

Odor Threshold prediction by means of the Monte Carlo method

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ABSTRACT

A large set of organic compounds (n=906) has been used as a basis to build up a model for the odor threshold (mg/m³). The statistical characteristics of the best model are the following: n=523, r²=0.647, RMSE=1.18 (training set); n=191, r²=0.610, RMSE=1.03, (calibration set); and n=192, r²=0.686, RMSE=1.06 (validation set). A mechanistic interpretation of the model is presented as the lists of statistical promoters of the increase and decrease in the odor threshold.

Keywords: QSPR/QSAR; odor threshold; Monte Carlo method; Optimal descriptor; CORAL software

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35 **1. Introduction**

36

37 The sense of smell and taste are generally referred as the chemical senses, as they give information
38 about the chemistry of the environment. From an evolutionary point of view, chemoreception is
39 generally thought to be one of the most primitive senses that was developed (Nei et al., 2008, Niimura,
40 2012). The potential of chemicals to impact the human olfactory system and to cause apparent health
41 effects was detected (Rosenkran and Cunningham, 2003; Orzi et al., 2015).

42 Coding and processing steps of odor receptor activation system, are in the olfactory bulb, one of the
43 limbic brain structures where part of the autonomic processes are regulated. Although the sense of
44 smell is a conscious perception, sensory information from the outer environment are brought by
45 sensory neurons centrally, where they constantly and unconsciously modulate also the activity of the
46 motor neurons of the autonomic nervous system. Our response to odors from the external environment
47 is both voluntary and involuntary and environments quality can be highly degraded by odor pollution
48 events, causing annoyance and difficulties in public health management. Measuring odor is
49 fundamental to predict environmental effects on the population, and perception, exposure-response
50 relationships, play a fundamental role for annoyance in the long-time exposure pathways, making it
51 enable for risk assessment in this field.

52 Odor intensity measurements are difficult due to the fact that presently no analytical instrument is
53 available to measure objectively odor intensity. Sensorial analysis, even performed by trained
54 assessors, are problematic as there are no fixed reference points and no easy comparison (Sell and
55 Pybus, 2006) and odor thresholds (the lowest concentration of an odorant that is perceivable by
56 human nose) are usually measured (Hoshika et al., 1993) to describe the odor potency of a specific
57 molecule.

58 The odor threshold is an important characteristic of a compound from the ecologic and biochemical
59 viewpoints too (Hobbs et al., 2001; Zahn et al., 2001; Yan et al., 2015; Hansen et al., 2016). Volatile
60 organic compounds are air pollutants that account for a substantial proportion of total pollutant
61 concentrations (Kumar et al., 2014; Brodzik et al., 2014; Yan et al., 2015). Most of these compounds
62 can easily cause pungent sensations even on very low concentration (Wu et al., 2015; Abraham et al.,
63 2012, 2016) and their environmental presence are complex to measure (Bianchi et al., 2013) and
64 difficult to describe (Capelli et al., 2012a).

65 Because of the interaction between these odorants, mixtures of many negligible odor pollutants can
66 generate a stronger odor impact (Le Berre et al., 2008). The odor pollution induced by these
67 substances lowers the quality of life (Palmiotto et al. 2014, Yan et al., 2015), causing potential threats

68 to human health both from a toxicological point of view (Capelli et al., 2012b) and from a stress-
69 related, psychosocial effect (Blanes-Vidal, 2014).

70 A substance with low odor threshold in the environment can be associated to a number of odors. Thus,
71 the ambiguous nature of odorant receptors along with various characteristics of olfactory data has
72 stimulated search for information about threshold data for odor of various compounds which have
73 applications in the field of bioscience, food chemistry and environmental pollution (Pal et al., 2014).
74 Fundamental research has well shown the effect of structural changes of odorant compounds to their
75 odor threshold values (Takeoka et al. 1995, 1996). For instance, pentyl acetate has the lowest odor
76 threshold of all the straight chain acetates; however, the addition of a methyl group in the 1-position
77 caused a 20-fold increase in the odor threshold. Systematic studies to define relationships between
78 structure and odor thresholds have been carried out on homologous series and isomers of substance
79 groups (for example see Boelens and van Gemert, 1986). Estimation of such effects is a complex
80 task: to determine which compounds contribute to the overall apple aroma, long-term analytical
81 methods should be utilized (ASTM E679 2004). In addition, this estimation involves measurement
82 of the compound's odor threshold in water (Teranishi et al., 1987).

83 The odor threshold value is defined by many factors including the temperature, pressure, presence /
84 absence of other substances and so on. Extraction of all these data is a time-consuming technical
85 problem. Consequently, development of computational models for the odor threshold is an attractive
86 alternative. Usually, such models are based on quantitative structure – property / activity relationships
87 (QSPRs/QSARs).

88 The QSAR models can also assist in the detection of potential odorant components from large
89 databases, which reduces the need for time-consuming synthesis and testing a large number of
90 compounds. Before any substance (pharmaceutical, cosmetic, chemical, etc.) can be brought into the
91 European market, its safety to human health and the environment must be evaluated. The
92 QSPR/QSAR paradigm is now supported by the Registration, Evaluation, and Authorization of
93 Chemicals (EC regulation 1907/2006; EU Regulation, 1223/2009), a legislative initiative of the
94 European Commission and the Organization for Economic Cooperation and Development (OECD,
95 2007).

96 The QSPR/QSAR models are used by the Food and Drug Administration (FDA) (Benigni and Zito,
97 2004; Valerio Jr, 2011) for minimizing the rate of false negatives and false positives saving
98 incalculable costs for manufacturers. The Council for International Organizations of Medical
99 Sciences (CIOMS) (Bankowski and Howard-Jones, 1986) also recommends the QSPR/QSAR
100 methods before animal experiments for the advancement of biological knowledge.

101 The predictive model for the odor threshold can find applications in many practical aspects: (i)
102 perfume manufacture; (ii) medicine; (iii) chemical technology; (iv) drug discovery and (v) regulation.
103 There are attempts to build up QSPR/QSAR for the odor threshold (Xu et al., 2012; Pal et al., 2014;
104 Polster and Schieberle, 2015). The aim of present work was building up a predictive model for the
105 odor threshold (mg/m^3) using the Monte Carlo method available via the CORAL software (CORAL,
106 2016).

107

108 **2. Method**

109 *2.1. Data*

110 The numerical data on the odor threshold (OT) were taken from a large database available (van
111 Gemert, 1999). Data distribution does not appear to be unimodal, symmetric. For several odorants,
112 several OT were available with large differences from average, median and mode values. For
113 example for hydrogen sulfide values of 0.23, 0.0042 and 0.012 mg/m^3 (average, median and mode
114 respectively) are obtained indicating not only large OT ranges, but also asymmetrical distributions.
115 In the present study the average OT values have been used, a classical approach for OT description
116 (see for example Chemoreception: In Ref. Sell and Pybus, 2006).

117 Three splits, into training, calibration, and validation sets were examined.

118

119 *2.2. Optimal descriptors*

120 Optimal descriptors (Toropov and Toropova, 2014, 2015a,b; Toropova and Toropov, 2013, 2014)
121 which are involved to build up the QSAR model for the pOT , (minus decimal logarithm of odor
122 Threshold) are the following:

$$123 \quad DCW(T^*, N^*) = \sum CW(s_k) + \sum CW(ss_k) + \sum CW(sss_k) + \quad (1)$$
$$CW(BOND) + CW(NOSP) + CW(HALO) + CW(PAIR)$$

124 In Eq. 1, the s_k , ss_k , and sss_k are combinations of one, two, and three “SMILES atoms”. The “SMILES
125 atom” is a fragment of the SMILES notation, which contains one symbol or two symbols, which
126 cannot be examined separately (e.g. ‘Cl’, ‘Br’, etc.). The $CW(s_k)$, $CW(ss_k)$, and $CW(sss_k)$ are
127 correlation weights of the above-mentioned “SMILES atoms”. The correlation weights are
128 coefficients, which are used to calculate the descriptor. The numerical data for the correlation weights
129 are obtained by the Monte Carlo method optimization procedure, which gives maximum correlation
130 coefficient between endpoint and the optimal descriptor. The BOND, NOSP, HALO, and PAIR are
131 global attributes of SMILES which reflects the presence of various kinds of chemical bonds (BOND);
132 the presence of nitrogen, oxygen, sulfur, and phosphorus (NOSP); the presence of halogens, i.e.
133 fluorine, chlorine, bromine, and iodine (HALO); and presence of various combinations of SMILES

134 atoms. The $CW(BOND)$, $CW(NOSP)$, and $CW(HALO)$, and $CW(PAIR)$ are correlation weights of the
135 global attributes of SMILES. The detailed description of the above- listed local (s_k , ss_k , and sss_k) and
136 global (BOND, NOSP, HALO, and PAIR) attributes of SMILES is available in the literature
137 (Toropova et al., 2015) as well as at web site of the CORAL software (CORAL, 2016).

138 The T is the threshold, i.e. a coefficient used to classify SMILES attributes into two classes (i) rare
139 or noise; and (ii) active. The rare attributes are blocked (their correlation weights are fixed zero). The
140 coefficient can be 1, 2, ..., M. The T^* is threshold which gives preferable statistical quality of the
141 model for the calibration set. The N is the number of epochs of the Monte Carlo optimization. The
142 N^* is the number which gives preferable statistical quality for the calibration set. The T^* and N^* are
143 calculated according to scheme suggested in works (Toropova et al., 2015). Having the numerical
144 data for the correlation weights, one can calculate the $DCW(T^*, N^*)$ for the training set and define
145 regression parameters C_0 and C_1 for the following model

$$146 \quad pOT = C_0 + C_1 \times DCW(T^*, N^*) \quad (2)$$

147 The predictive potential of the model calculated with Eq. 2 should be checked up with external
148 validation set (Toropova et al., 2015).

149

150 **3. Results and Discussion**

151

152 *3.1. QSAR models*

153 The Monte Carlo optimization with T^* and N^* which are selected according to scheme suggested in
154 work (Toropova et al., 2015) gives the following models:

155

$$156 \quad pOT = -5.5431(\pm 0.0035) + 0.1418(\pm 0.0001) * DCW(1,22) \quad (3)$$

157

$$158 \quad pOT = -5.4618(\pm 0.0036) + 0.1331(\pm 0.0001) * DCW(1,17) \quad (4)$$

159

$$160 \quad pOT = -5.5541(\pm 0.0033) + 0.1527(\pm 0.0001) * DCW(1,28) \quad (5)$$

161

162 Table 1 contains the statistical characteristics of the models for pOT calculated with Eqs. 3-5

163 [Table 1 around here]

164 *3.2. Mechanistic interpretation*

165 Table 2 contains the list of molecular features, which are statistically stable promoters of increase or
166 decrease of the pOT . These data selected according to the following principles: (i) molecular features
167 extracted from SMILES with large prevalence in the training and calibration sets; (ii) molecular

168 features which have positive correlation weights for all three runs of the Monte Carlo optimization;
169 and (iii) molecular features which have negative correlation weights for all three runs of the Monte
170 Carlo optimization.

171 One can see, there are stable promoters of the *pOT* increase related to all distributions. Table 3
172 contains interpretation of SMILES attributes as a molecular features.

173

174 3.3. Domain of applicability

175 Building up a model can be accompanied by probabilistic characteristics of the influence of the
176 parameters and/or conditions upon endpoint (Toropova et al., 2015).

177 An example of the probabilistic parameter is a statistical defect of an attribute. The defect can be
178 calculated as the following:

$$179 \text{defect}(A_k) = \frac{P_T(A_k) - P_C(A_k)}{N_T(A_k) + N_C(A_k)} \quad (6)$$

180 where $P_T(A_k)$ and $P_C(A_k)$ are probabilities of attribute A_k in the training and the calibration sets,
181 respectively; $N_T(A_k)$ and $N_C(A_k)$ are prevalence of the attribute A_k in the training and the calibration
182 sets, respectively. It should be noted if $N_C(A_k) = 0$ then the $\text{defect}_k = 1$.

183 The criterion to define the domain of applicability is the following: SMILES falls into domain of
184 applicability if defect (SMILES) calculated with Eq. 7 is less than average value of defects of
185 SMILES over the training set (inequality 8).

$$186 \text{defect}(SMILES) = \sum \text{defect}(A_k) \quad (7)$$

$$187 \text{defect}(SMILES) < 2 \times \overline{\text{defect}(SMILES)} \quad (8)$$

188 The sum of $\text{defect}(SMILES)$ is a measure of a distribution defect (DD):

$$189 DD = \sum_{Train\&Calib} \text{defect}(SMILES) \quad (9)$$

190 The suggested criteria (Eqs. 6-9) are calculated with data on the training and calibration sets without
191 any data on the external validation set. Our hypothesis is: if the first distribution D1 is characterized
192 by the distribution defect = DD1 and distribution D2 is characterized by the distribution defect =
193 DD2, then the distribution D1 is better than distribution D2 if DD1 is smaller than DD2.

194 The comparison of models from the literature (Pal et al., 2014; Cliff et al., 2011; Xu et al., 2012;
195 Hansen et al., 2016) shows that the CORAL software can be a satisfactory and useful tool to predict
196 the odor thresholds for a large variety of organic compounds (Table 4).

197 *Supplementary materials* section contains screenshots of the models calculated with the CORAL
198 software (<http://www.insilico.eu/coral>). Technical details related to total data set (i.e. three

199 distributions into the training, calibration, and validation sets examined in this work as well as
200 experimental and calculated values of odor threshold) are available on the request.

201 **4. Conclusions**

202 The suggested models for odor threshold have good statistical quality. It must be stressed that odor
203 measurements are complex, the odor sensitivity in population is lognormally distributed, and the
204 database used is comprehensive of all data in literature. The additional checking up of the approach
205 by means of comparison of three different distributions into the training, calibration, and validation
206 sets confirms the reproduction of the statistical quality for the three random events: i.e. building up
207 the QSAR model for random splits. The suggested models have mechanistic interpretation via groups
208 of promoters of increase/decrease for *pOT*. Finally, the special criteria for domain of applicability
209 and quality of distribution into the training and validation sets are defined for the approach. Thus, the
210 approach gives models in accordance with OECD principles (OECD, 2007).

211

212 **Software**

213 The CORAL software utilized in this work is free available on the Internet
214 (<http://www.insilico.eu/coral>).

215

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220 **References**

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341 Table 1

342 The statistical characteristics of QSAR models for odor threshold calculated with Eqs. 3-5

343

Split	n _{train}	r ² _{train}	q ² _{train}	Strain	n _{calib}	r ² _{calib}	Scalib	n _{valid}	r ² _{valid}	Svalid
1, Eq.3	523	0.647	0.645	1.18	191	0.610	1.03	192	0.686	1.06
2, Eq.4	479	0.600	0.597	1.26	214	0.667	1.00	213	0.587	1.08
3, Eq.5	523	0.644	0.642	1.15	191	0.618	1.16	192	0.664	1.01

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353 Table 2

354 Promoters of increase and decrease for *pOT*

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split	Attribute Ak	CW(Ak) in run 1	CW(Ak) in run 2	CW(Ak) in run 3	Frequency in training set	Frequency in calibration set	Defect(Ak)
	Promoters of <i>pOT</i> increase						
1	HALO00000000	3.58826	4.04993	3.72319	466	171	0.0000
2	HALO00000000	2.47709	2.71629	2.71376	425	190	0.0000
3	HALO00000000	2.64024	2.09750	2.39838	463	173	0.0000
1	C...C.....	0.51254	0.48620	0.26053	358	147	0.0002
2	C...C.....	0.36085	0.44673	0.15952	328	154	0.0001
3	C...C.....	0.05871	0.10788	0.12833	365	139	0.0001
1	NOSP01000000	5.32536	5.22168	5.37597	298	123	0.0002
2	NOSP01000000	5.16421	5.37228	5.15435	277	130	0.0001
3	NOSP01000000	4.29786	4.00130	4.16129	292	124	0.0002
1	++++O---B2==	1.33859	1.32759	1.12530	282	108	0.0001
2	++++O---B2==	0.52102	0.65489	0.50825	260	111	0.0001
3	++++O---B2==	1.73585	1.60045	1.50177	272	111	0.0002
	Promoters of <i>pOT</i> decrease						
1	(.....	-1.32289	-1.28659	-1.27594	369	128	0.0001
2	(.....	-0.97827	-1.12017	-1.23431	353	140	0.0002
3	(.....	-1.38277	-1.28759	-1.19665	373	133	0.0000
1	O.....	-2.14641	-2.06157	-2.14781	351	133	0.0001
2	O.....	-1.51134	-1.61069	-1.61012	321	144	0.0000
3	O.....	-1.89988	-1.83508	-1.81488	346	137	0.0001
1	1.....	-1.43485	-1.32913	-1.76430	247	65	0.0004
2	1.....	-0.69631	-0.79625	-0.70094	231	70	0.0005
3	1.....	-1.90238	-1.83671	-1.95808	247	65	0.0004
1	++++N---B2==	-3.45368	-3.42215	-3.44855	50	7	0.0010
2	++++N---B2==	-1.90281	-1.90455	-1.90284	40	9	0.0008
3	++++N---B2==	-3.29875	-3.30467	-3.30140	43	13	0.0003

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361 Table 3

362 Interpretation of the promoter of increase and decrease for *pOT*

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Attribute A_k	Comment
Promoter of <i>pOT</i> increase	
HALO00000000	Absence of F, Cl, Br
C...C.....	Presence of carbon – carbon bonds (sp ³)
NOSP01000000	Presence of oxygen together with absence of nitrogen, sulfur and phosphorus
++++O---B2==	Presence of oxygen and double bonds
Promoter of <i>pOT</i> decrease	
(.....	Branching in molecular skeleton
O.....	Presence of oxygen
l.....	Presence of rings
++++N---B2==	Presence of nitrogen and double bond

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369 Table 4

370 Statistical characteristics of QSAR models for odor threshold from the literature

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Reference	Number of compounds in training set	Determination coefficient for training set	Number of compounds in validation set	Determination coefficient for validation set
Cliff et al., 2011	10	0.728	-	-
Xu et al., 2012	40	0.8012-0.8767	10	0.648-0.7746
Pal et al., 2014	42	0.809	11	0.813
Hansen et al., 2016	94	0.74	21	0.77

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