European Twinning for research in Solar energy to (2) water (H2O) production and treatment technologies GA Number: 101079305 European Research Executive Agency REA.C3

Funded by the European Union

Fast Track School #1

Solar-driven water production & water treatment technologies and brine treatment processes

PALERMO, 10-11.01.2024

Sol2H2O

Isabel Oller Alberola; ioller@psa.es

Solar water treatment, integration of technologies for WW regeneration

PALERMO, 10-11.01.2024

Outline

Introduction & Motivation

New EU regulation on water reclamation

AOPs & photochemical processes

Simultaneous water decontamination & disinfection

Key messages

Introduction and Motivation

Exponential increase of population: The world population has already crossed the 8 billion mark at the end of 2023 and it is expected to increase to 9.7 billions in 2050

Where Water Stress Will Be Highest by 2040

Projected ratio of water withdrawals to water supply (water stress level) in 2040

Source: World Resources Institute via The Economist Intelligence Unit

Introduction and Motivation

Globally, 70% of Freshwater is Used for Agriculture

Share of freshwater withdrawals by sector (%) in 2014

World Development Indicator

40 %

By **2050**, feeding a planet of **9 billion people** will require an estimated 50% increase in agricultural production and a 15% increase in water withdrawals.

of the energy expenditures in small to medium-sized municipalities **go to pumping-moving-treating** of the urban water needed

> Forecasts of global water demand in 2050 show a **400%** increase for the manufacturing industry and a **140%** increase for thermal power generation (OECD, 2012).

> Water withdrawals for industrial use will have doubled by 2030, reaching a percentage of **22%** worldwide.

Introduction and Motivation: water reclamation, a global need

It is estimated an annual world production of **urban wastewater of 380 km**³, that is 15% of water withdrawal for agriculture (42 millions ha).

World urban wastewater production is estimated to increase 24% in 2030 and 51% in 2050.

Nutrients in urban wastewaters: 16.6 Tg (Tg = million of metric tons) of nitrogen; 3 Tg for phosphorous and 6.3 Tq for potassium. Total recovery of nutrients from urban wastewaters would compensate the 13.4% of the world demand for agriculture.

Introduction and Motivation: water reclamation, a global need

Benefits

- \triangleright It can improve the status of the environment both quantitatively, alleviating pressure by substituting abstraction, and qualitatively, relieving pressure of discharge from UWWTP to sensitive areas.
- Environmental
- Social
- Economic

- \triangleright Appropriate consideration for nutrients -in treated wastewater could also reduce the use of additional fertilizers resulting in savings for the environment, farmers and wastewater treatment.
- It is considered a reliable water supply, quite independent from seasonal drought and weather variability and able to cover peaks of water demand.

 \triangleright Lower investment costs and energy compared to alternative sources such as desalination or water transfer, also contributing to reduce greenhouse gas emissions.

Introduction: Contaminants of emerging concern

Introduction: Microbial water contamination

New challenge in UWW: Antibiotic Resistant Bacteria (ARB)

Definition: *Phenomenon by which a microorganism is no longer affected by an antimicrobial (previously sensitive), so that the usual medical treatments become ineffective and infections persist and can be transmitted to other people. WHO, Antimicrobial Resistance*

[http://www.who.int/medicines/publications/global-priority-list-antibiotic-resista](http://www.who.int/medicines/publications/global-priority-list-antibiotic-resistant-bacteria/en/) [nt-bacteria/en/](http://www.who.int/medicines/publications/global-priority-list-antibiotic-resistant-bacteria/en/)

Outline

Reclamation of Wastewater: CECs and OMCs

highest potential for uptake

lowest potential for uptake

by plants

by plants

OMCs translocation to plants and soil detection

^a Concentrations are given in µg/kg.

L.M. Madikizela et al. / Science of the Total Environment 636 (2018) 477-486

Fig. 2. Heat map showing the potential of the main crop species for CECs uptake. The highest potential for uptake is indicated with dark red; the lowest potential with dark green.

A. Christou et al. Environmental Research 170 (2019) 422-432

-

Reclamation of Wastewater: Pathogens

Regrowth during storage

Analysis of bacteria concentration in treated UWW after 24 h of storage.

New EC regulation on water reuse

New EC regulation on water reuse

Table 4 Validation monitoring of reclaimed water for agricultural irrigation

(*) The reference pathogens *Campylobacter*, Rotavirus and *Cryptosporidium* can also be used for validation monitoring purposes instead of the proposed indicator microorganisms. The following log_{10} reduction performance targets should then apply: *Campylobacter* (\geq 5.0), Rotavirus (\geq 6.0) and Cryptosporidium (≥ 5.0).

(**) Total coliphages is selected as the most appropriate viral indicator. However, if analysis of total coliphages is not feasible, at least one of them (F-specific or somatic coliphages) has to be analyzed.

(***) Clostridium perfringens spores is selected as the most appropriate protozoa indicator. However sporeforming sulfate-reducing bacteria is an alternative if the concentration of Clostridium perfringens spores does not allow to validate the requested log10 removal.

> Also risk assessment is required, but it's not currently defined

Outline

Solar treatments for water disinfection & decontamination

Advanced Oxidation Processes

TiO₂/UVA

(Carey et al., 1976) $TiO, \frac{hv}{m} \rightarrow TiO, (e^- + h^+)$ $h^+ + H_2O \rightarrow^{\bullet} O H + H^+$ $e^-+O, \rightarrow O^{\bullet-}_{2}$

Photo-Fenton

$$
Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + OH^- + {}^{\bullet}OH
$$

(J. Chem. Soc., 1894)

$$
Fe^{3+} + H_2O \xrightarrow{hv} Fe^{2+} + H^+ + {}^{\bullet}OH
$$

(several authors, early 90s)

- \triangleright H₂O₂
- \triangleright Persulfate
- \triangleright Peroxymonosulfate
- \triangleright Peracetic acid
- \triangleright Chlorine

- > No generation of HO by solar wavelengths reaching the Earth's surface
- > High efficiency on water microbial inactivation
- \triangleright Direct oxidation and complex microbial inactivation mechanisms

Ozone for water treatment

Disadvantages:

- **• Solubility and activity:** When the ozone dosage is too low, some germs and bacteria may survive, which is why higher ozone concentrations are used. If the water being treated is rich in organic compounds or total suspended solids, then ozone will decay more rapidly, as it reacts with these contaminants.
- **• High costs:** During corona discharge, around 85% of the energy is lost via heat waste, making ozone treatment extremely energy-intensive. Treating water with ozone is also energy-intensive because it requires high-class equipment, expensive technology, and an operator that knows how to work the complicated system.
- **• Reactivity and toxicity:** Ozone's reactivity with metals can cause issues in wastewater treatment pipes and containers, therefore corrosion-resistant materials such as stainless steel must be used, which adds to plant construction costs. Because of the high toxicity, ozone levels need to be constantly monitored.
- **• Reactivity and byproducts:** If the water contains bromide ions, ozone can react to form brominated compounds (like bromate ions/salts), which can cause human cancers. ozone leaves nothing behind; any ozone that doesn't react with the water contaminants is immediately broken down, leaving nothing to monitor after the disinfection process.

Photochemical processes: UV-C with oxidative

agents

(Goldstein et al., 2007)

 H_2O_2

Chlorine

(Chuang et al., Environ Sci Technol, 2017. 51(23) 13859-13868.)

 $0^{\bullet-}$ + $H_2O \rightarrow HO^{\bullet}$ + OH^-

Persulfate / Peroxymonosulfate

$$
S_2O_8^{2-} \xrightarrow{UVC} 2SO_4^-\longrightarrow BSO_4^-\longrightarrow HSO_5^- \xrightarrow{UVC} SO_4^-\longrightarrow HO^+\longrightarrow SO_4^{2-} + H^+ \qquad HSO_5^- + H^+ \rightarrow SO_4^-\longrightarrow H_{2}O
$$

(Sánchez-Montes et al. Environ, Sci.: Water Res. Technol., 2020, 6, 2553-2566) (Guerra-Rodríguez et al., Water 2018, 10, 1828)

- SO4⁻⁻ has a high oxidation potential (2.5-3.1 V) comparable or even higher than 'OH
- Reacts more selectively and efficiently with organic compounds that contain unsaturated bonds or $aromatic\pi electrons$
- \triangleright Half-life of sulfate radicals is supposed to be 30–40 µs
- Chlorine radicals oxidation potential 1.5V. High half-life

Solar Heterogeneous photocatalysis: TiO₂

Linearly dependent on the energy flux

but only \sim 5% of the whole solar

spectrum is available for $TiO₂$ band-gap.

♦ 75% of solar collector efficiency and 1% for the catalyst means 0.04% original solar photons are efficiently used. This is a rather inefficient process.

Mild catalyst working under mild conditions with mild oxidants.

- ♦ As concentration of contaminants and water ionic strength increase: slow kinetics and unpredictable mechanisms need to be solved.
- \blacklozenge TiO₂ efficiency improved with the addition of powerful oxidants or when doped (with iron, nitrogen...) to undertake practical applications.
- Pure $TiO₂$ can utilize only UV and new catalysts able to work with the visible component of \bullet the solar spectrum are needed.

Solar photocatalysis: enhancement strategies

DOPED TIO, & OTHER SEMICONDUCTORS

IMMOBILIZATION OF THE CATALYST IN SUBSTRATES

Uncoated external. $TiO₂$ coated internal

Lower microbial reduction efficiency than with suspended catalvst

Scaling-up challenge

Dehaya et al., Applied Catalysis B: Environmental 128 (2012) 126-134

Photochemical processes

Outline

Simultaneous water decontamination & disinfection

Simultaneous removal of CECs and Pathogens

Ozonation pilot plant

Science of the Total Environment 787 (2021) 147531 Science of the Total Environment 766 (2021) 144320

PANIWATER project

Assess the **efficiency of ozonation** for the treatment of **urban wastewater** (UWW), with three different O_3 doses. The parameters for the evaluation of the treatment were degradation of organic microcontaminants (OMCs), disinfection, generation of disinfection by-products (DBPs) and toxicity.

Municipal Urban Wastewater treatment plant "El Bobar" located in Almeria. South East of Spain.

- 0.2 gO₃/h L (25%). \bullet
	- 0.3 gO₃/h L (50%). \bullet
	- 0.4 gO₃/h L(100%). \cdot \bullet
- OMCs:UHPLC-QqQ-MS/MS. .
- DBPs:HS-GC-QqQ-MS/MS.
	- Bromate and bromide: IC. .
- Escherichia coli and Coliforms.
- Plate counting Chromocult[®] agar media.
-
- Cytotoxicity.
- Phytotoxicity.
- Bacteria survival.

Microcontaminants

SFX, TMP, DCF, ACE, CBZ. CAF

PANIWATER project

DEGRADATION OF OMCs

BROMATE

DISINFECTION

BROMOFORM

HALOACETIC ACIDS (HAAs)

Chlorinated HAAs = Constant

Brominated HAAs = Increased

BACTERIAL SURVIVAL

Outline

Key messages

- Water scarcity and bad water quality are problems affecting all over the world, which makes it crucial to find alternative water sources, such as municipal wastewater. Municipal wastewater treatment, jointly with desalination, mean a key strategy for trying to maintain high human life quality.
- ⬥ A deep evaluation on the specific problem to be solved must be done just to focus on the optimum treatment option.
- ⬥ Normally, different AOPs based technologies show highly efficiency as tertiary treatment for CECs & ARB elimination, but economic, health and life cycle assessments must support the final selection.
- Solar based AOPs are considered a sustainable and actual viable option for reducing contaminant impact on the Environment.
- ⬥ Water quality parameters monitoring as well as contaminant transfer to crops must be carried out for ensuring a "safe reuse". Risk Assessment studies are required

Sol2H20

Isabel Oller Alberola; ioller@psa.es

Prof. Sixto Malato

Dra. Isabel Oller

Polo López

Dra. Ana Ruiz Delgado

Dra. Samira Nahím Granados

Dra. M. Jesús Abeledo Lameiro

Alba Hernández Joyce Villachica (PhD student) Zanoletty (PhD student)

Kelly Castañeda (PhD student)

Paula Serrano Tarí (PhD student)

Isabel Espinoza

Elisa Ramos Carrión **Antonio Martos** Pavón (PhD student) (Técnico de laboratorio) (Técnico de planta)

Francisco Expósito (Técnico de planta)

