European Twinning for research in Solar energy to (2) water (H2O) production and treatment technologies

GA Number: 101079305

European Research Executive Agency REA.C3

Sol2H20











Fast Track School #1

Solar-driven water production & water treatment technologies and brine treatment processes

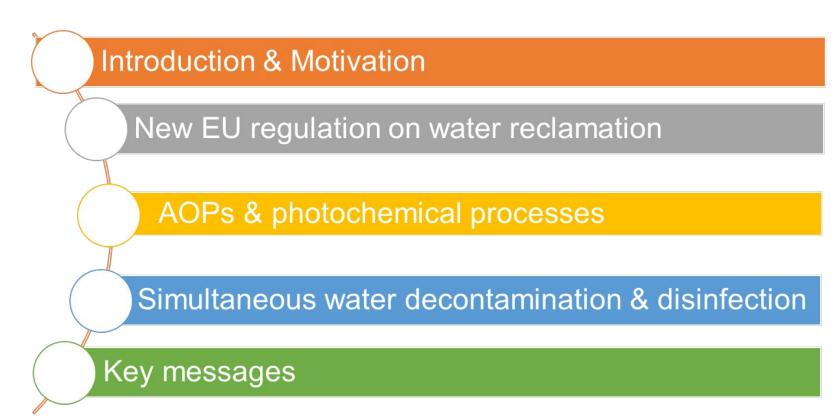




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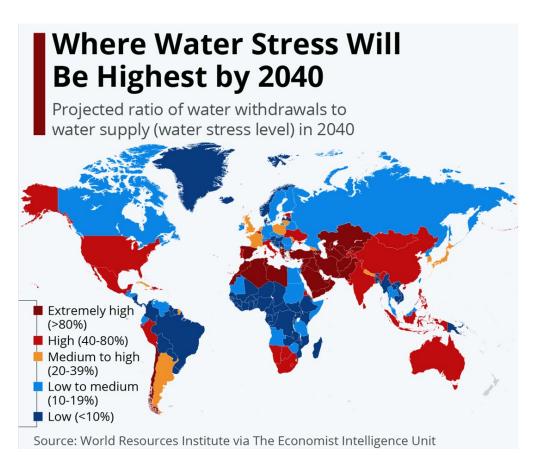
Solar water treatment, integration of technologies for WW regeneration

Outline



Introduction and Motivation

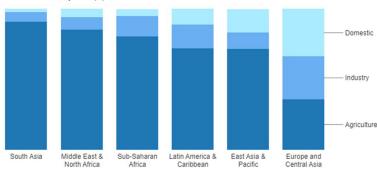
Exponential increase of population:
The world population has already crossed the 8 billion mark at the end of 2023 and it is expected to increase to 9.7 billions in 2050



Introduction and Motivation

Globally, 70% of Freshwater is Used for Agriculture

Share of freshwater withdrawals by sector (%) in 2014



By **2050**, feeding a planet of **9 billion people** will require an estimated 50% increase in agricultural production and a 15% increase in water withdrawals.



Source: World Development Indicator



40 %

of the energy expenditures in small to medium-sized municipalities **go to pumping-moving-treating** of the urban water needed

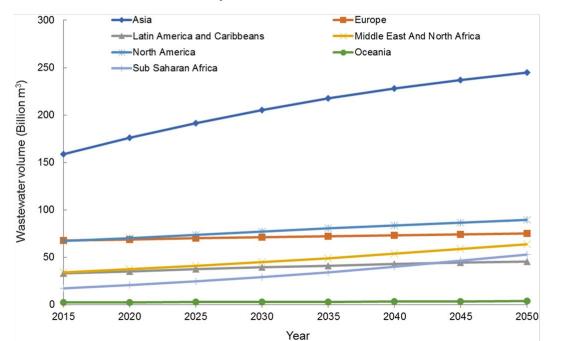
Forecasts of global water demand in 2050 show a 400% increase for the manufacturing industry and a 140% increase for thermal power generation (OECD, 2012).

Water withdrawals for industrial use will have doubled by 2030, reaching a percentage of 22% worldwide.

Introduction and Motivation: water reclamation, a global need

It is estimated an annual world production of **urban wastewater of 380 km**³, that is 15% of water withdrawal for agriculture (42 millions ha).

World urban wastewater production is estimated to increase 24% in 2030 and 51% in 2050.



Nutrients in urban wastewaters:
16.6 Tg (Tg = million of metric tons) of nitrogen;
3 Tg for phosphorous and 6.3 Tg for potassium.
Total recovery of nutrients from urban wastewaters would compensate the 13.4% of the world demand for agriculture.

Introduction and Motivation: water reclamation, a global need

Benefits

➤ It can improve the status of the environment both quantitatively, <u>alleviating pressure by substituting abstraction</u>, and qualitatively, <u>relieving pressure of discharge from UWWTP to sensitive</u> areas.

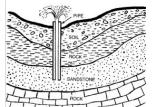
- ✓ Environmental
- ✓ Social
- ✓ Economic



- Appropriate consideration for <u>nutrients</u> in treated wastewater could also reduce the use of <u>additional fertilizers</u> resulting in savings for the environment, farmers and wastewater treatment.
- It is considered a <u>reliable water supply</u>, quite <u>independent</u> from seasonal drought and weather variability and able to cover peaks of water demand.



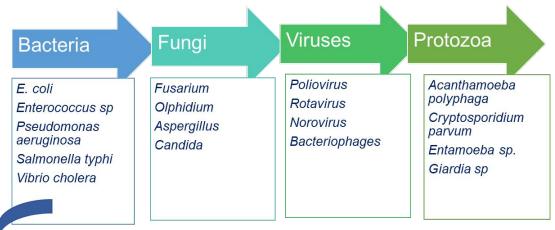
Lower investment costs and energy compared to alternative sources such as desalination or water transfer, also contributing to reduce greenhouse gas emissions.



Introduction: Contaminants of emerging concern



Introduction: Microbial water contamination



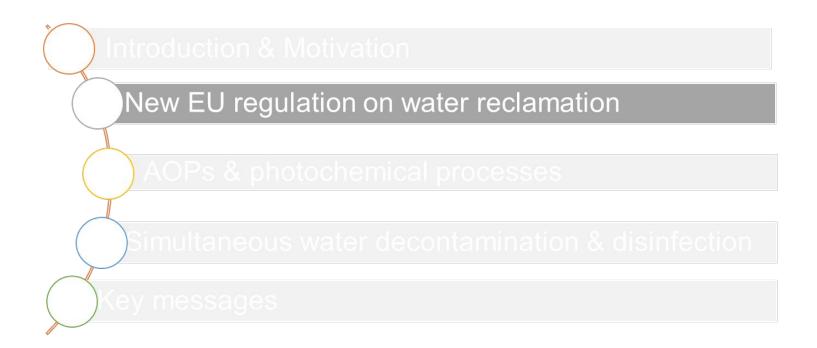
New challenge in UWW: Antibiotic Resistant Bacteria (ARB)

<u>Definition:</u> Phenomenon by which a microorganism is no longer affected by an antimicrobial (previously sensitive), so that the usual medical treatments become ineffective and infections persist and can be transmitted to other people. WHO, Antimicrobial Resistance

<u>http://www.who.int/medicines/publications/global-priority-list-antibiotic-rent-bacteria/en/</u>



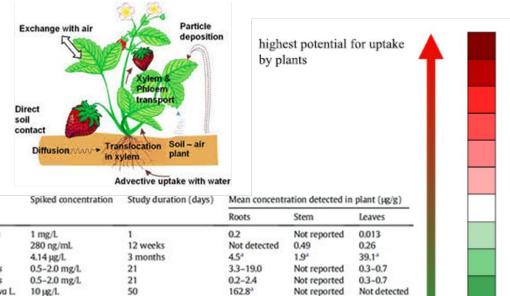
Outline



Reclamation of Wastewater: CECs and OMCs

OMCs translocation to plants and detection

Plant



	,	, , , , , , , , , , , , , , , , , , , ,	1000		
			Roots	Stem	Leaves
Typha latifolia	1 mg/L	1	0.2	Not reported	0.013
Brassicaceae	280 ng/ml.	12 weeks	Not detected	0.49	0.26
Cucumber	4.14 µg/L	3 months	4.5	1.94	39.1*
Scirpus validus	0.5-2.0 mg/L	21	3.3-19.0	Not reported	0.3-0.7
Scirpus validus	0.5-2.0 mg/L	21	0.2-2.4	Not reported	0.3-0.7
Medicago sativa L.	10 µg/L	50	162.83	Not reported	Not detected
Medicago sativa L.	10 µg/L	50	52.5*	Not reported	3.5*
Medicago sativa L.	10 µg/L	50	311.9	Not reported	23.5*
Medicago sativa L.	10 µg/L	50	28.9°	Not reported	28.3
	Brassicaceae Cucumber Scirpus validus Scirpus validus Medicago sativa L. Medicago sativa L. Medicago sativa L.	Brassicaceae 280 ng/ml. Cucumber 4.14 µg/L Scirpus validus 0.5-2.0 mg/L Scirpus validus 0.5-2.0 mg/L Medicago sativa L. 10 µg/L Medicago sativa L. 10 µg/L	Typha latifolia	Roots Roots Typha latifolia 1 mg/L 1 0.2	Roots Stem

Concentrations are given in µg/kg.

Pharmaceutical

lowest potential for uptake by plants

Fig. 2. Heat map showing the potential of the main crop species for CECs uptake. The highest potential for uptake is indicated with dark red; the lowest potential with dark green.

Crop Species

late-season potatoes

mid-season potatoes

spring potatoes

cucumber

tomatoes watermelons

melons pepper eggplant maize alfalfa peanuts haricot beans wheat

barley bananas walnut citrus and avocado fruit trees

pistachio

almonds table grapes

table olives

okra marrows

green beans

celery

spinach

lettuce cabbage

carrots radish

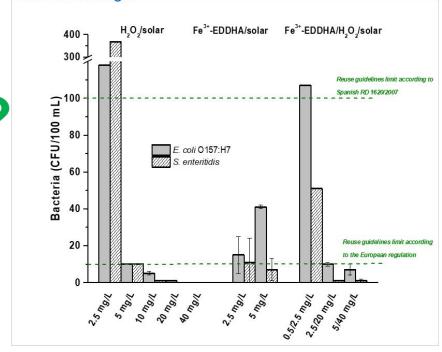
L.M. Madikizela et al. / Science of the Total Environment 636 (2018) 477–486

Reclamation of Wastewater: Pathogens

Regrowth during storage

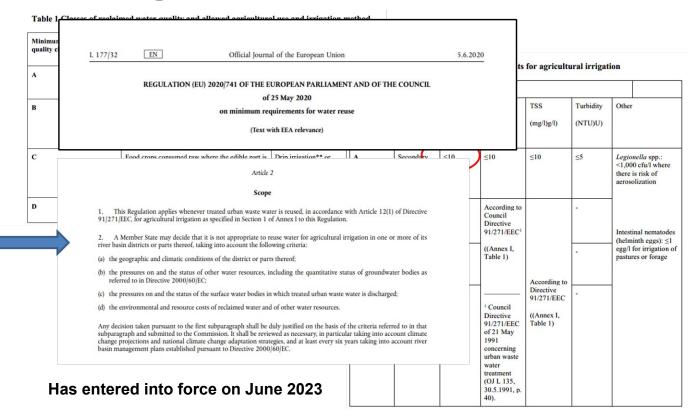
OR INHIBITION

Analysis of bacteria concentration in treated UWW after 24 h of storage.



S. Nahim-Granados, et al. Applied Catalysis B: Environmental 278 (2020) 119334

New EC regulation on water reuse



New EC regulation on water reuse

Table 4 Validation monitoring of reclaimed water for agricultural irrigation

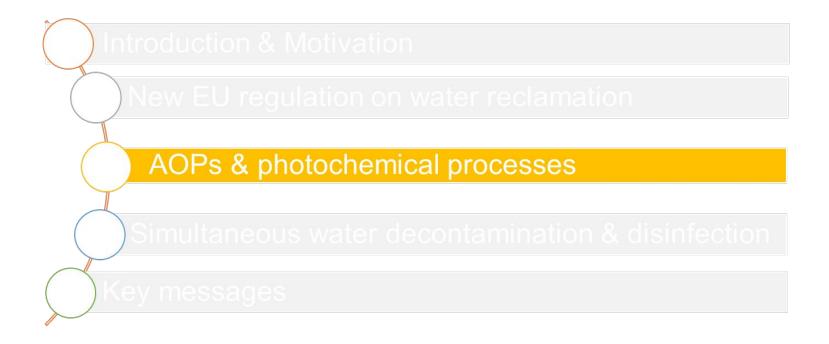
Reclaimed water quality class	Indicator microorganisms (*)	Performance targets for the treatment chain (log ₁₀ reduction)	
A	E. coli	≥ 5.0	
	Total coliphages/ F-specific coliphages/somatic coliphages/coliphages(**)	≥ 6.0	
	Clostridium perfringens spores/spore-forming sulfate-reducing bacteria(***)	≥ 4.0 (in case of Clostridium perfringens spores) ≥ 5.0 (in case of spore-forming sulfate-reducing bacteria)	



- (*) The reference pathogens *Campylobacter*, Rotavirus and *Cryptosporidium* can also be used for validation monitoring purposes instead of the proposed indicator microorganisms. The following \log_{10} reduction performance targets should then apply: *Campylobacter* (\geq 5.0), Rotavirus (\geq 6.0) and *Cryptosporidium* (\geq 5.0).
- (**) Total coliphages is selected as the most appropriate viral indicator. However, if analysis of total coliphages is not feasible, at least one of them (F-specific or somatic coliphages) has to be analyzed.
- (***) Clostridium perfringens spores is selected as the most appropriate protozoa indicator.
 However sporeforming sulfate-reducing bacteria is an alternative if the concentration of Clostridium perfringens spores does not allow to validate the requested log10 removal.

Also risk assessment is required, but it's not currently defined

Outline



Solar treatments for water disinfection & decontamination

Advanced Oxidation Processes



TiO₂/UVA

(Carey et al., 1976)

$$TiO_2 \xrightarrow{hv} TiO_2(e^- + h^+)$$

$$h^+ + H_2O \rightarrow {}^{\bullet}OH + H^+$$

$$e^- + O_2 \rightarrow O_2^{\bullet-}$$

Photo-Fenton

$$Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + OH^- + {}^{\bullet}OH$$

(J. Chem. Soc., 1894)
 $Fe^{3+} + H_2O \xrightarrow{hv} Fe^{2+} + H^+ + {}^{\bullet}OH$
(several authors, early 90s)

Photochemical processes

- \rightarrow H₂O₂
- > Persulfate
- Peroxymonosulfate
- Peracetic acid
- > Chlorine



- No generation of HO by solar wavelengths reaching the Earth's surface
- High efficiency on water microbial inactivation
- Direct oxidation and complex microbial inactivation mechanisms

Ozone for water treatment



Disadvantages:

- **Solubility and activity:** When the ozone dosage is too low, some germs and bacteria may survive, which is why higher ozone concentrations are used. If the water being treated is rich in organic compounds or total suspended solids, then ozone will decay more rapidly, as it reacts with these contaminants.
- **High costs:** During corona discharge, around 85% of the energy is lost via heat waste, making ozone treatment extremely energy-intensive. Treating water with ozone is also energy-intensive because it requires high-class equipment, expensive technology, and an operator that knows how to work the complicated system.
- Reactivity and toxicity: Ozone's reactivity with metals can cause issues in wastewater treatment pipes and containers, therefore corrosion-resistant materials such as stainless steel must be used, which adds to plant construction costs.
 Because of the high toxicity, ozone levels need to be constantly monitored.
- **Reactivity and byproducts:** If the water contains bromide ions, ozone can react to form brominated compounds (like bromate ions/salts), which can cause human cancers. ozone leaves nothing behind; any ozone that doesn't react with the water contaminants is immediately broken down, leaving nothing to monitor after the disinfection process.

Photochemical processes: UV-C with oxidative

agents

H_2O_2

$$H_2O_2 \xrightarrow{UVC} 2HO^{\bullet}$$

(Goldstein et al., 2007)

Chlorine

$$HClO \xrightarrow{UVC} HO^{\bullet} + Cl^{\bullet}$$

$$ClO^{-} \xrightarrow{UVC} O^{\bullet-} + Cl^{\bullet}$$

$$O^{\bullet-} + H_{2}O \rightarrow HO^{\bullet} + OH^{-}$$

(Chuang et al., Environ Sci Technol, 2017. 51(23) 13859-13868.)

Persulfate / Peroxymonosulfate

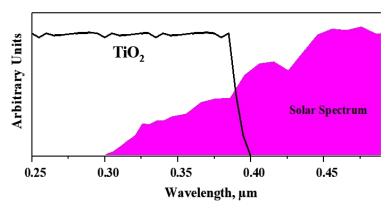
$$S_{2}O_{8}^{2-} \xrightarrow{UVC} 2 \underbrace{SO_{4}^{\bullet-}} + H_{2}O \xrightarrow{HO^{\bullet}} + SO_{4}^{2-} + H^{+} \quad HSO_{5}^{-} + H^{\bullet} \xrightarrow{SO_{4}^{\bullet-}} + H_{2}O$$

(Sánchez-Montes et al. Environ. Sci.: Water Res. Technol., 2020, 6, 2553–2566) (Guerra-Rodríquez et al., Water 2018, 10, 1828)



- SO4⁻⁻ has a high oxidation potential (2.5–3.1 V) comparable or even higher than OH
- Reacts more selectively and efficiently with organic compounds that contain unsaturated bonds or aromatic π electrons
- Half-life of sulfate radicals is supposed to be 30–40 μs
- Chlorine radicals oxidation potential 1.5V. High half-life

Solar Heterogeneous photocatalysis: TiO₂



Linearly dependent on the energy flux

but only ~5% of the whole solar

spectrum is available for TiO₂ band-gap.

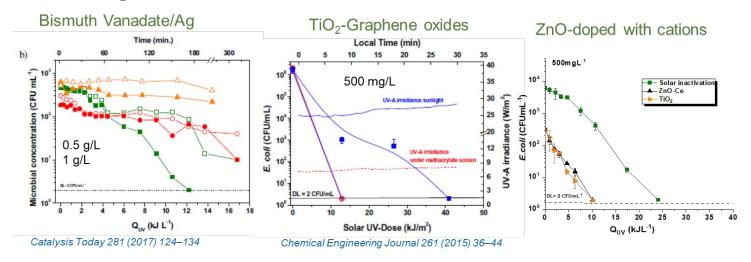
♦ 75% of solar collector efficiency and 1% for the catalyst means 0.04% original solar photons are efficiently used. This is a rather inefficient process.

Mild catalyst working under mild conditions with mild oxidants.

- ♦ As concentration of contaminants and water ionic strength increase: slow kinetics and unpredictable mechanisms need to be solved.
- ♦ TiO₂ efficiency improved with the addition of powerful oxidants or when doped (with iron, nitrogen...) to undertake practical applications.
- ◆ Pure TiO₂ can utilize only UV and new catalysts able to work with the visible component of the solar spectrum are needed.

Solar photocatalysis: enhancement strategies

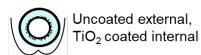
DOPED TiO₂ & OTHER SEMICONDUCTORS



IMMOBILIZATION OF THE CATALYST IN SUBSTRATES







- Lower microbial reduction efficiency than with suspended catalyst
- Scaling-up challenge

Photochemical processes

UV-C

 Low efficiency in CEC and OMC elimination. In crop irrigation, regenerated wastewater is normally stored along hours/days before being used.

Free Chlorine (Chlorination)

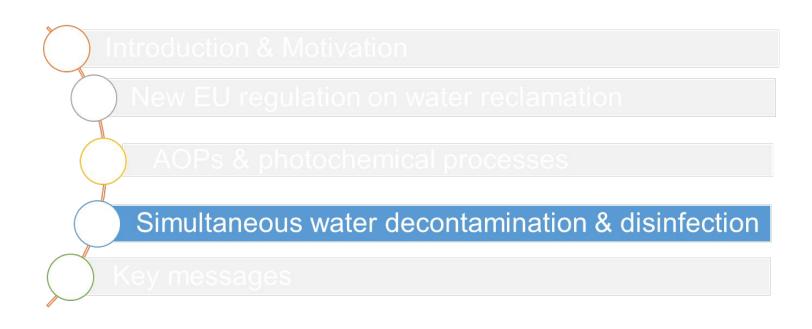
- Low efficiency in CEC and OMC elimination.
- Generation of toxic byproducts: trihalomethanes (THM), haloacetic acids.



UV-C/ free Chlorine

- Low studied combination for CEC and OMC elimination.
- Effect of UVC in THM formation.

Outline



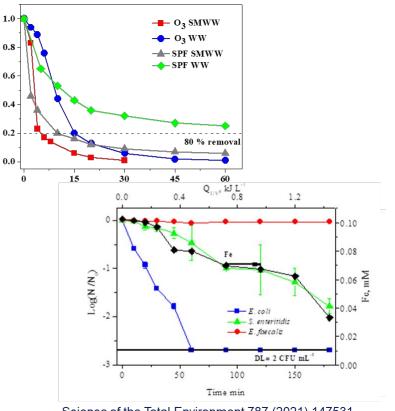
Simultaneous removal of CECs and Pathogens

Ozonation pilot plant



Raceway Pond Reactor





Science of the Total Environment 787 (2021) 147531 Science of the Total Environment 766 (2021) 144320

PANIWATER project



Assess the efficiency of ozonation for the treatment of urban wastewater (UWW), with three different O₃ doses. The parameters for the evaluation of the treatment were degradation of organic microcontaminants (OMCs), disinfection, generation of disinfection by-products (DBPs) and toxicity.









Municipal Urban Wastewater treatment plant "El Bobar" located in Almeria. South East of Spain.

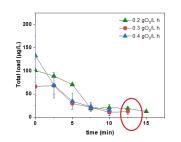
- $0.2 \text{ gO}_3/\text{h L } (25\%)$.
- OMCs:UHPLC-QqQ-MS/MS. **DBPs**:HS-GC-QqQ-MS/MS.
- $0.3 \text{ gO}_3/\text{h L } (50\%)$. 0.4 qO₃/h L(100%).
 - Bromate and bromide: IC.
- Escherichia coli and Coliforms.
- Plate counting Chromocult© agar media.
- Cytotoxicity.
- Phytotoxicity.
- Bacteria survival.

Microcontaminants

SFX, TMP, DCF, ACE, CBZ, CAF

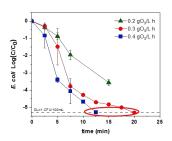
PANIWATER project

DEGRADATION OF OMCs



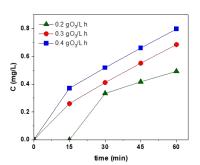
O ₃ doses (gO ₃ /h L)	time (min)	O ₃ Consumption (mg/L)
0.2	12.5	6.6
0.3	10.0	11.7
0.4	7.5	8.9

DISINFECTION

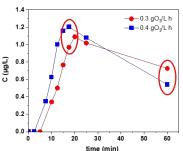


O ₃ doses (gO ₃ /L h)	time (min)	O ₃ Consumption (mg/L)
0.2	15.0	7.7
0.3	20.0	19.7
0.4	12.5	13.8

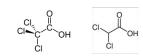
BROMATE



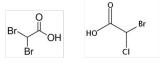




HALOACETIC ACIDS (HAAs)



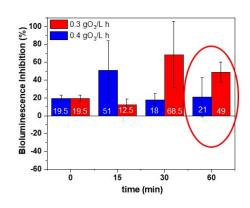
Chlorinated HAAs = Constant



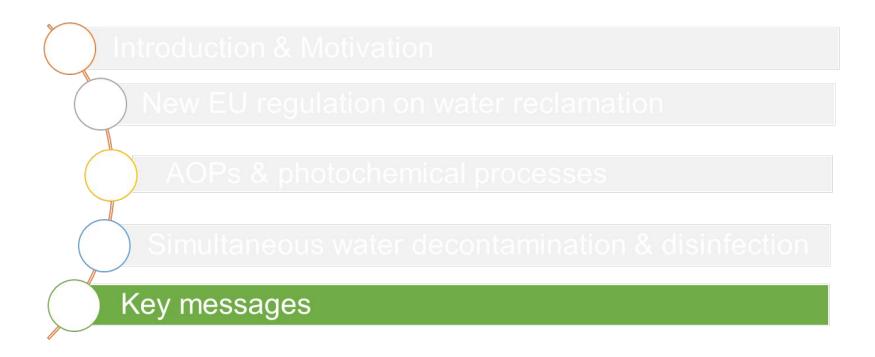
Brominated HAAs = Increased



BACTERIAL SURVIVAL



Outline



Key messages

- Water scarcity and bad water quality are problems affecting all over the world, which
 makes it crucial to find alternative water sources, such as municipal wastewater.
 Municipal wastewater treatment, jointly with desalination, mean a key strategy for
 trying to maintain high human life quality.
- A deep evaluation on the specific problem to be solved must be done just to focus on the optimum treatment option.
- Normally, different AOPs based technologies show highly efficiency as tertiary treatment for CECs & ARB elimination, but economic, health and life cycle assessments must support the final selection.
- Solar based AOPs are considered a sustainable and actual viable option for reducing contaminant impact on the Environment.
- Water quality parameters monitoring as well as contaminant transfer to crops must be carried out for ensuring a "safe reuse". Risk Assessment studies are required





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