

1 **Keywords**

2 Mixed crystals, luminescence, lattice parameters, clusterization, GdNbO₄, GdTaO₄

3

4 **1. Introduction**

5

6 Currently the fast and heavy crystals are needed for the application in scintillation detectors
7 [1]. Scintillators with activator emission are commonly used for such purposes. However, crystals
8 with intrinsic emission are preferable in case of cryogenic scintillators when activator emission
9 becomes inefficient due to the self-trapping of charge carriers by the host lattice at low temperature
10 [2,3]. REAO₄ compounds (A = Ta or Nb) demonstrate intrinsic luminescence attributed to the
11 emission of regular TaO₆ - and NbO₆ – groups and defect groups NbO₅V_O and TaO₅V_O (V_O –
12 oxygen vacancy) [4-7]. The brightest luminescence was observed in the rare earth niobates [4]. At
13 the same time, rare-earth tantalates demonstrate an extremely high density and stopping power [8].
14 Engineering of mixed compositions by Nb⁵⁺/Ta⁵⁺ substitution can be a tool to enhance the light
15 yield and other scintillation parameters by analogy with many other mixed systems studied recently
16 [9-16]. It was shown that the tendency to light yield increase in mixed crystals is linked to the
17 spatial distribution of substituted atoms in solid solutions which, in turn, determines their electronic
18 properties. Basing on luminescence properties study it was suggested that spatial inhomogeneities
19 of electronic structure affect the dynamics of relaxation, recombination, and localization of free
20 carriers [9,10,12]. This approach was confirmed by the dependence of light yield improvement in
21 mixed crystals on ionic radii difference between the substituted atoms [17]. Meanwhile, the
22 improvement of charge carriers transport efficiency to activator and a huge light yield increase was
23 linked to enveloping of the electron trap levels in the conduction band in some mixed crystals [18-
24 21].

25 Light output and decay times of the tantalato-niobates [RE(Nb_xTa_{1-x})O₄] strongly depends on
26 the Ta/Nb ratio [22]. Light output of Gd(Ta_{1-x}Nb_x)O₄ at room temperature shows about gradual

1 increase with Nb content. The fast decay component with the τ varied within 5 – 93 ns range is
2 accompanied by the slow component for all compositions except $\text{Gd}(\text{Ta}_{0.8}\text{Nb}_{0.2})\text{O}_4$. Recently, single
3 crystals with the latter composition were successfully grown by the Czochralski technique [23]. In
4 this scintillator the high density of 8.37 g/cm^3 is combined with the light output of 1400 ph./MeV
5 and very fast luminescence with the main decay time of 17 ns.

6 The present study focuses on structural and luminescence properties of $\text{Gd}(\text{Ta}_{1-x}\text{Nb}_x)\text{O}_4$ ($x =$
7 $0 \div 1$), including thermostimulated luminescence (TSL) curves analysis. We also discuss the
8 evidences of clusterization in these solid solutions basing on dependences of various physical
9 properties on the Ta/Nb ratio.

10

11 **2. Experimental**

12 **2.1. Raw materials**

13 Ceramic samples of gadolinium tantalate-based compounds with varying Nb content from 0
14 to 100 at.% were obtained by the solid-state synthesis (details are described in [22]). Gd_2O_3
15 (Stanford Materials Co, USA) and Ta_2O_5 , Nb_2O_5 (Lanhit, Russia) with purity not less than 4N were
16 used as starting materials. XRD data on the obtained samples indicate that all the $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$
17 samples crystallize in the M-fergusonite structure (space group $I 2/a$) with the main phase content of
18 95-97 wt.% [22]. After solid-state synthesis the pellets were prepared for the study of luminescent
19 and scintillation properties by polishing of their surfaces to the identical surface roughness
20 providing the same light collection conditions from all of the samples.

21

22 **2.2. Characterization techniques**

23 The luminescence spectra under excitation with synchrotron radiation (SR) with excitation
24 energy 22 eV have been obtained at the branch-line FINEST at MAX-lab, Lund [24]. The spectra
25 were measured using ARC Spectra Pro 300i monochromator equipped with Hamamatsu H6240-01

1 photon counting head in the temperature range 5 – 300 K. TSL curves were studied in the
2 temperature range $T = 100 - 550$ K. Samples were mounted into LINKAM THMS600 Stage. The
3 spectra were registered using a Shamrock 500i spectrograph equipped with a Newton EMCCD
4 DU970P. The heating rate at TSL measurements was 10 K/min.

5 The luminescence excitation spectra were measured at 300 K in the wavelength range 200-
6 350 nm using the deuterium Heraeus D 200 VUV lamp as an excitation source. The McPherson
7 Model 234/302 primary monochromator was used for the selection of excitation wavelength. The
8 luminescence was registered using the Shamrock 303i (Andor Technology) monochromator
9 equipped with the Hamamatsu H8259 photon counting head.

10

11 **3. Experimental results**

12 **3.1 Cell parameters**

13 Deviations of lattice constants from the Vegard's law were noticed previously in many
14 mixed crystals, e.g. in MgO-FeO, MgO-LiFeO₂ [25], and CdF₂-PbF₂ systems [26], and attributed to
15 the formation of inhomogeneities (clusters) enriched with one of substituted atoms. In Gd(Nb_xTa_{1-x}-
16 _x)O₄ the *b* cell parameter and the elementary cell volume determined recently in [22] demonstrate
17 similar positive deviations at $x = 0.2 - 0.6$ (Fig. 1).

18

19 [Fig. 1]

20

21 **3.2 Luminescence properties**

22 The luminescence spectra of Gd(Nb_xTa_{1-x})O₄ at $T = 300$ K are presented in Fig.2a. The
23 excitation energy of SR was 22 eV providing interband electron transitions. The excitation energy is
24 high enough to model the processes which takes place in a scintillator at the final stages of energy
25 relaxation including the photon multiplication, thermalization of hot charge carriers and their

1 migration to the emission centers. The photon multiplication process starts at $E_{ex} \geq 2E_g$. According
2 to [27] the band gap energies for gadolinium tantalate and niobate are 5.4 eV and 4.6 eV,
3 respectively. Therefore, the excitation energy exceeds the bandgap values of the studied samples by
4 the factor of 4.1-4.8. At $x \neq 0$ all the samples demonstrate a broad emission band with the maximum
5 around 450 nm. It was shown previously [28] that emission in rare earth tantalate-niobates possesses
6 the complex structure and consists of the short-wavelength band peaked at 415 nm and ascribed to
7 the exciton-like emission at regular oxyanion complexes NbO_6 , and the long-wavelength band
8 peaked at 460 nm, which is connected with the point defects, in particular, oxygen vacancies
9 (NbO_5V_O -group). Consequently, the $Gd(Nb_xTa_{1-x})O_4$ emission observed in this work at room
10 temperature can be ascribed to the defect-related emission.

11

12 [Fig. 2]

13

14 The intrinsic emission is not manifested, probably, due to the thermal quenching, or due to
15 efficient transfer from regular to defect states [4]. The broad but weak emission band in $GdTaO_4$
16 (Fig. 2a, curve 1) is red-shifted and linked to TaO_5V_O -group emission [28]. The several sharp peaks
17 at 380, 420, 440, 490, 550, 590 and 610 nm observed for the latter are related to emission of Eu^{3+}
18 and Tb^{3+} impurities. The peak at 310 nm linked with the emission of the cell-forming Gd^{3+} ion and
19 is most pronounced for the $GdTaO_4$. Less intensity of this band for Nb-containing samples should
20 be linked with more efficient energy transfer from Gd^{3+} ion to NbO_5V_O -group.

21 The excitation spectra of the emission at 460 nm are presented in Fig.3.

22

23 [Fig. 3]

24

1 The peak at 295 – 305 nm is located below the bandgap and is connected with the direct
2 excitation of the defect center. The low-energy shift of the bandgap with x is revealed in the
3 corresponding shift of the peak's maximum. The rise of intensity at shorter wavelengths relates to
4 the energy transfer from the region of fundamental absorption edge. The rise is not sharp and it does
5 not allow to trace the bandgap shift. It might be related to the defect origin of the emission center,
6 which is excited via consecutive capture of electrons and holes (recombination type of energy
7 transfer to emission centers, see e.g [29]).

8 The luminescence spectra of $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ ($x \neq 0$) at $T = 5$ K (Fig. 2b) are similar to those
9 at $T = 300$ K with maximum around 450 nm. Therefore, exciton-like emission of regular oxyanion
10 complexes NbO_6 are not observed for gadolinium tantaloniobates at low temperature as well. For
11 GdTaO_4 the broad band with maximum around 395 nm was detected (Fig. 2b, curve 1). In [4] the
12 intrinsic luminescence of GdTaO_4 at $T=77$ K was supposed similar to the defect-related emission in
13 YTaO_4 with maximum around 400-420 nm. Thus, the band observed in the present study (395 nm)
14 can be ascribed to intrinsic emission in GdTaO_4 . Note, this band is completely absent i) on the
15 $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ ($x \neq 0$) luminescence spectra at $T = 5$ K confirming the efficient transfer from
16 regular TaO_6 -groups to defect states and ii) on the GdTaO_4 luminescence spectrum at $T = 300$ K
17 confirming the thermal quenching of intrinsic luminescence.

18

19 **3.3 Thermostimulated luminescence**

20 Thermostimulated luminescence curves (Fig. 4a) provide an information on traps
21 modification in the solid solution. Up to five peaks can be seen at the TSL curves of $\text{Gd}(\text{Ta}_{1-x}\text{Nb}_x)\text{O}_4$.
22 Two TSL peaks are inherent for all samples – the weak one at 120-160 K and the most
23 intensive one at 230 -270 K. Therefore, these peaks relate to the defects, which origin does not
24 connected with Nb or Ta cations. We tentatively attribute them to oxygen vacancies. The shift of
25 these peaks to lower temperatures with increase of x corresponds to the decrease of activation
26 energy of the corresponding trap. The tendency is attributed to the decrease of the mean band gap

1 from GdTaO₄ to GdNbO₄. It is worth noting that the width of TSL peaks depends on x value as
2 well. The FWHM of the most prominent TSL peak at ~250 K depends non-linearly on x with the
3 maximum at x = 0.4 (Fig. 4b).

4

5 [Fig. 4]

6

7 **3.4 Light output.**

8 The efficient transfer from host to emission centers in solid solutions might cause a
9 significant increase of light output for intermediate x values of Gd(Nb_xTa_{1-x})O₄ at T = 16 K (Fig.
10 5a). The similar effect of light output enhancement has been previously observed in some mixed
11 oxide crystals at room temperature [9-16]. Here we demonstrate that the effect can be also observed
12 at low temperatures as well. Therefore, it can be utilized for the tailoring of cryogenic scintillators
13 with the improved scintillation performance. The linear dependence of light output at T = 300 K is
14 connected with the contribution of some additional processes to the light output formation, e.g. of
15 emission thermal quenching, which is observed for all studied crystals (Fig.5b). The temperature
16 dependence of the luminescence intensity can't be approximated by the Mott law for any of the
17 studied sample. Under 22 eV excitation the separated electrons and holes are created in the
18 conduction and valence band. Besides the radiative relaxation of electron-hole pairs at NbO₅V_O
19 emission center the separated charge carriers can also migrate to some other non-radiative
20 relaxation centers or to be captured by the traps. The presence of traps in Gd(Nb_xTa_{1-x})O₄ is
21 confirmed by the TSL curves presented in Fig.3. Relaxation of charge carriers via these competitive
22 to emission channels influences the temperature dependence of emission intensity.

23

24 [Fig. 5]

25

4 Discussion

The detailed consideration of dependences of physical properties of $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ mixed crystals reveals some evidences of spatial inhomogeneity in them, namely:

- loosening of the lattice at $x=0.2-0.4$ means that Ta and Nb are uniformly distributed in lattice in these compositions;

- the broadening of the main TSL peak for intermediate compositions indicates the wider distribution of activation energy of the trap due to the bandgap spatial fluctuations caused by the inhomogeneity in the solid solution;

- finally, the non-linearity for the light output at 16 K and the linear dependence at 300 K supports the hypothesis on correlation between spatial inhomogeneities in scintillation crystals and shortening of the carrier separation length in them [9,10,12].

As charge carriers in crystals are situated in periodic electrostatic field formed by crystal lattice ions, in a mixed crystal the amplitude of crystal field oscillations should increase in case of formation of clusters enriched by one of the substituted ions. The bandgap difference (ΔE_g) between the mixed crystal constituents is also matters, because clusters limit electron and hole separation at $\Delta E_g \gg kT$. This is the case of the considered system where ΔE_g between GdTaO_4 and GdNbO_4 is 0.8 eV [27]). As kT is by 19 times higher at 300 K compared to that at 16 K, the limitation of electron and hole separation at 16 K causes the increase of the light output in the intermediated compositions, while at 300 K no evidence of carrier separation length limitation is observed (possibly, due to thermal quenching of luminescence). Note that the cluster size must be large enough to influence the carrier separation length, which in dielectric crystals is from 1 nm up to several tens of nm depending on the relaxation speed on the stage of phonon scattering.

At the moment few data are available to distinguish the role of preparation procedure itself (solid state synthesis from powders) in cluster formation. According to the specifications, powder particle size of starting oxides used for the solid-state synthesis of ceramic samples is less than $6.6 \cdot 10^{-5}$ m (250 mesh). Since solid-state synthesis is a heterogeneous reaction, interaction took place on the particle surface. For correct description of the synthesis mechanism one should take

1 into account the diffusion mobility of the interacting ions. The diffusion mobility inversely
 2 proportional to the melting point (m.p.) of the compound [30] and increases in the row Gd (Gd₂O₃
 3 m.p. 2420 °C) < Ta (Ta₂O₅ m.p. 1872 °C) < Nb (Nb₂O₅ m.p. 1512 °C). Obviously, the interaction
 4 took place on the Gd₂O₃ particle surface while Ta₂O₅ and Nb₂O₅ were supplied to the interaction
 5 surface by diffusion. However, at the moment there is no available data on size of formed
 6 tantalonioabate grains, as well as Ta- or Nb- enriched clusters in them.

7 Instead, the nanoscale clusters in the considered system can be formed by intrinsic properties
 8 of the solid solution. Note, the band gap in GTO 5.4 eV and GNO 4.6 eV [27] differs by 0.8 eV.
 9 Numerical simulation in a model mixed system A_{1-x}B_xC [31] shows that sufficiently large E_g spatial
 10 fluctuations with the amplitude of up to $0.8 \Delta E_g$ are formed in the case of affinity between AA or
 11 BB atoms, e.g. a high probability to occupy neighboring positions in the lattice by atoms of the
 12 same sort. Meanwhile, at high AA or BB affinity the cluster size reaches hundreds of elementary
 13 cell volumes, while it is negligible in the case of AB affinity. At random distribution of A and B
 14 atoms the calculated E_g and cluster size are intermediate between the AA (BB) and AB affinity
 15 cases and the bandgap spatial fluctuations amplitude reaches $0.35 \Delta E_g$. Meanwhile, physical or
 16 chemical factors affecting such affinity were not discussed in [31].

17 Considering the chemical affinity as the capability of interaction between atoms or
 18 compounds, the Nb-Ta affinity (energy of chemical bond formation between Ta and Nb) in
 19 Gd(Nb_xTa_{1-x})O₄ is low because of similar electronegativities of 1.5 in Ta and 1.6 in Nb by the
 20 Pauling scale [32]. According to the empirical Pauling formula for the evaluation of Δ_{Ta-Nb} as the
 21 energy of chemical bond formation between Ta and Nb relatively to the hypothetical covalent bond

$$0.208 * (\Delta_{Ta-Nb})^{1/2} = | \chi_{Ta} - \chi_{Nb} |$$

22
 23 (χ_{Ta} and χ_{Nb} are electronegativities of Ta and Nb, respectively), Δ_{Ta-Nb} is around 30 kJ/mol.
 24 Accounting for the average chemical bond energy of several hundreds of kJ/mol, the Δ_{Ta-Nb} value is
 25 negligible. As no Nb-Nb or Ta-Ta chemical interaction can be expected too, random distribution of
 26 Nb and Ta in Gd(Nb_xTa_{1-x})O₄ is preferable, unless some unknown factors exist affecting the Nb-Ta

1 affinity, or the affinity between the same sort of atoms. However, the maximum deviations of cell
2 parameters and light output are shifted from $x = 0.5$ (expected maximum deviation at random
3 Ta/Nb distribution) to $x = 0.2 - 0.4$ pointing at some Nb-Nb affinity and formation of Nb-enriched
4 clusters [31]. Nevertheless, deeper structural studies (e.g. EXAFS) are needed to obtain a direct
5 confirmation of clusters formation, and other possible causes of Nb-Nb affinity should be analyzed.

6

7 **Conclusions**

8 The analysis of structure and luminescence properties of $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ mixed crystals on
9 Nb/Ta ratio indicates an inuniformity of substitutional ions distribution over the crystal lattice
10 resulting in the spatial band gap energy fluctuations. Meanwhile, the presence of positive light
11 output deviations just at low temperatures indicates that amplitudes of these fluctuations are higher
12 than the kT energy at ~ 16 K and significantly smaller compared to $E_g = 0.8$ eV between GdTaO_4
13 and GdNbO_4 . Despite no evidence of chemical affinity between Nb-Ta, or Nb-Nb (Ta-Ta) was
14 found, the maximal deviations of lattice parameters, light output at 16 K, and FWHM of the
15 strongest TSL peak from the linearity are observed at $x = 0.2 - 0.6$. This indicates a tendency to
16 formation of Nb enriched clusters over the lattice in $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$.

17

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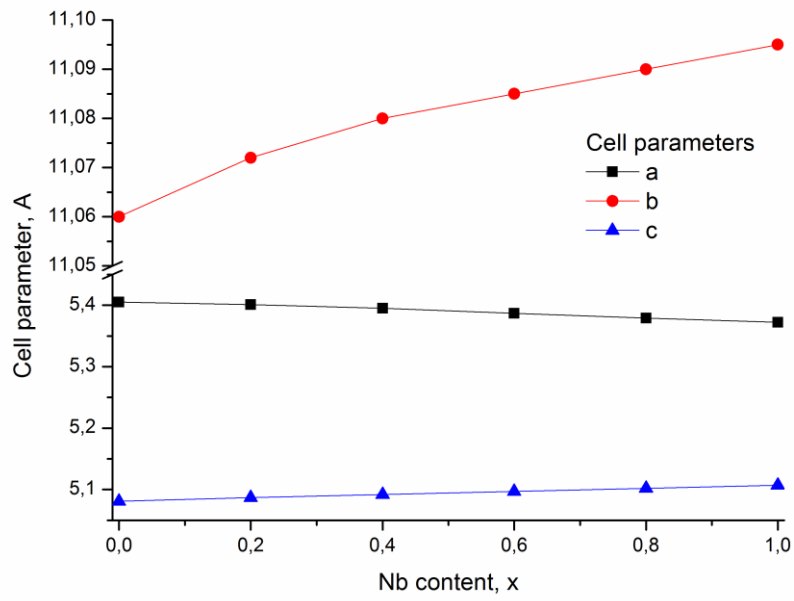
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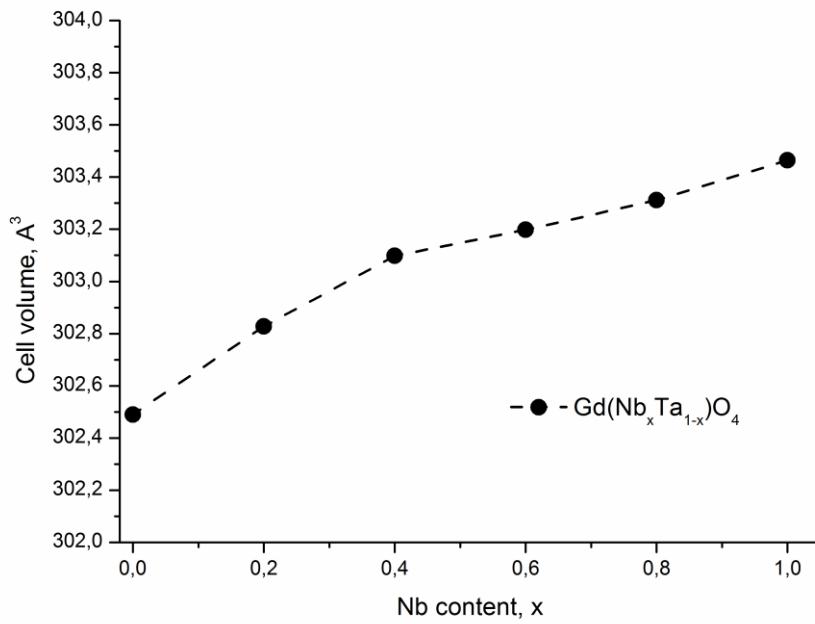
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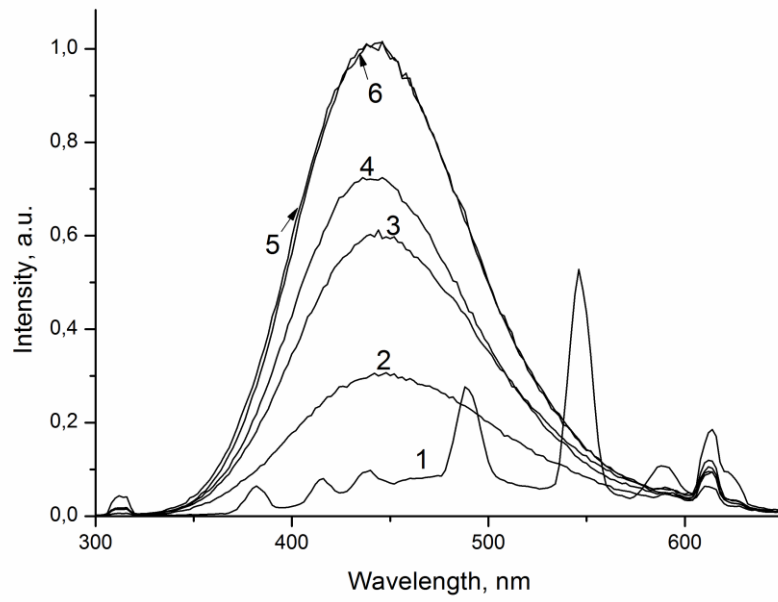


a

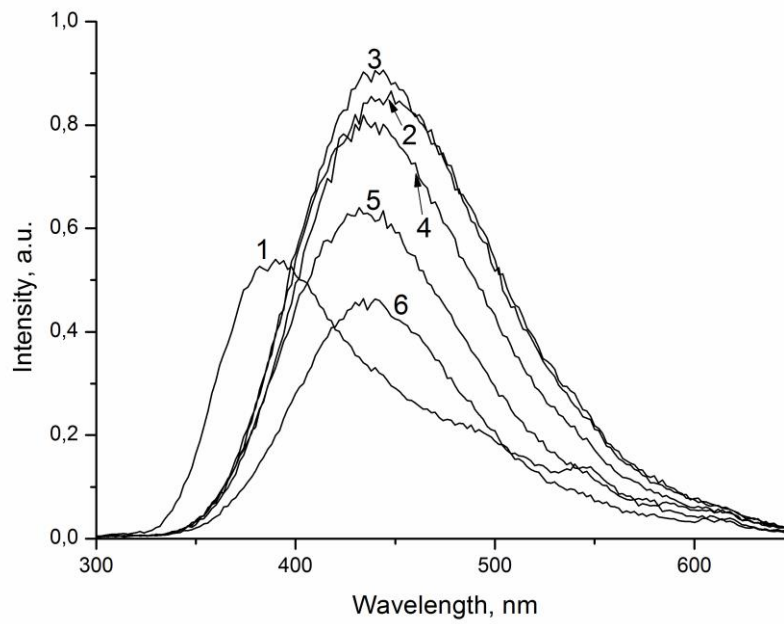


b

Fig. 1 Dependence of the cell parameters (a) and cell volume (b) on the x value for $Gd(Ta_{1-x}Nb_x)O_4$.



a



b

Fig. 2 Emission spectra of $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ ($x=0$ (1); $x=0.2$ (2); $x=0.4$ (2); $x=0.6$ (4); $x=0.8$ (5) and $x=1$ (6)) normalized by the maximum of the brightest curve under synchrotron radiation excitation, $E_{\text{ex}} = 22$ eV at $T = 300$ K (a) and $T = 5$ K (b).

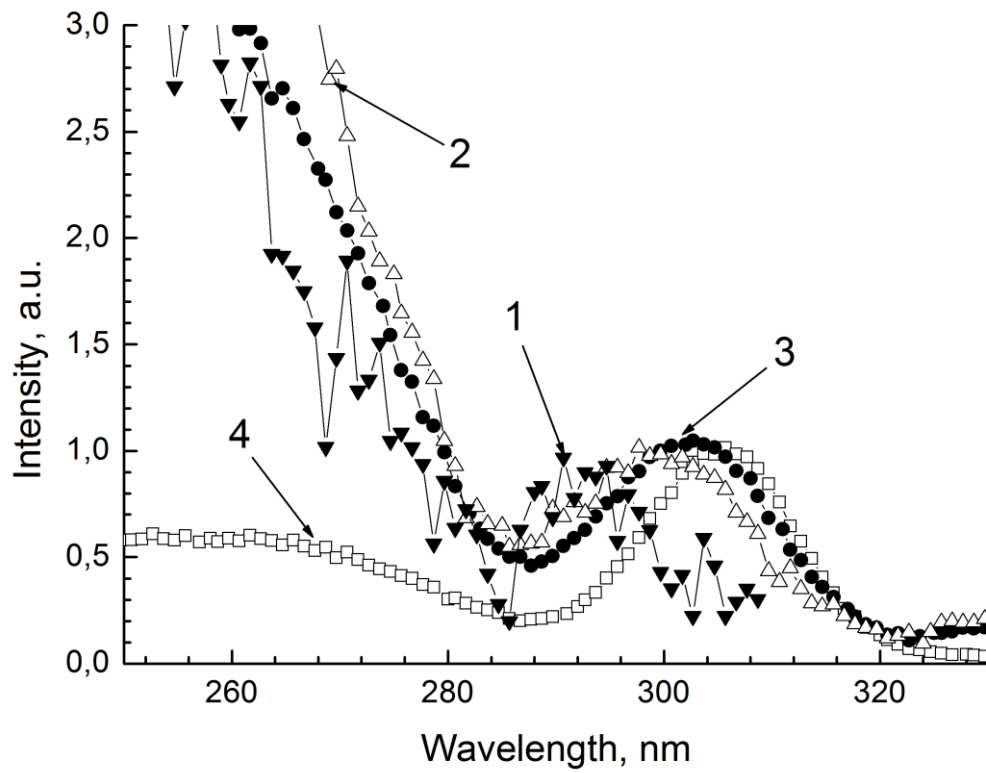
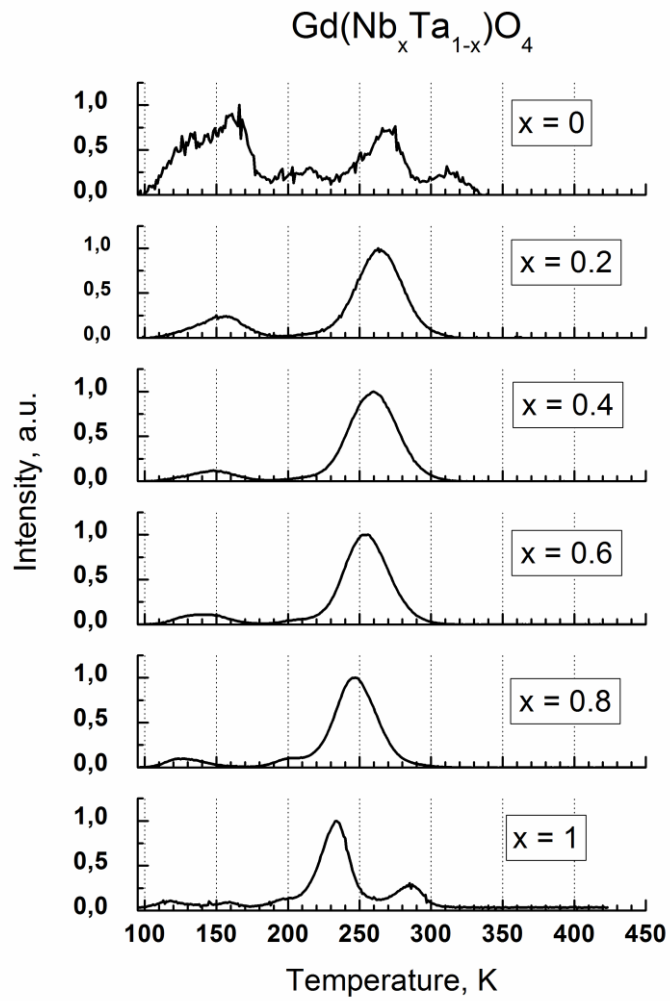
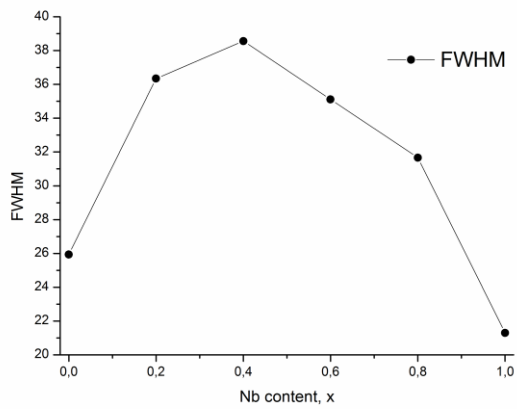


Fig. 3 Luminescence excitation spectra of $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ ($x=0.2$ (1); $x=0.4$ (2); $x=0.6$ (3) and $x=0.8$ (4)) at $\lambda_{\text{em}} = 460$ nm, $T = 300$ K

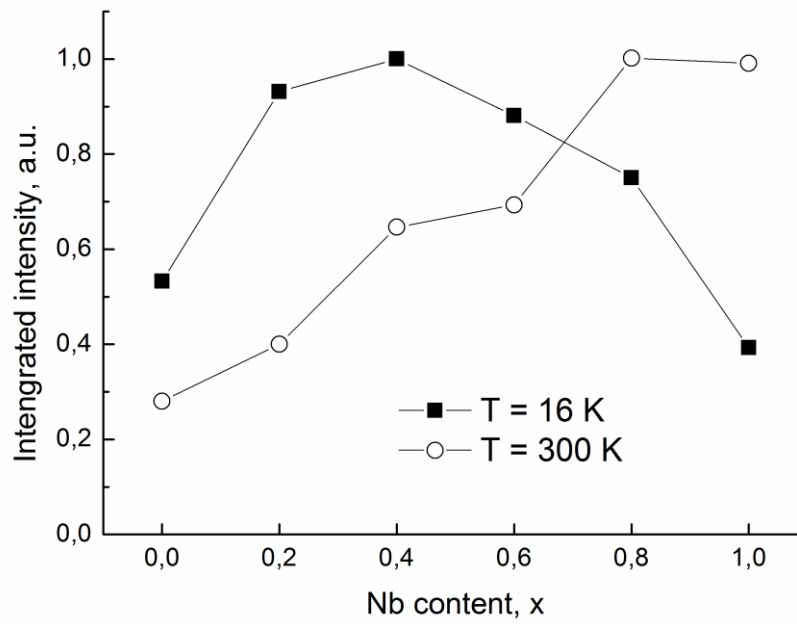


a

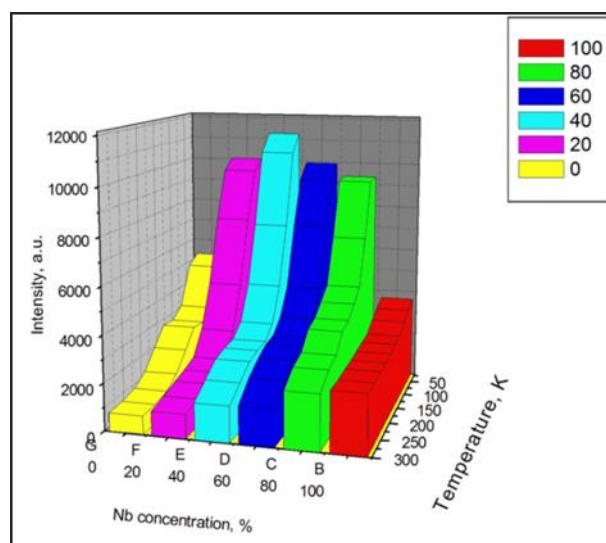


b

Fig. 4 TSL curves (a) and dependence of the ~ 250 K peak FWHM on the x value (b) in $\text{Gd}(\text{Ta}_{1-x}\text{Nb}_x)\text{O}_4$.



a



b

Fig. 5 Dependence of light output in $\text{Gd}(\text{Nb}_x\text{Ta}_{1-x})\text{O}_4$ on x value: a – under SR excitation at temperatures 16 K and 300 K, b – temperature dependence of light output under SR excitation, $E_{\text{ex}}=22$ eV.