1 ABACO-2: a comprehensive model for microalgae-bacteria consortia validated outdoor at

- 2 pilot-scale
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13 Abstract

Modelling microalgae-bacteria in wastewater treatment systems has gained significant attention in 14 15 the last few years. In this study, we present an enhanced version of the ABACO model, named 16 ABACO-2, which demonstrates improved accuracy through validation in outdoor pilot-scale systems. ABACO-2 enables the comprehensive characterization of microalgae-bacteria consortia dynamics, 17 allowing to predict the biomass concentration (microalgae, heterotrophic bacteria, and nitrifying 18 bacteria) and nutrient evolution. The updated version of the model incorporates new equations for 19 20 nutrient coefficient yields, oxygen mass balance, and microorganism cellular decay, while significantly reducing the number of calibrated parameters, simplifying the parameter identification. 21 22 Calibration and validation were performed using data from a 80 m² raceway reactor operated in a semicontinuous mode over an extensive period (May to November, total of 206 days) at a fixed 23 24 dilution rate of 0.2 day⁻¹ (corresponding to 5 days of hydraulic retention time), where untreated urban wastewater was used as culture medium. ABACO-2 exhibited robustness, accurately forecasting 25 biomass production, population dynamics, nutrient recovery, and prevailing culture conditions across 26 a wide range of environmental and water composition conditions. Mathematical models are essential 27 instruments for the industrial development and optimization of microalgae-related wastewater 28 29 treatment processes, thereby contributing to the sustainability of the wastewater treatment industry.

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32 **KEYWORDS:** Microalgae, wastewater, modeling

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34 **1. Introduction**

Water reuse and recycling have become crucial topics of discussion in recent decades, owing to 35 36 their significant environmental and social implications. With rapid industrialization and population 37 growth, the volume of wastewater generated annually has escalated (Angelakis and Gikas, 2014). contributing to water stress in various regions worldwide, particularly in southern Europe, including 38 Spain, Italy, and Greece (Strosser et al., 2012). Consequently, researchers have directed their efforts 39 toward developing innovative water remediation technologies. Among these technologies, 40 41 microalgae-based wastewater systems have emerged as a promising solution, capable of replacing traditional secondary and tertiary classical wastewater treatment processes (Abdel-Raouf et al., 42 2012). The utilization of microalgae in water remediation offers several advantages. Firstly, it 43 demands less energy compared to conventional methods. Secondly, it significantly reduces 44 45 greenhouse gas emissions. Additionally, microalgae systems eliminate residues effectively and yield 46 valuable biomass, which can be utilized in various industrial applications (Mohd Udaiyappan et al., 2017). 47

Microalgae are photosynthetic microorganisms capable of using CO₂ as a carbon source and light 48 49 as an energy source, in addition to using nitrogen (N) and phosphorus (P) present in wastewater to produce biomass. Moreover, thanks to the presence of bacteria, the organic matter's degradation is 50 ensured (Muñoz and Guieysse, 2006). The effectiveness of microalgae-based wastewater systems 51 has been demonstrated through pilot-scale testing in raceway reactors (Morillas-España et al., 52 2021b) and thin layers (Grivalský et al., 2019). These studies have proven that these systems are 53 robust and reliable, enabling the recovery of nutrients from wastewater and producing treated water 54 that meets the regulatory standards set by European legislation (Council Directive of May 1991 55 concerning urban wastewater treatment, 1991). One of the key factors contributing to the success 56 of this technology is its resilience in the face of significant variations in water composition and 57 58 weather conditions (Nordio et al., 2023). However, high costs and relatively low efficiency are 59 foreclosing the entire industrial development of these systems (Acién et al., 2014). The industrial development of these processes can be possible only after the improved understanding and 60

optimized management of the biological system allowing to study which are the main parameters
 influencing biomass productivity and the water remediation capacity (Solimeno and García, 2017).

63 In this framework, mathematical modelling serves as a valuable tool for describing these processes; 64 as it enables simulations and performance evaluations under different environmental and process conditions, aiding optimization and the development of control strategies (Solimeno and García, 65 2017). In the literature, it is possible to find many examples of microalgae models that evaluate the 66 growth rate as a function mainly of light, temperature, nutrients and pH (Lee et al., 2015). On the 67 68 contrary, a few examples of comprehensive models for microalgae-based wastewater treatments are available, defined as models that consider the effect of multiple process parameters and 69 70 biological mechanisms. Examples of recent comprehensive models are the Zambrano model (Zambrano et al., 2016), the BIOALGAE model (Solimeno et al., 2019), and the ALBA model (Casagli 71 72 et al., 2021). Nevertheless, it is challenging to obtain validation data over an extended period and 73 using urban wastewater in industrial facilities, so in the literature, some alternatives can be found, 74 such as validations achieved with digestates, synthetic waters, or small-scale reactors. However, it is crucial to develop models that accurately represent the biological system in real conditions (both 75 76 environmental and operational) since they are intended for use in commercial wastewater treatment 77 facilities. In Spain, for example, there are currently four industrial facilities using microalgae for urban wastewater treatment, employing reactors ranging from 0.5 to 1.0 ha (Masojídek et al., 2022). 78

79 To apply these models to large-scale production, it is important to balance complexity and realism. 80 For instance, very complex mathematical descriptions of all the metabolic reactions involved in photosynthesis, may not improve the model prediction accuracy, compared to the computational 81 cost required. Additionally, mechanisms that are not relevant for long-time series and continuous 82 83 conditions can be omitted from the model (Darvehei et al., 2018). Furthermore, in wastewater, thousands of bacterial groups can interact with microalgae, and it is fundamental to carefully select 84 85 the most relevant groups to reduce the number of equations and the overall complexity. Generally, 86 only a few groups are considered relevant, as they influence nutrient uptake from the culture 87 medium. The first model was proposed by Buhr & Miller in 1983 (Buhr and Miller, 1983), for example,

only two populations (algae and aerobic bacteria) were considered. However, afterwards, the leading
bacterial groups identified were the heterotrophic bacteria, which permitted the oxidation of the
organic matter, and nitrifying bacteria that compete with microalgae for the consumption of nitrogen
(together with phosporous and carbon) in some cases distinguished between ammonia-oxidizing
bacteria (AOB) and nitrite-oxidizing bacteria (NOB) (Aparicio et al., 2022a).

In this work, an improved version of the ABACO model (Sánchez-zurano et al., 2021), named the 93 ABACO-2 model is proposed, calibrated and validated for outdoor conditions. The original ABACO 94 model is a recent microalgae-bacteria model designed to represent the dynamics of these 95 96 populations in wastewater systems. This model underwent validation over three weeks using a 1L tubular reactor, fed with various types of wastewaters. While it served as a valid foundation for 97 98 subsequent modeling research, further improvements were necessary to validate it at the pilot-scale 99 level. Specifically, new equations are included to reduce the number of calibrated parameters, and 100 the use of the oxygen mass balance allows for improvement in the accuracy of the model. Equations 101 have been implemented using Python language and numpy packages while the calibration has been carried out with Scipy library and the optimization tool that uses the "Nelder-Mead" algorithm. The 102 103 model has been validated over an extensive period (May-November); data have been collected from a demonstrative pilot-scale raceway reactor of 12 m³ (80 m²) on which the prevailing strain was 104 Scenedesmus sp.. The reactor was operated in semi-continuous mode at a fixed rate of 0.2 day⁻¹ 105 using urban wastewater as culture medium. For this study, wastewater not pre-treated besides the 106 removal of large particles was used, so subjected to high variation in terms of composition. This 107 108 large variability of water nutrient concentration, together with different values of solar radiation and temperature typical of different year seasons, allowed to calibrate and validate the model in a wide 109 range of conditions (Section 2.2), so increasing its prediction accuracy and robustness. 110

111 **2. Material and methods**

112 **2.1. Raceway reactor and inoculum**

Experimental data were collected from an 80 m² (12 m³) raceway reactor working in semi-continuous
 mode between April and December, with a fixed dilution rate (reverse of the hydraulic retention time

time) of 0.2 day-1, meaning that every day a 20% of the total volume of the culture have been 115 removed and replaced with wastewater. The raceway was installed in the SABANA Demo Plant 116 located in the IFAPA research centre in La Cañada, Almería, Spain. It was composed of a double 117 channel of 40 m and a sump of 0.59 m³ for the gas injections through diffusers and an electric motor 118 connected to a paddlewheel system for culture mixing. The pH was monitored through sensors and 119 controlled through CO₂ injection in the reactor sump. Additionally, an independent airflow allowed 120 reducing the concentration of dissolved oxygen. The culture depth was fixed at 0.15 m. In order to 121 122 thoroughly monitor the culture dynamics, additional sensors have been installed for recording the dissolved oxygen (0-400% Sat), pH (0-14), temperature (0-80°C) and culture depth (4-40 cm). 123 Moreover, a meteorological station allowed for registering the weather conditions regarding solar 124 radiation and environmental temperature. 125

The chosen inoculum was Scenedemus sp. because it was demonstrated to be suitable microalgae 126 that can easily be grown in wastewater and a wide range of conditions (Fernández Sevilla et al., 127 2006). The strain was initially grown on a fertilizer medium (0.9 g·L⁻¹ of NaNO₃, 0.18 g·L⁻¹ of MgSO₄, 128 0.14 g·L⁻¹ of KH₂PO₄ and 0.003 g·L⁻¹ of Kerantol), first using columns of 0.1 m³ and then in a tubular 129 system of 3 m³ until it reached the concentration of 1 g•L⁻¹. The biomass was then used as inoculum 130 for the raceways. The culture was diluted with wastewater and kept in batch mode for one week, 131 after that being operated in semi-continuous mode until it reached a stable biomass concentration 132 (approximating a steady state condition). 133

134 **2.2.** Environmental conditions and water composition

As previously mentioned, environmental conditions in terms of temperature and radiation were continuously recorded throughout the entire period. The Photosynthetically Active Radiation (PAR) registered were ranged from 2000 μ E·m⁻²·s⁻¹ during the months of April and May, gradually decreasing during the colder seasons with peaks at 1200 μ E·m⁻²·s⁻¹, with an average of 620 μ E·m⁻ ²·s⁻¹ and 350 μ E·m⁻²·s⁻¹respectevely. Regarding temperature, the highest values were recorded during the summer season (July-August), reaching peaks of 38°C, while in the spring season (April-

May), they ranged between 25-35 °C, and in the autumn season (September-November), between
25-12 °C.

143 Regarding the culture medium, it consisted of wastewater collected from the University of Almería during the entire data collection period, except for August when the water was sourced from the 144 primary water treatment plant in the city of Almeria. In both instances, the water underwent pre-145 treatment, involving the removal of solid particles through an industrial filter (Azud Helix, 200 µm), 146 before being introduced into the culture. During the study period, the nutrient concentration of the 147 148 wastewater varied over a wide range. Specifically, ammonium (NH₄⁺) concentration varies between 10 - 400 g·m⁻³, nitrate (NO₃⁻) concentration ranged from 0- 13 g·m⁻³, phosphate (PO₄²⁻) between 30 149 - 76 g·m⁻³, while the chemical oxygen demand (COD) between $100 - 600 \text{ gO}_2 \text{ ·m}^{-3}$. 150

151 **2.3. Biomass concentration and nutrients analysis**

The influent wastewater and the filtered effluent were analysed in terms of nutrient content (N-NO₃⁻, 152 153 N-NH₄⁺, P-PO₄³⁻) and COD. The biomass concentration in the culture was daily measured through the dry weight (DW) method. The culture was collected in the morning after the reactor sump, and 154 100 mL were filtered in a 0.5 µm filter and let dry for 24h at 80°C in an oven. Regarding the nutrients, 155 they were analysed through colourimetric methodologies in a spectrophotometer according to 156 157 standard procedures (Standard IC 74246, Standard IC 38364, Standard IC 59755). The total COD was measured with Hanch-Lange kits (LCI-400) and the biodegradable soluble organic matter 158 (BSMO) was estimated as a percentage of the total COD as reported in the literature by Pasztor I. 159 et al., 2009. 160

161

2.4. Data collection and analysis

Experimental online data were collected every second by a set of sensors, connected to a Programming Logic Controller (PLC) and a Supervisory Control and Data Acquisition (SCADA) system. On the contrary, data coming from laboratory analysis as described in the previous section, were collected once a day. Given the big amount of data available, it was necessary to perform a prior data analysis following a procedure inspired by the "Cross Industry Standard Process Alliance

for Data Mining" (CRISP-DM) approach (Ncr and Clinton, 1999). This methodology is one of the most used among data mining problems and it is composed of six main steps: (i) Business understanding, (ii) Data understanding, (iii) Data preparation, (iv) Modelling, (v) Evaluation, and (vi) Deployment. In this research, the first five steps have been developed as briefly described below:

Business understanding: the objective is to develop a model that can describe the evolution
 of microalgae-bacteria populations in wastewater-related systems. The aim is to develop a
 tool that allows the simulation of the variation of biomass and nutrient concentration with time
 as a function of environmental and process parameters.

II. <u>Data understanding</u>: data have been collected as described in the previous section, studied,
 and analysed.

177 III. <u>Data preparation</u>: Datasets have been ordered, cleaned, and prepared for the modelling step.

- IV. <u>Modelling</u>: the biological system has been modelled as described in the next sections.
 Stepping back to data preparation is often necessary. A part of the experimental data set has
 been used for the identification of the calibration parameters.
- 181 V. <u>Evaluation</u>: the developed model has been validated with long-term outdoor dataset. If the 182 quality of the model was not enough to reach the defined objective, the data preparation and 183 modelling part has been reviewed.
- 184 VI. <u>Deployment</u>: this step was not addressed in this research. However, a web interface for 185 model utilization will be developed in future works.
- 186 **3. Model development**

187 **3.1. Model concept**

This work considers three microbial groups: microalgae, heterotrophic bacteria and nitrifying bacteria as they are the main actors in the nutrient uptake and the O_2/CO_2 fluxes (Figure 1). During the day, microalgae perform photosynthesis consuming the inorganic carbon and fixing nitrogen and phosphorus while producing O_2 . The preferred nitrogen form for microalgae growth is NH_4^+ , which is highly present in urban wastewater. Microalgae compete with nitrifying bacteria for the uptake of this compound since they use it to transform it into NO_3^- during the nitrification. The nitrification process

involved the oxidation of NH₄⁺ to NO₂⁻ by AOB and then NOB transform NO₂⁻ into NO₃⁻. For this 194 study, it is considered that the nitrification is complete because it was not registered a significant 195 concentration of NO₂ in the culture (consistently below 5 g·m⁻³). Moreover, microalgae can use NO₃⁻ 196 197 as a form of nitrogen, but its consumption takes place only when ammonium is found below a given threshold (prior experimental analysis have estimated it as 80 g·m⁻³). Heterotrophic bacteria are 198 199 considered the leading bacteria group as they are the main ones responsible for the degradation of organic matter. For the present work, the COD has been fractionated as proposed by Pasztor I. et 200 201 al., 2009. Briefly, the total COD can be divided into two main fractions: the biodegradable (readily 202 and slowly) and the non-biodegradable (soluble and particulate). Heterotrophic bacteria can consume only the readily biodegradable organic matter estimated as 22% of the total COD and it 203 204 will be called BSMO (biodegradable soluble organic matter), as proposed by the same authors. Summarizing, the BSMO concentration is decreased in the culture due to the heterotroph's activity 205 206 and it can be increased due to the cellular death and decay of the microorganisms present in the 207 culture. Regarding the gas fluxes, the inorganic carbon necessary for microalgae growth is partially 208 provided by the on-demand injection of CO₂ for pH control and the natural release of CO₂ given by 209 bacteria during respiration. This study assumes that microalgae are never limited by inorganic 210 carbon concentration, as CO₂ injection always ensures enough carbon availability for microalgal growth, as already demonstrated by previous studies (Posadas et al., 2015). Additionally, 211 212 experimental data performed into the system indicate that the liquid bulk alkalinity into the medium 213 is never exhausted, preventing the loss of injected CO_2 used for pH control. On the contrary, the O_2 214 is produced during photosynthesis by microalgae and used by bacteria for their respiration, and it is partially removed from the culture broth due to mass transfer phenomena, mainly aeration into the 215 sump installed on the reactor. 216

217 Main changes implemented from ABACO model

Despite the ABACO model served as the starting point for the development of ABACO-2, significant
modifications have been implemented to enhance prediction accuracy and process understanding.
Indeed, the ABACO model proposed by Sánchez-zurano et al., 2021, can be considered a

preliminary study, conducted with a limited expertise regarding microalgae cultivation phenomena. Furthermore, this model was developed using a restricted dataset and calibrated using data from laboratory-scale experiments conducted under controlled conditions. In contrast, with ABACO-2, the intention is to calibrate parameters under industrial conditions, utilizing a more extensive dataset that encompasses diverse operational and climatic scenarios.

The foremost modification involved the integration of models to account for cell death and respiration, 226 coupled with the variation in BSMO content within the culture, subsequently reduced by heterotrophic 227 228 bacterial activity. This refinement also led to a reduction in the number of parameters necessitating calibration. As an additional parameter reduction strategy, it was assumed that phosphate 229 consumption by bacteria is minimal and primarily relevant for microalgae. Consequently, phosphate 230 consumption yields for these microorganisms were excluded from the calibration set. Regarding 231 nutrient yields, equations were introduced to describe the dynamics of nutrient uptake by algae as a 232 function of their concentration in the medium. In this context, process rates were adjusted to consider 233 that NO₃ consumption by algae is significant only when NH₄ levels are substantially reduced. 234 Furthermore, in relation to nutrients, a correction parameter was incorporated into the Monod 235 236 equations to account for nutrient accumulation by microalgae, preventing zero growth in such 237 scenarios. In the context of refining the calibration process, parameters associated with nutrient consumption by bacteria (originally calibrated) were set and adopted from the Activated Sludge 238 Models (ASM), and an oxygen balance was introduced. The O₂ concentration is a continuous 239 240 measuremet within the reactor and it significantly facilitated parameter recognition during the calibration process. Finally, the parameters of the cardinal temperature equations were adjusted 241 242 using those proposed by Casagli et al., 2021, as they are more representative, having been calibrated while considering winter seasons. 243

244 **3.2. Model components**

245 This section summarizes the main model components:

• S_{NH4} [g_{NH4}•m⁻³]: <u>ammonium</u>. It is present in the influent, and it is consumed especially by 247 microalgae and nitrifying bacteria and in a lower amount by heterotrophic bacteria.

S_{NO3} [g_{NO3}•m⁻³]: <u>nitrate</u>. This form of nitrogen is generally null in the influent, but it is generated by
 the nitrifying bacteria during the nitrification process. Nitrate is consumed by microalgae when
 ammonium concentration in the medium is low.

S_{PO4} [g_{PO4}•m⁻³]: <u>phosphate</u>. Phosphorus is present as a dissolved component in the water inlet.
 Its consumption is mainly due to the activity of microalgae, while the uptake from bacteria is considered negligible.

- S_{BSMO} [g_{BSMO}•m⁻³]: <u>biodegradable soluble organic matter</u>. This is a fraction of the total COD,
 assumed as 22%. It is consumed by heterotrophic bacteria and generated during the cellular
 decay of both microalgae and bacteria.
- S₀₂ [g₀₂•m⁻³]: <u>dissolved oxygen</u>. Oxygen is produced by microalgae during photosynthesis and
 consumed by microalgal respiration and by the activity of both bacterial populations. Moreover,
 the dissolved oxygen can be stripped to the atmosphere by bubbling air into the reactor sump.
- X_{alg} [g_{alg}•m⁻³]: <u>microalgae biomass</u>. Microalgae proliferate starting from an initial inoculum, thus
 microalgae biomass is produced by fixing nitrogen and phosphorus, also consuming CO₂ while
 producing oxygen. Microalgae concentration in the inlet wastewater is considered negligible,
 while a given amount is harvested every day. Moreover, their growth decreases due to cellular
 death.
- X_{nit} [g_{nit}•m⁻³]: <u>nitrifying bacteria biomass</u>. Nitrifying bacteria proliferate starting from an initial inoculum by consuming nitrogen in the form of ammonium and releasing nitrate. It is assumed that their concentration entering the system is negligible, while a given concentration is exiting during the harvesting. Moreover, their growth decreases due to cellular death.
- X_{het} [g_{het}•m⁻³]: <u>heterotrophic bacteria biomass</u>. Heterotrophic bacteria proliferate starting from an initial inoculum by consuming the BSMO and nitrogen in the form of ammonium. It is assumed that their concentration entering the system is negligible, while a given concentration is exiting during the harvesting. Moreover, their growth decreases due to cellular death.
- 273 **3.3. Boundary conditions**

274 Concentrations must be always positive or equal to zero. This boundary condition can be expressed 275 as in equation (1): when a concentration is approaching zero (assuming ε in the order of 10⁻⁸), its 276 derivative has to be equal or more than zero, meaning that it cannot generate negative matters.

277
$$if X_i \le \varepsilon \rightarrow \dot{X}_i \Big|_{X_i=0} \ge 0$$
 (1)

As a result, all the balances implemented for this model have been implemented according to equation (2), guaranteeing the boundary conditions to be satisfied.

280
$$\dot{X} = f(x, y) \cdot \frac{X}{X + \varepsilon}$$
 (2)

281 **3.4. Biological processes**

Table 1 summarizes the processes taken into consideration of the microalgae and the bacterial growth, while Table 2 is the relative matrix of the stoichiometric parameters. The mass balances for the microorganism's growth have been built according to equation (3):

$$Inlet - Outlet + Reaction = Accumulation (3)$$

where the *Inlet* and *Outlet* are the flowrates in $[m^{-3} \cdot s^{-1}]$ in and out of the system, generically defined as (4) and (5):

$$Inlet = Q_d X_{in} \quad (4)$$

$$0utlet = Q_h X_{out}$$
(5)

where Q_d is the dilution flow rate in $[m^{-3} \cdot s^{-1}]$, Q_h is the harvesting flow rate in $[m^{-3} \cdot s^{-1}]$, X_{in}/X_{out} (or S_i/S_{out}) is the concentration of component inlet or outlet in $[g \cdot m^{-3}]$.

292 The reaction (r_i, [g•m⁻³•day⁻¹]) term can be obtained by summing the product of the yield coefficients,

vi (Tables 2 and 5) and the process rate, ρ_{i} as described in (6) (Henze et al., 2000)

294
$$r_i = \sum_i v_{i,j} \rho_j \quad (6)$$

In summary, the processes considered in the ABACO-2 model are:

- ρ₁: <u>microalgae growth in NH₄</u>[±]. Microalgae grow photosynthetically using NH₄⁺ as a nutrient source, and contemporarily consuming PO₄³⁺ and CO₂ while producing O₂.
- p₂: microalgae growth in NO₃⁻. Microalgae grow photosynthetically using NO₃⁻ as a nutrient source, and contemporarily consuming PO₄³⁺ and CO₂ while producing O₂. The growth in this nitrogen source is activated only once the medium is decreasing in NH₄⁺ concentration.
- 301 ρ_3 : <u>microalgae decay</u>. This process includes both the algal biomass loss (decay), increasing the 302 BSMO concentration in the medium, and the algal respiration that leads to a consumption of 303 oxygen all over the entire process.
- ρ_4 : <u>nitrifying bacteria growth</u>. Nitrifying bacteria growth consumes NH₄⁺ and O₂ and produces NO₃⁻.
- p₅: <u>nitrifying bacteria decay</u>. Bacterial biomass loss due to their decay; it leads to an increase in
 the BSMO concentration.
- 308 ρ_6 : <u>heterotrophic bacteria growth</u>. Heterotrophic bacteria growth consuming the BSMO, O₂ and 309 NH₄⁺.
- p₇: <u>heterotrophic decay</u>. Bacterial biomass loss due to their decay (it leads to an increase in the
 BSMO concentration).

312 **3.4.1.** Photosynthesis and respiration

The growth rate as a function of light was modelled using the equation proposed by Molina (Grima et al., 1994), as described in equation (7), where I_k in [μ E•m⁻²•s⁻¹] is the irradiance constant that represents the equivalent irradiance necessary to reach half of the maximal growth rate, *n* is the shape constant and I_{av} is the average light inside the reactor in [μ E•m⁻²•s⁻¹].

317
$$\mu(I_{av}) = \frac{I_{av}^{\ n}}{I_k^{\ n} + I_{av}^{\ n}} \quad (7)$$

The average light inside the culture was expressed following equation (8), and it depends on the incident light I_0 [µE•m⁻²•s⁻¹], the extinction coefficient K_a [m²•g⁻¹], the algal biomass concentration (X_{alg}) in [g•m⁻³] and the culture depth h [m].

321
$$I_{av} = \frac{I_0}{K_a X_{alg} h} \left(1 - \exp\left(-K_a X_{alg} h\right) \right)$$
(8)

Moreover, the average light was used to express the microalgal endogenous respiration as given by equation (9); where m_{max} and m_{min} are the maximum and the minimum respiration in [day⁻¹], l_{kr} is the irradiance necessary to stop photosynthesis and let begin the respiration and n_r is the shape form for respiration.

326
$$m_{alg} = m_{min} + \frac{m_{max} I_{av}^{n_r}}{I_{kr}^{n_r} + I_{av}^{n_r}} \quad (9)$$

Finally, bacterial decay has been taken into consideration as a constant effect during the cultivation process. m_{nit} and m_{het} [day⁻¹] have been modelled as a percentage of the maximum growth rate (as summarized in Table 4) corrected by a coefficient dependent on the temperature θ , as described in equation (10).

$$\theta = \theta_i (T - 20^\circ C) \quad (10)$$

Where θ_i are specific parameters that depend on the bacterial population considered (Table 3).

333 **3.4.2.** Influence of pH, temperature, dissolved oxygen and nutrients

334 As described in Table 1, for each microorganism, the growth rate depends on a maximum specific 335 growth rate µ value multiplied by a series of normalized factors that depends on the culture conditions of temperature, pH, O₂ and nutrient concentration. The growth rates of microalgae exhibit a bell-336 shaped function in response to temperature and pH. Initially, as temperature (or pH) increases from 337 low values, the growth rate rapidly increases until it reaches its maximum, corresponding to the 338 optimal parameter value. However, beyond the optimum, the growth rate decreases sharply with 339 340 further increases in temperature (or pH). The pH parameters, minimum, maximum, and optimal values, were determined through laboratory measurements using the photo-respirometric method 341 (Sánchez Zurano et al., 2021); Notably, bacterial parameters vary from those of microalgae, as their 342 growth is favoured by higher values of pH (Table 4). In contrast, temperature parameters were 343

derived from previous modeling studies (Casagli et al., 2021), with the optimal temperature aligning
 closely with that of microalgae.

The growth dependence can be described through a cardinal equation with inflexion, developed for the first time by Bernard et alt. (Bernard and Rémond, 2012) (11). Similarly, the model proposed by Ippoliti et al. was used to describe the pH dependence (Ippoliti et al., 2016) (12).

349
$$\overline{\mu(T)} = \frac{(T - T_{max})(T - T_{min})^2}{(T_{opt} - T_{min})[(T_{opt} - T_{min})(T - T_{opt}) - (T_{opt} - T_{max})(T_{opt} + T_{min} - 2T)]}$$
(11)

350
$$\overline{\mu(pH)} = \frac{(pH - pH_{min})(pH - pH_{max})^2}{(pH_{opt} - pH_{min})[(pH_{opt} - pH_{min})(pH - pH_{opt}) - (pH_{opt} - pH_{max})(pH_{opt} + pH_{min} - 2pH)]}$$
(12)

Regarding the effect of dissolved oxygen, it is known that high concentrations are inhibitory for microalgal photosynthesis. According to previous studies on *Scenedemsus sp.*, the growth rate can be reduced by 25% when the concentration is increased up to 150% Sat, while below 250% Sat the photosynthesis is completely stopped. This effect was modelled using the equation proposed by Costache et al., 2013 and reported in equation (13). On the contrary, oxygen has been modelled as a nutrient source for bacteria growth, as described below.

357
$$\overline{\mu(O_2)_{alg}} = 1 - \left(\frac{S_{O_2}}{S_{O_2,max}}\right)^2 \quad (13)$$

Finally, the influence of nutrient concentration on the growth rate was taken into account. As 358 mentioned, nitrogen is a fundamental macronutrient that must be provided to microalgae to ensure 359 360 their growth. The inorganic nitrogen can be assimilated into acids for the protein formations in many 361 forms, such as NH⁺₄, NO⁻₂ and NO⁻₃. However, in this study, the main nitrogen form present in 362 wastewater was ammonium, while nitrate was formed only after the complete nitrification process. Ammonium is the favoured nitrogen form for microalgae as it requires less energy to be assimilated. 363 Only after a given concentration threshold do microalgae begin to consume NO₃, which will be 364 365 transformed into NH₄ to be assimilated into the cells. Phosphate is another fundamental macronutrient for microalgal growth, as it is necessary for the synthesis of RNA into the nucleotides, 366 while it is assumed that this component is not consumed by bacteria. The growth rate as a function 367

of the substrate concentration, has been modelled using the Monod equation, as described in (14),
(Monod, 1949)

$$\overline{\mu(S_i)} = \frac{S_i}{S_i + K_s} \quad (14)$$

371 The Monod equations for the nitrogen and the phosphorus for the process n.1 and 2 in Table 1 have been modified by the inclusion of a correction factor. With this modification, zero-growth when no 372 373 longer nitrogen/phosphate are present in the medium was avoided. Indeed, it is already known that microalgae can store nutrients in cells guaranteeing their survival and growth even when the medium 374 375 is limited in nutrients. This fact can be represented by more complex models such as the Droop 376 model (Droop, 1970), which considers the cells quota of the limiting element. However, quotas are 377 difficult to be estimated as they required specific laboratory techniques. For this reason, in this work, a simplified description of this phenomenon was chosen by correcting the concentrations in the 378 Monod equation as the sum of the component available in the medium and the one present in the 379 algal biomass (equal to 10% in nitrogen and 2% in phosphorus, multiplied by the "assimilation" factor 380 α) (14), (15). 381

 $S_N = S_{N,medium} + X_{alg} * 0.1 * \alpha \quad (15)$

383

 $S_P = S_{P.medium} + X_{alg} * 0.02 * \alpha$ (16)

The kinetic parameters used for the Monod equation are summarized in Table 3.

385 **3.4.3. Nutrient yields**

Nutrient yield can be defined as the amount of nutrients consumed from the medium per gram of biomass produced. In the literature, it can be found that the nutrient yield for algae is not constant, but changes depending on the nutrient amount present in the medium. More specifically, it was found that the nutrient uptake rate is higher at lower nutrient concentrations until it is reached a maximum. This can be explained by some biological mechanisms like the "luxury uptake" (Solovchenko et al., 2019): microalgae store a larger amount of nutrients than the ones necessary for immediate growth. It is possible to suppose that nutrients yield not only depends on the nutrient concentrations in the

medium but even on environmental conditions and process parameters. However, modelling this 393 phenomenon is complex, and in the literature can be found different results that mainly depend on 394 the strain used and the type of experiment performed. The equations developed by Zurano et al., 395 396 2021 were taken as a good approximation of ammonium and phosphorus consumption rates by microalgae, as described in equation (17), where S is the substrate consumed by microalgae (NH₄⁺, 397 NO_3 or PO_4^{-3}). This was developed as a combination between and hyperbolic and a cardinal 398 equation. The hyperbolic equation, typically employed for describing microbial growth kinetics, helps 399 400 explain the increase in nitrogen and phosphorous coefficient yields with higher nitrogen or phosphorous concentrations in the medium. Furthermore, to account for the observed peaks in both 401 nitrogen and phosphorous coefficient yields, the cardinal equation has been applied within 402 predefined minimum and maximum ranges. The cardinal model enables the definition of maximum, 403 minimum, and optimal conditions for any variable, and it characterizes the influence of these 404 variables on the biological system's performance as a Gaussian function. All the parameters values 405 406 are summarized in Table 5.

407
$$Y_{s,alg} = \left[\frac{Y_{max} S^{t_s}}{S_N^{t_s} K_{s,Y_s}^{t_s}}\right] + \left[\frac{(S - S_{max}) (S - S_{min})^2}{(S_{opt} - S_{min}) \left(\left((S_{opt} - S_{min})(S_N - S_{opt})\right) - \left((S_{opt} - S_{max})(S_{opt} + S_{min} - 2S)\right)\right)}\right] (17)$$

Regarding the nutrient yield of bacteria, the ones proposed by the ASM models (Henze et al., 2000)
were considered a good approximation, and they are summarized in Table 3.

410 **3.4.4. Dissolved oxygen**

During the day, microalgae produce oxygen through photosynthesis, which is partially consumed for algal and bacteria respiration. At the same time, dissolved oxygen can be desorbed to the atmosphere according to two different phenomena: (i) natural mass transfer from the culture to the atmosphere in the reactor channels and paddlewheel; (ii) oxygen release and consecutive reduction of the culture dissolved oxygen thanks to the bubbling of air in the reactor sump. The two phenomena are represented by two different mass transfer coefficients K_{Ia} (equal to 1.0 and 110 h⁻¹ respectively) and Henry law as described in (18):

$$m_{O_2} = K la_i (H_{O_2} P_{O_2} - S_{O_2})$$
(18)

419 **4. Model parameters**

418

420 **4.1. Calibration procedure**

Figure 2 represents the calibration strategy adopted. The experimental dataset used to calibrate the model parameters was selected to include data from different seasons, such as summer, winter, and intermediate seasons (28/04-15/05, 9/08-15/08, 1/11-15/11). In total, the calibration days chosen were 41 (20% of the total amount of data). In this way, it was possible to address the parameters by accounting for various climatic conditions. The final set of parameters was chosen once the objective function described in (19) was minimized.

427
$$Obj = \sum \frac{\sum (y_{sim} - y_{exp})^2}{\sigma_{exp}}$$
(19)

where y_{sim} is the model output, y_{exp} the experimental data and σ_{exp} is the experimental data standard deviation. The experimental data used for the model calibration regarded X_{tot} , S_{NH4} , S_{PO4} , S_{NO3} , S_{BSMO} , S_{O2} where X_{tot} was defined as the sum of X_{alg} , X_{nit} and X_{het} and evaluated experimentally as the total dry weight. The model calibration was carried out using Scipy library in Python and "Nelder-Mead" algorithm which is a robust algorithm mainly used for solving unconstrained optimization problems (Gao and Han, 2012). The list of calibrated parameters with their corresponding values is presented in

435 Table 6.

436 4.2. Sensitivity analysis

Table 7 presents the findings of a sensitivity analysis that examined all biological and process parameters together with the associated standard deviation. In this analysis, each parameter was individually variated by +/-20% from its nominal value, and the percentage error (20) between the nominal parameter value (y_{nom}) and the variated parameter value (y_{var}) was evaluated.

441 %
$$err = \frac{\sum |y_{nom} - y_{var}|}{\sum y_{var}} * 100$$
 (20)

Results indicate that the most sensible parameters are the ones for microalgae growth rate as a 442 function of light (I_{av}, K_a, n), the maximum growth rates of all the organisms (µ_{max,alg}, µ_{max,nit}, µ_{max,het}) 443 and the nutrient yield of NH₄⁺ and NO₃⁻ (Y_{NH4,alg}, Y_{NO3,nit}, Y_{NH4,nit}). Additionally, the cardinal parameters 444 (T_{max}, T_{min}, T_{opt}) of temperature and pH (pH_{max}, pH_{min}, pH_{opt}) show to be highly sensitive. Given their 445 significant impact on the final prediction, these parameters should be carefully selected based on 446 the biological system analysed and the climatic conditions. Notably, only a few of the nutrient yields 447 were deemed relevant to model error. 448

449

4.3. Parameters uncertainty and error propagation

Once the most sensible parameters have been identified, it was possible to calculate the model 450 451 variance and the confidence interval (Denis Dochain, 2001). From the sensitivity analysis, it was possible to define a sensitivity matrix as (21), which collects the functions of the given output y by 452 variating the parameter pi. 453

454
$$S = \left[\frac{\delta y}{\delta p_1}; \frac{\delta y}{\delta p_2}; \dots; \frac{\delta y}{\delta p_j}\right] (21)$$

The standard deviation of the parameter can be calculated as (22), where C_{ij} is the covariance matrix 455 456 and p_i is the associated parameter:

457
$$\vartheta_j^2 = p_j \sqrt{C_{j,j}} \quad (22)$$

458 The covariance matrix is the inverse of the Fisher information matrix (23), defined as the variance of the score function (Fujita et al., n.d.). 459

460
$$F = C^{-1}$$
 (23)

And it can be calculated starting from the sensitivity analysis according to (24): 461

462
$$F = S^T Q^{-1} S$$
 (24)

where Q is the array of the measured standard deviation. 463

464 Once the covariance of the parameter has been evaluated, it was possible to estimate the model 465 error propagation of the output variable y at the given instant time t as (25):

$$\sigma_{y}(t) = \sqrt{\sum_{i=1}^{m} S_{i}(t)^{2} \vartheta_{p_{j}}^{2}} \quad (25)$$

467 The model confidence interval at 95% has been calculated on the model output as (26):

468
$$[y_i - 1.95\sigma_y; y_i + 1.95\sigma_y]$$
 (26)

469 **5. Model validation**

466

The model prediction accuracy has been evaluated by calculating the normalized squared root error (NRMSE) and Theil's inequality coefficient (TIC) (H. Theil. et al., 1959), as described in (26), (27). Results are reported in Table 8, it is important to note that when the TIC is lower than 0.3, it is possible to consider a good agreement between the experimental data and the model predictions. Additionally, Figure 3 to Figure 5 represent the model estimation and the respective experimental data, as described in the next session.

476
$$NRMSE = \frac{\sqrt{\Sigma(y_{sim} - y_{exp})^2}}{(y_{exp,max} - y_{exp,min})} \quad (26)$$

477
$$TIC = \frac{\sqrt{\Sigma(y_{sim} - y_{exp})^2}}{\Sigma y_{sim}^2 + \Sigma y_{exp}^2} \quad (27)$$

Overall, it is possible to affirm that the model can accurately reproduce the biological system. The
ABACO-2 model is remarkably accurate for describing the total biomass concentration and the
nutrient concentration evolution (NRMSE between 0.14 and 0.23, TIC between 0.16 and 0.24).
Additionally, the model accurately can trace the dissolved oxygen in the culture (NRMSE= 0.14,
TIC=0.21).

483 6. Discussion

484 **6.1. Simulation results**

Although microalgae-bacteria consortia are considered a promising technology for wastewater 485 treatment, they still address several challenges. An accurate microalgae-bacteria model is a powerful 486 487 tool to overcome the bottlenecks of this technology (Aparicio et al., 2023). The ABACO-2 model aims 488 to act as a tool for robust and accurate prediction of the evolution of biomass concentration in a microalgae-bacteria system, and, therefore, to differentiate between the evolution of both 489 populations in the face of operational and environmental conditions. Figure 3A represents the 490 evolution of the total biomass concentration from 15th May to 15th November. The dots represent the 491 492 experimental data while the model is shown with a solid line, and the shading is the model confidence 493 interval at 95%. Total biomass is mean the sum of the contributions of algae and bacteria that can be approximated to the biomass experimentally evaluated through the dry weight method. The 494 results showed that the model reproduces the trend of the experimental data, with a NRMSE=0.21 495 and TIC=0.16 (Table 8). The concentration of heterotrophic and nitrifying bacteria over the study 496 period is shown in Figure 3B. According to the simulations, heterotrophic bacteria exhibit higher 497 concentrations than nitrifying bacteria as they vary between 0 and 80 g_{het}•m⁻³, whereas nitrifying 498 499 bacteria range from 0 to 10 g_{nit}•m⁻³. Regarding heterotrophic bacteria, the strong fluctuations 500 observed could be explained by the large variability in the COD concentration in the influent (100 -600 $qO_2 \cdot m^{-3}$). This variability arises from the use of two different types of water sources, one from 501 502 the University and the other from the city, with the latter typically containing a higher organic matter 503 content. The concentration over the months of the nitrifying bacteria was lower, considering their 504 slower maximum growth rate compared to the one of heterotrophic bacteria. Results show that the 505 concentration of nitrifying bacteria increased from October to November. This increase may be due to a reduction in the aeration rate in this specific period, which decreased from 200 L•min⁻¹ (set-point 506 in normal operations) to 50 L•min⁻¹. During that months, the dissolved oxygen concentration 507 508 increased in the culture, as the aeration was insufficient to remove it efficiently. Previous studies have shown that dissolved oxygen concentration strongly influences the growth of microalgae, as it 509 has an inhibitory effect on photosynthetic activity (Rossi et al., 2020a). Thus, by decreasing the 510 concentration of microalgae, nitrifying activity is favoured, as both populations compete for the N-511 NH₄⁺ present in the medium. Previous authors suggested that competition for N-NH₄⁺ is the most 512

frequently negative interaction between microalgae and AOB (Aparicio et al., 2022b). The microalgal 513 biomass concentration is represented in Figure 3C. Algal productivity is primarily influenced by 514 variations in light and temperature throughout the seasons (Muñoz and Bernard, 2021). During 515 spring, the biomass concentration is approximately 0.7 g_{alo} ·L⁻¹, while in summer it can reach higher 516 values of up to 1.5 g_{alg}•L⁻¹. However, during the colder seasons, it decreases to 0.3 g_{alg}•L⁻¹. Overall, 517 the model effectively captures the evolution of biomass concentration, highlighting the prevalence of 518 algae biomass compared to bacterial biomass within the culture. Although there is a lack of 519 520 experimental data on bacterial concentration, this outcome remains reasonable, supported by analysis conducted in previous studies on similar systems (Sánchez Zurano et al., 2020). 521

Figure 4 represents the PO₄³⁻ NO₃, BSMO and NH₄⁺ concentration in g•m⁻³ respectively. In Figure 522 4A it is shown that the phosphate concentration in the culture can vary between 10 and 60 g_{PO4}•m⁻³. 523 The uptake of this component depends only on the activity of microalgae, given that the influence of 524 bacteria can be considered negligible. In previous studies, it has been demonstrated that the 525 consumption of phosphate is efficient, but not sufficient to lower it to a concentration below the 526 minimum required for the waster discharge (Nordio et al., 2023). Figure 4B represents the NO₃⁻ 527 528 concentration, that remained constant from May to October, however, from October to November, it 529 was observed an increase in the NO₃ concentration, mainly generated during the nitrification process by nitrifiers. The concentration of this compound can vary greatly (between 0 and 300 g_{NO3}•m⁻³) 530 depending on the activity of the nitrifying bacteria, which, as already explained, was enhanced at the 531 532 end of the study. Nitrate increase and accumulation in the system can be considered one of the main 533 challenges in microalgae-bacteria-based systems. A decrease in microalgae activity leads to an 534 increase in nitrifying activity, which results in the accumulation of nitrate in the medium. The nitrate generated must be consumed by the microalgae. However, as long as ammonium is available in the 535 536 medium, it will not be consumed or will be consumed slowly, in breach of discharge regulations. 537 Therefore, ensuring correct microalgae activity is essential to achieve treated water at the end of the 538 process. Regarding the organic matter, its degradation is due to the activity of the heterotrophic bacteria and it can be present with a concentration of up to 250 g_{O2} •m⁻³ in the culture (Figure 4C). 539 Finally, in Figure 4D there is the evolution of the ammonium concentration. It is possible to observe 540

that, despite the high NH_4^+ concentration entering the system with the wastewater, in the outlet its concentration is mostly lower than 60 g_{NH4} •m⁻³, meaning that microalgae and nitrifiers can uptake this nutrient with a high efficiency. The peak generated by the simulation is mainly due to the high concentration of this compound entering the system during the dilution/harvesting process at a specific time of the day.

546 Concluding, Figure 5 represents the simulation of the dissolved oxygen in the culture. Specifically, 547 Figure 5A shows the experimental and simulated values of dissolved oxygen concentration along 548 the entire study period, while Figure 5B shows the representation of the dissolved oxygen 549 concentration in a shorter period. The concentration of dissolved oxygen can reach a high 550 concentration during the day due to the microalgal photosynthesis (up to 25 mg_{O2}•L⁻¹), while it 551 decreases to anoxic conditions during the night due to the couple effect the algae and bacteria 552 respiration.

The results obtained show that the evolution of nutrients in the system together with the simulated biomass concentration agree with those obtained in the experimental data, demonstrating the usefulness of ABACO-2 in microalgae-based systems for wastewater treatment, and its potential on an industrial scale.

6.2. Case study: evaluating microalgae-bacteria consortia as function of the operational conditions

Studying the populations living in wastewater systems treated with microalgae poses a significant 559 challenge, primarily because there are no fully validated protocols to effectively differentiate between 560 bacterial and microalgal communities. The primary method for assessing biomass in raceway 561 reactors is dry weight, encompassing contributions from both algae and bacteria. Separating them 562 remains challenging yet significant, as their ratios impact various process outcomes, such as 563 biomass quality and water remediation efficiency. Some methods, like successive filtrations based 564 on cell size differences, have been explored, though they often result in a notable presence of 565 bacteria clinging to microalgae due to cell aggregation (Sánchez-Zurano et al., 2020). Alternative 566 methods, including flow cytometry techniques (FCM), prove valuable in assessing the relative 567

composition of mixed microorganism populations, encompassing both prokaryotes and eukaryotes. 568 569 This approach discriminates between groups by analyzing intrinsic characteristics of individual cells, such as size, complexity, and autofluorescence. Additionally, molecular identification techniques like 570 571 amplification of 16S and 18S rDNA sequences serve to evaluate microbial community structure (Barreiro-Vescovo et al., 2021). Alongside these methods, photo-respirometry, based on traditional 572 respirometry, has been employed to discern population differences (Rossi et al., 2018). However, 573 these methods lack a direct correlation in biomass concentration $(q \cdot L^{-1})$, which is more 574 straightforward to interpret. 575

In this context, mathematical models offer a useful tool of indirectly study how the balance between 576 577 populations evolves. Operational conditions, notably cultivation height, dilution/harvesting strategy, and oxygen removal capacity, exibilit a substantial influence. This section presents a case study 578 employing the ABACO-2 model to assess how the proportion between algae and bacteria shifts 579 based on the operational conditions. Simulations have been carried out using the same solar 580 radiation and temperature registered for the validation of the model. On the contrary, the inlet values 581 of nutrient concentration have been maintained constant (as a average values measured in the 582 583 wastewater medium) in order to avoid their influence in the evaluation of the process conditions (180 mg·L⁻¹ NH₄⁺, 30 mg·L⁻¹ PO₄³⁻, 80 mg·L⁻¹ BSMO, 3.4 mg·L⁻¹ NO₃⁻). 584

585 6.2.1. Culture height

The cultivation height is one of the fundamental parameters to consider when operating racewaytype reactors as it significantly influences the penetrative capacity of light within the cultivation. It has been demonstrated that light reaches the cells only in the first three centimeters of culture, while the rest remains in a state of darkness due to an effect of autoshading, and, therefore, photosynthetically inactive. Furthermore, light penetration depends on other factors, such as the extinction coefficient (K_a), which can vary from cultivation to cultivation and depends on the property of microalgae to scatter the received light (Barceló-Villalobos et al., 2019).

In general, facilities that aim to treat large quantities of water prefer to operate at a rather high culture
height, around 30 cm. However, this could be a disadvantage in terms of producing high-quality

biomass, as the dark zone favors the growth of bacterial populations over phototrophic ones. Figure 595 596 6 A,B and C show the populations varying with cultivation height among the seasons (summer, spring and autumn). As expected, at 8 cm, the algal productivity is favored at the expense of the 597 598 amount of treated water, with the maximum concentration reached in spring as it is the period when the maximum irradiance is reached (Figure 6A). The possibility of being able to increase productivity 599 by reducing the culture height has already been studied through the development of new reactors 600 called "thin layers" that operate at around 2 cm; they have also been tested for treating wastewater 601 602 in previous studies (Morillas-España et al., 2021a). On the contrary, microalgal concentration 603 decreases as much as the culture height increases, mitigating the effects related to the different 604 seasons, in favour of the proliferation of the bacterial population (Figure 6B,C). This is particularly evident in the case of nitrifiers. An increased water depth and greater availability of ammonium, 605 606 owing to a reduced uptake capacity by microalgae, enable an increase in their activity. Previous 607 works demonstrated that the depth of the culture has a striking effect on the composition of the 608 microalgae-bacteria consortia, specially in relative abundance of nitrifiers (Figure 6C). Remarkably, 609 at an 8 cm depth where algae activity is more significant, ammonium consumption mainly occurs 610 within the algae population, resulting in minimal involvement of nitrifying bacteria. Conversely, 611 variations in cultivation height appear to have little effect on the concentration of heterotrophic bacteria (Figure 6B). This is because they do not compete with microalgae for nutrient uptake like 612 nitrifying bacteria; their primary substrate is organic carbon. 613

614 In conclusion, the choice of the optimal culture height is of relevance and its regulation mainly depend 615 on the specific goals of the remediation process. These goals may involve achieving high-quality, 616 productive biomass or ensuring efficient water treatment on a larger scale. While a lower culture height is generally recommended to facilitate light penetration, further studies should be carried to 617 618 avoid heat accumulation and the cellular death. Moreover, it's essential to recognize that significantly 619 increasing the culture depth could enhance the activity of nitrifying bacteria and lead to the accumulation of NO₃⁻ within the reactor. Based on the outcomes of this simulation study, a culture 620 height of 15 cm appears as a good compromise between treated water quantity, productivity, and 621 balance between algal and bacterial populations. 622

6.2.2. Dilution/harvesting rate

The harvesting/dilution factor is defined as the reverse of the hydraulic retention time (in day⁻¹) and 624 625 it plays a crucial role in microalgae industrial production. When deciding on the best approach to 626 adopt, key considerations include the optimal dilution rate and its application method - whether through continuous or semi-continuous mode. In the first case, dilution occurs gradually over the 12 627 hours of daylight, while in the latter, both dilution and harvesting take place at specific moments of 628 the day. Figure 6 D, E and F illustrate how varying the dilution rate can impact the productivity and 629 630 composition of algal-bacterial populations, assuming to operate the reactor in a semi-continous mode. The optimal dilution rates for algae range between 0.15 to 0.2 day⁻¹, exceeding these rates 631 leads to a significant decrease in productivity (Figure 6D). This finding is consistent with previous 632 research in similar raceway reactors, which suggested a fixed optimal dilution rate of 0.2 day⁻¹ 633 throughout the year (Morillas-España et al., 2020). Concerning the bacterial population, it is evident 634 that nitrifying activity decreases significantly beyond a dilution rate of 0.2 day⁻¹. This phenomenon, 635 known as "culture washout," indicates that the growth rate of nitrifying bacteria lags behind the 636 dilution rate, potentially affecting process effectiveness (Figure 6F). Conversely, a contrasting 637 positive impact is observed on heterotrophic bacteria with increased dilution rates (Figure 6E). This 638 639 could be attributed to their rapid growth rate and the enhanced organic matter influx, promoting proliferation. 640

In summary, the simulations presented in this study underscore the critical importance of selecting 641 642 the optimal dilution factor for the efficient operation of raceway reactors. Traditionally, dilution factors are determined experimentally through batchs, which establish the maximum growth rate and 643 consequently the dilution rate. Alternatively, some studies have suggested comparing reactors at 644 different dilutions operated in parallel. While these methods are effective, they can be laborious and 645 time-consuming. Furthermore, conventional laboratory techniques primarily evaluate productivity in 646 647 terms of dry weight without delving into the dynamics of the populations involved. Therefore, our 648 research demonstrates that a simulation-based approach offers a viable alternative for identifying 649 the most effective dilution and harvesting strategies.

650 6.2.3. Air desorption

The concentration of oxygen in a microalgae culture is crucial in biological systems. When oxygen 651 652 levels surpass air saturation, they can induce inhibitory effects due to the diffusion of dissolved 653 oxygen through microalgae membranes, resulting in oxidative stress on the cells. This inhibitory impact becomes more pronounced with prolonged exposure to elevated oxygen levels (Antonino 654 Baez and Joseph Shiloach, 2014). Microalgae, as photosynthetic organisms, generate oxygen while 655 consuming carbon dioxide, leading to its accumulation in the culture. Studies, such as the one 656 657 conducted by Rossi et al. in 2020, have shown that excessive oxygen accumulation in cultures can hinder microalgal cell growth (Rossi et al., 2020b). However, there is a scarcity of research assessing 658 how oxygen levels actually influence productivity on a large scale. The prevailing assumption is that 659 oxygen is naturally removed through channels or within the paddlewheel zone. Nonetheless, our 660 experimental data presented herein emphasize the necessity of air injection to reduce oxygen levels 661 in the reactor. 662

Figure 6G, H, and I illustrate the simulation results in scenarios both without air injection and with a 663 Kla equal to 110 h⁻¹. Once again, it is evident that forced air input is advantageous in promoting algal 664 population production compared to bacterial populations (Figure 6G). Specifically, when air is not 665 removed, microalgal concentration drastically decreases by 65%, underscoring the significance of 666 air injection in the reactor to lower dissolved oxygen levels. Concerning the bacteria, it's notable that 667 air injection predominantly affects nitrifying population, with heterotrophic ones being less affected. 668 Heterotrophic batteries exhibit a slight decrease in concentration when air isn't injected. Conversely, 669 nitrifying batteries show significant growth when oxygen isn't removed from the reactor using 670 compressed air. These results align with the model validation discussed earlier and may offer an 671 explanation for occasional NO⁻³ accumulations in cultures. As previously explained, validation data 672 indicated increased nitrification activity when air injection was reduced from 200 L min⁻¹ to 50 L min⁻¹ 673 674 ¹, unfavoring algal growth but favoring nitrification. These simulations confirm this trend, although 675 further experimental studies are needed for confirmation.

676 In conclusion, these simulations reaffirm the critical significance of oxygen removal from the reactor,

not only for enhancing productivity but also for ensuring effective water remediation.

678 Conclusions

679 ABACO-2 is a comprehensive model for microalgae-bacteria consortia in wastewater systems. The model was calibrated and validated in a pilot-scale wastewater treatment reactor, exposed to 680 environmental changes and fed with real urban wastewater (with daily changes in the concentration 681 682 of nitrogen, phosphorus and organic matter), over a long period (May-November). The model allowed to predict the biomass, dissolved oxygen and nutrient concentration evolution with high 683 684 accuracy. Overall, the use of the ABACO-2 model's relative simplicity allows for good predictions 685 while offering advantages in terms of understanding, practicality, efficiency, and versatility. Concluding, the ABACO-2 model can be considered a useful biological model for the description of 686 algae-bacteria in wastewater systems. In order to increase the robusteness, in the future it will be 687 688 necessary to carry out additional validation studies in several data set (accounting for different climatologies and wastewater types) and in higher industrial scales. 689

690 **Credit authorship contribution statement**

R. Nordio: Investigation, Formal Analysis, Validation, Software, Data Curation, Visualization, Writing
– Original Draft; E. Rodríguez-Miranda: Investigation, Formal Analysis, Validation, Software, Data
Curation, Writing – Original Draft; F. Casagli: Formal analysis, Validation; A. Sanchez-Zurano:
Investigation, Writing – Original Draft; J. L. Guzmán: Conceptualization, Supervision, Writing Review & Editing, Funding Acquisition; G. Acien: Conceptualization, Supervision, Writing - Review
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857 Tables

858 Table 1.- ABACO-2 model process rates

| n. | Process | Process rate [g•m ⁻³ •day ⁻¹] |
|----|-------------------------------|--|
| 1 | Microalgae growth on NH4 | $\mu_{max,alg} \cdot \mu(I_{av}) \cdot \overline{\mu(T)} \cdot \overline{\mu(pH)} \cdot \overline{\mu(O_2)_{alg}} \cdot \overline{\mu(N - NH_4)} \cdot \overline{\mu(P - PO_4)} \cdot X_{alg}$ |
| 2 | Microalgae growth on NO3 | $\mu_{max,alg} \cdot \mu(I_{av}) \cdot \overline{\mu(T)} \cdot \overline{\mu(pH)} \cdot \overline{\mu(O_2)_{alg}} \cdot \overline{\mu(N - NO_3)} \cdot \overline{\mu(P - PO_4)} \cdot \left(1 - \overline{\mu(N - NH_4)}\right) \cdot X_{alg}$ |
| 3 | Microalgae decay | $m \cdot X_{alg}$ |
| 4 | Nitrifying bacteria growth | $\mu_{max,nit} \cdot \overline{\mu_{nit}(T)} \cdot \overline{\mu_{nit}(pH)} \cdot \overline{\mu_{nit}(O_2)} \cdot \overline{\mu_{nit}(N - NH_4)} \cdot X_{nit}$ |
| 5 | Nitrifying bacteria decay | $	heta_{nit} \cdot m_{nit} \cdot X_{nit}$ |
| 6 | Heterotrophic bacteria growth | $\mu_{max,het} \cdot \overline{\mu_{het}(T)} \cdot \overline{\mu_{het}(pH)} \cdot \overline{\mu_{het}(O_2)} \cdot \overline{\mu_{het}(N - NH_4)} \cdot \overline{\mu_{het}(BSMO)} \cdot X_{het}$ |
| 7 | Heterotrophic bacteria decay | $	heta_{het} \cdot m_{het} \cdot X_{het}$ |

859

860 Table 2.- ABACO-2 yields matrix

| _ | Component \rightarrow i | SNH4 | S _{NO3} | SPO4 | SBSMO | S _{O2} | Xalg | X _{nit} | Xhet |
|---|--------------------------------------|--------------------------------------|--------------------------------------|--------------------------------------|---------------------------------------|-------------------------------------|--------------------------------------|---|--------------------------|
| j | Process ↓ | [g _{NH4} •m ⁻³] | [g _{NO3} •m ⁻³] | [g _{PO4} •m ⁻³] | [g _{BSMO} •m ⁻³] | [g _{O2} •m ⁻³] | [g _{alg} •m ⁻³] | [g _{nit} •m ⁻ ³] | [ghet •m ⁻³] |
| 1 | Microalgae growth on NH4 | $-Y_{NH_4,alg}$ | | $-Y_{PO_4,alg}$ | | $+Y_{O2,alg}$ | 1 | - | |
| 2 | Microalgae growth on NO ₃ | | $-Y_{NO_3,alg}$ | $-Y_{PO_4,alg}$ | | $+Y_{02,alg}$ | 1 | | |
| 3 | Microalgae decay | | | | $1 - f_{alg}$ | $-Y_{O2,alg}$ | -1 | | |
| 4 | Nitrifying bacteria growth | $-Y_{NH_4,nit}$ | $+Y_{NO_3,nit}$ | | | $-Y_{O2,nit}$ | | 1 | |
| 5 | Nitrifying bacteria decay | | | | $1 - f_{bac}$ | | | -1 | |
| 6 | Heterotrophic bacteria growth | $-Y_{NH_4,het}$ | | | $-Y_{BSMO}$ | $-Y_{O2,het}$ | | | 1 |
| 7 | Heterotrophic bacteria decay | | | | $1 - f_{bac}$ | | | | -1 |

861

862

863 Table 3.- Stoichiometric coefficients.

| Parameter | Value | Units | Source | |
|-------------------|-------|-------------------------------|----------------------|---|
| $Y_{NH_4,nit}$ | 7.9 | $g_{NH_4} \cdot g_{nit}^{-1}$ | (Henze et al., 2000) | - |
| $Y_{NO_3,nit}$ | 26.7 | $g_{NO_3} \cdot g_{nit}^{-1}$ | (Henze et al., 2000) | |
| $Y_{NH_4,het}$ | 0.16 | $g_{NH_4} \cdot g_{het}^{-1}$ | (Henze et al., 2000) | |
| Y _{BSMO} | 2.3 | $g_{BSMO} \cdot g_{het}^{-1}$ | (Henze et al., 2000) | |

| Y _{O2,alg} | 1.33 | $g_{O_2} \cdot g_{alg}^{-1}$ | |
|---------------------|-------|------------------------------|-------------------------|
| Y _{02,nit} | 12.44 | $g_{O_2} \cdot g_{nit}^{-1}$ | (Henze et al., 2000) |
| Y _{02,het} | 0.4 | $g_{O_2} \cdot g_{het}^{-1}$ | (Henze et al., 2000) |
| f _{alg} | 0.1 | - | (Solimeno et al., 2019) |
| f _{bac} | 0.1 | - | (Solimeno et al., 2019) |
| $	heta_{het}$ | 1.07 | °C | (Casagli et al., 2021) |
| $	heta_{nit}$ | 1.1 | °C | (Casagli et al., 2021) |
| | | | |

865 Table 4.- Microalgae kinetic parameters.

| Parameter | Value | Units | Source |
|-----------------------|----------|-----------------------------------|-------------------------------|
| Microalgae kinetic pa | rameters | | |
| I _K | 168 | $\mu E \cdot m^{-2} \cdot s^{-1}$ | (Sánchez Zurano et al., 2021) |
| n | 1.7 | - | (Sánchez Zurano et al., 2021) |
| K _a | 0.08 | $m^2 g^{-1}$ | This study |
| I_{K_r} | 134 | $\mu E \cdot m^{-2} \cdot s^{-1}$ | (Sánchez Zurano et al., 2021) |
| n_r | 1.4 | - | (Sánchez Zurano et al., 2021) |
| T _{min,alg} | -10 | °C | (Casagli et al., 2021) |
| T _{max,alg} | 38 | °C | (Casagli et al., 2021) |
| $T_{opt,alg}$ | 20 | °C | (Casagli et al., 2021) |
| $pH_{min,alg}$ | 1.8 | - | (Sánchez Zurano et al., 2021) |
| $pH_{max,alg}$ | 12.9 | _ | (Sánchez Zurano et al., 2021) |
| $pH_{opt,alg}$ | 8.5 | _ | (Sánchez Zurano et al., 2021) |
| S _{02,max} | 22.68 | $g_{O_2} \cdot m^{-3}$ | (Sánchez Zurano et al., 2021) |
| Ζ | 4.15 | _ | (Sánchez Zurano et al., 2021) |
| $K_{S,NH_4,alg}$ | 1.98 | $g_{NH_4}\cdot m^{-3}$ | (Zurano et al., 2021) |
| $K_{i,NH_4,alg}$ | 734 | $g_{_{NH_4}}\cdot m^{-3}$ | (Zurano et al., 2021) |
| $n_{NH_4,alg}$ | 2 | - | (Zurano et al., 2021) |

| $K_{s,NO_3,alg}$ | 12.26 | $g_{NO_3} \cdot m^{-3}$ | (Zurano et al., 2021) |
|--------------------------|-------------------|-------------------------|-------------------------------|
| $K_{i,NO_3,alg}$ | 1713 | $g_{NO_3} \cdot m^{-3}$ | (Zurano et al., 2021) |
| $n_{NO_3,alg}$ | 2 | _ | (Zurano et al., 2021) |
| $K_{S,PO_4,alg}$ | 1.31 | $g_{PO_4}\cdot m^{-3}$ | (Zurano et al., 2021) |
| Heterotropic bacteria | kinetic parameter | S | |
| T _{min,het} | -3 | °C | (Casagli et al., 2021) |
| T _{max,het} | 42 | °C | (Casagli et al., 2021) |
| $T_{opt,het}$ | 25 | °C | (Casagli et al., 2021) |
| $pH_{min,het}$ | 6 | _ | (Sánchez Zurano et al., 2021) |
| $pH_{max,het}$ | 12 | _ | (Sánchez Zurano et al., 2021) |
| $pH_{opt,het}$ | 9 | _ | (Sánchez Zurano et al., 2021) |
| $K_{s,O_2,het}$ | 1.98 | $g_{0_2} \cdot m^{-3}$ | (Sánchez Zurano et al., 2021) |
| $K_{S,NH_4,het}$ | 0.64 | $g_{NH_4}\cdot m^{-3}$ | (Henze et al., 2000) |
| $K_{S,BSMO,het}$ | 0.299 | $g_{BMSO} \cdot m^{-3}$ | (Henze et al., 2000) |
| Nitrifying bacteria kine | etic parameters | | |
| T _{min,nit} | -8 | °C | (Casagli et al., 2021) |
| T _{max,nit} | 38 | °C | (Casagli et al., 2021) |
| T _{opt,nit} | 20 | °C | (Casagli et al., 2021) |
| $pH_{min,nit}$ | 2 | _ | (Sánchez Zurano et al., 2021) |
| $pH_{max,nit}$ | 13.4 | _ | (Sánchez Zurano et al., 2021) |
| $pH_{opt,nit}$ | 9 | _ | (Sánchez Zurano et al., 2021) |
| $K_{s,o_2,nit}$ | 1.080 | $g_{0_2} \cdot m^{-3}$ | (Henze et al., 2000) |
| $K_{s,o_2,nit}$ | 104.9 | $g_{0_2} \cdot m^{-3}$ | (Henze et al., 2000) |
| $K_{S,NH_4,nit}$ | 1.28 | $g_{NH_4}\cdot m^{-3}$ | (Henze et al., 2000) |

| 867 | Table 5 Microalgae nutrient yield parameters. | |
|-----|---|--|
|-----|---|--|

| Parameter | Value | Units | Source | Parameter | Value | Units | Source | Parameter | Value | Units | Source |
|-----------------------|-------|------------------------------|--------------------|-----------------------|-------|--------------------------|--------------------|-----------------------|-------|-------------------------------|--------------------|
| Y _{NH4} ,max | 0.77 | $g_{\rm NH_4} \cdot g_{alg}$ | Calibrated | Y _{NO3} ,max | 0.44 | $g_{NO_3} \cdot g_{alg}$ | Calibrated | Y _{PO4} ,max | 0.001 | g_{PO_4} . g_{alg} | Calibrated |
| K_{S,NH_4} | 32 | $g_{_{NH_4}} \cdot m^{-3}$ | | K_{S,No_3} | 141 | $g_{NO_3} \cdot m^{-3}$ | | К _{S,PO4} , | 10 | g_{PO_4} $\cdot m^{-3}$ | |
| t_{NH_4} | 2 | - | (Zuran | t_{NO_3} | 2 | - | (Zurano et al., | t _{PPO4,} | 2.14 | • m • | (Zurano et al., |
| NH _{4max} | 102 | $g_{_{NH_4}} \cdot m^{-3}$ | o et al., 2021) | NO _{3max} | 102 | $g_{NO_3} \cdot m^{-3}$ | 2021) | PO _{4max} | 69 | g_{PO_4} $\cdot m^{-3}$ | 2021) |
| NH_{4min} | 12 | $g_{_{NH_4}} \cdot m^{-3}$ | | NO _{3min} | 12 | $g_{NO_3} \cdot m^{-3}$ | | PO_{4min} | 6 | $g_{\scriptscriptstyle PO_4}$ | |
| NH _{4opt} | 71 | $g_{_{NH_4}} \cdot m^{-3}$ | | NO _{3opt} | 71 | $g_{NO_3}\cdot m^{-3}$ | | PO _{4opt} | 47 | $\cdot m^{-3}$ g_{PO_4} | |
| | | | | | | | | | | $\cdot m^{-3}$ | |

| 870 | Table 6 List of calibrated parameters and their corresponding values |
|-----|--|
| 0.0 | |

| Parameter | Description | Value | Units |
|-----------------------|--|----------------------|-------------------------------|
| $\mu_{max,alg}$ | Maximum algal growth rate | 1.5 | day^{-1} |
| m_{min} | Minimum algal respiration rate | 0.1 | day^{-1} |
| m_{max} | Maximum algal respiration rate | 0.008 | day^{-1} |
| $\mu_{max,nit}$ | Maximum nitrifying bacteria growth rate | 0.75 | day^{-1} |
| m _{nit} | Nitrifying bacteria decay | 0.05 $\mu_{max,nit}$ | day^{-1} |
| $\mu_{max,het}$ | Maximum heterotrophic bacteria growth rate | 3.4 | day^{-1} |
| m_{het} | Heterotrophic bacteria decay | 0.2 $\mu_{max,het}$ | day^{-1} |
| Y _{NH4} ,max | NH4 ⁺ microalage nutrient yield max | 0.6 | $g_{NH_4} \cdot g_{alg}^{-1}$ |
| Y _{NO3} ,max | NO3 ⁻ microalage nutrient yield max | 0.1 | $g_{NO_3} \cdot g_{alg}^{-1}$ |
| $Y_{PO_4,max}$ | $PO_{4^{3-}}$ microalage nutrient yield max | 0.004 | $g_{PO_4} \cdot g_{alg}^{-1}$ |
| Kl _a | O ₂ natural mass transfer | 0.1 | h^{-1} |
| α | Nutrient assimilation coefficient | 1 | _ |

| Parameter | Units | Nominal | Standard | Most affected parameters |
|-----------------------|-----------------------------------|----------------------|--|--|
| | | value | deviation | |
| I_k | $\mu E \cdot m^{-2} \cdot s^{-1}$ | 168 | 0.02 | $X_{alg}, X_{het}, X_{nit}, S_{NH_4}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_O$ |
| K _a | $m^2 g^{-1}$ | 0.08 | 1.4E-5 | $X_{alg}, X_{het}, X_{nit}, S_{NH_4}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_O$ |
| n | _ | 1.7 | 1.6E-4 | $X_{alg}, X_{het}, X_{nit}, S_{NO_3}, S_{PO_4}, S_{O_2}$ |
| $\mu_{max,alg}$ | day^{-1} | 1.5 | 5.38E-4 | $X_{alg}, X_{het}, X_{nit}, S_{NH_4}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_O$ |
| $\mu_{max,het}$ | day^{-1} | 3.4 | 0.0016 | X_{het}, S_{BSMO} |
| $\mu_{max,nit}$ | day^{-1} | 0.75 | 0.0004 | X_{nit} , S_{NO_3} , S_{NH_4} |
| m_{min} | day^{-1} | 0.1 | 8E-6 | $X_{alg}, X_{het}, X_{nit}, S_{NO_3}, S_{BSMO}, S_{O_2}$ |
| m _{nit} | day^{-1} | 0.05 $\mu_{max,nit}$ | 8.34E-5 | $X_{het}, X_{nit}, S_{NO3}S_{BSMO},$ |
| m _{het} | day^{-1} | 0.2 $\mu_{max,het}$ | 6.8E-5 | $X_{het}, X_{nit}, S_{NO3}S_{BSMO},$ |
| 0 _{2,max} | $mg_{O_2} \cdot l^{-1}$ | 22.68 | 0.0011 | $X_{alg}, X_{het}, X_{nit}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_{O_2}$ |
| Y _{NH4} ,max | _ | 0.6 | 0.001 | X_{het} , X_{nit} S_{NH_4} , S_{NO_3} |
| Y_{NH_4nit} | _ | 0.4 | 0.016 | X_{nit} , S_{NH_4} |
| Y _{NO3} ,nit | _ | 26.76 | 0.11 | S _{NO3} |
| Cardinal | Units | Opt, max, | Standard | Most affected parameters |
| parameters | | min | deviation | |
| T_{alg} | °C | 38,20,-10 | 0.0026 | $X_{alg}, X_{het}, X_{nit}, S_{NH_4}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_C$ |
| pH_{alg} | _ | 8.5, 12.9, 1.8 | 0.0032 | $X_{alg}, X_{het}, X_{nit}, S_{NH_4}, S_{NO_3}, S_{PO_4}, S_{BSMO}, S_C$ |
| T _{nit} | °C | 20,38,-8 | 0.04 | X_{het} , X_{nit} S_{NH_4} , S_{NO_3} , S_{BSMO} |
| pH _{nit} | _ | 9, 13.4, 2 | 0.024 | X_{het} , X_{nit} S_{NH_4} , S_{NO_3} , S_{BSMO} |
| T _{het} | °C | 25,42,-3 | 0.034 $X_{het}, X_{nit}, S_{NO_3}, S_{BSMO}$ | |
| pH_{het} | _ | 9, 12, 6 | 0.0026 | $X_{het}, X_{nit}, S_{NO_3}, S_{BSMO}$ |

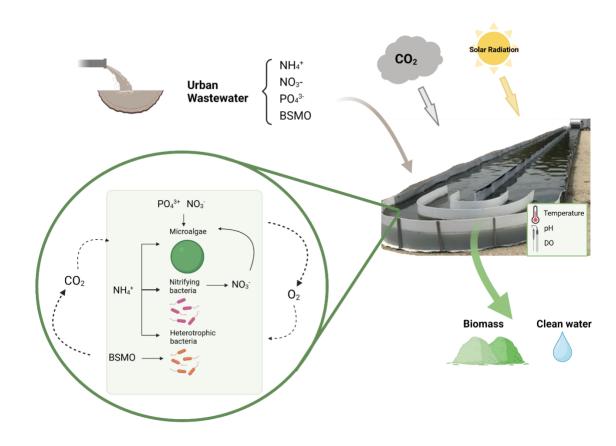
872 Table 7.- List of most sensible parameters

| | 875 | Table 8 Validation errors of the ABACO-2 model. |
|--|-----|---|
|--|-----|---|

| Parameter | NRMSE | TIC |
|--|-------|------|
| $X_{tot} [g \cdot m^{-3}]$ | 0.21 | 0.16 |
| $S_{NH_4} \left[g \cdot m^{-3} \right]$ | 0.22 | 0.55 |
| $S_{NO_3}\left[g\cdot m^{-3} ight]$ | 0.15 | 0.21 |
| $S_{PO_4}\left[g\cdot m^{-3} ight]$ | 0.23 | 0.24 |
| $S_{BSMO} \left[g \cdot m^{-3}\right]$ | 0.21 | 0.2 |
| $S_{O_2}\left[g\cdot m^{-3}\right]$ | 0.14 | 0.21 |
| | | |



877 Figures



878

Figure 1.- Schematic description of the biological mechanisms taking in place in microalgae-bacteria
 wastewater systems.

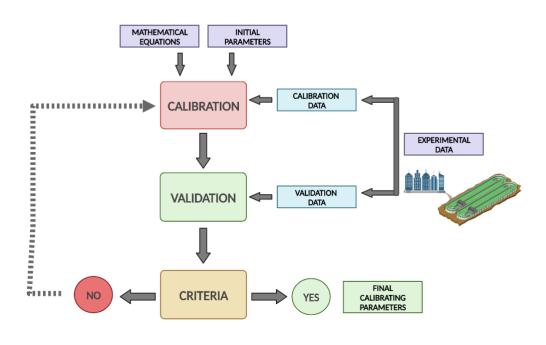






Figure 2.- Calibration methodology used in the present work.

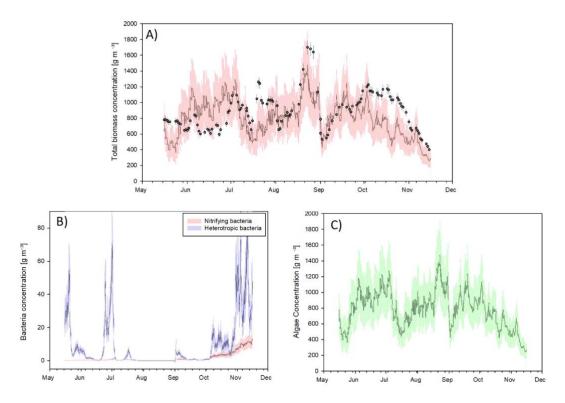


Figure 3.- Biomass concentration evolution. A) Total biomass concentration (sum of algae and bacteria,
 simulated (continuous line) and experimental (scatter plot); B) Nitrifying and heterotropic concentration; C)
 Algae concentration. The model shade is the model confidence interval at 95%.

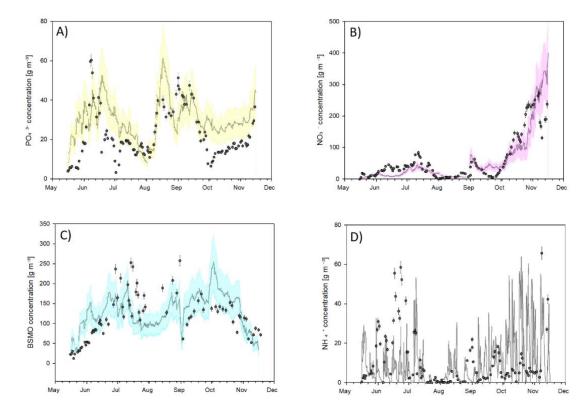


Figure 4.- Variation of culture nutrients A) PO₄³⁻, B) NO₃⁻, C) NH₄⁺ and D) BSMO concentration, in g•m⁻³
 respectively (experimental, scatter plot; model prediction, continuous line).

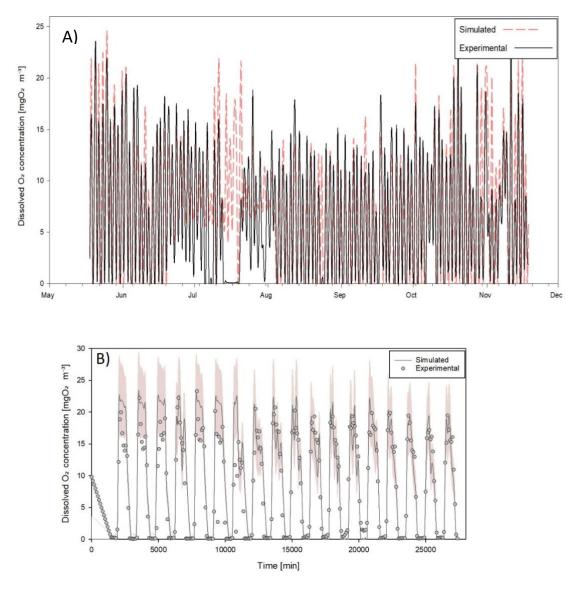
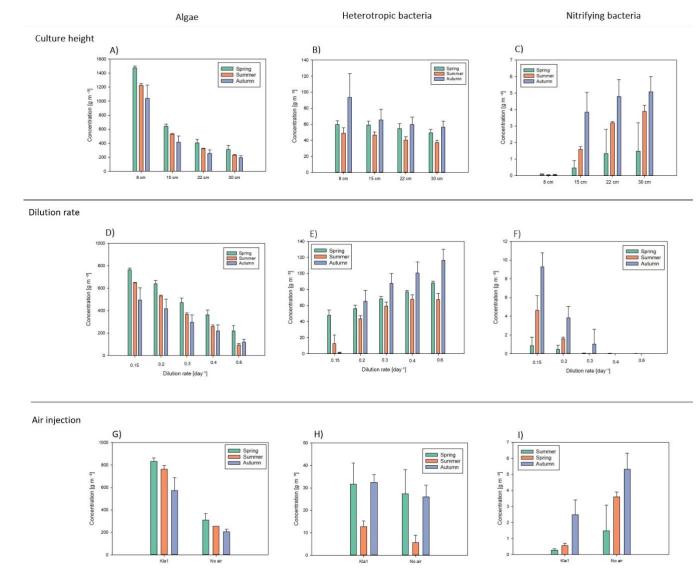


Figure 5.- Simulation of the dissolved oxygen in the culture. A) Experimental and simulated dissolved oxygen along the entire study period, B) Representation of the oxygen in a shorter period.



- 899 Figure 6: Application of ABACO-2 model in a case-study. Algae, heterotropic and nitrifying bacteria
- concentration eveolution depending on: A), B), C) culture height; D), E), F) Dilution rate; G), H), I) with orwithout ijecting air into the system