

**New cross sections, indices of refraction,
and reflectance spectra of atmospheric interest**

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Abstract

A brief review is presented of the indices of refraction and cross sections that are on the HITRAN 2004 compilation. New measurements of the refractive indices of ternary ($\text{H}_2\text{O}/\text{H}_2\text{SO}_4/\text{HNO}_3$) droplets by Myhre [2003,2005] and the indices of supercooled water by Wagner [2005] are presented and compared to previous data. New infrared cross sections of species of atmospheric interest (e.g., the measurements of HFC-125 and HFC-143a of Lonardo and Masciarelli [2000]) are also discussed.

We finally propose to link to the “other listings” portion of the HITRAN web site several established reflectance data bases, since many current and future remote sensing experiments are influenced by ground emission and reflectance contributions.

Outline of Presentation

Discussion of new

Indices of Refraction

Cross sections

Links to

Digital surface reflectance atlases

HITRAN 2004 Refractive Indices



Water, ice

Sulfuric acid (stratospheric sulfate)

Nitric acid

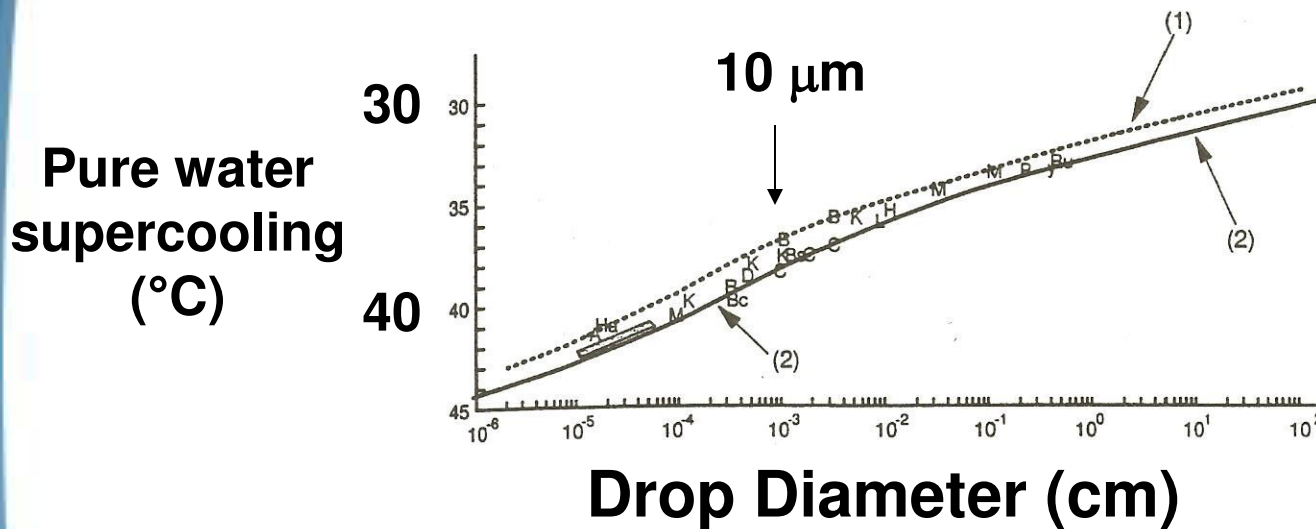
**Solid hydrates (NAM, NAD, NAT)
nitric acid mono, di, tri hydrate**

**Ternary solution droplets
(H₂SO₄, H₂O, HNO₃)**

**Tropospheric (sodium chloride,
sea salt, ammonium sulfate, organic soot,
quartz, sand)**

Supercooled Water - Motivation

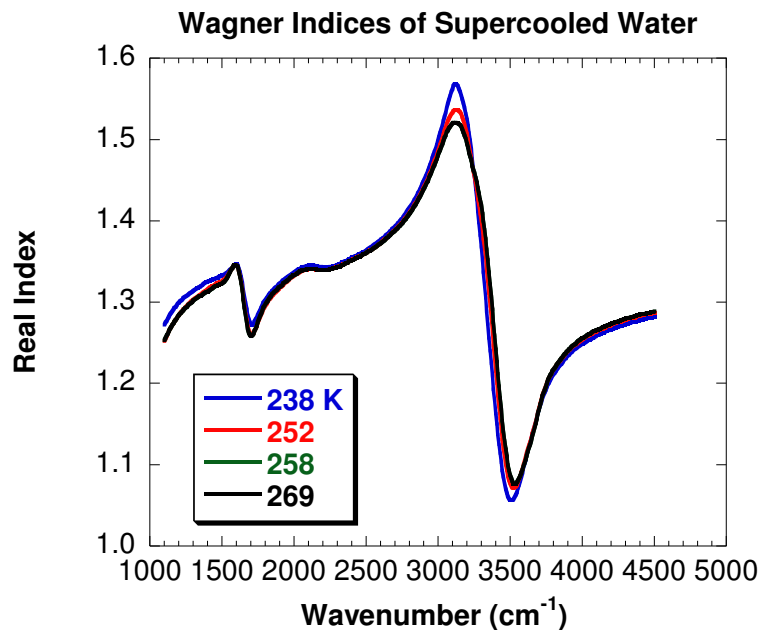
Pure water – 10 μm drop nucleates at -38°C



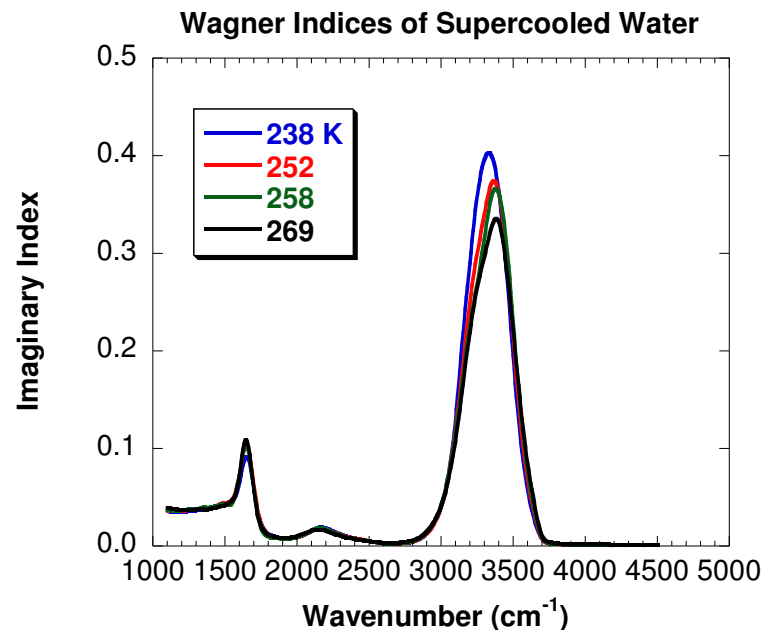
Pruppacher and Klett, *Microphysics of Clouds and Precipitation*, p213, 1998

Water droplet + desert dust nucleates at -15°C
 (Sassen, *Nature*, 434, p456, 2005)

Supercooled Water Indices



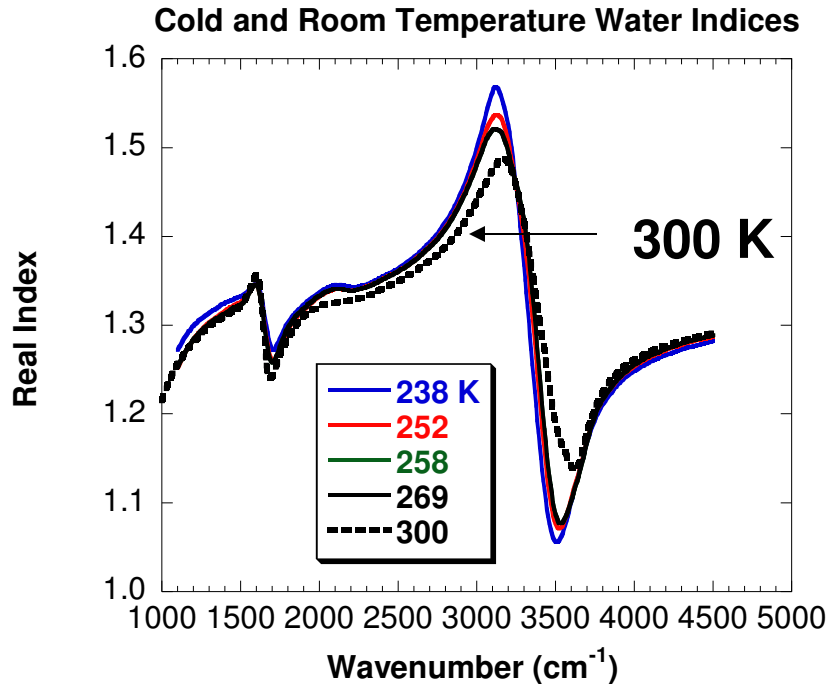
Real Index



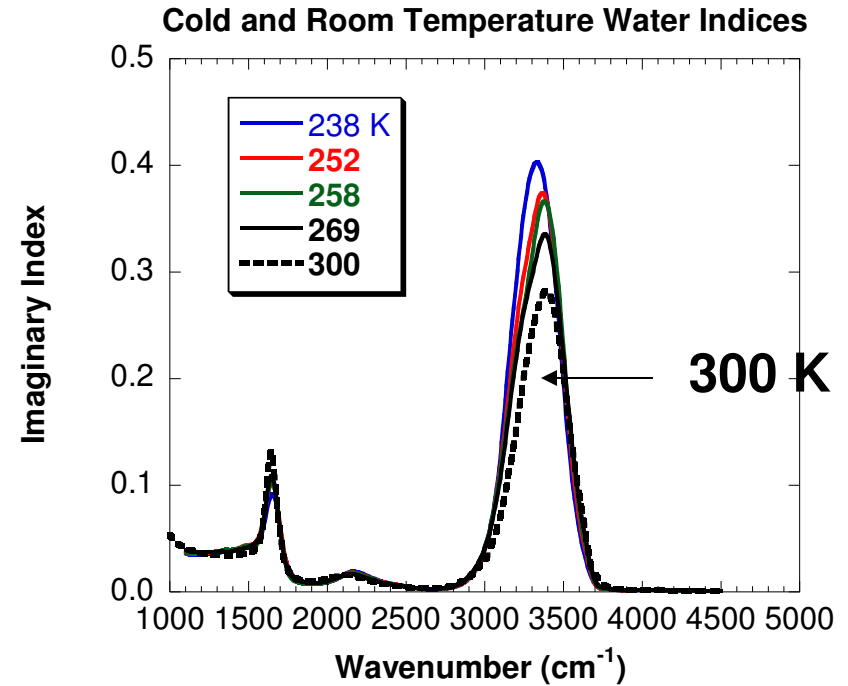
Imaginary Index

Wagner et al., Mid-infrared Extinction Spectra and Optical Constants of Supercooled Water Droplets, *J. Phys. Chem.*, 109, p7099, 2005.

Comparison to Room Temperature Data



Real Index



Imaginary Index

Downing and Williams, Optical constants of water in the infrared, *J. Geophys. Res.*, 80, 1656-1661, 1975.

Ternary Indices - Motivation



Polar Stratospheric Clouds (PSCs)

Need to know PSC composition and size

**Composition: Liquid droplets (ternary),
Solid hydrates (e.g. NAT, NAD)**

Heterogeneous chemistry rates of reaction:

γ = reaction probability (unitless)

A = PSC surface area ($\mu\text{m}^2 \text{cm}^{-3}$)

v = mean molecular speed (cm sec^{-1})

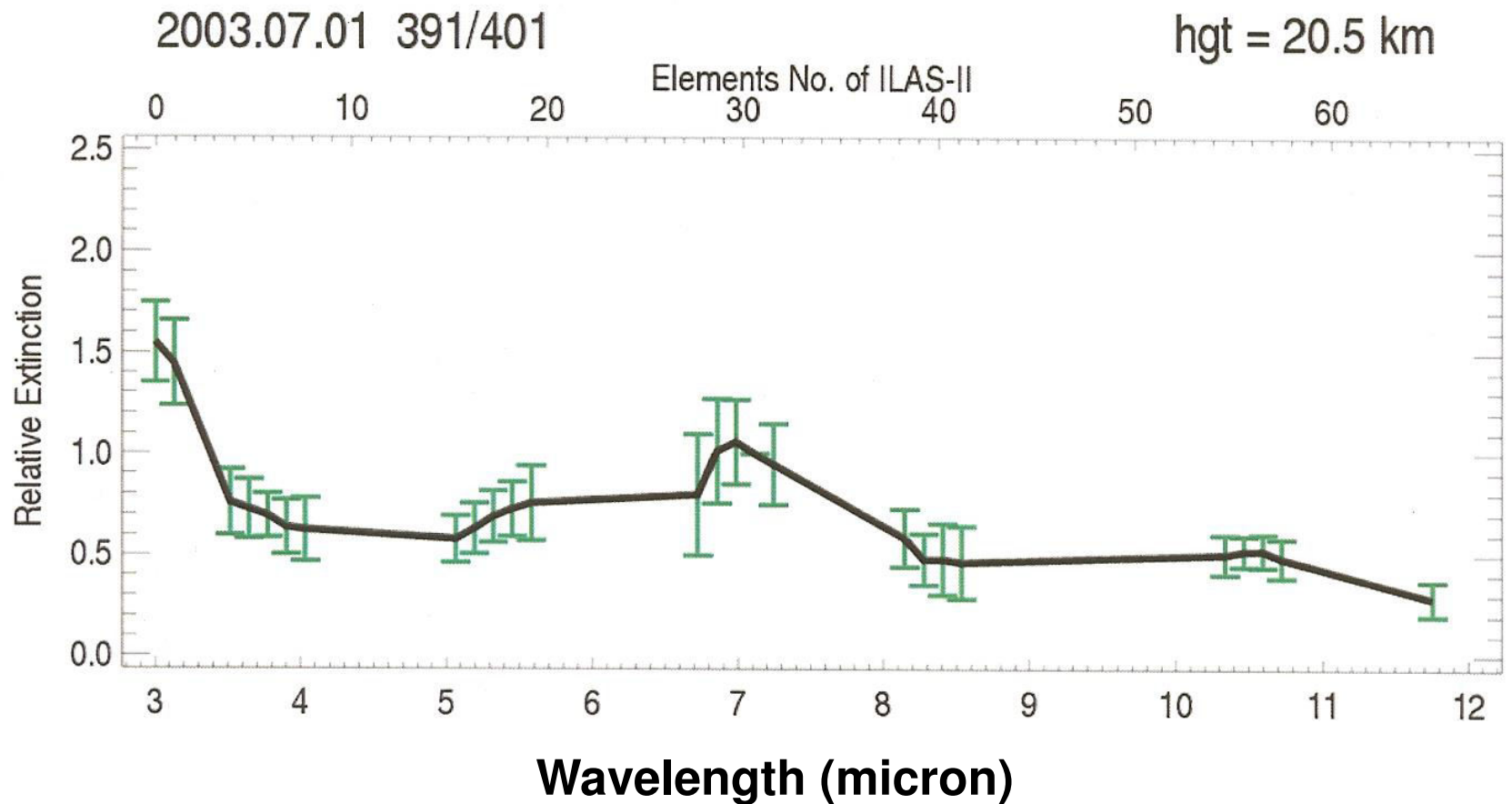
r = rate constant (sec^{-1})

$r = 10^{-8} \gamma A v/4$

γ is a function of PSC composition type

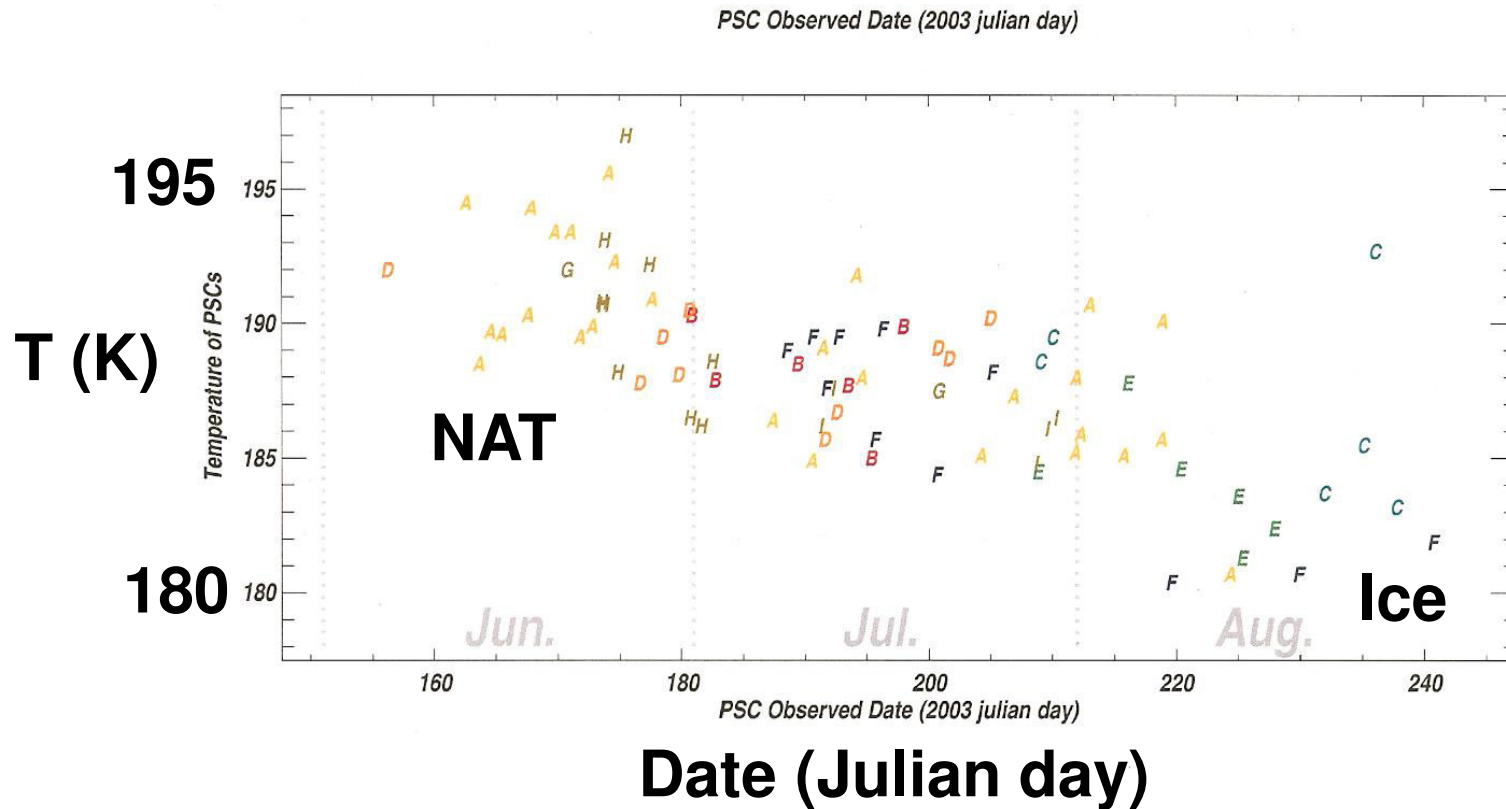
Denitrification: depends upon particle size

Example: observed PSC spectrum



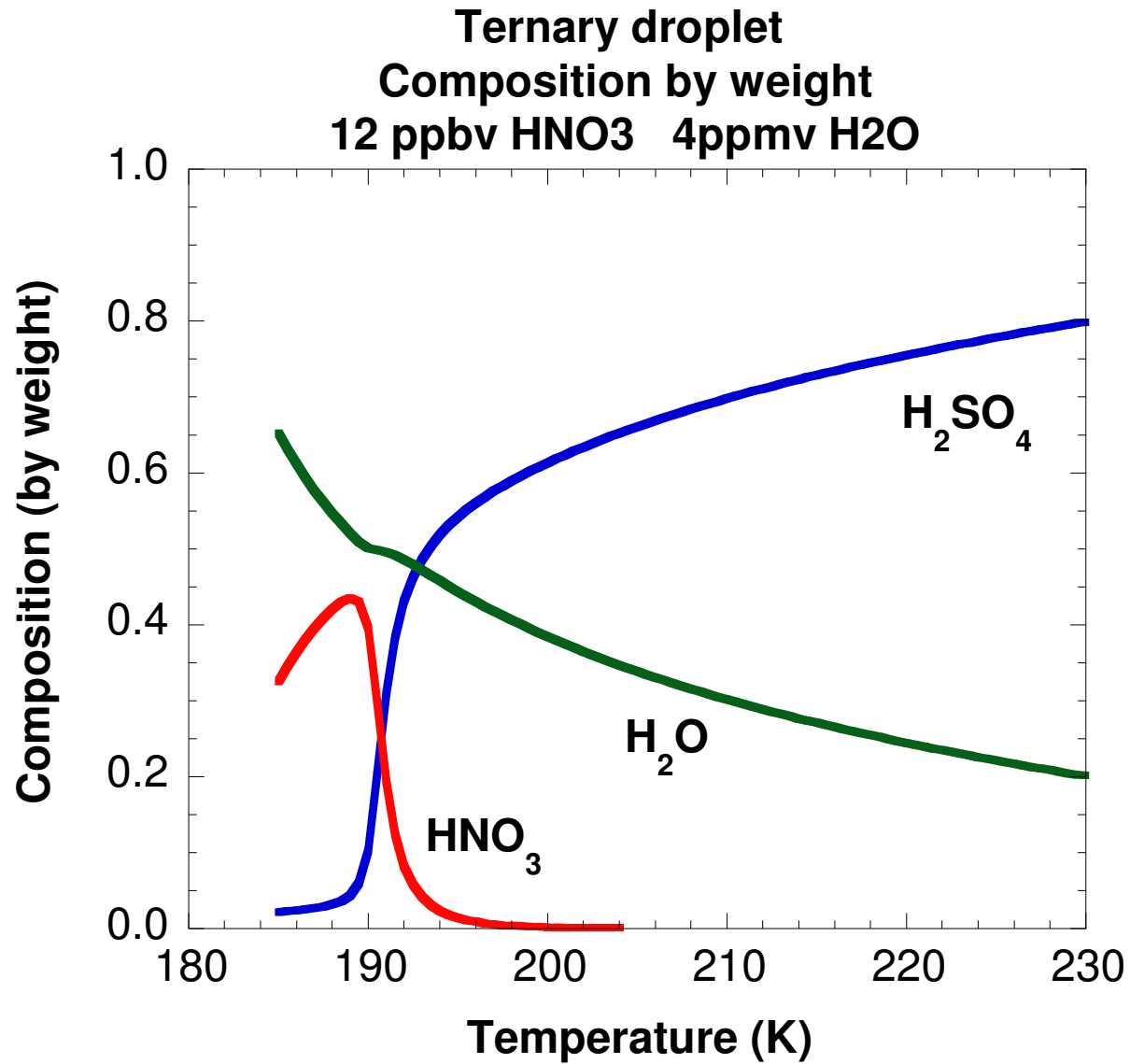
Kim et al., PSCs Observed by the ILAS-II in the Antarctic Region: Dual Compositions and Variation of Compositions during June-August of 2003, *JGR*, in press, 2006.

Example: PSC composition determination



Kim et al., PSCs Observed by the ILAS-II in the Antarctic Region: Dual Compositions and Variation of Compositions during June-August of 2003, *JGR*, in press, 2006.

Different letters denote different composition types.



Carslaw, Luo, Peter, *Geo Res Lett*, 22, p1877, 1995

New Binary and Ternary Indices



C. E. Lund Myhre, D. H. Christensen, F. M. Nicolaisen, and C. J. Nielsen, Spectroscopic Study of Aqueous H₂SO₄ at Different Temperatures and Compositions: Variations in Dissociation and Optical Properties, *J. Phys. Chem. A.*, 107, 1979-1991, 2003.

C.E. Lund Myhre, H. Grothe, A. A. Gola, and C. J. Nielsen, Optical Constants of HNO₃/H₂O and H₂SO₄/HNO₃/H₂O at Low Temperatures in the Infrared Region, *J. Phys. Chem. A*, 109, 7166-7171, 2005.

Myhre (2003, 2005) measurements



Binary ($\text{H}_2\text{SO}_4/\text{H}_2\text{O}$) – 32 cases

Weight % H_2SO_4 = 81, 81, 81, 76, 76, 76, 76, 76, 72, 72
72, 72, 72, 72, 72, 65, 65, 65, 65, 58,
58, 58, 48, 48, 48, 48, 38, 38, 38, 38,
38, 38

Temperature(k) = 298, 273, 267, 298, 273, 233, 213, 203, 298, 253,
245, 233, 223, 213, 203, 298, 263, 243, 223, 298,
243, 233, 298, 273, 234, 213, 298, 277, 257, 243,
223, 213

Binary ($\text{HNO}_3/\text{H}_2\text{O}$)

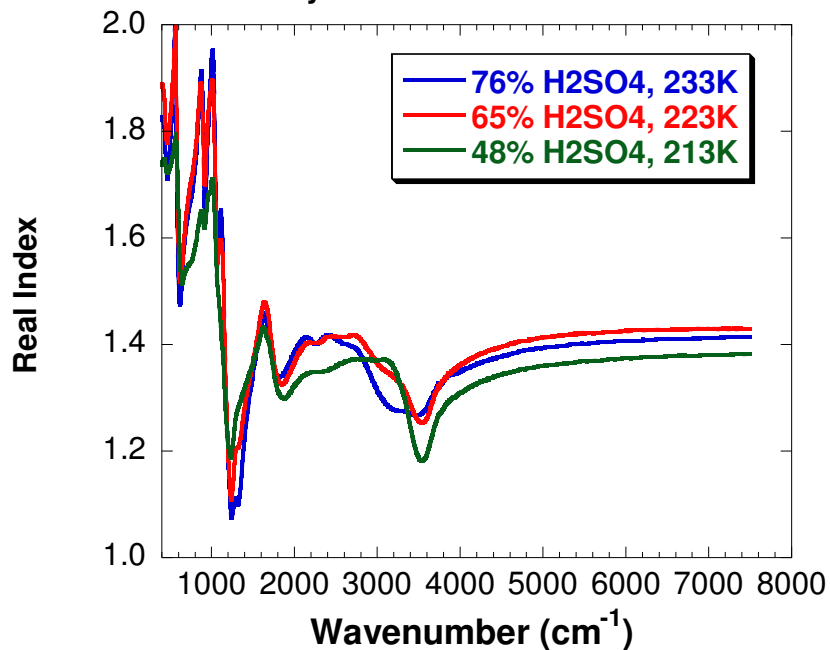
30% HNO_3 , 70 % H_2O T= 223, 233, 243, 253, 273, 293 K
54% HNO_3 , 46 % H_2O T= 243, 248, 253, 273, 293 K
64% HNO_3 , 36% H_2O T= 238, 243, 253, 273, 293 K

Ternary ($\text{HNO}_3/\text{H}_2\text{SO}_4/\text{H}_2\text{O}$)

17% HNO_3 , 25% H_2SO_4 T=183, 193, 203, 213, 223, 253, 273, 293 K
23% HNO_3 , 21% H_2SO_4 T=203, 213, 223, 253, 273 K
46% HNO_3 , 4% H_2SO_4 T=223, 253, 273, 293 K

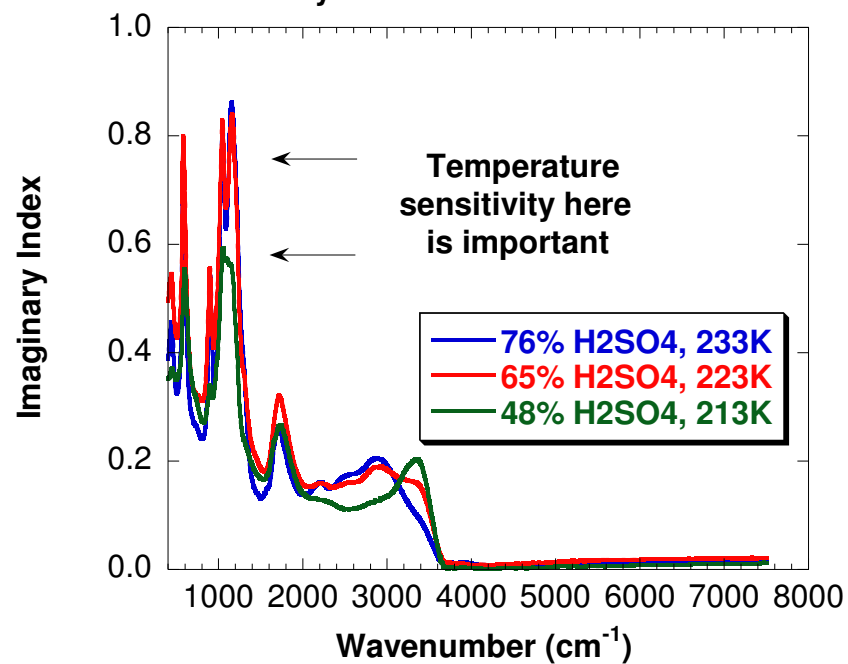
Examples of Myhre H₂SO₄/H₂O Indices

Myhre H₂SO₄/H₂O Indices



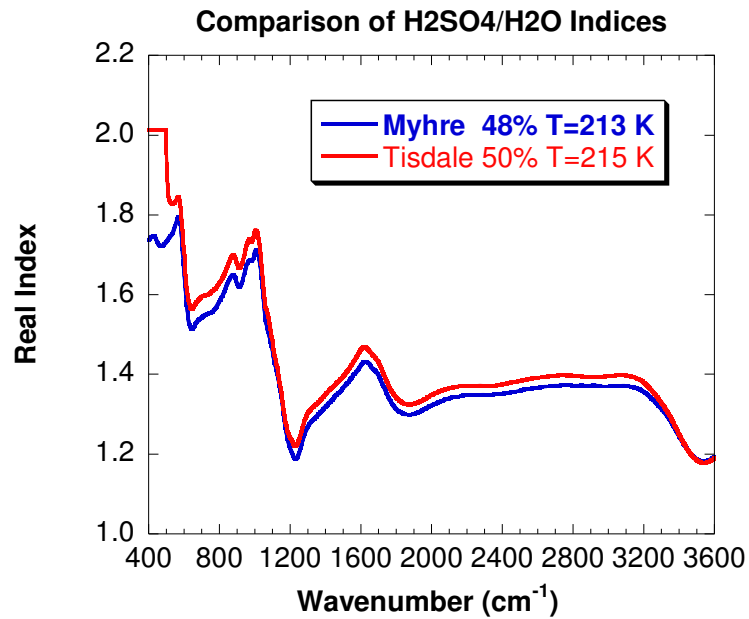
Real Index

Myhre H₂SO₄/H₂O Indices

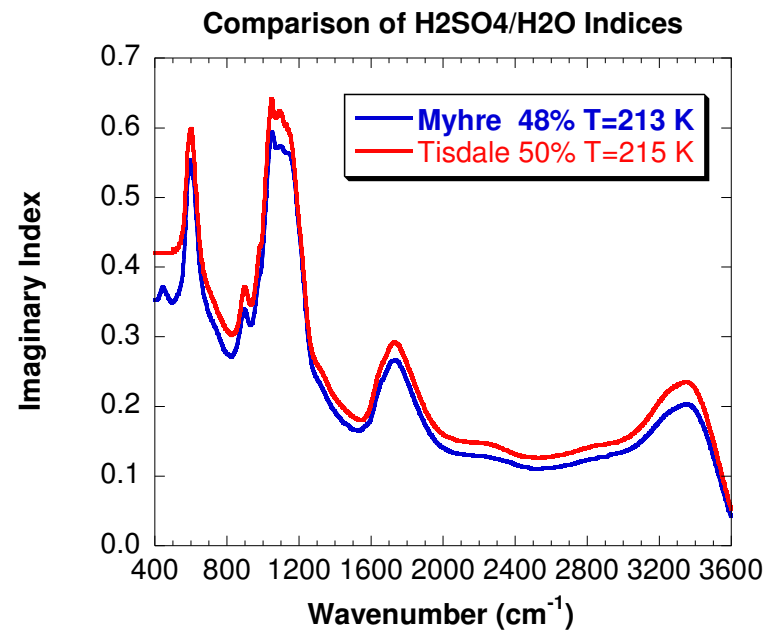


Imaginary Index

Comparison of H₂SO₄/H₂O Indices



Real Index

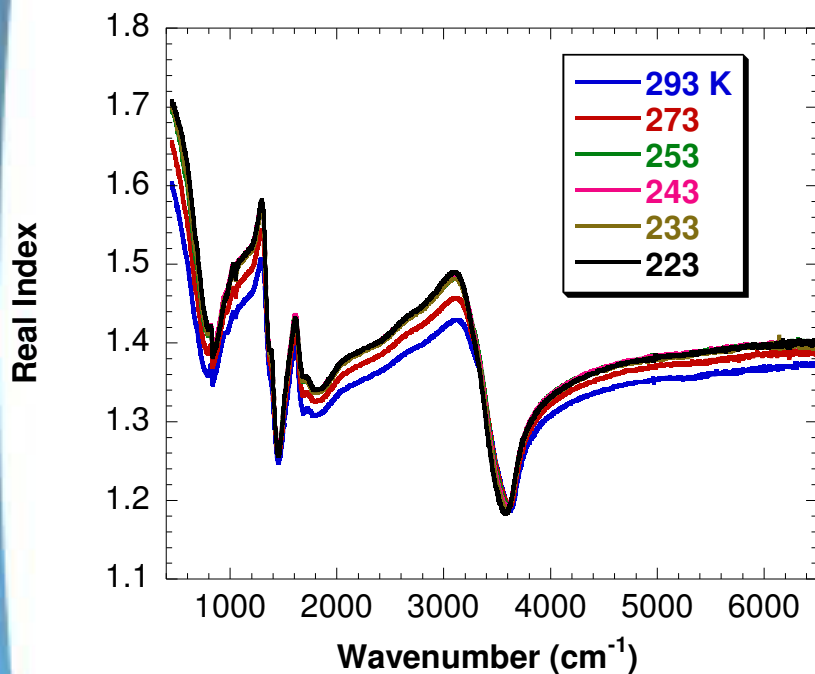


Imaginary Index

Examples of Myhre HNO₃/H₂O Indices

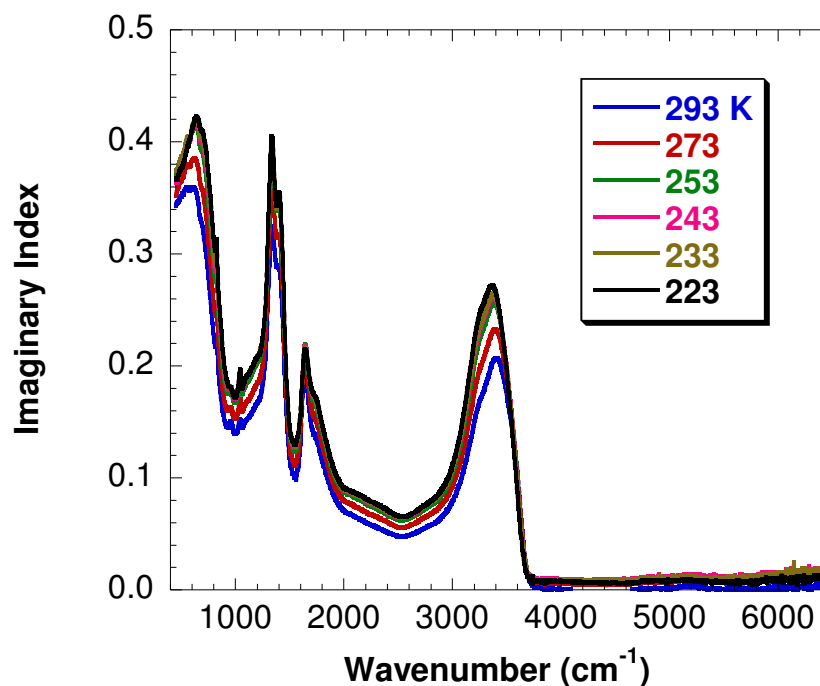
Small Temperature Dependence

Myhre H₂O/HNO₃ Indices
30 wt% HNO₃ - 70 wt% H₂O



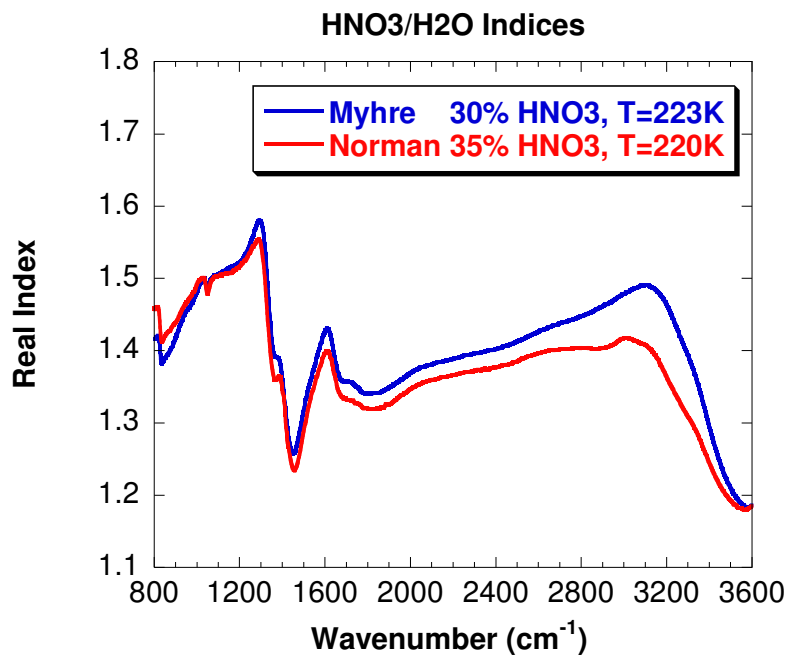
Real Index

Myhre H₂O/HNO₃ Indices
30 wt% HNO₃ - 70 wt% H₂O

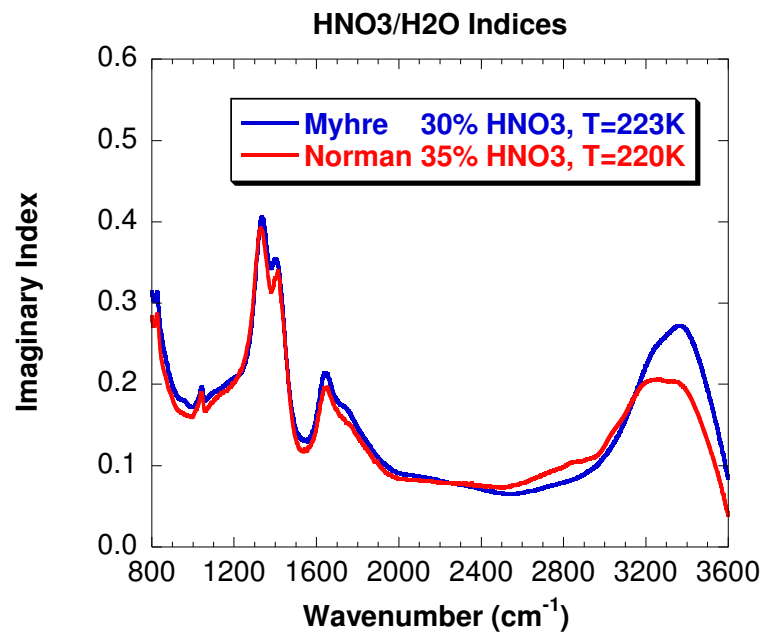


Imaginary Index

Comparisons of HNO₃/H₂O Indices

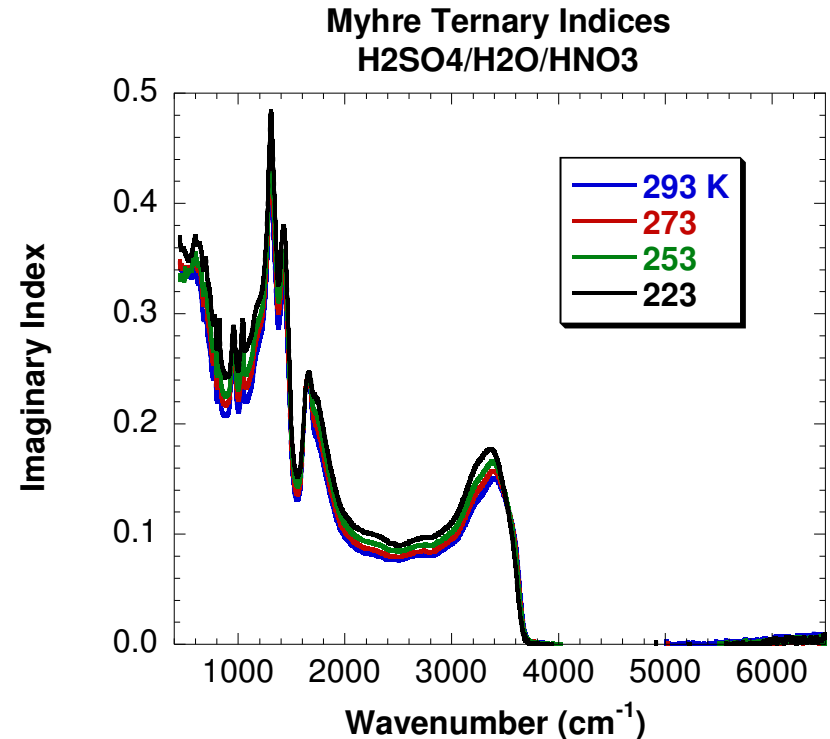
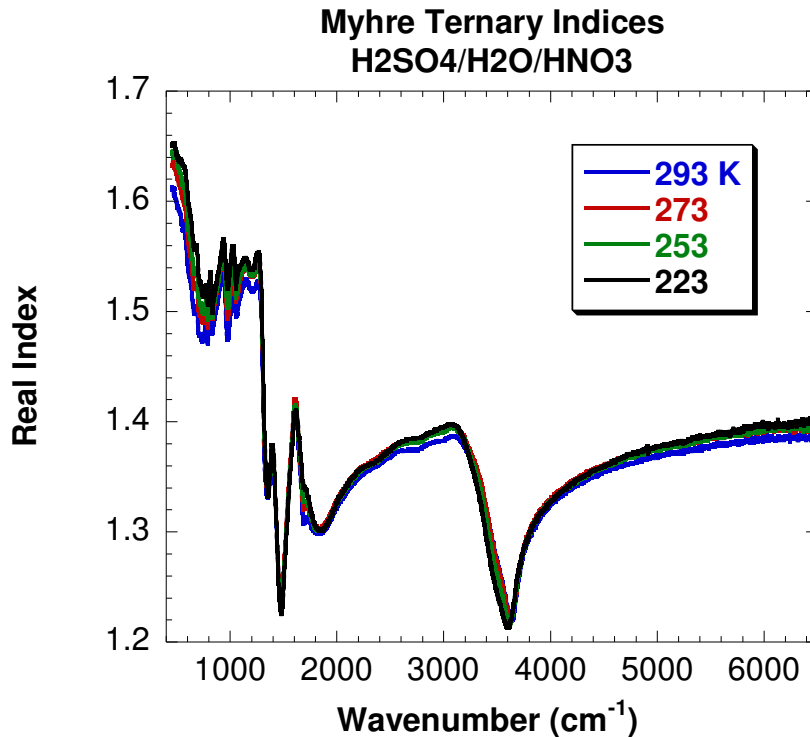


Real Index



Imaginary Index

Examples of Myhre Ternary Indices Small Temperature Dependence



Real Index

Imaginary Index

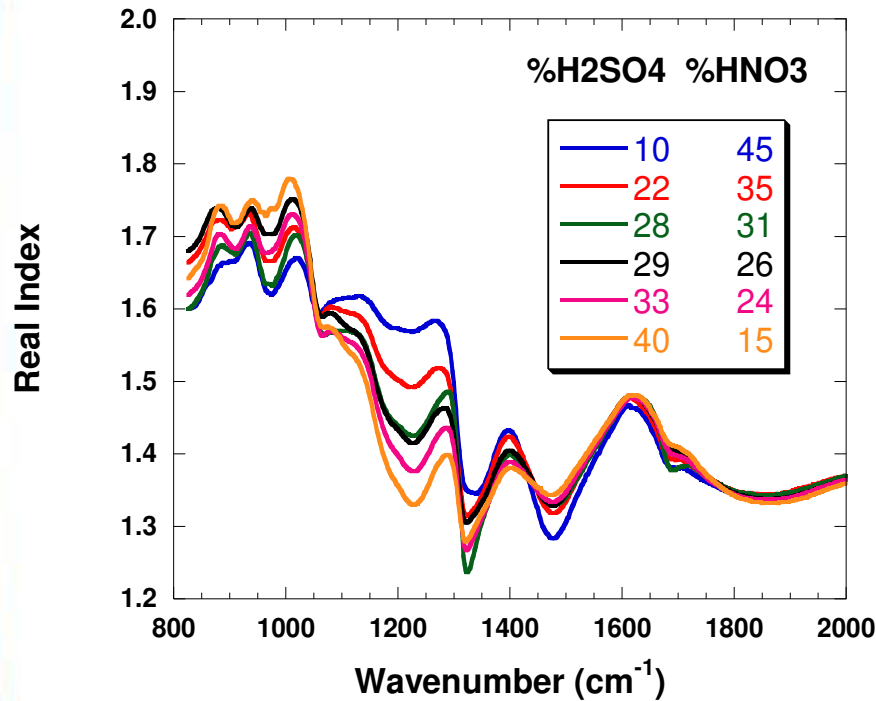
46 % HNO₃, 4% H₂SO₄

HITRAN 2004 Included Norman's Ternary Indices at 220 K

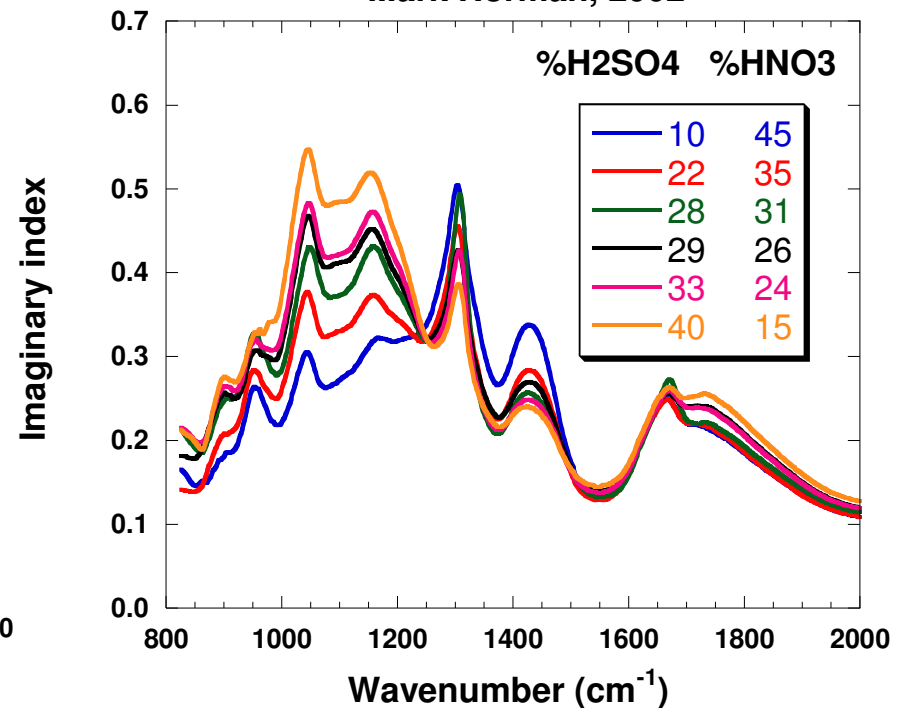


Ternary ($\text{H}_2\text{O}/\text{H}_2\text{SO}_4/\text{HNO}_3$) Indices at 220 K
Mark Norman, 2002

Ternary ($\text{H}_2\text{O}/\text{H}_2\text{SO}_4/\text{HNO}_3$)
Imaginary Indices at 220 K
Mark Norman, 2002



Real



Imaginary

Comparisons of Myhre data to other data sets



Percent differences in H₂SO₄/H₂O indices

| <u>Data set</u> | <u>Real Index</u> | <u>Imaginary Index</u> |
|----------------------------|-------------------|------------------------|
| Niedziela | 2% | 10-20% |
| Biermann | | |
| 1000-3500 cm ⁻¹ | 10% | 10-20% |
| ν < 1000 cm ⁻¹ | problematic | |
| Tisdale | 3% | 10% |

Which data sets to use?



Wagner et al., A quantitative test of infrared optical constants for supercooled sulphuric and nitric acid droplet aerosols, *Atmos. Chem. Phys.*, 3, 1147-1164, 2003.

Compared cloud chamber measurements of $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$ and $\text{H}_2\text{SO}_4/\text{HNO}_3$ extinction spectra with spectra calculated using published indices.

Good agreement

| | |
|-------------------------|--|
| Niedziela (1999) | $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$ |
| Norman (1999) | $\text{H}_2\text{SO}_4/\text{HNO}_3$ |

How to use the data?



Mixing rule for the complex index has been proposed:

$$k(\nu, T, W_s, W_n) = (W_s / (W_s + W_n)) k_s + (W_n / (W_s + W_n)) k_n$$

complex indices k_s and k_n , weight percents W_s and W_n of $\text{H}_2\text{SO}_4/\text{H}_2\text{O}$ (s) and $\text{HNO}_3/\text{H}_2\text{O}$ (n) components

This mixing rule is likely not adequate.

Myhre et al. (2005) state that:

“a reliable model of the optical constants of ternary $\text{H}_2\text{SO}_4/\text{HNO}_3/\text{H}_2\text{O}$ solutions requires a thorough understanding of ionic speciation in the system”

Cross Sections



HITRAN 2004 - 28 molecules

Rothman et al., The HITRAN 2004 molecular spectroscopic database, *J. Quant. Spect. and Radiat. Transfer*, 96, 139-204, 2005.

Data to be added to HITRAN:

Lonardo and Masciarelli, Infrared absorption cross-sections and integrated absorption intensities of HFC-125 and HFC-143a, *J. Quant. Spect. and Radiat. Transf.*, 66, p129-142, 2000.

16 HFC 125 and 19 HFC 143a sets

Temperature range: 203 – 293 K

Pressure: pure vapor, 50, 200, 800 hPa

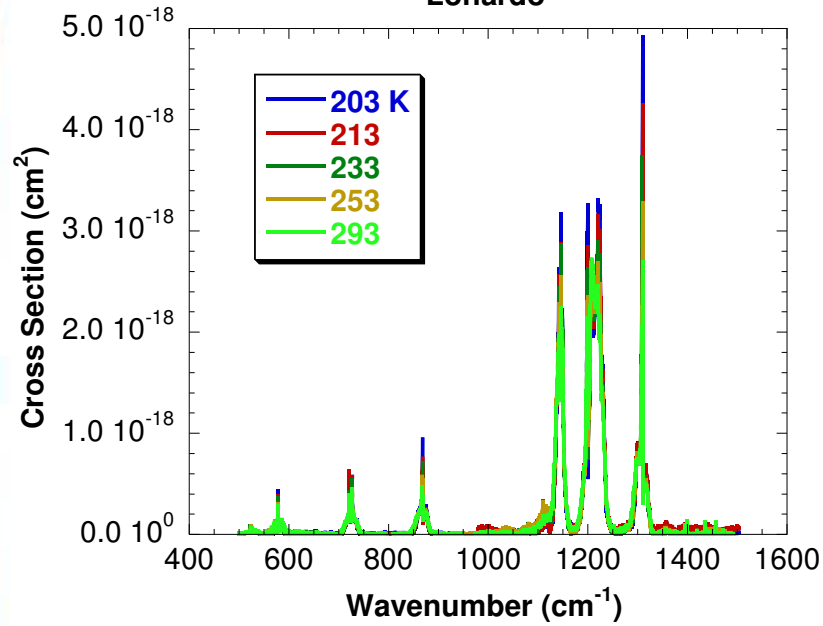
Spectral resolution: 0.03 cm⁻¹

HFC-125

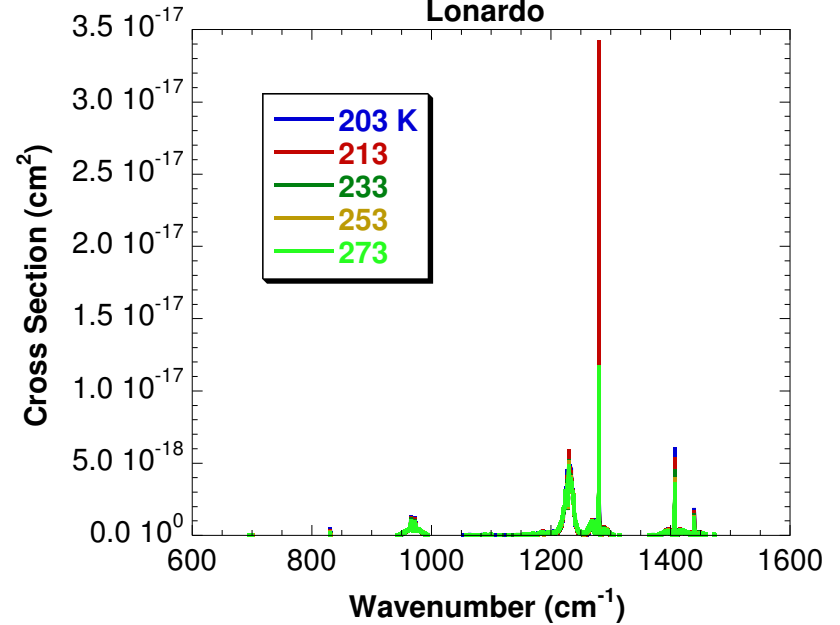
HFC-143a



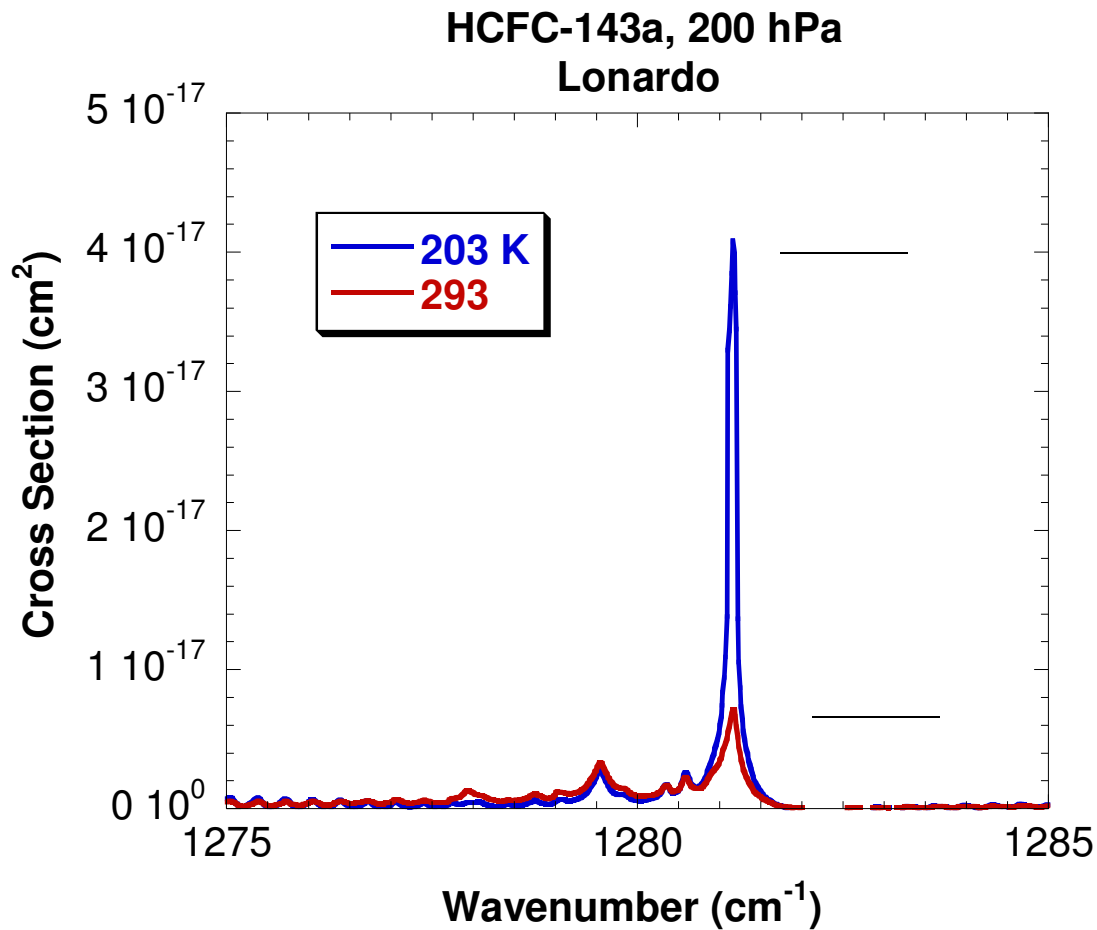
HCFC-125, 200 hPa
Lonardo



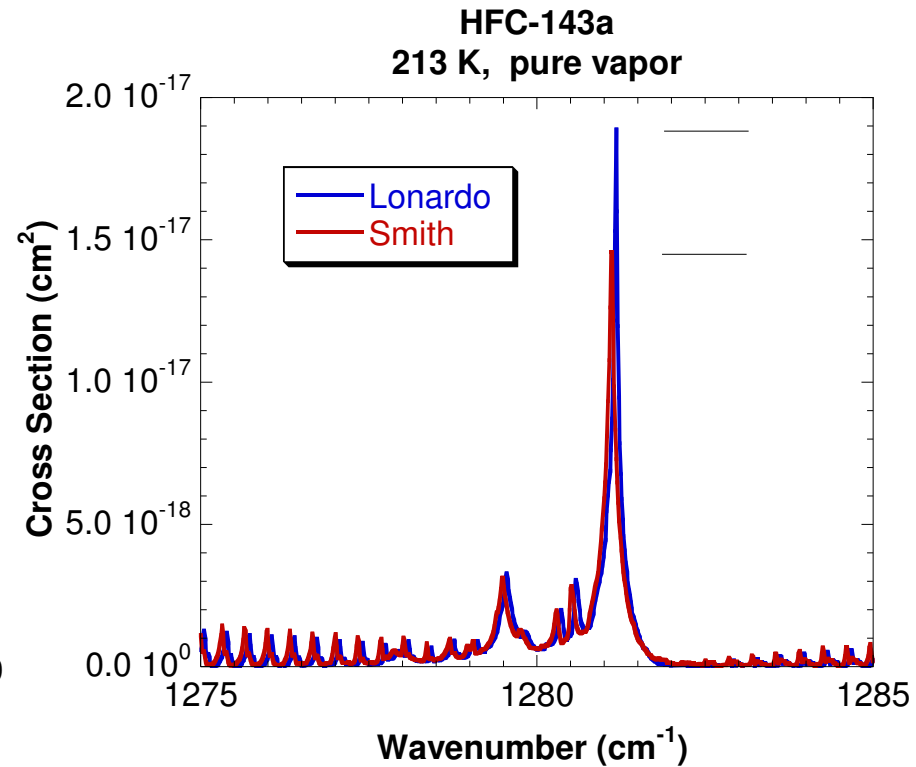
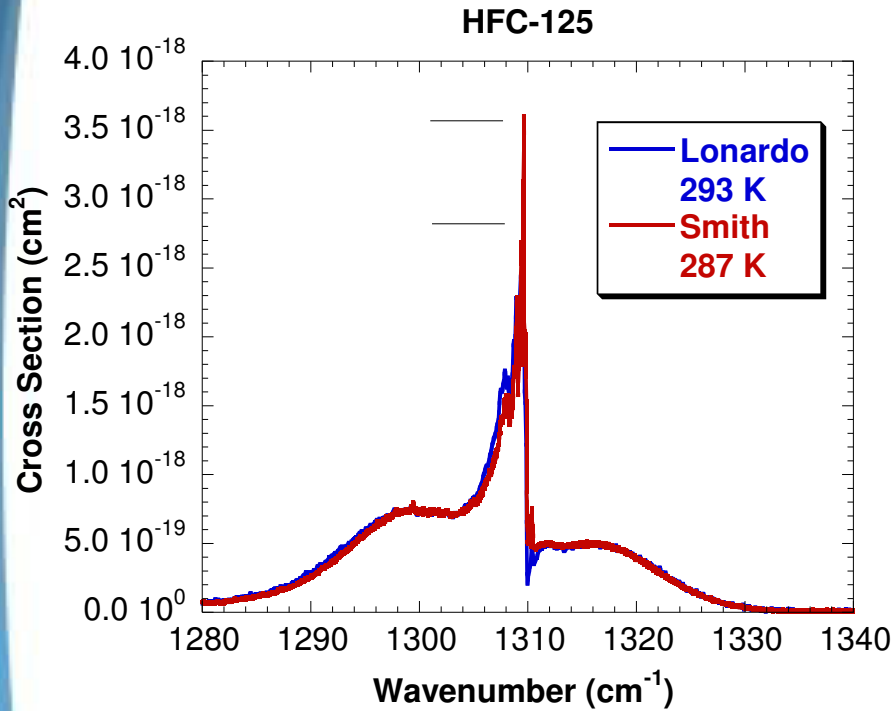
HCFC-143a, 200 hPa
Lonardo



HFC-143a Temperature Dependence



Comparisons to Other Data



Reflectance Spectra - Motivation



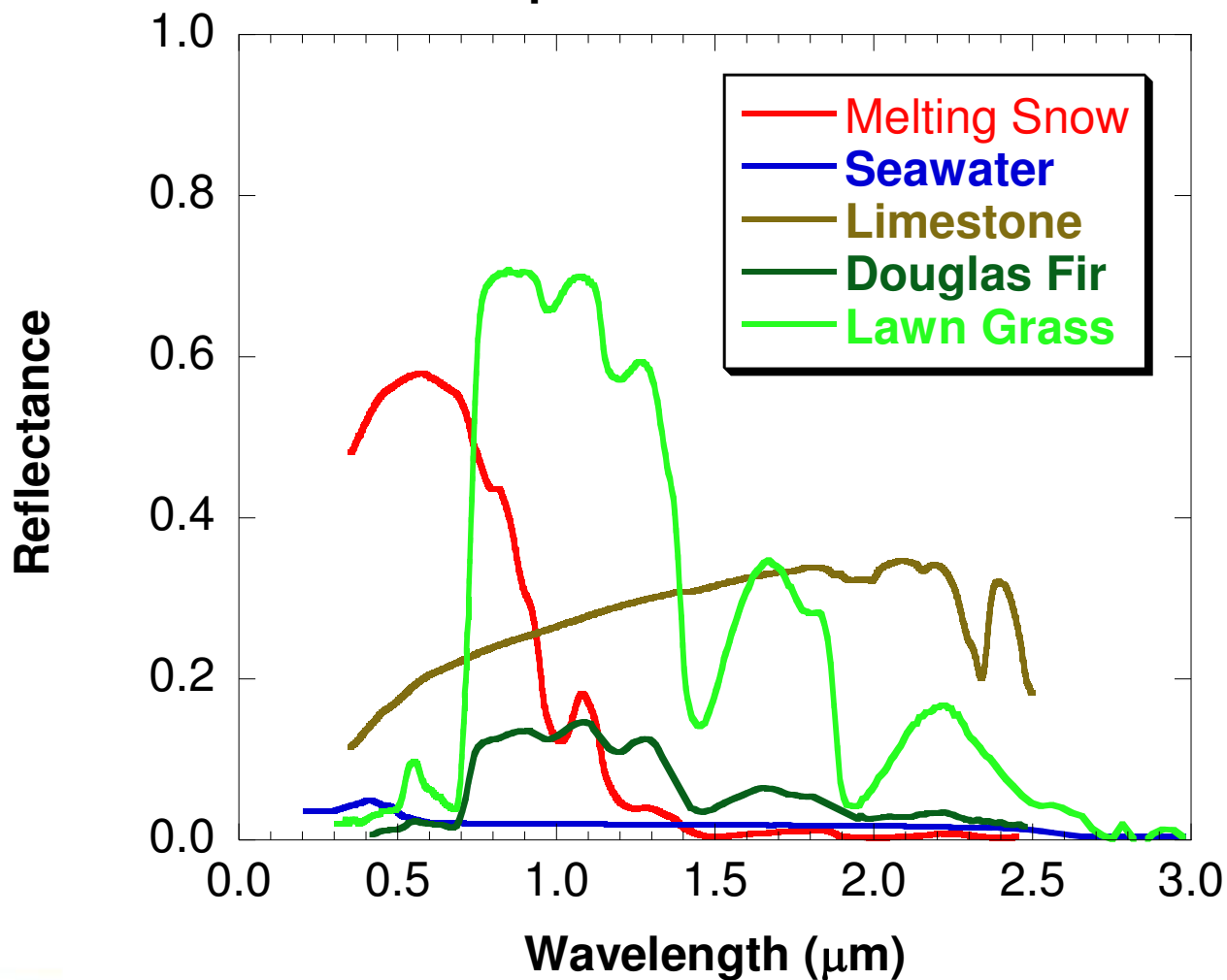
Recent interest focuses on the troposphere

To probe into the troposphere, one frequently encounters radiative transfer physics that is influenced by the surface

Nadir retrievals over bright surfaces are difficult
- MODIS experiment can not detect desert dust over bright deserts

Reflectance Spectra

USGS Spectral Reflectance Data



Reflectance Spectra



USGS Digital Spectral Library

<http://pubs.usgs.gov/of/2003/ofr-03-395/ofr-03-395.html>

Over 800 spectra, including:

13 mineral classes

borate, carbonate, chloride, element, halide,..

7 series

olivine, garnet, scapolite,..

Representative spectra of

H₂O ice, kerogen, desert varnish, evaporite,..

Vegetation

trees, shrubs, grasses, flowers,..

Man-made materials

roofing material, plastics, paint,..

Spectral range: 0.2 to 5.2 μm

Reflectance Spectra



ASTER spectral library
Simon.J. Hook@jpl.nasa.gov

<http://speclib.jpl.nasa.gov/>

The ASTER spectral library includes data from three other spectral libraries: the Johns Hopkins University (JHU) Spectral Library, the Jet Propulsion Laboratory (JPL) Spectral Library, and the United States Geological Survey (USGS - Reston) Spectral Library

~2000 spectra of natural and man-made materials

Conclusions



Refractive Indices

There's still work to be done with the ternary indices!

Cross sections

Tropospheric species will drive interest

Surface reflectance data

Link to existing libraries