

Development of a System to Measure Trace Amounts of Helium in Air, Nitrogen, Hydrogen, or Deuterium

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What is the Problem?

- Need to be able to separate and then quantify ppb amounts of He from essentially pure D₂.

Why is this Difficult?

- Most mass spectrometers do not have sufficient mass resolution to separate He⁴ from D₂. ($\Delta M = 0.025$ AMU)
- Most measuring instruments do not have sufficient dynamic range (need $> 10^9$)
- Ambient air is 5 ppm He⁴.

Masses of light ions

		MKS		MKS
	atomic mass	mass #	positive ion mass	mass #
atomic hydrogen	1.007825	50.39	1.00727	50.36
molecular hydrogen	2.01565	100.78	2.01510	100.76
atomic deuterium	2.014102	100.71	2.01355	100.68
molecular deuterium	4.028204	201.41	4.02765	201.38
tritium	3.016049	150.80	3.01550	150.78
He-3	3.016029	150.80	3.01548	150.77
HD	3.02193	151.096	3.02083	151.04
He-4	4.002603	200.13	4.00205	200.10

Mass of electron = 0.00054858 amu = 9.10940×10^{-31} kg

Mass of neutron = 1.008665 amu = 1.67495×10^{-27} kg

Mass of proton = 1.007277 amu = 1.6726×10^{-27} kg

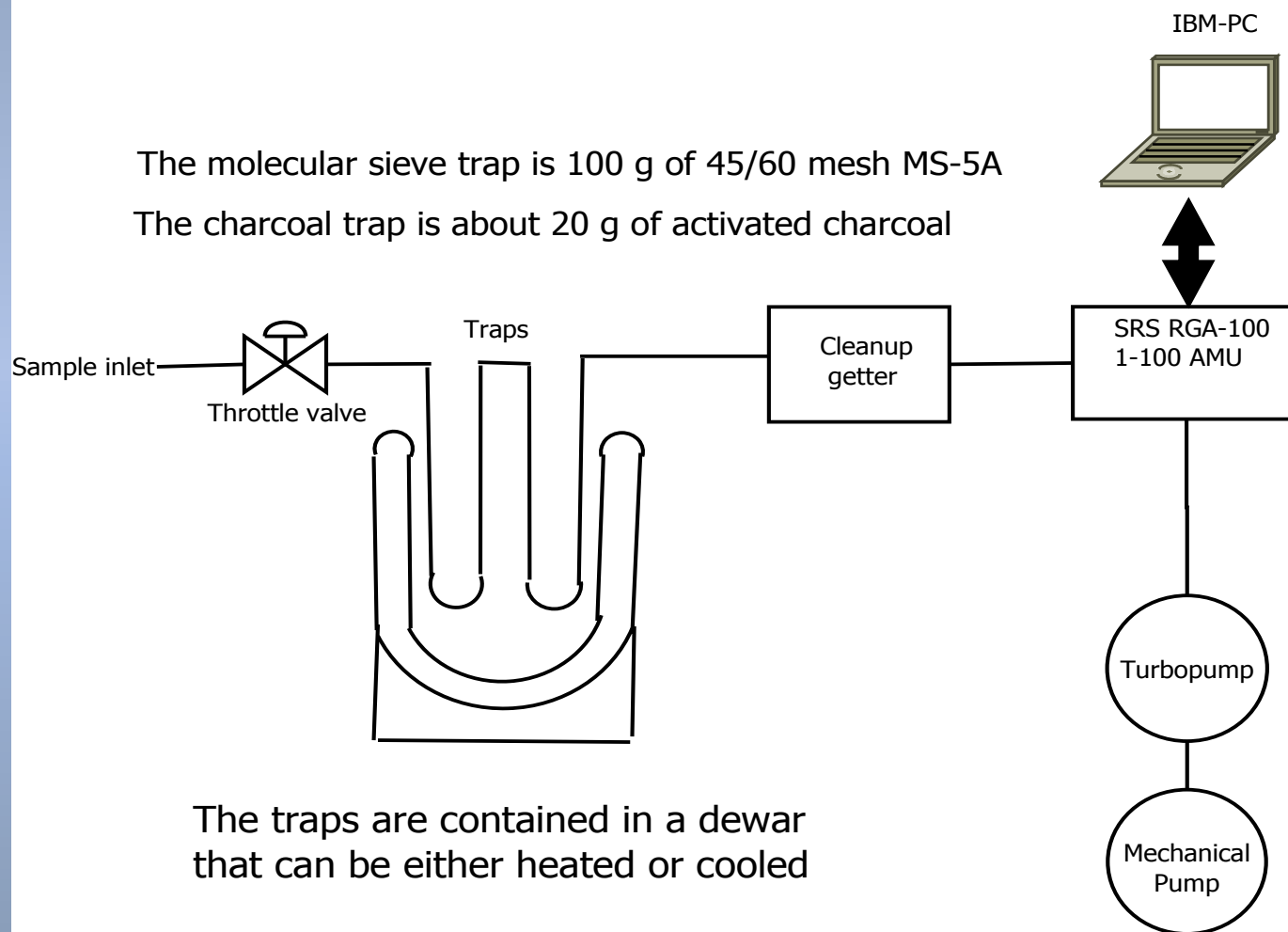
Atomic mass unit= 1.0000 amu

Concept of system for analysis of helium in air samples

There are two traps proposed, one is activated charcoal and the other is MS-5A

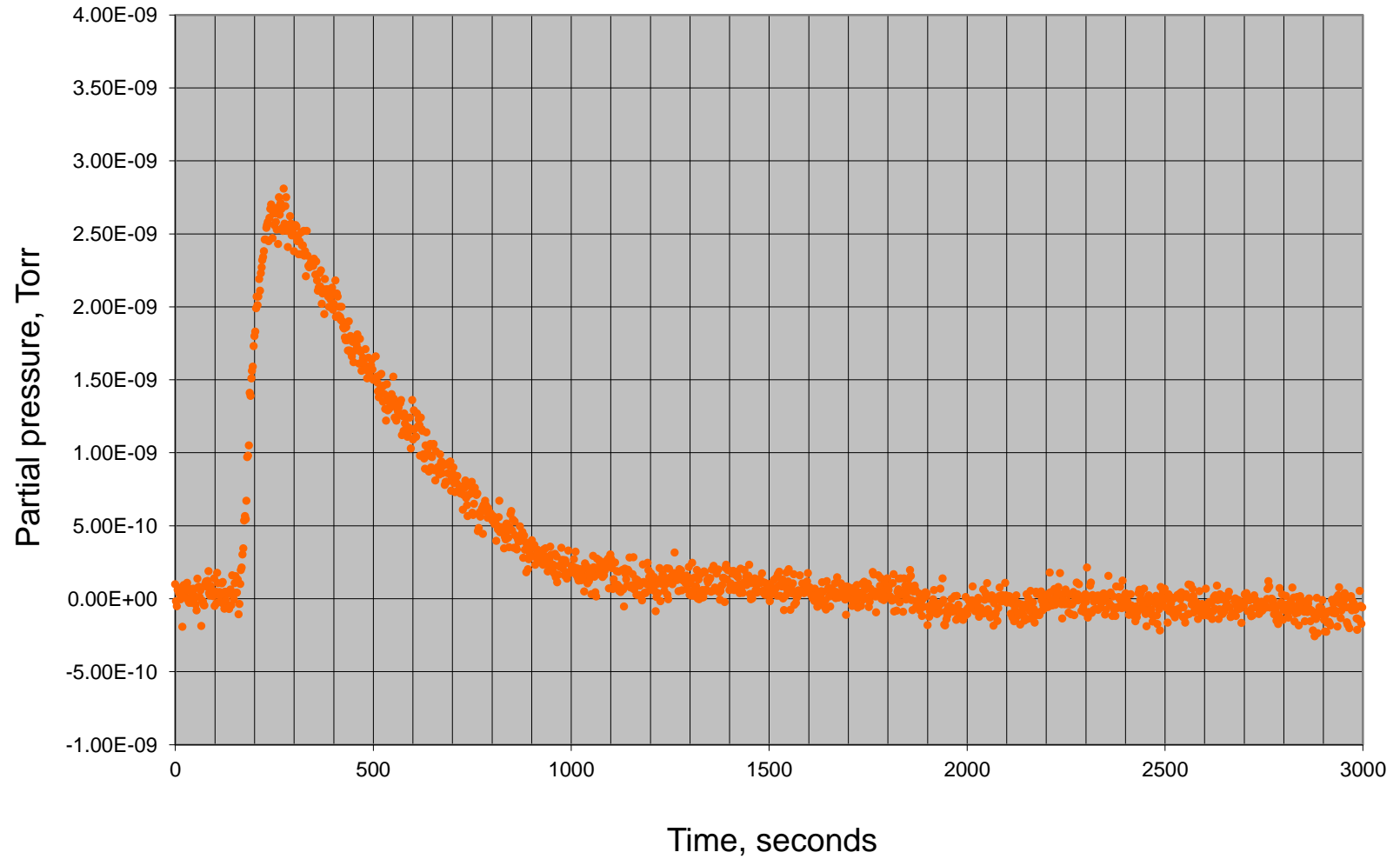
The molecular sieve trap is 100 g of 45/60 mesh MS-5A

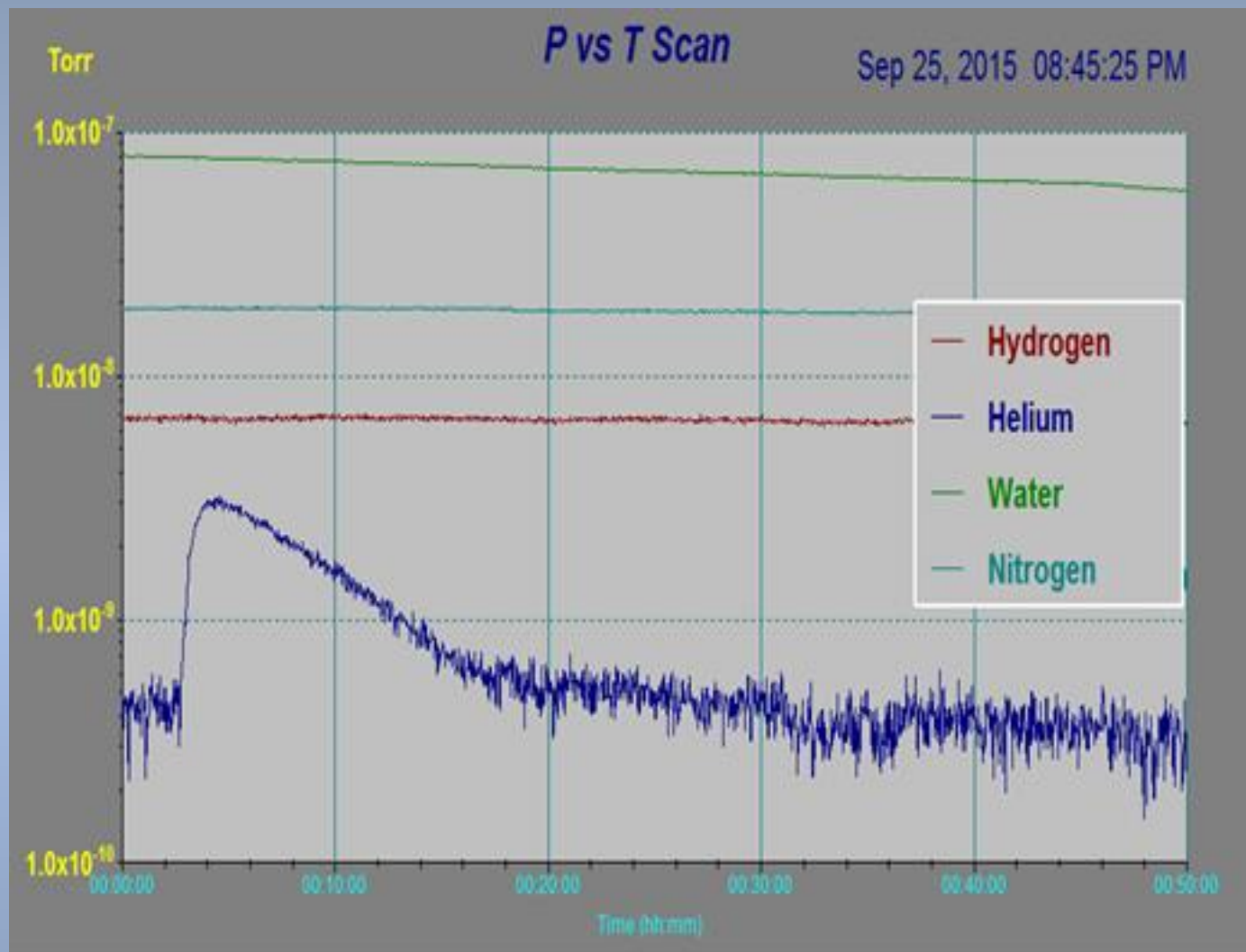
The charcoal trap is about 20 g of activated charcoal



The traps are contained in a dewar that can be either heated or cooled

9/25/15 Net helium signal from 350 cc air at 200 Torr (7e4 Torr-cc)

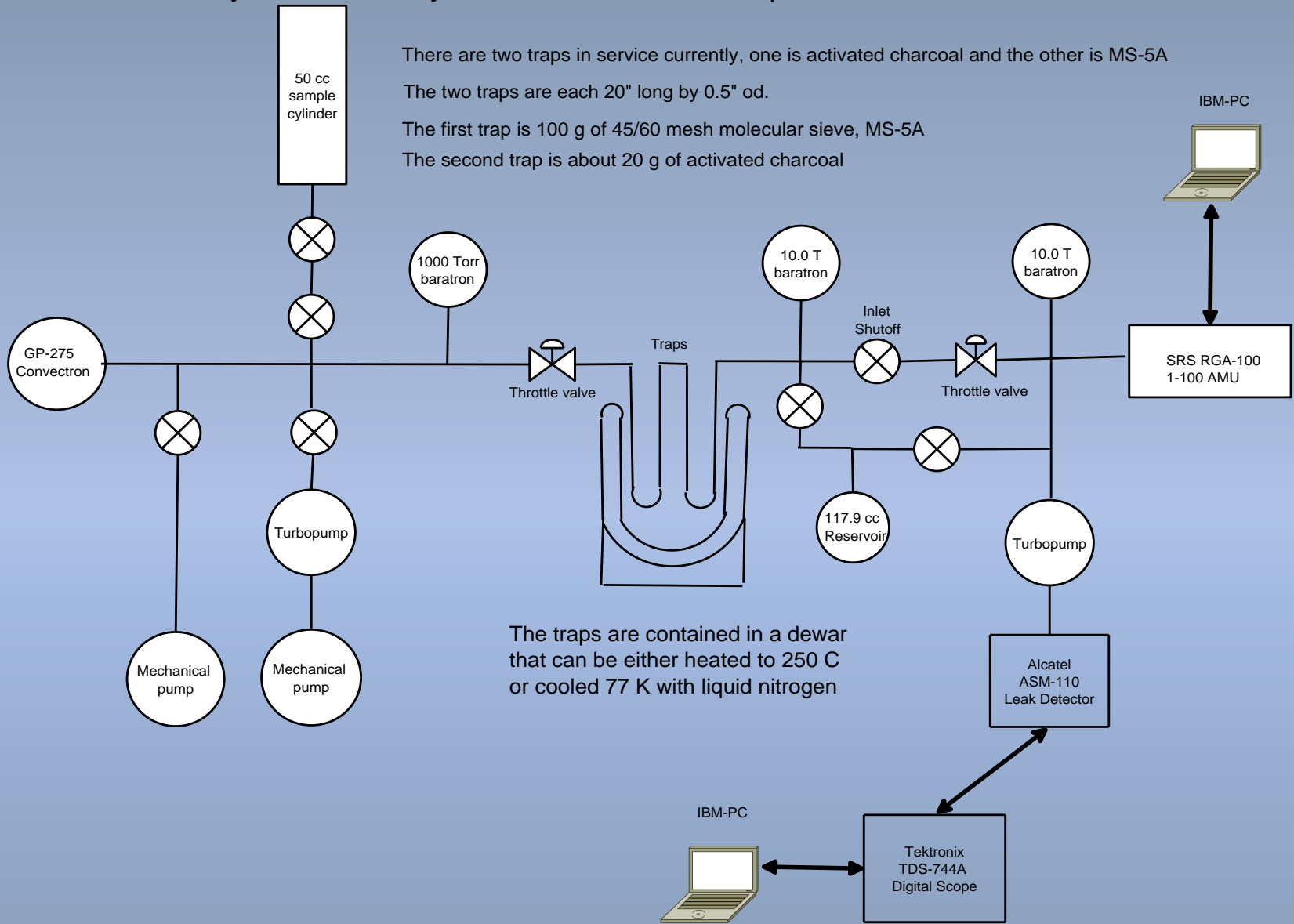




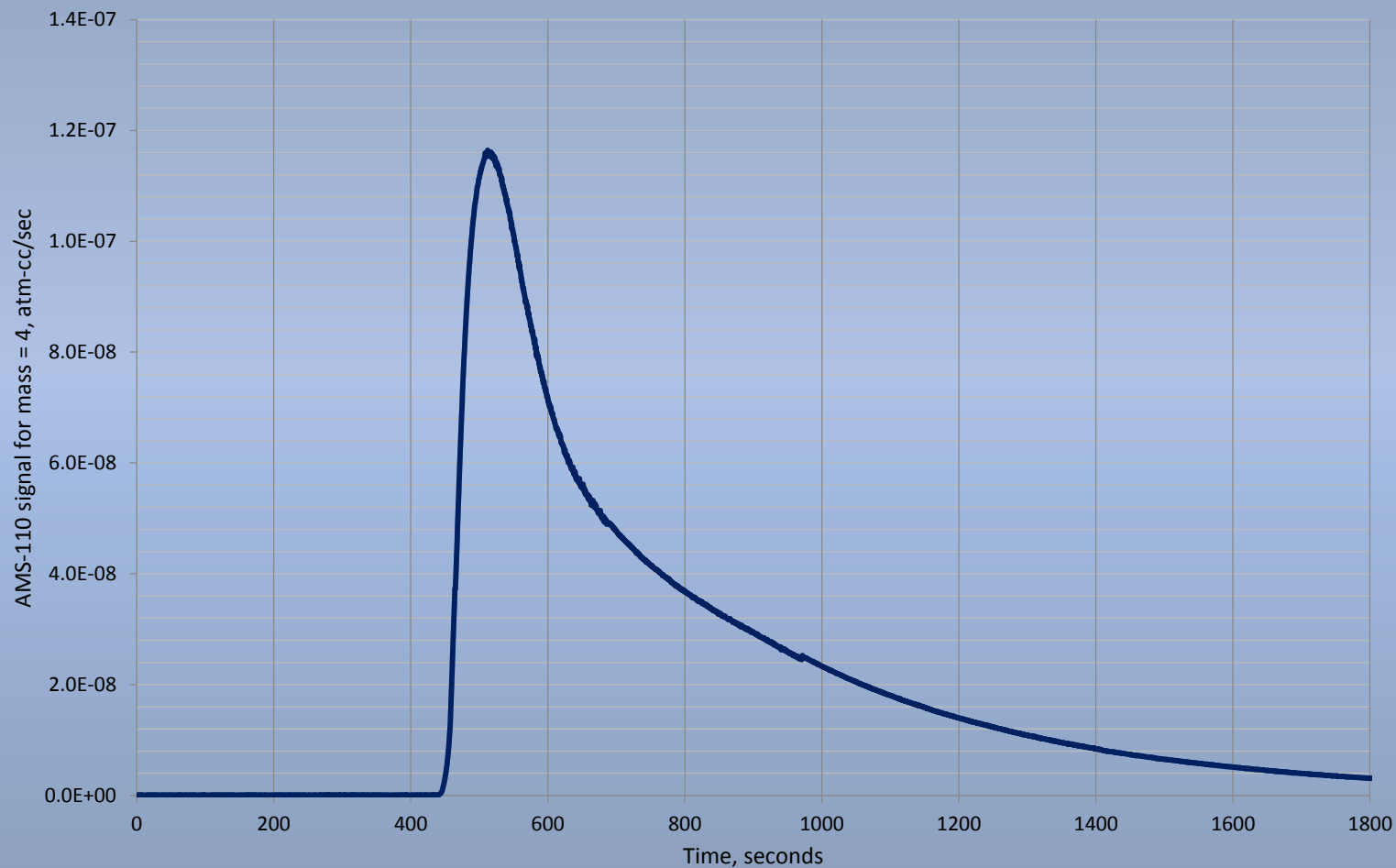
Next, we made several modifications to improve the overall efficiency

- Added a magnetic sector helium leak detector.
 - This instrument has quite a bit more sensitivity for mass-4 helium relative to that for the RGA.
 - The effective background and variability is also much better than that for the RGA.
- Added a TDS-744A digital oscilloscope to log data from the leak detector.
 - This device has only 8 bit resolution, but can be operated in “high resolution” mode that averages the signal between samples yielding 11-12 bit effective resolution.
- Added several pressure gauges and vacuum pumps to facilitate the sample gas handling
- The SRS-100 RGA was retained to verify that there were no gasses other than mass=4 passing through the traps.

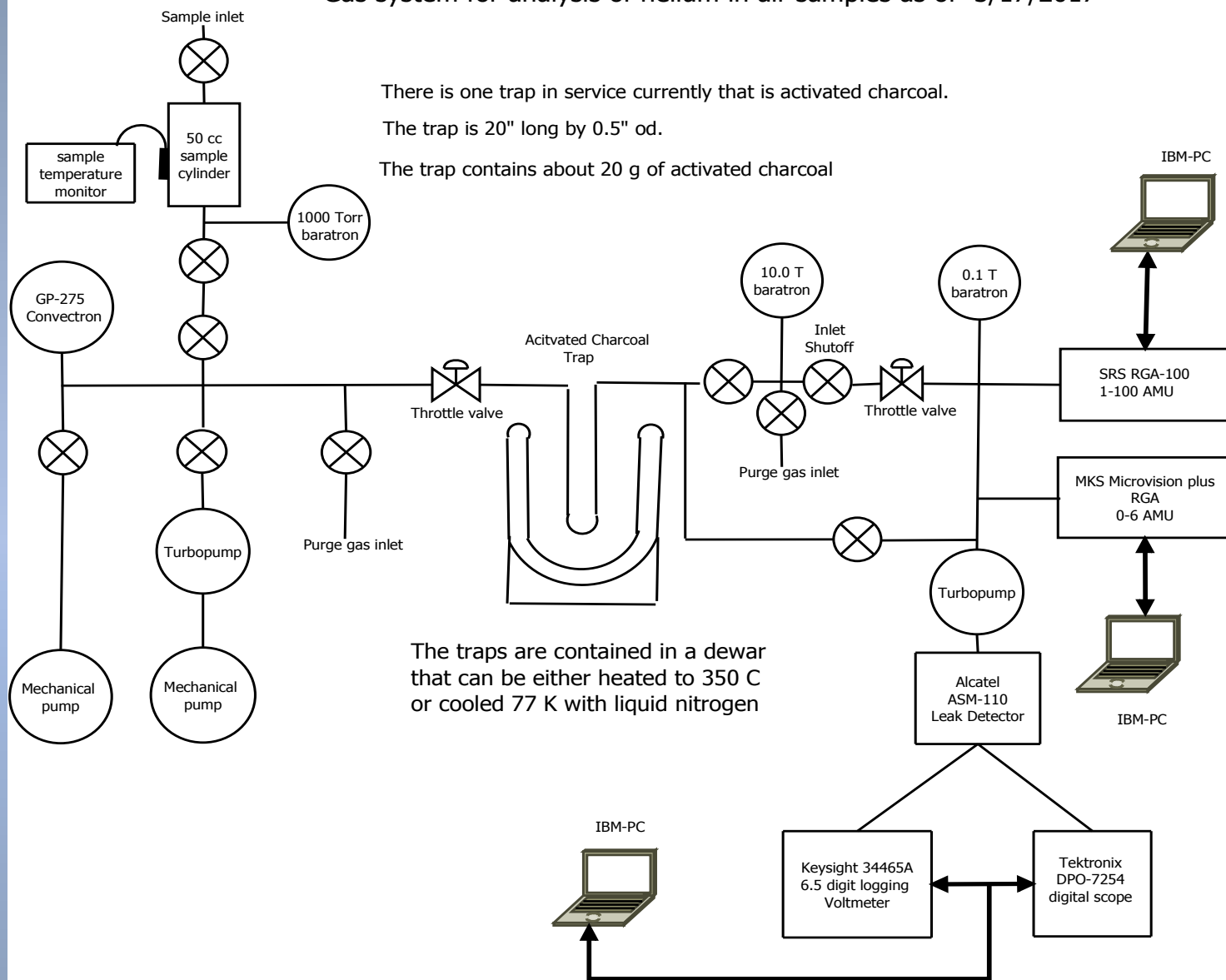
Gas system for analysis of helium in air samples as of 10/15/2015



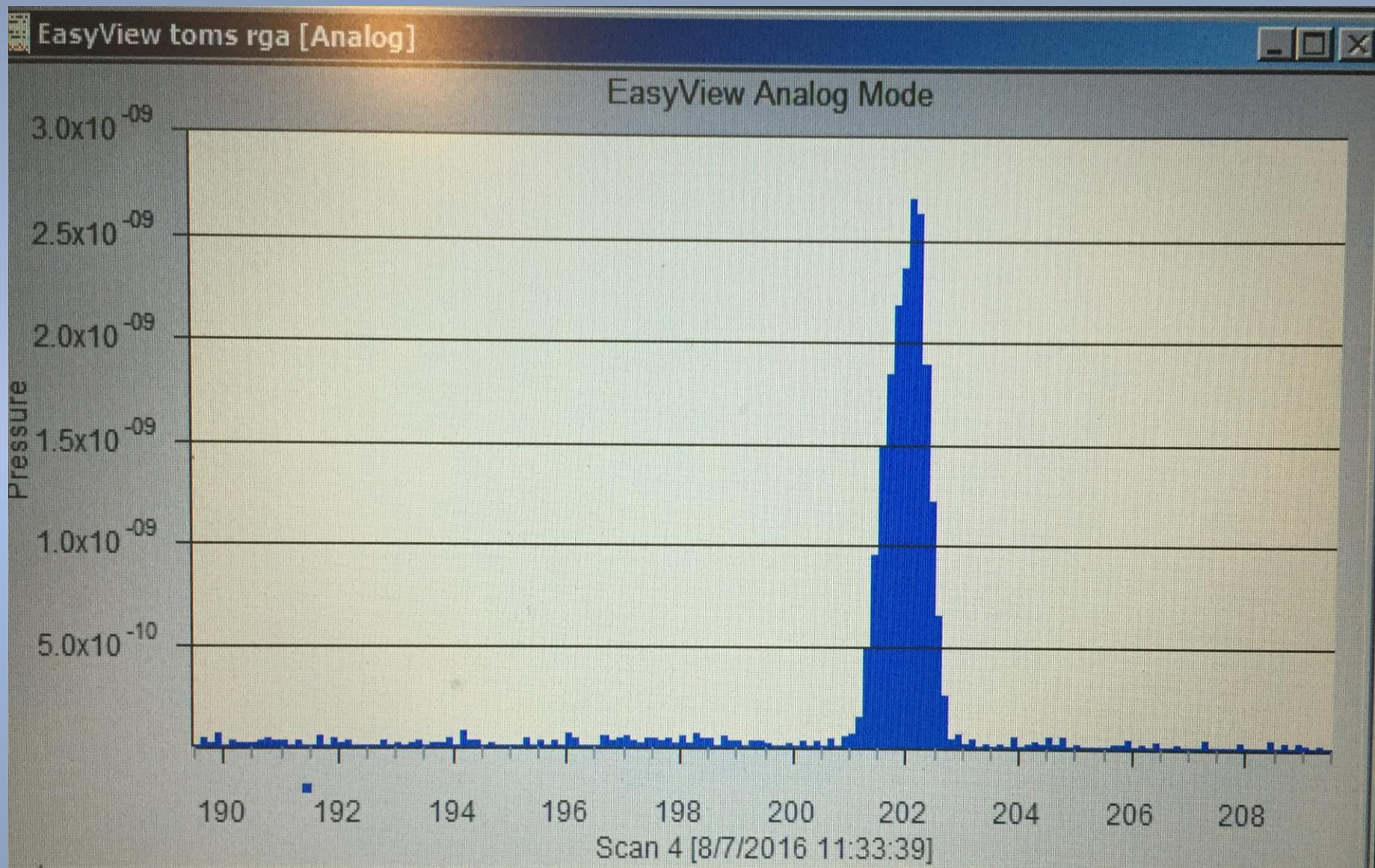
ASM-110 signal for mass=4 from 10 cc of air at 2585 Torr (2.6×10^4 Torr-cc) 9-30-15
This sample was about 1/3 the size of that shown on the previous plot.



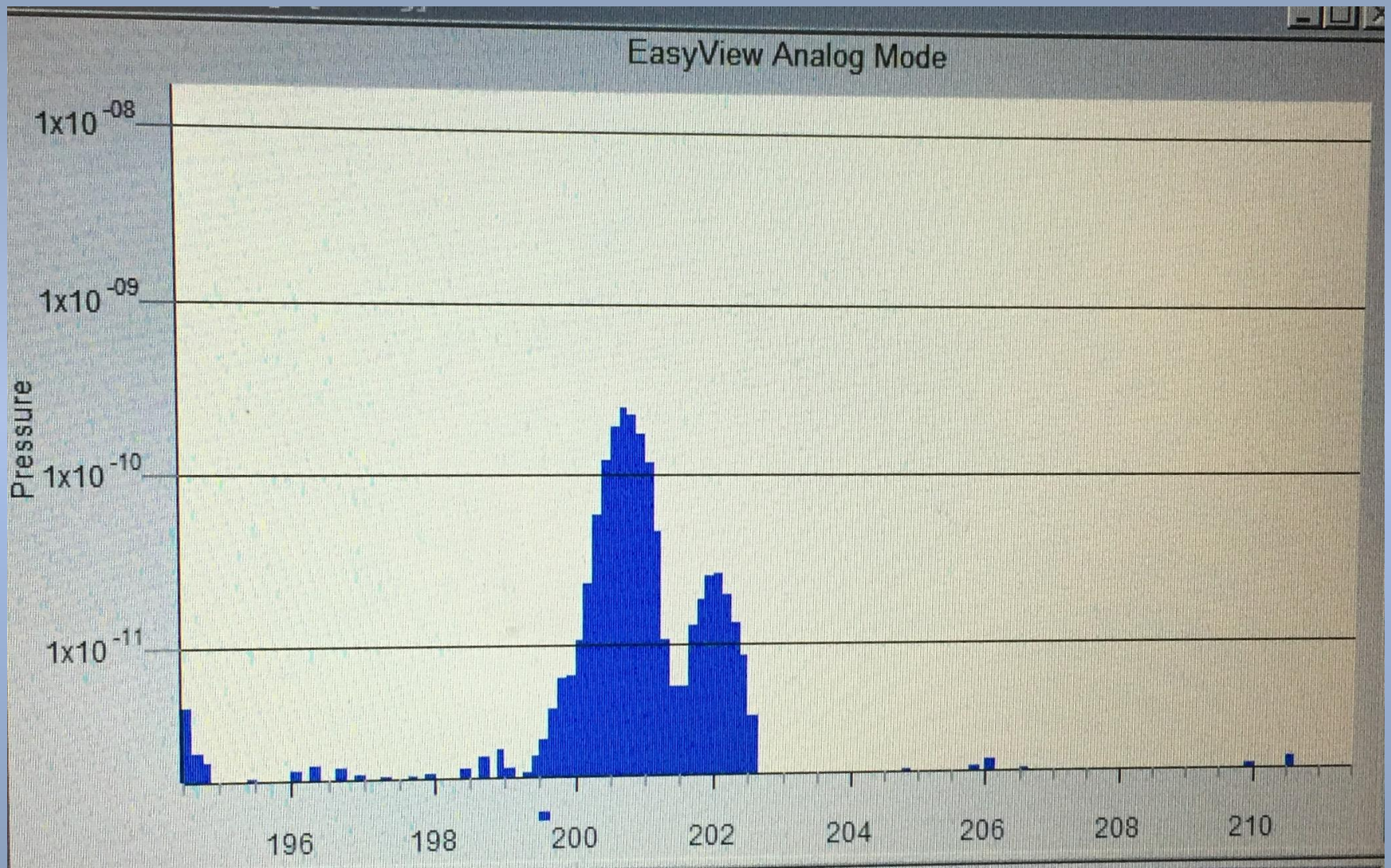
Gas system for analysis of helium in air samples as of 3/17/2017



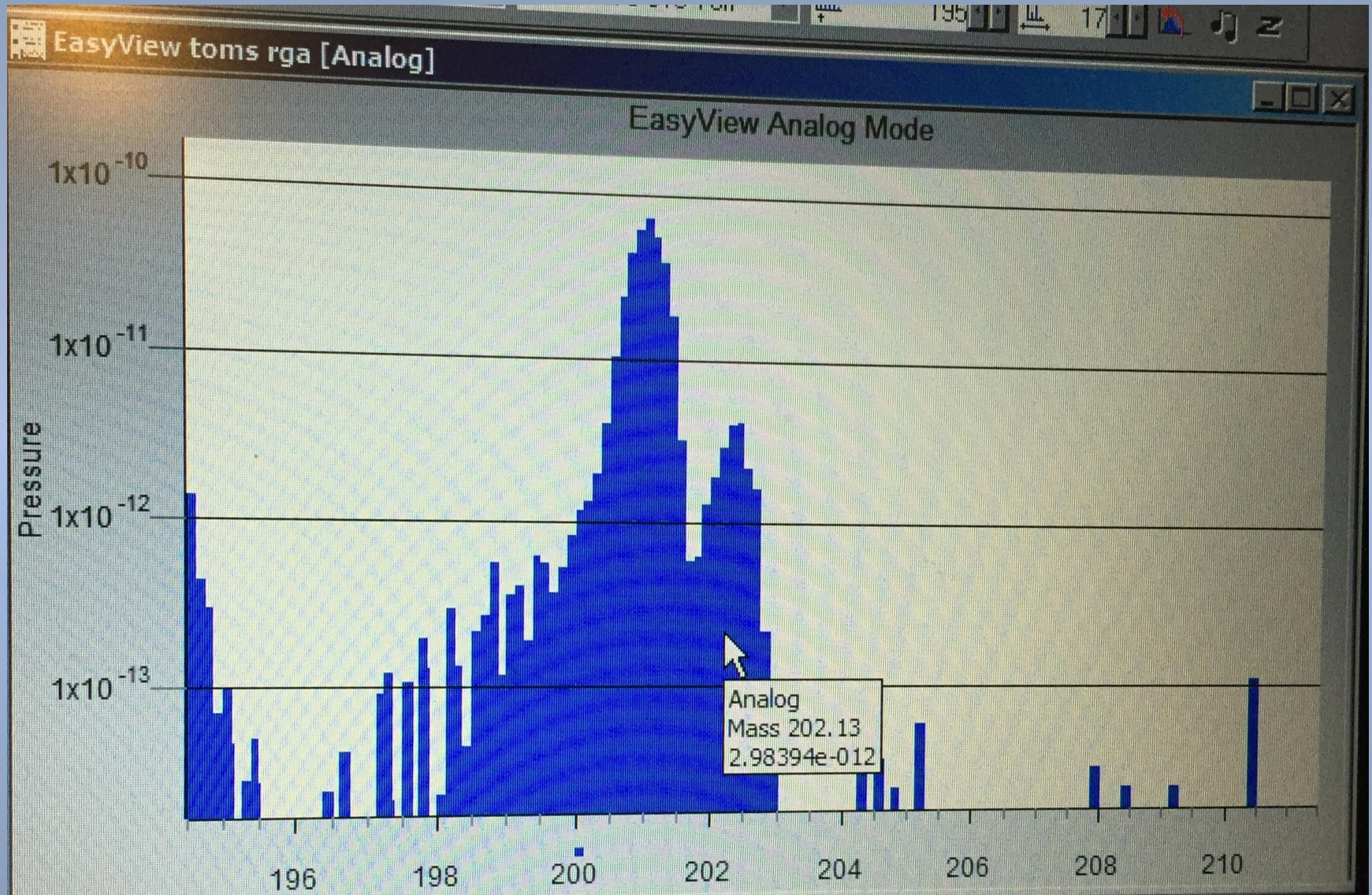
Deuterium Response using the MKS Microvision RGA



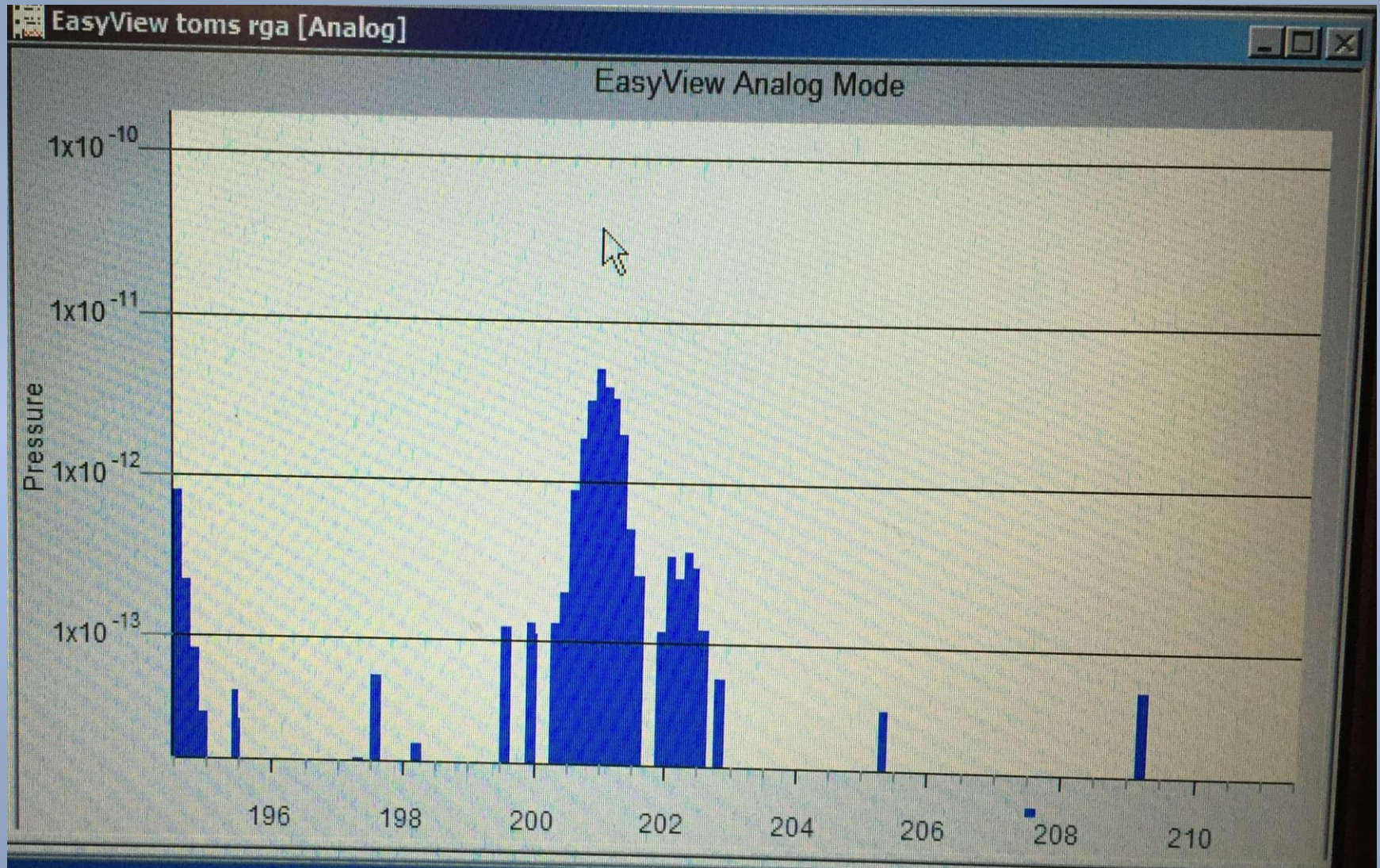
Mixture of He⁴ and D₂



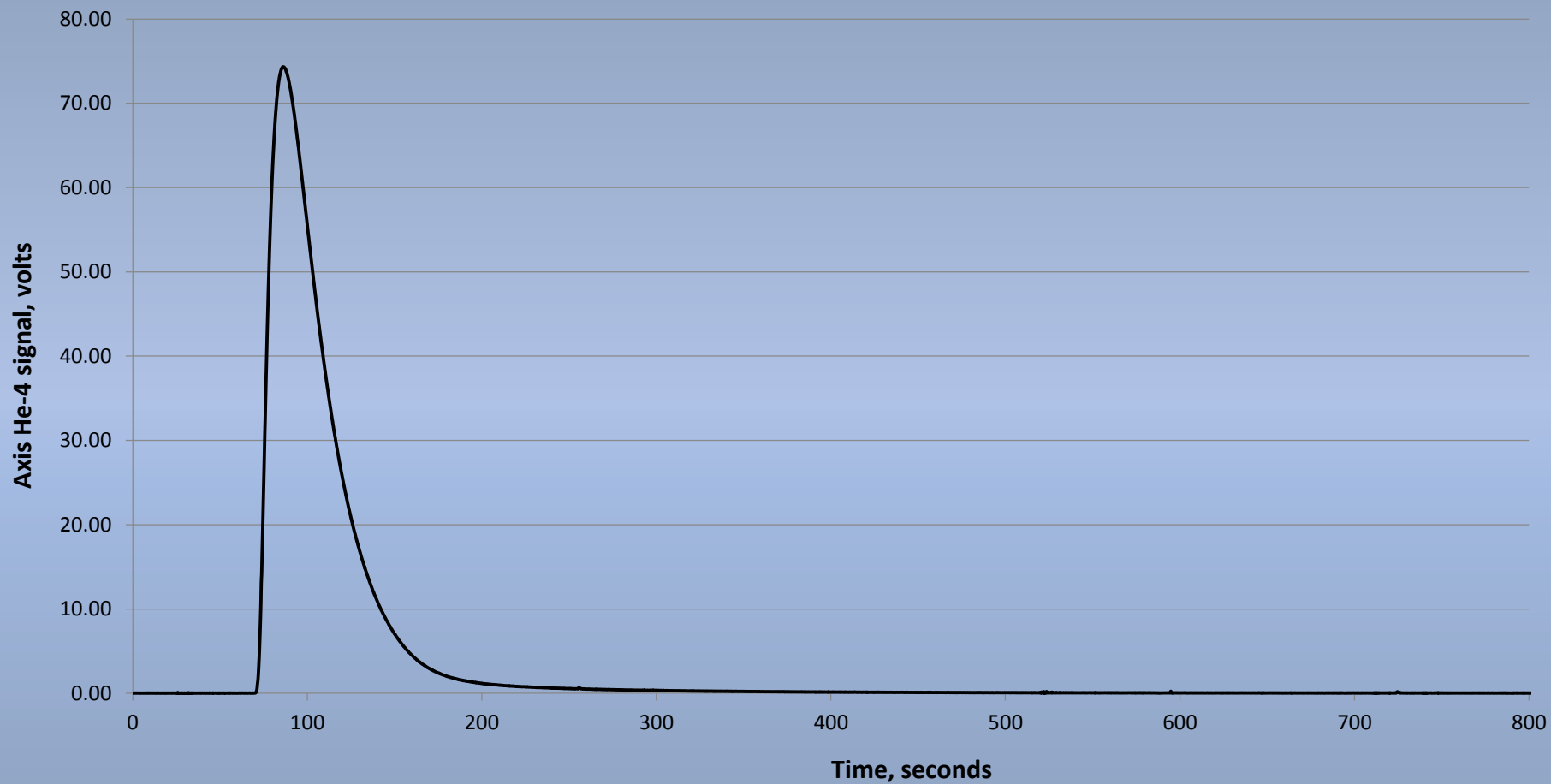
Same mixture at lower pressure showing low mass tailing



Again at even lower pressure



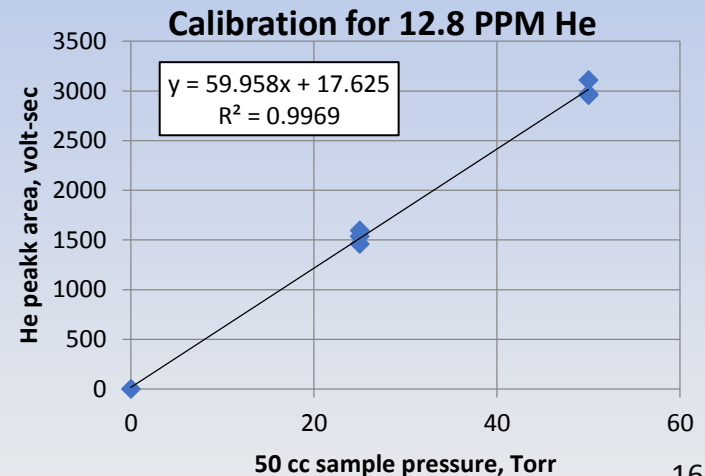
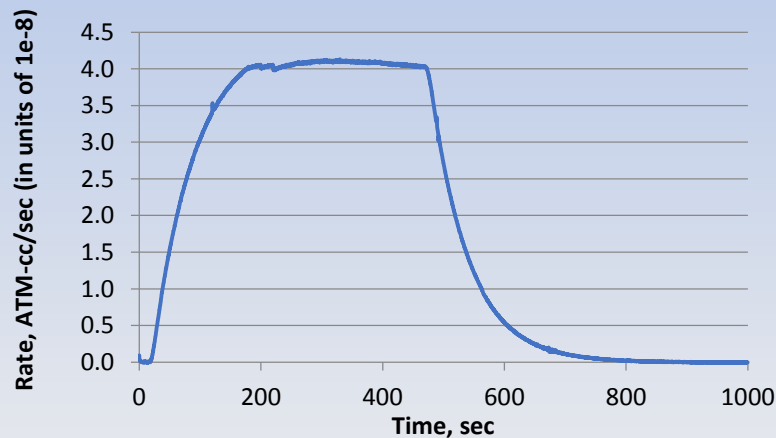
Run 127 50 cc @ 25 Torr 12.8 PPM He-4 (1250 Torr-cc)



Calibration

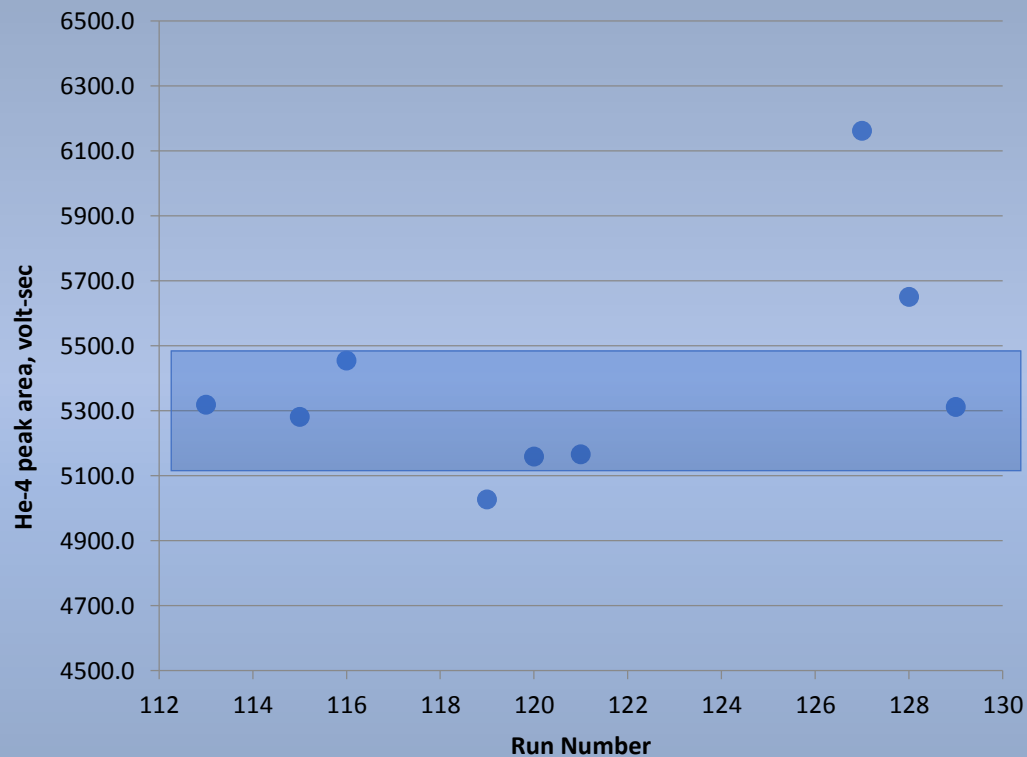
We have two methods to calibrate the system:

1. We can verify that the calibration of the helium leak detector is correct using a known calibrated leak. The integral, over time, of the measured sample “leak rate” gives the number of atm-cc of He in the sample. The ratio of this to the total number of atm-cc in the sample is the concentration of helium in the sample.
2. We can prepare known standards of helium that are similar in concentration to that expected in the samples and then analyze the sample and standard using the same volumes and pressures for both. The concentration of helium in the sample is given by the ratio of the integrated signal over time for the standard to that for the sample.

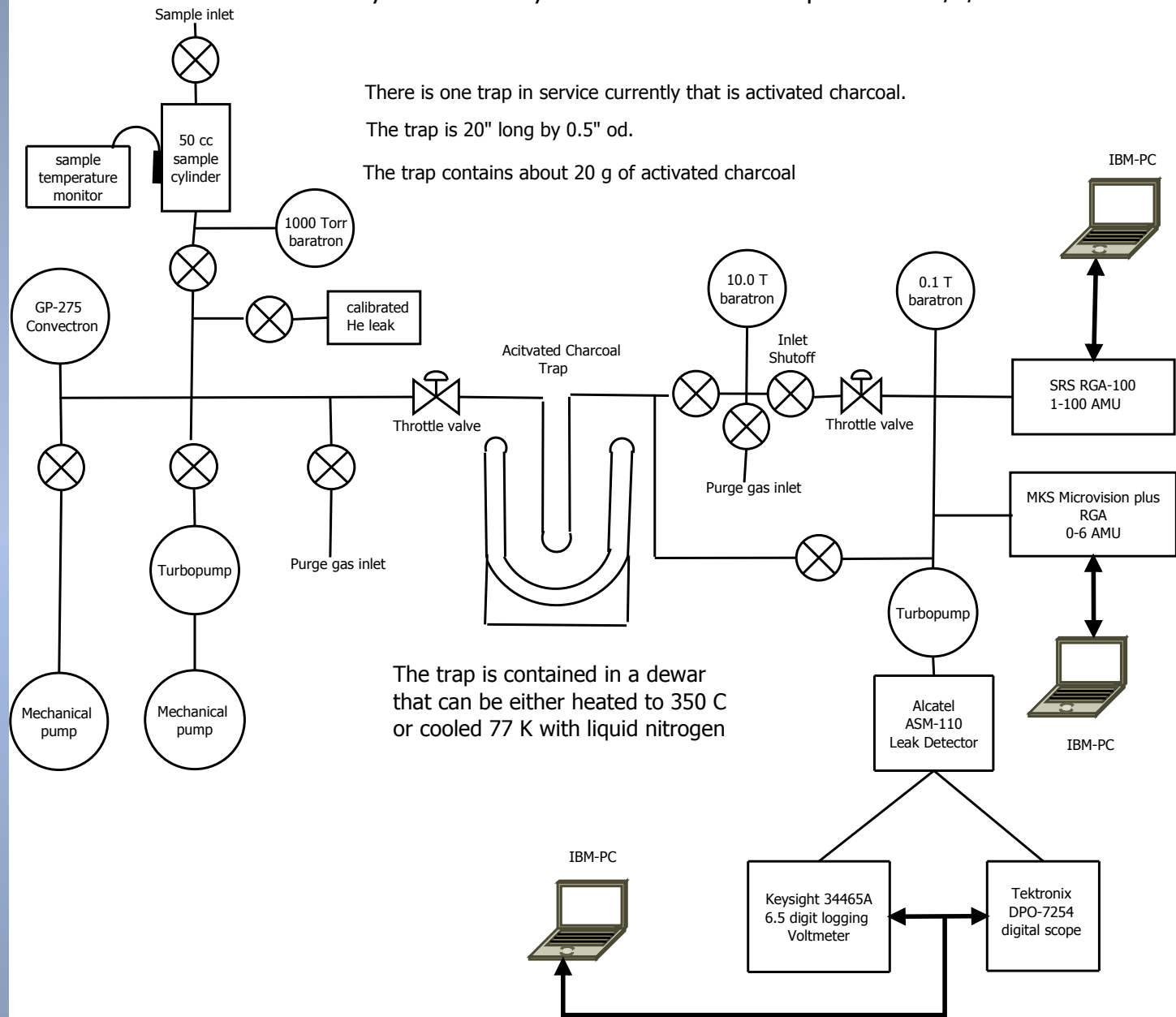


12.8 PPM std normalized to 50 cc @ 50 Torr

Date	pressure	area	norm area
3/25/17	25.00	2659.5	5319.0
3/29/17	25.00	2640.5	5281.0
3/31/17	25.01	2526.2	5454.4
4/2/17	24.96	2509.4	5026.8
4/7/17	25.01	2580.6	5159.2
4/7/17	25.03	2586.1	5166.0
4/29/17	25.00	3081.0	6162.0
4/30/17	25.01	2826.0	5650.5
4/30/17	25.00	2655.8	5311.6
avg			5296.1
stdev			192.7
rsd			3.6%
+1 stdev			5488.7
-1 stdev			5103.4



Gas system for analysis of helium in air samples as of 5/9/2017



Estimation of the total amount of helium and heat produced in the recent gas discharge experiments

- Gas Discharge system contains about 2400 cc of D_2 at 350 Torr, or 8.4×10^5 Torr-cc.
- The analyzed samples contain 2500 Torr-cc, or about 0.3% of the total.
- The last 5 samples analyzed ranged from 20 to 70 ppb He with a fairly large uncertainty.
- This would correspond to 0.017 to 0.059 Torr-cc of He. or about 5.5×10^{14} to 1.9×10^{15} helium atoms.
- If the reaction was $d+d \rightarrow He$, and the helium survived, the helium would represent 2100 to 7100 Joules of heat spread over 24 hours.
- This would correspond to 24-82 mW while the discharge is several watts.

Summary

- A system has been developed that can measure small amounts of helium in gas samples.
- Currently, we can measure the amount of helium in air at ambient levels (5.26 ppm) with an uncertainty of about 10% using a sample of 50 cc at 50 Torr pressure to make the measurement.
- The minimum detectable amount of helium is estimated to be about 3 ppb for the same size sample.
- The measurement requires about 1 hour to complete, but then requires several hours to warm and regenerate the traps to prepare for the next sample.
- We have run about 150 analyses to date with most of the samples being standards and blanks.
- We detect small amounts of helium in almost every gas sample we run.
- We may consider adding a getter in the future, although it does not seem to be necessary at this time.
- The most recent standard runs have a scatter of ~5% at the one standard deviation level.
- The the baseline measurement is equivalent to about 1×10^{-10} atm-cc/sec with an uncertainty of about 30%.
- This value integrated over 100 sec that the He signal is being measured corresponds to about 1×10^{-8} atm-cc or 7.6×10^{-6} torr-cc, equivalent to 3 ppb in our standard sample.

We would like to acknowledge the continued support and encouragement for this work by:

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Questions?

Estimate the amount of He in various air samples and the energy release if it came from fusion

Assuming the concentration of He in the air samples is 5.26 ppmv

Malcolm Fowler, 10/7/2015

volume cc	pressure			moles	moles of He	atoms of He	energy release, based on $D+D \rightarrow He$	
	PSIA	Torr	Atm				MeV	joules
		2585.0340						
10	50.0	14	3.40136	0.00139	7.31718E-09	4.40714E+15	1.05771E+17	16.9
350	3.9	200	0.26316	0.00377	1.98141E-08	1.19341E+16	2.86417E+17	45.9
1725	19.4	1001	1.31711	0.09292	4.88765E-07	2.94383E+17	7.06520E+18	1131.8
1725	12.3	636	0.83684	0.05904	3.10544E-07	1.87041E+17	4.48898E+18	719.1
50	12.3	636	0.83684	0.00171	9.00128E-09	5.42147E+15	1.30115E+17	20.8
100	12.3	636	0.83684	0.00342	1.80026E-08	1.08429E+16	2.60231E+17	41.7
150	12.3	636	0.83684	0.00513	2.70038E-08	1.62644E+16	3.90346E+17	62.5
check:	24451	14.7	760	1.00000	5.26002E-06	3.16811E+18	7.60347E+19	12180.8



DLS-20

Ultra high resolution quadrupole mass spectrometer featuring 20mm pole diameter – specifically designed for the analysis of hydrogen and Helium isotopes and light gases.

DLS 20. Showing He and D separation

