Jillian F.
Reviews in Mineralogy (1999), 38, 393-432 | Language: English, Database: CAplus
A review with many references
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2
Uranium in river water
By: Palmer, M. R.; Edmond, J. M.
Geochimica et Cosmochimica Acta (1993), 57(20), 4947-55 | Language: English, Database: CAplus
Access Options
The concentration of dissolved uranium has been determined in over 250 river waters from the Orinoco (Amazon) and Ganges (India) basins. The U concentrations are largely determined by dissolution of limestones, although weathering of black shales represents an important addnl. source in some basins. In shield terrains, the level of dissolved U is transport limited. Data from the Amazon indicate that floodplains do not represent a significant source of U in river waters. In addition, the authors have determined dissolved
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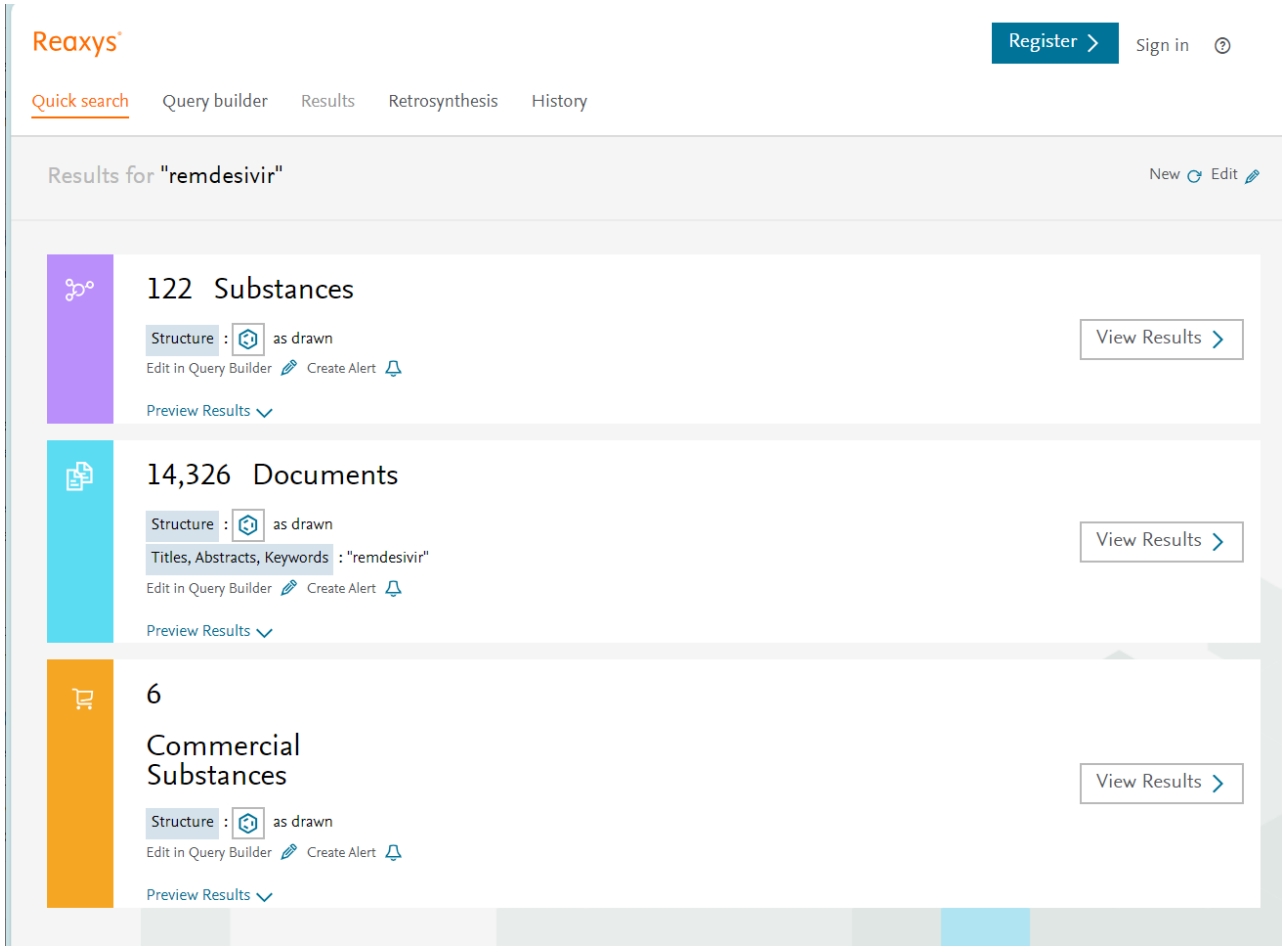

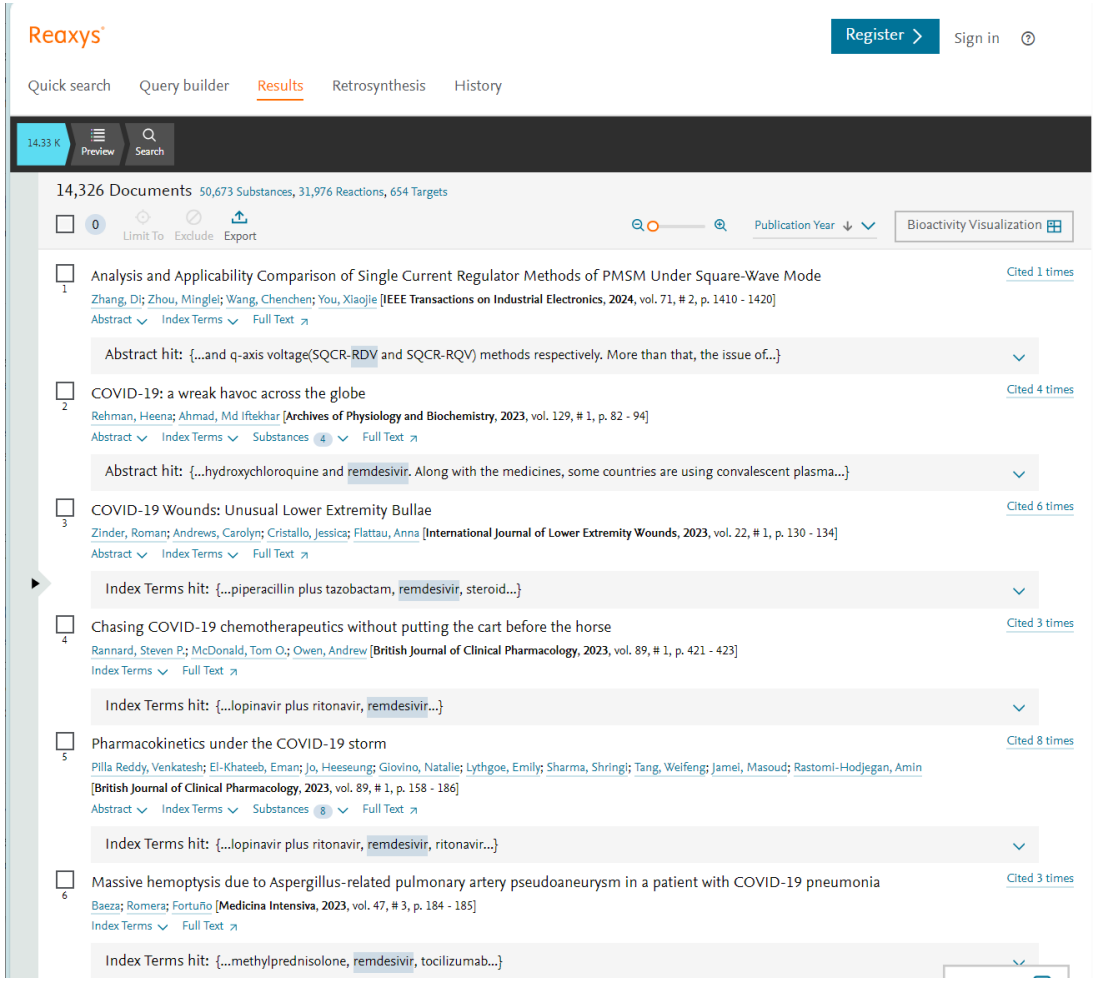

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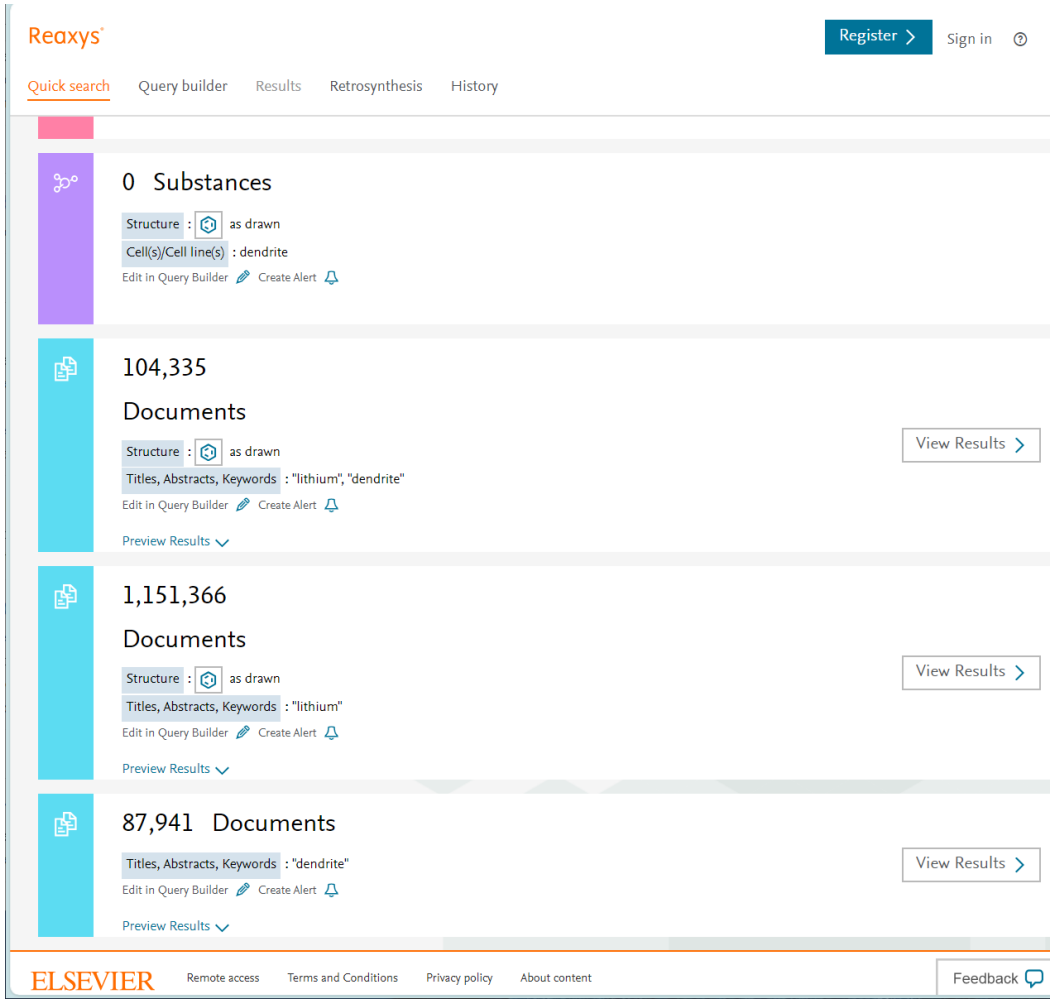
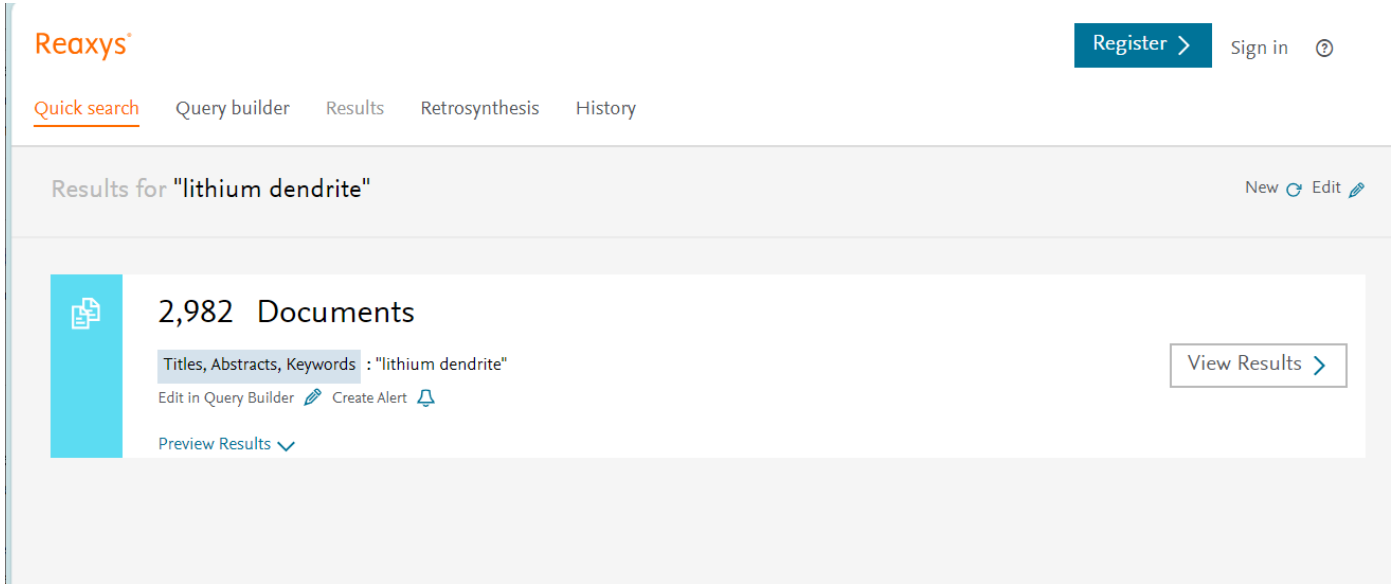
Lithium Dendrite v. “Lithium Dendrite”





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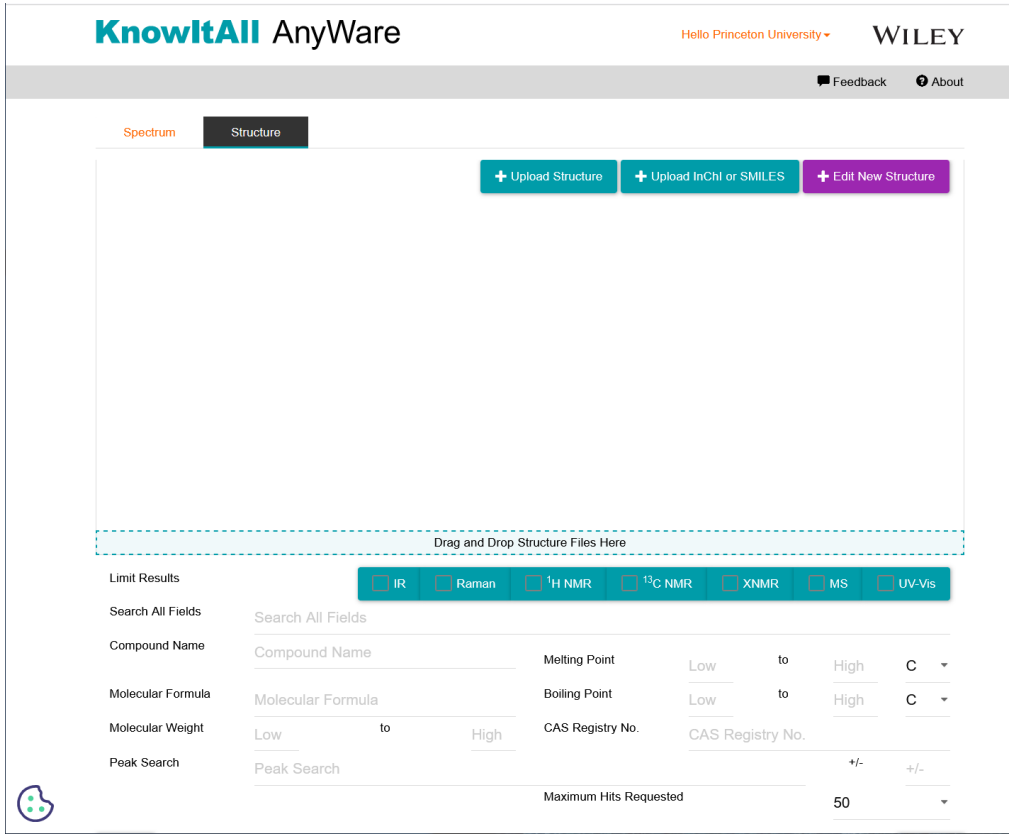

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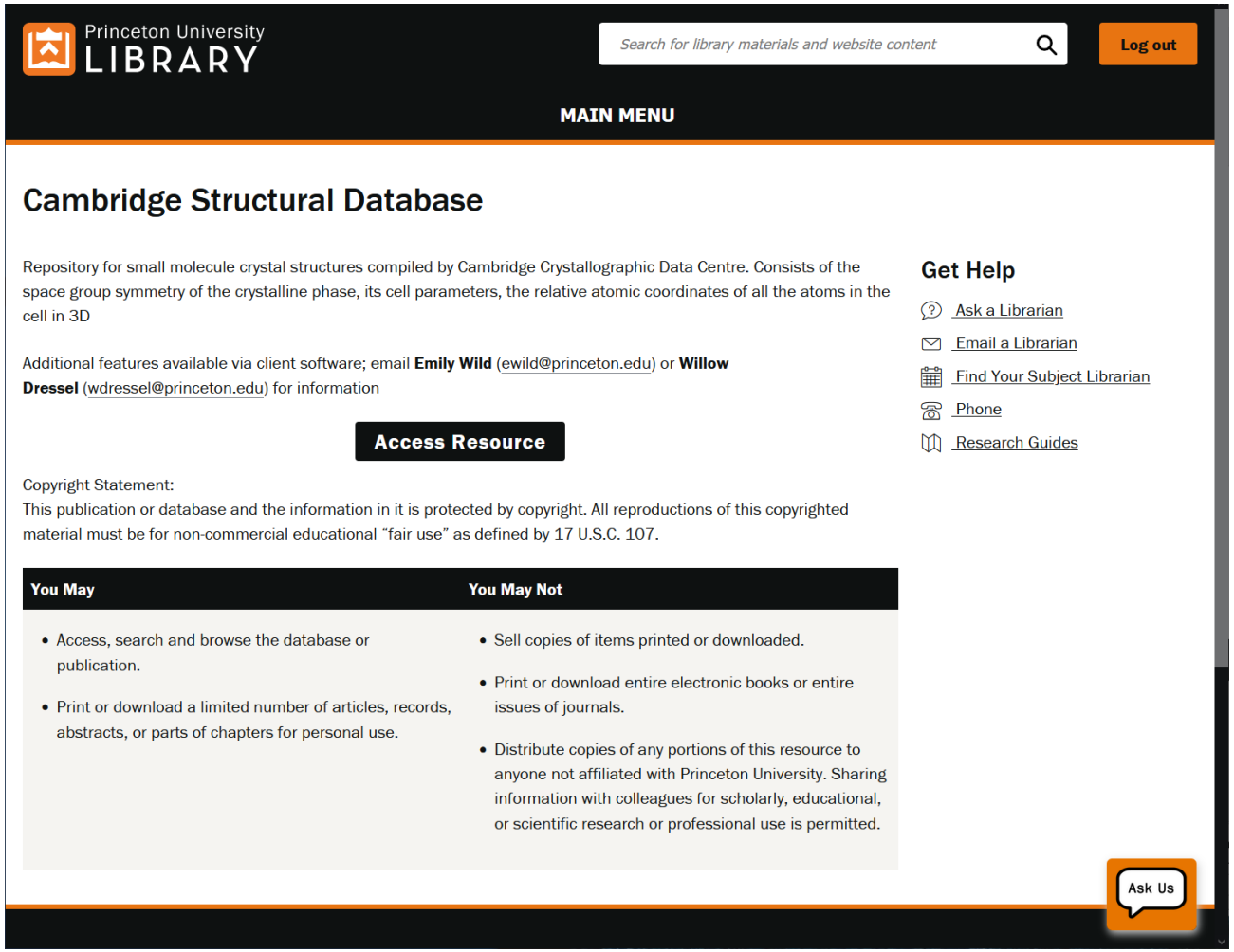

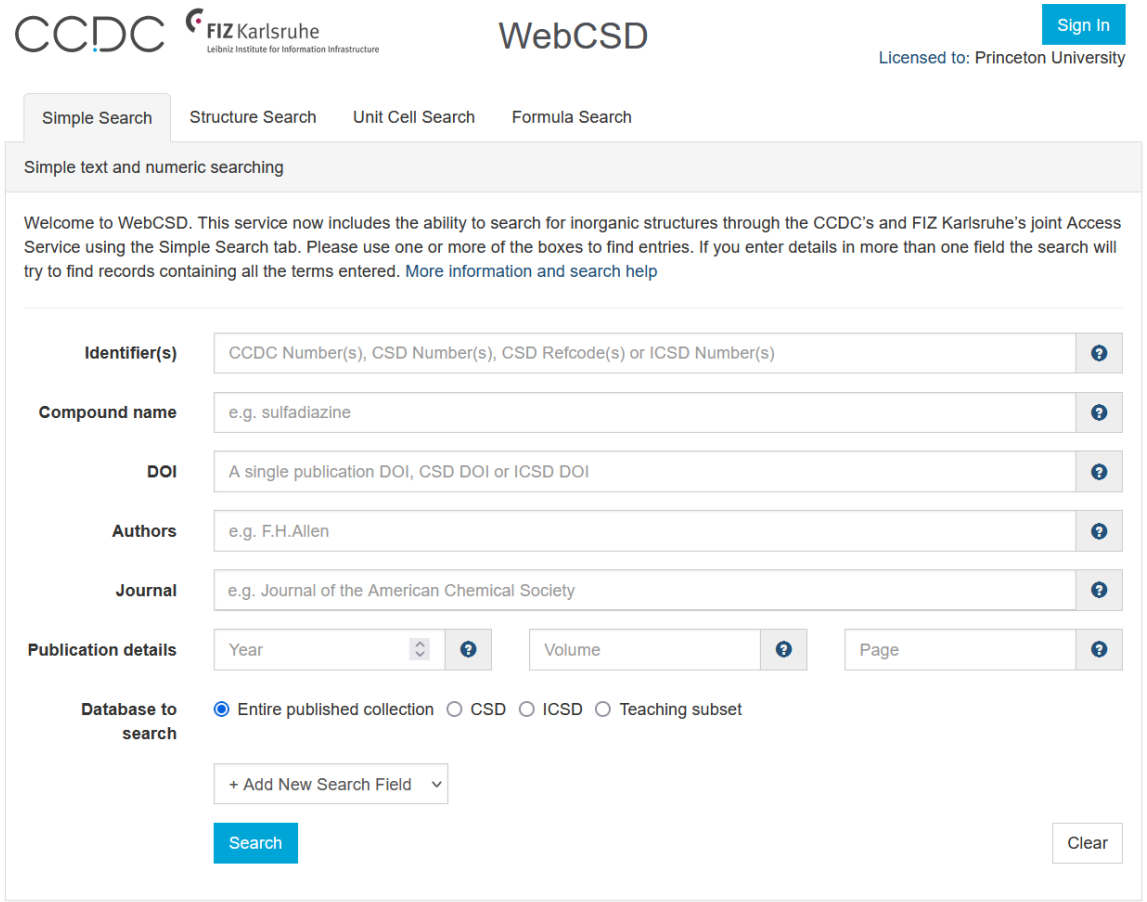

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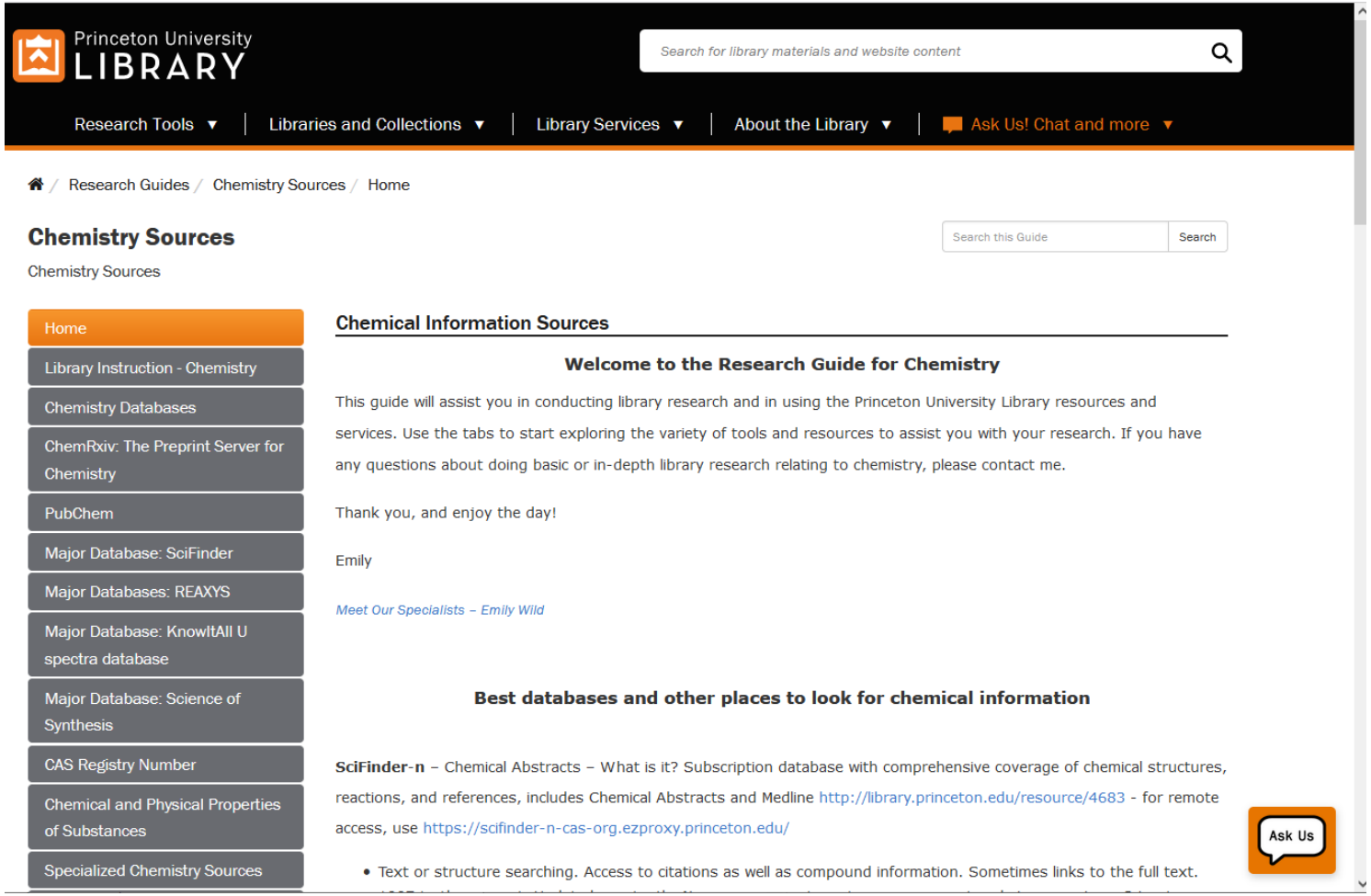

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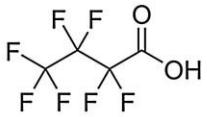
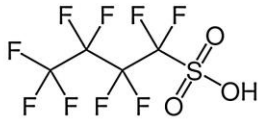
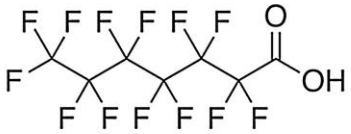
My most frequently asked chemistry questions = PFAS (left) and Elements (right)



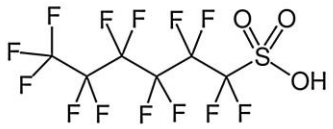
Perfluorobutanoic acid (PFBA)



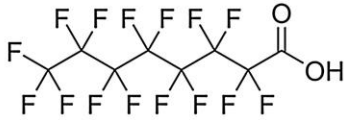
Perfluorobutane sulfonic acid (PFBS)



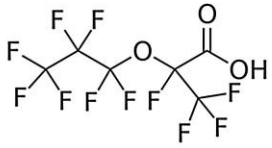
Perfluoroheptanoic acid (PFHpA)



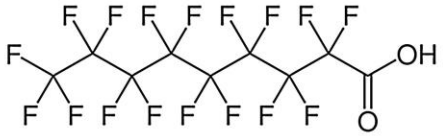
Perfluorohexane sulfonic acid (PFHxS)



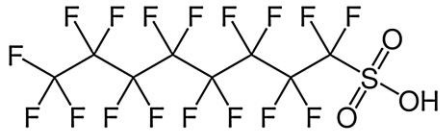
Perfluorooctanoic acid (PFOA)



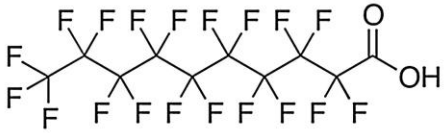
Hexafluoropropylene oxide dimer acid (HFPO-DA or GenX)



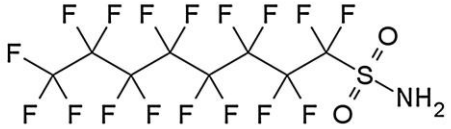
Perfluorononancanoic acid (PFNA)



Perfluorooctane sulfonic acid (PFOS)



Perfluorodecanoic acid (PFDA)



Perfluorooctanesulfonamide (PFOSA)

IUPAC Periodic Table of the Elements

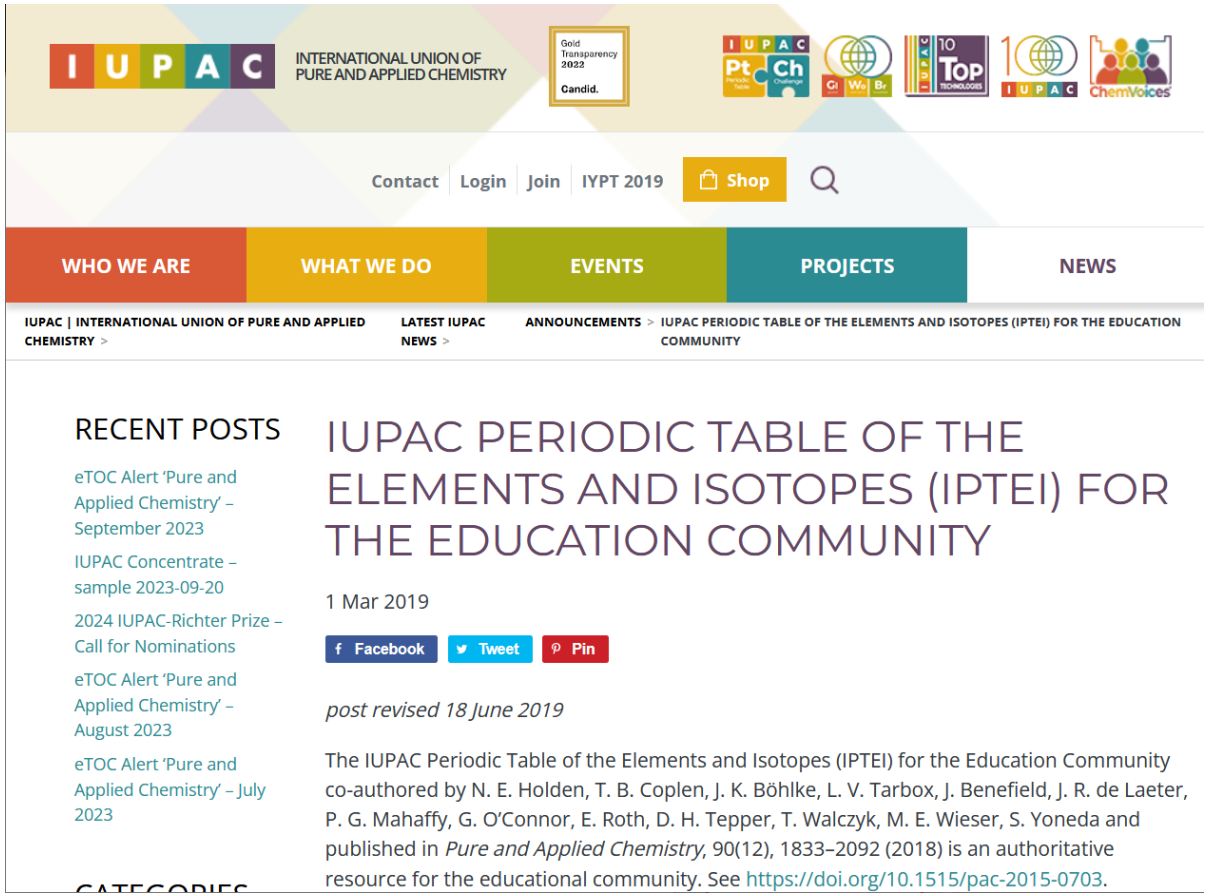
1																	18
1 H hydrogen 1.00080 ±0.00002	2		Key: atomic number Symbol name abridged standard atomic weight									13	14	15	16	17	2 He helium 4.00026 ±0.00001
3 Li lithium 6.94 ±0.06	4 Be beryllium 9.0122 ±0.0001											5 B boron 10.81 ±0.02	6 C carbon 12.011 ±0.002	7 N nitrogen 14.007 ±0.001	8 O oxygen 15.999 ±0.001	9 F fluorine 18.998 ±0.001	10 Ne neon 20.180 ±0.001
11 Na sodium 22.990 ±0.001	12 Mg magnesium 24.305 ±0.002	3	4	5	6	7	8	9	10	11	12	13 Al aluminium 26.982 ±0.001	14 Si silicon 28.085 ±0.001	15 P phosphorus 30.974 ±0.001	16 S sulfur 32.06 ±0.02	17 Cl chlorine 35.45 ±0.01	18 Ar argon 39.95 ±0.16
19 K potassium 39.098 ±0.001	20 Ca calcium 40.078 ±0.004	21 Sc scandium 44.956 ±0.001	22 Ti titanium 47.867 ±0.001	23 V vanadium 50.942 ±0.001	24 Cr chromium 51.996 ±0.001	25 Mn manganese 54.938 ±0.001	26 Fe iron 55.845 ±0.002	27 Co cobalt 58.933 ±0.001	28 Ni nickel 58.693 ±0.001	29 Cu copper 63.546 ±0.003	30 Zn zinc 65.38 ±0.02	31 Ga gallium 69.723 ±0.001	32 Ge germanium 72.630 ±0.008	33 As arsenic 74.922 ±0.001	34 Se selenium 78.971 ±0.008	35 Br bromine 79.904 ±0.003	36 Kr krypton 83.798 ±0.002
37 Rb rubidium 85.468 ±0.001	38 Sr strontium 87.62 ±0.01	39 Y yttrium 88.906 ±0.001	40 Zr zirconium 91.224 ±0.002	41 Nb niobium 92.906 ±0.001	42 Mo molybdenum 95.95 ±0.01	43 Tc technetium [97]	44 Ru ruthenium 101.07 ±0.02	45 Rh rhodium 102.91 ±0.01	46 Pd palladium 106.42 ±0.01	47 Ag silver 107.87 ±0.01	48 Cd cadmium 112.41 ±0.01	49 In indium 114.82 ±0.01	50 Sn tin 118.71 ±0.01	51 Sb antimony 121.76 ±0.01	52 Te tellurium 127.60 ±0.03	53 I iodine 126.90 ±0.01	54 Xe xenon 131.29 ±0.01
55 Cs caesium 132.91 ±0.01	56 Ba barium 137.33 ±0.01	57-71 lanthanoids	72 Hf hafnium 178.49 ±0.01	73 Ta tantalum 180.95 ±0.01	74 W tungsten 183.84 ±0.01	75 Re rhenium 186.21 ±0.01	76 Os osmium 190.23 ±0.03	77 Ir iridium 192.22 ±0.01	78 Pt platinum 195.08 ±0.02	79 Au gold 196.97 ±0.01	80 Hg mercury 200.59 ±0.01	81 Tl thallium 204.38 ±0.01	82 Pb lead 207.2 ±1.1	83 Bi bismuth 208.98 ±0.01	84 Po polonium [209]	85 At astatine [210]	86 Rn radon [222]
87 Fr francium [223]	88 Ra radium [226]	89-103 actinoids	104 Rf rutherfordium [267]	105 Db dubnium [268]	106 Sg seaborgium [269]	107 Bh bohrium [270]	108 Hs hassium [269]	109 Mt meitnerium [277]	110 Ds darmstadtium [281]	111 Rg roentgenium [282]	112 Cn copernicium [285]	113 Nh nihonium [286]	114 Fl flerovium [290]	115 Mc moscovium [290]	116 Lv livermorium [293]	117 Ts tennessine [294]	118 Og oganesson [294]

57 La lanthanum 138.91 ±0.01	58 Ce cerium 140.12 ±0.01	59 Pr praseodymium 140.91 ±0.01	60 Nd neodymium 144.24 ±0.01	61 Pm promethium [145]	62 Sm samarium 150.36 ±0.02	63 Eu europium 151.96 ±0.01	64 Gd gadolinium 157.25 ±0.03	65 Tb terbium 158.93 ±0.01	66 Dy dysprosium 162.50 ±0.01	67 Ho holmium 164.93 ±0.01	68 Er erbium 167.26 ±0.01	69 Tm thulium 168.93 ±0.01	70 Yb ytterbium 173.05 ±0.02	71 Lu lutetium 174.97 ±0.01
89 Ac actinium [227]	90 Th thorium 232.04 ±0.01	91 Pa protactinium 231.04 ±0.01	92 U uranium 238.03 ±0.01	93 Np neptunium [237]	94 Pu plutonium [244]	95 Am americium [243]	96 Cm curium [247]	97 Bk berkelium [247]	98 Cf californium [251]	99 Es einsteinium [252]	100 Fm fermium [257]	101 Md mendelevium [258]	102 No nobelium [259]	103 Lr lawrencium [262]

INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY

For notes and updates to this table, see www.iupac.org. This version is dated 4 May 2022.

https://iupac.org/what-we-do/periodic-table-of-elements/



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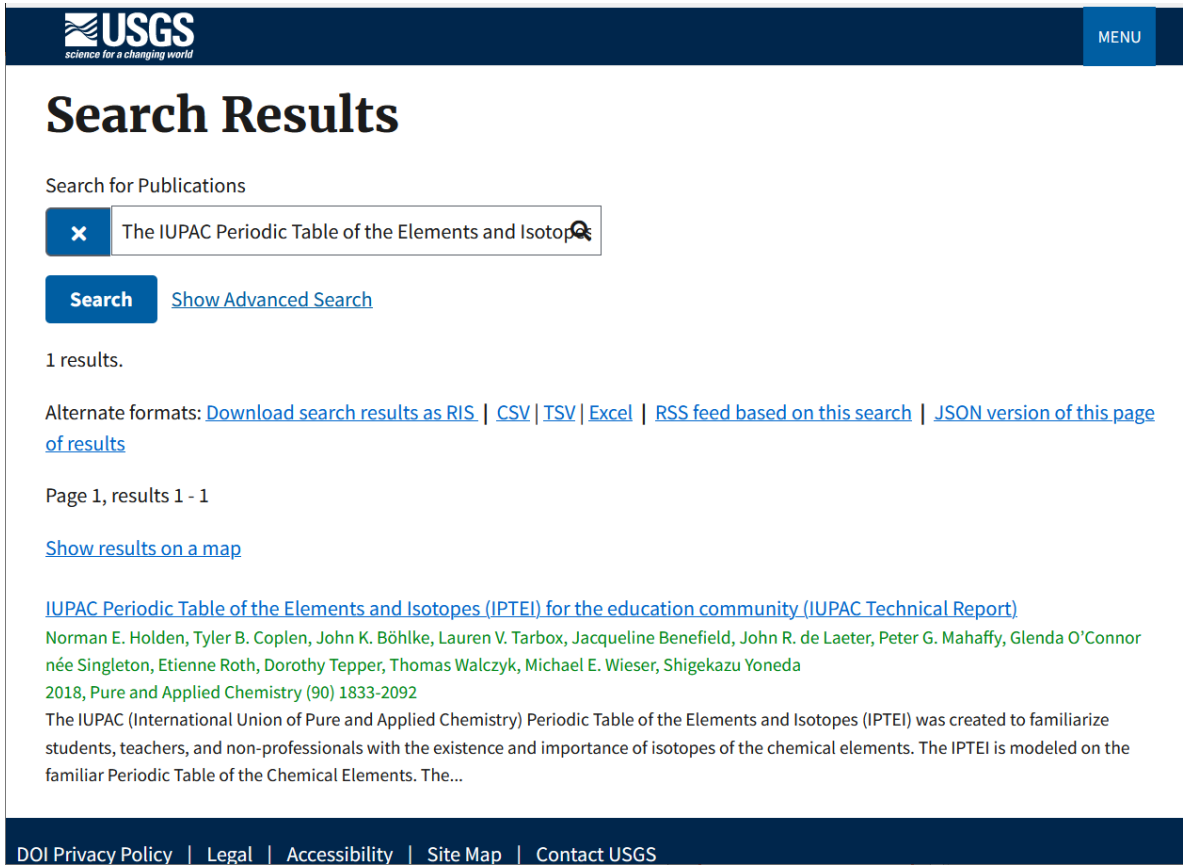

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U.S. Geological Survey (USGS)

https://www.usgs.gov/news/national-news-release/tap-water-study-detects-pfas-forever-chemicals-across-us

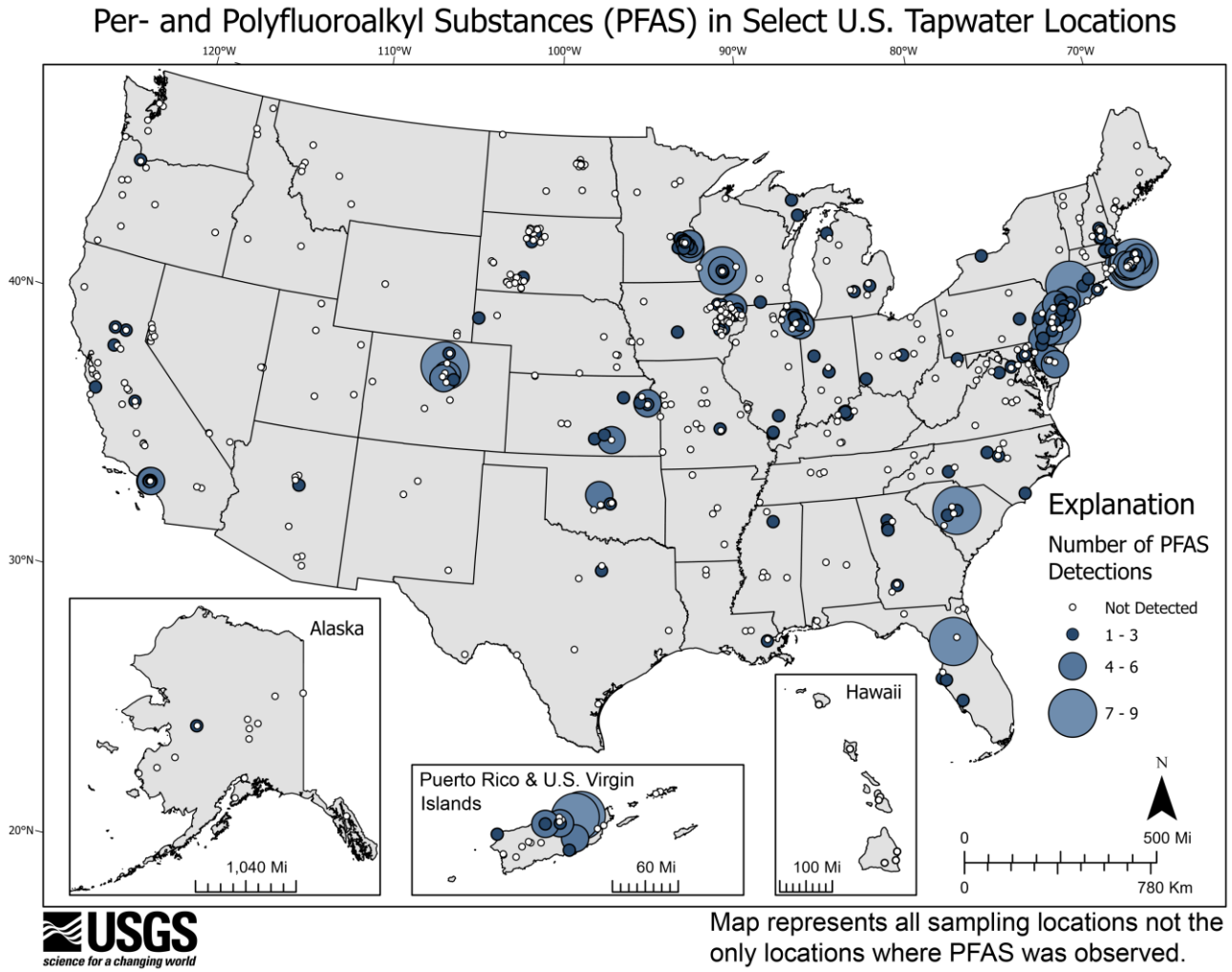
Emily was the USGS Hydrologist that pulled all the data for Leslie this publication
https://pubs.usgs.gov/publication/sir20085227

Emily was the USGS Librarian (Physical Scientist for this publication

"Nationwide reconnaissance of contaminants of emerging concern in source and treated drinking waters of the United States"
https://pubs.usgs.gov/publication/70189315

Recommended reading

"Contaminants of emerging concern in the environment: Where we have been and what does the future hold?"
https://pubs.usgs.gov/publication/70201175



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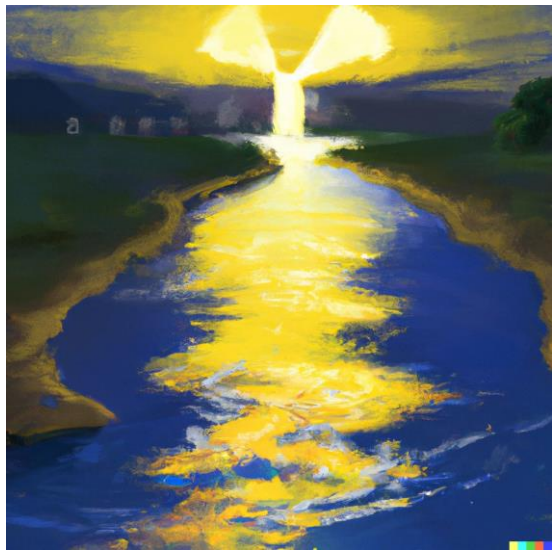

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