

20 years of atomistic simulation in cement

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MOTIVATION: The C-S-H (Calcium silicate hydrate) gel is the most important hydration product of cement based materials. Much of the existing knowledge on the nanostructure of C-S-H has been gained from structural comparisons with crystalline calcium silicate hydrates (see Fig.1). These crystalline structures have been the starting point for the few existing computational works on C-S-H gel [1,2]. However, theoretical predictions of the Bulk and Young modulus overestimate largely the values obtained experimentally [3-7]. This work aims precisely to shed light on the origin of such a discrepancy.

Figure 1: Examples of C-S-H crystal structures: (a) Tobermorite T9R and (b) Jennite

Dependence on the Composition

To assess the influence of compositional variations on the mechanical properties of the C-S-H gel, we have analysed different C-S-H crystals (see Table 2), covering the Ca/Si range that is typically found in cement pastes (from approx 0.5 to approx 2).

Crystal	Chemical formula	Bulk	Young	Ca/Si	B/B _{exp}	Y/Y _{exp}
Jennite	C ₆ H ₈ O ₁₂ S ₂ Si ₂ O ₁₁ ·H ₂ O	1.7	1.21	1.1	11.5(1)	14.5(1)
Tobermorite T9R	C ₄ H ₆ O ₁₀ Si ₂ O ₇	0.9	0.9	0.6	10.7(1)	11.8(1)
Tobermorite T14	C ₄ H ₆ O ₁₀ Si ₂ O ₇ ·H ₂ O	0.9	1.21	0.6	10.7(1)	11.8(1)
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Experimental value C-S-H gel	-	-	-	-	1.1	1.1
Experimental value concrete	-	-	-	-	1.1	1.1

Note: B = Bulk modulus, Y = Young's modulus. The values in parentheses are standard deviations. The values in parentheses are standard deviations. The values in parentheses are standard deviations.

CONCLUSIONS In this work the mechanical properties of the cementitious C-S-H gel have been computed. Firstly the bulk and Young modulus of several perfect C-S-H crystals have been computed. Our calculations have shown that both Bulk (B) and Young modulus (Y) increase with water molecules enter into its composition. This last point was nonetheless expected to us. Nevertheless neither the water content nor variations of Ca/Si ratios can explain by their own the difference between the experimental and theoretical values. Secondly we have introduced defects in the crystalline structures of T14 and Jennite; in fact, cures the art admits that the C-S-H gel can be described as a mixture of tobermorite-type and jennite structures [13].

References

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- [2] L. Laguerre in "Nanotechnology in Construction" Symposium Proceedings, P. J.M. Barthelemy Editor (PSC Pub, 2004).
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- [5] P. Lecoq, E.A. Lindgren, M. Rosen and H. Sato, *Mater. Sci. Eng.*, 27, 61-64, 2000.
- [6] P. Lecoq, private communication.
- [7] M. Jaber, acknowledges the grant received from the EITE. Thanks are due to the Basque Government for funding NANOCATERS consortium of the project MONACEM (Ref MAT2005-G036E) within the framework of the "Plan Nacional de Ciencia y Tecnología". Finally
- [8] G. Cor
- [9] A. Gholami
- [10] J. M. Barthelemy
- [11] H. J. M. Pelling
- [12] A. Gholami
- [13] G. Cor

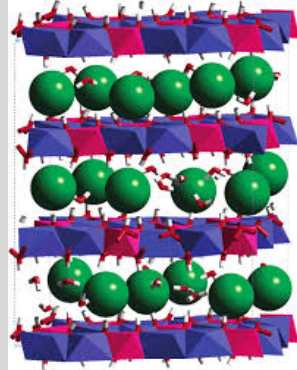
Acknowledgments

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Before 2005: early studies by the clay community

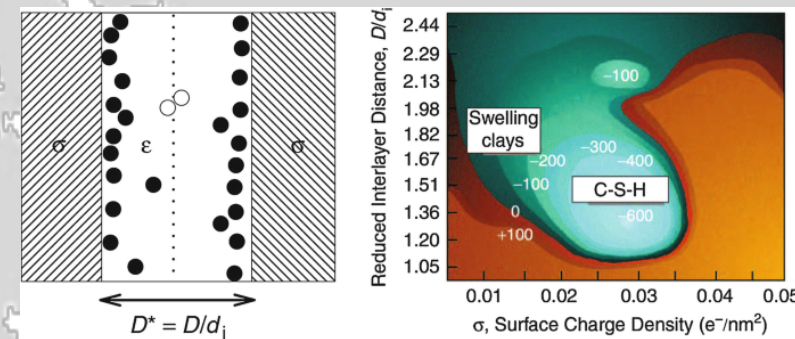


Urbana Champaign
Kirkpatrick & Kalinichev



First MD studies on tobermorite,
ettingite, protlandite...

France
Faucon & Nonat,
R.Pellenq, Labbez



Primitive model to investigate
cohesive forces

2005 to 2012: specialised studies

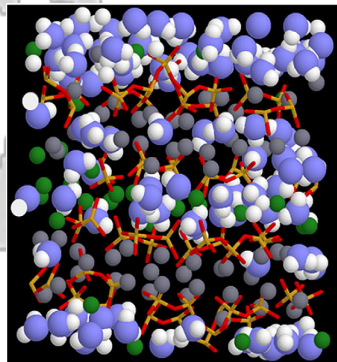
First calculations on hybrid systems

Vandervilt
Florence Sanchez



MIT CSHub

Ulm, Pellenq, et al. (also me)



First “realistic” C-S-H model
Disseminate simulations

France
Labbez



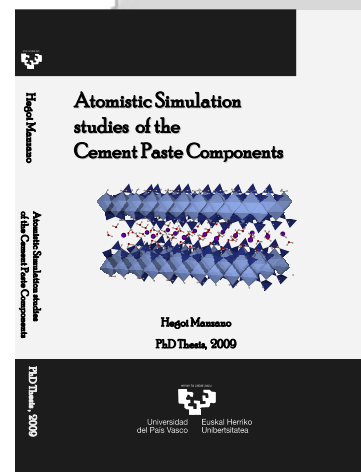
PSI (Switzerland)
Churakov



Basque Country

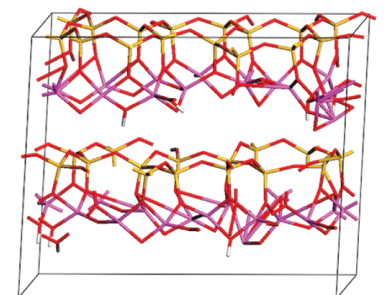
Dolado, Ayuela, Manzano

First determination of elastic properties
First calculations on clinker phases
First PhD thesis



Melbourne

White and Provis

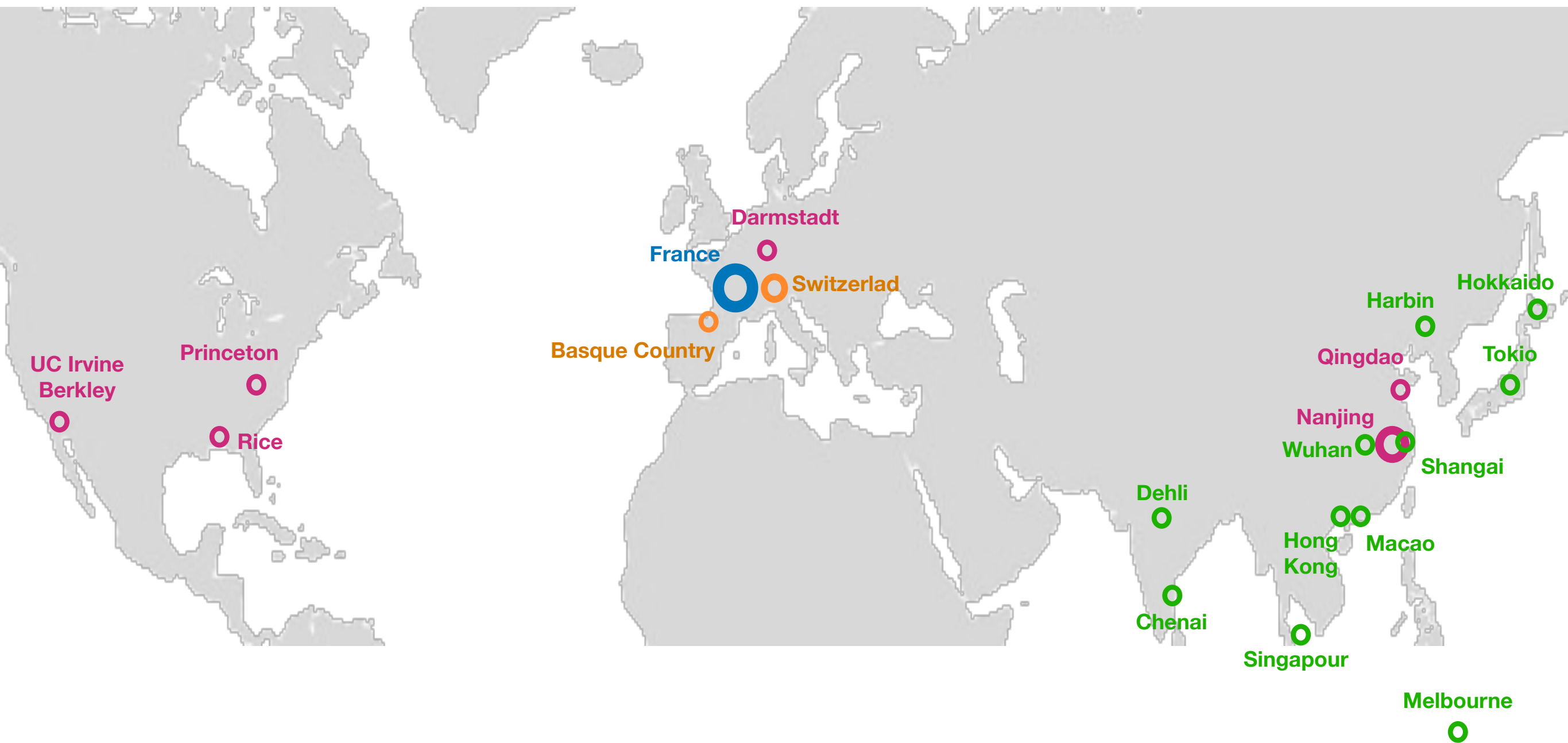


First calculations on SCM

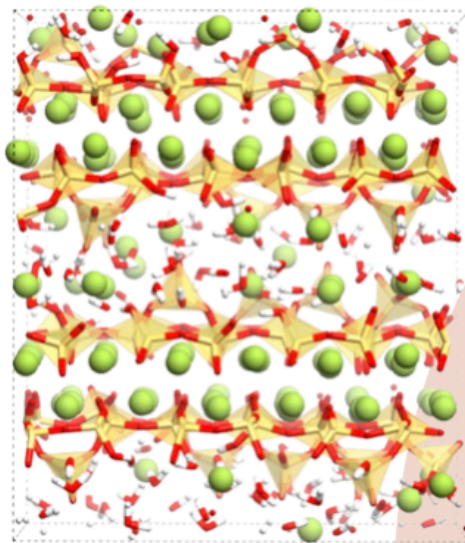
2012 to 2017: generalization of the technique



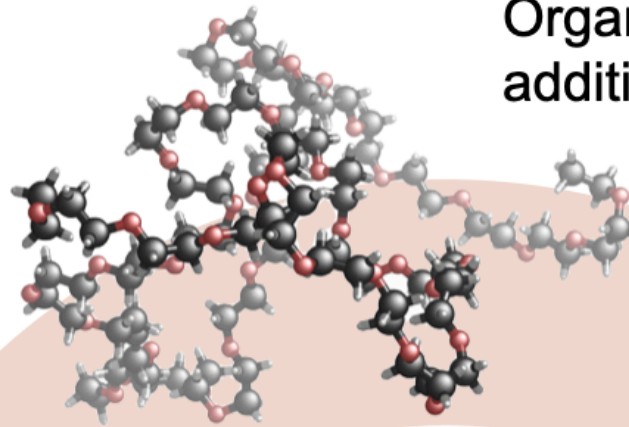
2018 to present: the BOOM!



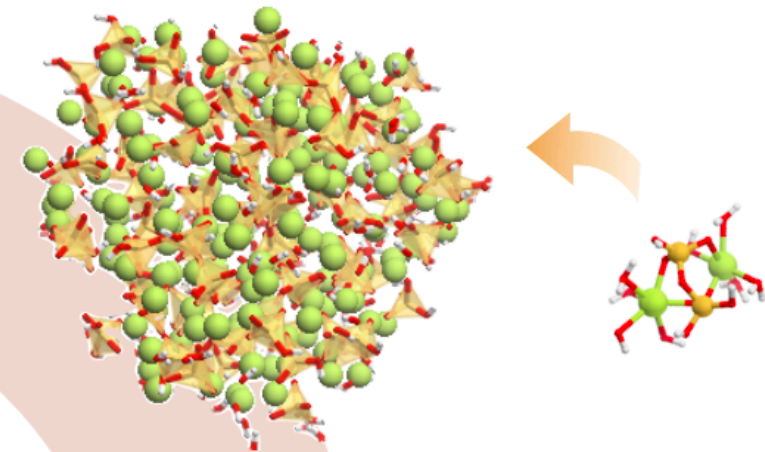
C-S-H structure and properties: confined water, ion transport, etc



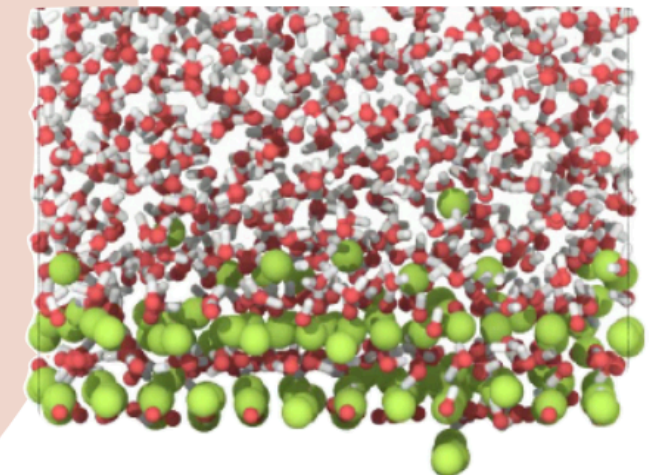
Organic additives



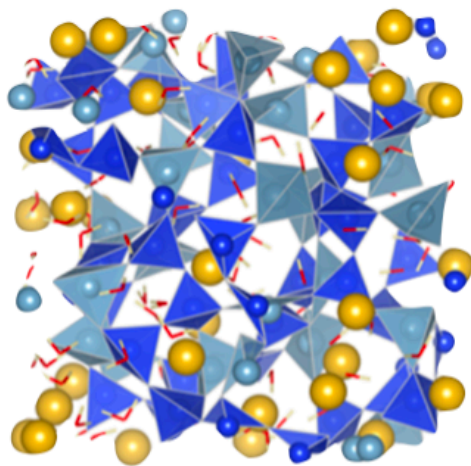
Nucleation and growth of hydrates



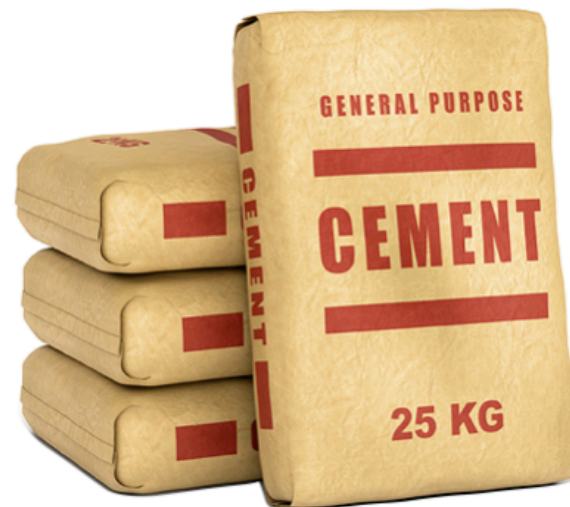
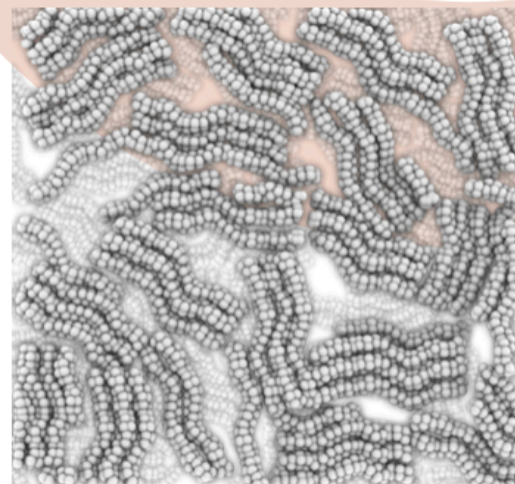
Water/solid interfaces, hydration and dissolution



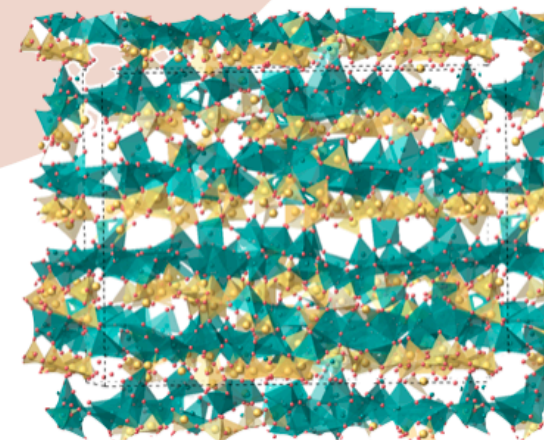
Alternative binders: geopolimers, Mg-based

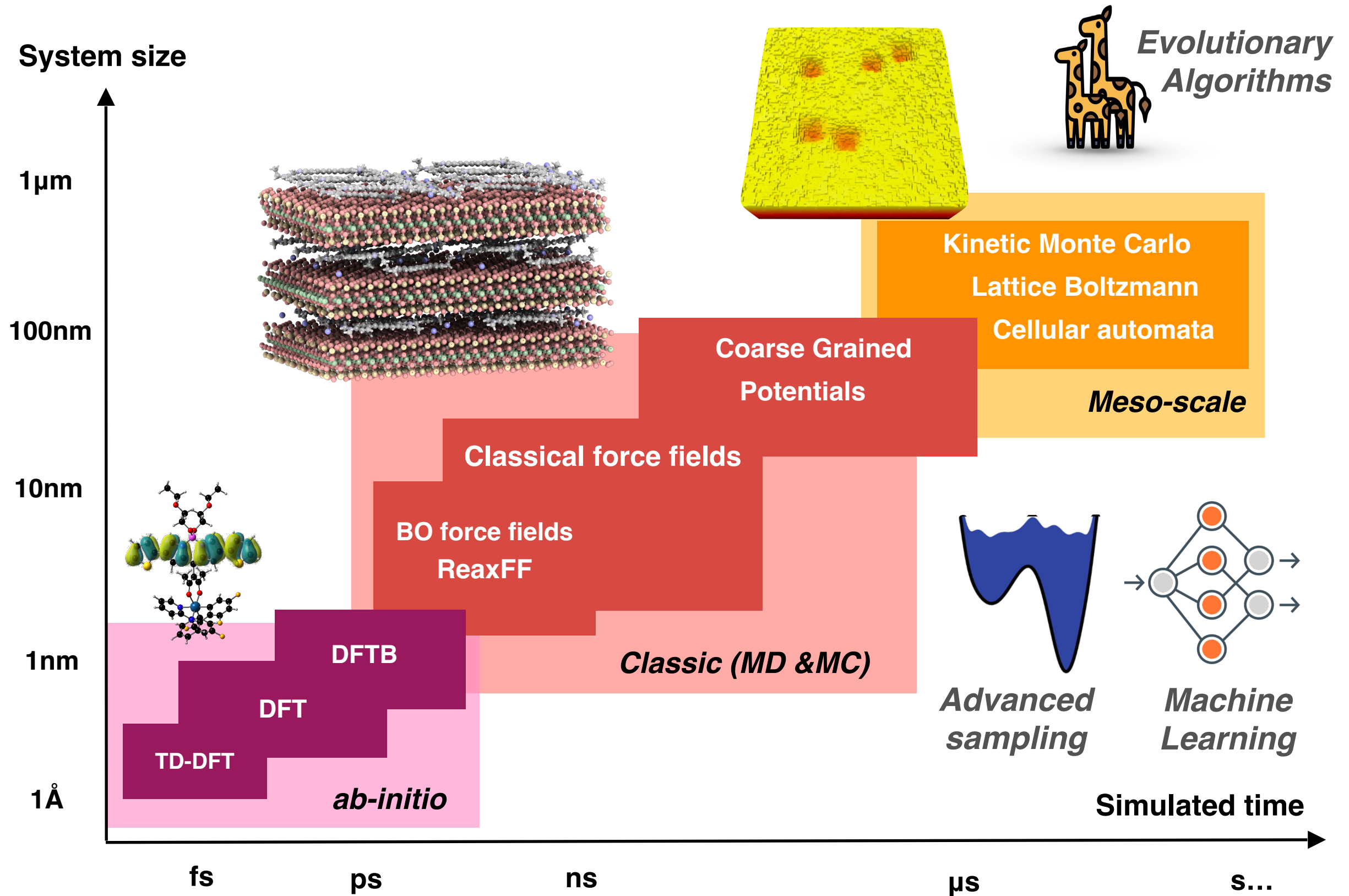


Coarse-grain and mesoscale modelling



SCM: fly ashes, slags, metakaolin





My 2005 self



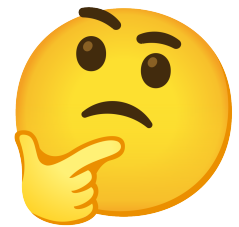
“We can go beyond laboratory limitations, do thousands of simulations to test materials and calculate properties, and design cement from electrons to buildings”

My 2005 self



“We can go beyond laboratory limitations, do thousands of simulations to test materials and calculate properties, and design cement from electrons to buildings”

20 years later



“In practice, the number of atomistic simulation studies that truly imply a practical advance or guide the design of cement towards enhanced performance is limited”

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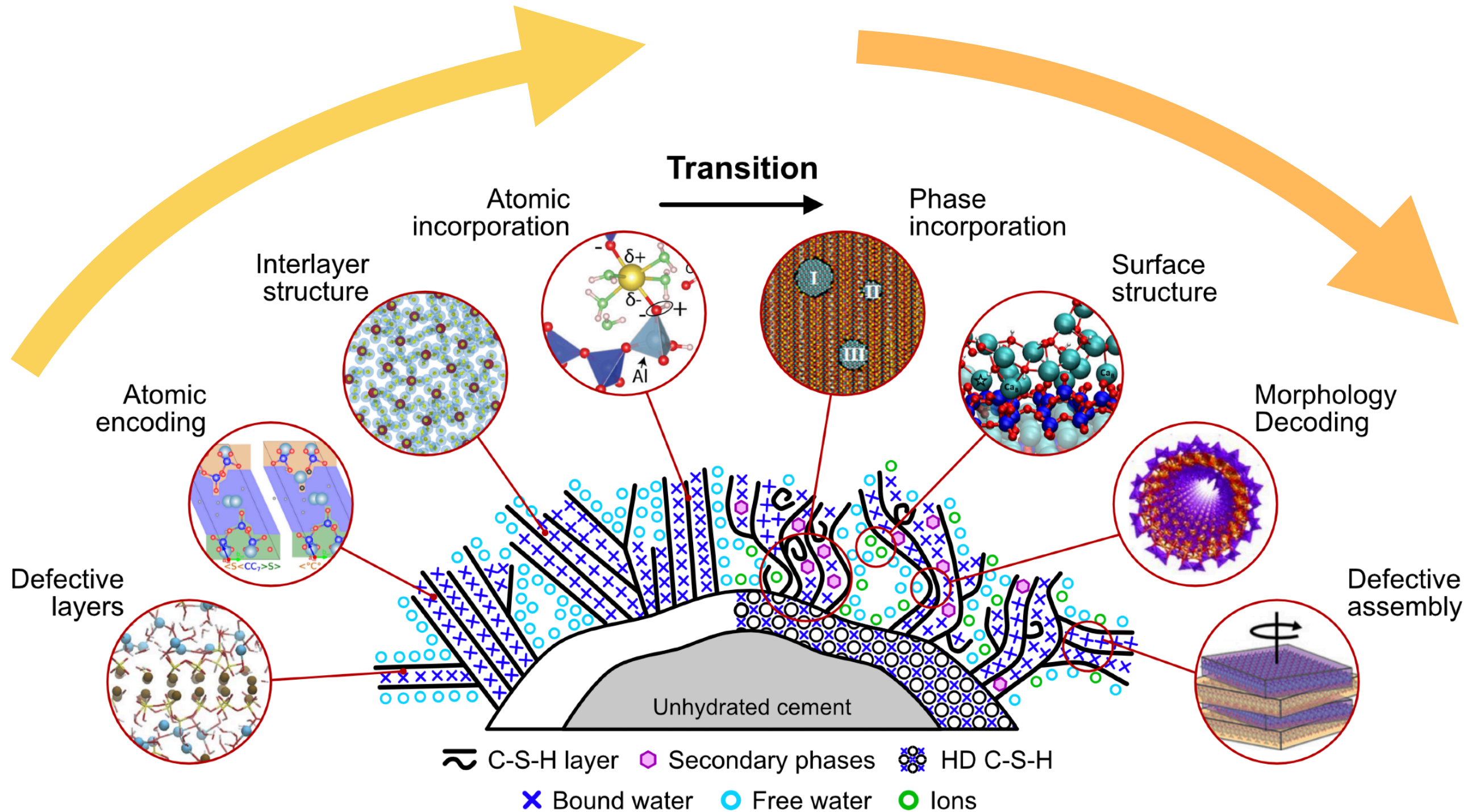
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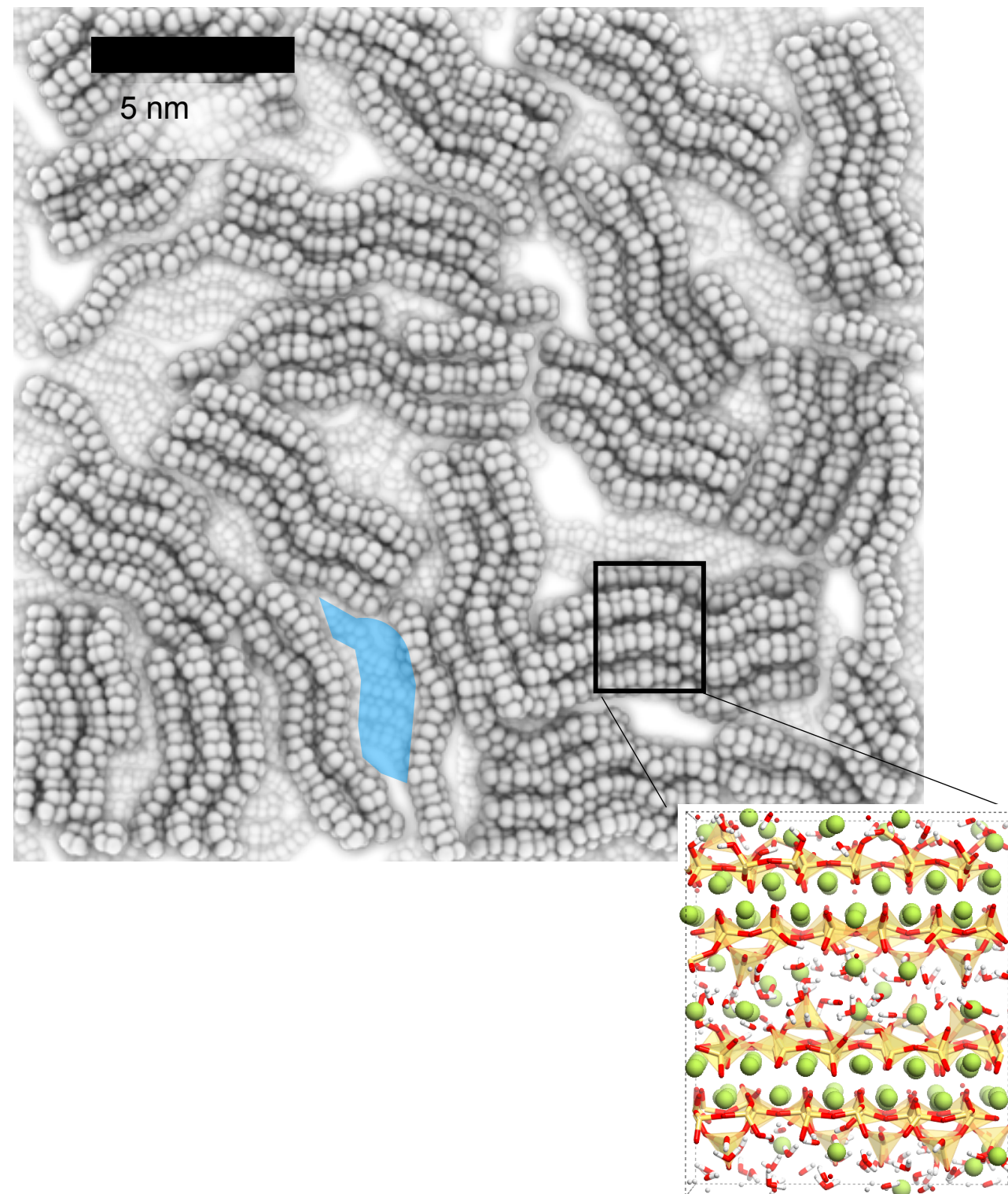
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Intrinsic scale limitations, lack of contextualization, incomplete models, lack of experimental focus,...



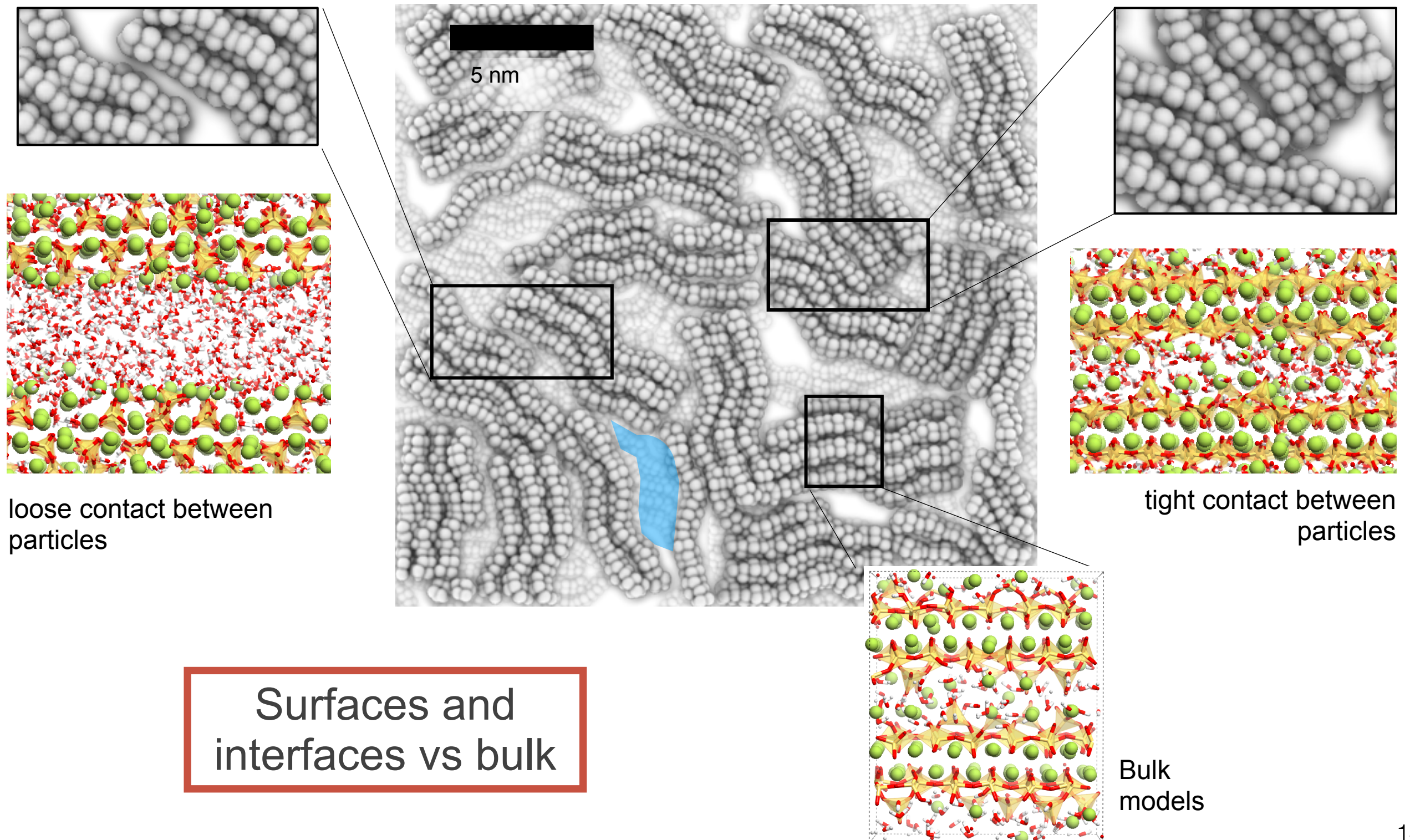
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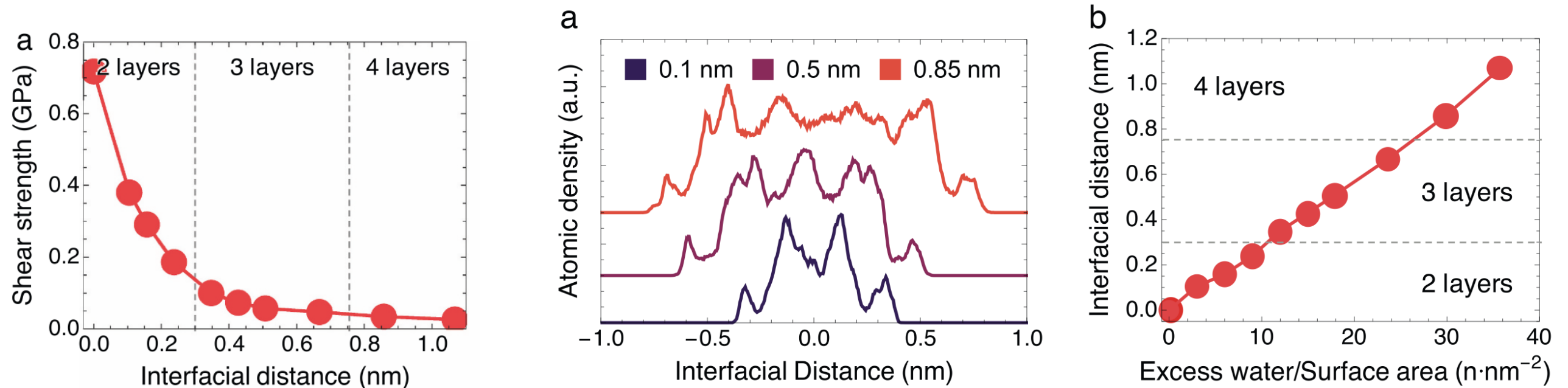


Calcium Silicate Hydrate: interfaces

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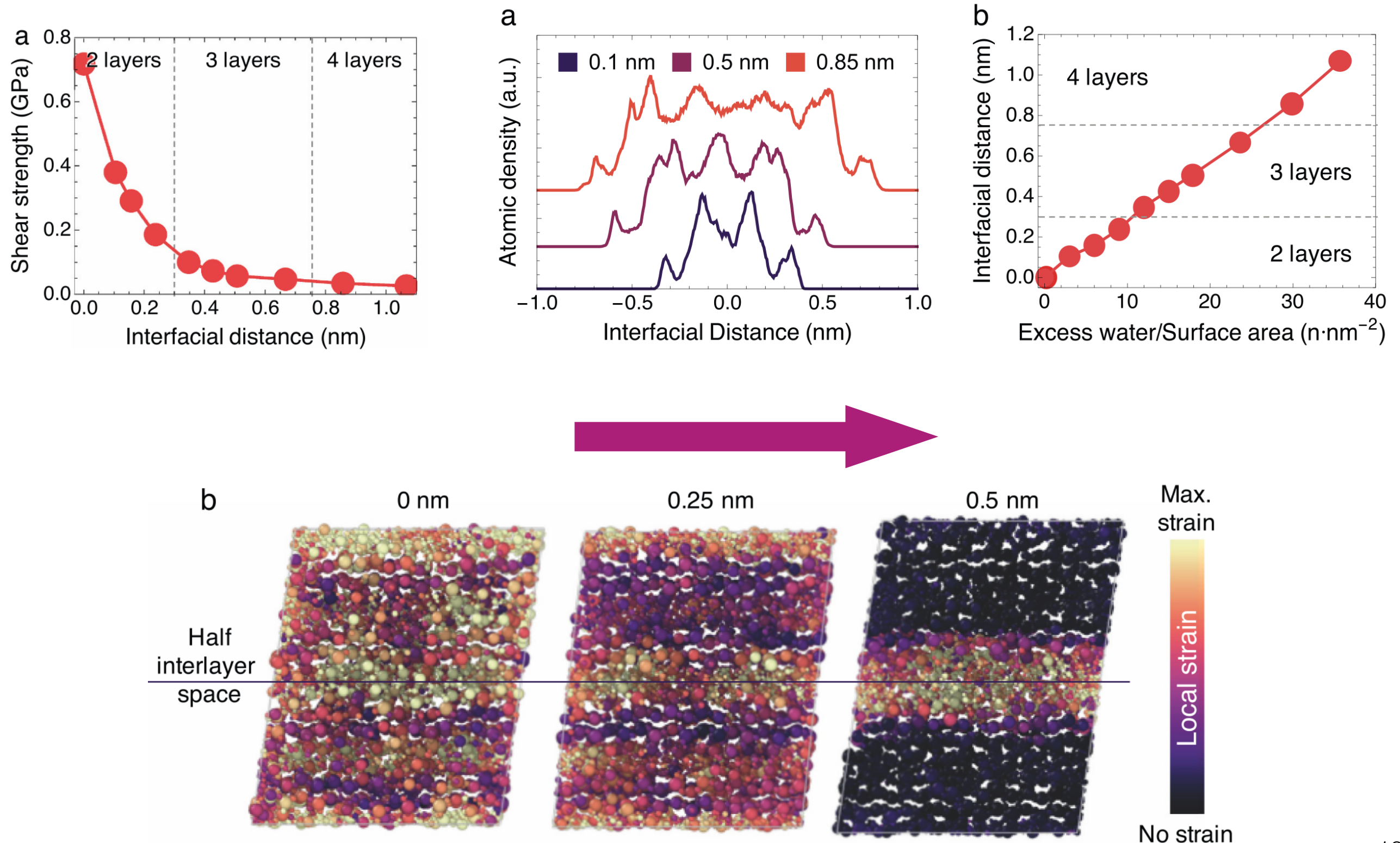


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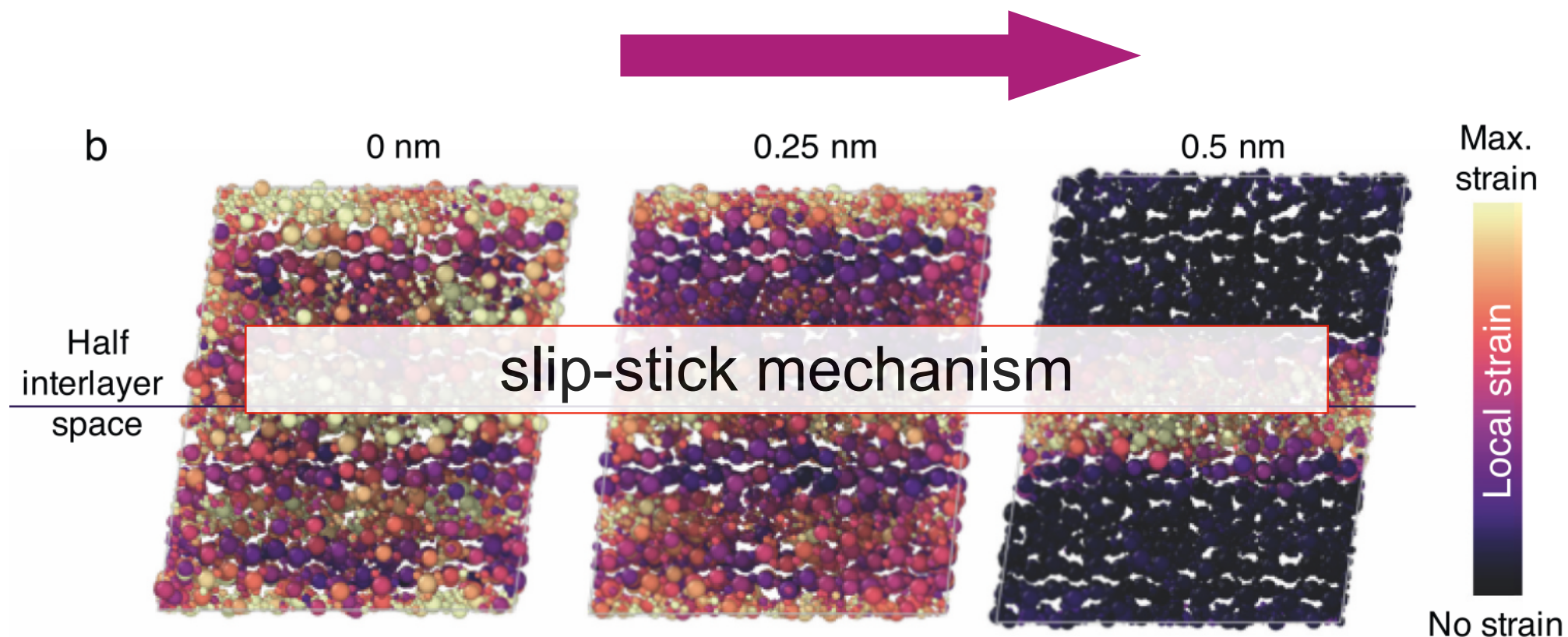
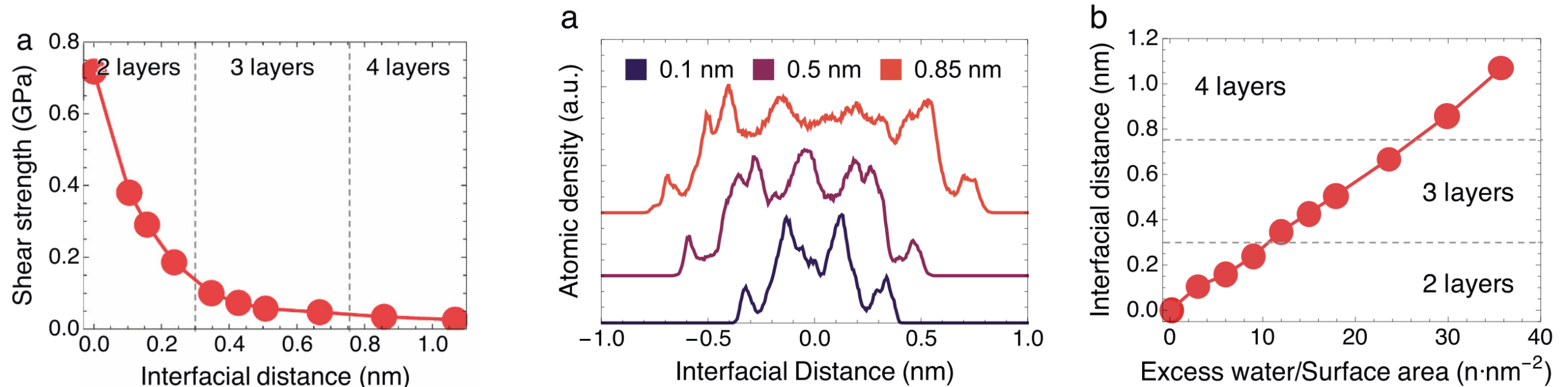
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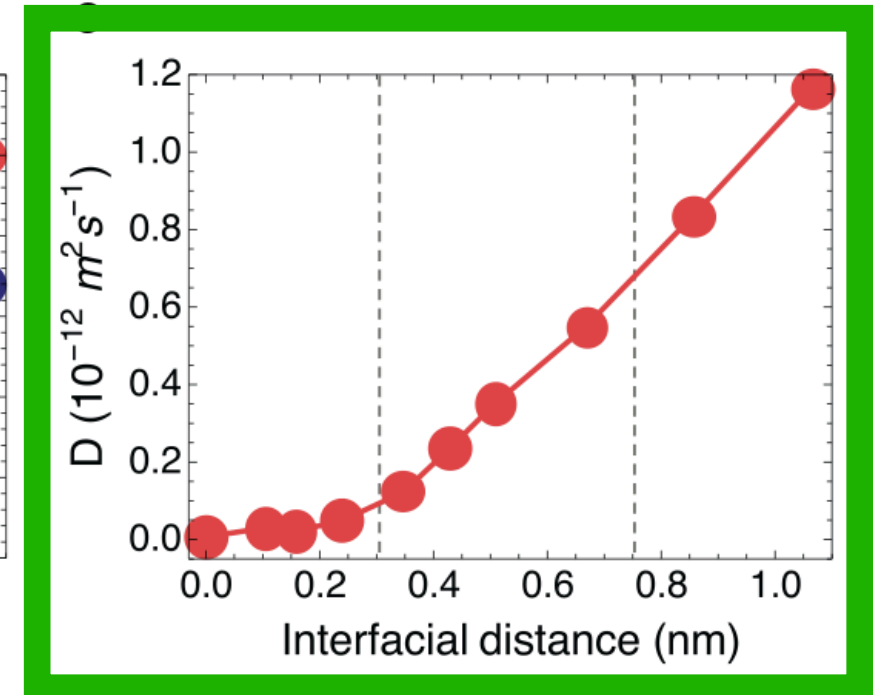
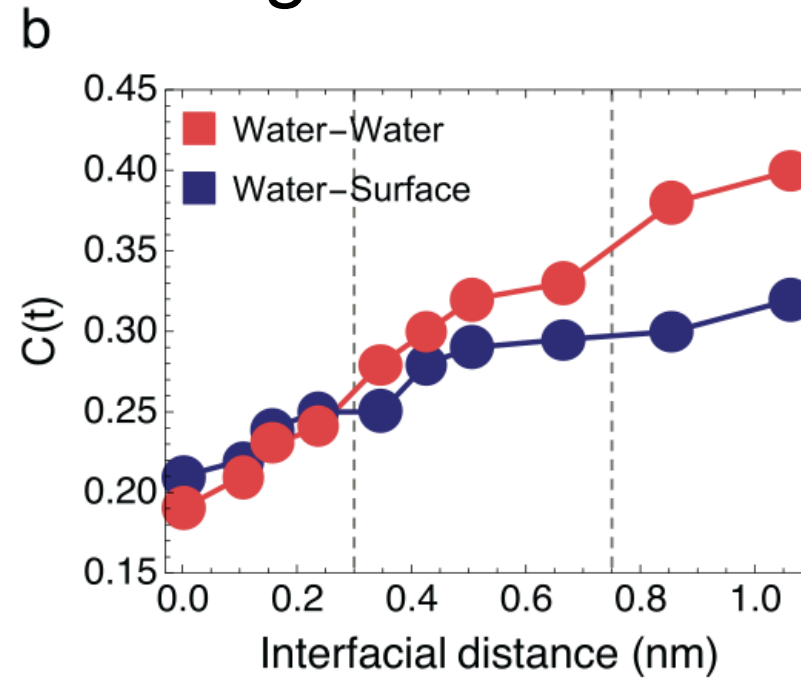
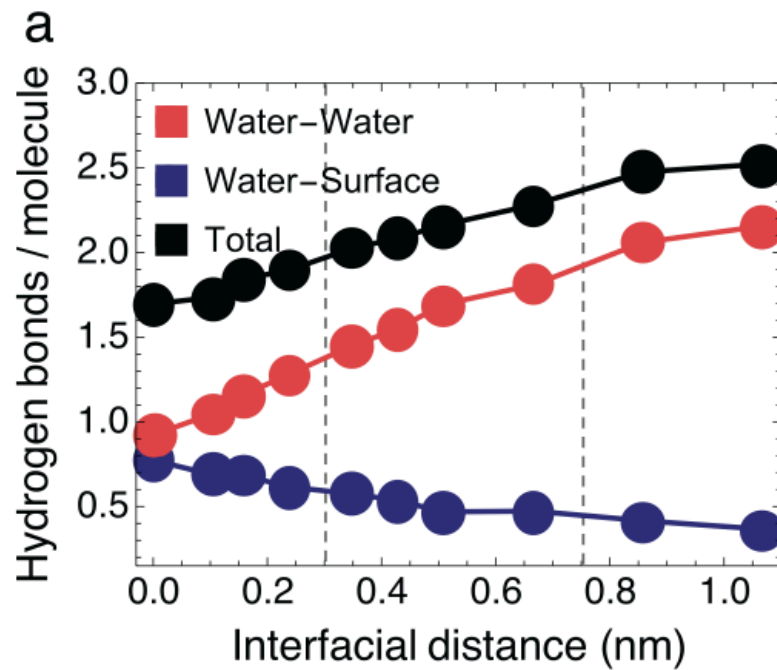


Calcium Silicate Hydrate: interfaces

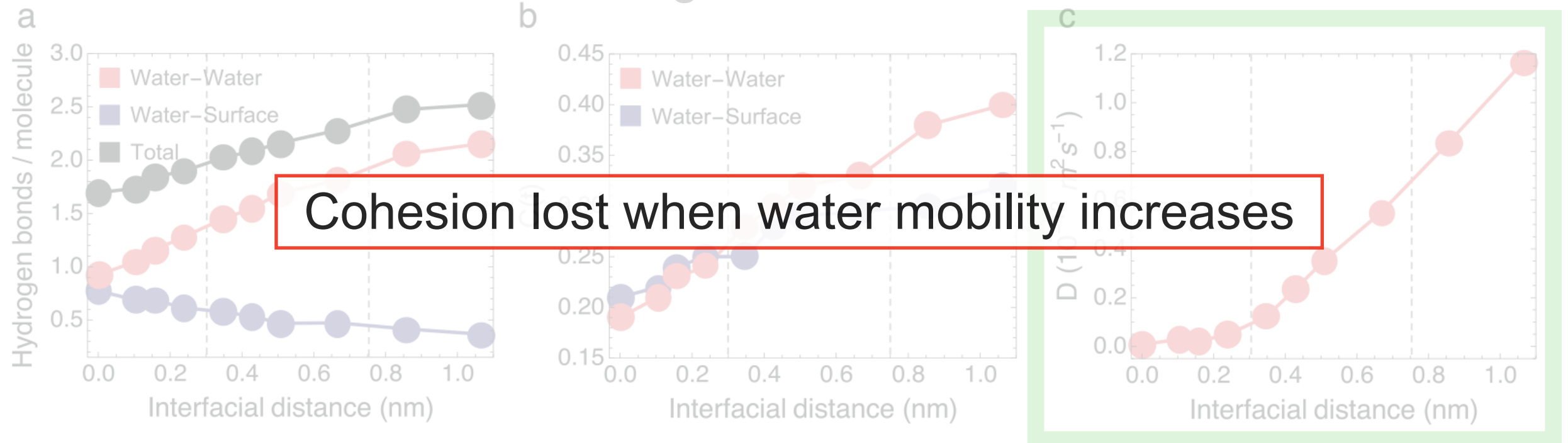
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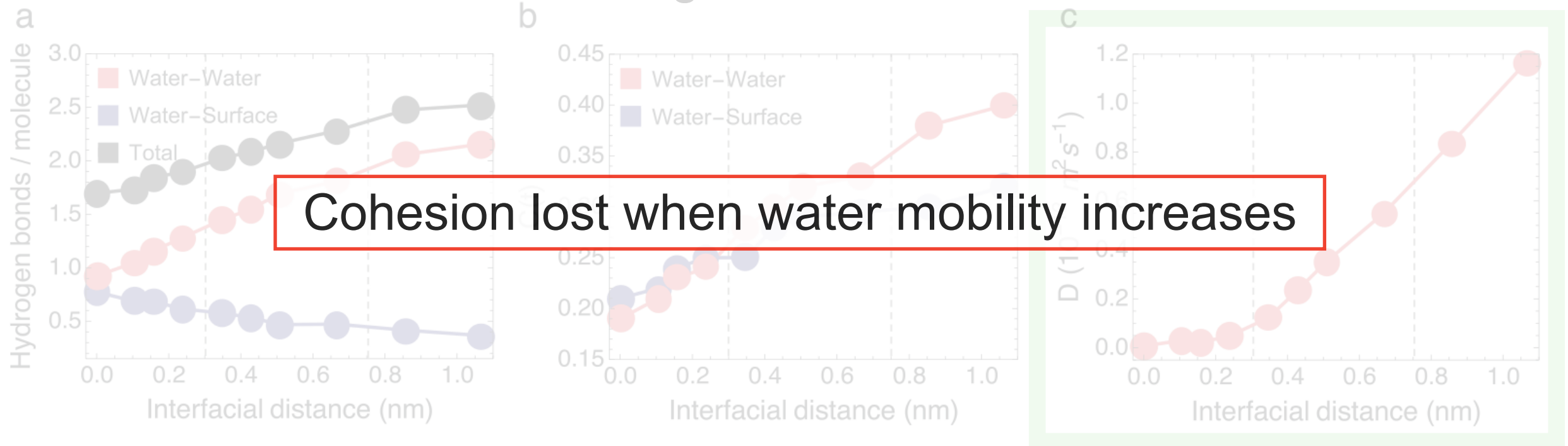
What changes at the interface?



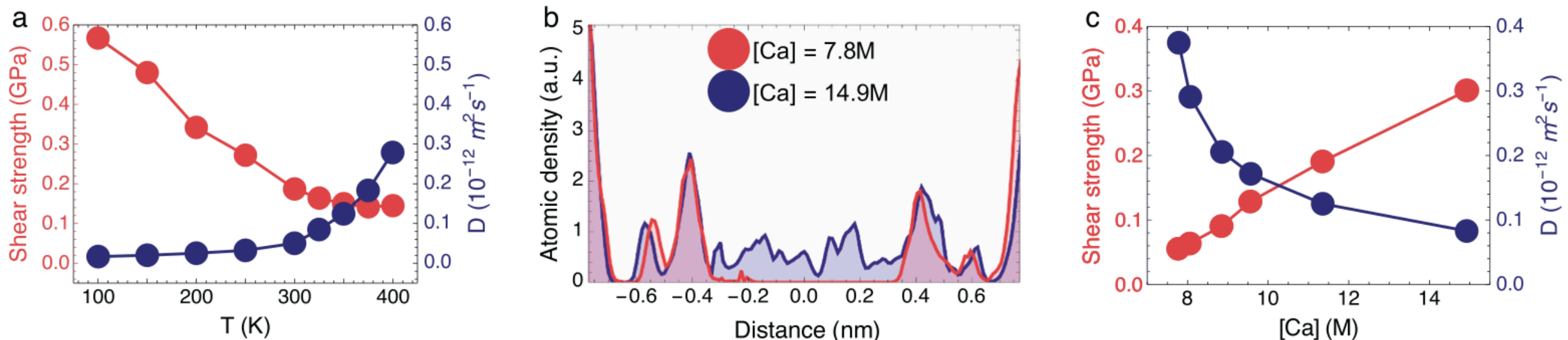
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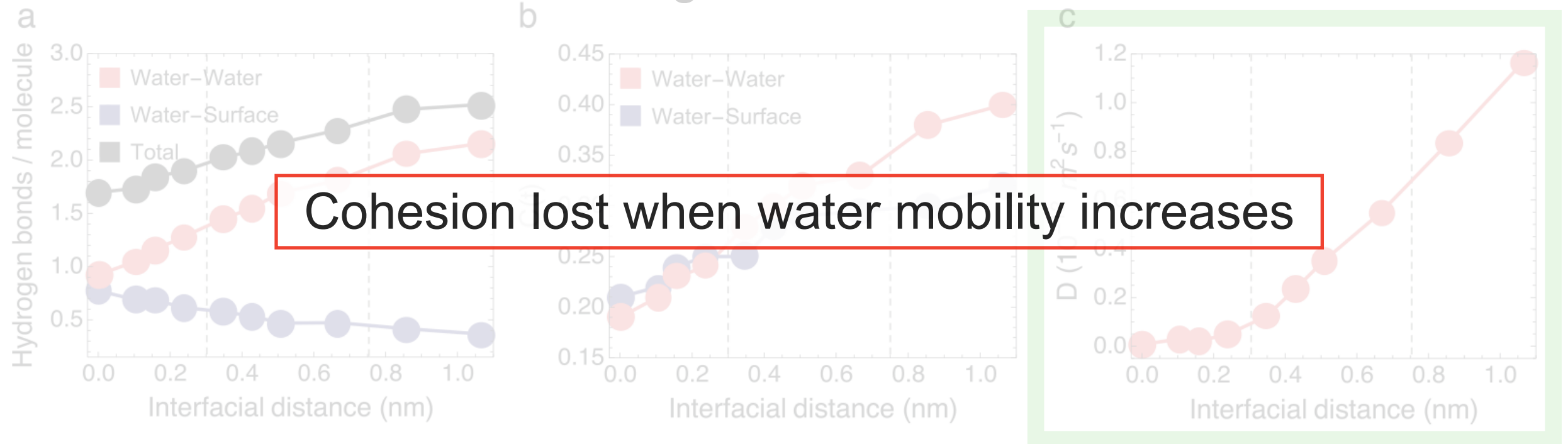
What changes at the interface?



Control water mobility to test the mechanism



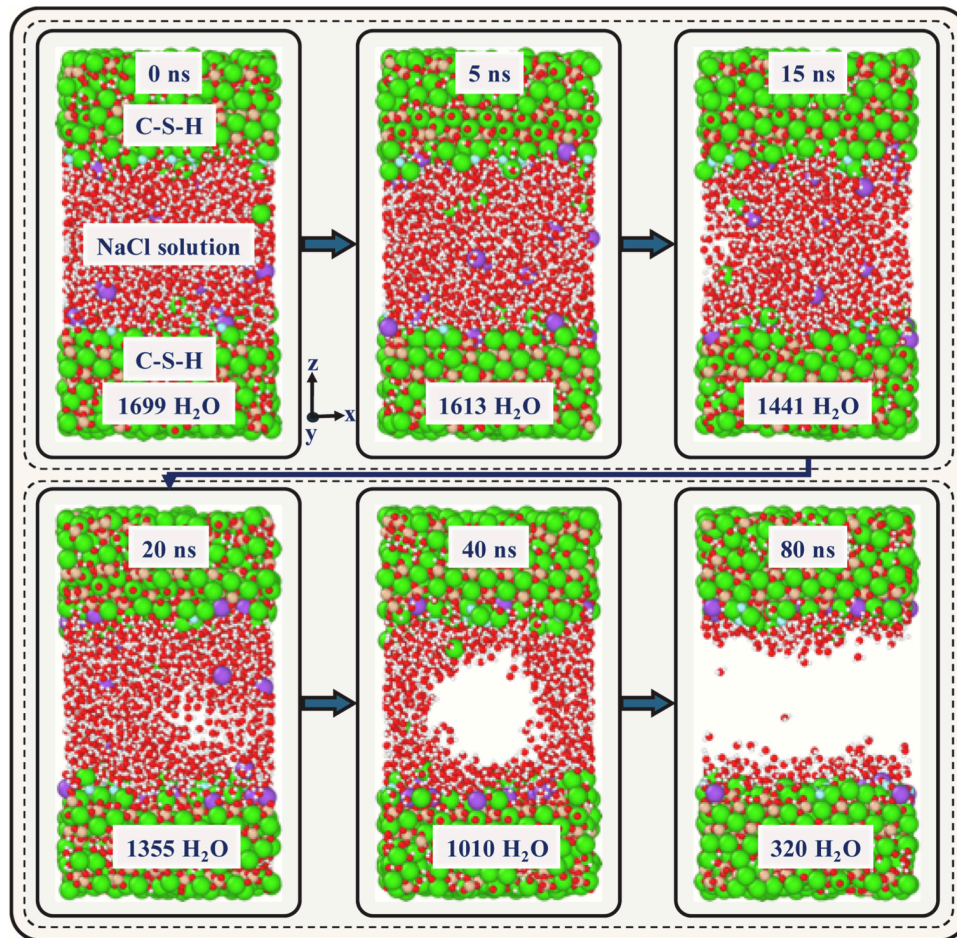
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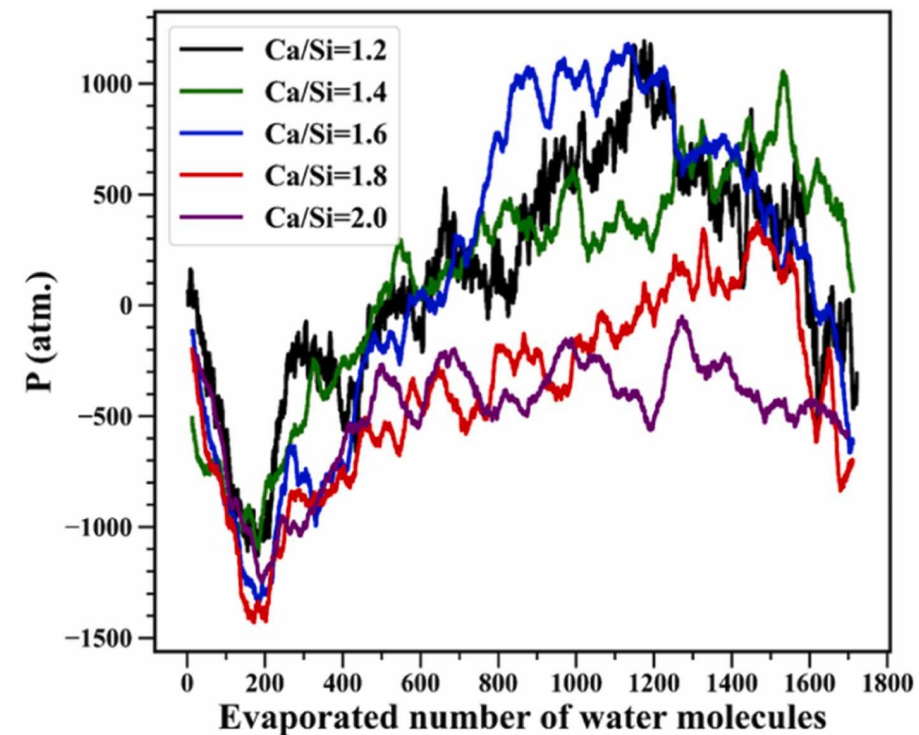
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Umar Hayat, Eduardo Duque-Redondo, Ming-Feng Kai, Hegoi Manzano, Muhammad Riaz Ahmad, You Dong, Jian-Guo Dai, Desorption of water from aqueous solution confined in CSH gel pore: A molecular dynamics study Construction and Building Materials 490, 142602

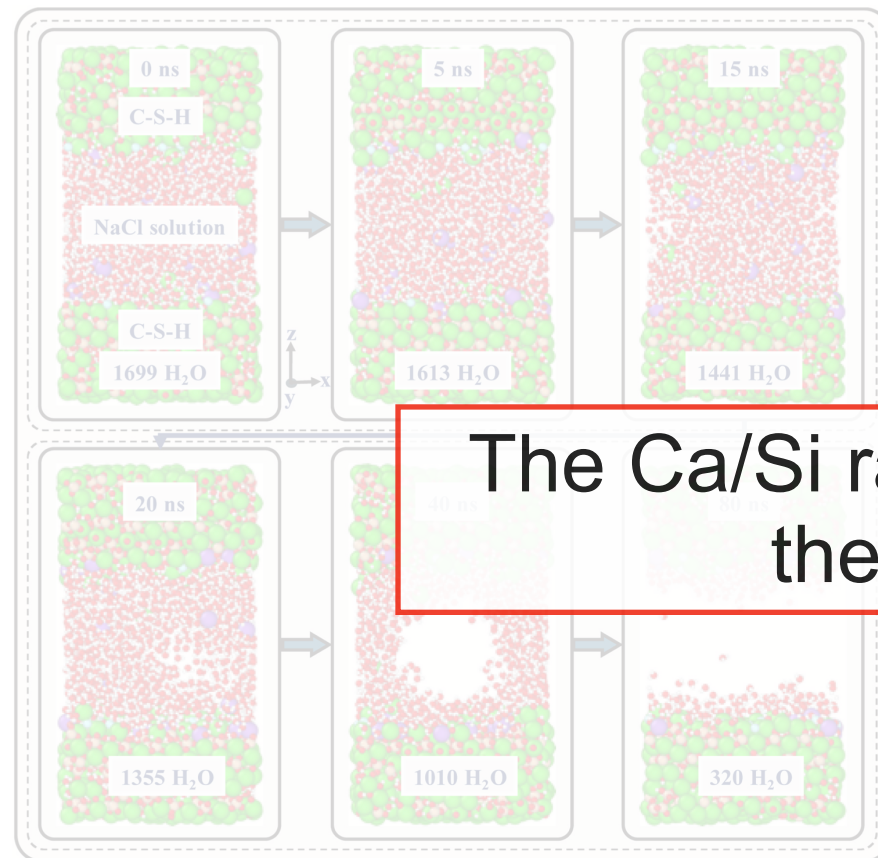


The Ca/Si ratio does not have significant impact

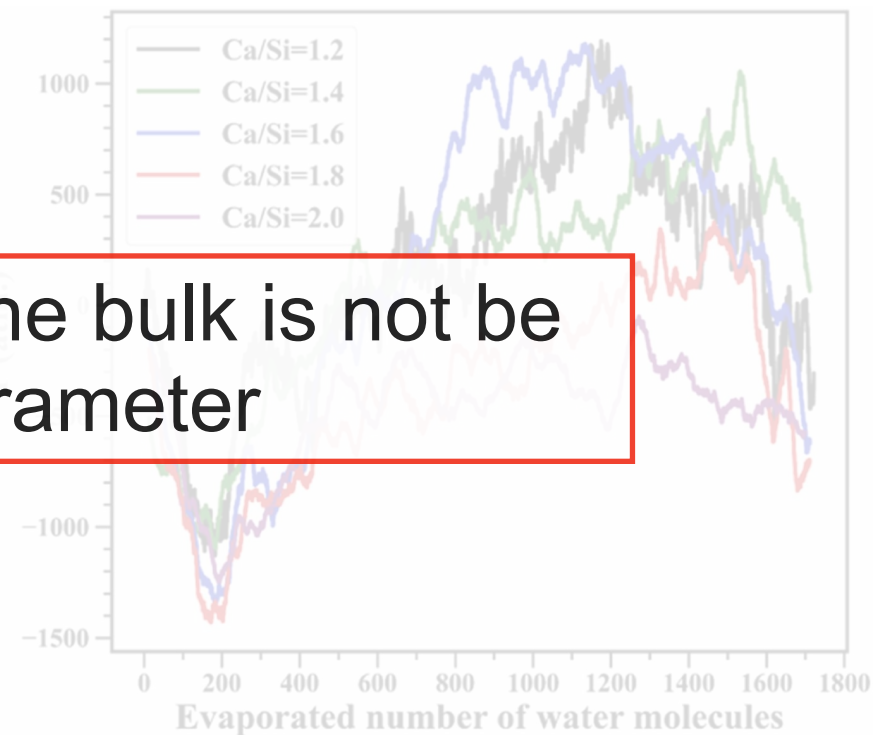


Calcium Silicate Hydrate: Microstructure

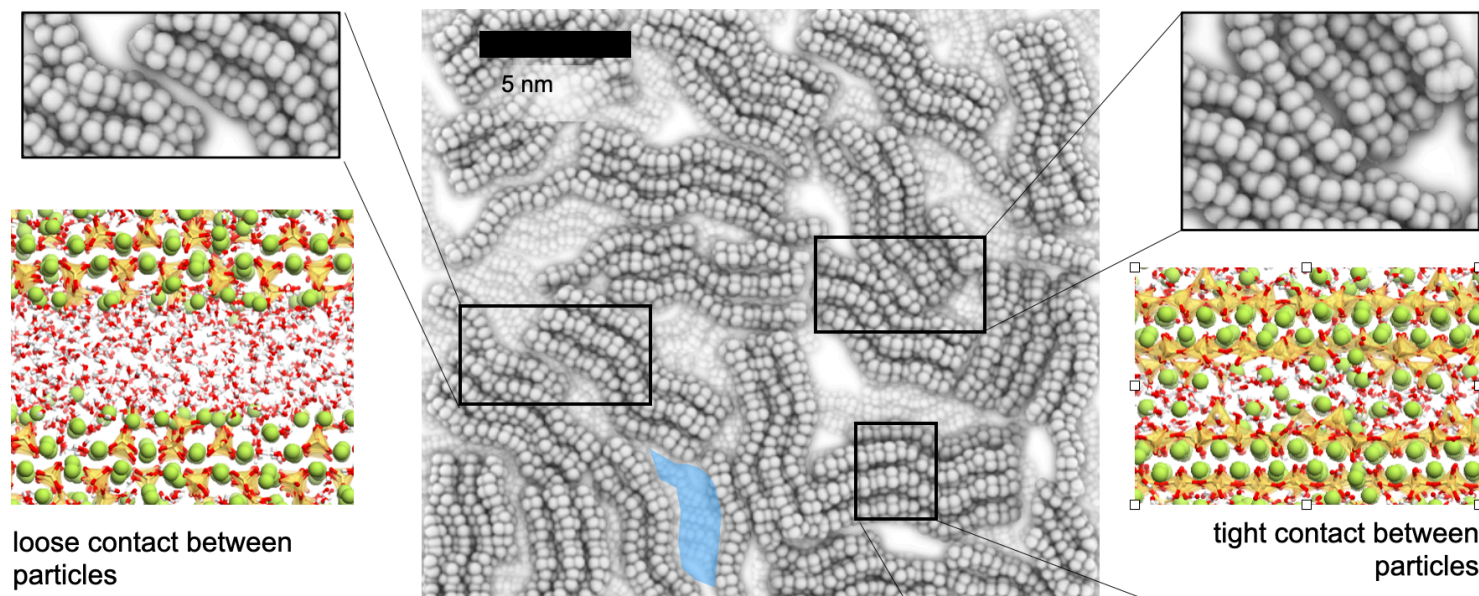
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The Ca/Si ratio does not have significant impact



The Ca/Si ratio of the bulk is not be the key parameter



My 2005 self



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20 years later

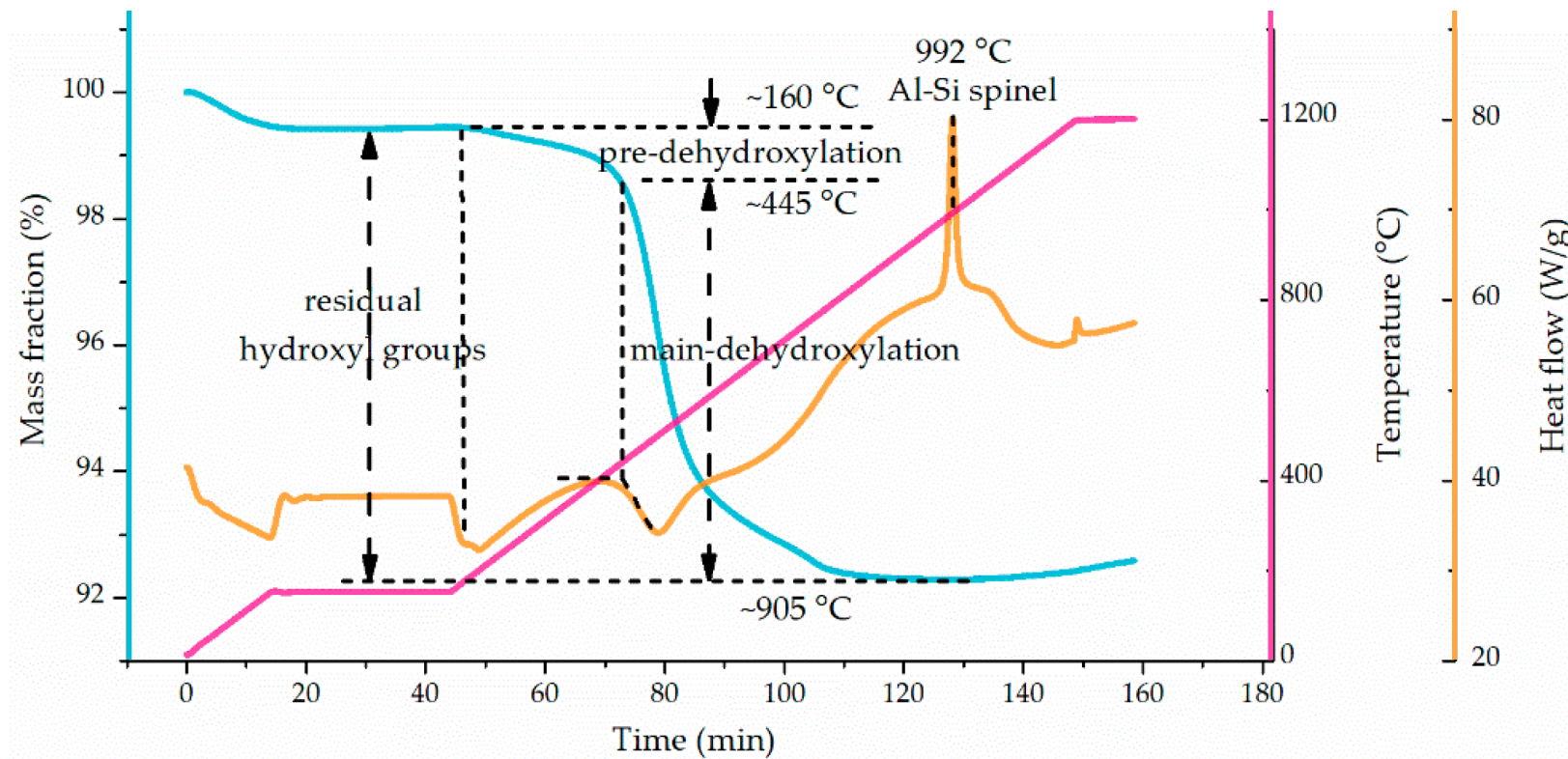


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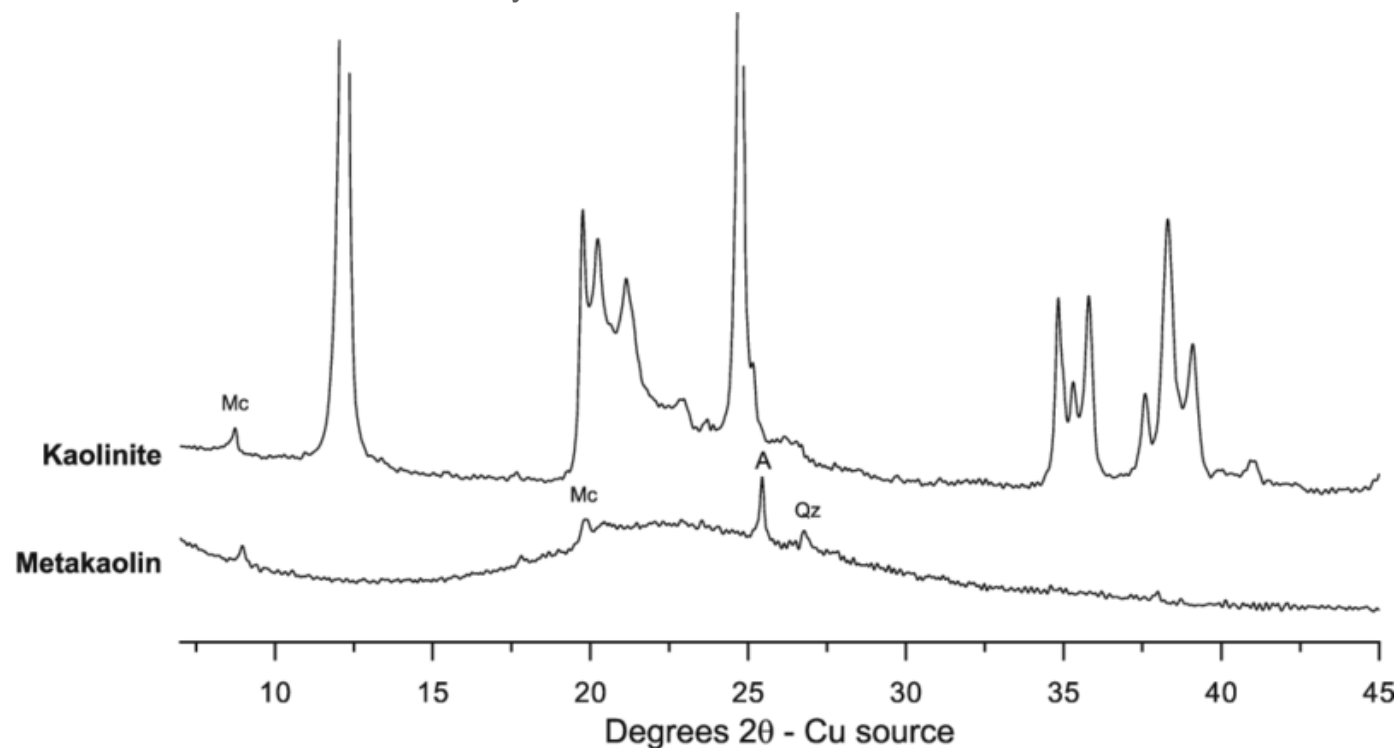
Intrinsic scale limitations, lack of contextualization, incomplete models, lack of experimental focus,...

Kaolinite to Metakaolin

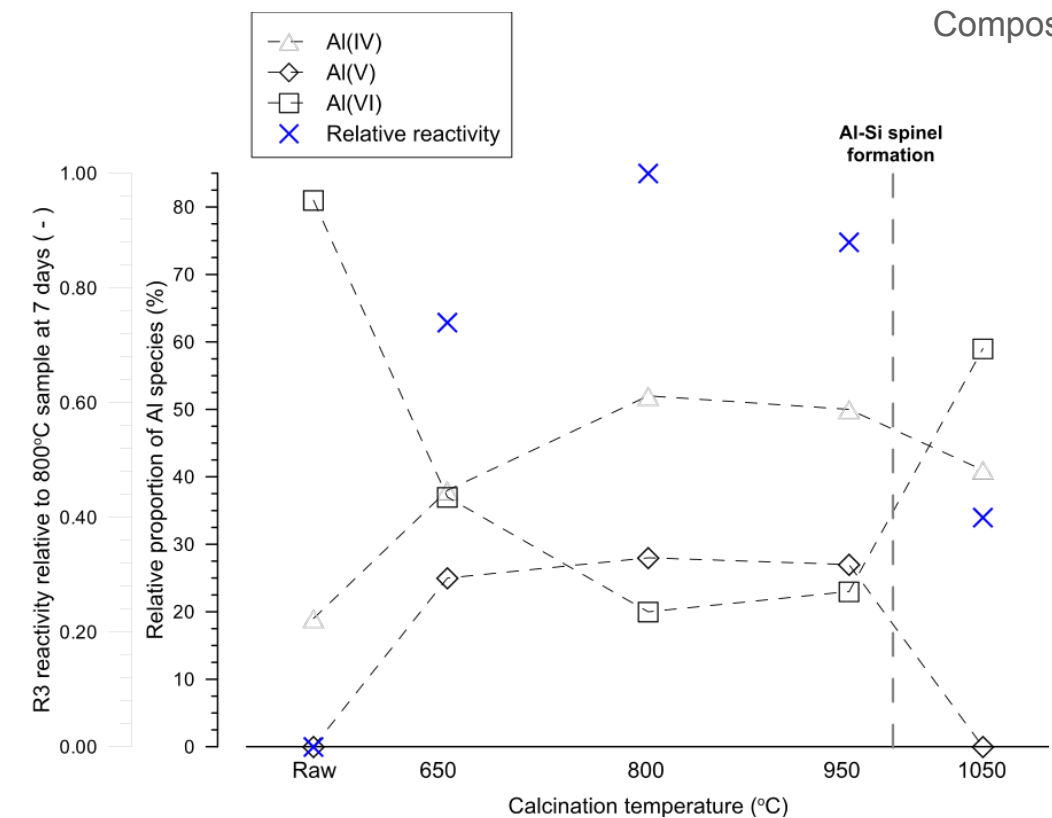


Adapted from: Cheng, et al. (2019).
Dehydroxylation and structural
distortion of kaolinite as a high-
temperature sorbent in the furnace.
Minerals, 9(10), 587

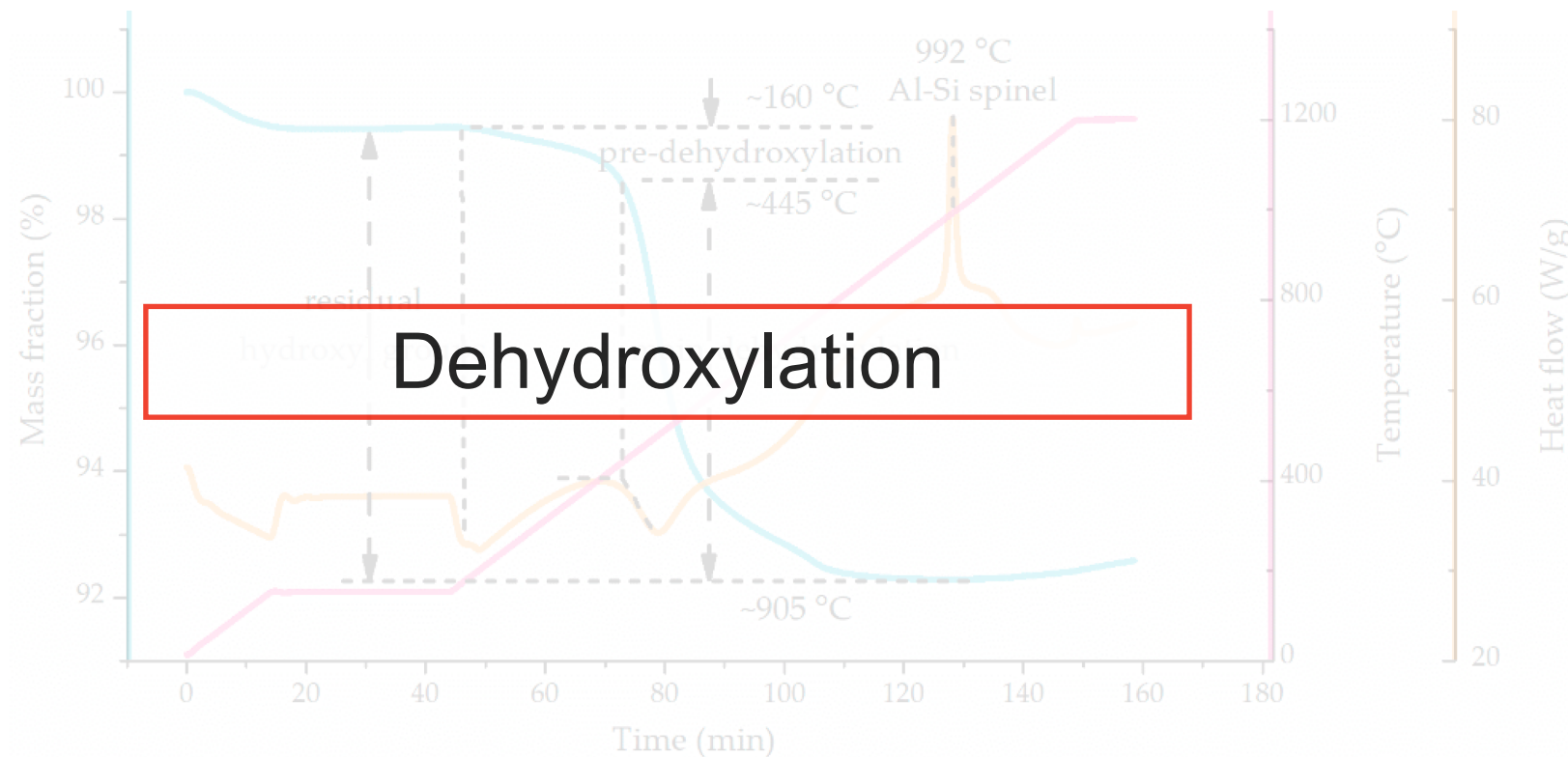
Zunino, F., & Scrivener, K. (2022). Oxidation of pyrite (FeS_2) and troilite (FeS) impurities
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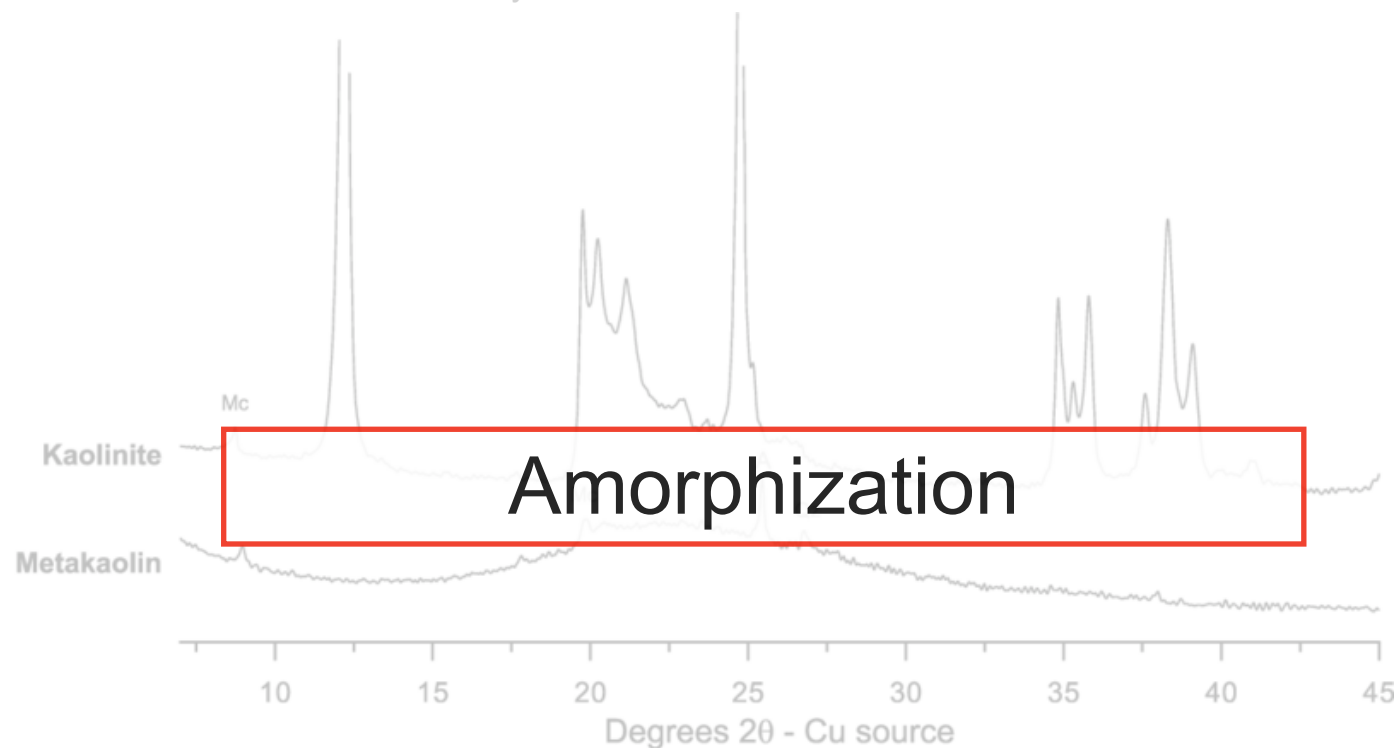


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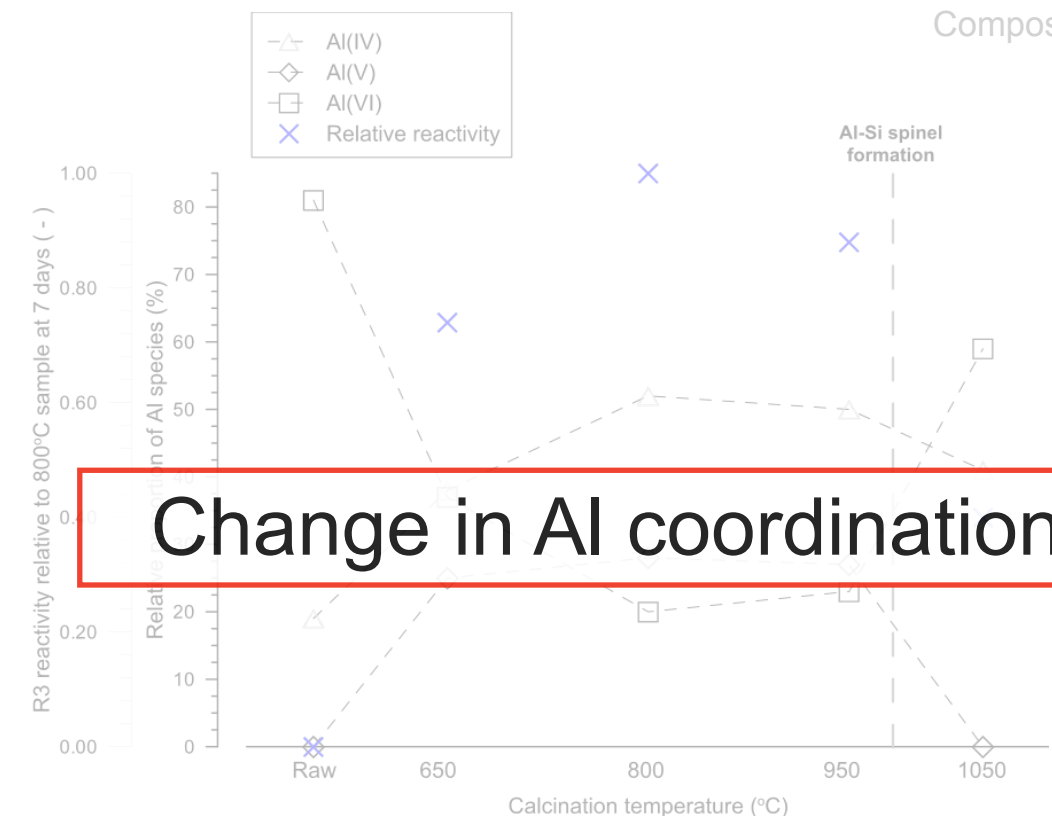


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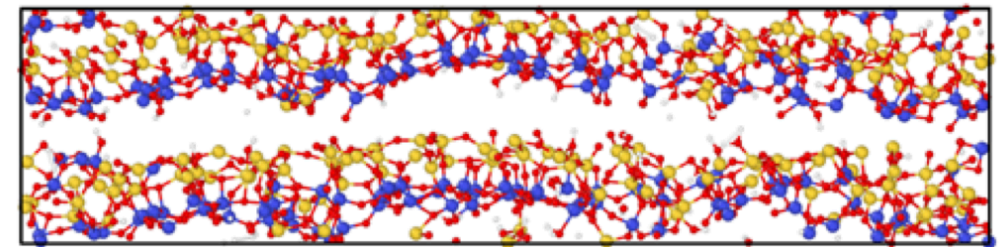
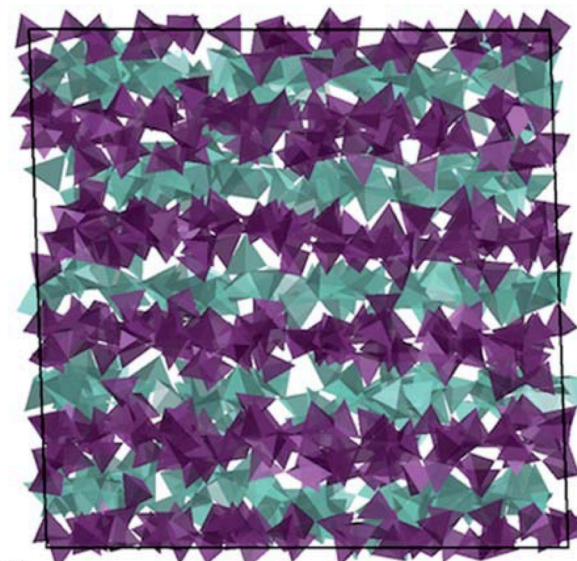
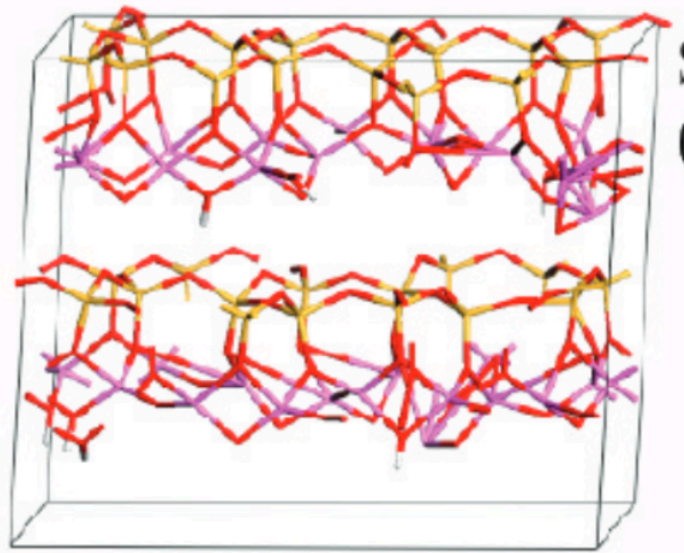


Metakaolin: current models

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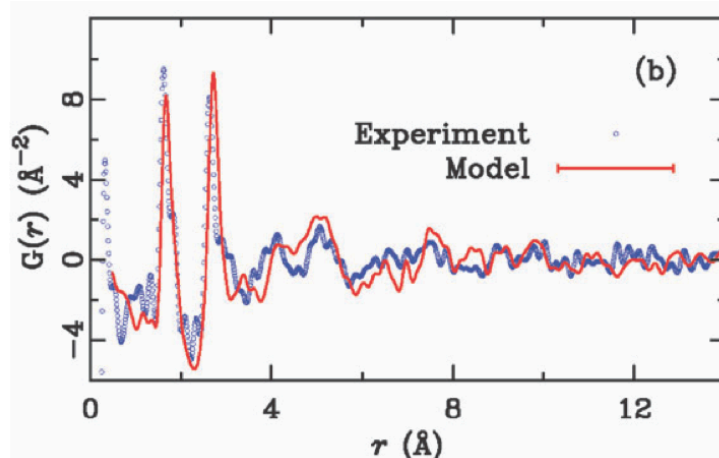
Muraleedharan, Murali Gopal, et al. "Elucidating thermally induced structural and chemical transformations in kaolinite using reactive molecular dynamics simulations and X-ray scattering measurements." Chemistry of Materials 32.2 (2019):



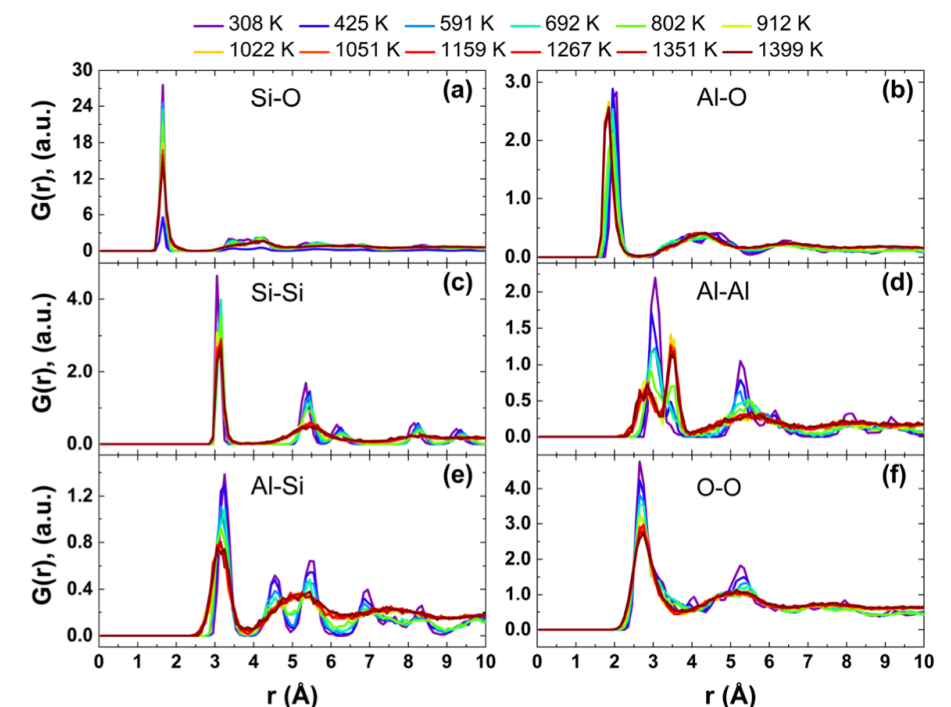
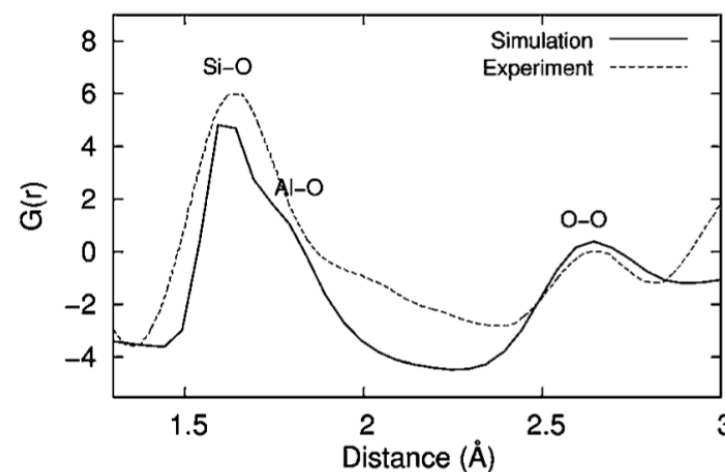
Dehydroxylation: spontaneous with T MD from 298L to 1300K, monolayer

Dehydroxylation: progressive by hand

DFT relaxation (0K)



MD 1000K

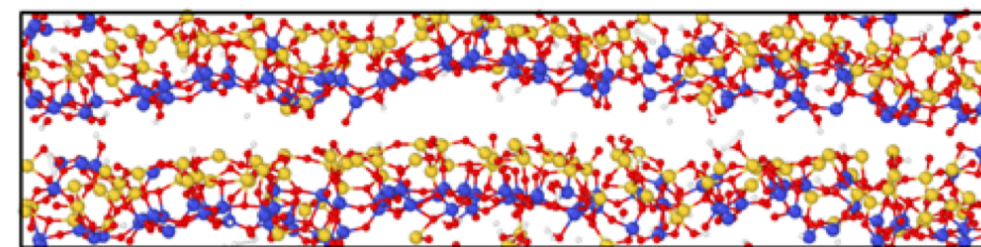
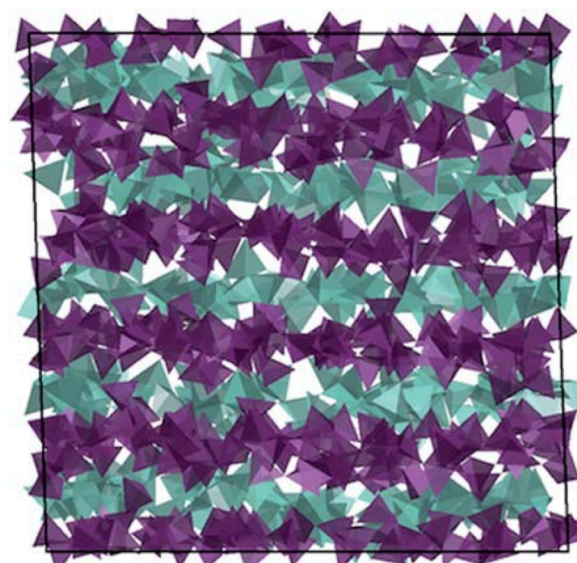
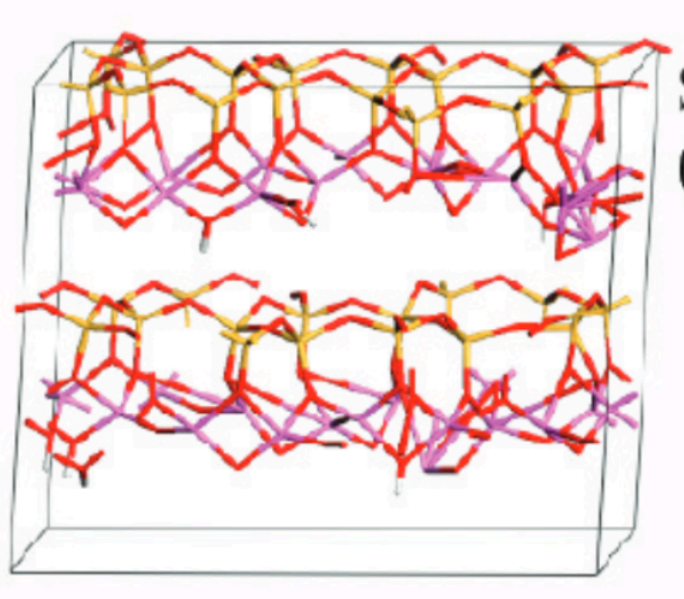


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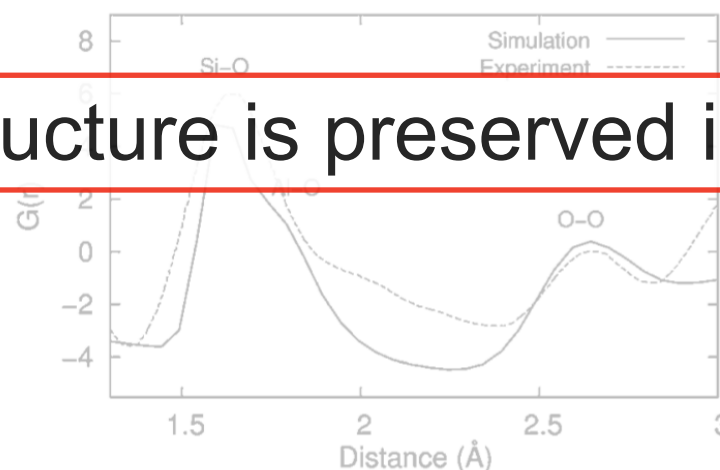
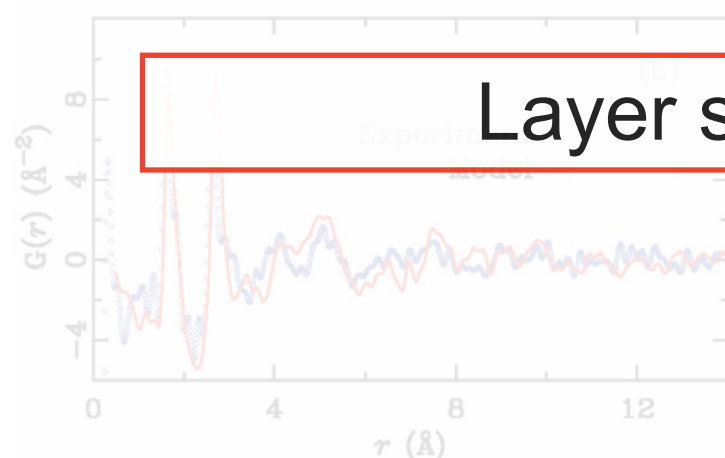


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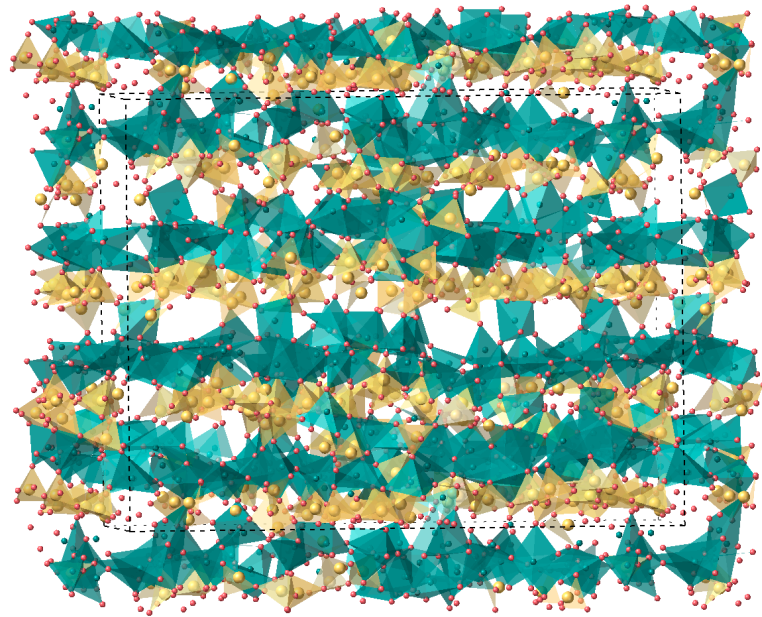
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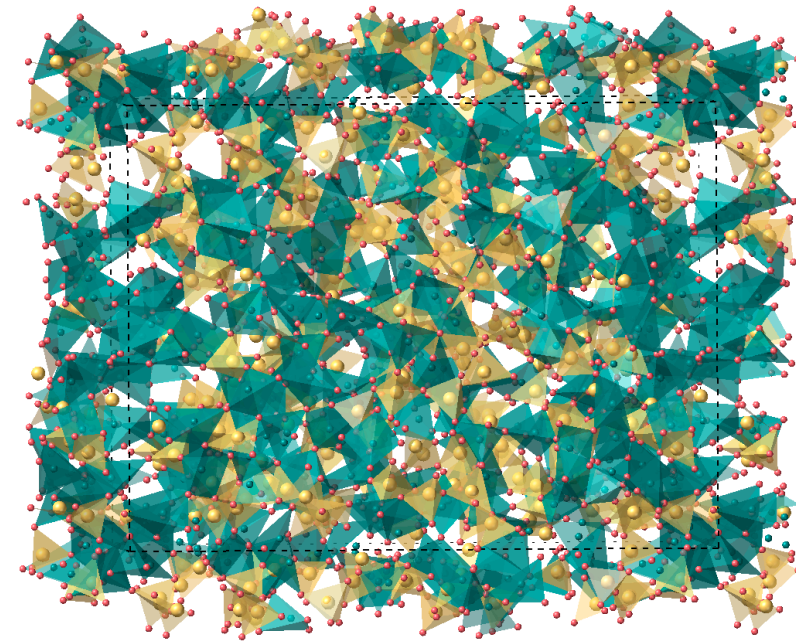
Layer structure is preserved in current models



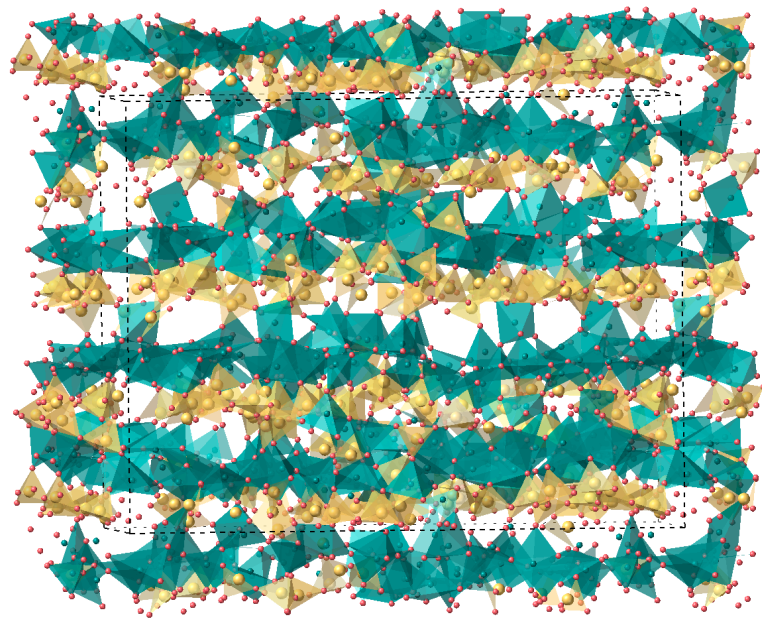
Dehydroxylation protocols



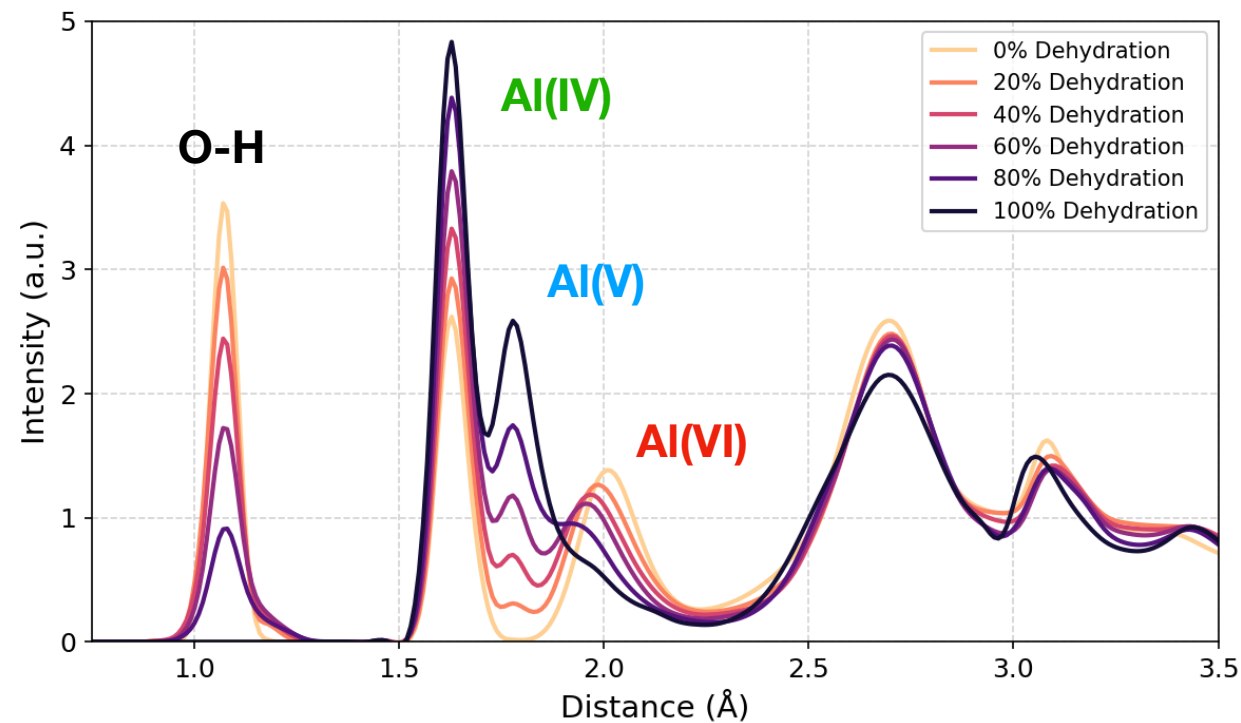
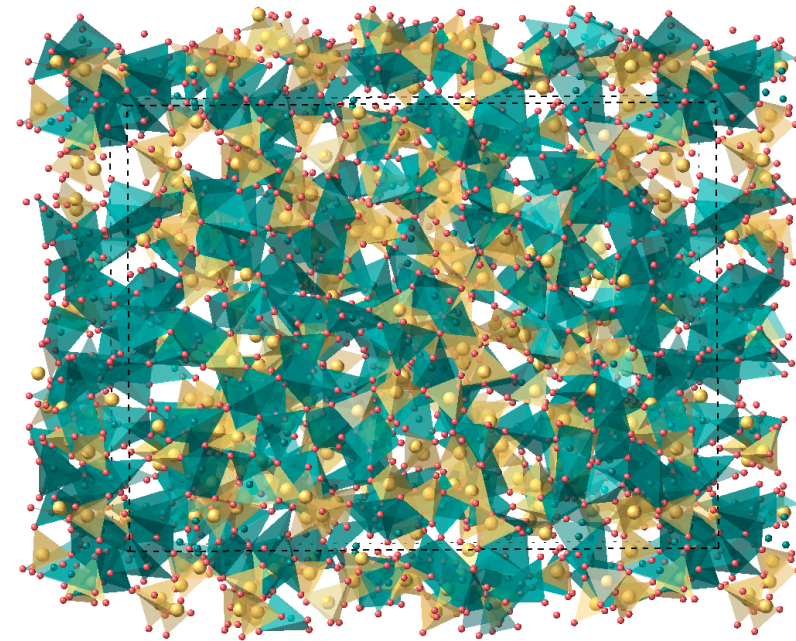
Melting - quenching



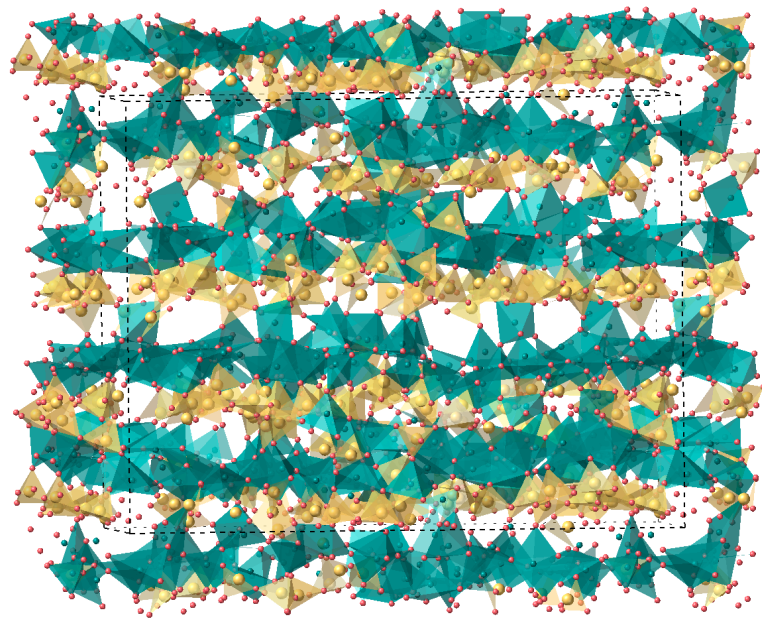
Dehydroxylation protocols



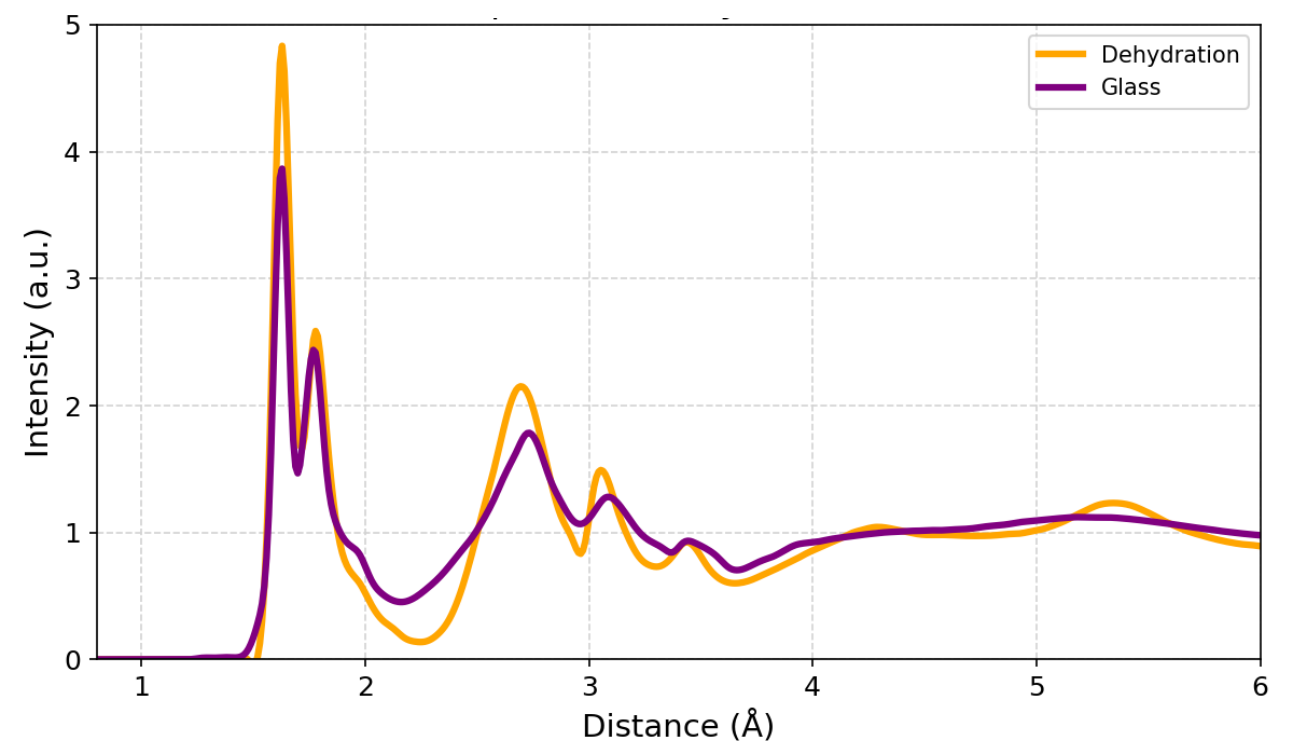
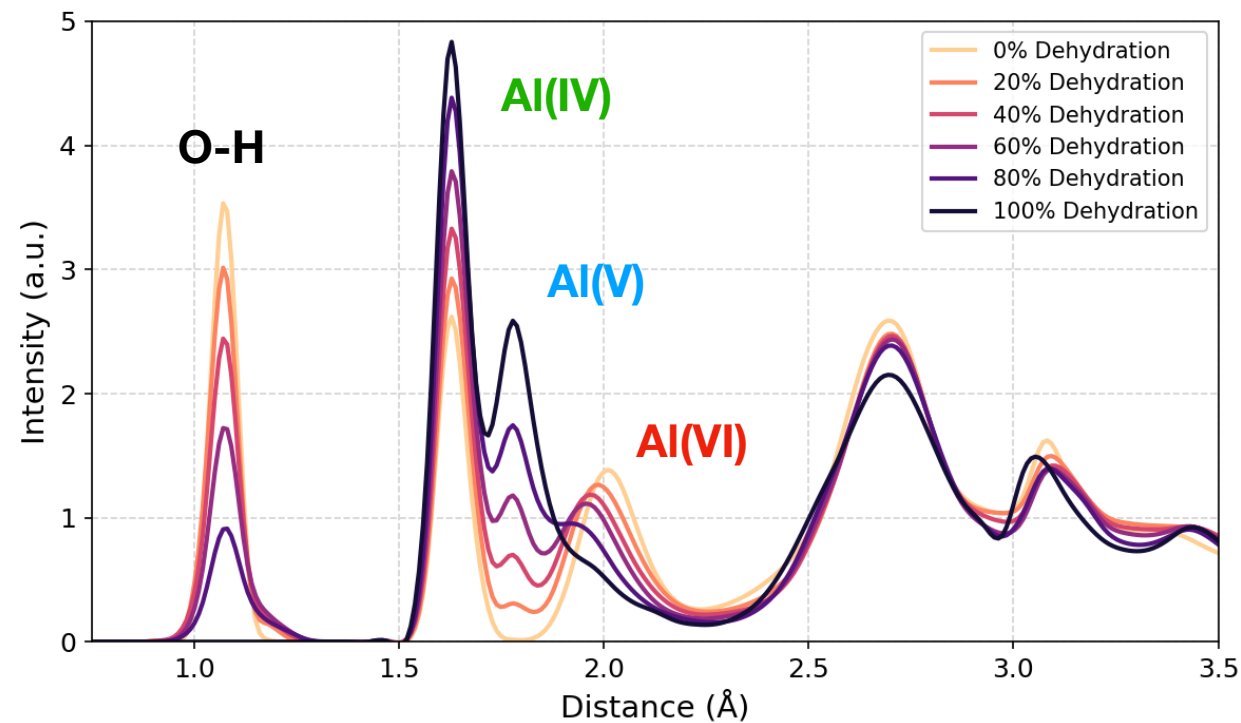
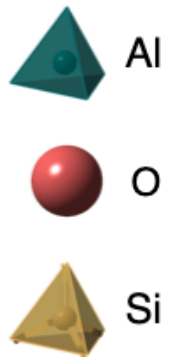
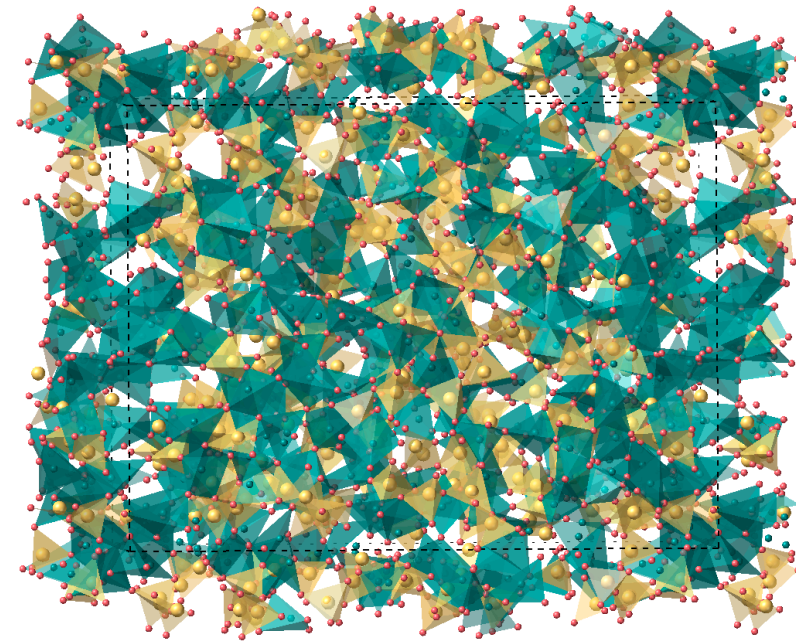
Melting - quenching



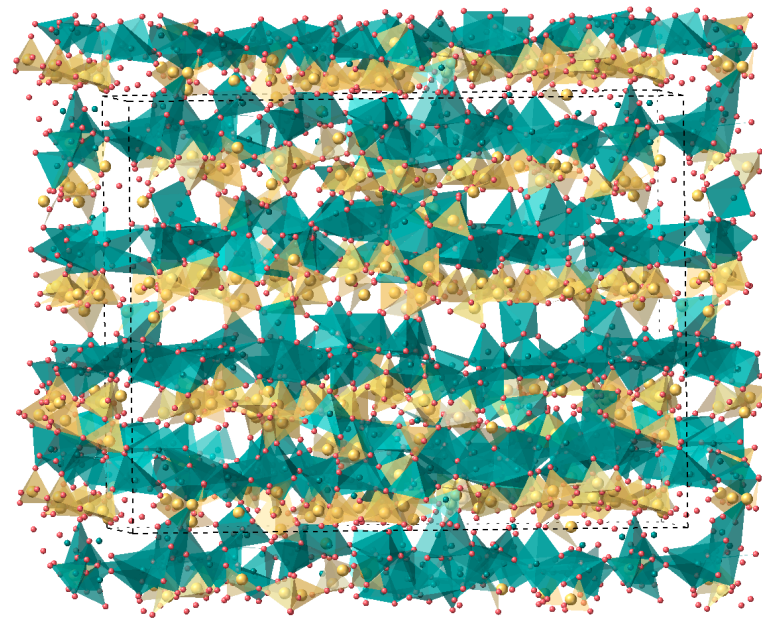
Dehydroxylation protocols



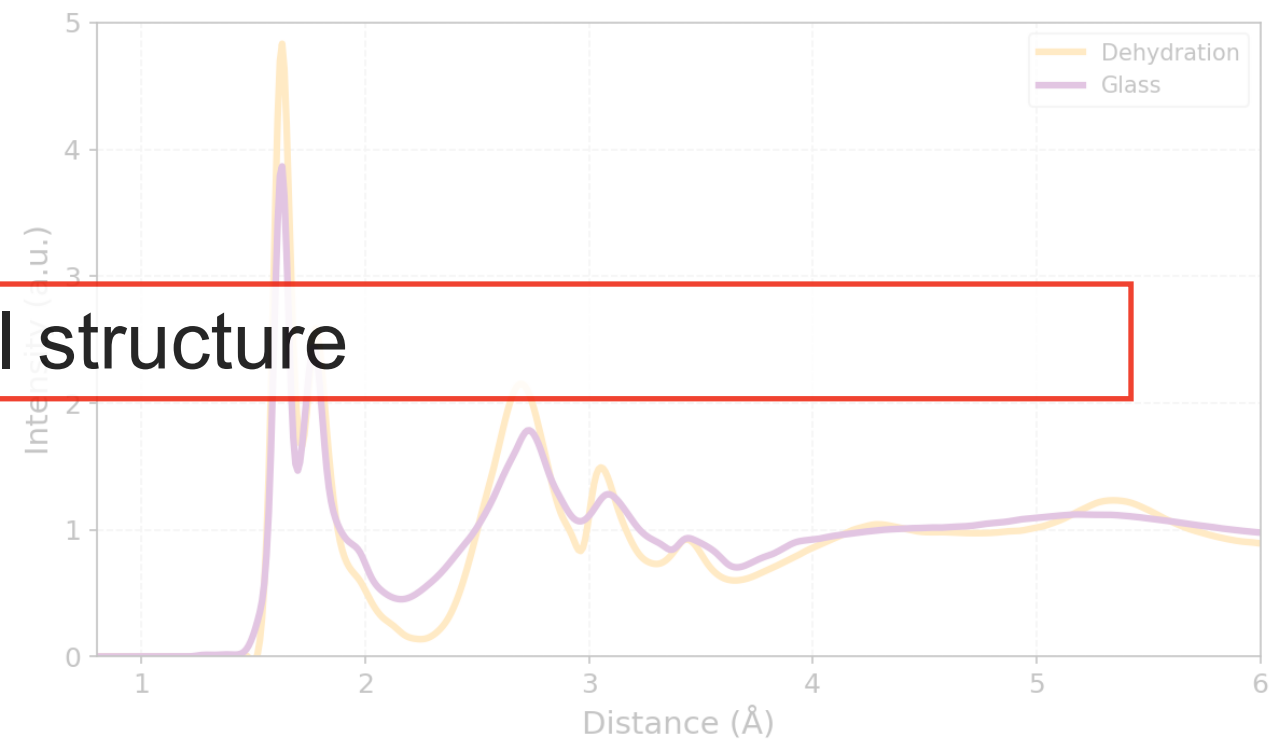
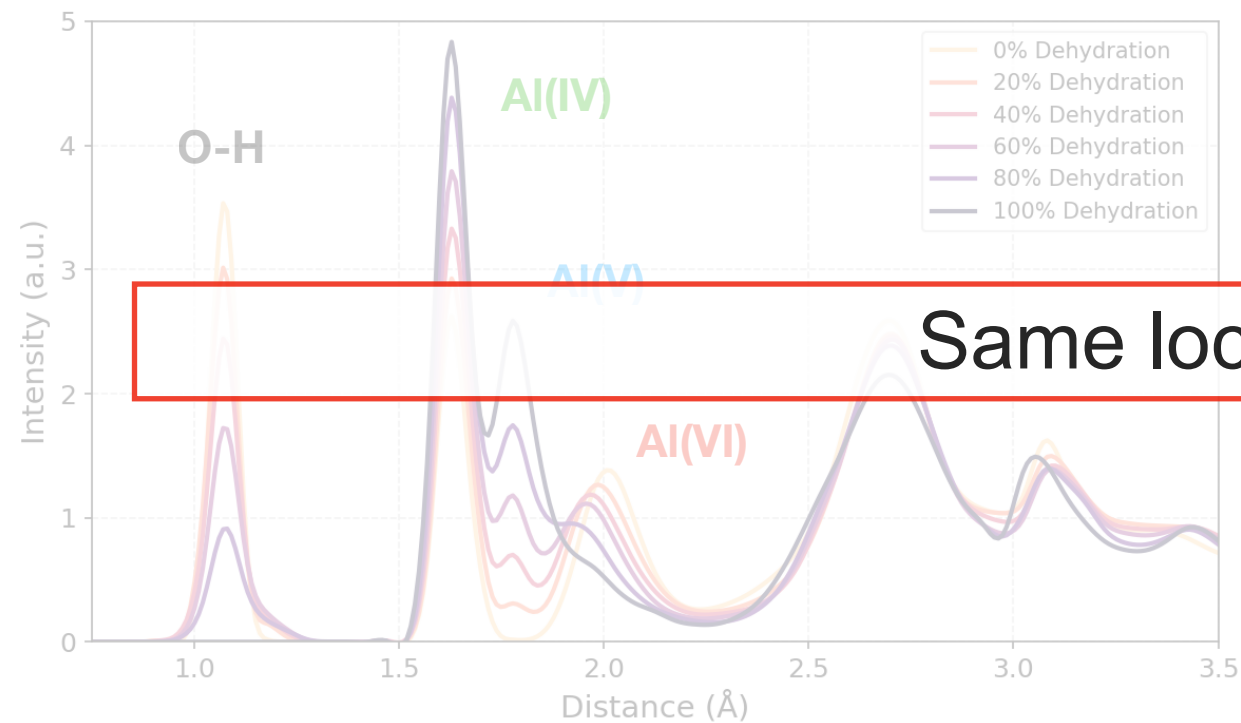
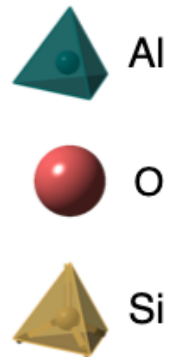
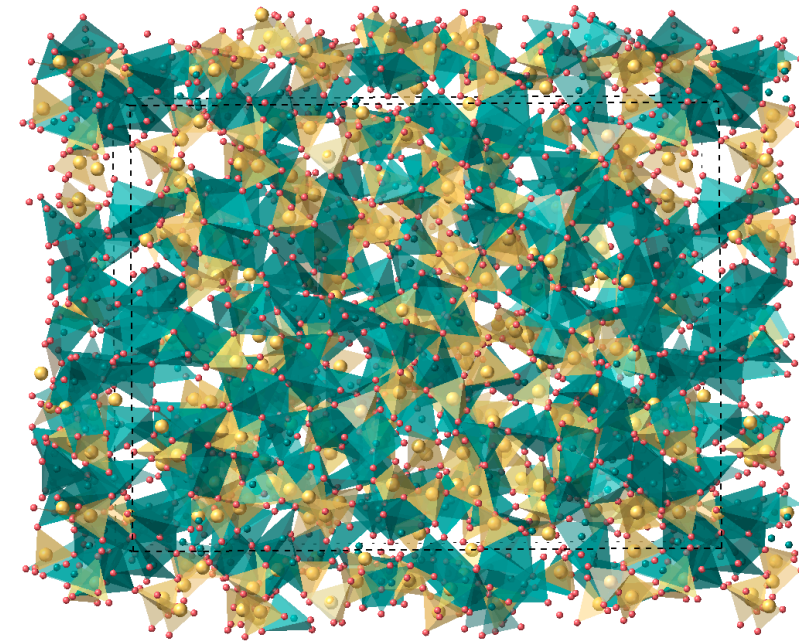
Melting - quenching



Dehydroxylation protocols

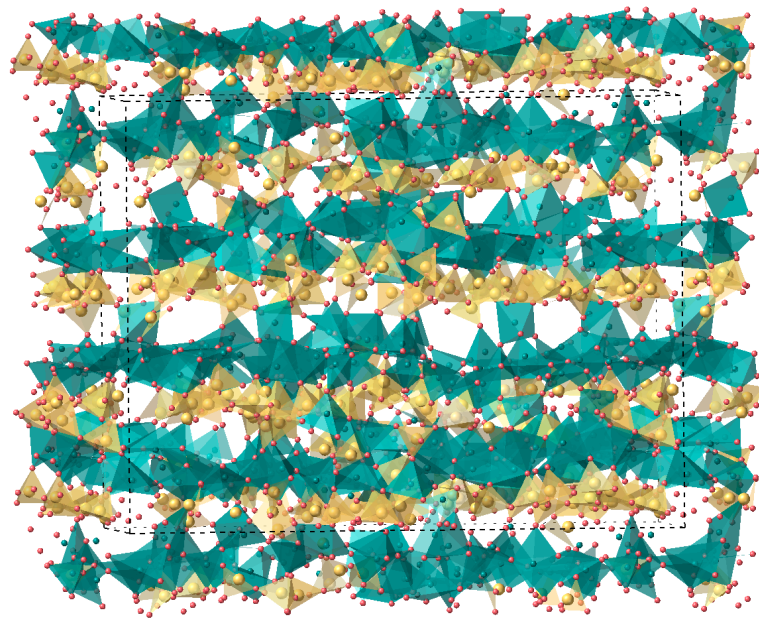


Melting - quenching

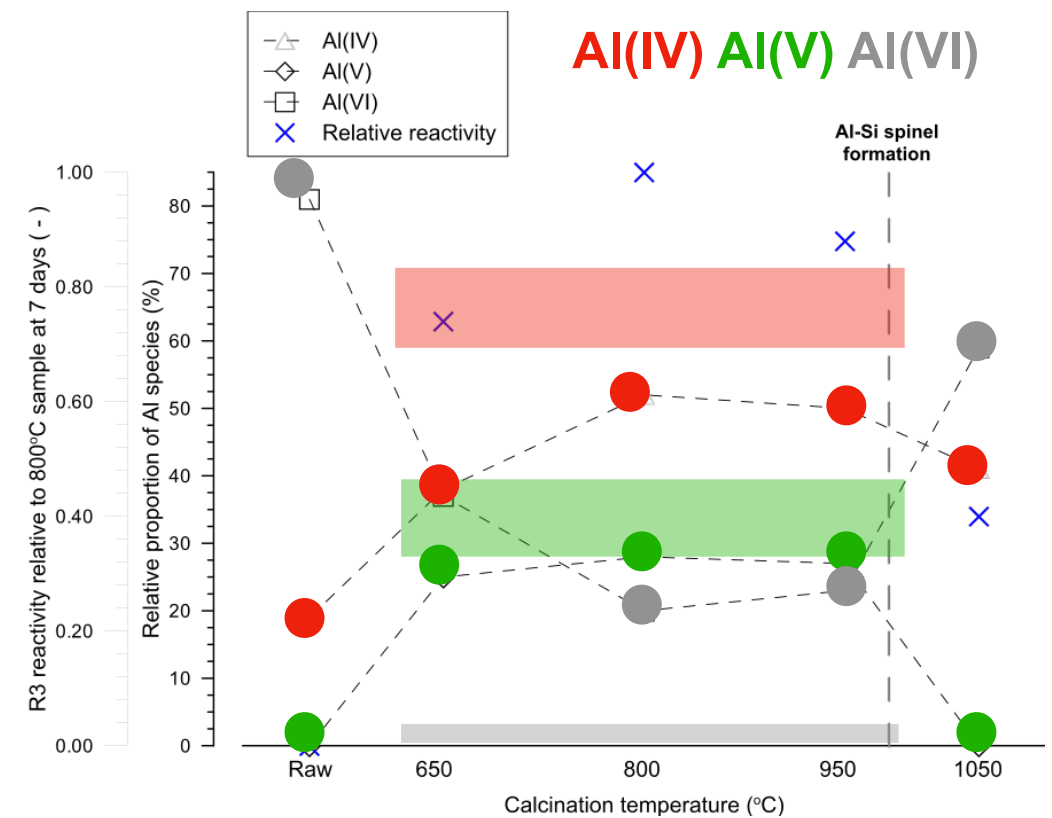
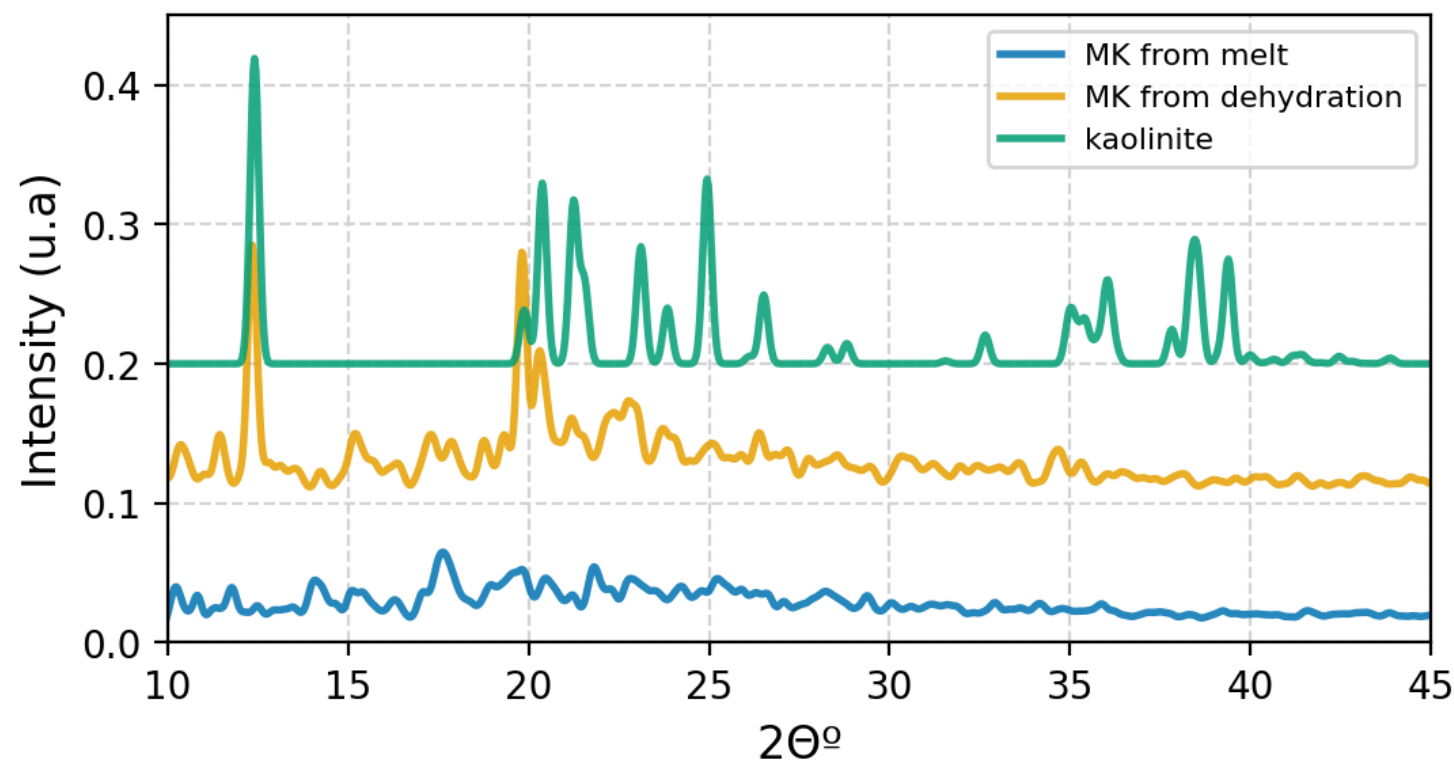
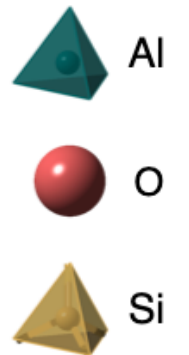
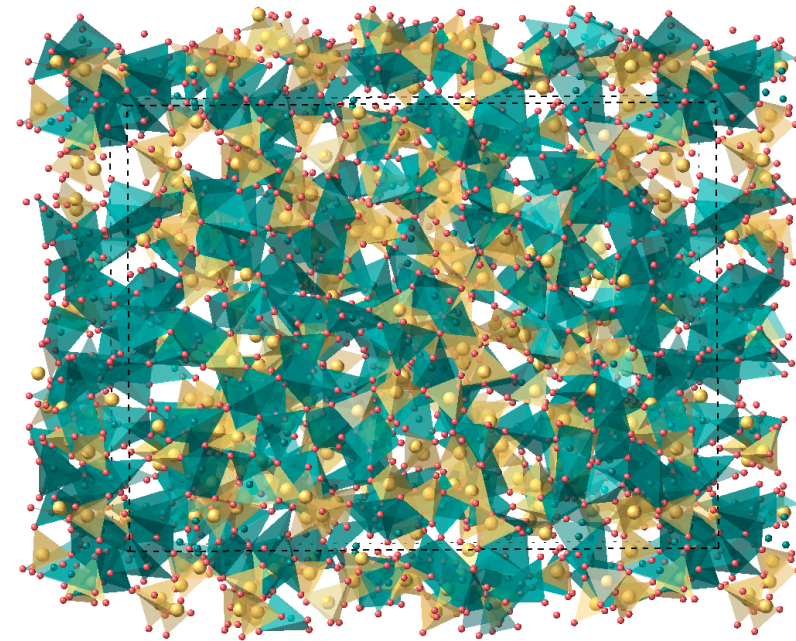


Same local structure

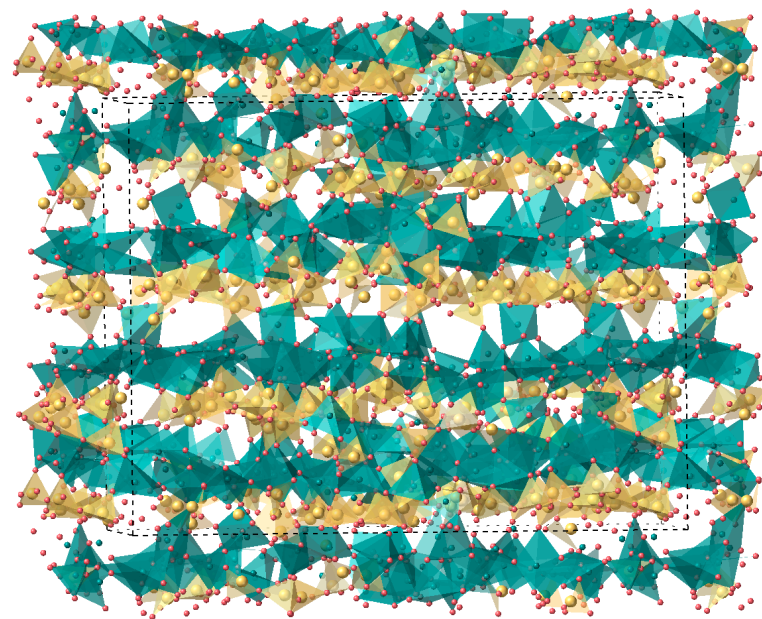
Dehydroxylation protocols



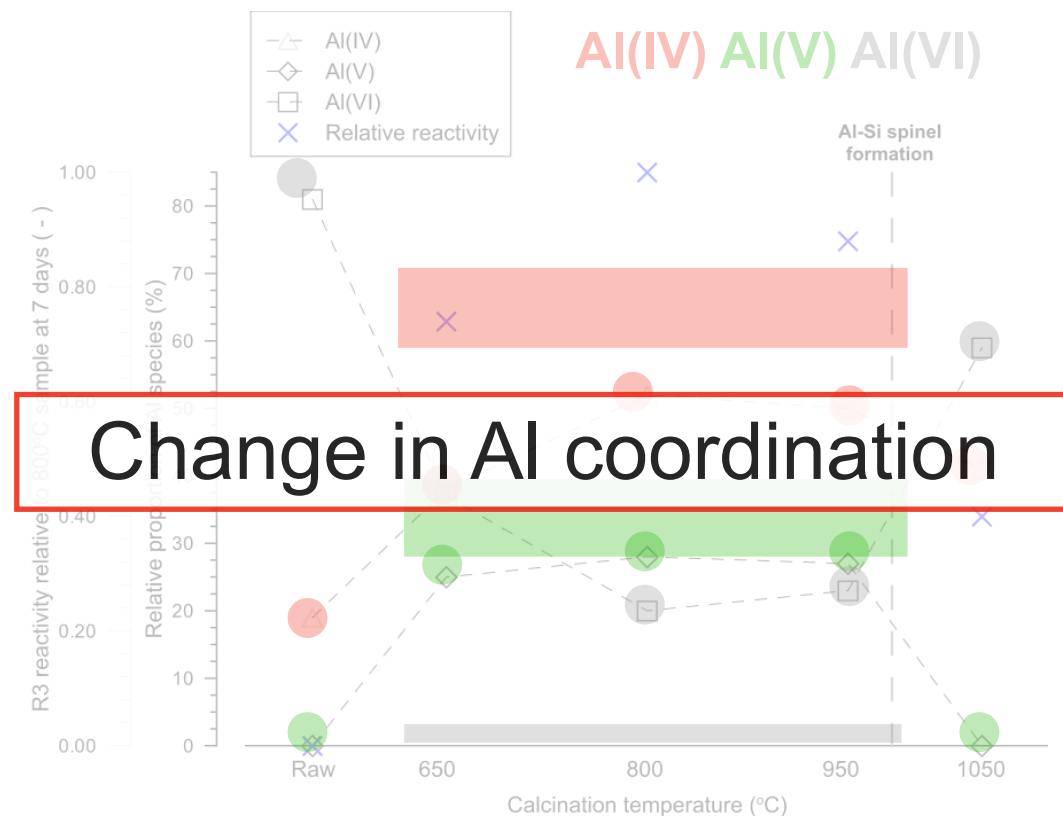
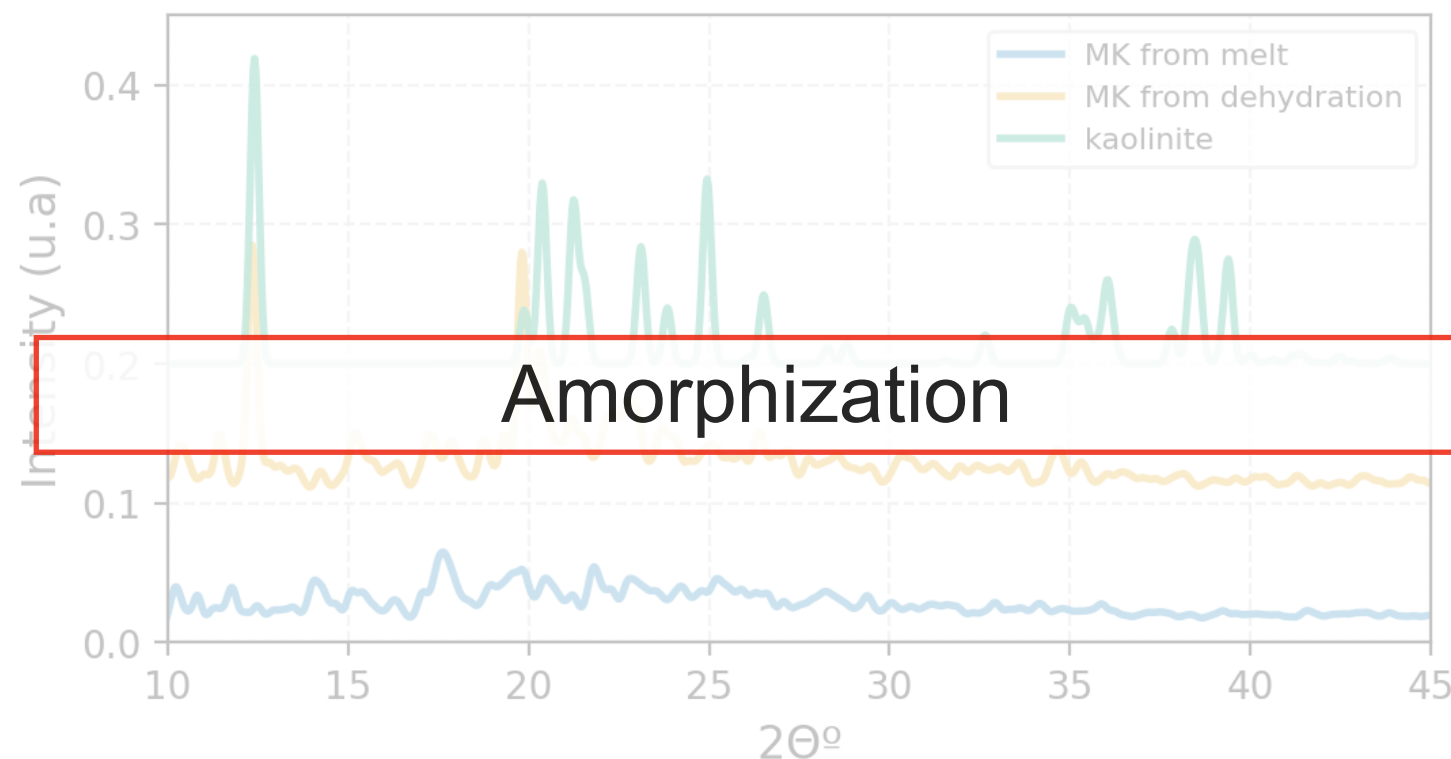
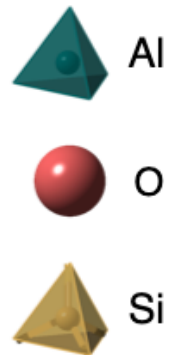
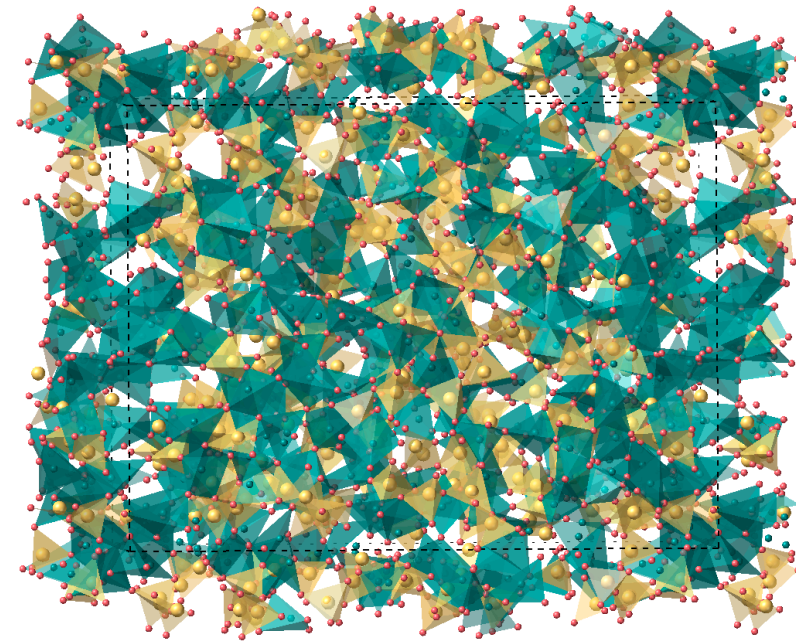
Melting - quenching



Dehydroxylation protocols



Melting - quenching



My 2005 self



“We can go beyond laboratory limitations, do thousands of simulations to test materials and calculate properties, and design cement from electrons to buildings”

20 years later



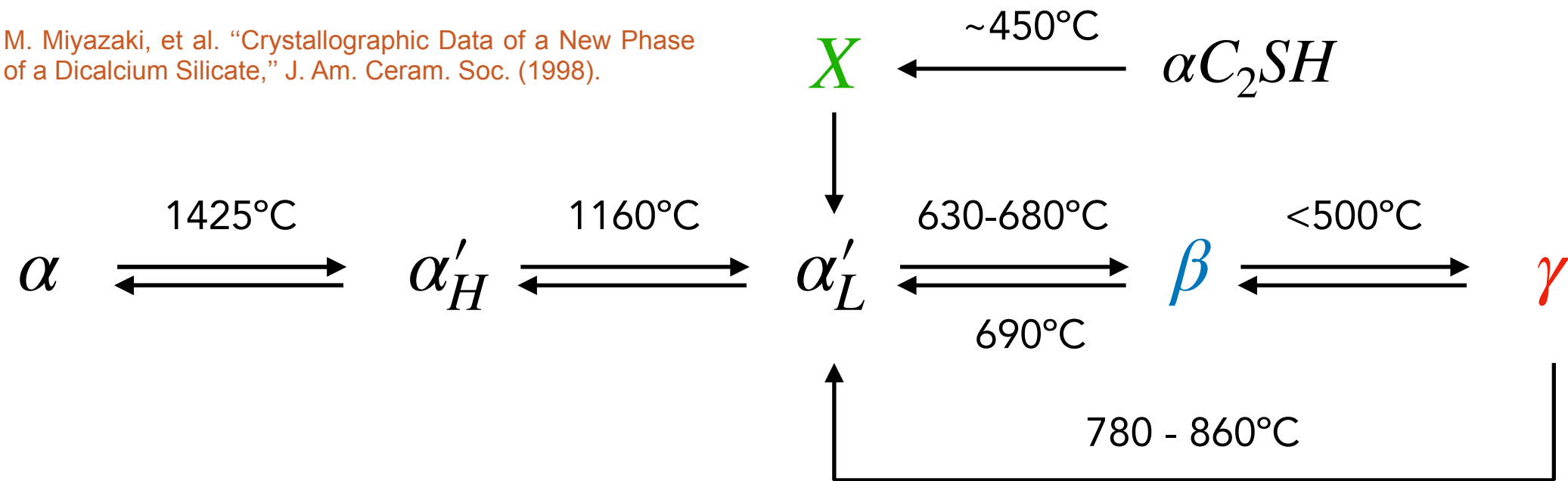
“In practice, the number of atomistic simulation studies that truly imply a practical advance or guide the design of cement towards enhanced performance is limited”

Duque-Redondo, E., de Souza, F. B., Geng, G., & Manzano, H. (2026). A critical review and perspectives on atomistic models of non-crystalline cementitious materials. *Cement and Concrete Research*, 199, 108067.

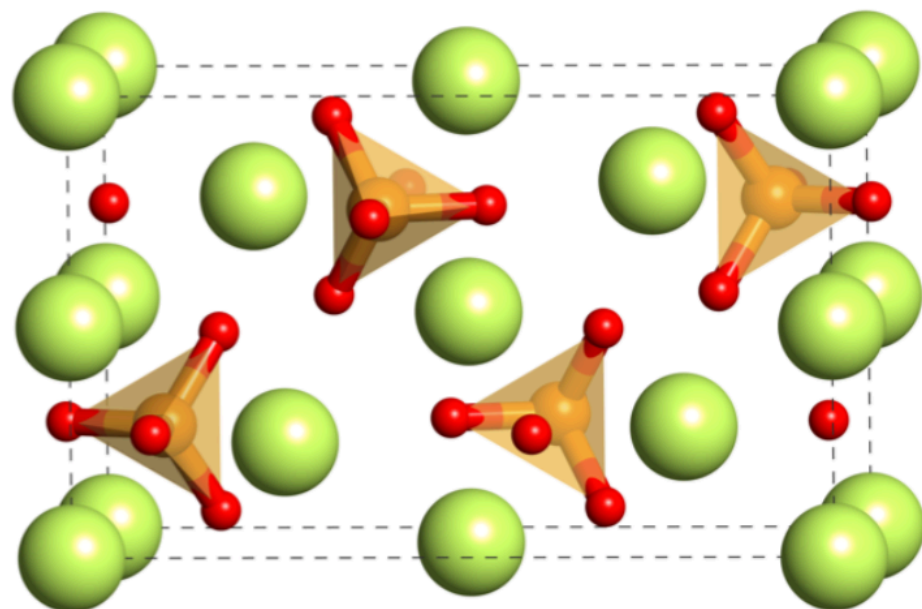
Intrinsic scale limitations, lack of contextualization, incomplete models, lack of experimental focus,...

Case of study: belitic cements

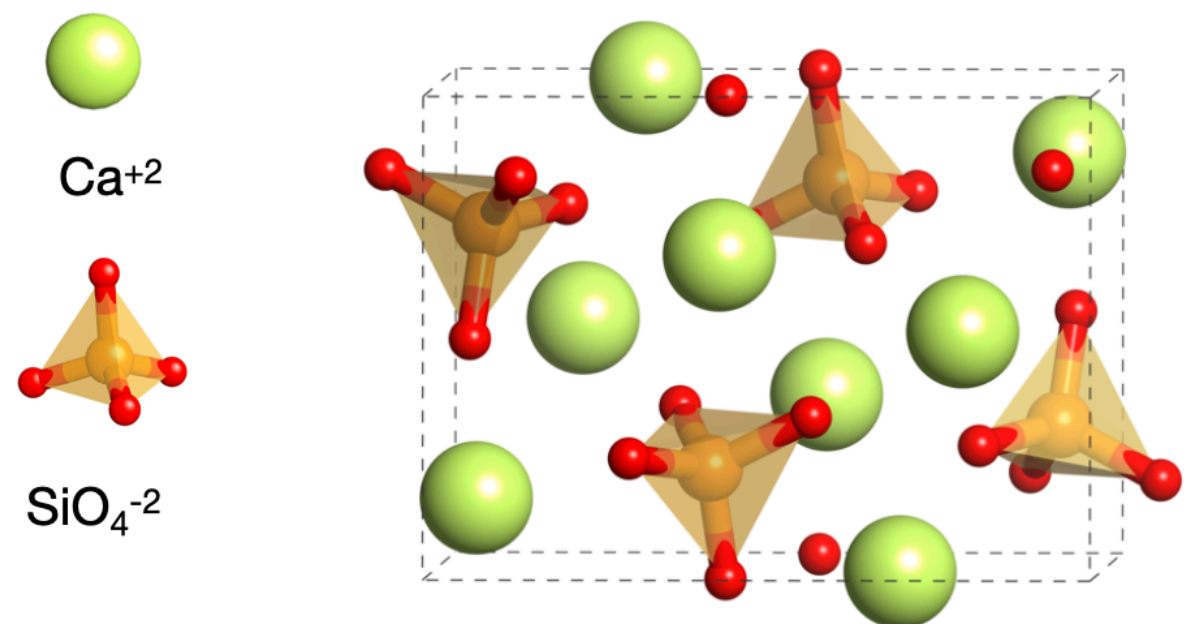
M. Miyazaki, et al. "Crystallographic Data of a New Phase of a Dicalcium Silicate," J. Am. Ceram. Soc. (1998).

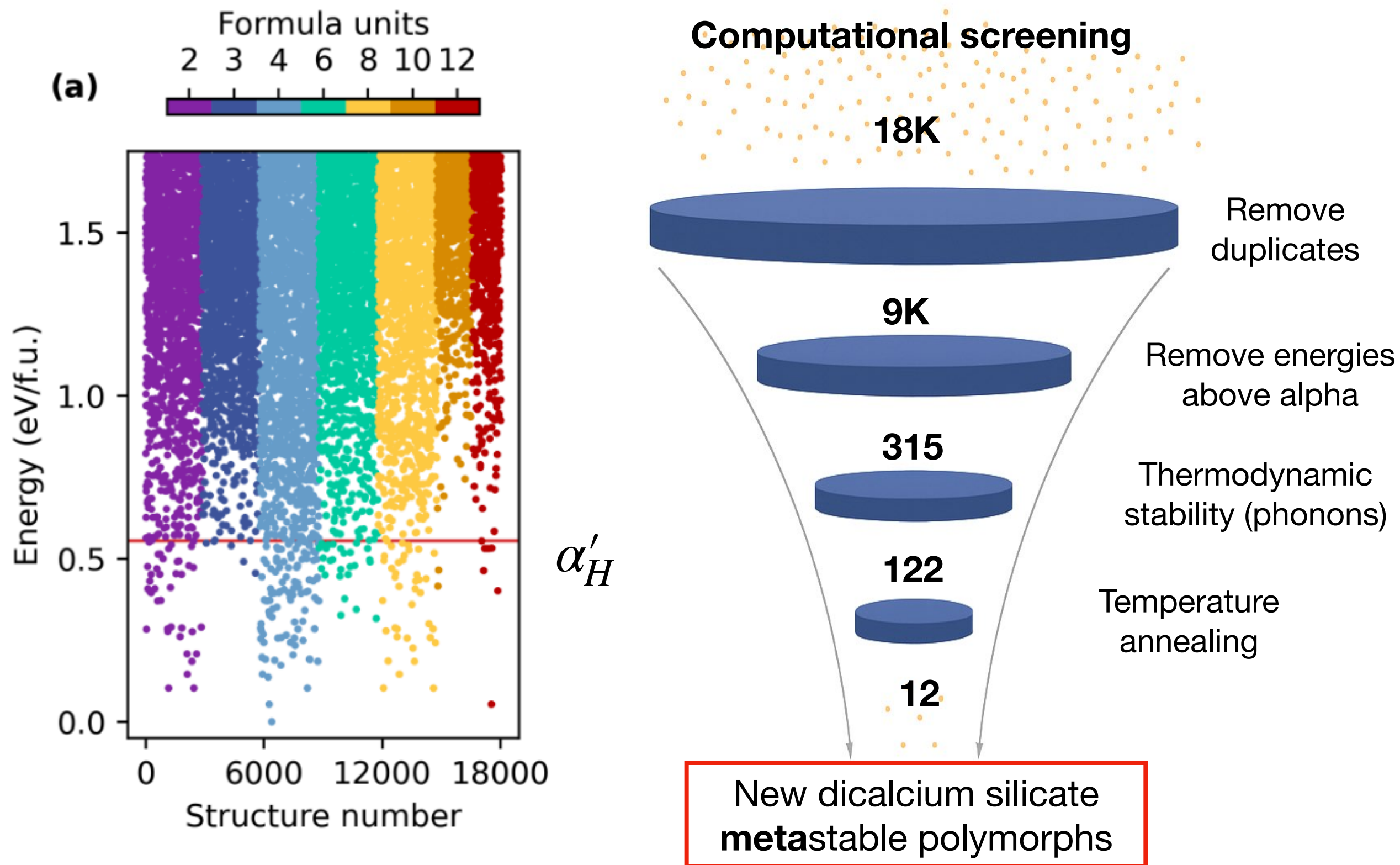


βC_2S ($r_{diss} = 18 \mu\text{mol m}^{-2} \text{s}^{-1}$)

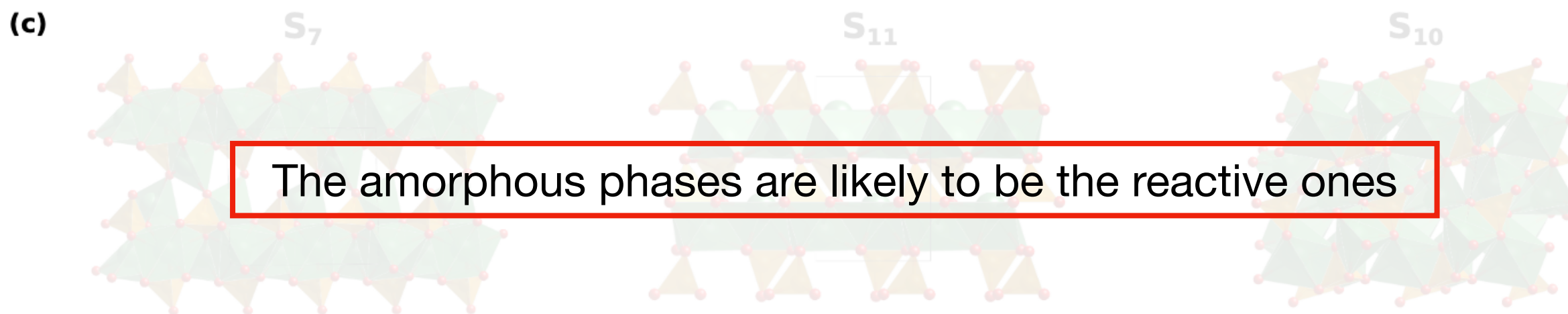
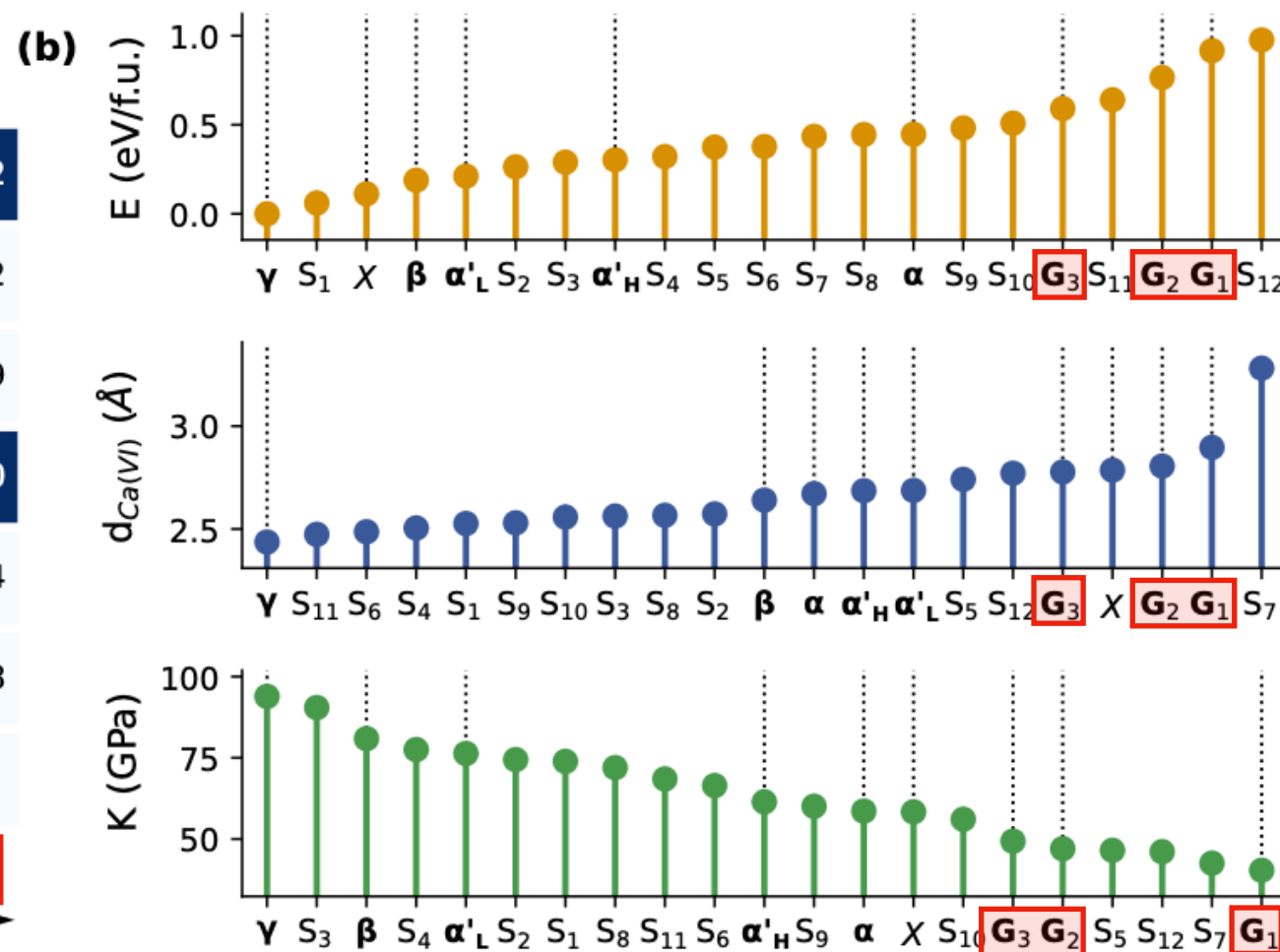
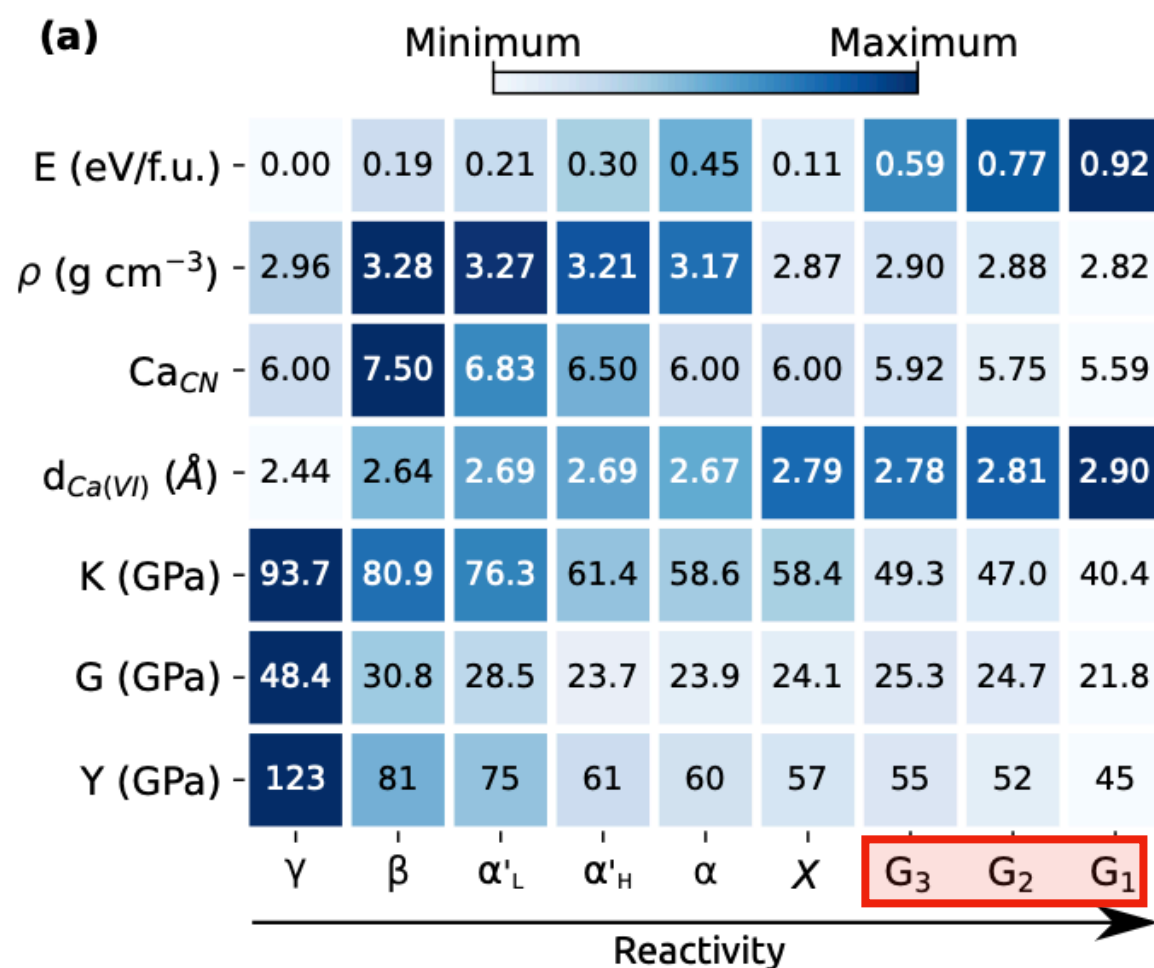


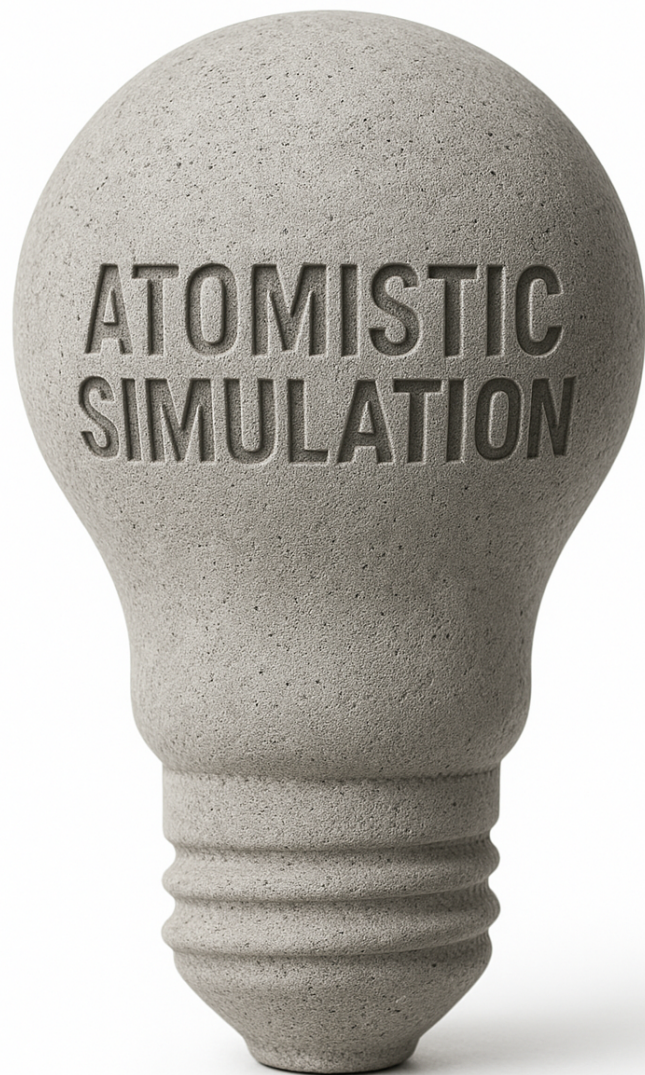
γC_2S ($r_{diss} = 3 \times 10^{-4} \mu\text{mol m}^{-2} \text{s}^{-1}$)





Dicalcium Silicate: searching for new polymorphs



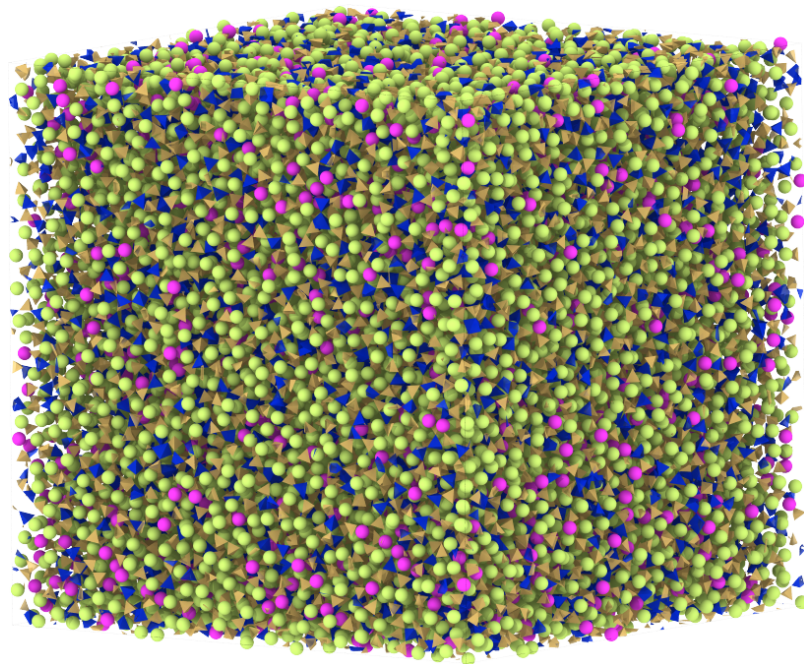


Better models
Better methods
Focused studies
Increasing contextualization and
interaction with experiments

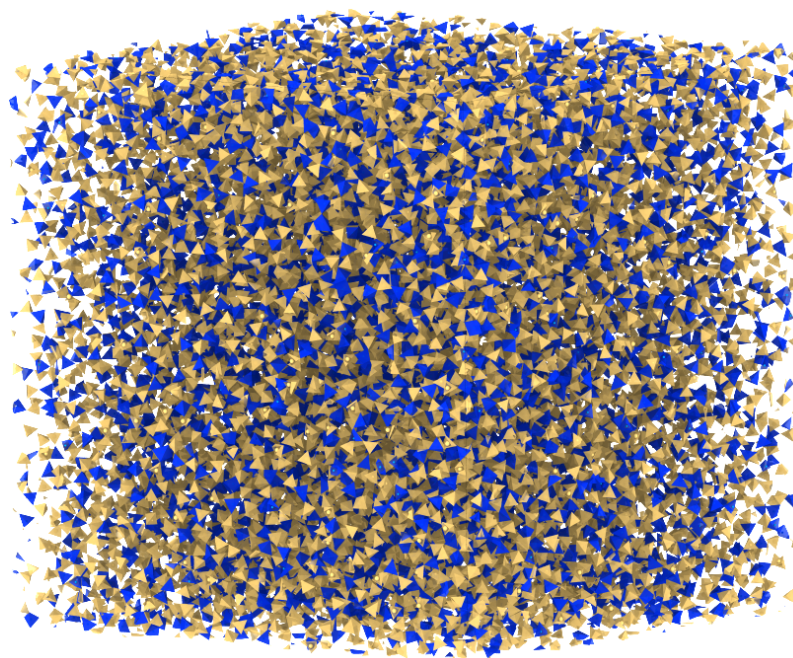
Fly ashes and slags: current models

Qi Zhai, Macro Bertani, Hegoi Manzano, Takayasu Ito, Koji Ohara, Kiyofumi Kurumisawa, The changes in the reactivity of synthetic aluminate silicate-based slag in alkaline environment induced by **minor** components, under review

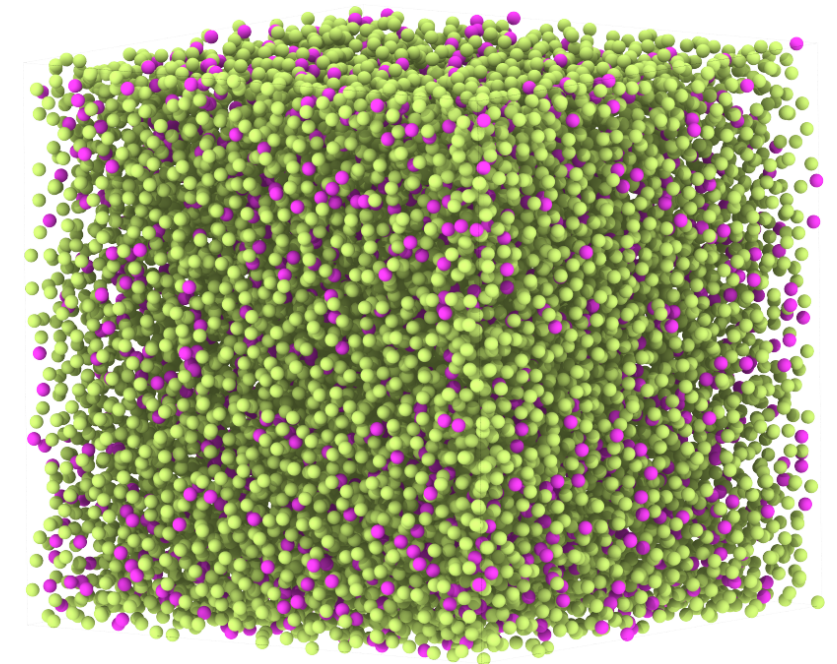
Full slag model



Al/Si T-O-T network



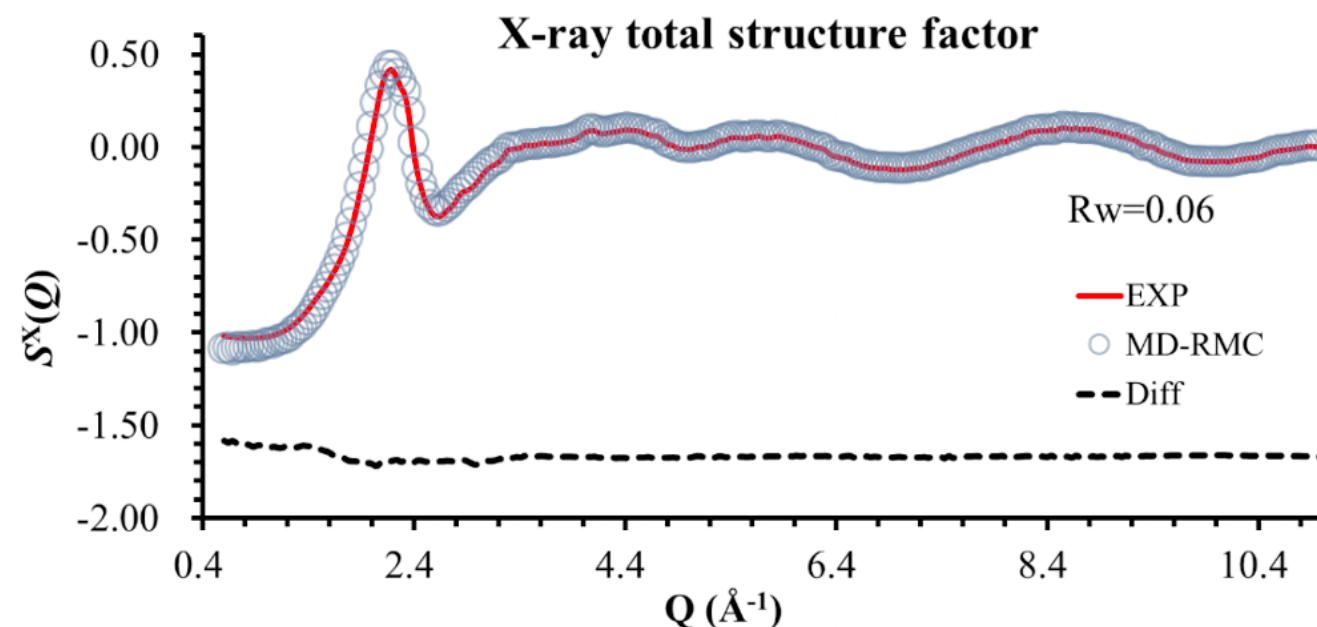
Ca-Mg network modifiers



~70,000 atoms!

70 substitutions!

Molecular Dynamics +
Reverse Monte Carlo

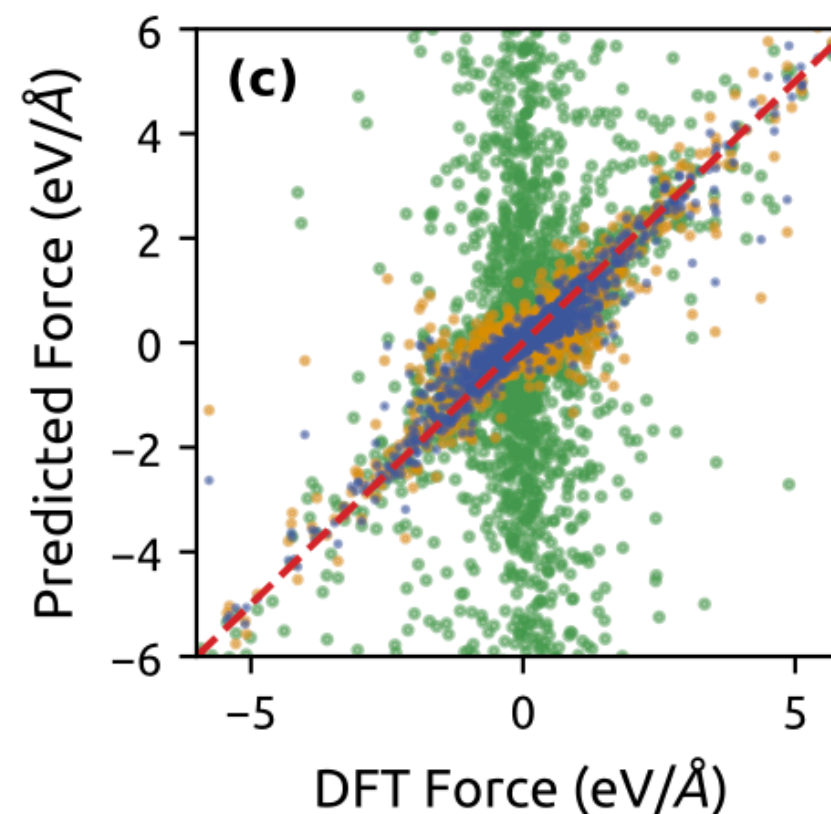
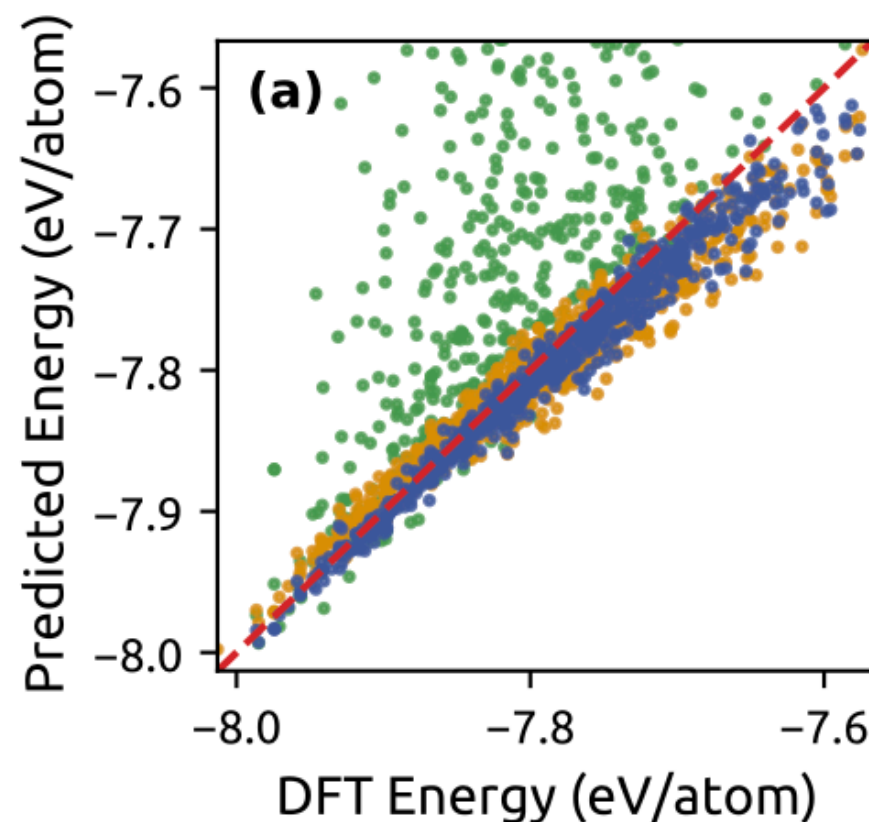


20K calculations; DFT @ 10h ~ 23 years CPU time

20K calculations; MLP @ 10' ~ 0.4 years CPU time

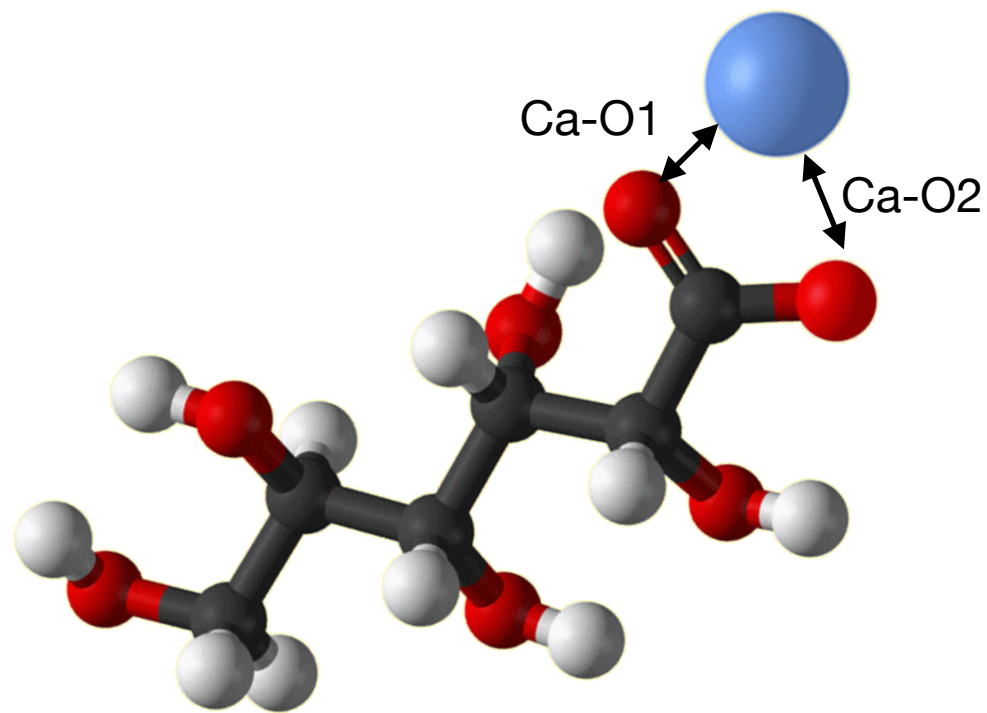
Transfer Learning
of **ænet MLP**

	C_1S	C_2S	C_3S	SiO_2	CaO	Dimers
ReaxFF	4000	14000	4000	2000	1000	-
DFT	1291	5662	1447	750	300	500

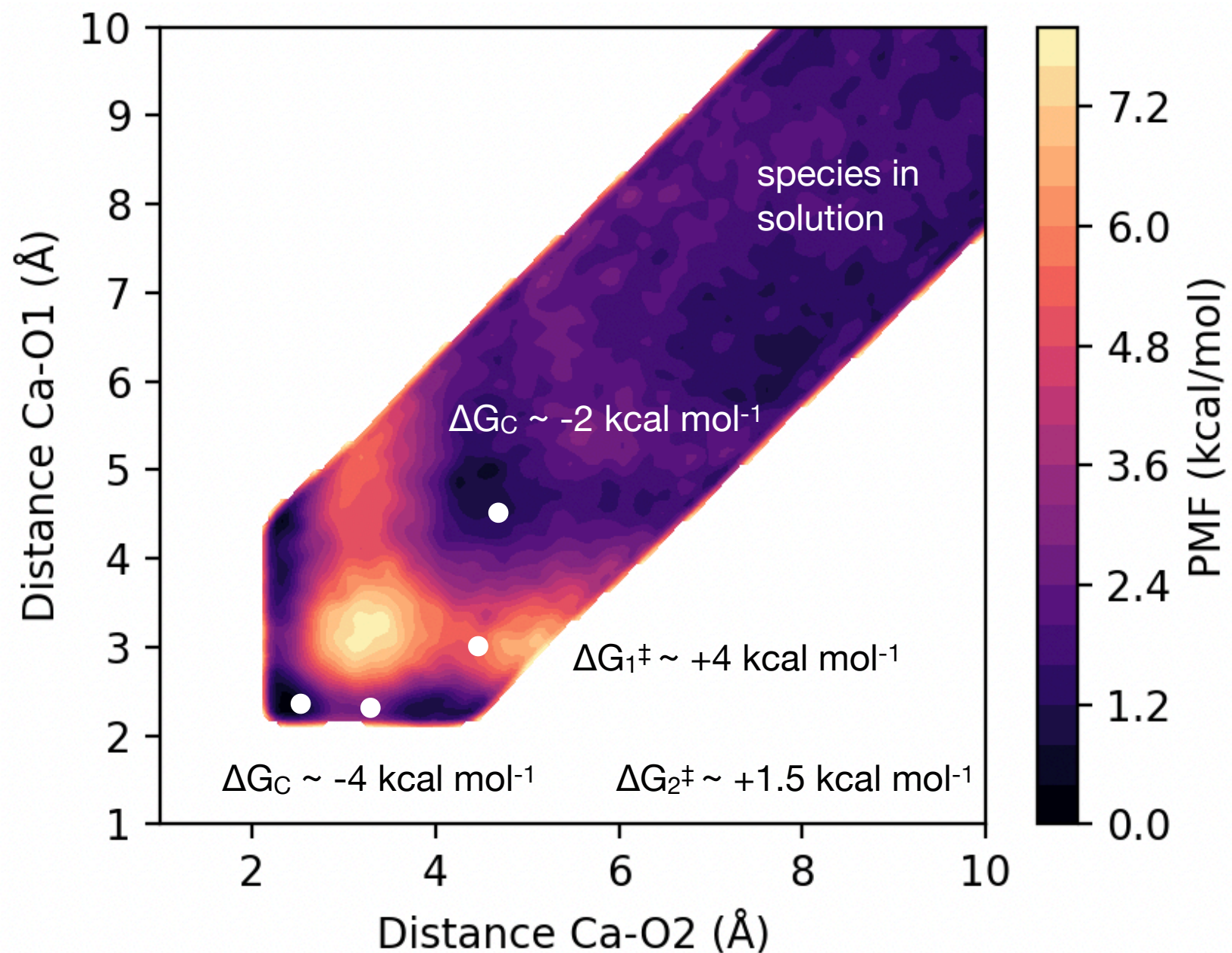


MLP - direct
MLP -transfer
ReaxFF

Complexation of ions by chemical admixtures Adsorption of chemical admixtures on surfaces



Jiale Huang, Hegoi Manzano et al. Ongoing work



Duque-Redondo, E., de Souza, F. B., Geng, G., & Manzano, H. (2026). A critical review and perspectives on atomistic models of non-crystalline cementitious materials. *Cement and Concrete Research*, 199, 108067.

Duque-Redondo, E., Masoero, E., & Manzano, H. (2022). Nanoscale shear cohesion between cement hydrates: The role of water diffusivity under structural and electrostatic confinement. *Cement and Concrete Research*, 154, 106716.

López-Zorrilla, J., Aretxabaleta, X. M., & Manzano, H. (2024). Exploring the polymorphism of dicalcium silicates using transfer learning enhanced machine learning atomic potentials.

López-Zorrilla, J., Duque-Redondo, ... & Manzano, H. (2025). Mechanistic insights into calcium-silicate-hydrate nucleation from molecular dynamics simulations. *Cement and Concrete Research*, 198

Aretxabaleta, X. M., López-Zorrilla, J., Etxebarria, I., & Manzano, H. (2023). Multi-step nucleation pathway of CSH during cement hydration from atomistic simulations. *Nature Communications*, 14(1)

Manzano, H., A. Ayuela, and J. S. Dolado. (2007) "On the formation of cementitious C–S–H nanoparticles." *Journal of computer-aided materials design* 14(1)

Qi Zhai, Macro Bertani, Hegoi Manzano, Takayasu Ito, Koji Ohara, Kiyofumi Kurumisawa, The changes in the reactivity of synthetic aluminate silicate-based slag in alkaline environment induced by minor components, under review

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See you in 20 years!

Hegoi Manzano

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Faculty of Science and Engineering
University of the Basque Country UPV/EHU

